Natural Product Synthesis

Total Synthesis of Polyoxygenated Cembrenes**

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Dedicated to Professor Helmut Schwarz on the occasion of his 65th birthday

Numerous new biologically active and structurally interesting diterpenes containing a characteristic oxygenated 14-membered carbon ring system, which is derived formally from (+)-cembrene (1),^[1] have been isolated from terrestrial and marine sources over the last few years (Scheme 1).^[2] The



Scheme 1. (+)-Cembrene and oxygenated derivatives.

oxygenated cembrenes display amongst others anti-HIV activity, anti-inflammatory properties, and neuro- and cyto-toxicity.^[3] Sarcophytol A (**2**), isolated from soft coral, for example inhibits tumor growth and serves as the lead structure for the development of other pharmacologically active compounds.^[4]

It is therefore essential for ongoing investigations that efficient strategies are developed for the synthesis of oxygenated, especially the more complex polyoxygenated, cembrenes,^[5] which would allow general access to differently substituted compounds, and thereby make them available for testing. We describe here a stereoselective total synthetic access to the bicyclic polyoxygenated cembrene **3** with six stereogenic elements and four oxygen atoms. Compound **3** was structurally characterized as part of biosynthetic studies on polyoxygenated cembrenes occurring in Greek tobacco plants.^[6] In addition an epimer of compound **3** was prepared.

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- [**] This work was supported by the Deutsche Forschungsgemeinschaft and the Fonds der Chemischen Industrie. We thank BASF, Bayer-Schering, Evonik, Symrise, and Wacker for the provision of chemicals.
- Supporting information for this article is available on the WWW under http://www.angewandte.com or from the author.

Retrosynthetically, cembrene **3** may be broken down into the two almost equally sized building blocks **5** and **6** that can be coupled with one another by a nucleophilic epoxide opening (Scheme 2).^[7] The tertiary alcohol function in **5** can



Scheme 2. Retrosynthetic analysis of compound 3.

be constructed in a highly stereoselective domino multicomponent allylation^[8] of the aliphatic ketone **8**, whereas the dithiane **6** is accessible by a modified Myers alkylation.^[9] The 14-membered macrocycle is constructed by ring-closing metathesis.^[10] The benzyl ether **10** with d.r. = 95:5 is obtained by allylation of the ketone **8** with the silyl ether **9**^[8c] and allyl silane **7** under trifluoromethanesulfonic acid catalysis (Scheme 3). Reductive cleavage of the benzyl group with subsequent *p*-methoxybenzyl protection of the resulting tertiary alcohol and Sharpless dihydroxylation^[11] afforded the diol **11** with d.r. = 91:9, which was transformed into the epoxide **5**.^[12]

In the construction of the building block **6**, initially a diastereoselective α -alkylation of the pseudoephedrine amide **16** with the triflate **13** was carried out, followed by oxidation of the secondary alcohol function to give the ketoamide **17**, which could be enriched to d.r. = 99:1 by crystallization (Scheme 4).^[9a] Triflate **13** was synthesized in quantitative yield from **12** by reaction with Tf₂O, and amide **16** was accessible from **14** in 94% yield by acylation with the acid chloride **15**. Reductive amide cleavage of **17** with LDA and a borane–ammonia complex,^[9] and oxidation of the primary alcohol to the aldehyde with subsequent Wittig–Horner

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Scheme 3. a) 20 mol% TfOH, CH_2Cl_2 , $-196 \rightarrow -90$ °C; b) Li, DBBP, THF, $-78 \rightarrow -45$ °C, 71% over 2 steps; c) PMBOC(=NH)CCl_3, 7 mol % La(OTf)_3, toluene, RT, 77%; d) 1 mol% K_2OSO_2(OH)_4, 5 mol% (DHQD)_2Pyr, K_3[Fe(CN)_6], K_2CO_3, tBuOH/H_2O (1:1), 0°C; e) PivCl, pyridine, CH_2Cl_2 , 0°C \rightarrow RT, 64% over 2 steps; f) MsCl, NEt_3, cat. DMAP, CH_2Cl_2 , RT, 83%; g) K_2CO_3, MeOH, RT, quant. DBBP=4,4'-di*tert*-butylbiphenyl, (DHQD)_2Pyr=hydroquinidine-(2,5-diphenyl-4,6-pyrimidinediyl)diether, DMAP=4-(dimethylamino)pyridine, Ms = methanesulfonyl, Piv = pivalyl, PMB = 4-methoxybenzyl, THF = tetrahydrofuran, Tf = trifluoromethanesulfonyl.



