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Tetrahedron: Asymmetry

Synthesis of chiral monodentate binaphthophosphepine ligands and their application in asymmetric hydrogenations

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Abstract—A general synthesis of chiral monodentate 4,5-dihydro-3*H*-dinaphthophosphepines 4 and a detailed study of the catalytic performance of the resulting ligands 4a–n in benchmark hydrogenation reactions is presented. Hydrogenation of methyl α -acetam-idocinnamate 11 and methyl α -acetamidoacrylate 13 proceeded with enantioselectivities up to 95% and 94%, respectively. The best enantioselectivity for the rhodium-catalyzed hydrogenation of dimethyl itaconate 15 was 88%. © 2004 Elsevier Ltd. All rights reserved.

1. Introduction

Transition metal-catalyzed asymmetric reactions offer an efficient and elegant possibility for the synthesis of enantiomerically pure compounds.¹ Among the different transition metal-catalyzed enantioselective reactions, hydrogenations of olefins, imines and ketones have been used extensively over the last two decades and are likely to provide the most important accesses to pharmaceutical intermediates.² In general, optically active bidentate phosphine ligands have been viewed as essential in order to achieve high selectivities and excellent control in a number of catalytic asymmetric hydrogenation reactions.³

However more recently, based on the important discovery by Reetz and co-workers (phosphites 1),⁴ Feringa et al. (phosphoramidites 2),⁵ Pringle et al. (phosphonites 3)⁶ monodentate ligands have become increasingly important for various catalytic asymmetric hydrogenations.⁷ A selection of monodentate ligands that have recently been obtained (2002–2004) is presented in Scheme 1. To date, phosphites and phosphoramidites with a 2,2'-binaphthol core have attracted considerable interest.⁸ Furthermore, biphenyl-based monodentate ligands



Scheme 1. Selection of recent chiral monodentate phosphorus containing ligands.

7 were introduced by Chen and Xiao⁹ and Ojima et al.¹⁰ Recently, Zhou et al.¹¹ also developed a monophosphoramidite ligand named SIPHOS **8** containing a 1,1'-spirobiindane-7,7'-diol backbone.

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Apart from phosphites and phosphoramidites, monodentate phosphine ligands are also of interest due to their often increased stability against hydrolysis and easier purification when compared to the corresponding P– O ligands. An interesting example of a monodentate secondary phosphine 9,¹² which gives excellent enantioselectivities in the Rh-catalyzed hydrogenation of itaconates, has been reported by Helmchen et al.

Similar to the 2,2'-binaphthol core, the corresponding carba analogue (2,2'-bismethylenbinaphthyl) proved to be a structural unit, which induces high stereoselectivity in asymmetric hydrogenations.¹³ Therefore, Gladiali et al.,¹⁴ Chi and Zhang¹⁵ and ourselves¹⁶ introduced independently new monodentate phosphines based on a 4,5-dihydro-3*H*-dinaphtho[2,1-*c*;1',2'-*e*]phosphepine **4**.

In previous communications we briefly described the synthesis of ligands 4^{16} and 5^{17} and their use in different benchmark hydrogenation reactions. Herein, we report a full account of our work on the synthesis and characterization of 14 monodentate ligands of this type and their application in the hydrogenation of dehydroamino acid methyl esters and dimethyl itaconate.

2. Results and discussion

Our general synthesis of the different 4,5-dihydro-3*H*dinaphtho[2,1-*c*;1',2'-*e*]phosphepine ligands **4** started from enantiomerically pure 2,2'-binaphthol, which is available on a large scale and is relatively inexpensive. Esterification with trifluoromethanesulfonic acid anhydride in the presence of pyridine gave the corresponding diester in 99% yield. Subsequent nickel-catalyzed Grignard reaction with methyl magnesium bromide in diethyl ether leads to enantiomerically pure (ee >99%) 2,2'-dimethylbinaphthyl in 95% yield.

As shown in Scheme 2, two synthetic routes were used to prepare 14 different 4,5-dihydro-3H-dinaphtho[2,1c;1',2'-e]phosphepine ligands 4 from homochiral 2,2'dimethylbinaphthyl. The first synthesis involved the double metallation of 2,2'-dimethylbinaphthyl with *n*butyl lithium in the presence of TMEDA (tetramethylethylenediamine) and quenching with commercially available dichlorophosphines gave ligands 4a (P-ethyl), 4c (P-tert butyl) and 4d (P-phenyl) in 60-83% yield. On the other hand double metallation, quenching with diethylaminodichloro-phosphine and subsequent reaction with HCl gave the corresponding chlorophosphepine 10 in 80% yield. It is interesting to note that this chlorophosphepine is a valuable building block for the synthesis of a variety of substituted 4,5-dihydro-3Hdinaphtho[2,1-c;1',2'-e]phosphepines by simple Grignard reaction. Hence, ligands 4b (P-isopropyl), and 4e-n (P-aryl) were prepared without optimization in 62-76% yield. In general, the Grignard reagents were prepared from commercially available alkyl or aryl halides and 1.2 equiv of magnesium turnings. In the case of the 3,5-di-t-butyl-1-bromobenzene and 2-bromonaphthalene, the preparation of the Grignard solution caused some problems because of the sterically demand-



Scheme 2. General route to 4,5-dihydro-3*H*-dinaphtho[2,1-*c*;1', 2'-*e*]phosphepine ligands **4**.

ing substituents. Hence, ligand **4h** $(3,5-(t-Bu)_2-C_6H_3-P)$ gave a somewhat lower yield (50%) and **4l** (P-2-naphthyl) was obtained with a lower purity. Advantageously, monodentate phosphepines **4b**, **4c** and **4e** (4-CH₃O-C₆H₄-P) can be purified by simple crystallization from toluene or toluene/methanol.

All ligands were characterized by NMR, IR, mass spectroscopy and elemental analysis. In addition, the structure of the isopropyl-substituted **4b** and the 4-meth-oxyphenyl-substituted **4e** phosphepine ligands were confirmed by X-ray crystallography (Figs. 1 and 2).¹⁸

As expected in both compounds **4b** and **4e**, the phosphorus atom is arranged in a trigonal pyramidal structure.



Figure 1. ORTEP plot of compound 4b. Hydrogen atoms are omitted for clarity. The thermal ellipsoids correspond to 30% probability.



Figure 2. ORTEP plot of compound **4e**. Only one of the both symmetry-independent molecules of the asymmetric unit is depicted. Hydrogen atoms are omitted for clarity. The thermal ellipsoids correspond to 30% probability.

 Table 1. Selected bond lengths [Å] and angles [°] of 4b and 4e

Ligano	1 4b	Ligand 4	le ^a
P-C1	1.871(3)	P1-C20	1.861(6)
		(P2–C40)	1.867(7)
P-C18	1.867(4)	P1-C7	1.839(6)
		(P2–C27)	1.842(6)
P-C19	1.845(4)	P1-C1	1.829(7)
		(P2–C21)	1.819(6)
C19-P-C18	100.4(2)	C1-P1-C7	102.2(3)
		(C21-P2-C27)	104.3(3)
C19-P-C1	100.6(2)	C1-P1-C20	104.1(3)
		(C21-P2-C40)	101.1(3)
C18-P-C1	98.38(15)	C7-P1-C20	99.2(3)
		(C27–P2–C40)	98.5(3)

^a Values of the second molecule in the asymmetric unit are in brackets.

Selected bond lengths and angles are shown in Table 1. The exocyclic P–C bond (**4b**: P–C19; **4e**: P1–C1) is slightly shorter in both molecules then the P–C bonds in the phosphepine ring.

¹H and ¹³C NMR spectra of ligands **4** show the nonequivalence of the methylene groups and both naphthalene ring systems of the 1,1'-binaphthalene-2,2'bis(methylene) backbone. Two ¹³C doublets for the CH₂ groups are observed were different P–C coupling constants. For the diastereotopic CH₂ protons, in general, four sets of double doublets are displayed in the ¹H NMR spectra due to geminal ¹H,¹H and ³¹P,¹H couplings. The ³¹P spectra of **4a–n** show a signal in the range of $\delta = -3$ to +28.

With a number of chiral monodentate ligands in hand, we were interested in their catalytic behaviour. As model reactions, the asymmetric hydrogenation of methyl (Z)- α -acetamidocinnamate 11 (Table 2), methyl α -acetamidoacrylate 13 (Table 3) and dimethyl itaconate 15 (Table 4) were studied at ambient pressure. All these reactions are generally accepted as typical benchmark tests for chiral ligands.

Initial hydrogenation experiments with methyl α -acetamidocinnamate **11** as substrate were carried out using different solvents. The best results were obtained in toluene, ethyl acetate or a mixture of toluene/SDS (sodium dodecylsulfonate) as the reaction medium. Thus, these solvents were tested with most of the prepared ligands **4a–n**. Selected results of this study are shown in Table 2. As a pre-catalyst, a mixture of 1 mol% [Rh(COD)₂]BF₄ and 2 mol% of the respective ligand was used. In order to compare catalyst activities, the time necessary for achieving 50% conversion (*t*/2) was measured.

