New Esters of 4-Amino-5-chloro-2-methoxybenzoic Acid as Potent Agonists and Antagonists for 5-HT₄ Receptors

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A number of benzoates derived from 4-amino-5-chloro-2-methoxybenzoic acid and substituted 1-piperidineethanol were synthesized and found to be potent 5-HT₄ receptor agonists in the electrically-stimulated myenteric plexus and longitudinal muscle of the guinea pig ileum and the rat esophagus muscle. Monosubstitution of the piperidine ring with Me, OH, NH-Ac, or $CONH_2$ groups gave compounds equipotent to **7a** (ML 10302), a 5-HT₄ receptor agonist previously reported to have nanomolar affinity. 7a,k were as potent as serotonin (5-HT) but had maximal responses which were only 60-80% of that of 5-HT, suggesting a partial agonist profile for these compounds. Binding assays were performed with [3H]GR 113808 in the rat striatum, and several of these compounds were found to have nanomolar affinity for 5-HT₄ receptors (**7a**, $K_i = 1.07 \pm 0.5$ nM; **7k**, $K_i = 1.0 \pm 0.3$ nM). The introduction of two methyl groups on the piperidine ring brought about a dramatic change in the pharmacological profile of 2-[(cis- and trans-3,5-dimethylpiperidinyl)ethyl]-4-amino-5-chloro-2-methoxybenzoate, 7g,h. **7g** ($K_i = 0.26 \pm 0.06$ nM) inhibited the relaxant action of 5-HT in the rat esophagus muscle with a pA_2 value of 8.6. The advantage of the ester function was demonstrated by comparing the activity of several such compounds at 5-HT₄ receptors with those of the corresponding amidic derivatives. This difference was less marked when the basic moiety was sterically constrained as in the quinuclidine and tropane moieties. Structural analyses of 7a,g were performed by determining their X-ray crystal structures and by molecular modeling (SYBYL). A relatively limited number of minimum energy conformers was found for both compounds. They were characterized by the *cis* folded conformation of the ethyl chain and by the orientation of the lone pair of the nitrogen atom pointing out of the molecule as seen in conformationallyconstrained benzamides such as zacopride and renzapride. A hypothetical model for the 5- HT_4 receptor with two sites for the binding of agonist and antagonist molecules was proposed.

Recent data on the cloning of the 5-HT₄ receptor^{1,2} showed that it is a member of the G-protein-coupled receptor family and confirmed the early report of Dumuis³ on the existence of a serotoninergic receptor linked positively to adenylyl cyclase in the central nervous system (CNS). This receptor has been shown to be similar to that discovered by Clarke^{4a} in the gastrointestinal (GI) system where it was shown to be responsible for the stimulation of gut motility by 5-HT. Stimulation of these receptors provided a mechanism to explain the pharmacological action of gastrokinetic drugs such as cisapride, renzapride, and zacopride which enhance GI propulsion and can correct intestinal motility disorders.⁵ In the isolated guinea pig ileum, Clarke^{4a} showed that 5-HT₄ receptors were implicated in the first phase of the biphasic concentration curve to 5-HT, while 5-HT₃ receptors mediated the second phase. This finding explained the initial confusion surrounding the role of 5-HT₃ receptor antagonists^{4b} in the GI tract because a number of these compounds were also demonstrated⁶ to be 5-HT₄ receptor agonists. It was shown that the action of these receptors was mediated by the cholinergic system and that they were neuronally located, either directly on cholinergic nerves or upstream.⁷ To date, 5-HT₄ receptors have been found in several peripheral tissues: rat esophagus,⁴ colon,⁸ and

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urinary bladder⁹ where they are implicated in the contraction or relaxation of smooth muscles, and they have also been identified in porcine¹⁰ and human¹¹ atria where they mediate an increase in contractile force. Moreover, 5-HT₄ receptors have been found in rat, guinea pig, and human brain¹² with high receptor concentrations in the striatum and substantia nigra. Considerable progress has been made in the localization of 5-HT₄ receptors and in the study of their functions with the development of potent antagonists such as SDZ 205-557 (1),¹³ SB 204070 (2),¹⁴ SB 207710 (3),^{15a} GR 113808 (4),16 and carbazimidamide derivatives,17c in particular which have been used as radiolabeled ligands.^{12a,15b,c} More recently, SB 207266 (5), an amidic derivative of a tricyclic indole moiety, was reported as the first potent, selective 5-HT₄ receptor antagonist active orally.^{18a}

Thus, the mechanism of action of the gastric prokinetic drugs has been determined, and a number of them have been found to be 5-HT₄ receptor agonists.⁶ Many of these compounds are members of the generic benzamide family, derived from metoclopramide,¹⁹ a drug well-known for gastric prokinetic activity. These compounds, such as renzapride,²⁰ BRL 24682 (4-amino-5-chloro-2-methoxy-*N*-(*endo*-8-methylazabicyclo[3.2.1]octan-3-yl)benzamide),²¹ and zacopride,²² are amides of 4-amino-5-chloro-2-methoxybenzoic acid and possess a relatively rigid basic structure. These structural characteristics are in common with those of the 5-HT₃

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Chart 1



^{*a*} (a) 1,2-Dibromoethane, THF, reflux, DBU; (b) amine, K_2CO_3 , DMF, rt or 50 °C (method A) or amine, toluene, reflux (method B); (c) trityl chloride, CH₂Cl₂, pyridine; (d) CDI, THF; (e) alcohol, DBU, THF, reflux.

receptor antagonists and explain why almost all of these compounds are active at both receptor types. However, it has been clearly demonstrated that 5-HT₃ receptor antagonist activity is not implicated in the gastrokinetic effect of these drugs as several selective 5-HT₃ receptor antagonists are devoid of gastrokinetic properties.²³ The search for selective 5-HT₄ receptor agonists equipotent to serotonin is an important goal to develop new compounds to correct intestinal motility disorders.²⁴ Moreover, the presence of 5-HT₄ receptors which are supposed to facilitate the release of acetylcholine has been shown in the CNS, and thus 5-HT₄ receptor agonists may have a role in improving cognitive function.^{12b,25}

In recent years, several reports have described new compounds with 5-HT₄ receptor agonist activity possessing better selectivity than the previous compounds. Thus SC-53116 (6),²⁶ the quaternized butyl derivative of renzapride,²⁷ and a macrocyclic compound²⁸ were the first reported benzamides which surpassed the selectivity of the other benzamides for this receptor. ML 10302 (7a), a compound structurally closely related to 1, was also shown by us to be a selective 5-HT₄ receptor agonist²⁹ with nanomolar affinity in the electricallyevoked contraction model in the guinea pig ileum, while 5-HT₄ receptor antagonist properties have been reported previously.^{30,31} More recently, several ketone derivatives³² and some potent carbazimidamines^{17a,b} were reported as selective and metabolically-stable 5-HT₄ receptor agonists.

In our early approach to the design of new 5-HT₄ receptor agonists or antagonists, we were interested in

the dramatic variation of the pharmacological activity observed with 1 compared to metoclopramide, due only to the substitution of the amidic function by an ester function. It is well-known $^{\rm 33}$ that the amidic function of the benzamides is implicated in an intramolecular hydrogen bond with the *o*-methoxy group, and it was considered worthwhile to examine the influence of this structural property using different reference benzamides. Preliminary reports^{29a,34} from our laboratory demonstrated that the affinities of the corresponding esters for 5-HT₃ receptors were little affected with regard to those of the amidic derivatives, while an increase in affinity for 5-HT₄ receptors was generally observed. Moreover, it seemed that the magnitude of the variation was related to the flexibility of the basic framework.

We present, herein, structural variations³⁵ carried out with compounds **7** and **8** and the surprising data concerning the development of potent antagonist molecules designed using small structural variations on the piperidine ring. Some comparisons between the biological activities of the esters and the corresponding amides (compounds **9a**-**h**) are also presented.

Chemistry

Compounds **7** and **8** were synthesized according to the pathway described in Scheme 1. 4-Amino-5-chloro-2-methoxybenzoic acid was transformed to the 2-bromo-ethyl ester **10** by the reaction between the acid and 1,2-dibromoethane in THF in the presence of DBU (1,8-diazabicyclo[5.4.0]undec-7-ene). Compound **10** was then condensed with appropriate piperidines or cyclic amines

Scheme 2^a



^a (a) LiAlH₄, THF; (b) CH₂Cl₂, MnO₂, 24 h, rt; (c) Mg, THF, 1-(3-chloropropyl)piperidine, 60 °C, 24 h; (d) CH₂Cl₂, MnO₂, 40–50 °C, 7 days; (e) acetone, HCl, rt.

in DMF in the presence of K_2CO_3 (method A) or in toluene at reflux (method B) to provide compounds 7a-s and 8a-g, respectively. The amines were commercially available or prepared by previously-described methods. The esters derived from tropine (compound **9e**), (*R*)- and (S)-3-quinuclidinol (compounds 9c,d), and 2-(dimethylamino)ethanol (compound 9a) were prepared from 5-chloro-2-methoxy-4-(tritylamino)benzoic acid (11) (Scheme 1) which was synthesized by the reaction between trityl chloride and the acid in the presence of pyridine. The acid obtained was activated with 1,1'carbonyldiimidazole in THF to give the imidazolide which was isolated and reacted with alcohol by heating in THF in the presence of DBU to provide compounds **9a**,**c**–**e**. The derivatives of 3,5-dimethylpiperidine were obtained as a 4:1 mixture of the cis (compound 7g) and trans (compound 7h) isomers which were easily separated by column chromatography and characterized by their ¹H NMR spectra. The (+)- and (-)-3-hydroxypiperidine derivatives (compounds 7k,l) were synthesized from (+)- and (-)-3-hydroxypiperidine which was resolved according to a process already reported.³⁶

The amides reported in Table 3 (compounds **9b**,**f**–**h**) were synthesized from 4-amino-5-chloro-2-methoxybenzoic acid³⁷ using the mixed anhydride method. In the course of this work, it was deemed worthwhile to prepare the bioisosteric ketone **14** to compare its biological activity with that of **7a**. The synthesis of **14** was carried out according to the method presented in Scheme 2: the tritylated acid **11** was reduced to the corresponding alcohol and oxidized by MnO_2 to the aldehyde **12**. Condensation with 2-(1-piperidinyl)ethylmagnesium chloride provided the alcohol **13** which was oxidized by MnO_2 to the ketone **14**.

Biological Methods

The assessment of the affinity of the compounds for 5-HT₄ receptors was performed using radioligand binding assays with [3H]GR 113808 and rat striatal membranes.38 The existence of 5-HT₄ receptors has been clearly demonstrated in the CNS, and there are data to show that the 5-HT₄ receptors in the brain are identical with those in the periphery, in particular, those in the intestine.²⁵ Consequently, binding assays in the brain constitute a valuable test to select compounds which will be active in the GI tract. Previously, several 5-HT₄ receptor agonists and antagonists have also been described as potent 5-HT₃ receptor antagonists, and consequently, it was mandatory to evaluate the selectivity for both receptors. For this purpose, the most active compounds selected as ligands for 5-HT₄ receptors were evaluated in 5-HT₃ receptor binding assays using [³H]-BRL 43694³⁹ and rat posterior cortex membranes.

The facilitatory role of 5-HT₄ receptors in the peristaltic reflex was demonstrated by Craig and Clarke⁴ who found that both 5-HT₃ and 5-HT₄ receptor agonists induced peristalsis in the guinea pig isolated ileum maintained at a subthreshold intraluminal pressure. The 5-HT₄ receptor agonist activity of tested compounds was measured in the electrically-stimulated myenteric plexus and longitudinal muscle of the guinea pig ileum, and it was assessed as the concentration which gave a 50% increase in the response to electrical stimulation (EC₅₀, nM) with regard to its maximum. Antagonist activity (IC₅₀, nM) was calculated as the concentration which produced a 50% reduction of the 5-HT-induced contractions.

