

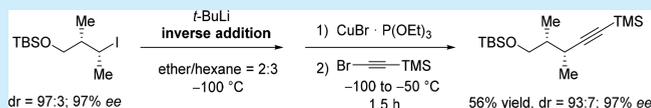
Diastereoselective Copper-Mediated Cross-Couplings between Stereodefined Secondary Alkylcoppers with Bromoalkynes

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S Supporting Information

ABSTRACT: A copper(I)-mediated cross-coupling of stereodefined secondary alkylolithiums with bromoalkynes provided stereodefined alkynes with high diastereoselectivity (dr up to 98:2). This cross-coupling was extended to various secondary alkylolithiums bearing a remote oxygen functionality, and the alkyne synthesis was also performed with optically enriched alkyl iodides (up to 99% ee) providing, after cross-coupling, alkynes bearing two stereocenters (dr = 93:7; up to 99% ee).



Cross-coupling reactions have been successfully used to prepare polyfunctional alkynes. Thus, the Sonogashira¹ and Negishi² cross-couplings allow linking alkynyl moieties with C_{sp}²-unsaturated halides in the presence of copper³ or palladium⁴ catalysts. The alkyne acts as a nucleophile in these reactions. Alternatively, 1-halogenoalkynes react with zinc organometallics or mixed copper–zinc reagents,⁵ providing alkylated alkynes. The use of catalytic amounts of copper salts⁶ or palladium catalysts⁷ in such cross-couplings has been reported. However, the preparation of alkynes bearing chiral centers in the α -position is still a challenge.⁸

Recently, we have reported that secondary alkyl iodides such as **1** undergo an I/Li exchange, leading to the corresponding secondary alkylolithium reagents of type **2** with retention of configuration using *t*-BuLi (2.5 equiv, inverse addition).⁹ We have shown that these secondary alkylolithium reagents undergo a stereoretentive transmetalation to the corresponding alkylcopper reagents of type **3**. The use of CuBr·P(OEt)₃¹⁰ as a soluble copper salt was essential for achieving high stereoselectivity.¹¹ These secondary alkylcopper reagents react with activated alkynes, allylic halides, and epoxides with retention of configuration.^{9d,11}

Herein, we report a successful cross-coupling reaction of stereodefined secondary alkylcopper reagents of type **3** with 1-bromoalkynes (**4**), leading to alkylated alkynes of type **5** bearing a chiral side chain with high retention of configuration (up to 98:2; Scheme 1).

In preliminary experiments, we have studied the thermal and configurational stability of such alkylcopper species **3**. Thus, we have treated the diastereomerically enriched secondary alkyl iodide (*anti*-**1a**; dr = 1:99) with *t*-BuLi (hexane–ether, –100 °C, 1 min) followed by the addition of CuBr·P(OEt)₃ (2 equiv, –100 °C), leading to the corresponding alkylcopper reagent *anti*-**3a** (Table 1). After stirring the solution at –70 °C for 10 min or for 1 h, benzoyl chloride (3.0 equiv)¹² was added at –70 °C, leading to the desired ketone *anti*-**6** in 59% yield (dr = 3:97; entry 1) and 38% yield (dr = 6:94; entry 2), respectively. This result indicated the high configurational stability of *anti*-**3a** at –70 °C.

Scheme 1. Enantioselective Synthesis of Stereodefined Alkynes Starting from Optically Enriched Iodides via I/Li-Exchange, Transmetalation to Copper Reagents, and Quench with 1-Bromoalkynes