Although toluene is an unusual solvent for asymmetric hydrogenation reactions of α -amino acid precursors, when applying P-aryl substituted 4,5-dihydro-3Hdinaphtho[2,1-c;1',2'-e]phosphepines **4d**-n as ligands, enantioselectivities between 70% and 90% ee were observed. Interestingly, the addition of the tenside SDS led in most cases to a slight increase (5-10%) in the enantioselectivity (Table 2, compare entries 4–5, 9–10, 14–15, 18–19).¹⁹ The reason for this effect is still unclear. Notably, the catalyst activity in ethyl acetate is significantly higher when compared to toluene because of the formation of arene complexes in the latter case (Table 2, entry 6 vs 7; 20 vs 21).²⁰ Thus, it is possible to decrease the catalyst concentration by one order of magnitude without a significant change of selectivity. Notably, the substituents on the phenyl group displayed a significant influence on the hydrogenation activity. While electron-withdrawing substituents decrease the activity (vide supra), electron-donating groups increase it. In toluene, high catalyst activities were obtained using the simple alkyl-substituted ligands 4a and 4b, as well as the 4-methoxyphenyl-substituted ligand 4e and the 3,5- $(t-Bu)_2$ -phenyl-substituted ligand 4h (TOF = 1.000- $3.000 \,\mathrm{h^{-1}}$ at 50% conversion).

A comparison of the structures of ligands 4 demonstrated the necessity of a P-aryl group for obtaining good enantioselectivity. Hence, alkyl-substituted phosphepine ligands only gave disappointing enantioselectivities (<50%) in the model reaction. Interestingly, both the substitution of the electron-donating (OMe) as well as the electron-withdrawing substituents (F, CF₃) at the 4-position of the phenyl group resulted in a decrease in enantioselectivity (Table 2, entries 4, 6, 8, 16). Similar substitutions at the 2-position of the phenyl group also led to a slight decrease in the enantioselectivity (Table 2, entries 13, 14, 18). When using bulky substituents at the 3- and 5-position (**4h**), a further increase of the enantioselectivity above 95% ee was not observed (Table 2, entry 11).

Next, the asymmetric hydrogenation of methyl α -acetamidoacrylate 13 was investigated in more detail. Again the influence of different solvents and ligands studied. Only the best results for the respective ligands are shown in Table 3. In general, the ee's were significantly lower when compared to the hydrogenation of methyl α -acetamidocinnamate 11. For example the alkyl-substituted derivatives **4a–c** gave 5–20% ee (Table 3, entries 1–3) while the phenyl-substituted ligand **4d** led only to 67% ee (Table 3, entry 4). The influence of different substituents on the phenyl group can not be rationalized easily. The best enantioselectivity (94% ee) was obtained in the

Table 2. Asymmetric hydrogenation of methyl α -acetamidocinnamate 11

		$L = \bigcup_{i=1}^{p \cdot R} P \cdot R$	$\begin{array}{c} 0 \\ Ph \\ \hline \\ NHAc \\ 11 \end{array} \qquad \begin{array}{c} H_2 \\ Rh/2 \\ \hline \\ Rh/2 \end{array}$	Ph *	OMe	
Entry	Ligand	R =	Solvent	t/2 [min]	Conversion %	Ee % (<i>R</i>)
1 ^a	4 a	Et	Toluene	6	>99	47
2 ^a	4b	<i>i</i> -Pr	Toluene	4	>99	50
3 ^a	4c	t-Bu	Toluene	31	>99	20
4 ^a	4d	Ph	Toluene	50	>99	90
5 ^b	4d	Ph	Toluene + SDS	33	>99	95
6 ^a	4 e	$4-CH_3O-C_6H_4$	Toluene	2	>99	74
7^{a}	4 e	$4-CH_3O-C_6H_4$	Ethyl acetate	4	>99	88
8 ^a	4f	$4-CF_3-C_6H_4$	Toluene	36	>99	82
9 ^a	4g	3,5-(CH ₃) ₂ -C ₆ H ₃	Toluene	17	>99	67
10 ^b	4g	3,5-(CH ₃) ₂ -C ₆ H ₃	Toluene + SDS	15	>99	81
11 ^a	4h	$3,5-(t-Bu)_2-C_6H_3$	Toluene	2	>99	95
12 ^b	4h	$3,5-(t-Bu)_2-C_6H_3$	Toluene + SDS	2	>99	95
13 ^a	4i	$2-CH_3O-C_6H_4$	Toluene	59	>99	64
14 ^a	4j	$2-F-C_6H_4$	Toluene	13	>99	87
15 ^b	4j	$2-F-C_6H_4$	Toluene + SDS	53	>99	91
16 ^a	4k	$4-F-C_6H_4$	Toluene	160	>99	73
17 ^b	4k	$4-F-C_6H_4$	Toluene + SDS	126	>99	86
18 ^a	41	2-Naphthyl	Toluene	22	>99	88
19 ^b	41	2-Naphthyl	Toluene + SDS	50	>99	94
20 ^a	4m	$3-CH_3O-C_6H_4$	Toluene	36	>99	83
21 ^a	4 m	$3-CH_3O-C_6H_4$	Ethyl acetate	5	>99	88
22 ^a	4n	3,4-(CH ₃ O) ₂ C ₆ H ₃	Toluene	—	>99	93

^a Reaction conditions: 1.0mmol substrate; 0.01mmol [Rh(COD)₂]BF₄; cat.:ligand = 1:2; 15mL solvent; 25°C, 1 bar H₂.

^b Reaction conditions like ^a + 0.2 mmol SDS.

Table 3.	Asymmetric	hydrogenation	of methyl α -ad	cetamidoacrvlate 13
		J		

		$L = \bigcup_{P \in \mathbb{R}} P \cdot R$	O CH ₂ NHAc 13 OMe H ₂ Rh / 21	CH ₃ NHAc 14	ОМе	
Entry	Ligand	R =	Solvent	t/2 [min]	Conversion %	Ee % (<i>R</i>)
1 ^a	4a	Et	Toluene	6	>99	5
2^{a}	4b	<i>i</i> -Pr	Toluene	4	>99	20
3 ^a	4c	t-Bu	Toluene		99	15 (S)
4 ^a	4d	Ph	Toluene	18	>99	67
5 ^a	4 e	$4-CH_3O-C_6H_4$	Ethyl acetate	2	>99	51
6 ^a	4f	$4-CF_3-C_6H_4$	Ethyl acetate	2	>99	57
$7^{\rm a}$	4g	$3,5-(CH_3)_2-C_6H_3$	Ethyl acetate	2	>99	43
8^{a}	4h	$3,5-(t-Bu)_2-C_6H_3$	Toluene	1	>99	94
9 ^b	4i	$2-CH_3O-C_6H_4$	Toluene	58	>99	55
$10^{\rm a}$	4j	$2-F-C_6H_4$	Toluene	7	>99	76
11 ^b	4j	$2-F-C_6H_4$	Toluene + SDS	20	>99	84
12 ^a	4k	$4-F-C_6H_4$	Toluene	95	>99	49
13 ^b	4k	$4-F-C_6H_4$	Toluene + SDS	106	>99	73
14 ^a	41	2-Naphthyl	Toluene	34	97	64
15 ^b	41	2-Naphthyl	Toluene + SDS	33	97	75
16 ^a	4m	$3-CH_3O-C_6H_4$	Toluene	16	>99	75
17 ^b	4m	3-CH ₃ O-C ₆ H ₄	Toluene + SDS	17	>99	86
18 ^a	4n	3,4-(CH ₃ O) ₂ -C ₆ H ₃	Ethyl acetate		98	56

^a Reaction conditions: 1.0mmol substrate; 0.01mmol [Rh(COD)₂]BF₄; cat.:ligand = 1:2; 15mL solvent; 25°C, 1 bar H₂.

^b Reaction conditions like ^a + 0.2 mmol SDS.

presence of **4h** with bulky *tert*-butyl groups at the 3- and 5-position of the phenyl ring (Table 3, entry 8). Comparably high selectivities (84–86% ee) were also obtained

with the 2-fluorophenyl- and 3-methoxyphenyl-substituted ligands **4j** and **4m** (Table 3, entries 11 and 17). With regard to activity, similar trends were observed

Table 4. Asymmetric hydrogenation of dimethyl itaconate 15

		$L = \bigcup_{P \in \mathbb{R}} P \cdot R$	CH ₂ COOMe H ₂ Rh / 2 Rh / 2 15	L CH ₃ COOMe 16	COOMe	
Entry	Ligand	R =	Solvent	t/2 [min]	Conversion %	Ee % (S)
1 ^a	4 a	Et	Toluene	180	>99	54
2^{a}	4b	<i>i</i> -Pr	Dichloromethane	_	>99	65
3 ^a	4c	t-Bu	Dichloromethane	_	98	62
4 ^a	4 d	Ph	Dichloromethane	56	>99	86
5 ^a	4 e	$4-CH_3O-C_6H_4$	Ethyl acetate	6	>99	79
6	4 e	$4-CH_3O-C_6H_4$	Dichloromethane	_	>99	86
7^{a}	4 f	$4-CF_3-C_6H_4$	Ethyl acetate	14	>99	63
8 ^a	4g	3,5-(CH ₃) ₂ -C ₆ H ₃	Ethyl acetate	26	>99	67
9 ^a	4h	$3,5-(t-Bu)_2-C_6H_3$	Dichloromethane	16	>99	83
10^{a}	4i	$2-CH_3O-C_6H_4$	Dichloromethane	_	7	22
11 ^a	4j	$2-F-C_6H_4$	Dichloromethane	16	86	77
12 ^a	4k	$4-F-C_6H_4$	Ethyl acetate	13	>99	65
13 ^a	41	2-Naphthyl	Dichloromethane	34	29	70
14 ^a	4 m	$3-CH_3O-C_6H_4$	Dichloromethane	_	>99	88
15 ^a	4n	3,4-(CH ₃ O) ₂ -C ₆ H ₃	Dichloromethane		32	66

^a Reaction conditions: 0.5mmol substrate; 0.005mmol [Rh(COD)₂]BF₄; cat.:ligand=1:2; 7.5mL solvent; 25°C, 1 bar H₂.

when compared to the hydrogenation of methyl α -acetamidocinnamate **11**. Hence, good catalyst activity (TOF = 1.000-6.000 h⁻¹ at 50% conversion) was seen with ligands **4a**, **4b**, **4e**–**h** and **4j**. Pleasingly, **4h**, which showed the best enantioselectivity induces also the highest activity.