5-HT₄ receptors are also present in the rat esophagus where they mediate the muscle relaxation.⁴⁰ Several compounds were evaluated for their ability to relax carbachol-contracted rat esophageal muscularis mucosae. 5-HT₄ receptor agonist activity (EC₅₀) and the pA₂ value of several antagonist compounds were evaluated from the concentration–effect curves according to previously-reported methods.⁴¹

Biological Results and Discussion

The results of the compounds tested are reported in Tables 1–3. The values for 5-HT₄ receptor activation of the compounds were compared to those of the reference compounds: BIMU 8 (*endo-N*-(8-methyl-8-azabicyclo[3.2.1]oct-3-yl)-2,3-dihydro-2-oxo-3-(prop-2-yl)-1*H*-benzimidazole-1-carboxamide),^{3b} **1**, **3**, and RS 23597-190 (**15**).⁴²

The data from the binding assays for the piperidine derivatives 7 presented in Table 1 highlighted their potent affinity for 5-HT₄ receptors since a number of compounds, regardless of the nature of the substituent on the heterocycle, possessed nanomolar affinity for these receptors and were much more potent than the previously-described amides derived from 4-amino-5chloro-2-methoxybenzoic acid, such as zacopride, renzapride, and 6. The structural differences to the benzamide references compounds reside in the presence of the ester function and in the totally flexible basic moieties. Compound 7a (ML 10302), the 5-HT₄ receptor agonist activity of which was reported previously,²⁹ is structurally closely related to 14, which has been described as a potent 5-HT₄ receptor antagonist.⁴² However, **7a** was clearly more potent in the binding assays, indicating that the distance between the benzene ring and the basic nitrogen is a crucial structural parameter in this class of compounds for recognition by the 5-HT₄ receptor. Examination of the results in Table 1 showed that affinity for this receptor is unrelated to the nature of the substituent in the monosubstituted piperidine because groups as different as Me, OH, NHAc, CONH₂, and Bz on the 4 position gave compounds **7b**,**i**,**m**–**o** which were approximately equipotent. The affinity was not much affected by the substituent position (compounds 7c,d,j); on the other hand, modification of the basicity of the nitrogen atom (compound

Table 1. Pharmacological Activity of the Benzoates 7 at 5-HT₄ Receptors



	\mathbf{R}_1	\mathbf{R}_2	Х	binding assays <i>K</i> _i , nM, ^a 5-HT4	functional activity at 5-HT ₄ receptors	
compd					EC_{50} , nM^b	IC ₅₀ , nM^c
7a	Н	Н	CH_2	1.07 ± 0.5	4 [3.7-4.3] (80%)	$\mathbf{n}\mathbf{e}^{d}$
7b	4-Me	Н	CH_2	4.08 ± 0.5	3 [2-5] (57%)	ne
7c	3-Me	Н	CH_2	0.93 ± 0.05	15 [9-23] (54%)	ne
7d	2-Me	Н	CH_2	7.8 ± 0.6	6 [5-8] (82%)	ne
7e	2-Me	6-Me	CH_2	39.5 ± 7.2	NT^e	NT
7f	3-Me	3-Me	CH_2	1.72 ± 0.33	I ^f	150
7g	3-Me	5-Me, <i>cis</i>	CH_2	0.26 ± 0.06	Ι	11 [9-14]
7g 7h	3-Me	5-Me, trans	CH_2	2.32 ± 0.6	Ι	13 [8-16]
7i	4-OH	Н	CH_2	1.42 ± 0.4	14 [13-17] (72%)	ne
7j 7k	3-OH (±)	Н	CH_2	4.8 ± 0.5	5 [4-7] (64%)	ne
7ĸ	3-OH (+)	Н	CH_2	1 ± 0.3	4 [3-5] (62%)	
71	3-OH (-)	Н	CH_2	2 ± 0.2	10 [8-13] (61%)	
7m	4-NHAc	Н	CH_2	1.05 ± 0.36	13 [10-17] (59%)	ne
7n	4-CONH2	Н	CH_2	1.43 ± 0.32	19 [16-24] (65%)	ne
70	4-Bz	Н	CH_2	7.64 ± 0.38	I	100 [75-150]
7р	Н	Н	0	35.6 ± 3.1	113 [87-146] (53%)	I
7 q	3-Me	5-Me, <i>cis</i>	0	6.28 ± 0.32	I	Ι
7r	3-Me	5-Me, trans	0	NT	Ι	300
7s	Н	Н	N-Ph	NT	Ι	100 [70-165]
5-HT				187 ± 33	5 [3-8] (100%)	Ι
BIMU 8				257 ± 43	43 [29-63] (86%)	Ι
1 (SDZ 205557)				5.5 ± 0.7	I	77 [55-130]
4 (GR 113808)				0.15 ± 0.03	I	13 [9-18]
15 (RS 23597-190)				$\textbf{28.1} \pm \textbf{6.9}$	30 [17-49] (52%)	ne

^{*a*} [³H]GR 113808 was used as the radioligand in rat striatal membranes, nonspecific binding was determined with **7a** (10 μ M), and $K_i \pm$ SEM values were determined from the Cheng–Prussof equation. ^{*b*} The agonist activity was assessed as the concentration which gave a 50% increase in the response to electrical stimulation (EC₅₀) with regard to the maximum in the guinea pig ileum. The activity is expressed as percent of the maximum 5-HT response; 95% confidence limits are in brackets. ^{*c*} The antagonist activity (IC₅₀) was calculated as the concentration which produced a 50% reduction in the response to 5-HT in the guinea pig ileum; 95% confidence limits are in brackets. ^{*d*} ne = not capable of evaluation. ^{*e*} NT = not tested. ^{*f*} I = inactive up to 10⁻⁵ M.

7p) produced a clear decrease in affinity. Introduction of an additional substitution such as a methyl group had a favorable effect with the *cis*-3,5-dimethylpiperidine derivative (compound **7g**), while the *trans*-2,6-dimethyl-substituted compound **7e** was less active.

In the isolated guinea pig ileum preparation, the monosubstituted piperidine derivatives (compounds 7a**d**,**i**–**n**,**p**) displayed an agonist profile. They were equipotent to 5-HT and more potent than the reference compounds such as BIMU 8²⁵ (Table 1), zacopride,²⁵ renzapride,²⁰ and **6**.²⁶ However, their efficacy as agonists was found to be 60-80% of that of 5-HT, suggesting a partial agonist profile. Maximal contraction of the guinea pig ileum produced by compound 7a was 80% of that of 5-HT, and it was the most potent compound tested. The agonist profiles of compounds 7a,i,j were confirmed by evaluating their activity at 5-HT₄ receptors in the rat esophagus preparation. The potencies of 7a $(EC_{50} = 2.4 \text{ nM} (80\%))$, **7i** $(EC_{50} = 3 \text{ nM} (56\%))$, and **7j** $(EC_{50} = 2.7 \text{ nM} (67\%))$ were shown to be comparable to that of 5-HT (EC₅₀ = 2.3 nM (100%)); nevertheless, these compounds were again apparent partial agonists in this system. **1** acted as a competitive antagonist in this preparation with slopes not significantly different from unity and had pA_2 values of 7.7 \pm 0.1 and 7.6 \pm 0.2 when it was evaluated against 5-HT and compound 7a, respectively. 15, reported to be a 5-HT₄ receptor antagonist⁴² in the esophageal muscularis mucosae preparation, was found, unexpectedly, to have partial

agonist activity (EC $_{50}$ = 30 nM (52%)) in the electricallystimulated longitudinal muscle of the guinea pig ileum (Table 1).

The apparent partial agonist activity of compound 7a contrasted markedly with the antagonist profile of 1 which had an IC_{50} (\pm SE) value of 77 \pm 8 nM in the guinea pig ileum and 57 ± 10 nM in rat esophagus using 5-HT as the agonist. These two molecules are structurally similar, indicating, doubtless, a key role of the basic framework in the pharmacological profile for the 5-HT₄ receptor. As has already been suggested, a suitable orientation of the lone pair of the basic nitrogen atom^{29a} seems to be required for receptor stimulation. This orientation depends upon the conformational flexibility and steric hindrance of the basic chain. It is possible that the 1-piperidineethyl chain of compounds derived from **7a** adopts a markedly favorable conformation in the receptor site where the basic nitrogen atom occupies the correct orientation to stimulate the receptor. The steric hindrance of the less conformationally-restricted chain of 1 could disturb this orientation and induce a change in the pharmacological profile without marked modification of the affinity. The antagonist activity of the 3,5-dimethylpiperidine derivatives seemed to confirm this hypothesis. They were isolated as the cis and trans isomers (compounds 7g,h) and found to act as potent antagonists of 5-HT-mediated contractions in the guinea pig ileum preparation ($IC_{50} = 11$ and 13 nM, respectively). Competitive antagonist activity was con**Table 2.** Pharmacological Activity of the Benzoates ${\bf 8}$ at $5\text{-}\text{HT}_4$ Receptors



Compd	NR_1R_2	Binding assays <i>K</i> i, nM ^a , 5-HT ₄	Functional activity at EC ₅₀ , nM ^b	5-HT ₄ receptors IC ₅₀ , nM ^C
8a	—n	5.35± 0.31	40 [27-59] (50%)	~200
8b	-N	0.92 ± 0.03	4.4 [3.8-5.2] (59%)	n.e ^d
8c	—n	9.1 ± 0.6	29 [23-38] (65%)	n.e
8d	_N+	6.2 ± 0.29	17 [14-21] (76%)	ne
8e	—n:	6.03 ± 0.3	5.3 [4-8] (78%)	~20
8f	~ <u>N</u>	5.45 ± 0.35	7.8 [5-12] (61%)	ne
8g	N	9.3 ± 0.3	40 [32-51] (63%)	ne

a-d See footnotes to Table 1.

firmed in the rat tunica muscularis mucosae preparation where the compounds inhibited the relaxant activity of 5-HT with pA_2 values of 8.6 and 8.2, respectively, confirming the superiority of the *cis* derivative. **7g** (K_i = 0.26 ± 0.06 nM) was found to be equipotent to **4** (K_i = 0.15 ± 0.03 nM) in the binding assays and the guinea pig ileum preparation (IC₅₀ = 13 nM). The 3,3'-dimethyl derivative (compound **7f**) was also found to be an antagonist, although its antagonist potency was much lower than its affinity for 5-HT₄ receptors (IC₅₀ = 150 nM, $K_i = 1.72$ nM).

Several structural variations of compound 7 were also produced to evaluate the influence of the size and the nature of the cyclic amine on the pharmacological profile, and these results are reported in Table 2. It was noteworthy that all compounds had high affinity for 5-HT₄ receptors and behaved as agonists while possessing marked structural differences such as the size of the ring (compounds 8a-c), the existence of a bicyclic structure (compounds 8f,g), or the presence of one charge on the basic nitrogen atom (compounds **8d.e**). The introduction of a conformational constraint in the aza ring with an ethylene bridge (compounds 8f,g) did not affect the profile and potency of the pharmacological activity, which were similar to those of the parent compound. Moreover, the quaternary compounds 8d,e, which possess a totally-rigid basic framework, were found to be agonists for 5-HT₄ receptors with good efficacy (EC₅₀ = 17 nM (76%), and 5.3 nM (78%), respectively). These results agree with the report of King²⁷ which showed the favorable role of the quaternary nitrogen atom on the 5-HT₄ receptor agonist activity of renzapride.

