

**MANNICH REACTION IN THE SYNTHESIS OF  
N,S-CONTAINING HETEROCYCLES. 13\*. ONE-POT  
METHOD FOR PREPARING PYRIMIDO[4,3-*b*]-  
[1,3,5]THIADIAZINES BY REACTION OF ALDEHYDES,  
CYANTHIOACETAMIDE, FORMALDEHYDE,  
AND PRIMARY AMINES**

**V. V. Dotsenko<sup>1\*\*</sup>, K. A. Frolov<sup>1</sup>, and S. G. Krivokolysko<sup>1</sup>**

*Successive reaction of cyanothioacetamide with aliphatic or aromatic aldehydes, formaldehyde, and primary amines gives the corresponding 3,7-disubstituted 3,4,7,8-tetrahydro-2H,6H-pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitriles.*

**Keywords:** cyanothioacetamide, pyrimido[4,3-*b*][1,3,5]thiadiazines, aminomethylation, Mannich reaction, multicomponent condensation.

The double aminomethylation of different N,S-dinucleophiles including dithiocarbamates, primary thioamides, and 2-mercaptoazoles (azines) can be deservedly considered as one of the most simple and convenient methods for constructing the 1,3,5-thiadiazine system [2-4]. This route was used successfully in the synthesis of condensed thiadiazines including substituted bis(pyrido[2,1-*b*][1,3,5]thiadiazin-7-yl)methanes [5], 1,2,4-triazolo[3,4-*b*][1,3,5]thiadiazines [6], imidazo[2,1-*b*][1,3,5]thiadiazines [7-16], 1,2,4-triazino[3,2-*b*][1,3,5]thiadiazines [15], thiazolo[3',4':1,5][1,2,4]triazolo[3,4-*b*][1,3,5]thiadiazines [17], 1,3,5-thiadiazino[3,2-*a*]benzimidazoles [18], cyclopenta[*g*]pyrido[2,1-*b*][1,3,5]thiadiazines [19], pyrimido[2,1-*b*][1,3,5]thiadiazines [20], and pyrido[2,1-*b*][1,3,5]thiadiazines [21].

We have previously reported the preparation of 8-aryl-3,4,7,8-tetrahydro-2H,6H-pyrimido[4,3-*b*]-[1,3,5]thiadiazine-9-carbonitriles **1** by treatment of 3-aryl-2-cyanoprop-2-enethioamides **2** with primary amines and HCHO [22] (method A), by recyclization of 4H-thiopyrans **3** under the Mannich reaction conditions [23] (method B), or by the aminomethylation of tetrahydropyridine-2-thiolates **4** [24] (method C). The said methods are not free from limitations and drawbacks. Method A allows preparation of pyrimidothiadiazines **1** with acceptable yields (up to 65%) but is limited to a range of products consisting only of 8-aryl derivatives of compound **1** since the 3-alkyl analogs of the starting  $\alpha,\beta$ -unsaturated thioamides **2** are unknown at this time.

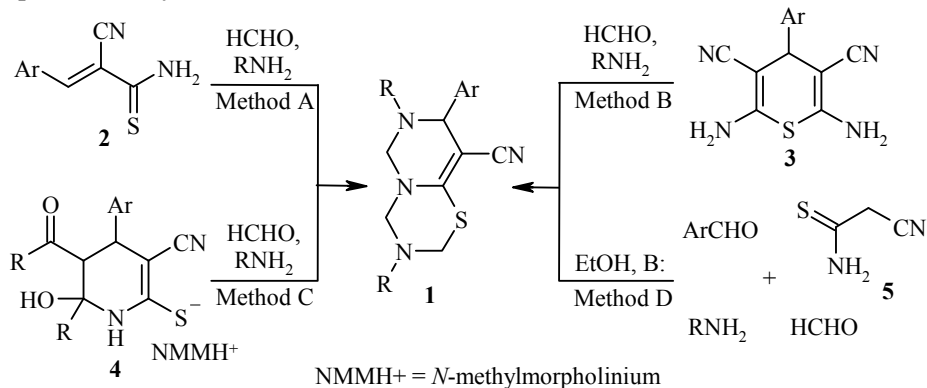
\*For Communication 12, see [1].

\*\*To whom correspondence should be addressed, e-mail: Victor\_Dotsenko@bigmir.net.