Initially, the asymmetric hydrogenation of dimethyl itaconate **15** proved to be the most difficult test reaction (Table 4). Here, in toluene and toluene/SDS, a significantly lower catalyst activity was observed (compare entry 1 in Tables 2–4). In addition, in most cases lower enantioselectivities were observed. Surprisingly, the highest observed enantioselectivity was when using toluene as the solvent for ligand **4a** (Table 4, entry 1).

However, by changing the solvent to dichloromethane or ethyl acetate moderate to good enantioselectivities were obtained. For example, the phenyl- and *p*-methoxy-phenyl-substituted derivatives **4d** and **4e** gave 86%ee.

The best selectivity (88% ee) seen was by applying **4m** as ligand.

3. Conclusion

In conclusion, the synthesis and characterization of 14 different enantiomerically pure monodentate dinaphtho-phosphepine ligands 4 has been described in detail. Several members of this class are presented here for the first time. In general, this class of chiral ligands is easily available in 3-5 steps from commercially available 2,2'-binaphthol on 10g scale. By using the corresponding chlorodinaphtho-phosphepine 10, a toolbox of chiral ligands can be synthesized in one step. Arylsubstituted dinaphthophosphepine ligands **4d–n** led to good enantioselectivities in various rhodium-catalyzed hydrogenations. Hence, reactions of methyl α -acetamidocinnamate **11** proceeded up to 95% ee and for methyl α -acetamidoacrylate **13** up to 94% ee, while the best enantioselectivity for the hydrogenation of dimethyl itaconate **15** obtained was 88%.

4. Experimental

¹H, ¹³C and ³¹P NMR spectra were recorded on a Bruker spectrometer ARX 400 (¹H: 400.13 MHz, ¹³C: 100.6 MHz, ³¹P: 162.0 MHz). The calibration of ¹H and ¹³C spectra was carried out on solvent signals [δ (CDCl₃) = 7.27 and 77.0; δ (C₆D₆) = 7.15 and 128.0; δ (CD₃COCD₃) = 2.05 and 29.8]. The ³¹P chemical shifts are referenced to 85% H₃PO₄. The ¹H and ¹³C NMR signals were assigned by DEPT and two-dimensional ¹H, ¹H COSY and ¹H, ¹³C correlation spectra. The differentiation between the signals of both nonequivalent naphthalenes, if possible, and the assignment of these rings to the CH₂ and CH₂ groups, respectively, were based on long-range correlation (HMBC) and ¹H, ¹H NOESY spectra recorded for compounds 4c and 4d. Mass spectra were recorded on an AMD 402 spectrometer. Elemental analyses were performed in a Leco CHNS-932. Optical rotations were measured on a Gyromat-HP polarimeter. IR spectra were recorded as KBr pellets or Nujol mulls on a Nicolet Magna 550. All experiments were carried out under an argon atmosphere. Enantiomerically pure 2,2'-dimethylbinaphthyl was obtained in 92% overall yield from enantiomerically pure 2,2'-binaphthol via esterification with trifluoromethanesulfonic acid anhydride in pyridine and subsequent Ni-catalyzed Grignard reaction with MeMgBr.

4.1. (S)-2,2'-Dimethylbinaphthyl



Yield: 95%. ¹H NMR (400 MHz, acetone- d_6): $\delta = 2.01$ (s, 6H, CH₃); 6.97 (d"q", $J_{7,8} = 8.5$ Hz, $3 \times J = 1.2$ Hz, 2H, H-8); 7.22 (ddd, $J_{7,8} = 8.5 \text{ Hz}$, $J_{6,7} = 6.6 \text{ Hz}$, $J_{5,7} = 1.2 \text{ Hz}$, 2H, H-7); 7.41 (ddd, $J_{5,6} = 8.3 \text{ Hz}$, $J_{6,7} = 6.6 \,\text{Hz}, \quad J_{6,8} = 1.2 \,\text{Hz}, \quad 2\text{H}, \quad \text{H-6}); \quad 7.57 \quad (d,$ $J_{3,4} = 8.5$ Hz, 2H, H-3); 7.95 (br d, 2H, H-4,5). ¹³C NMR (100.6 MHz, acetone- d_6): $\delta = 20.1$ (CH₃); 125.9 (C-6); 126.0 (C-8); 127.0 (C-7); 128.5, 129.1 (C-4,5); 129.5 (C-3), 133.4, 133.6 (C-9,10); 134.9 (C-1); 135.8 (C-2). IR (Nujol, KBr): 3045w, 2953s, 2857s, 1617w, 1594w, 1506w, 1463s, 1421w, 1378m, 1352w, 1261w, 1221w, 1158w, 1143w, 1028w, 958w, 898w, 863w, 810s, 796m, 783m, 775m, 742s, 720m, 662w, 621m, 574w, 483w, 416m. MS (ESI): m/z (%) = 282 ([M⁺], 100); 267 $([M^+-Me], 50); 252 ([M^+-2Me], 27); 239 (12); 133$ (25); 126 (15); 69 (8); 58 (7); 28 (13). Anal. Calcd for $C_{22}H_{18}$ (%): C, 93.57; H, 6.43. Found: C, 93.22; H, 6.58. $[\alpha]_D^{23} = +14.5$ (*c* 1, toluene).

4.2. General procedure for the synthesis of 4

Procedure A: Metallation¹⁵ of enantiomerically pure 2,2'-dimethyl-binaphthyl with *n*-BuLi/TMEDA afforded the crystalline dilithio species in 70–80% yield. Starting from 12 mmol dilithium salt of homochiral 2,2'-dimethylbinaphthyl in 35 mL hexane, 13.6 mmol phenyl dichlorophosphine in 15 mL hexane was added at 0 °C. After 2h refluxing, the reaction mixture was quenched with water/toluene. The organic layer was separated and dried over MgSO₄. Ligand **4** was purified by column chromatography in toluene or crystallized from toluene/ methanol.

Procedure B: Starting from 30 mmol of the dilithium salt of 2,2'-dimethyl-1,1'-binaphthyl in 100 mL hexane, 34 mmol Cl₂PNEt₂ in 30 mL hexane was added at 0 °C (30–45 min). After 2h refluxing, the hexane was removed and 100 mL toluene added. LiCl was filtrated under inert conditions and washed with 50 mL toluene. The toluene was then removed from the filtrate. The yellow solid compound was suspended in 200 mL cyclohexane. HCl was bubbled through the stirred suspension for 1 h at 0 °C and then argon for an additional hour through the reaction mixture. The precipitate was removed by filtration to give a clear yellow solution. Rotation of the filtrate gave the corresponding chlorophosphine **10** (light yellow foam) in 80% yield.

Compound **10** (6.2mmol) was dissolved in 20mL THF. The corresponding Grignard solution (1.1equiv) prepared from Mg and aryl or alkylhalides in THF was added during 30min. After 2h refluxing, the reaction mixture was quenched with water/toluene. The organic layer was separated and dried over MgSO₄. Ligand **4** was purified by column chromatography in toluene or crystallized from toluene/methanol.

Compounds 4a, 4c and 4d were synthesized by procedure A, ligands 4b and 4e-n by procedure B.

4.2.1. 4-Ethyl-4,5-dihydro-3H-dinaphtho[2,1-c;1',2'-e]-phosphepine 4a.