The presence of an ester function in the pharmacologically-active molecules could be an important drawback for their use as drugs because of the putative short half-lives of such compounds. Initially, the advantage of the presence of the ester function over that of the amidic group was demonstrated by the superior activity of **1** with regard to metoclopramide. It seemed to us

worthwhile to compare the pharmacological activity at 5-HT₄ receptors of a number of amidic derivatives and the corresponding esters in order to understand the influence of this latter function. Examination of the results reported in Table 3 showed in all cases, except for BRL 24682, the superiority of the ester derivatives over the amidic derivatives and that the difference in the pharmacological activities depended essentially upon the structure of the basic framework. Thus, we observed a dramatic drop in the affinity of the amidic compounds derived from 1-piperidinoethylamine (compounds **9f**-**h**) and *N*,*N*-dimethyl- and *N*,*N*-diethylethylenediamines (compounds 9b and metoclopramide) for 5-HT₄ receptors. In particular, compound 9h, the amidic analogue of the potent 5-HT₄ receptor antagonists 7g,h, was 500-fold less potent than the ester derivatives in the binding assays and possessed partial agonist activity in the micromolar range in the ileum preparation. On the other hand, the effect was less marked for derivatives with a conformationally-constrained basic moiety such as quinuclidine or tropane. Thus the R and S enantiomers of zacopride, a wellknown 5-HT₄ receptor full agonist,⁴³ had potencies similar to those of the esters (compounds 9c,d) derived from the *R* and *S* enantiomers of 3-quinuclidinol, while BRL 24682 and the corresponding ester 9e had similar affinity for 5-HT₄ receptors. These data confirm the results reported by Gaster^{18a} which showed the superiority of the indolic esters of (1-butyl-4-piperidinyl)methanol over the corresponding amides. On the other hand, high potency was retained with the tricyclic amidic derivative 5 which was demonstrated to be a potent and selective 5-HT₄ receptor antagonist with oral activity.^{18b}

The advantages of compounds such as 7a,b,d,i-k reside in their equipotency to 5-HT in the in vitro preparations and their selectivity for 5-HT₄ receptors. The results reported in Table 4 show the low affinity of these compounds for 5-HT₃ receptors and confirm their advantage over the 5-HT₄ receptor agonist benzamides which are also potent 5-HT₃ receptor antagonists. It is well-known that this latter receptor type is implicated to the same extent as 5-HT₄ receptors in mediating intestinal contractions, and the 5-HT₃ receptor antagonist properties of the benzamides could be an important drawback for the development of gastrokinetic drugs. On the other hand, the presence of an ester function in compound 7a and its derivatives does not explain the drop in affinity for 5-HT₃ receptors because we have previously shown that the corresponding esters of a number of 5-HT₃ receptor antagonists were equipotent to the amidic derivatives at 5-HT₃ receptors.³⁴ In the course of this study, we prepared the ketone 14, the bioisosteric derivative of the ester 7a recently described by Clarke,³² to show the influence of the oxygen atom on the pharmacological activity with regard to the benzoate derivative. However, we saw a marked decrease in the affinity of this compound for 5-HT₄ receptors ($K_i = 103 \pm 65$ nM) and in its pharmacological activity on the guinea pig ileum (EC₅₀ = 115 nM, 60%) of the maximal effect of 5-HT). These data confirm the importance of the ester function in the benzoate derivatives of 1-piperidineethanol for recognition by 5-HT₄ receptors, although several aromatic ketones, analogues of the benzoate esters of 4-amino-5-chloro-2-methoxyTable 3. Pharmacological Activity of the Derivatives 9 at 5-HT₄ Receptors



^{*a*-*f*} See footnotes to Table 1. ^{*g*} See ref 8.

Table 4. Affinity of Compounds 7a,c,g,h,j for 5-HT₃ Receptors^a

compd	binding assays 5-HT3, Ki, nM		
7a	730 ± 75		
7c	>1000		
7g 7h	>1000		
$7\mathbf{\tilde{h}}$	>1000		
7j	>1000		

^a [³H]BRL-43694 was used as the radioligand, and the binding assays were carried out using rat posterior cortex (30 min, -25°C); nonspecific binding was determined with GR 38032F (10 μ M).

benzoic acid but possesing a different basic chain, have been reported to be 5-HT₄ receptor agonists³² or antagonists⁴⁴ in the rat esophageal muscularis mucosae preparation. In particular, a series of derivatives of 3-(4-piperidinyl)propiophenone, derived from the benzoates described by Gaster,^{14a} were shown to be potent partial agonists at these receptors. In this case, substitution of the oxygen atom by a methylene group reversed the pharmacological profile without decreasing the affinity.

Structural and Conformational Analysis

In order to understand the dramatic change from an agonist to an antagonist profile of the molecule 7a by the addition of two methyl groups on the piperidine ring (compound 7g), we carried out structural analyses to examine the structural properties of both molecules. The orientation of the phenyl ring was found to be markedly different between the X-ray crystal structures of the hydrochloride salt forms of the agonist 7a and the antagonist **7g** (Figure 1). The methoxy group of **7g** was found to be in the *syn* configuration with regard to the

carbonyl function, while it was in the anti position in compound **7a** as has been observed for orthopramide derivatives. However, it was coplanar with the aromatic ring in both compounds. On the other hand, both compounds had a similar orientation of the piperidine ring with regard to the benzoate moiety. This was due to the folded conformation of the ethyl chains of compounds 7a,g, the values of the torsion angles $\tau_1 =$ C(20)-O(7)-C(19)-C(18) and $\tau_2 = O(7)-C(19)-C(18)-C(18)$ N(1) being 289° and 260° for 7a and 179° and 62° for 7g, respectively. Consequently, the molecules possessed a folded shape with the hydrogen atom on the quaternary nitrogen atom pointing toward the outside of the molecule as seen in benzamides such as zacopride. renzapride, and 6. Moreover, the distance between the oxygen atom O(21) of the carbonyl group and the protonated nitrogen atom N(1), which appears to be an important structural parameter of the 5-HT₄ receptor pharmacophore, was 4.80 Å in **7a**, close to that of the two minimum energy conformers of zacopride (5.1 and 4.79 Å) reported previously.⁴⁵ These data confirm the structurally-closely-related pharmacophores of the 5-HT₃ and 5-HT₄ receptor antagonists and agonists, respectively. However, the marked structural similarity between 7a,g, which possess opposite pharmacological profiles, was intriguing. The only structural difference between compounds 7a,g resided in the change of the orientation of the aromatic ring which brought about the different configuration of the *o*-methoxy group with regard to that of the carbonyl function and the difference in the spatial position of the chlorine atom. However, these data cannot satisfactorily explain the reversal of the pharmacological profile. To date, no structural data have been reported on 5-HT₄ receptor antagonists, but



CI(31)

N(30



Figure 1. ORTEP drawing of compounds 7a (a) and 7g (b).

Table 5. Values for the Torsion Angles τ_1 and τ_2 and Energies of the Minimum Energy Conformers of Compounds **7a**,**g**

compd	conformer	τ_1 , deg	τ_2 , deg	energy, kcal mol ⁻¹
7a	1	280	300	-5.36
	2	280	180	-5.33
	3	60	60	-4.54
	4	80	180	-5.43
	5	180	60	-6.05
	6	180	180	-6.37
7g	1	280	180	-4.81
-	2	300	300	-5.01
	3	300	180	-4.33
	4	60	60	-5.53
	5	80	180	-4.55
	6	180	60	-5.38
	7	180	300	-3.47
	8	180	180	-5.54

it is likely that potent compounds such as **2** and **4** have an extended conformation because the large substituents on positions 1 and 4 of the piperidine ring prefer to occupy diequatorial stereochemistry.

The structural analyses of **7a**,**g** were performed with SYBYL software 6.03 using the Random Search and Grid Search programs, and the values of the minimum energy conformers are presented in Table 5. This study showed that there was a relatively limited number of minimun energy conformers, such as the extended conformers 6 and 8 ($\tau_1 = 180^\circ$, $\tau_2 = 180^\circ$) and those resulting from the *cis*-folded conformation of the ethyl chain; on the other hand, it did not show essential structural differences between compounds **7a**,**g**.

Therefore, we assume that the additional methyl of **7g** and, consequently, the steric hindrance are implicated in the differences in the pharmacological profiles. Thus, one of the *cis*-staggered conformers of **7a** may represent the active form at the binding site of the 5-HT₄ receptor, producing an agonist response. This would be due essentially to the particular orientation of the lone pair which is directed outside the molecule as has been observed for the 5-HT₄ receptor agonist benzamides such as zacopride and renzapride. For compound **7g**, it is probable that the folded conformations cannot occupy the agonist site because the presence of the methyl groups on the piperidine ring ham-

pers the binding to the receptor site by steric hindrance. The presence of both methyl groups seems to be essential for antagonist activity because the monomethylpiperidine derivatives, such as compounds 7b-d, are potent agonists in the pharmacological assays.

The existence of two binding sites on the 5-HT₄ receptor (Figure 2) is therefore suggested to explain the agonist and antagonist activities of 7a,g, respectively. Thus, 5-HT₄ receptor agonist or antagonist molecules in the energetically-stable-extended conformation would bind with high affinity to the first site. The agonist effect is mediated by the propensity of the molecule to adopt the folded conformation in the binding site, bringing about a conformational rearrangement of the ligand-receptor complex and the binding of the molecule in another conformation to the second site (Figure 2). In the course of this process, G-protein decoupling would occur, mediating the agonist effect. Any structural feature such as steric hindrance of the ligand would block this process and, consequently, block activation of the receptor. This hypothesis could explain the antagonist effect of compound 7g which would be hampered by the two methyl groups on the piperidine ring from adopting the folded conformation and thus could only bind to the first site. The high antagonist potency of compounds such as 2 and 4 could be due to their inability to occupy a folded conformation with a suitable orientation of the nitrogen atom. On the other hand, we suggest that 5-HT₄ receptor agonists such as benzamides, with the tied-back basic cyclic structure, may bind directly to the second site, and thus their full agonist properties result from their inability to occupy the first site.

Conclusion

The data presented here demonstrate the potential of monosubstituted 1-piperidinoethyl esters derived from 4-amino-5-chloro-2-methoxybenzoates to act as potent agonists for 5-HT₄ receptors in the CNS, guinea pig ileum, and rat esophagus. The presence of two methyl groups with *cis* stereochemistry on positions 3 and 5 of the piperidine ring increased the affinity for



Figure 2. Hypothetical model of the binding of 5-HT₄ receptor ligands to the receptor site to explain the agonist and antagonist activities of the 1-piperidineethanol benzoates **7a**,**g**, respectively.

5-HT₄ receptors and induced a dramatic change of the pharmacological profile as compound 7g is a potent antagonist for this receptor. The pharmacological activities of several esters and the corresponding amides at 5-HT₄ receptors were compared, and as has been reported previously, the role of the ester function in increasing the potency and selectivity for the 5-HT₄ receptor was confirmed. This point was particularly marked with the esters of 2-piperidinoethanol which were 100-fold more active than the corresponding amides. 7a,k,g, with nanomolar affinity, constitute interesting pharmacological tools for the understanding of the role of 5-HT₄ receptors. Comparison of the structural analyses of 7a,g did not provide an explanation for the reversal of the pharmacological profile and suggested the involvement of steric hindrance. The 5-HT₄ receptor agonist activity of these compounds could be explained by a dynamic process whereby the first phase is the recognition and binding of the molecule to the receptor followed by a conformational rearrangement of the molecule and the receptor site, mediating the coupling of G-proteins and thus the agonist activity. Compounds such as 7g possess high affinity for the first binding site but are unable to occupy the second site because of the steric hindrance of the methyl groups, thus resulting in an antagonist action at 5-HT₄ receptors. The possibility of an equilibrium between the two sites could explain the partial agonist profile of the piperidine derivatives, and further experiments are in progress to test this hypothesis.

Experimental Section

Chemistry. Melting points were determined on a Mettler FP 61 melting point apparatus and are uncorrected. ¹H and ¹³C NMR spectra were recorded on a Bruker AC 200 spectrometer at 200 and 50 MHz, respectively. Chemical shifts are reported in parts per million (δ) relative to tetramethylsilane as the internal standard, and signals are quoted as s (singlet), ds (dedoublet singlet), d (doublet), dd (dedoublet doublet), t (triplet), dt (dedoublet triplet), q (quartet), br s (broad singlet), or m (multiplet). Elemental analyses were performed at the CNRS's analysis services in Châtenay-Malabry (France) and were within 0.4% of the theoretical values unless otherwise noted.

Materials. Tetrahydrofuran (THF) was distilled from sodium/benzophenone. Acetonitrile, dimethylformamide, toluene, and the usual solvents were purchased from SDS (Paris, France). Column chromatography was performed on Merck silica gel 60 (70/230 mesh). Thin-layer chromatography was done on silica gel 60F-254 (0.26 mm thickness) plates.

4-Amino-5-chloro-2-methoxybenzoic acid, trityl chloride, the mixture of *cis*- and *trans*-3,5-dimethylpiperidine, 4-benzylpiperidine, hexamethylenimine, 1,2,3,6-tetrahydropyridine, *cis*-2,6-dimethylpiperidine, the mixture of *cis*- and *trans*-2,6-dimethylpiperidine, 3,3-dimethylpiperidine, 3-methylpiperidine, 2-methylpiperidine, 3,3-dimethylpiperidine, 4-hydroxypiperidine, 3-hydroxypiperidine, 4-piperidinecarboxamide, quinuclidine, 3-azabicyclo[3.2.2]nonane, *N*,*N*-dimethylenediamine, morpholine, pyrrolidine, 1-(3-chloropropyl)piperidine, and MnO₂ were purchased from Aldrich (France).