Scheme 4. a) Tf_2O , CH_2Cl_2 , $-78 \rightarrow 0^{\circ}C$; b) NEt₃, THF, $0^{\circ}C$, 97%; c) LDA, LiCl, THF, $-78^{\circ}C \rightarrow RT$, then 13, $-78 \rightarrow -20^{\circ}C$, 83% over 2 steps; d) (COCl)₂, DMSO, CH_2Cl_2 , $-78^{\circ}C$, then NEt₃, RT, 79%; e) BH₃·NH₃, LDA, THF, $0^{\circ}C \rightarrow RT$, 89%; f) IBX, DMSO, MS 4 Å, RT, quant.; g) (EtO)₂P(O)CH₂CO₂Et, NaH, THF, $0^{\circ}C \rightarrow RT$, 89%; h) DIBAL-H, CH_2Cl_2 , $-78^{\circ}C$, quant.; i) MnO₂, CH_2Cl_2 , RT; j) HS(CH₂)₃SH, BF₃·Et₂O, Et₂O, $-40^{\circ}C$, 75% over 2 stages. DIBAL-H=diisobutylaluminum hydride, DMSO=dimethyl sulfoxide, IBX = 1-hydroxy-1,2-benziodoxol-3(1H)-one 1-oxide, LDA=lithium diisopropylamide, MS = molecular sieves.

olefination led to the α , β -unsaturated ester **18** in 79 % yield over three steps. Compound **18** was reduced to the allyl alcohol with DIBAL-H, then oxidized to the α , β -unsaturated aldehyde with MnO₂, and subsequently converted into the dithiane **6** with propane-1,3-dithiol.

The secondary alcohol **19** was then obtained in 92 % yield from the dithiane **6** and the epoxide **5** by nucleophilic epoxide opening^[7] and was converted into the primary alcohol **20** by acetylation and cleavage of the silyl protecting group (Scheme 5). Compound **20** was oxidized to the corresponding aldehyde in 88 % yield by using SO₃·pyridine under mild Parikh–Doering conditions^[13] and subsequently converted directly to the triene **21** by a Wittig carbonyl olefination in 89 % yield. Only decomposition occurred under Dess–Martin, Swern, and TPAP oxidation conditions, presumably owing to the sensitivity towards oxidation of the dithiane and *p*methoxybenzyl ether groups in **20**. The ring-closing metathesis was carried out with the Grubbs catalyst **22**^[14] (7 mol%) and provided the macrocycle **23** as a uniform *Z* diastereoisomer in an astonishingly good yield of 89%.

The orientation of the newly formed double bond and the relative configuration of the stereogenic centers in **23** were determined unambiguously by X-ray structure analysis.^[15] The absolute configurations of the stereogenic centers in **23** were derived from the known configurations of **10** and **17**.^[8,9a] Since the corresponding stereogenic centers in the product **23** have an *R* configuration at C-10 and an *S* configuration at C-17 as in **10** and **17**, respectively, it is also confirmed that no intermediate isomerization has occurred. The use of other catalysts for the metathesis, such as the Fürstner catalyst Neolyst M1,^[16] afforded complex mixtures of dimers. Interestingly, the protective groups of the hydroxy groups at C-8 and C-10 have a large influence upon the metathesis. No ring-closure could be achieved by using a triisopropylsilyl group at C-8, which is presumably attributable to steric reasons.

For the diastereofacial introduction of the methyl groups at C-1 of ketone 4 it was necessary to carry out a reprotection of the macrocycle 23 by using the sterically demanding triisopropylsilyl protecting group at C-13. For this purpose the acetyl group was removed under basic conditions, and the resulting secondary alcohol was converted over two steps to 24 with triisopropylsilyl triflate in 80% yield. Subsequent oxidative cleavage of the p-methoxybenzyl ether with dichloro-5,6-dicyano-p-benzoquinone and of the dithiane with bis(trifluoroacetoxy)iodobenzene^[17] led to the ketone 4 in 80% yield over two steps. Ketone 4 was epoxidized regioselectively with dimethyldioxirane^[18] leading to the formation of two diastereoisomers 25a and 25b in a ratio of 40:60. Under acidic conditions, the main diastereoisomer 25b underwent a nucleophilic intramolecular epoxide ring opening leading to 26, which contains a tetrahydrofuran moiety (Scheme 6).^[19]. A regioselective 1,2-addition of methyllithium to the carbonyl group in 26, which however in spite of the sterically demanding triisopropylsilyl group at the neighboring position took place with only low diastereoselectivity, led to the diols 27 and 28 in a ratio of 40:60. As expected, the main product 28 was formed by an attack anti to the substituent at C-3.^[20] Fortunately, it was possible to separate the two diastereoisomers 27 and 28. The final steps in