Yield: 74%. ¹H NMR (400 MHz, C_6D_6): $\delta = 0.93$ (dt, 3H, $J_{P,H} = 15.2 \text{ Hz}, J_{H,H} = 7.6 \text{ Hz}, \text{ CH}_3$; 1.08–1.15 (m, 2H, CH_2CH_3); 1.98 (dd, 1H, $J_{H,H} = 11.7$ Hz, $J_{P,H} = 4.0$ Hz, CH₂(a)); 2.18 (dd, 1H, $J_{H,H} = 14.3$ Hz, $J_{P,H} = 6.4$ Hz, CH₂(a)); 2.54 (dd, 1H, $J_{P,H} = 8.2$ Hz, CH₂(b)); 2.55 (dd, 1H, $J_{P,H} = 17.2$ Hz, CH₂(b)); 6.95–7.01 (m, 2H, H-7,7'); 7.01 (d, 1H, H-8); 7.16-7.22 (m, 2H, H-6,6'); 7.39-7.46 (2d, 1dd, 3H, H-3,3',8'); 7.69–7.76 (4d, 4H, H-4,4',5,5'). ¹³C NMR (100.6 MHz, C₆D₆): $\delta = 10.1$ (d, $J = 16.2 \text{ Hz}, \text{ CH}_3$; 19.0 (d, $J = 17.2 \text{ Hz}, \text{ CH}_2\text{CH}_3$); 29.1 (d, J = 22.0 Hz, CH₂); 32.5 (d, J = 18.0 Hz, CH₂); 125.17, 125.2 (C-6,6'); 126.3, 126.4 (C-7,7'); 127.07, 127.15 (C-8,8'); 127.4 (d, C-3); 127.6–128.6 (C-3',4,4',5,5'); 132.8–133.9 (C-1,1',2,2',9,9',10,10'). ³¹P NMR (162 MHz, C₆D₆): δ = 8.5. IR (Nujol): 2953s, 2843s, 1618w, 1591w, 1507m, 1457s, 1422w, 1377m, 1247w, 1224w, 1160w, 1144w, 1063w, 1025m, 993w, 958w, 935m, 864m, 828s, 821s, 804s, 775m, 747s, 676m, 658w, 634w, 620w, 566m, 520m. MS (ESI): m/z (%) = 340 ([M⁺], 100); 311 ([M⁺-Et], 47); 276 (23); 265 (63); 252 (7); 183 (12); 170 (14); 138 (21); 91 (2); 75 (1); 63 (2); 41 (2). HRMS calcd for C₂₄H₂₁P: 340.13809. Found: 340.13792. [α]²³_D = +301 (*c* 1, toluene).

4.2.2. 4-*i*-Propyl-4,5-dihydro-3*H*-dinaphtho[2,1-*c*;1',2'-*e*]-phosphepine 4b.



Yield: 81%. ¹H NMR (400 MHz, C₆D₆): $\delta = 0.79$ (dd, 3H, $J_{P,H} = 13.9$ Hz, $J_{H,H} = 7.0$ Hz, CH₃); 1.06 (dd, 3H, $J_{P,H} = 14.8$ Hz, $J_{H,H} = 7.0$ Hz, CH₃); 1.29 (m, 1H, CH); 2.12 (dd, 1H, $J_{H,H} = 11.5$ Hz, $J_{P,H} = 4.0$ Hz, CH₂(a)); 2.43 (dd, 1H, $J_{H,H} = 14.5$ Hz, $J_{P,H} = 6.8$ Hz, CH₂(a)); 2.53 (dd, 1H, $J_{P,H} = 8.1$ Hz, CH₂(b)); 2.63 (dd, 1H, $J_{P,H} = 17.3$ Hz, CH₂(b)); 6.95–7.01 (m, 2H, H-7,7'); 7.10 (d, 1H, H-8); 7.17–7.22 (m, 2H, H-6,6'); 7.39–7.48 (2d, 1dd, 3H, H-3,3',8'); 7.68–7.76 (4d, 4H, H-4,4',5,5'); ¹³C NMR (100.6 MHz, C₆D₆): $\delta = 18.9$ (d, J = 17.2 Hz, CH₃); 20.1 (d, J = 18.1 Hz, CH₃); 24.8 (d, J = 15.3 Hz, CH); 28.2 (d, J = 24.8 Hz, CH₂); 31.8 (d, J = 20.0 Hz, CH₂); 125.2 (C-6,6'); 126.3, 126.4 (C-7,7'), 127.1, 127.2 (C-8,8'); 127.6 (C-3), 127.7–128.6 (C-3',4,4',5,5'); 132.8–134.4 (C-1,1',2,2',9,9',10,10'). ³¹P NMR (162 MHz, C₆D₆): $\delta = 21.8$. IR (Nujol): 2945s, 2851s, 1616w, 1589w, 1507w, 1463s, 1419w, 1378s, 1329w, 1248w, 1224w, 1187w, 1144w, 1025w, 948w, 938w, 864m, 831m, 824m, 750m, 673w, 658w, 628w, 566w, 520w. MS (ESI): m/z (%) = 354 ([M⁺], 86); 311 ([M⁺-Pr], 100); 283 (35); 265 (63); 183 (11); 138 (14); 91 (5); 57 (2); 43 (13). Anal. Calcd for C₂₅H₂₃P (%): C, 84.72; H, 6.54; P, 8.74. Found: C, 84.43; H, 6.49; P, 8.69. [α]²_D = +218 (c 0.5, toluene).

4.2.3. 4-*t*-Butyl-4,5-dihydro-3*H*-dinaphtho[2,1-*c*;1',2'-*e*]phosphepine 4c.



Yield: 60%. ¹H NMR (400 MHz, CDCl₃): $\delta = 0.94$ (d, 9H, J = 11.9Hz, CH₃); 2.54–2.66 (m, 4H, CH₂); 6.96 (d, 1H, H-8), 7.01-7.13 (m, 3H, H-7,7',8'); 7.23-7.30 (m, 2H, H-6,6'); 7.33 (d, 1H); 7.50 (dd, 1H), (H-3,3'); 7.72–7.78 (m, 4H, H-4,4',5,5'). ¹³C NMR (100.6MHz, CDCl₃): $\delta = 27.2$ (d, J = 23.8 Hz, CH₂); 27.3 (d, $J = 23.8 \text{ Hz}, \text{ CH}_2$; 28.3 (d, $J = 13.3 \text{ Hz}, \text{ CH}_3$); 29.3 (d, $J = 20.0 \text{ Hz}, C(CH_3)_3$; 124.7, 124.8 (C-6,6'); 125.7, 125.8 (C-7,7'); 126.6, 126.8 (C-8,8'); 127.4 (d, C-3); 127.6 (C-4'); 128.0, 128.2, 128.3 (C-4,5,5'); 129.0 (C-3'); 132.2 (d, C-1'); 132.05, 132.15, 132.5, 133.1 (C-9,9',10,10'); 132.8 (C-1); 134.5 (C-2'); 134.9 (C-2). ³¹P NMR (162 MHz, CDCl₃): δ = 29.8. IR (Nujol): 2935s, 2860s, 1615w, 1591w, 1507w, 1458s, 1420w, 1377m, 1359m, 1331w, 1248m, 1225w, 1206w, 1173m, 1143w, 1063w, 1026w, 963w, 948w, 933m, 868m, 832s, 812w, 791w, 773w, 750s, 739w, 674m, 621m, 567w, 521m. MS (ESI): m/z (%) = 368 ([M⁺], 57); 311 ([M⁺-Bu], 100); 283 (20); 265 (61); 183 (4); 138 (5); 91 (1); 57 (4); 41 (4). Anal. Calcd for C₂₆H₂₅P (%): C, 84.75; H, 6.84; P, 8.41. Found: C, 84.21; H, 6.70; P, 8.34. $[\alpha]_D^{24} = +304 \ (c \ 0.5, \ toluene).$

4.2.4. 4-Phenyl-4,5-dihydro-3H-dinaphtho[2,1-c;1',2'-e]-phosphepine 4d.



Yield: 83%. ¹H NMR (400 MHz, CDCl₃): δ = 2.66 (dd, 1H, $J_{H,H}$ = 14.3 Hz, $J_{P,H}$ = 4.8 Hz, CH₂(a)); 2.77 (dd, 1H, $J_{P,H}$ = 11.3 Hz, CH₂(b)); 2.82 (dd, 1H, $J_{H,H}$ = 11.5 Hz, $J_{P,H}$ = 3.0 Hz, CH₂(a)); 2.91 (dd, 1H, $J_{P,H}$ = 14.3 Hz, CH₂(b)); 6.81 (d, 1H, H-3'); 7.11–

7.22 (m, 9H, H-7,7',8,8', Ph); 7.30-7.36 (m, 2H, H-6,6'); 7.57 (dd, 1H), 7.61 (d, 1H), (H-3,4'); 7.79 (d, 1H, H-5'); 7.83–7.86 (2d, 2H, H-4,5'). ¹³C NMR (100.6 MHz, CDCl₃): $\delta = 30.3$ (d, J = 16.2 Hz, CH₂); 32.2 (d, *J* = 22.0 Hz, CH₂); 124.8, 125.0 (C-6,6'); 125.8, 125.9 (C-7,7'); 126.6, 126.7 (C-8,8'); 127.4 (d, C-3); 127.5 (C-4'); 128.0 (C-3'); 128.1-128.7 (C-4,5,5', m-Ph, p-Ph); 131.6 (d, o-Ph); 132.2 (d, C-1'); 131.9, 132.3, 132.7, 132.8 (C-9,9',10,10'); 133.5 (d, C-2'); 133.7 (d, C-1); 134.3 (C-2); 137.5 (d, *i*-Ph). ${}^{31}P$ NMR (162 MHz, CDCl₃): δ = 7.8. IR (Nujol): 2948s, 2870s, 1616w, 1590w, 1507w, 1463s, 1377s, 1248w, 1159w, 1101w, 1063w, 1025m, 934m, 864m, 827s, 749s, 695m, 669w, 657w, 620w, 566w, 520w. MS (ESI): m/z (%) = 388 ([M⁺], 100); 282 ([M⁺-P-C₆H₅], 20); 265 (20); 194 (5); 155 (10); 91 (3); 39 (1). Anal. Calcd for $C_{28}H_{21}P$ (%): C, 86.58; H, 5.45; P, 7.79. Found: C, 85.99; H, 5.47; P, 8.03. $[\alpha]_D^{23} = -27$ (c 0.5, toluene).