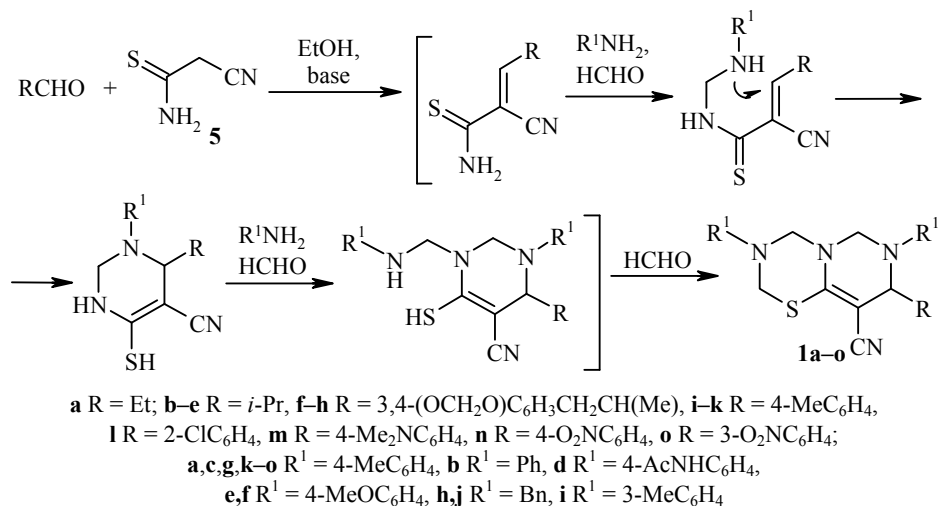
<sup>1</sup>"ChemEx Laboratory", Vladimir Dal' East-Ukrainian National University, 20-A/7 Molodezhny Kv., Lugansk 91034, Ukraine.

This limitation is removed with the preparation of thiadiazines **1** by the aminomethylation of 4*H*-thiopyrans **3** using method B, but, in this case, the yields of the target products are lower and the route itself is atom inefficient. Method C gives the lowest yields of thiadiazines **1** (up to 28%) and seems the least rational.

In the earlier study [23], we declared the possible preparation of pyrimido[4,3-*b*][1,3,5]thiadiazines **1** by a multicomponent condensation of aromatic aldehydes, cyanothioacetamide **5**, *p*-toluidine, and HCHO (method D). In searches for the optimum approach to the pyrimido[4,3-*b*][1,3,5]thiadiazines **1** synthesis, we have studied and summerized in the present work the overall possibilities for a multicomponent approach and, in particular, the possibility of varying the substituents in the 8 position of the pyrimido[4,3-*b*][1,3,5]thiadiazine system **1** and the introduction of aliphatic aldehydes into this reaction.



It was found that successive reaction of cyanothioacetamide (**5**) with aldehydes, primary amines, and formaldehyde occurs under mild conditions (short term heating in aqueous EtOH in the presence of catalytic amounts of base) and leads to the pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitriles **1a-o** in 11-70% yield (Table 1). Both aromatic and aliphatic aldehydes can be used in the reaction.



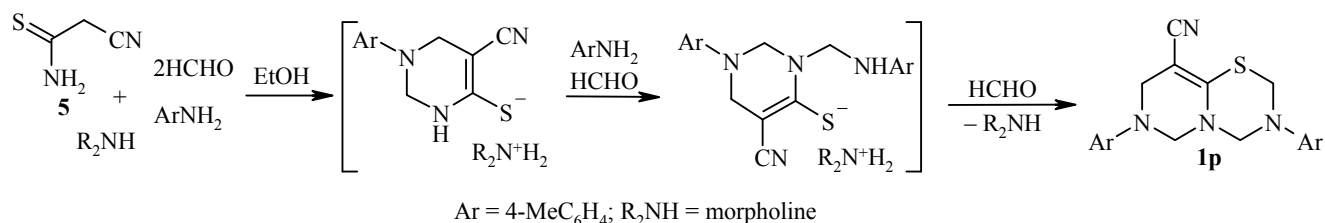
In the latter case, the previously unknown 8-alkylpyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitriles **1a-g** are formed in low yields (up to 40%). The use of aromatic aldehydes in this reaction usually gives higher yields of the target products **1i-o** when compared with yields using aliphatic aldehydes. Compound **1h** could not be prepared in a pure state. In this case, synthesis by the general method gives an uncrystallizable oil which contains about 73% of the target product according to HPLC-MS and <sup>1</sup>H NMR spectroscopy. Attempts to purify the