4-Acetamidopiperidine was synthesized from 4-amino-1benzylpiperidine⁴⁶ by acetylation and debenzylation; 8-azabicyclo[3.2.1]octane was prepared from *N*-benzyltropinone (Janssen) by the Wolf–Kischner reduction⁴⁷ followed by debenzylation. (*R*)- and (*S*)-3-quinuclidinol were prepared according to the previously-reported process.⁴⁸

N-Methyl-2-(1-piperidinyl)ethylamine. A solution of 1-(2-chloroethyl)piperidine hydrochloride (20 g, 100 mmol) in distilled water (30 mL) was added to a stirred solution of 40% aqueous methylamine (77.4 mL) for 20 min. The mixture became warm during the addition. After 30 min, the reaction mixture was made basic by the addition of 37 g of NaOH pellets, and the compound was separated as a yellow oil. The cooled solution was extracted with ether (4×50 mL), and the combined organic layers were evaporated *in vacuo* to afford the product as a pale yellow oil. The oil was distilled to give 9 g (58%) of a colorless oil: bp 54–58 °C (1 mmHg); ¹H NMR (CDCl₃) δ 2.5 (t, 2H), 2.45–2.2 (m, 9H), 1.65–1.25 (m, 6H).

2-Bromoethyl 4-Amino-5-chloro-2-methoxybenzoate (10). A mixture of 4-amino-5-chloro-2-methoxybenzoic acid (20.1 g, 100 mmol) in 1,2-dibromoethane (100 mL) and dry THF (200 mL) was heated to reflux. Then 15.2 g (100 mmol) of 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) was added dropwise over 2 h, and the resulting mixture was maintained under reflux for 3 h. The cooled mixture was filtered, and the filtrate was concentrated *in vacuo*. The residue was diluted with CH₂Cl₂ (500 mL), washed with water, and dried over MgSO₄. The solution was filtered through a short column of silica gel and eluted with CH₂Cl₂. Evaporation of the combined eluates and recrystallization of the solid residue from Et₂O/pentane gave **10** (22 g, 71%): mp 137 °C; R_f 0.63 (CH₂Cl₂/MeOH, 95: 5); ¹H NMR (CDCl₃) δ 7.8 (s, 1H, ArH), 6.22 (s, 1H, ArH), 4.46 (t, J = 6 Hz, 2H, OCH₂), 4.45 (s, 2H, NH₂), 3.78 (s, 3H, OCH₃), 3.54 (t, J = 6 Hz, 2H, CH₂Br). Anal. (C₁₀H₁₁NO₃BrCl).

5-Chloro-2-methoxy-4-(tritylamino)benzoic Acid (11). Trityl chloride (33.3 g, 119 mmol) in dry CH₂Cl₂ (60 mL) was added portionwise at 0 °C to a mixture of 20 g (99 mmol) of 4-amino-5-chloro-2-methoxybenzoic acid in 100 mL of pyridine. The mixture was stirred overnight at room temperature, and the solvent was evaporated. The residue was taken up in CH₂Cl₂ and washed several times with water. The organic layer was dried over MgSO₄, and the solvent was evaporated. The solid was recrystallized from a mixture of CH₂Cl₂/EtOH to yield a white solid 38 g (86%): mp 214 °C; R_r 0.58 (CH₂Cl₂/MeOH, 90:10); ¹H NMR (CDCl₃) δ 8.02 (s, 1H, ArH), 7.4–7.18 (m, 15H, tritylH), 6.34 (br s, 1H, NH), 5.65 (s, 1H, ArH), 3.29 (s, 3H, OCH₃). Anal. ($C_{27}H_{22}NO_3Cl$).

General Method A: Preparation of Compounds 7a,g,h, m,o,p,s and 8a-c,f. 2-Piperidinoethyl 4-amino-5-chloro-2-methoxybenzoate Hydrochloride (7a). A mixture of compound 10 (1.54 g, 5 mmol), piperidine (0.85 g, 10 mmol), and K₂CO₃ (1.38 g, 10 mmol) in dry DMF (20 mL) was stirred for 18 h at room temperature. The resulting precipitate was filtered off, and the filtrate was concentrated in vacuo. The residue was dissolved in ethyl acetate (100 mL), washed with water, and dried over MgSO₄. The resulting solution was then treated with 3 N HCl ethyl acetate solution to afford the crude hydrochloride salt 7a (1.4 g, 80%) which was recrystallized from *i*-PrOH/*i*-Pr₂O: mp > 200 °C; R_f 0.72 (CH₂Cl₂/MeOH/NH₄-OH, 99:0.9:0.1); ¹H NMR (CD₃OD) δ 7.8 (s, 1H, ArH), 6.5 (s, 1H, ArH), 4.6-4.5 (m, 2H, OCH₂), 3.7 (s, 3H, OCH₃), 3.75-2.9 (m, 6H), 2.1–1.4; ¹³C NMR (CD₃OD) δ 166.17 (C=O), 162.29 (C₂), 152.20 (C₄), 134.65 (C₆), 110.71–107.67 (C₁–C₅), 98.88 (C₃), 59.78, 57.27, 56.64, 55.19, 24.41, 22.71. Anal. $(C_{15}H_{21}N_2O_3Cl\cdot HCl).$

2-(*cis*-3,5-Dimethylpiperidino)ethyl 4-Amino-5-chloro-2-methoxybenzoate (7g) and 2-(*trans*-3,5-Dimethylpiperidino)ethyl 4-Amino-5-chloro-2-methoxybenzoate (7h). As described for the preparation of **7a**, compound **10** (3.08 g, 10 mmol) was reacted with the commercially-available mixture of *cis*- and *trans*-3,5-dimethylpiperidine (2.26 g, 20 mmol) in dry DMF at 50 °C for 5 h. After evaporation of the solvent, the residue was dissolved in a minimum amount of methylene chloride and chromatographed on a column of silica gel with a mixture of $Et_2O/CH_2Cl_2/MeOH$ (7:3:0.1); the *trans* isomer **7h** (0.5 g, 15%) was eluted first followed by the *cis* isomer **7g** (1.7 g, 50%).

7g: mp 115 °C; $R_f 0.2$ (Et₂O/CH₂Cl₂/MeOH, 7:3:0.1); ¹H NMR (CDCl₃) δ 7.8 (s, 1H, ArH), 6.25 (s, 1H, ArH), 4.39 (s, 2H, NH₂), 4.35 (t, J = 6.3 Hz, 2H, OCH₂), 3.81 (s, 3H, OCH₃), 2.88 (m, 2H_{eq}, 2,6-piperidinyl H), 2.71 (t, J = 6.3 Hz, 2H, CH₂N), 1.71–1.52 (m, 5H), 0.84 (d, J = 5.8 Hz, 6H, CH₃ × 2), 0.52 (m, 1H_{ax}, 4-piperidinyl H); ¹³C NMR (CDCl₃) δ 164.5 (C=O), 160.2 (C₂), 147.9 (C₄), 133.09 (C₆), 109.69–109.32 (C₁–C₅), 98.01 (C₃), 61.96, 61.77, 56.82, 55.85 (OCH₃), 41.87, 30.96, 19.51 (CH₃ × 2). Anal. (C₁₇H₂₅N₂O₃Cl).

7h: mp 114–115 °C; R_f 0.3 (Et₂O/CH₂Cl₂/MeOH, 7:3:0.1); ¹H NMR (CDCl₃) δ 7.82 (s, 1H, ArH), 6.28 (s, 1H, ArH), 4.44 (s, 2H, NH₂), 4.33 (t, J = 6.3 Hz, 2H, OCH₂), 3.84 (s, 3H, OCH₃), 2.64 (m, 2H, CH₂N), 2.48 (m, 2H_{eq}, 2,6-piperidinyl H), 2.14 (m, 2H_{ax}, 2,6-piperidinyl H), 1.99 (m, 2H, 3,5-piperidinyl H), 1.26 (t, J = 5.8 Hz, 2H, 4-piperidinyl H), 0.94 (d, J = 6.9 Hz, 6H, CH₃ × 2); ¹³C NMR (CDCl₃) δ 164.48 (C=O), 160.16 (C₂), 147.84 (C₄), 133.17 (C₆), 109.77–109.64 (C₁–C₅), 98.17 (C₃), 61.96, 61.17, 57.07, 55.92 (OCH₃), 38.85, 27.34, 19.07 (CH₃ × 2). Anal. (C₁₇H₂₅N₂O₃Cl).

2-(4-Acetamidopiperidino)ethyl 4-Amino-5-chloro-2methoxybenzoate (7m). As described for the preparation of **7a**, compound **10** was reacted with 4-acetamidopiperidine. Purification by chromatography on silica gel (CH₂Cl₂/MeOH, 7:1) and recrystallization from AcOEt/hexane gave **7m** (1.1 g, 60%): mp 135 °C; R_f 0.35 (CH₂Cl₂/MeOH, 85:15); ¹H NMR (CDCl₃) δ 7.80 (s, 1H, ArH), 6.27 (s, 1H, ArH), 5.36 (d, 1H, NH), 4.48 (s, 2H, NH₂), 4.34 (t, J = 6 Hz, 2H, OCH₂), 3.75 (d, 2H, NH₂), 4.34 (t, J = 6 Hz, 2H, OCH₂), 3.75 (c, 2, 6piperidinyl H), 2.72 (t, J = 6 Hz, 2H, CH₂N), 2.24 (m, 2H_{ax}, 2,6-piperidinyl H), 1.96 (s, 3H, CH₃), 1.9 (m, 2H_{eq}, 3,5piperidinyl H), 1.45 (m, 2H_{ax}, 3,5-piperidinyl H); ¹³C NMR (CDCl₃) δ 169.4 (CONH), 164.5 (CO₂), 160.2 (C₂), 147.9 (C₄), 133.3 (C₆), 109.9–109.6 (C₁–C₅), 98.2 (C₃), 62.2, 56.7, 56.0, 46.4, 52.6, 32.2, 23.5 (CH₃). Anal. (C₁₇H₂₄N₃O₄Cl).

2-(4-Benzylpiperidino)ethyl 4-Amino-5-chloro-2-methoxybenzoate Hydrochloride (70). As described for the preparation of **7a**, compound **10** (1.23 g, 4 mmol) was reacted with 4-benzylpiperidine (0.7 g, 4.0 mmol) in dry DMF to give **7o**, transformed into the hydrochoride salt and recrystallized from MeOH/AcOEt/hexane: 1.2 g (68%); mp 170 °C; R_f 0.33 (CH₂Cl₂/MeOH/NH₃(aq), 95:5:0.1); ¹H NMR (CD₃OD) δ 7.86 (s, 1H, ArH), 7.37–7.22 (m, 5H, PhH), 6.52 (s, 1H, ArH), 4.60 (t, J = 5 Hz, 2H, OCH₂), 3.85 (s, 3H, OCH₃), 3.71 (m, 2H), Yang et al.

3.55 (t, J = 5 Hz, 2H, CH₂N), 3.12 (m, 2H), 2.69 (d, J = 7 Hz, 2H, CH₂Ph), 1.97 (m, 3H), 1.58 (m, 2H). Anal. (C₂₂H₂₇N₂O₃-Cl·HCl).

2-Morpholinoethyl 4-Amino-5-chloro-2-methoxybenzoate (7p). As described previously for **7a**, compound **10** (1.23 g, 4 mmol) was reacted with morpholine in dry DMF to give **7p** (58%): mp 80 °C; ¹H NMR (CDCl₃) δ 7.9 (s, 1H, ArH), 6.35 (s, 1H, ArH), 4.5 (s, 2H, NH₂), 4.42 (t, *J* = 5 Hz, 2H, OCH₂), 3.85 (s, 3H, OCH₃), 3.75 (m, 4H, CH₂OCH₂), 2.8 (t, *J* = 5 Hz, 2H, NCH₂), 2.6 (m, 4H, CH₂NCH₂). Anal. (C₁₄H₁₉N₂O₄-Cl·0.5H₂O).