4.2.5. 4-(4-Methoxy)-phenyl-4,5-dihydro-3*H*-dinaphtho-[2,1-*c*;1',2'-*e*]phosphepine 4e.



Yield: 76%. ¹H NMR (400 MHz, CDCl₃): δ = 2.68 (dd, 1H, $J_{H,H} = 14.5 \text{ Hz}$, $J_{P,H} = 4.2 \text{ Hz}$, $CH_2(a)$; 2.87 (dd, 1H, $J_{P,H} = 11.0$ Hz, $CH_2(b)$; 2.92–3.00 (dq, 2H, $J_{H,H} = 12.0$ Hz, $J_{P,H} = 14.8$ Hz, $J_{P,H} = 2.2$ Hz, CH'_2); 3.82 (s, 3H, CH₃O); 6.84 (m, 2H, m-PC₆H₄); 6.95 (d, 1H, H-3'); 7.17 (m, 2H, o-PC₆H₄); 7.23-7.28 (m, 3H), 7.31 (d, 1H), (H-7,7',8,8'); 7.42–7.46 (m, 2H, H-6,6'); 7.66 (dd, 1H, H-3'); 7.77 (d, 1H, H-4'); 7.91–7.96 (3d, 3H, H-4,5,5'). ¹³C NMR (100.6 MHz, CDCl₃): δ = 30.7 (d, J = 13.5 Hz, CH₂); 32.6 (d, J = 20.0 Hz, CH₂); 55.2 (CH₃O); 113.9 (d, *m*-PC₆H₄); 124.9, 125.0 (C-6,6'); 125.8, 125.9 (C-7,7'); 126.7, 126.7 (C-8,8'); 127.3 (d, C-3); 127.4 (C-4'); 128.17, 128.25, 128.35, 128.41 (C-3',4,5,5'); 132.0–132.9 (C-1',9,9',10,10'); 133.4 (C-2'); 133.6 (d, C-1); 133.7 (d, o-PC₆H₄); 134.3 (C-2); 160.6 $(p-PC_6H_4)$. ³¹P NMR (162 MHz, CD₂Cl₂): $\delta = 5.7$. IR (Nujol): 2945s, 2865s, 1591m, 1567w, 1497m, 1464s, 1377s, 1281m, 1246m, 1176m, 1092w, 1023m, 960w, 933w, 865w, 822m, 808m, 755m, 729m, 695w, 671w, 621w, 600w, 567w, 526w. MS (ESI): m/z (%) = 418 $([M^+], 100); 335 ([M^+-MeOPh], 45); 265 (30); 235$ (15); 169 (100); 147 (63); 119 (30); 69 (70); 44 (17). Anal. Calcd for C₂₉H₂₃PO (%): C, 83.20; H, 5.51; P, 7.41. Found: C, 82.78; H, 5.58; P, 7.59. $[\alpha]_D^{23} = -35$ (c 0.01, toluene).

4.2.6. 4-(4-Trifluoromethyl)-phenyl-4,5-dihydro-3*H*-dina-phtho[2,1-*c*;1',2'-*e*]phosphepine 4f.

Yield: 70%. ¹H NMR (400 MHz, CD₂Cl₂): δ = 2.83 (dd, 1H, $J_{H,H}$ = 17.2 Hz, $J_{P,H}$ = 3.5 Hz, CH₂(a)); 2.86 (d, 2H,



 $J_{P,H} = 8.3 \text{ Hz}, \text{ CH}_2$; 3.14 (dd, 1H, $J_{P,H} = 11.5 \text{ Hz}, \text{CH}_2$ (b)); 6.92 (d, 1H, H-3'); 7.15–7.25 (m, 6H, H-7,7',8,8', o-PC₆H₄); 7.36–7.45 (m, 2H, H-6,6'); 7.55 (d, 2H, m-PC₆H₄); 7.67–7.72 (2d, 2H, H-3,4'); 7.85–8.00 (3d, 3H, H-4,5,5'). ¹³C NMR (100.6 MHz, CD₂Cl₂): $\delta = 30.5$ (d, J = 18.1 Hz, CH₂); 32.0 (d, J = 24.0 Hz, CH₂); 116.0 (q, CF₃); 125.3, 125.6 (C-6,6'); 126.2, 126.3 (C-7,7'); 126.8, 126.9 (C-8,8'); 127.5 (C-3); 127.8 (C-4'); 128.1, 128.5, 128.7, 128.8 (C-3',4,5,5'); 130.2 (q, *p*-PC₆H₄); 131.7 (d, *o*-PC₆H₄); 132.2–133.6 (C-(1,1',2',9,9',10,10'); 134.2 (C-2); 143.9 (d,*i*-PC₆H₄). ³¹PNMR (162 MHz, CD₂Cl₂): δ = 8.2. IR (Nujol): 2958s, 2830s, 1607m, 1592w, 1507m, 1463s, 1395m, 1377m, 1324s, 1249w, 1224w, 1168s, 1131s, 1060s, 1015m, 958m, 935m, 865m, 822s, 803m, 750s, 696m, 670w, 658w, 620w, 597w, 566w, 521w, 503w. MS (ESI): m/z $(\%) = 456 ([M^+], 100); 437 ([M^+-F], 2); 276 (16); 265$ (27); 228 (6); 183 (3); 155 (7); 132 (5); 107 (3); 91 (2); 43 (1). HRMS calcd for $C_{29}H_{20}PF_3$: 456.12549. Found: 456.12298. $[\alpha]_D^{23} = -34.3$ (*c* 0.01, toluene).

4.2.7. 4-(3,5-Dimethyl)-phenyl-4,5-dihydro-3*H*-dinaphtho[2,1-*c*;1',2'-*e*]phosphepine 4g.



Yield: 83%. ¹H NMR 400 MHz, CD₂Cl₂): $\delta = 2.15$ (s, 6H, CH₃); 2.81 (d, 2H, $J_{P,H} = 8.2 \text{ Hz}$, CH₂); 2.86 (dd, 1H, $J_{H,H} = 11.5 \text{ Hz}$, $J_{P,H} = 3.0 \text{ Hz}$, $CH'_{2}(a)$; 3.05 (dd, 1H, $J_{P,H} = 16.8 \text{ Hz}$, $CH'_{2}(b)$; 6.90 (d, 2H, $o-PC_{6}H_{3}$); 6.97 (d, 1H, p-PC₆H₃); 7.00 (d, 1H, H-3'); 7.18–7.28 (m, 4H, H-7,7',8,8'); 7.39-7.47 (m, 2H, H-6,6'); 7.69 (dd, 1H); 7.76 (d, 1H), (H-3,4'); 7.90-8.00 (3d, 3H, H-4,5,5'). ¹³C NMR (100.6 MHz, CD₂Cl₂): $\delta = 21.3$ (d, J = 3.0 Hz; 30.8 (d, J = 17.2 Hz, CH₂); 32.1 (d, $J = 23.0 \text{ Hz}, \text{ CH}'_2$; 125.1, 125.3 (C-6,6'); 126.1, 126.2 (C-7,7'); 126.8, 126.9 (C-8,8'); 127.6 (d, C-3); 127.7 (C-4'); 128.4–128.6 (C-3,4,5,5'); 129.3 (d, o-PC₆H₃); 130.5 (p-PC₆H₃); 132.3–133.2 (C-1',9,9',10,10'); 134.0 (d, C-1); 134.3, 134.7 (C-2,2'); 137.9 (d, *i*-PC₆H₃); 138.0 $(m-PC_6H_3)$. ³¹P NMR (162 MHz, CD₂Cl₂): δ = 7.8. IR (Nujol): 2948s, 2870s, 1616w, 1590w, 1507w, 1463s, 1377s, 1248w, 1159w, 1101w, 1063w, 1025m, 934m, 864m, 827s, 749s, 695m, 669w, 657w, 620w, 566w, 520w. MS (ESI): m/z (%) = 388 ([M⁺], 100); 282 ($[M^+-P-C_6H_5]$, 20); 265 (20); 194 (5); 155 (10); 91 (3); 39 (1). HRMS calcd for $C_{30}H_{25}P$: 416.16940. Found: 416.16273. Optical rotation: $[\alpha]_{\rm D}^{26} = +12$ (c 0.5, toluene).

4.2.8. 4-(3,5-Di-*t*-butyl)-phenyl-4,5-dihydro-3*H*-dinaph-tho[2,1-*c*;1',2'-*e*]phosphepine 4h.