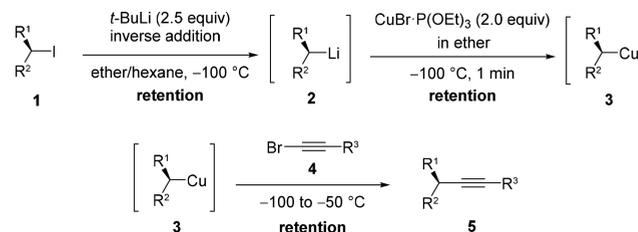
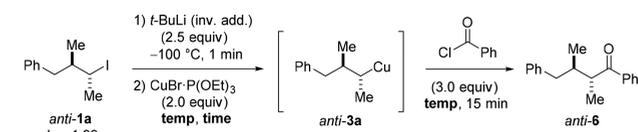


Table 1. Effect of Temperature and Time on the Stability of Secondary Alkylcopper Reagent *anti*-3a****



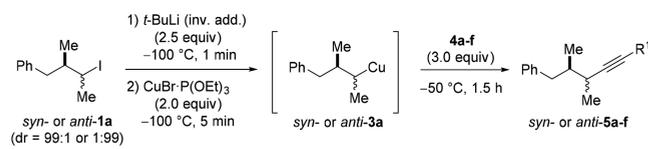
entry	temp (°C)	time (min)	yield (%) ^a	dr ^a
1	–70	10	59	3:97
2	–70	60	38	6:94
3	–50	10	50	5:95
4	–50	60	37	7:93
5	–30	10	41	6:94
6	–30	60	26	50:50

^aThe diastereoselectivity (dr; *syn/anti* ratio) and yield were determined by capillary GC using undecane as internal standard.

Stirring the copper intermediate (*anti*-**3a**) at –50 °C (for 10 min or 1 h, entries 3 and 4) or at –30 °C for 10 min (entry 5) led, after benzoylation, to a yield decrease of *anti*-**6**, but only to a marginal loss of diastereoselectivity. However, stirring *anti*-**3a**

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Table 2. Stereoselective Preparation of Secondary *syn*- and *anti*-Alkylated Alkynes (5) Obtained by the Cross-Coupling of *syn*- and *anti*-Alkylcopper (3a) with 1-Bromoalkynes of Type 4



entry	alkylcopper	electrophile	product of type 5 ^{a,b}
1	<i>syn</i> -3a	Br-C≡C-TMS 4a	<i>syn</i> -5a: 42%; dr = 93:7
2	<i>anti</i> -3a	Br-C≡C-TMS 4a	<i>anti</i> -5a: 66%; dr = 2:98
3	<i>syn</i> -3a	Br-C≡C-TIPS 4b	<i>syn</i> -5b: 29%; dr = 98:2
4	<i>anti</i> -3a	Br-C≡C-TIPS 4b	<i>anti</i> -5b: 46%; dr = 4:96
5	<i>syn</i> -3a	Br-C≡C-Ph 4c	<i>syn</i> -5c: 43%; dr = 93:7
6	<i>anti</i> -3a	Br-C≡C-Ph 4c	<i>anti</i> -5c: 51%; dr = 4:96
7	<i>syn</i> -3a	Br-C≡C-Bu 4d	<i>syn</i> -5d: 44%; dr = 97:3
8	<i>anti</i> -3a	Br-C≡C-Bu 4d	<i>anti</i> -5d: 46%; dr = 8:92
9	<i>syn</i> -3a	Br-C≡C-OEt 4e	<i>syn</i> -5e: 40%; dr = 97:3
10	<i>anti</i> -3a	Br-C≡C-OEt 4e	<i>anti</i> -5e: 43%; dr = 8:92
11	<i>syn</i> -3a	Br-C≡C-OEt 4f	<i>syn</i> -5f: 38%; dr = 89:11
12	<i>anti</i> -3a	Br-C≡C-OEt 4f	<i>anti</i> -5f: 51%; dr = 4:96

^aIsolated yield. ^bdr (*syn/anti* ratio) was determined by ¹H and ¹³C NMR analysis.

at $-30\text{ }^{\circ}\text{C}$ for 1 h led to a complete epimerization, as well as to a significant yield decrease (entry 6). This indicated that secondary alkylcoppers of type 3 are configurationally stable at $-30\text{ }^{\circ}\text{C}$ for only a short time.