2-(4-Phenylpiperazino)ethyl 4-Amino-5-chloro-2-methoxybenzoate Hydrochloride (7s). As described previously for **7a**, compound **10** (1.23 g, 4 mmol) was reacted with 4-phenylpiperazine (0.65 g, 4.0 mmol) in dry DMF to give **7s**, transformed into the hydrochloride and recrystallized from the MeOH/AcOEt/hexane mixture (1.0 g, 59%): mp 224 °C; R_f 0.44 (CH₂Cl₂/MeOH/NH₃(aq), 95:5:0.1); ¹H NMR (CD₃OD) δ 7.91 (s, 1H, ArH), 7.34 (m, 2H, PhH), 7.09 (m, 2H, PhH), 6.99 (m, 1H, PhH), 6.54 (s, 1H, ArH), 4.68 (t, J = 5 Hz, 2H, OCH₂), 3.89 (s, 3H, OCH₃), 3.7–3.3 (m, 10H). Anal. (C₂₀H₂₄N₃O₃Cl·-HCl).

2-Pyrrolidinoethyl 4-Amino-5-chloro-2-methoxybenzoate (8a). As described for the preparation of **7a**, compound **10** (1.23 g, 4 mmol) was reacted with pyrrolidine to give **8a** as an oil: ¹H NMR (CDCl₃) δ 7.81 (s, 1H, ArH), 6.27 (s, 1H, ArH), 4.37 (t, 2H, J = 6.1 Hz, 2H, OCH₂), 3.82 (s, 3H, OCH₃), 2.83 (t, 2H, NCH₂, J = 6.1 Hz), 2.7–2.5 (m, 4H), 1.9–1.65 (m, 4H). Anal. (C₁₄H₁₉N₂O₃Cl·0.5H₂O).

7-(Hexamethyleneimino)ethyl 4-Amino-5-chloro-2methoxybenzoate Hydrochloride (8b). As described for the preparation of **7a**, compound **10** (1.54 g, 5 mmol) was reacted with hexamethylenimine (0.5 g, 5.0 mmol) in dry DMF to give **8b**, transformed into the hydrochoride salt and recrystallized from MeOH/AcOEt (1.1 g, 86%): mp 140 °C; R_f 0.44 (CH₂Cl₂/MeOH/NH₄OH, 99:1:0.1); ¹H NMR (CD₃OD) δ 7.65 (s, 1H, ArH), 6.34 (s, 1H, ArH), 4.5 (t, J = 5 Hz, 2H, OCH₂), 3.67 (s, 3H, OCH₃), 3.44 (t, J = 5 Hz, 2H, CH₂N), 3.21 (m, 4H), 1.88 (m, 4H), 1.73 (m, 4H). Anal. (C₁₆H₂₃N₂O₃Cl· HCl·0.5H₂O).

2-(1,2,3,6-Tetrahydropyridino)ethyl 4-Amino-5-chloro-2-methoxybenzoate (8c). As described for the preparation of **7a**, compound **10** was reacted with 1,2,3,6-tetrahydropyridine to give **8c** as a white solid (yield 10%) after recrystallization from hexane/AcOEt: mp 110 °C; R_f 0.37 (CH₂Cl₂/MeOH, 95:5); ¹H NMR (CDCl₃) δ 7.82 (s, 1H, ArH), 6.27 (s, 1H, ArH), 5.8–5.55 (m, 2H, CH=CH), 4.44 (br s, 2H, NH₂), 4.40 (t, J = 6.2 Hz, 2H, OCH₂), 3.84 (s, 3H, OCH₃), 3.14–3.07 (m, 2H), 2.83 (t, J = 6.2 Hz, 2H, CH₂N), 2.68 (t, J = 5.6 Hz, 2H), 2.2– 2.1 (m, 2H). Anal. (C₁₅H₁₉N₂O₃Cl).

2-(3-Azabicyclo[3.2.2]nonan-3-yl)ethyl 4-Amino-5-chloro-2-methoxybenzoate Hydrochloride (8f). It was prepared according to method A from compound **10** (2.45 g, 8.0 mmol) and 3-azabicyclo[3.2.2]nonane (1.0 g, 8.0 mmol) in dry DMF, transformed into the hydrochloride salt, and recrystallized from *i*-PrOH/*i*-Pr₂O: 2.2 g (78%); mp 224 °C; R_f 0.63 (CH₂Cl₂/MeOH/NH₄OH, 99:1:0.1); ¹H NMR (CD₃OD) δ 7.60 (s, 1H, ArH), 6.35 (s, 1H, ArH), 4.54 (t, J = 5 Hz, 2H, OCH₂), 3.65 (s, 3H, OCH₃), 3.40 (t, J = 5 Hz, 2H, CH₂N), 3.7–3.0 (m, 4H), 2.1 (br, 2H), 1.9–1.6 (m, 8H). Anal. (C₁₈H₂₅N₂O₃Cl·HCl).

General Method B: Preparation of Compounds 7b– f,i–l,n,q–r and 8d,e,g. 2-(*cis*-2,6-Dimethylmorpholino)ethyl 4-Amino-5-chloro-2-methoxybenzoate Hydrochloride (7q) and 2-(*trans*-2,6-Dimethylmorpholino)ethyl 4-Amino-5-chloro-2-methoxybenzoate Hydrochloride (7r). A mixture of compound 10 (1.54 g, 5 mmol) and the commercially-available *cis*, *trans*-2,6-dimethylmorpholine (1.5 g, 13 mmol) in dry toluene (20 mL) was heated to reflux for 5 h. After evaporation of the toluene, the residue was dissolved in CH_2CI_2 (100 mL). The organic layer was washed with water and brine, dried over MgSO₄, and concentrated. The mixture of *cis* and *trans* isomers was separated by flash chromatography on silica gel and eluted with $CH_2CI_2/Et_2O/MeOH$ (5:5:0.1) to give 7r (0.4 g, 23%) and 7q (0.85 g, 50%). Compounds 7q,r were converted into HCl salts by a 4 N HCl ether solution. **7q:** mp 224 °C; $R_f 0.4$ (Et₂O/CH₂Cl₂/MeOH, 7:3:0.1); ¹H NMR (CDCl₃) δ 7.8 (s, 1H, ArH), 6.25 (s, 1H, ArH), 4.52 (s, 2H, NH₂), 4.37 (t, J = 6.3 Hz, 2H, OCH₂), 3,8 (s, 3H, OCH₃), 3.68 (m, 2H, 2,6-morpholinyl H), 2.80 (m, 2H_{eq}, 3,5-morpholinyl H), 2.7 (t, J = 6.3 Hz, 2H, CH₂N), 1.86 (m, 2H_{ax}, 3,5-morpholinyl H), 1.15 (d, J = 6.4 Hz, 3H, CH₃ × 2); ¹³C NMR (CDCl₃) δ 164.38 (C=O), 160.17 (C₂), 147.80 (C₄), 133.18 (C₆), 109.78, 109.36, 98.09 (C₃), 71.54, 61.63, 59.59, 56.60, 55.94 (OCH₃), 19.03 (CH₃ × 2). Anal. (C₁₆H₂₃N₂O₄Cl·HCl·0.5H₂O).

7r: mp >170 °C; R_f 0.5 (Et₂O/CH₂Cl₂/MeOH, 7:3:0.1); ¹H NMR (CDCl₃) δ 7.77 (s, 1H, ArH), 6.26 (s, 1H, ArH), 4.59 (s, 2H, NH₂), 4.30 (t, J = 6.3 Hz, 2H, OCH₂), 3.97 (m, 2H, 2,6-morpholinyl H), 3.77 (s, 3H, OCH₃), 2.64–2.50 (m, 4H), 2.21 (m, 2H), 1.28 (d, J = 6.4 Hz, 6H, CH₃ × 2); ¹³C NMR (CDCl₃) δ 164.38 (C=O), 160.15 (C₂), 147.95 (C₄), 133.10 (C₆), 109.71–109.36 (C₁–C₅), 98.10 (C₃), 66.53, 61.45, 58.78, 56.83, 55.89 (OCH₃), 18.01 (CH₃ × 2). Anal. (C₁₆H₂₃N₂O₄Cl·HCl·0.5H₂O).

2-(4-Methylpiperidino)ethyl 4-Amino-5-chloro-2-methoxybenzoate (7b). It was prepared according to method B from compound **10** and 4-methylpiperidine in dry CH₃CN. Recrystallization from AcOEt/hexane afforded **7b** (1.2 g, 73%): mp 123 °C; R_f 0.49 (CH₂Cl₂/MeOH/NH₃ (aq), 9:1:0.1); ¹H NMR (CDCl₃) δ 7.73 (s, 1H, ArH), 6.20 (s, 1H, ArH), 4.47 (s, 2H, NH₂), 4.30 (t, J = 6.2 Hz, 2H, OCH₂), 3.75 (s, 3H, OCH₃), 2.85 (m, 2H), 2.65 (t, J = 6.2 Hz, 2H, CH₂N), 2.01 (m, 2H), 1.52 (m, 2H), 1.2–1.1 (m, 3H), 0.85 (d, J = 6 Hz, 3H, CH₃). Anal. (C₁₆H₂₃N₂O₃Cl).

2-(3-Methylpiperidino)ethyl 4-Amino-5-chloro-2-methoxybenzoate Hydrochloride (7c). It was prepared according to method B from compound **10** (0.92 g, 3 mmol) and 3-methylpiperidine (0.74 g, 7.5 mmol) to give crude compound **7c** as a colorless foam, which was converted into the HCl salt by a 4 N HCl ether solution. Recrystallization from *i*-PrOH/ *i*-Pr₂O gave **7c** (0.7 g, 64%): mp 205 °C; R_f 0.5 (CH₂Cl₂/MeOH/ NH₃(aq), 9:1:0.1); ¹H NMR (CDCl₃) δ 7.68 (s, 1H, ArH), 6.15 (s, 1H, ArH), 4.4 (s, 2H, NH₂), 4.35 (t, J = 6 Hz, 2H, OCH₂), 3.72 (s, 3H, OCH₃), 2.85 (m, 2H), 2.68 (t, J = 6 Hz, 2H, CH₂N), 1.8 (m, 1H), 1.7–1.5 (m, 5H), 0.65–0.9 (m, 4H); ¹³C NMR (CDCl₃) δ 164.36 (C=O), 160.12 (C₂), 148.14 (C₄), 133.08 (C₆), 109.67, 108.84, 97.96 (C₃), 61.74, 61.23, 56.86, 55.85 (OCH₃), 53.91, 32.31, 30.57, 24.92, 19.94 (CH₃). Anal. (C₁₆H₂₃N₂O₃-Cl·HCl).

2-(2-Methylpiperidino)ethyl 4-Amino-5-chloro-2-methoxybenzoate Hydrochloride (7d). It was prepared according to method B from compound **10** (0.92 g, 3 mmol) and 2-methylpiperidine (0.74 g, 7.5 mmol) as a colorless foam, which was converted into the HCl salt by a 4 N HCl ether solution. Recrystallization from *i*-PrOH/*i*-Pr₂O gave **7d** (0.9 g, 80%): mp 185 °C; R_r 0.47 (CH₂Cl₂/MeOH/NH₄OH, 9:1:0.1); ¹H NMR (CDCl₃) δ 7.8 (s, 1H, ArH), 6.25 (s, 1H, ArH), 4.47 (s, 2H, NH₂), 4.35 (t, J = 6.2 Hz, 2H, OCH₂), 3.82 (s, 3H, OCH₃), 3.0–2.7 (m, 3H), 2.35 (m, 2H), 1.6 (m, 4H), 1.25 (m, 2H), 1.1 (d, J = 6 Hz, 3H, CH₃); ¹³C NMR (CDCl₃) δ 164.48 (C=O), 160.25 (C₂), 147.71 (C₄), 133.31 (C₆), 109.87, 109.46, 98.18 (C₃), 61.77, 56.02–55.96, 53.29, 52.06, 34.64, 26.12, 24.03, 19.56 (CH₃). Anal. (C₁₆H₂₃N₂O₃Cl·HCl).

