

Palladium-Catalyzed Alkynylative Lactonization of Unsaturated Bicyclic Carboxylic Acids: Synthesis of Fused Polycyclic γ -Lactone Compounds

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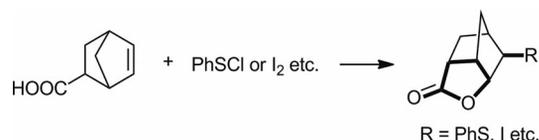
Keywords: Synthetic methods / Polycycles / Lactones / Alkynes / Palladium

Unsaturated bicyclic carboxylic acids undergo palladium-catalyzed coupling with bromoalkynes to produce γ -alkynyl lactones in moderate to good yields with excellent chemo- and regioselectivity. The reaction conditions were extremely mild, and functional groups such as methyl, methoxy, chloro,

and bromo were tolerated under the optimized reaction conditions. Moreover, γ -chloroalkenyl lactones were readily synthesized through chloropalladation/carboesterification of electron-deficient C=C bonds by using molecular oxygen as the sole oxidant.

Introduction

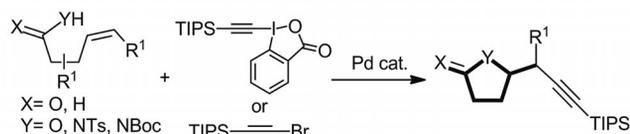
The lactone unit is one of the most abundant and relevant esters in natural products and pharmaceuticals. Despite the existence of numerous methods for the synthesis and derivatization of lactone rings, the development of new, more efficient methods is of great importance.^[1] Haloalkynes represent an attractive alternative to aryl halides or terminal alkyne derivatives because of their higher reactivity, tolerance under milder conditions, their availability from inexpensive acetylides, and because additional oxidant and base are not required in several applications.^[2] Moreover, highly functionalized alkenynes have been prepared through the bromoalkynylation of internal alkynes.^[3] Further research on haloalkynes has revealed other possibilities for the synthesis of complex molecules. Norbornene derivatives are an appealing group of organic molecules because of their strained structure and high reactivity. They tend to coordinate to transition metals and have potential applications in various areas of industry; as such, they have attracted considerable research interest from organic chemists.^[4] To date, various synthetic methods have been developed for the synthesis of phenylsulfenyl lactones and halolactones through intramolecular lactonization of unsaturated carboxylic acids (Scheme 1).^[5]



Scheme 1. Usual lactonization procedure for bicyclic carboxylic acids.

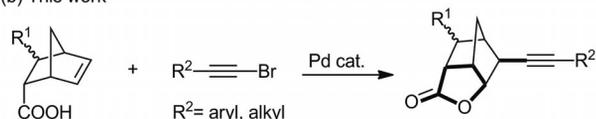
Recent research into the transition-metal-catalyzed reactions of norbornenes by the group of Jiang^[6] and others^[7] led us to investigate the intermolecular alkynylation and carboesterification of this class of molecules. The Pd-catalyzed intramolecular oxy- and aminoalkynylation of nonactivated olefins with the use of a benziodoxolone-derived hypervalent iodine reagent or (2-bromoethynyl)triisopropylsilane as an alkyne transfer reagent was reported by Waser and co-workers for the preparation of tetrahydrofurans and pyrrolidines (Scheme 2, a).^[8] Unfortunately, when using phenylethynyl bromides or phenylethylethynyl bromides as substrates, complex mixtures of products were obtained.

(a) Reported work



Ts = *para*-toluenesulfonyl, Boc = *tert*-butoxycarbonyl, TIPS = triisopropylsilyl

(b) This work



Scheme 2. Pd-catalyzed oxyalkynylation of olefins with haloalkynes.

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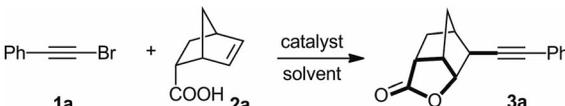
Supporting information for this article is available on the WWW under <http://dx.doi.org/10.1002/ejoc.201300037>.