Angew. Chem. Int. Ed. 2008, 47, 5246-5249

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Scheme 5. a) 6. *n*BuLi, THF, HMPA, $-78 \rightarrow -55$ °C, then 5. THF, -55 °C \rightarrow RT, 92%; b) Ac₂O, pyridine, cat. DMAP, RT, quant.; c) TBAF·3 H₂O, THF, RT, quant.; d) SO₃·pyridine, NEt₃, DMSO, RT, 88%; e) KHMDS, Ph₃PMeBr, THF, -78 °C, then 0 °C, 89%; f) 7 mol % 22, CH₂Cl₂, 40 °C, 89%; g) NaOMe, MeOH, Et₂O, 40 °C, 89%; h) TIPSOTf, 2,6-lutidine, CH₂Cl₂, 0 °C, 90%; i) DDQ, CH₂Cl₂, buffer pH 7, 0 °C, 81%; j) (CF₃CO₂)₂IPh, MeOH/THF/H₂O (10:5:1), RT, 99%. DDQ = 2,3-dichloro-5,6-dicyano-*p*-benzoquinone, HMPA = hexamethylphosphoric triamide, KHMDS = potassium bis(trimethylsilyl)amide, TBAF = tetra-*n*-butylammonium fluoride, TIPS = triisopropylsilyl.



Scheme 6. a) DMDO, acetone, CH_2Cl_2 , -85 °C; b) 10 mol % CSA, CH_2Cl_2 , $-78 \rightarrow 0$ °C, 49% for 26 over 2 steps via 25 b, 40% for 25 a; c) MeLi, Et_2O , -110 °C \rightarrow RT, 30% for 27, 45% for 28; d) TBAF·3 H₂O, THF, RT; e) 10 mol % TPAP, NMO, MS 4 Å, CH_2Cl_2 , RT, 74% for 29 over 2 steps, 77% for 3 over 2 steps. DMDO = dimethyldioxirane, TPAP = tetra-*n*-butylammonium perruthenate, NMO = *N*-methylmorpholine *N*-oxide.

the synthesis of the polyoxygenated cembrene **3** comprised cleavage of the TIPS group and oxidation of the resulting secondary alcohol to the ketone in 73% yield over two steps by using tetra-*n*-butylammonium perruthenate and *N*-meth-ylmorpholine-*N*-oxide. The diastereoisomer **29** was obtained analogously from **27**.

Since NOE investigations on the molecules to determine the relative configuration of the newly formed stereogenic centers were generally poorly predictive, an X-ray structure analysis was carried out on the final product 3.^[21]. Compound 3 was synthesized in an overall yield of 2.7% over 24 steps; this corresponds to an average yield of 86% for each step. The analytical data of the synthetic material is identical to that for the polyoxygenated cembrene described by Wahlberg et al.^[6,22]. In conclusion, by using ringclosing metathesis for the construction of the 14-membered carbon ring, a stereoselective Tietze allylation and a stereoselective Myers alkylation, we have developed a new and efficient route to complex polyoxygenated cembrenes, which offers a high degree of flexibility in respect of stereochemical variations and functionalization.

Received: February 7, 2008 Published online: June 4, 2008

Keywords: allylation · cembrenes · epoxidation · metathesis · terpenoids

- [1] W. G. Dauben, W. E. Thiessen, P. R. Resnick, J. Am. Chem. Soc. 1962, 84, 2015– 2016.
- [2] a) J. R. Hanson, Nat. Prod. Rep. 2006, 23, 875–885;
 b) J. R. Hanson, Nat. Prod. Rep. 2005, 22, 594–602; c) I. Wahlberg, C. R. Enzell, Nat. Prod. Rep. 1987, 4, 237–276.
- [3] a) M. I. Nieto, N. González, J. Rodríguez, R. G. Kerr, C. Jiménez, *Tetrahedron* 2006, 62, 11747–11754; b) P. Radhika, P. R. Rao, J. Archana, N. K. Rao, *Biol. Pharm. Bull.* 2005, 28, 1311–1313; c) A. D. Rodríguez, I. C. Piña, A. L. Acosta, C. L.

Barnes, *Tetrahedron* **2001**, *57*, 93–107; d) M. A. Rashid, K. R. Gustafson, M. R. Boyd, *J. Nat. Prod.* **2000**, *63*, 531–533.

[4] a) I. Katsuyama, H. Fahmy, J. K. Zjawiony, S. I. Khalifa, R. W. Kilada, T. Konoshima. M. Takasaki, H. Tokuda, *J. Nat. Prod.* 2002, 65, 1809–1814; b) T. Zhang, Z. Liu, Y. Li, *Synthesis* 2001,

393–398; c) J. Lan, Z. Liu, H. Yuan, L. Peng, W.-D. Z. Li, Y. Li, Y. Li, A. S. C. Chan, *Tetrahedron Lett.* **2000**, *41*, 2181–2184.