TABLE 1. Physicochemical and Spectroscopic Characteristics of Compounds **1a-p**

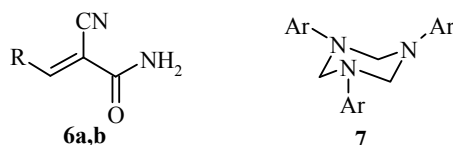
| Com-<br>pound | Empirical<br>formula                                            | Found, %              |                     |                       | Mp, °C                                         | IR spectrum,<br>ν, cm <sup>-1</sup>   | Yield,<br>%      |
|---------------|-----------------------------------------------------------------|-----------------------|---------------------|-----------------------|------------------------------------------------|---------------------------------------|------------------|
|               |                                                                 | Calculated, %         |                     |                       |                                                |                                       |                  |
|               |                                                                 | C                     | H                   | N                     |                                                |                                       |                  |
| <b>1a</b>     | C <sub>23</sub> H <sub>26</sub> N <sub>4</sub> S                | <u>70.65</u><br>70.73 | <u>6.75</u><br>6.71 | <u>14.50</u><br>14.35 | 175-176 (decomp.)<br>(Me <sub>2</sub> CO)      | 2165 (C≡N)                            | 17               |
| <b>1b</b>     | C <sub>22</sub> H <sub>24</sub> N <sub>4</sub> S                | <u>69.88</u><br>70.18 | <u>6.39</u><br>6.42 | <u>14.99</u><br>14.88 | 140-142 (decomp.)<br>(Me <sub>2</sub> CO-EtOH) | 2173 (C≡N)                            | 29               |
| <b>1c</b>     | C <sub>24</sub> H <sub>28</sub> N <sub>4</sub> S                | <u>71.08</u><br>71.25 | <u>7.09</u><br>6.98 | <u>14.00</u><br>13.85 | 171-173 (decomp.)<br>(Me <sub>2</sub> CO)      | 2164 (C≡N)                            | 39               |
| <b>1d</b>     | C <sub>26</sub> H <sub>30</sub> N <sub>6</sub> O <sub>2</sub> S | <u>63.23</u><br>63.65 | <u>6.20</u><br>6.16 | <u>17.26</u><br>17.13 | 169-171 (decomp.)<br>(Me <sub>2</sub> CO)      | 2167 (C≡N)<br>1675 (C=O)<br>3300 (NH) | 19               |
| <b>1e</b>     | C <sub>24</sub> H <sub>28</sub> N <sub>4</sub> O <sub>2</sub> S | <u>65.65</u><br>66.03 | <u>6.52</u><br>6.46 | <u>12.53</u><br>12.38 | 152-153 (decomp.)<br>(Me <sub>2</sub> CO-EtOH) | 2172 (C≡N)                            | 26               |
| <b>1f</b>     | C <sub>31</sub> H <sub>32</sub> N <sub>4</sub> O <sub>4</sub> S | <u>66.40</u><br>66.89 | <u>5.81</u><br>5.79 | <u>10.14</u><br>10.06 | 164-165 (decomp.)<br>(DMF-Me <sub>2</sub> CO)  | 2167 (C≡N)                            | 30               |
| <b>1g</b>     | C <sub>31</sub> H <sub>32</sub> N <sub>4</sub> O <sub>2</sub> S | <u>70.53</u><br>70.96 | <u>6.18</u><br>6.15 | <u>10.79</u><br>10.68 | 185-187 (decomp.)<br>(Me <sub>2</sub> CO)      | 2170 (C≡N)                            | 37               |
| <b>1h</b>     | —                                                               | —                     | —                   | —                     | —                                              | 2170 (C≡N)                            | -                |
| <b>1i</b>     | C <sub>28</sub> H <sub>28</sub> N <sub>4</sub> S                | <u>74.03</u><br>74.30 | <u>6.28</u><br>6.24 | <u>12.49</u><br>12.38 | 177-179<br>(DMF-EtOH)                          | 2176 (C≡N)                            | 62               |
| <b>1j</b>     | C <sub>28</sub> H <sub>28</sub> N <sub>4</sub> S                | <u>74.14</u><br>74.30 | <u>6.22</u><br>6.24 | <u>12.50</u><br>12.38 | 128-131 (DMF)<br>(125-128 [22])                | 2176 (C≡N)                            | 65               |
| <b>1k</b>     | C <sub>28</sub> H <sub>28</sub> N <sub>4</sub> S                | <u>74.16</u><br>74.30 | <u>6.25</u><br>6.24 | <u>12.53</u><br>12.38 | 177-176<br>(DMF-EtOH)<br>(178-181 [22])        | 2172 (C≡N)                            | 70               |
| <b>1l</b>     | C <sub>27</sub> H <sub>25</sub> ClN <sub>4</sub> S              | <u>68.32</u><br>68.56 | <u>5.27</u><br>5.33 | <u>12.01</u><br>11.84 | 210-212<br>(DMF-EtOH)<br>(210-211 [22])        | 2167 (C≡N)                            | 66               |
| <b>1m</b>     | C <sub>29</sub> H <sub>31</sub> N <sub>5</sub> S                | <u>72.11</u><br>72.32 | <u>6.50</u><br>6.49 | <u>14.73</u><br>14.54 | 165-167 (decomp.)<br>(DMF-Me <sub>2</sub> CO)  | 2165 (C≡N)                            | 26               |
| <b>1n</b>     | C <sub>27</sub> H <sub>25</sub> N <sub>5</sub> O <sub>2</sub> S | <u>66.75</u><br>67.06 | <u>5.25</u><br>5.21 | <u>14.69</u><br>14.48 | 213-215 (decomp.)<br>(DMF-EtOH)                | 2167 (C≡N)                            | 36               |
| <b>1o</b>     | C <sub>27</sub> H <sub>25</sub> N <sub>5</sub> O <sub>2</sub> S | <u>66.88</u><br>67.06 | <u>5.22</u><br>5.21 | <u>14.65</u><br>14.48 | 205-207 (decomp.)<br>(DMF-EtOH)                | 2168 (C≡N)                            | 33               |
| <b>1p</b>     | C <sub>21</sub> H <sub>22</sub> N <sub>4</sub> S                | <u>69.37</u><br>69.58 | <u>6.15</u><br>6.12 | <u>15.55</u><br>15.46 | 195-197 (decomp.)<br>(DMF-Me <sub>2</sub> CO)  | 2158 (C≡N)                            | 11 (A)<br>13 (B) |