Herein, we report a general protocol for the alkynylation and carboesterification of unsaturated bicycle carboxylic acids to give functional γ -alkynyl lactone derivatives (Scheme 2, b). This approach is another utilization of alkyl or aryl bromoalkynes for the formation of lactone rings under palladium catalysis. As such, bicycle alkynyl lactones may be readily converted into alkynyl hydrins, γ -oxocarboxylic acids, and epoxides.^[9] Furthermore, we extended this methodology to other internal electron-deficient alkynes and successfully obtained the corresponding chloropalladation/carboesterification products in good yields.

Results and Discussion

At the onset, through the screening of a variety of conditions, we found that several palladium salts could effect the alkynylative lactonization of phenylethynyl bromide (**1a**) with *endo*-norborn-5-ene-2-carboxylic acid (**2a**, Table 1). In experiments to optimize the reaction, we found that the solvent played a very important role in the process (Table 1, entries 1–3). The best results were obtained with a catalytic amount of Pd(OAc)₂ in CH₃CN. Under these conditions, conversion was complete within 8 h at 50 °C without the addition of base or ligand (Table 1, entry 3). The alkynylative lactonization of **2a** with **1a** under Pd(PPh₃)₄ catalysis led to low conversion and the starting material was recovered (Table 1, entry 5). The use of Pd₂(dba)₃ as the catalyst gave good results (Table 1, entry 6). The addition of a base such as K₂CO₃ did not disturb the reaction (Table 1, entry 7). We next focused on PPh₃ and 1,10-phenanthroline (phen) as ligands, and both hampered the oxyalkynylation process (Table 1, entries 8 and 9). Notably, the reaction was sluggish when CuI was employed instead of a palladium salt (Table 1, entry 10). The reaction did not occur without a palladium catalyst (Table 1, entry 11).

Table 1. Optimization of the reaction conditions.^[a]

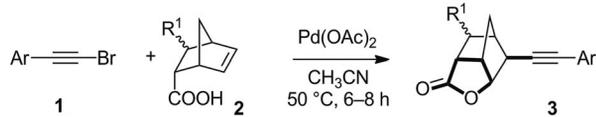


Entry	Catalyst	Solvent	Additive	Yield [%] ^[b]
1	Pd(OAc) ₂	DMSO	–	35
2	Pd(OAc) ₂	NEt ₃	–	–
3	Pd(OAc) ₂	CH ₃ CN	–	93
4	PdCl ₂	CH ₃ CN	–	83
5	Pd(PPh ₃) ₄	CH ₃ CN	–	42
6	Pd ₂ (dba) ₃	CH ₃ CN	–	91
7	Pd(OAc) ₂	CH ₃ CN	K ₂ CO ₃	85
8	Pd(OAc) ₂	CH ₃ CN	PPh ₃	45
9	Pd(OAc) ₂	CH ₃ CN	phen	28
10 ^[c]	CuI	CH ₃ CN	–	–
11 ^[d]	–	CH ₃ CN	–	–

[a] Reaction conditions: phenylethynyl bromide (1.0 mmol), norbornene (1.5 mmol), catalyst (5 mol-%), and additive (10 mol-%) in solvent (3 mL) at 50 °C for 8 h. [b] Isolated yield. [c] 1,4-Diphenylbuta-1,3-diyne was obtained as the major product. [d] Starting material was recovered.

Using the Pd(OAc)₂-catalyzed system (Table 1, entry 3), we examined the generality of the synthesis of substituted fused polycyclic γ -lactone compounds. Varying the aryl substituent of bromoalkynes **2** (Table 2) showed that substrates bearing electron-donating (Table 2, **3b–g**) and electron-withdrawing groups (Table 2, **3h–l**) were well tolerated. Notably, the C–Br bond remained intact when a bromine substituent was introduced on the aryl ring, and therefore the product may undergo additional functionalization at the C–Br bond (Table 2, **3k** and **3p**). Next, the effects of the unsaturated carboxylic acid moiety in the synthesis of γ -alkynyl lactones were examined. The reaction of 3-(methoxycarbonyl)bicyclo[2.2.1]hept-5-ene-2-carboxylic acid under the standard reaction conditions afforded the corresponding γ -alkynyl lactones in good yields (Table 2, **3n–p**). In addition to aryl bromoalkynes, aliphatic bromoalkynes were also found to be suitable substrates for the alkynylative lactonization reaction (Scheme 3). When 2-(6-bromohex-5-ynyl)isoindoline-1,3-dione, 1-(2-bromoethynyl)cyclohex-1-ene, 1-bromooct-1-yne, and 1-(3-bromoprop-2-ynyl)benzene were employed, the desired products were formed and isolated in excellent yields (Scheme 3, **3r–u**).