- [5] M. A. Tius, Chem. Rev. 1988, 88, 719-732.
- [6] I. Wahlberg, I. Forsblom, C. Vogt, A.-M. Eklund, T. Nishida, C. R. Enzell, J.-E. Berg, J. Org. Chem. 1985, 50, 4527-4538.
- [7] a) A. B. Smith III, S. M. Pitram, A. M. Boldi, M. J. Gaunt, C. Sfouggatakis, W. H. Moser, *J. Am. Chem. Soc.* 2003, *125*, 14435–14445; b) L. F. Tietze, H. Geissler, J. A. Gewert, U. Jakobi, *Synlett* 1994, 511–512; c) B.-T. Gröbel, D. Seebach, *Synthesis* 1977, 357–402.
- [8] a) L. F. Tietze, T. Kinzel, S. Schmatz, J. Am. Chem. Soc. 2008, 130, 4386-4395; b) L. F. Tietze, T. Kinzel, S. Schmatz, J. Am. Chem. Soc. 2006, 128, 11483-11495; c) L. F. Tietze, S. Hölsken, J. Adrio, T. Kinzel, C. Wegner, Synthesis 2004, 2236-2239.
- [9] a) L. F. Tietze, C. Raith, C. C. Brazel, S. Hölsken, *Synthesis* 2008, 229–236; b) A. G. Myers, B. H. Yang, H. Chen, L. McKinstry, D. J. Kopecky, J. L. Gleason, *J. Am. Chem. Soc.* 1997, *119*, 6496–6511.
- [10] A. Gradillas, J. Pérez-Castells, Angew. Chem. 2006, 118, 6232– 6247; Angew. Chem. Int. Ed. 2006, 45, 6086–6101.
- [11] a) Q.-H. Fan, Y.-M. Li, A. S. C. Chan, *Chem. Rev.* 2002, 102, 3385–3466; b) H. C. Kolb, M. S. VanNieuwenhze, K. B. Sharpless, *Chem. Rev.* 1994, 94, 2483–2547.
- [12] a) A. B. Smith III, D.-S. Kim, J. Org. Chem. 2006, 71, 2547–2557;
 b) L. A. Paquette, J. Chang, Z. Liu, J. Org. Chem. 2004, 69, 6441–6448.
- [13] J. R. Parikh, W. v. E. Doering, J. Am. Chem. Soc. 1967, 89, 5505 5507.

- [14] M. Scholl, S. Ding, C. W. Lee, R. H. Grubbs, Org. Lett. 1999, 1, 953–956.
- [15] Single-crystal structure analysis of compound **23**: $C_{32}H_{48}O_4S_2$, monoclinic, space group *P*2(1), *T*=133(2) K, *a*=11.0076(5), *b* = 9.2718(3), *c*=30.9119(16) Å, *V*=3133.3(2) Å³, *Z*=2, ρ = 1.189 Mg m⁻³, *R*₁=0.0582, *wR*₂=0.0662.^[21]
- [16] a) A. Fürstner, O. Guth, A. Düffels, G. Seidel, M. Liebl, B. Gabor, R. Mynott, *Chem. Eur. J.* 2001, *7*, 4811–4820; b) A. Fürstner, *Angew. Chem.* 2000, *112*, 3140–3172; *Angew. Chem. Int. Ed.* 2000, *39*, 3012–3043.
- [17] G. Stork, K. Zhao, Tetrahedron Lett. 1989, 30, 287-290.
- [18] M. Gibert, M. Ferrer, F. Sánchez-Baeza, A. Messeguer, *Tetrahedron* **1997**, *53*, 8643–8650.
- [19] A. D. Rodríguez, J. J. Soto, C. L. Barnes, J. Org. Chem. 2000, 65, 7700-7702.
- [20] P. C. Astles, E. J. Thomas, J. Chem. Soc. Perkin Trans. 1 1997, 845–856.
- [21] Single-crystal structure analysis of compound **3**: $C_{20}H_{34}O_4$, monoclinic, space group P2(1), T=133(2) K, a=6.3368(5), b=13.5693(10), c=11.4017(8) Å, V=971.21(13) Å³, Z=2, $\rho=1.157$ Mg m⁻³, $R_1=0.0589$, $wR_2=0.1008$. CCDC-676776 (23) and CCDC-676777 (3) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.
- [22] Rotation of the synthesized compound **3**: $[\alpha]_D^{20} = +39.4^\circ$ (c = 0.35, chloroform); literature value: $[\alpha]_D^{20} = +38^\circ$ (c = 0.18, chloroform).^[6]