Yield: 61%. ¹H NMR (400 MHz, CDCl₃): δ = 1.26 (m, 18H, CH₃); 2.73 (dd, 1H, $J_{H,H}$ = 14.3 Hz, $J_{P,H}$ = 4.5 Hz, CH₂(a)); 2.88 (dd, $J_{P,H} = 11.3$ Hz, CH₂(b)); 2.91–3.00 (m, 2H, CH₂'); 7.02 (d, 1H, H-3'); 7.11 (2d, 2H, H-8,8'); 7.21-7.31 (m, 4H, H-7,7', o-PC₆H₃); 7.38-7.44 (m, 3H, H-6,6', p-PC₆H₃); 7.66 (dd, 1H), 7.76 (d, 1H), (H-3,4'); 7.88–7.95 (3d, 3H, H-4,5,5'). ¹³C NMR (100.6 MHz, CDCl₃): $\delta = 31.0$ (d, J = 21.0 Hz, CH₂); 31.4 (CH₃); 32.1 (d, J = 21.0 Hz, CH₂); 35.0 (CMe₃); 121.4 (p-PC₆H₃); 124.8, 125.0 (C-6,6'); 125.8, 125.9 (C-7,7'); 126.4 (d, o-PC₆H₃); 126.6, 126.7 (C-8,8'); 127.4 (d, C-3); 127.4 (d, C-4'); 128.1, 128.2, 128.4, 128.5 (C-3',4,5,5'); 132.0, 132.3, 132.8, 132.9 (C-9,9',10,10'); 132.1 (d, C-1'); 133.6 (d, C-1); 133.7 (C-2'); 134.6 (C-2); 135.5 (*i*-PC₆H₄); 150.9 (*m*-PC₆H₃). ³¹P NMR (162 MHz, CDCl₃): δ = 8.0. IR (Nujol): 2968s, 2874s, 1595w, 1563w, 1507w, 1462s, 1377m, 1364m, 1248w, 1217w, 1146w, 1027w, 932w, 864w, 810m, 741m, 705w, 660w, 622w, 566w, 543w. MS (ESI): m/z (%) = 500 ([M⁺], 100); 444 ([M⁺-Bu], 6); 387 ([M⁺-2Bu], 1); 311 (2); 265 (10); 183 (1); 141 (1); 91 (1); 57 (6). HRMS calcd for $C_{36}H_{37}P$: 500.65274. Found: 500.65029. $[\alpha]_D^{26} = +48$ $(c \ 0.5, toluene).$

4.2.9. 4-(2-Methoxy)-phenyl-4,5-dihydro-3*H*-dinaphtho[2, 1-*c*;1',2'-*e*]phosphepine 4i.



Yield: 62%. ¹H NMR (400 MHz, CDCl₃): δ = 2.75 (dd, 1H, $J_{H,H} = 14.0 \text{ Hz}$, $J_{P,H} = 13.3 \text{ Hz}$, $CH_2(a)$; 2.83 (dd, 1H $J_{H,H} = 11.5 \text{ Hz}, J_{P,H} = 2.4 \text{ Hz}, \text{ CH}'_2(a)$); 3.03 (dd, 2H, $J_{H,H} = 10.8 \text{ Hz}$, $CH'_{2}(b)$); 3.20 (dd, 1H, $J_{P,H} = 2.4 \text{ Hz}, \text{ CH}_2(b)); 3.94 \text{ (s, 3H, CH}_3O); 6.87 \text{ (d,}$ 1H, H-3); 6.90-6.96 (m, 2H, m-, m'-PC₆H₄); 7.20-7.34 (m, 6H, H-7,7',8,8', o'-, p-PC₆H₄); 7.39–7.45 (m, 2H, H-6,6'); 7.65 (d, 1H), 7.71 (dd, 1H), (H-3,4'); 7.83–7.93 (3d, 3H, H-4,5,5'). ¹³C NMR (100.6 MHz, CDCl₃), (not all signals are given): $\delta = 28.9$ (d, J = 13.3 Hz, CH₂); 29.2 (d, J = 19.2 Hz, CH₂); 55.6 (CH₃O); 110.0 $(m-PC_6H_4)$; 120.8 $(m'-PC_6H_4)$; 124.9, 125.0 (C-6,6'); 125.7, 125.9 (C-7,7'); 126.6, 126.7 (C-8,8'); 132.1 (d, C-1'); 132.7 (d, C-2'); 133.9 (d, C-1); 134.3 (C-2); 160.2 (d, o-PC₆H₄). ³¹P NMR (162 MHz, CDCl₃): $\delta = -2.32$. IR (Nujol): 3050s, 2726w, 1591w, 1570w, 1507w, 1459s, 1430m, 1377m, 1239m, 1175w, 1161w, 1021m, 935w, 866w, 827m, 809m, 747s. MS (ESI): m/z $(\%) = 418 ([M^+], 100); 403 (4); 387 (6); 311 (2); 281$

(39); 265 (44); 209 (8); 141 (14); 91 (18). HRMS calcd for $C_{29}H_{23}OP$: 418.14865. Found: 418.14840. $[\alpha]_D^{25} = -189$ (*c* 0.5, toluene).

4.2.10. 4-(2-Fluoro)-phenyl-4,5-dihydro-3*H*-dinaphtho-[2,1-*c*;1',2'-*e*]phosphepine 4j.



Yield: 67%. ¹H NMR (400 MHz, CDCl₃): δ = 2.82 ('t', 1H, $J_{H,H} = 14.5 \text{ Hz}$, $J_{P,H} = 14.2 \text{ Hz}$, $CH'_2(a)$; 2.87 (dd, 1H, $J_{H,H} = 11.5$ Hz, $J_{P,H} = 2.5$ Hz, CH₂(a)); 3.09 (dd, 1H, $J_{P,H} = 16.7$ Hz, CH₂(b)); 3.13 (dd, 1H, $J_{P,H} = 4.5$ Hz, CH₂(b)); 6.95–7.00 (m, 3H, H-3', o'-, m'-PC₆H₄); 7.08–7.18 (m, 1H, m-PC₆H₄); 7.20–7.34 (m, 5H, H-7,7',8,8', p-PC₆H₄); 7.44–7.50 (m, 2H, H-6,6'); 7.69 (d, 1H), 7.71 (dd, 1H), (H-3,4'); 7.90 (d, 1H), 7.99 (2d, 2H), (H-4,5,5'). ¹³C NMR (100.6 MHz, CDCl₃), (not all signals are given): $\delta = 29.2$ (dd, $J_{P,C} = 16.2 \text{ Hz}, \quad J_{F,C} = 2.8 \text{ Hz}, \quad CH_2$; 29.8 (dd, $J_{P,C} = 22.5 \text{ Hz}, \quad J_{F,C} = 2.5 \text{ Hz}, \quad CH'_2$); 115.1 (d, *m*-PC₆H₄); 124.1 (d, *m*'-PC₆H₄); 124.8, 125.1 (C-6,6'); 125.8, 125.9 (C-7,7'); 126.6, 126.7 (C-8,8'); 127.3 (d, $J_{P,C} = 2.0 \text{ Hz}$; 127.5 (C-4'); 128.1–128.4 (C-3',4,5,5'); 130.1 (d, p-PC₆H₄); 131.2 (dd, o'-PC₆H₄); 132.5 (d, C-1'); 132.9 (d, C-2'); 133.8 (C-2); 133.8 (d, C-1); 163.3 (dd, *o*-PC₆H₄). ³¹P NMR (162MHz, CDCl₃): $\delta = -3.2$ (d, $J_{P,F} = 34.5 \text{ Hz}$). IR (Nujol): 2668w, 1909w, 1710w, 1591w, 1507w, 1467s, 1377m, 1260w, 1213w, 1160w, 1072w, 1024w, 935w, 864w, 822m, 751m, 673w, 566w, 520w. MS (ESI): m/z (%) = 406 ([M⁺], 100); 311 (2); 279 (12); 265 (27); 183 (4); 141 (4); 96 (2); 75 (3). HRMS calcd for C₂₈H₂₀PF: 406.12454. Found: 406.12466. $[\alpha]_{\rm D}^{23} = -157 \ (c \ 0.5, \ toluene).$

4.2.11. 4-(4-Fluoro)-phenyl-4,5-dihydro-3*H*-dinaphtho-[2,1-*c*;1',2'-*e*]phosphepine 4k.