compound **1h** or to separate it from the mixture using column chromatography were unsuccessful. In the case of aliphatic aldehydes it was found that morpholine is preferred to tertiary amine catalysts such as *N*-methylmorpholine or triethylamine.

A very unexpected result was obtained in the condensation of acetaldehyde, thioamide **5**, *p*-toluidine, and formaldehyde. Instead of the expected 8-methyl-3,7-di(4-methylphenyl)-3,4,7,8-tetrahydro-2*H*,6*H*-pyrimido-[4,3-*b*][1,3,5]thiadiazine-9-carbonitrile its 8-unsubstituted analog **1p** was formed. Compound **1p** was also found as an admixture (up to 20% according to <sup>1</sup>H NMR data) in unpurified samples of compounds **1a,c**. Clearly formation of the pyrimidothiadiazine system precedes the *N*-aminomethylation and retro-Knoevenagel reaction. A possible reason for the anomalous reaction course is the ease of the retro-reaction and the comparable rate of the reaction of the cyanothioacetamide with formaldehyde and aliphatic aldehydes. Compound **1p** was also obtained by the direct aminomethylation of the cyanothioacetamide (**5**).



Finally, attempts to broaden the boundaries of the use of this route and to exchange the cyanothioacetamide (**5**) in this reaction for cyanoacetamide or to introduce into the reaction 3-aryl-2-cyano-prop-2-enamides **6a,b** in place of compounds **2** were not successful. Hence under the said conditions only the perhydro-1,3,5-triazine **7** (a known product of *p*-toluidine condensation with formaldehyde [25]) was isolated in place of the 3-(2-chlorophenyl)-2-cyanoprop-2-enamide (**6a**) aminomethylation products. Polycomponent condensation of furfural, cyanoacetamide, HCHO, and *p*-toluidine gives a mixture of the unsaturated amide **6b** (R = 2-furyl) and the perhydro-1,3,5-triazine **7**.