2-(*cis*-2,6-Dimethylpiperidino)ethyl 4-Amino-5-chloro-**2**-methoxybenzoate (7e). It was prepared from compound **10** (0.92 g, 3 mmol) and 2,6-dimethylpiperidine (1.0 g, 9 mmol) as a colorless foam. It was triturated with Et₂O and recrystallized from AcOEt/Hexane to give **7e** (0.4 g, 39%): mp 109– 111 °C; R_f 0.3 (CH₂Cl₂/MeOH/NH₄OH, 9:1:0.1); ¹H NMR (CDCl₃) δ 7.8 (s, 1H, ArH), 6.25 (s, 1H, ArH), 4.47 (s, 2H, NH₂), 4.25 (t, J = 6 Hz, 2H, OCH₂), 3.8 (s, 3H, OCH₃), 3.0 (t, J = 6Hz, 2H, CH₂N), 2.5 (m, 2H, 2,6-piperidinyl H), 1.5–1.7 (m, 3H), 1.4–1.2 (m, 3H), 1.15 (d, J = 6 Hz, 6H, CH₃ × 2); ¹³C NMR (CDCl₃) δ 164.42 (C=O), 160.31 (C₂), 147.75 (C₄), 133.26 (C₆), 109.81, 109.4, 98.13 (C₃), 61.8, 56.72, 56.01 (OCH₃), 46.33, 34.44, 24.59, 21.55 (CH₃). Anal. (C₁₇H₂₅N₂O₃Cl).

2-(3,3-Dimethylpiperidino)ethyl 4-Amino-5-chloro-2methoxybenzoate Hydrochloride (7f). It was prepared according to method B from compound **10** and 3,3-dimethylpiperidine in toluene and transformed into the hydrochloride salt (1.5 g, 80%): mp 260 °C dec; R_f 0.5 (CH₂Cl₂/MeOH/NH₄OH, 9:1:0.1); ¹H NMR (DMSO) δ 10.3 (s, 1H, HCl), 7.65 (s, 1H, ArH), 6.48 (s, 1H, ArH), 6.26 (s, 2H, NH₂), 4.55 (m, 2H, OCH₂), 3.68 (s, 3H, OCH₃), 3.5–3.1 (m, 4H), 2.9–2.7 (m, 2H), 2.0–1.6 (m, 2H), 1.5–1.1 (m, 2H), 1.1 (s, 3H, CH₃), 0.90 (s, 3H, CH₃). Anal. ($C_{17}H_{25}N_2O_3Cl$ ·HCl·0.5H₂O).

2-(4-Hydroxylpiperidino)ethyl 4-Amino-5-chloro-2methoxybenzoate Hydrochloride (7i). It was prepared according to method B from compound **10** (1.54 g, 5 mmol) and 4-hydroxypiperidine (1.26 g, 12.5 mmol) and transformed into the hydrochloride salt which was recrystallized from the *i*-PrOH/CHCl₃ mixture (1.2 g, 66%): mp 130–150 °C; R_f 0.35 (CH₂Cl₂/MeOH/NH₄OH, 9:10.1); ¹H NMR (CD₃OD) δ 7.75 (s, 1H, ArH), 6.4 (s, 1H, ArH), 4.5 (t, J = 5 Hz, 2H, OCH₂), 3.75 (s, 3H, OCH₃), 3.7–3.6 (m, 1H), 3.5 (t, J = 5 Hz, 2H, CH₂N), 3.5–3.0 (m, 4H), 1.88 (m, 4H), 1.73 (m, 4H). Anal. (C₁₅H₂₁N₂O₄-Cl·HCl·H₂O).

2-(3-Hydroxylpiperidino)ethyl 4-Amino-5-chloro-2methoxybenzoate Hydrochloride (7j). It was prepared according to method B from compound **10** (1.54 g, 5 mmol) and 3-hydroxypiperidine as a colorless foam, which was then converted into the HCl salt by a 4 N HCl ether solution and recrystallized from a EtOH/CHCl₃ mixture to give 1.0 g (55%) of **7j**: mp 130–150 °C; R_f 0.31 (CH₂Cl₂/MeOH/NH₄OH, 9:1: 0.1); ¹H NMR (DMSO) δ 7.62 (s, 1H, ArH), 6.48 (s, 1H, ArH), 6.19 (s, 2H, NH₂), 4.61 (d, 1H, OH), 4.23 (m, 2H, OCH₂), 3.77 (s, 3H, OCH₃), 3.58–3.31 (m, 1H, CH), 2.92 (m, 1H), 2.75 (m, 1H), 2.63 (m, 2H, NCH₂), 2.10–1.72 (m, 3H), 1.72–1.56 (m, 3H), 1.56–1.25 (m, 1H), 1.25–0.93 (m, 1H). Anal. (C₁₅H₂₁N₂O₄-Cl·HCl·H₂O).

2-(3-(*S*)-Hydroxylpiperidino)ethyl 4-Amino-5-chloro-2-methoxybenzoate Hydrochloride (7k). It was prepared according to method B from compound **10** (1.6 g, 5.3 mmol) and 3-(*S*)-hydroxypiperidine (1.3g, 12.9 mmol) and then purified by chromatography on silica gel (CH₂Cl₂/MeOH, 9:1). The residue was further purified through the formation of the hydrochloride salt. It was treated with base to give **7k** (0.54 g, 30%): mp 95–98 °C; $[\alpha]^{20} = +3^{\circ}$ (c = 1, MeOH); ¹H NMR (DMSO) δ 7.60 (s, 1H, ArH), 6.47 (s, 1H, ArH), 6.20 (s, 2H, NH₂), 4.66 (d, 1H, OH), 4.25 (m, 2H, OCH₂), 3.78 (s, 3H, OCH₃), 3.63–3.28 (m, 1H, CH), 2.94 (m, 1H), 2.77 (m, 1H), 2.66 (m, 2H, NCH₂), 2.09–1.72 (m, 3H), 1.72–1.53 (m, 3H), 1.53–125 (m, 1H), 1.25–0.91 (m, 1H). Anal. (C₁₅H₂₁N₂O₄Cl·HCl·H₂O).

2-(3-(*R***)-Hydroxylpiperidino)ethyl 4-Amino-5-chloro-2-methoxybenzoate Hydrochloride (7l).** It was prepared according to method B from compound **10** (1.6 g, 5.2 mmol) and 3-(*R*)-hydroxypiperidine (1.3 g, 12.9 mmol) and then purified by chromatography on silica gel (CH₂Cl₂/MeOH, 9:1). The residue was further purified through the formation of the hydrochloride salt. It was treated with base to give **7l** (0.54 g, 30%): mp 95–98 °C; $[\alpha]^{20} = -3.5^{\circ}$ (c = 1, MeOH); ¹H NMR (DMSO) δ 7.60 (s, 1H, ArH), 6.47 (s, 1H, ArH), 6.21 (s, 2H, NH₂), 4.66 (d, 1H, OH), 4.25 (m, 2H, OCH₂), 3.78 (s, 3H, OCH₃), 3.60–3.34 (m, 1H, CH), 2.94 (m, 1H), 2.78 (m, 1H), 2.66 (m, 2H, NCH₂), 2.09–1.72 (m, 3H), 1.72–1.55 (m, 3H), 1.55–1.31 (m, 1H), 1.25–0.97 (m, 1H). Anal. (C₁₅H₂₁N₂O₄Cl·HCl·0.5H₂O).

2-(4-Carboxamidopiperidino)ethyl 4-Amino-5-chloro-2-methoxybenzoate (7n). It was prepared according to method B from compound **10** (1.0 g, 3.2 mmol) and 4-piperidinecarboxamide (0.82 g, 6.4 mmol) in dry DMF at 50 °C for 5 h. Purification by flash chromatography (CH₂Cl₂/MeOH, 9:1) and recrystallization from CHCl₃/hexane yielded **7n** (0.8 g, 71%): mp 137 °C; R_f 0.4 (CH₂Cl₂/MeOH, 8:2); ¹H NMR (CDCl₃) δ 7.79 (s, 1H, ArH), 6.27 (s, 1H, ArH), 5.61 (s, 2H, CONH₂), 4.51 (s, 2H, NH₂), 4.34 (t, J = 6.0 Hz, OCH₂), 3.82 (s, 3H, OCH₃), 3.01 (m, 2H), 2.72 (t, J = 6.0 Hz, OCH₂), 3.82 (s, 3H, OCH₃), 3.01 (m, 2H), 2.72 (t, J = 6.0 Hz, OCH₂), δ 175.7 (CONH), 169.0 (CO₂), 164.5 (C₂), 154.0 (C₄), 136.7 (C₆), 112.9–111 (C₁-C₅), 101.3 (C₃), 65.2, 60.5–58.9, 57.0, 45.8, 32.1. Anal. (C₁₆H_{22N₃O₄Cl).}

[2-[(4-Amino-5-chloro-2-methoxybenzoyl)oxy]ethyl]pyridinium Bromide (8d). A solution of compound 10 (0.62 g, 2.0 mmol) in pyridine (20 mL) was refluxed for 5 h; 20 mL of ether was added to the cooled reaction mixture, and the crude compound 8d was collected by filtration. Recrystallization from EtOH/*i*-Pr₂O gave 8d (0.6 g, 78%) as a white solid: mp 223 °C; ¹H NMR (D₂O) δ 8.75 (m, 2H, pyridinyl H), 8.35 (m, 1H, pyridinyl H), 7.83 (m, 2H, pyridinyl H), 7.01 (s, 1H, ArH), 6.0 (s, 1H, ArH), 4.7 (m, 2H), 4.4 (m, 2H), 3.4 (s, 3H, OCH_3). Anal. ($C_{15}H_{16}N_2O_3ClBr$).

[2-(4-Amino-5-chloro-2-methoxybenzoxy)ethyl]quinuclidinium Bromide (8e). A solution of compound 10 (1.24 g, 4.0 mmol) and quinuclidine (0.44 g, 4.0 mmol) in THF (25 mL) was refluxed for 24 h. The solution was cooled and filtered to give **8e** (1.4 g, 71%) after recrystallization from EtOH/*i*-Pr₂O: mp >260 °C; ¹H NMR (CD₃OD) δ 7.64 (s, 1H, ArH), 6.38 (s, 1H, ArH), 4.54 (m, 2H, OCH₂), 3.7 (s, 3H, OCH₃), 3.54 (m, 8H), 2.08 (m, 1H), 1.95 (m, 6H). Anal. (C₁₇H₂₄N₂O₃ClBr).

2-(8-Azabicyclo[3.2.1]octan-8-yl)ethyl 4-Amino-5-chloro-2-methoxybenzoate (8g). It was prepared according to method B from compound **10** and 8-azabicyclo[3.2.1]octane. The crude product was purified by chromatography on silica gel (CH₂Cl₂/MeOH, 85:15) and by recrystallization from an AcOEt/hexane mixture to give **8g** (1.0 g, 59%): mp 122 °C; R_f 0.46 (CH₂Cl₂/MeOH, 85:15): ¹H NMR (CDCl₃) δ 8.00 (s, 1H, ArH), 6.46 (s, 1H, ArH), 4.79 (s, 2H, NH₂), 4.53 (t, J = 6.4 Hz, 2H, OCH₂), 3.99 (s, 3H, OCH₃), 3.45 (m, 2H), 2.89 (t, J = 6.4 Hz, 2H, CH₂N), 2.2–1.5 (m, 10H); ¹³C NMR (CDCl₃) δ 164.5 (C=O), 160.1 (C₂), 147.8 (C₄), 133.1 (C₆), 109.7–109.6 (C₁–C₅), 98.1 (C₃), 64.0, 60.2, 55.9 (OCH₃), 51.1, 30.5, 26.3, 16.4. Anal. (C₁₇H₂₃N₂O₃Cl).

(*R*)-1-Azabicyclo[2.2.2]octan-3-yl 4-Amino-5-chloro-2methoxybenzoate (9c). It was prepared from 5-chloro-4-(tritylamino)-2-methoxybenzoic acid imidazolide which was synthesized according to the following process: Carbonyldiimidazole (3.2 g, 20 mmol) in dry THF (40 mL) was added under an inert atmosphere to a solution of the benzoic acid derivative 11 (4.44 g, 10 mmol) in 200 mL of dry THF. The reaction mixture was refluxed for 3 h, and the solvent was removed *in vacuo*. The residue was taken up with ether and filtrated to give 3.77 g (74%) of product as a white powder: ¹H NMR (CDCl₃) δ 7.73 (s, 1H, ArH), 7.28 (s, 1H), 7.24 (s, 15H, tritylH), 7.21 (s, 1H), 6.95 (s, 1H), 6.21 (br s, 1H, NH), 5.60 (s, 1H, ArH), 2.98 (s, 3H, OCH₃).