Yield: 45%. ¹H NMR (400 MHz, CDCl₃): δ = 2.71 (dd, 1H, $J_{H,H}$ = 14.3 Hz, $J_{P,H}$ = 4.8 Hz, CH₂(a)); 2.87–2.94 (m, 2H, CH₂(a), CH₂(b)); 3.02 (dd, 1H, $J_{H,H}$ = 16.6 Hz, $J_{P,H}$ = 11.7 Hz, CH₂'(b)); 6.94 (d, 1H, H-3'); 7.04 (m, 2H, m-PC₆H₄); 7.21–7.35 (m, 6H, H-7,7',8,8', *o*-PC₆H₄); 7.45–7.51 (m, 2H, H-6,6'); 7.70 (dd, 1H), 7.79 (d, 1H), (H-3,4'); 7.93–8.01 (m, 3H, H-4,5,5'). ¹³C NMR (100.6 MHz, CDCl₃): δ = 30.6 (d, J = 15.3 Hz, CH₂); 32.5 (d, J = 22.0 Hz, CH₂'); 115.3 (dd, $J_{F,C}$ = 21.0 Hz, $J_{P,C}$ = 6.6 Hz, m-PC₆H₄); 124.9, 125.1 (C-6,6'); 125.9, 126.0 (C-7,7'); 126.6 (C-8,8'); 127.3 (d, C-3); 127.6 (C-4'); 127.9 (C-3'); 128.2, 128.3, 128.4 (C-4,5,5'); 131.9, 132.3, 132.8, 132.8 (C-9,9',10,10'); 132.2 (d, C-1'); 133.2 (C-2'); 133.6 (d, C-1); 133.8 (dd, $J_{P,C} = 20.0$ Hz, $J_{F,C} = 7.6$ Hz, o-PC₆H₄); 134.0 (C-2); 134.7 (d, $J_{P,C} = 83.9$ Hz, i-PC₆H₄); 163.4 (d, $J_{F,C} = 249$ Hz, p-PC₆H₄). ³¹P NMR (162 MHz, CDCl₃): $\delta = 5.1$ (d, $J_{P,F} = 2.8$ Hz). IR (Nujol): 3046w, 1588m, 1507w, 1495s, 1462m, 1377w, 1234m, 1160w, 1092w, 1026w, 935w, 865w, 822m, 751m, 671w, 566w, 513w. MS (ESI): m/z (%) = 406 ([M⁺], 100); 311 (2); 279 (12); 265 (27); 183 (4); 141 (4); 96 (2); 75 (3). HRMS calcd for C₂₈H₂₀PF: 406.12867. Found: 406.12866. $[\alpha]_D^{23} = -16$ (c 0.5, toluene).

4.2.12. 4-(2-Naphthyl)-4,5-dihydro-3*H*-dinaphtho[2,1*c*;1',2'-*e*]phosphepine 4l.



Yield: 80%. ¹H NMR (400 MHz, CDCl₃), (only signals for the CH₂ groups are given): $\delta = 2.87$ (dd, 1H, $J_{\rm H,H} = 14.5$ Hz, $J_{\rm P,H} = 4.8$ Hz, CH₂(a)); 2.96 (dd, 1H, $J_{\rm P,H} = 11.8$ Hz, CH₂(b)); 3.07 (dd, 1H, $J_{\rm H,H} = 11.7$ Hz, $J_{\rm P,H} = 2.8$ Hz, CH₂(a)); 3.13 (dd, 1H, $J_{\rm P,H} = 16.1$ Hz, CH₂(b)). ¹³C NMR (100.6 MHz, CDCl₃), (not all signals are given): $\delta = 30.5$ (d, J = 16.5 Hz, CH₂); 32.1 (d, J = 23.0 Hz, CH₂); 138.1 (d, 2-Naph). ³¹P NMR (162 MHz, CDCl₃): $\delta = 3.82$. IR (Nujol): 2752w, 1916w, 1695w, 1591w, 1506m, 1462s, 1377m, 1338w, 1268w, 1218w, 1159w, 1025w, 932w, 862w, 811s, 742s, 681w, 660w, 636w, 476m. MS (ESI): m/z (%) = 438 ([M⁺], 100); 309 ([M⁺-naphth], 3); 265 (24); 157 (8); 141 (4); 132 (7). HRMS calcd for C₃₂H₂₃P: 438.15375. Found: 438.15349. [α]_P²⁴ = -89 (*c* 0.5, toluene).





Yield: 61%. ¹H NMR (400 MHz, CDCl₃): 2.79 (dd, 1H, $J_{H,H} = 14.5$ Hz, $J_{P,H} = 4.5$ Hz, $CH_2(a)$); 2.89 (dd, 1H, $J_{P,H} = 11.6$ Hz, $CH_2(b)$); 2.93 (dd, 1H, $J_{H,H} = 11.7$ Hz, $J_{P,H} = 3.3$ Hz, $CH'_2(a)$); 3.02 (dd, 1H, $J_{P,H} = 16.7$ Hz, $CH'_2(b)$); 3.75 (s, 3H, CH₃O); 6.81–6.88 (m, 3H, *o*-, *o'*-, p-PC₆H₄); 6.99 (d, 1H, H-3'); 7.17–7.31 (m, 5H, H-7,7',8,8', *m'*-PC₆H₄); 7.39–7.46 (m, 2H, H-6,6'); 7.68 (d, 1H, H-3); 7.74 (d, 1H, H-4'); 7.87–7.95 (m, 3H, H-4,5,5'). ¹³C NMR (100.6 MHz, CDCl₃): $\delta = 30.5$ (d, J = 15.4Hz, CH₂); 32.1 (d, J = 21.0Hz, CH'₂); 55.2 (CH₃O); 114.5 (*p*-PC₆H₄); 116.8 (d, *o*-PC₆H₄); 124.0 (d, *o'*-PC₆H₄); 124.9, 125.1 (C-6,6'); 125.8, 125.9 (C-7,7'); 126.6, 126.7 (C-8,8'); 127.3 (d, C-3); 127.4 (C-4'); 128.0–128.4 (C-3',4,5,5'); 129.2 (d, *m'*-PC₆H₄); 131.9, 132.2, 132.8, 132.8 (C-9,9',10,10'); 132.3 (d, C-1'); 133.6 (d, C-2'); 133.7 (d, C-1); 134.3 (C-2); 139.2 (d, *i*-PC₆H₄); 159.1 (d, *m*-PC₆H₄). ³¹P NMR (162 MHz, CDCl₃): δ = 7.8. IR (KBr): 3049m, 3005 m, 2952 m, 2912 m, 2832 m, 1617 w, 1590 s, 1572 s, 1507 s, 1479 s, 1462 m, 1414 s, 1358 m, 1317 w, 1284 m, 1247 s, 1181 m, 1160 m, 1145 m, 1102 w, 1041 s, 992 m, 960 w, 934 m, 864 m, 814 s, 773 s, 748 s, 691 s, 658 m, 621 m, 565 m, 541 w, 520 m, 465 m, 417 m. MS (ESI): *m*/*z* (%) = 418 ([M⁺], 100); 281 (6); 276 (30); 265 (31); 252 (5); 108 (5). HRMS calcd for C₂₉H₂₃OP: 418.14865. Found: 418.14343. [α]_D²⁴ = -15 (*c* 0.3, toluene).

4.2.14. 4-(3,4-Dimethoxy)-phenyl-4,5-dihydro-3*H*-dinaphtho[2,1-*c*;1',2'-*e*]phosphepine 4n.



Yield: 67%. ¹H NMR (400 MHz, CDCl₃): δ = 2.68 (dd, 1H, $J_{H,H} = 14.3$ Hz, $J_{P,H} = 4.5$ Hz, $CH_2(a)$); 2.89 (dd, 1H, $J_{P,H} = 11.7$ Hz, $CH_2(b)$); 2.96 (dq, 2H, $J_{H,H} = 11.7$ Hz, CH'_2); 3.73 (s, 3H, CH_3O); 3.90 (s, 3H, CH_3O); 6.70 (dd, 1H, *o*-PC₆H₃); 6.81 (d, 1H, *m'*- PC_6H_3 ; 6.89 (m, 1H, o'-PC_6H_3); 7.03 (d, 1H, H-3'); 7.17-7.32 (m, 4H, H-7,7',8,8'); 7.40-7.47 (m, 2H, H- $\delta = 31.1$ (d, J = 15.5 Hz, CH₂); 32.7 (d, J = 22.0 Hz, CH₂); 55.8, 55.8 (2CH₃O); 110.9 (d, m'-PC₆H₃); 114.6 (d, o-PC₆H₃); 124.9, 125.0 (C-6, 6'); 125.6 (d, o'-PC₆H₃); 125.9, 125.9 (C-7,7'); 126.62, 126.65 (C-8, 8'); 127.3 (d, C-3); 127.4 (C-4); 128.0, 128.2, 128.4, 128.5 (C-3',4,5,5'); 132.0, 132.3, 132.8, 132.8 (C-9, 9',10,10'); 132.2 (d, C-1'); 133.5 (d, C-1); 133.6 (C-2'); 134.3 (C-2); 148.5 (d, *m*-PC₆H₃); 149.1 (*p*-PC₆H₃). ³¹P NMR (162 MHz, CDCl₃): $\delta = 6.0$. IR (KBr): 3056 m, 3002 m, 2955 m, 2835 m, 1592 s, 1506 s, 1463 s, 1441 s, 1397 m, 1359 w, 1329 m, 1254 s, 1231 s, 1176 s, 1144 m, 1124 s, 1027 s, 912 w, 867 w, 853 w, 814 m, 745 s, 683 w, 660 w, 622 w, 581 w, 544 w, 468 w, 419 m. MS (ESI): m/z (%) = 448 ([M⁺], 100); 433 (7); 276 (17); 265 (19); 151 (15). HRMS calcd for $C_{30}H_{25}O_2P$: 448.15921. Found: 448.15980. $[\alpha]_D^{24} = -33.5$ (*c* 0.4, toluene).