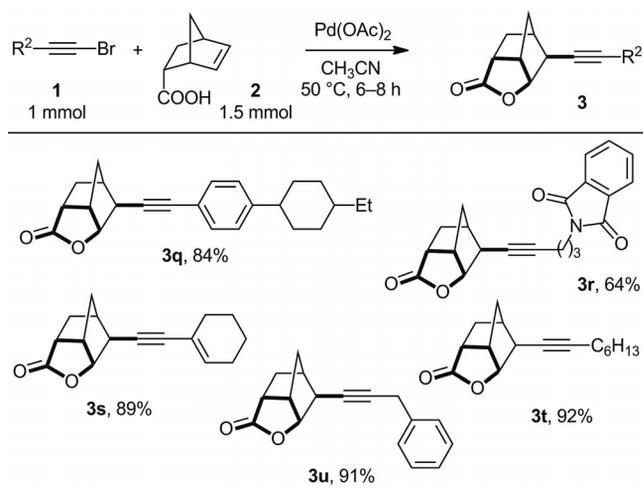
Table 2. Scope of the alkynylative lactonization reaction.^[a]



Entry	Ar	R ¹ (2)	Product	Yield [%] ^[b]
1	C ₆ H ₅	H (2a)	3a	93
2	4-MeC ₆ H ₄	H (2a)	3b	95
3	3-MeC ₆ H ₄	H (2a)	3c	92
4	2,4-Me ₂ C ₆ H ₃	H (2a)	3d	83
5	4-EtC ₆ H ₄	H (2a)	3e	87
6	4- <i>t</i> BuC ₆ H ₄	H (2a)	3f	95
7	4-EtOC ₆ H ₄	H (2a)	3g	90
8	2-ClC ₆ H ₄	H (2a)	3h	83
9	3-ClC ₆ H ₄	H (2a)	3i	87
10	4-ClC ₆ H ₄	H (2a)	3j	88
11	4-BrC ₆ H ₄	H (2a)	3k	87
12	4-FC ₆ H ₄	H (2a)	3l	86
13	4-PhC ₆ H ₄	H (2a)	3m	92
14	4-EtC ₆ H ₄	COOMe (2b)	3n	85
15	2-ClC ₆ H ₄	COOMe (2b)	3o	81
16	4-BrC ₆ H ₄	COOMe (2b)	3p	83

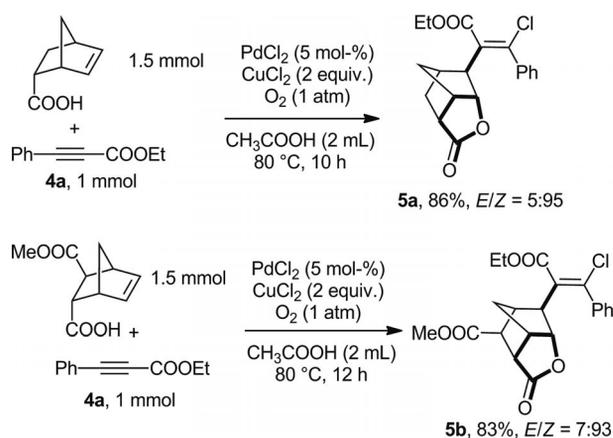
[a] Reaction conditions: Pd(OAc)₂ (5 mol-%), norbornene (1.5 mmol), bromoalkyne (1.0 mmol), and CH₃CN (3 mL) at 50 °C for 6–8 h. [b] Isolated yield.