(R)-3-Quinuclidinol (0.13 g, 1 mmol) and DBU (0.15 mL, 1 mmol) were added dropwise to a solution of 0.51 g (1 mmol) of the previous compound in dry THF. The reaction mixture was refluxed overnight. The solvent was removed in vacuo, and the residue was taken up in CH₂Cl₂ and washed with water. The organic layer was dried over MgSO₄ and evaporated to dryness to give a solid which was purified by chromatography on silica gel and eluted with a CHCl₃/MeOH (95:5) mixture to give the pure R-tritylated benzoate derivative (64%). The ester (0.8 g, 1.44 mmol) was dissolved in pure CHCl₃, and 2 equiv of concentrated HCl was added. After stirring for 2 h at 25 °C, the solvent was removed *in vacuo* and the corresponding amino derivative was purified on a silica gel column and eluted with CHCl₃/MeOH (80:20) to give the pure hydrochloride salt **9c** (60%): mp 244 °C; TLC R_f (on silica gel) 0.28 (CH₂Cl₂/ MeOH, 80:20); $[\alpha]_{546} = -36.8^{\circ}$ (c = 1.0, MeOH); ¹H NMR (CD₃-OD) & 7.91 (s, 1H, ArH), 6.64 (s, 1H, ArH), 5.31 (m, 1H), 4.18 (m, 1H), 3.98 (s, 3H, OCH₃), 3.6–3.4 (m, 5H), 2.6–2.1 (m, 5H). Anal. ($C_{15}H_{19}N_2O_3Cl\cdot HCl$).

(*S*)-1-Azabicyclo[2.2.2]octan-3-yl 4-Amino-5-chloro-2methoxybenzoate (9d). 9d was prepared from (*S*)-3-quinuclidinol (1 mmol) as described for 9c. After the column chromatography purification, the pure hydrochloride salt was obtained (59%): mp 240 °C; $[\alpha]_{546} = +40.6^{\circ}$ (*c* = 1.0, MeOH); ¹H NMR (CD₃OD) δ 7.91 (s, 1H, ArH), 6.64 (s, 1H, ArH), 5.31 (m, 1H), 4.18 (m, 1H), 3.98 (s, 3H, OCH₃), 3.6–3.4 (m, 5H), 2.6–2.1 (m, 5H). Anal. (C₁₅H₁₉N₂O₃Cl·HCl).

2-(Dimethylamino)ethyl 4-Amino-5-chloro-2-methoxybenzoate (9a). The compound was prepared according to the previous process reported for **9c** from 2-(dimethylamino)ethanol and the benzoic acid **11**. It was isolated as a hydrochloride salt after recrystallization from the *i*-Pr₂O/ MeOH mixture (global yield 13%): mp 230 °C; ¹H NMR (CDCl₃) δ 7.82 (s, 1H, ArH), 6.47 (s, 1H, ArH), 4.54 (m, 2H, OCH₂), 3.8 (s, 3H, OCH₃), 3.49 (m, 2H, N-CH₂), 2.94 (s, 6H, N(CH₃)₂). Anal. (C₁₂H₁₈N₂O₃Cl·2H₂O).

8-Methyl-8-azabicyclo[3.2.1]octan-3-yl 4-Amino-5-chloro-2-methoxybenzoate (9e). It was prepared according to the previous process reported for **9c** from tropine and the benzoic acid **11** with modifications. BuLi in THF at -10 °C was used in place of DBU. Compound **9e** was isolated as a hydrochloride salt after recrystallization from the MeOH/*i*-Pr₂O mixture (global yield 19%): mp >260 °C; ¹H NMR (CD₃OD) δ 7.71 (s, 1H, ArH), 6.51 (s, 1H, ArH), 5.21 (m, 1H), 4–3.85 (m, 2H), 3.82 (s, 3H, OCH₃), 2.82 (s, 3H, N-CH₃), 2.65–2.3 (m, 6H), 2.27–2.07 (m, 2H). Anal. (C₁₆H₂₁N₂O₃Cl·HCl·H₂O).

4-Amino-5-chloro-2-methoxy-N-methyl-N-(7-piperidinoethyl)benzamide Hydrochloride (9g). Triethylamine (1 g, 9.9 mmol) was added to a suspension of 4-amino-5-chloro-2-methoxybenzoic acid (2 g, 9.9 mmol) in dry THF (100 mL). The mixture was heated to dissolve the reactants and cooled to between -5 and -10 °C. Ethyl chloroformate (1.07 g, 9.9 mmol) was slowly added, and the mixture was stirred for 30 min. A solution of N-methyl-2-(1-piperidinyl)ethylamine (1.54 g, 9.9 mmol) in dry CH_2Cl_2 (5 mL) was added slowly, the temperature was allowed to rise to 25 °C, and the mixture was left stirring overnight. After evaporation of the solvent, the residue was taken up with water and extracted with CH₂Cl₂. The organic layers were washed with NaOH (4%) and brine and then dried over Na2SO4 and evaporated to afford a pale yellow oil which was purified by column chromatography on silica gel by eluting with CH₂Cl₂/MeOH (90:10) to afford 0.62 g (19%) of the free base. Treatment of the base with a 2 N HCl/MeOH solution and precipitation by *i*-Pr₂O give a solid which was recrystallized from the *i*-Pr₂O/MeOH mixture to yield 0.38 g (11%) of the hydrochloride salt: mp 254–256 °C; $R_f 0.098$ (CH₂Cl₂/MeOH, 90:10); ¹H NMR (CD₃OD) δ 7.19 (s, 1H, ArH), 6.53 (s, 1H, ArH), 3.87 (3.87, J = 6 Hz, 3H), 3.8 (s, 3H, OCH₃), 3.68 (m, 2H), 3.37 (t, *J* = 6 Hz, 3H), 3.06 (m, 2H), 2.96 (s, 3H, N-CH₃), 2.1-1.4 (m, 6H). Anal. (C₁₆H₂₄O₂ClN₃. HCl·0.5H₂O).

4-Amino-5-chloro-*N*-[**2-(dimethylamino)ethyl]-2-meth-oxybenzamide Hydrochloride (9b).** This compound was synthesized from *N*,*N*-dimethylethylenediamine (0.88 g, 0.01 mol) according to the previous method described for **9g**. The free base was treated with a solution of HCl (3 N) in AcOEt to give **9b** as a hydrochloride salt (2.3 g, 77%): mp 220 °C; ¹H NMR (CD₃OD) δ 7.8 (s, 1H, ArH), 6.65 (s, 1H, ArH), 3.8 (s, 3H, OCH₃), 3.6 (t, *J* = 5.7 Hz, 2H), 3.2 (t, *J* = 5.7 Hz, 2H), 2.8 (s, 6H, CMe₂). Anal. (C₁₂H₁₈N₃O₂Cl·HCl).

4-Amino-5-chloro-N-[2-(3,5-dimethylpiperidino)ethyl]-2-methoxybenzamide (9h). A mixture of 2-bromoethylamine hydrobromide (2.05 g, 10 mmol), the commerciallyavailable mixture of cis- and trans-3,5-dimethylpiperidine (2.26 g, 20 mmol), and K₂CO₃ (1.38 g, 10 mmol) in MeOH (20 mL) was stirred for 3 h at room temperature. After evaporation of MeOH, the residue was dissolved in CH₂Cl₂ and water. The organic layer was washed with brine and dried over MgSO₄. Evaporation of the solvent under reduced pressure afforded cis, trans-1-(2-aminoethyl)-3,5-dimethylpiperidine as a colorless oil, which was then diluted with THF. This solution was added dropwise to a mixture of Et₃N (1.01 g, 10 mmol), ClCO₂Et (1.08 g, 10 mmol), and 4-amino-5-chloro-2-methoxybenzoic acid (2.01 g, 10 mmol) in THF (50 mL) at 0 °C for 30 min. The reaction mixture was warmed to room temperature and stirred for 4 h. The solvent was removed, and the residue was partitioned between aqueous AcOEt and Na₂CO₃ solutions. The organic layer was washed with brine, dried over MgSO₄, and evaporated. The residue was recrystallized from ether to yield 9h (0.8 g, 24%) as a mixture of the *cis/trans* isomers (70:30): mp 178–179 °C; ¹H NMR (CDCl₃) δ 8.2 (m, 1H, CONH), 8.1 (s, 1H, ArH), 6.2 (s, 1H, ArH), 4.4 (s, 2H, NH₂), 3.8 (s, 3H, OCH₃), 3.5 (m, 2H, NHCH₂), 2.8 (m, 1.4H_{eq}, cis-2,6-piperidinyl H), 2.55-2.35 (m, 2H, CH₂N), 2.3-1.2 (m, 6H), 0.95 (d, J =6 Hz, 1.8H, trans-dimethyl), 0.85 (d, J = 6 Hz, 4.2H, cisdimethyl), 0.55 (m, 0.7H_{ax}, 4-piperidinyl H); ¹³C NMR (CDCl₃) δ 164.6 (C=O), 157.8 (C₂), 146.9 (C₄), 133.2 (C₆), 112.9-111.6 (C₁-C₅), 98.1 (C₃), 61.6 and 61.1, 57.4 and 57.0, 56.2 (OCH₃), 42.5, 39.4, 37.0, 31.7 and 27.7, 19.8 and 19.4. Anal. $(C_{17}H_{26}N_3O_2Cl).$

4-Amino-5-chloro-*N***·(2-piperidinoethyl)-2-methoxybenzamide (9f).** This compound was synthesized from 2-piperidinoethylamine and 4-amino-5-chloro-2-methoxybenzoic acid according to the process described for **9g** and isolated as an amorphous solid (0.6 g, 27%): ¹H NMR (CDCl₃) δ 8.2 (br s, 1H, CONH), 8.08 (s, 1H, ArH), 6.3 (s, 1H, ArH), 4.4 (s, 2H, NH₂), 3.85 (s, 3H, OCH₃), 3.5 (m, 2H, NHCH₂), 2.48 (t, J = 6 Hz, 2H), 2.4 (m, 4H, CH₂N), 1.35–1.65 (m, 6H); ¹³C NMR (CDCl₃) δ 164.32 (C=O), 157.6 (C₂), 146.48 (C₄), 133.02 (C₆), 113.04–111.54 (C₁–C₅), 97.97 (C₃), 57.25, 56.01, 54.28, 36.66, 26.31, 24.48. Anal. (C₁₅H₂₂N₃O₂Cl).

5-Chloro-2-methoxy-4-(tritylamino)benzaldehyde (12). 11 (14 g, 32 mmol) in anhydrous THF (150 mL) was added dropwise to a magnetically-stirred suspension of LiAlH₄ (1.63 g, 43 mmol) in anhydrous THF (60 mL) cooled with an ice bath. After the end of the addition, the mixture was left at room temperature for 4 h. The excess of LiAlH₄ was quenched by the successive dropwise additions of 1.63 mL of water, 1.63 mL of 15% aqueous NaOH, 1.5 mL of water, and solid MgSO₄. The solution was filtered through Celite, and the filtrate was concentrated under reduced pressure. The residue was purified by column chromatography on neutral alumina by eluting successively with CH2Cl2 and CH2Cl2/MeOH (95:5) to yield 8.09 g (59%) of 5-chloro-2-methoxy-4-(tritylamino)benzyl alcohol as a white foaming solid: Rf 0.62 (CH2Cl2/MeOH, 95:5); ¹H NMR (CDCl₃) δ 7.4–7.18 (m, 15H, tritylH), 7.09 (s, 1H, ArH), 5.82 (br s, 1H, NH), 5.67(s, 1H, ArH), 4.42 (d, 2H, CH₂OH), 3.14 (s, 3H, OCH₃), 1.97 (t, 1H, OH). Anal. (C₂₇H₂₄NO₂Cl).

MnO₂ (1.4 g, 16.2 mmol) was added to a solution of the alcohol prepared above (0.7 g, 1.62 mmol) in anhydrous CH₂Cl₂ (20 mL). The reaction mixture was stirred at room temperature for 24 h. Then the suspension was filtered through Celite and the filtrate concentrated *in vacuo*. The yellow foamy solid obtained was recrystallized from hexane/CH₂Cl₂ to give 0.59 g (86%) of **12** as white-yellow prisms: mp 198 °C; R_f 0.8 (CH₂Cl₂); ¹H NMR (CDCl₃) δ 10 (s, 1H, CHO), 7.73 (s, 1H, ArH), 7.4–7.18 (m, 15H, tritylH), 6.4 (br s, 1H, NH), 5.58 (s, 1H, ArH), 3.17 (s, 3H, OCH₃). Anal. (C₂₇H₂₂-NO₂Cl).