**6 a** R = 2-ClC<sub>6</sub>H<sub>4</sub>, **b** 2-furyl; **7** Ar = 4-MeC<sub>6</sub>H<sub>4</sub>

Hence the 3,4,7,8-tetrahydro-2*H*,6*H*-pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitriles can be successfully prepared by condensation of cyanothioacetamide with aliphatic or aromatic aldehydes, formaldehyde, and primary amines. A similar route, in contrast to other methods, allows the preparation of pyrimido[4,3-*b*]-[1,3,5]thiadiazines unsubstituted in position 8 or with alkyl substituents in this position in acceptable yields. Cyanoacetamide and its arylmethyldene derivatives do not take part in a Mannich reaction under these conditions.

## EXPERIMENTAL

IR spectra were recorded on an IKS-29 spectrophotometer in nujol. <sup>1</sup>H NMR spectra were recorded on Varian Unity Plus (400 MHz) and Bruker DRX-500 (500 MHz) spectrometers and <sup>13</sup>C NMR spectra on a Bruker DRX-500 (125 MHz) using DMSO-*d*<sub>6</sub> with TMS as internal standard. HPLC-MS analysis was carried out on an Agilent 1100 liquid chromatograph with DAD and Sedex 75 ELSD detectors combined with an Agilent LC/MSD VL mass spectrometer in electrospray ionization mode. Elemental analysis was performed on a Carlo-Erba 1106 Elemental Analyzer with an accuracy of measurement of ± 0.25%. Monitoring of the purity of the compounds obtained was carried out by TLC using Silufol UV-254 plates with acetone–hexane (1:1) as eluent and visualized by UV light or iodine vapor. Melting points for the prepared compounds were determined on a Kofler hot bench and are not corrected.

The starting cyanothioacetamide (**5**) was prepared by a known method [26].

### 3,4,7,8-Tetrahydro-2*H*,6*H*-pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitriles (**1a-o**) (General Method).

The base (2-3 drops, morpholine in the case of the aliphatic aldehydes, *N*-methylmorpholine or Et<sub>3</sub>N in the case of aromatic aldehydes) was added to a mixture of cyanothioacetamide (**5**) (0.5 g, 5.0 mmol) and the corresponding aldehyde (5.0 mmol) in EtOH (15-20 ml). The product was stirred for 10-20 min at 20°C and the corresponding amine (10.5 mmol) and 37% formalin, free from paraformaldehyde admixture (2.5-3.0 ml, 34-40 mmol) were added. The mixture was then refluxed with vigorous stirring for 1-3 min, a further 15 ml of EtOH added, and stirring continued for 3-5 h at 20°C. The precipitate formed after 24 h was filtered off (in the case of tar formation the solution was decanted) and immediately recrystallized from acetone, a mixture of acetone–EtOH (1: 1), or dissolved in hot DMF with subsequent reprecipitation using EtOH or acetone.

**8-Ethyl-3,7-di(4-methylphenyl)-3,4,7,8-tetrahydro-2*H*,6*H*-pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitrile (**1a**).** Colorless crystals. <sup>1</sup>H NMR spectrum (400 MHz), δ, ppm (*J*, Hz): 1.00 (3H, t, <sup>3</sup>*J* = 7.3, CH<sub>2</sub>CH<sub>3</sub>); 1.62-1.75 (2H, m, CH<sub>2</sub>CH<sub>3</sub>); 2.25 (3H, s, CH<sub>3</sub>); 2.26 (3H, s, CH<sub>3</sub>); 3.70 (1H, dd, <sup>3</sup>*J* = 9.1, <sup>3</sup>*J* = 9.1, H-8); 4.39 (1H, d, <sup>2</sup>*J* = 12.9) and 4.62 (1H, d, <sup>2</sup>*J* = 12.9, NCH<sub>2</sub>N); 4.75 (1H, d, <sup>2</sup>*J* = 12.9) and 4.88 (1H, d,

$^2J = 12.9$ ,  $\text{NCH}_2\text{N}$ ); 4.99 (1H, d,  $^2J = 12.6$ ) and 5.17 (1H, d,  $^2J = 12.6$ ,  $\text{SCH}_2\text{N}$ ); 6.80 (2H, d,  $^3J = 7.9$ , H Ar); 6.95 (2H, d,  $^3J = 8.2$ , H Ar); 6.96 (2H, d,  $^3J = 7.9$ , H Ar); 7.00 (2H, d,  $^3J = 8.2$ , H Ar).