The above studies dealt only with bromoalkynes as the reactive point in the substrates. To extend the substrate scope, we used other electronic-deficient compounds, such as ethyl 3-phenylpropiolate, to investigate the possibility of this transformation (Scheme 4). The reaction of *endo*-norborn-5-ene-2-carboxylic acid with ethyl 3-phenylpropiolate afforded chloropalladation/carboesterification product **5a** with high *Z/E* selectivity. In addition, 3-(methoxycarbonyl)-

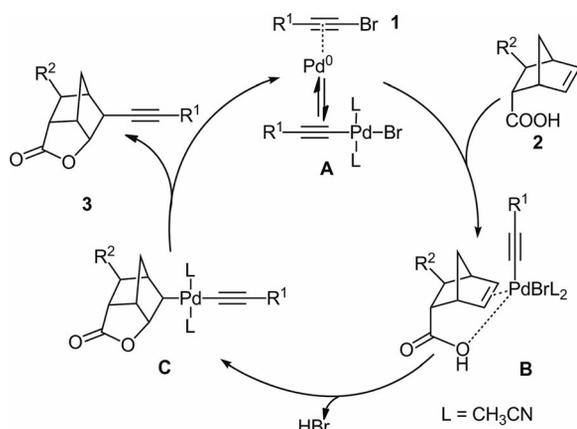


Scheme 3. Pd-catalyzed oxyalkynylation of bicyclo[2.2.1]hept-5-ene-2-carboxylic acid with bromoalkynes.

bicyclo[2.2.1]hept-5-ene-2-carboxylic acid was equally adaptable to the applied reaction conditions and formed corresponding product **5b** in good yield.



Scheme 4. Synthesis of γ -chloroalkenyl lactone products.



Scheme 5. Proposed mechanism for the alkylnylative lactonization process.

The basic mechanistic features of the alkylnylative lactonization reaction are reasonably well established and are shown in Scheme 5.^[6,8] Bromoalkyne **1** first undergoes oxidative addition with the low-valent palladium to give alkylnyl palladium(II) intermediate **A**. Intermediate **A** coordinates to norbornene **2** to generate intermediate **B**, which is followed by *trans* oxypalladation to give alkylnyl alkyl palladium intermediate **C** with well-defined stereochemistry. Subsequent reductive elimination of **C** generates γ -alkynyl lactone product **3** and the active catalyst species.

Conclusions

In conclusion, we have developed the Pd-catalyzed alkylnylative lactonization of unsaturated bicyclic carboxylic acids by using common bromoalkynes as substrates. This protocol is compatible with a large variety of bromoalkynes, and fused polycyclic γ -alkynyl lactones and γ -chloroalkenyl lactones were obtained in good yields. Further synthetic application of these fused polycyclic γ -alkynyl lactones and direct alkylnylative lactonization of straight-chain unsaturated carboxylic acids is ongoing. The results will be reported in due course.

Experimental Section

Typical Procedure for the Reaction of Bromoalkynes with Unsaturated Bicyclic Carboxylic Acids: Pd(OAc)₂ (12 mg, 0.05 mmol), CH₃CN (3 mL), *endo*-norborn-5-ene-2-carboxylic acid (208 mg, 1.5 mmol), and phenylethyne bromide (180 mg, 1.0 mmol) were added successively to a Schlenk tube. After stirring for 8 h at 50 °C, the solution was filtered through a small amount of silica gel. The residue was purified by silica gel preparative TLC (*n*-hexane/ethyl acetate = 10:1), which furnished **3a** (221 mg, 93%) as a pale-yellow oil.

Supporting Information (see footnote on the first page of this article): Full experimental procedures, characterization data, ¹H NMR and ¹³C NMR spectra, MS data, and IR spectra.

Acknowledgments

The work was financially supported by the Guangdong Science and Technology Department of China through the Youth Science Research Fund of Wuyi University (Key Project No. 2012B091000077, Major Program No. 2012A090300004). The authors thank the Jiang group for fruitful discussions and for recording the NMR spectra.

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Received: January 10, 2013

Published Online: March 21, 2013