After having determined the configurational stability of the alkylcopper reagents 3, we have performed a cross-coupling of the diastereomerically defined *syn*- and *anti*-secondary alkylcopper reagents (3a) with a range of alkynyl bromides of

type 4, leading to the corresponding *syn*- and *anti*-alkynes of type 5.

Typically, the secondary alkylcopper reagent (*syn*- or *anti*-3a) was generated as described above and treated with a bromoalkyne of type 4 (3.0 equiv) at $-100\text{ }^{\circ}\text{C}$. The reaction mixture was allowed to warm to $-50\text{ }^{\circ}\text{C}$ within 10 min and was stirred at this temperature for 1.5 h to ensure complete conversion. After workup, the alkynes 5a–f were isolated in 29–66% yields (Table 2). In most cases, a high retention of configuration between the two adjacent centers was observed, and only the cross-coupling of 4f with *syn*-3a led to more than 10% epimerization (entry 11).

This cross-coupling was extended to a range of secondary alkylcopper reagents of type 3 prepared by an I/Li-exchange reaction of the secondary alkyl iodides 1. Thus, the reaction of the *syn*- and *anti*-alkyl iodides (*syn*-1b and *anti*-1b) with *t*-BuLi (2.5 equiv, $-100\text{ }^{\circ}\text{C}$), followed by the addition of CuBr·P(OEt)₃ (2.0 equiv, $-100\text{ }^{\circ}\text{C}$, 5 min), provided the intermediate alkylcopper reagents (*syn*- and *anti*-3b), which underwent a smooth cross-coupling with 1-bromo-2-(trimethylsilyl)ethyne (4a), leading to the expected alkynes *syn*- and *anti*-7a in 48–53% yield with a dr of 89:11 and 9:91, respectively (entries 1 and 2, Table 3). Similarly, the *syn*- and *anti*-alkyl

Table 3. Scope of Diastereoconvergent Quenching of Secondary *syn*- and *anti*-Alkylcoppers (3) with 1-Bromoalkyne (4a) Leading to Alkylated Alkynes of Type 7

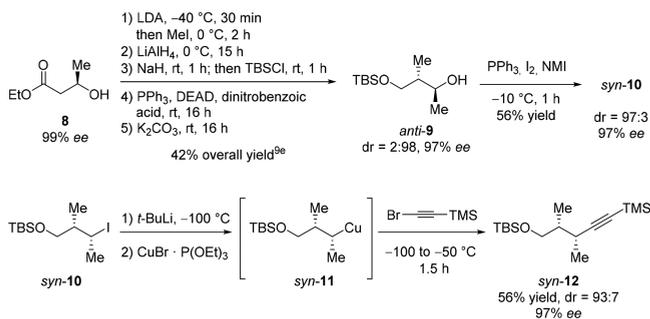
entry	alkyl iodide of type 1	alkylcopper of type 3	product of type 5 ^{a,b}
1	<i>syn</i> -1b, dr = 98:2	<i>syn</i> -3b	<i>syn</i> -7a 53%; dr = 89:11
2	<i>anti</i> -1b, dr = 1:99	<i>anti</i> -3b	<i>anti</i> -7a 48%; dr = 9:91
3	<i>syn</i> -1c, dr = 99:1	<i>syn</i> -3c	<i>syn</i> -7b 58%; dr = 87:13
4	<i>anti</i> -1c, dr = 2:98	<i>anti</i> -3c	<i>anti</i> -7b 51%; dr = 15:85
5	<i>syn</i> -1d, dr = 99:1	<i>syn</i> -3d	<i>syn</i> -7c 67%; dr = 91:9

^aIsolated yield. ^bdr (*syn/anti* ratio) was determined by ¹H and ¹³C NMR analysis.

iodides (*syn*- and *anti*-1c) were converted to the corresponding copper reagents and gave, after cross-coupling with 4a, the diastereomerically enriched alkynes *syn*-7b (58% yield; dr = 87:13; entry 3) and *anti*-7b (51% yield; dr = 15:85; entry 4). The use of a more bulky silyl protecting group such as TBDPS (1d) improved the diastereoselectivity and yield of the cross-coupling reaction with 4a, leading to the *syn*-product *syn*-7c in 67% yield (dr = 91:9, entry 5).