**8-Isopropyl-3,7-diphenyl-3,4,7,8-tetrahydro-2H,6H-pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitrile (1b).** Beige powder.  $^1\text{H}$  NMR spectrum (400 MHz),  $\delta$ , ppm ( $J$ , Hz): 1.01 (3H, d,  $^3J = 6.2$ ,  $\text{CH}_3$ ); 1.09 (3H, d,  $^3J = 6.2$ ,  $\text{CH}_3$ ); 1.94-1.96 (1H, m,  $\text{CH}(\text{CH}_3)_2$ ); 3.45 (1H, d,  $^3J = 9.1$ , H-8); 4.48 (1H, d,  $^2J = 12.9$ ) and 4.74 (1H, d,  $^2J = 12.9$ ,  $\text{NCH}_2\text{N}$ ); 4.82 (1H, d,  $^2J = 12.9$ ) and 5.07 (1H, d,  $^2J = 12.9$ ,  $\text{NCH}_2\text{N}$ ); 5.03 (1H, d,  $^2J = 12.6$ ) and 5.25 (1H, d,  $^2J = 12.6$ ,  $\text{SCH}_2\text{N}$ ); 6.83-7.26 (10H, m, H Ph).

**8-Isopropyl-3,7-di(4-methylphenyl)-3,4,7,8-tetrahydro-2H,6H-pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitrile (1c).** Colorless crystals.  $^1\text{H}$  NMR spectrum (400 MHz),  $\delta$ , ppm ( $J$ , Hz): 1.00 (3H, d,  $^3J = 6.3$ ,  $\text{CH}_3$ ); 1.07 (3H, d,  $^3J = 6.3$ ,  $\text{CH}_3$ ); 1.83-1.87 (1H, m,  $\text{CH}(\text{CH}_3)_2$ ); 2.25 (6H, br. s,  $2\text{ArCH}_3$ ); 3.30 (1H, d,  $^3J = 8.6$ , H-8); 4.40 (1H, d,  $^2J = 13.0$ ) and 4.56 (1H, d,  $^2J = 13.0$ ,  $\text{NCH}_2\text{N}$ ); 4.72 (1H, d,  $^2J = 13.5$ ) and 4.82 (1H, d,  $^2J = 13.5$ ,  $\text{NCH}_2\text{N}$ ); 4.92 (1H, d,  $^2J = 13.1$ ) and 5.16 (1H, d,  $^2J = 13.1$ ,  $\text{SCH}_2\text{N}$ ); 6.70-6.94 (8H, m, H Ar).

**3,7-Di(4-acetamidophenyl)-8-isopropyl-3,4,7,8-tetrahydro-2H,6H-pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitrile (1d).** Colorless amorphous powder.  $^1\text{H}$  NMR spectrum (500 MHz),  $\delta$ , ppm ( $J$ , Hz): 0.99 (3H, d,  $^3J = 6.2$ ,  $\text{CH}_3$ ); 1.06 (3H, d,  $^3J = 6.2$ ,  $\text{CH}_3$ ); 1.90-1.95 (1H, m,  $\text{CH}(\text{CH}_3)_2$ ); 2.02 (3H, s,  $\text{NHCOCH}_3$ ); 2.04 (3H, s,  $\text{NHCOCH}_3$ ); 3.42 (1H, br. d,  $^3J = 9.3$ , H-8); 4.47 (1H, d,  $^2J = 13.2$ ) and 4.67 (1H, d,  $^2J = 13.2$ ,  $\text{NCH}_2\text{N}$ ); 4.79 (1H, d,  $^2J = 12.7$ ) and 4.98 (1H, d,  $^2J = 12.7$ ,  $\text{NCH}_2\text{N}$ ); 5.09 (1H, d,  $^2J = 12.7$ ) and 5.25 (1H, d,  $^2J = 12.7$ ,  $\text{SCH}_2\text{N}$ ); 6.85 (2H, d,  $^3J = 8.8$ , H Ar); 7.10 (2H, d,  $^3J = 8.8$ , H Ar); 7.36 (2H, d,  $^3J = 8.8$ , H Ar); 7.47 (2H, d,  $^3J = 8.8$ , H Ar); 9.75 (1H, s,  $\text{NHAc}$ ); 9.82 (1H, s,  $\text{NHAc}$ ).