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References

- (a) Noyori, R. Asymmetric Catalysis in Organic Synthesis; Wiley: New York, 1994; (b) Transition Metals for Organic Synthesis; Beller, M., Bolm, C., Eds.; Wiley-VCH: Weinheim, 1998; (c) Comprehensive Asymmetric Catalysis; Jacobsen, E. N., Pfaltz, A., Yamamoto, H., Eds.; Springer: Berlin, 1999.
- Reviews: (a) MacCarthy, M.; Guiry, P. J. Tetrahedron 2001, 57, 3809–3843; (b) Zhang, X. Enantiomer 1999, 4, 541–555; (c) Gladiali, S.; Fabbri, D. Chem. Ber./Recl. 1997, 130, 543–554; (d) Lagasse, F.; Kagan, H. B. Chem. Pharm. Bull. 2000, 48, 315–324.
- Comprehensive Asymmetric Catalysis; Jacobsen, E. N., Pfaltz, A., Yamamoto, H., Eds.; Springer: Berlin, 1999; *Transition Metals for Organic Synthesis*; Beller, M., Bolm, C., Eds.; Wiley-VCH: Weinheim, 1998.
- (a) Reetz, M. T.; Sell, T. *Tetrahedron Lett.* 2000, 41, 6333–6336; (b) Reetz, M. T.; Mehler, G. Angew. Chem., *Int. Ed.* 2000, 39, 3889–3891.
- (a) Arnold, L. A.; Imobos, R.; Manoli, A.; de Vries, A. H. M.; Naasz, R.; Feringa, B. *Tetrahedron* 2000, 56, 2865–2878; (b) van den Berg, M.; Minnaard, A. J.; Schudde, E. P.; van Esch, J.; de Vries, A. H. M.; de Vries, J. G.; Feringa, B. L. J. Am. Chem. Soc. 2000, 122, 11539–11540; (c) Minnaard, A. J.; vanden Berg, M.; Schudde, E. P.; van Esch, J.; de Vries, A. H. M.; de Vries, J. G.; Feringa, B. Chim. Oggi 2001, 19, 12–13.
- (a) Claver, C.; Fernandez, E.; Gillon, A.; Heslop, K.; Hyett, D. J.; Martorell, A.; Orpen, A. G.; Pringle, P. G. *Chem. Commun.* 2000, 961–962; (b) Martorell, A.; Naasz, R.; Feringa, B. L.; Pringle, P. G. *Tetrahedron: Asymmetry* 2001, *12*, 2497–2499.
- Recent review about monophosphines: (a) Komarov, I. V.; Börner, A. Angew. Chem., Int. Ed. 2001, 40, 1197–1200; some successful previous examples: (b) Guillen, F.; Fiaud, J.-F. Tetrahedron Lett. 1999, 40, 2939–2942; (c) Knowles, W. S.; Sabacky, M. J.; Vineyard, B. D. J. Chem. Soc., Chem. Commun. 1972, 10–11; (d) Morrison, J. D.; Burnett, R. E.; Aguiar, A. M.; Morrow, C. J.; Phillips, C. J. Am. Chem. Soc. 1971, 93, 1301–1303; (e) Burk, M. J.; Feaster, J. E.; Harlow, R. L. Tetrahedron: Asymmetry 1991, 2, 569–592; (f) Marinetti, A.; Mathey, F.; Ricard, I. Organometallics 1993, 12, 1207–1212.
- 8. (a) Alexakis, A.; Benhhaim, C.; Rosset, S.; Human, M. J. Am. Chem. Soc. 2002, 124, 5262-5263; (b) Boezio, A. A.; Charette, A. B. J. Am. Chem. Soc. 2003, 125, 1692-1693; (c) Piarulli, U.; Claverie, C.; Daudos, P.; Gennari, C.; Minnaard, A. J.; Feringa, B. L. Org. Lett. 2003, 23, 4493-4496; (d) Jia, X.; Li, X.; Shi, Q.; Yao, X.; Chan, A. S. J. Org. Chem. 2003, 68, 4539-4541; (e) Li, X.; Jia, X.; Lu, G.; Au-Yeung, T. T.-L.; Lam, K.-H.; Lo, T. W. H.; Chan, A. S. Tetrahedron: Asymmetry 2003, 14, 2687–2691; (f) Au-Yeung, T. T.-L.; Chan, S.-S.; Chan, A. S. Adv. Synth. Catal. 2003, 345, 537-555; (g) Reetz, M. T.; Oka, H.; Goddard, R. Synthesis 2003, 12, 1809-1814; (h) Jerphagon, T.; Renaud, J.-L.; Demonchaux, P.; Ferreira, A.; Bruneau, C. Adv. Synth. Catal. 2004, 346, 33-36; (i) van Zijl, A. W.; Arnold, L. A.; Minnaard, A. J.; Feringa, B. L. Adv. Synth. Catal. 2004, 346, 413-420.
- 9. Chen, W.; Xiao, J. Tetrahedron Lett. 2001, 42, 8737-8740.
- Hua, Z.; Vassar, V. C.; Ojima, I. Org. Lett. 2003, 21, 3831–3834.
- (a) Hu, A.-G.; Fu, Y.; Xie, J.-H.; Zhou, H.; Wang, L.-X.; Zhou, Q.-L. Angew. Chem., Int. Ed. 2002, 41, 2348–2350;
 (b) Fu, Y.; Xie, J.-H.; Hu, A.-G.; Zhou, H.; Wang, L.-X.; Zhou, Q.-L. Chem. Commun. 2002, 480.
- Ostermeier, M.; Prieß, J.; Helmchen, G. Angew. Chem., Int. Ed. 2002, 41, 612–617.

- For bidentate ligands with 4,5-dihydro-3*H*-dinaphtho-[2,1-c;1',2'-e]phosphepine core see: (a) Tang, W.; Wang, W.; Chi, Y.; Zhang, X. Angew. Chem., Int. Ed. 2003, 42, 3506–3509; (b) Chi, Y.; Zhou, Y.-G.; Zhang, X. J. Org. Chem. 2003, 68, 4120–4122; (c) Xiao, D.; Zhang, X. Angew. Chem., Int. Ed. 2001, 40, 3425–3428; (d) Xiao, D.; Zhang, Z.; Zhang, X. Org. Lett. 1999, 1, 1679–1681.
- Gladiali, S. et al. unpublished work; see also Gladiali, S.; Dore, A.; Fabbri, D.; De Lucchi, O.; Manassero, M. *Tetrahedron: Asymmetry* 1994, 5, 511–514.
- 15. Chi, Y.; Zhang, X. Tetrahedron Lett. 2002, 43, 4849-4852.
- Junge, K.; Oehme, G.; Monsees, A.; Riermeier, T.; Dingerdissen, U.; Beller, M. *Tetrahedron Lett.* 2002, 43, 4977–4980.
- Junge, K.; Oehme, G.; Monsees, A.; Riermeier, T.; Dingerdissen, U.; Beller, M. J. Organomet. Chem. 2003, 675, 91–96.
- 18. X-ray data were collected with a STOE-IPDS diffractometer using graphite-monochromated Mo-K α radiation. The structures were solved by direct methods (SHELXS-86: Sheldrick, G. M. *Acta Crystallogr.* **1990**, *A46*, 467) and refined by full-matrix least-squares techniques against F^2 (SHELXL-93: Sheldrick, G. M. University of Göttingen, Germany, **1993**). XP (BRUKER AXS) was used for structure representations. Crystal data of compound **4b**: C₂₅H₂₃P, M = 354.40, tetragonal, space group *P4*₃, a = 11.275(2), c = 15.783(3)Å, V = 2006.4(6)Å³, T =

 μ (Mo-K α) = 0.142 mm⁻¹, *Z* = 4. 293(2)K, $D_c =$ 1.173 g cm^{-3} , 10,876 reflections measured, 3212 were independent of symmetry and 2201 were observed $[I > 2\sigma(I)], R_1 = 0.0426, wR_2(all data) = 0.0967, 235$ parameters. Crystal data of compound 4e: C₂₉H₂₃OP, M = 418.44, monoclinic, space group $P2_1$, a = 9.503(2), b = 16.723(3),c = 14.149(3)Å, $\beta = 91.32(3)^{\circ},$ V =2247.9(8)Å³, T = 293(2)K, Z = 4, μ (Mo-K α) = 0.141 mm⁻¹, $D_c = 1.236$ g cm⁻³; 8352 reflections measured, 4700 were independent of symmetry and 3477 were observed $[I > 2\sigma(I)]$, $R_1 = 0.0568$, wR_2 (all data) = 0.1275, 559 parameters. Crystallographic data for the structures in this paper have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication number CCDC 237562 and CCDC 237563. Copies of the data can be obtained free of charge, on application to CCDC, 12 Union Road, Cambridge, CB2 1EZ. UK [fax: +44(0)1223-336033 or e-mail: deposit@ccdc.ac.uk].

- For the beneficial effect of SDS in asymmetric hydrogenations see for example: (a) Grassert, I.; Paetzold, E.; Oehme, G. *Tetrahedron* 1993, 49, 6605–6612; (b) Kumar, A.; Oehme, G.; Roque, J. P.; Schwarze, M.; Selke, R. *Angew. Chem., Int. Ed. Engl.* 1994, 33, 2197– 2199.
- Heller, D.; Drexler, H.-J.; Spannenberg, A.; Heller, B.; You, J.; Baumann, W. Angew. Chem., Int. Ed. 2002, 41, 777–780.