We have also performed such a cross-coupling with an optically enriched alkyl iodide *syn*-10 (see Scheme 2). Starting from the commercially available (+)-*R*-hydroxybutyrate 8 (99%

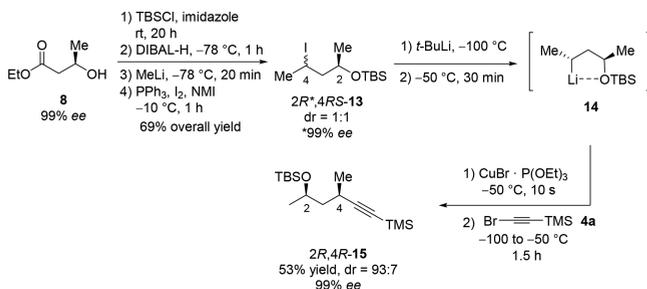
Scheme 2. Preparation of the Chiral Alkyne *syn*-12 Starting from Optically Enriched Alcohol *anti*-9



ee), we have prepared the alcohol *anti*-9 (dr = 2:98) in 5 steps and 42% overall yield.^{9e} This alcohol was converted to the corresponding iodide (*syn*-10) with complete inversion of configuration (dr = 97:3) using an Appel reaction.⁹ An I/Li exchange, followed by a transmetalation with $\text{CuBr}\cdot\text{P}(\text{OEt})_3$, furnished the intermediate copper reagent *syn*-11, which was converted, after a cross-coupling with **4a**, to the *syn*-alkyne *syn*-12 in 56% yield and dr = 93:7 (97% *ee*)¹³.

Furthermore, the 1,3-functionalized secondary alkyl iodide **2R,4RS-13** was prepared starting from **8** in 4 steps^{9c} and 69% overall yield (see Scheme 3). The addition of **2R,4RS-13** to a

Scheme 3. Preparation of the Chiral Alkyne *syn*-15 Starting from Optically Enriched Alkyl Iodide (**13**)



solution of *t*-BuLi (inverse addition, 2.5 equiv, $-100\text{ }^{\circ}\text{C}$) led, after epimerization ($-50\text{ }^{\circ}\text{C}$, 30 min),^{9c} to the chelate-stabilized lithium reagent **14**. Subsequent transmetalation with $\text{CuBr}\cdot\text{P}(\text{OEt})_3$ led to the corresponding alkylcopper reagent, which underwent a smooth cross-coupling reaction with **4a**, leading to the enantiomerically enriched alkyne **2R,4R-15** in 53% yield and dr = 93:7 (99% *ee*)¹³.

In summary, we have shown that α -chiral alkynes are readily obtained with high diastereoselectivity using a stereoselective I/Li exchange at $-100\text{ }^{\circ}\text{C}$, followed by a retentive transmetalation with $\text{CuBr}\cdot\text{P}(\text{OEt})_3$ and further cross-coupling with various bromoalkynes. This method allows preparation of α -chiral alkynes with two stereocontrolled centers in up to 99% *ee*. Extensions of this method are currently underway in our laboratories.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.8b00699.

Full experimental details; ^1H and ^{13}C NMR spectra (PDF)

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Notes

The authors declare no competing financial interest.

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(12) Benzoyl chloride was used as an electrophile, because the *anti/syn* ratio of the ketone **6** could be determined by GC analysis.

(13) The enantiomeric excess (*ee*) was determined by chiral GC analysis. For details, see [Supporting Information](#).