1-[5-Chloro-2-methoxy-4-(tritylamino)phenyl]-4-piperidinobutanol (13). Magnesium turnings (0.34 g, 14 mmol) were covered with 3 mL of dry THF under an Ar atmosphere. Iodine crystals and 1 or 2 drops of ethyl bromide were added. As soon as the reaction had started, the suspension was heated at 60 °C and 1-(3-chloropropyl)piperidine (2.27 g, 14 mmol) in 3 mL of dry THF was added dropwise to maintain gentle reflux. When the addition was over, the reaction mixture was refluxed with stirring for 30 min and cooled to 0 °C. Compound 12 (3 g, 6.97 mmol) was added portionwise, and the mixture was heated at 60 °C for 24 h. The reaction mixture was hydrolyzed by a solution of 16.5 g of ammonium chloride in 40 mL of water. The organic layer was separated, the aqueous phase was extracted with ether (3×60 mL), and the combined extracts were dried over MgSO₄. After concentration in vacuo, the residue was purified by column chromatography on neutral alumina with AcOEt as the eluant to give 13 as a foamy white solid (1.7 g, 44%): R_f (neutral alumina) 0.43 (AcOEt); ¹H NMR (CDCl₃) δ 7.4–7.18 (m, 16H, ArH, tritylH), 5.7 (br s, 1H, NH), 5.61 (s, 1H, ArH), 4.7-4.6 (m, 1H, ArCHO), 3.08 (s, 3H, OCH₃), 2.7-2.6 (m, 6H), 1.9-1.3 (m, 10H). Anal. $(C_{35}H_{39}N_2O_2Cl \cdot 0.5H_2O).$

1-(5-Chloro-2-methoxy-4-aminophenyl)-4-piperidino**butanone (14).** MnO_2 (2.6 g, 30 mmol) was added to the previous alcohol 13 (1.7 g, 3 mmol) in 50 mL of dry CH₂Cl₂. The mixture was stirred overnight at room temperature and heated at 40–50 °C for 7 days in the presence of an additional quantity of MnO₂ (2.6 g). The mixture was filtered through Celite and purified on neutral alumina using AcOEt as the eluant to yield a yellow-orange solid (1.6 g). The compound was dissolved in a mixture of 40 mL of dry acetone with 2.1 equiv of 12.4 N HCl and stirred at room temperature for 2 h. After evaporation of the solvent, the ketone was converted into the hydrochloride salt with a 2 N HCl/MeOH solution and precipitated by the addition of *i*-Pr₂O. It was recrystallized from *i*-Pr₂O/MeOH to afford 0.26 g (25%) of a clear pink solid: mp 242–244 °C; R_f 0.074 (CH₂Cl₂/MeOH, 90:10); ¹H NMR (CD₃OD) & 7.64 (s, 1H, ArH), 6.37 (s, 1H, ArH), 3.78 (s, 3H, OCH₃), 3.4-2.8 (m, 6H), 2.05-1.86 (m, 2H, ArCOCH₂), 1.86-1.4 (m, 6H); ¹³C NMR (CD₃OD) δ 198.07 (C=O), 162.06, 152.06, 132.608, 117.3, 111.77, 98.03, 58.09, 56.19 (OCH₃), 54.37, 41.048, 24.35, 22.76, 20.12; IR (cm⁻¹, CH₂Cl₂) 1620 (C=O); MS (CI, NH₃) m/z 312 (MH⁺) (free base, M = 311). Anal. (C $_{16}H_{23}N_2O_2Cl$ ·HCl·0.25H₂O).

Conformational Studies. Conformational studies were performed using the SYBYL 6.03 and 6.1 programs of Tripos. Energies were calculated using the Tripos force field and MOPAC charges. Current minimizations were performed using a gradient termination of 0.05 kcal mol⁻¹. Conformational random search studies were performed using 1000 cycles and an energy cutoff of 30 kcal mol⁻¹. Conformational grid searches were performed with rotational steps of 20°.

X-ray Crystallography. A selected crystal was set up on a Nonius CAD4 automatic diffractometer. Unit cell dimensions with estimated standard deviations were obtained from least-squares refinements of the setting angles of 25 wellcentered reflections. Two standard reflections were monitored periodically and showed no change during data collection. Computations were performed using CRYSTALS (University of Oxford, U.K.). The structures were solved by a direct method using the SHELX 86 (University of Göttingen, Germany) program and successive Fourier maps. For compounds 7a,g, all the hydrogen atoms could be found on difference maps. Non-hydrogen atoms were not refined, and they were given overall isotropic thermal parameters. Full matrix leastsquares refinements were carried out by minimizing the function $\sum w(|F_0| - |F_c|)^2$, where F_0 and F_c are the observed and calculated structure factors. Unit weight was used. Models reached convergence with $R = \sum (||F_0| - |F_c|)/\sum |F_0|$ and $R_w =$ $[\Sigma w(|F_0| - |F_c|)^2 / \Sigma w(F_0)^2]^{1/2}$ having values of 0.054 and 0.052, respectively, for 7a and values of 0.044 and 0.046, respectively, for 7g. Criteria for a satisfactory complete analysis were the ratios of the rms shift to standard deviation being less than 0.1 and no significant features in the final difference map.

7a: $C_{15}H_{21}N_{2}O_{3}Cl$ ·HCl, MW = 349.26, crystals are parallelepiped and colorless and belong to the monoclinic system, space group *P*21/*c*; cell parameters, a = 12.490 Å, b = 10.805 Å, c = 12.838 Å; $\beta = 92.53^{\circ}$, V = 1731 Å³, d = 1.34 g/cm³.

7g: $C_{17}H_{25}N_2O_3CI$ ·HCl, MW = 377.31, crystals are parallelepiped and colorless and belong to the monoclinic system, space group *P*21/*c*, cell parameters, a = 6.887 Å, b = 19.926Å, c = 14.219 Å; $\beta = 96.52^{\circ}$, V = 1948 Å³, d = 1.29 g/cm³.

Tables of atomic coordinates, bond distances, and bond angles are available from the authors.

Pharmacological Methods. 1. Binding Assays. Male Sprague–Dawley rats from Janvier Laboratories (France) were used. Animals were housed at 22 ± 1 °C, with 55% humidity, on a 12 h light–dark cycle with free access to food and water for 4 days before the experiments.

2. 5-HT₃ Receptors. Membranes were prepared from rat posterior cortex according to the procedure described by Hall.⁴⁹ [³H]BRL 43694 (61 Ci/mmol) was purchased from NEN Research Products. GR 38032F was a generous gift from Glaxo (U.K.). All other chemicals and reagents were commercially available from Sigma. The binding of 1.2 nM [³H]-BRL 43694 ($K_D = 1.5$ nM, $B_{max} = 30$ fmol/mg of protein for 5-HT₃ receptors) was measured using membranes (100 μ L aliquots equivalent to 0.95 mg of protein) suspended in a final volume of 0.5 mL of 50 mM Hepes buffer, pH 8.4, and incubated at 25 °C for 30 min; 7–11 concentrations of each drug were used in triplicate. Nonspecific binding was determined by the addition of 10 μ M GR 38032F in duplicate and represented less than 50% of the total binding. Total binding was defined in quadruplicate.

3. 5-HT₄ Receptors. Membranes were prepared from rat striatum and olfactory tubercules which were pooled separately and stored at -80 °C. Tissues were thawed at 0 °C and homogenized in 15 vol of ice-cold 50 mM Hepes (pH 7.4) using a Polytron homogenizer and then centrifuged once at 40000*g* at 4 °C for 15 min. The resulting pellet was resuspended in 5 vol of Hepes to a final concentration of 4-5 mg of protein/mL. Membrane aliquots of 2.6 mL were kept frozen at -80 °C for subsequent use.

Binding assays were performed according to the method described^{12a} previously with modifications. The binding of [³H]-GR 113808 (85 Ci/mmol; purchased from Amersham) was measured using membranes (50 μ L aliquots equivalent to 0.1–0.2 mg of protein) suspended in a final volume of 0.5 mL of 50

mM Hepes buffer (pH 7.4) and incubated at 25 °C for 30 min. Seven concentrations of each drug were used, and the assay was done in triplicate. Nonspecific binding was defined with 10 μ M **7a** in duplicate and represented less than 10% of the total binding. Total binding was defined in quadruplicate.

For each assay the bound radioactivity was separated by vacuum filtration through Whatman GF/B glass filters, presoaked in 0.1% poly(ethylenimine), using a Brandel cell harvester. The filters were then washed twice with 5 mL of 50 mM Tris-HCl (pH 8.4) at room temperature and dried. The filters were placed in poly(ethylene) vials to which was added 4 mL of a scintillation cocktail (Beckman, Ready-Safe), and after equilibration, the radioactivity was determined using liquid scintillation spectrometry. The data were analyzed by a computer-assisted curve-fitting program in Lotus 1.2.3 to provide IC₅₀, K_i , and r^2 values, K_i values being calculated from the Cheng–Prussof equation.

4. Protein Estimation. The protein concentration of the rat posterior cortex membranes was determined by the method of Lowry⁵⁰ using bovine serum albumin as the standard. The protein concentrations of the rat striatum and olfactory tubercle membrane preparations were determined by the method of Bradford.⁵¹

5. 5-HT₄ Receptor Activity. Myenteric Plexus and Longitudinal Muscle Preparation from Guinea Pig Ileum. Tissues were obtained from male Crl (HA) SPF guinea pigs (400–600 g). According to Craig and Clarke,^{4a} segments (2 cm long, 10 cm from the ileocecal junction) of myenteric plexus and longitudinal muscle from the ileum were carefully dissected and mounted in a 20 mL organ bath containing a warm (37 °C), aerated (95% O₂, 5% CO₂) Krebs solution. The tissue was stretched with a 0.5 g weight, and contractions elicited by transmural electrical stimulation (0.2 Hz, pulse duration 1.5 ms) were recorded by means of an isotonic transducer connected to a polygraph (Gemini, Basile, Italy). The preparations were subjected to supramaximal voltage stimulation. After stabilization and incubation with 10⁻⁷ M phenoxybenzamine, the voltage was reduced to obtain contractions 50% of those elicited by supramaximal stimulation. In control experiments, at least three fully reproducible concentration-effect curves to 5-HT (contact time about 60 s) were obtained. After two fully reproducible cumulative concentration-effect curves to 5-HT, a concentration-effect curve to the 5-HT₄ receptor agonist under test was carried out (contact time 60-120 s) and related to the maximal effect of 5-HT (100%). In antagonist studies, the antagonist was added 30 min before the last agonist concentration-effect curve, several concentrations of antagonists were tested, and the IC₅₀ value was calculated.

Rat Esophageal Tunica Muscularis Mucosae. Tissues were obtained from male Crl (CD) Br rats (225-250 g). According to Baxter and Clarke,^{4a} the rat esophageal segment (2 cm of intrathoracic esophagus, proximal to the diaphragm) was removed and the outer muscle layers were carefully dissected away. The segment, i.e., the tunica muscularis mucosae, was mounted, under 1 g of tension, in a tissue bath containing a warm (37 °C), aerated (95% O₂, 5% CO₂) Krebs solution, and contractions were recorded by means of an isotonic transducer connected to a polygraph. After 60 min of equilibration, tissues were incubated with 10^{-4} M pargyline (30 min). In order to reach a steady state of tonus, preparations were contracted with cumulative submaximal concentrations of carbachol ($10^{-6}-4 \times 10^{-6}$ M). In control experiments, three fully reproducible contractions to carbachol and then three concentration-effect curves to 5-HT (inhibition of carbachol contractions) were carried out (contact time about 60 s). After two reproductible cumulative concentration-effect curves to 5-HT, a concentration-effect curve to the 5-HT₄ receptor agonist under test (contact time 60-120 s) was carried out cumulatively and related to the maximal effect of 5-HT (100%). In antagonist studies, the compound was added 30 min before the last agonist concentration-effect curve.

The pA_2 values, as defined by Arunlakshana and Schild,^{41a} were obtained from the plot (Cr - 1) vs negative log of the antagonist concentration. The computer analyses were done as described by Tallarida and Murray.^{41b}

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