**8-Isopropyl-3,7-di(4-methoxyphenyl)-3,4,7,8-tetrahydro-2H,6H-pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitrile (1e).** White amorphous powder.  $^1\text{H}$  NMR spectrum (400 MHz),  $\delta$ , ppm ( $J$ , Hz): 1.00 (3H, d,  $^3J = 6.6$ ,  $\text{CH}_3$ ); 1.07 (3H, d,  $^3J = 6.6$ ,  $\text{CH}_3$ ); 1.84-1.88 (1H, m,  $\text{CH}(\text{CH}_3)_2$ ); 3.20 (1H, d,  $^3J = 9.4$ , H-8); 3.71 (3H, s,  $\text{OCH}_3$ ); 3.74 (3H, s,  $\text{OCH}_3$ ); 4.43 (1H, d,  $^2J = 13.2$ ) and 4.47 (1H, d,  $^2J = 13.2$ ,  $\text{NCH}_2\text{N}$ ); 4.74-4.76 (2H, m,  $\text{NCH}_2\text{N}$ ); 4.93 (1H, d,  $^2J = 12.6$ ) and 5.21 (1H, d,  $^2J = 12.6$ ,  $\text{SCH}_2\text{N}$ ); 6.68-6.98 (8H, m, H Ar).

**8-[2-(1,3-Benzodioxol-5-yl)-1-methylethyl]-3,7-di(4-methoxyphenyl)-3,4,7,8-tetrahydro-2H,6H-pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitrile (1f).** White amorphous powder.  $^1\text{H}$  NMR spectrum (500 MHz),  $\delta$ , ppm ( $J$ , Hz) (ratio of main and minor diastereomers 3:1): 0.81 (2.25H, d,  $^3J = 6.4$ ,  $\text{CHCH}_3$ ); 0.89 (0.75H, d,  $^3J = 6.8$ ,  $\text{CHCH}_3$ )\*; 1.97-2.05 (1H, m,  $\text{CHCH}_3$ )\*<sup>2</sup>; 2.18-2.26 (1H, m)\*<sup>2</sup>; 2.99 (0.75H, dd,  $^2J = 13.2$ ,  $^3J = 3.4$ ), and 3.05 (0.25H, dd,  $^2J = 13.1$ ,  $^3J = 3.0$ ,  $\text{ArCH}_2\text{CH}$ )\*; 3.41 (0.25H, d,  $^3J = 8.8$ , H-8)\*; 3.46 (0.75H, d,  $^3J = 8.3$ , H-8); 3.71 (2.25H, s,  $\text{OCH}_3$ ); 3.72 (0.75H, s,  $\text{OCH}_3$ )\*; 3.73 (0.75H, s,  $\text{OCH}_3$ )\*; 3.74 (2.25H, s,  $\text{OCH}_3$ ); 4.51 (0.75H, d,  $^2J = 13.2$ ) and 4.55 (0.75H, d,  $^2J = 13.2$ ,  $\text{NCH}_2\text{N}$ ); 4.63 (0.5H, dd,  $^2J = 13.2$ ,  $^2J = 13.2$ ,  $\text{NCH}_2\text{N}$ )\*; 4.79-4.88 (2H, m,  $\text{NCH}_2\text{N}$ )\*<sup>2</sup>; 5.00 (0.25H, d,  $^2J = 12.7$ )\* and 5.27 (0.25H, d,  $^2J = 12.7$ ,  $\text{NCH}_2\text{S}$ )\*; 5.04 (0.75H, d,  $^2J = 12.7$ ) and 5.29 (0.75H, d,  $^2J = 12.7$ ,  $\text{NCH}_2\text{S}$ ); 5.98 (1.5H, s,  $\text{OCH}_2\text{O}$ ); 5.99 (0.5H, s,  $\text{OCH}_2\text{O}$ )\*; 6.60-7.05 (11H, m, H Ar)\*<sup>2</sup>.  $^{13}\text{C}$  NMR spectrum,  $\delta$ , ppm: 16.0 ( $\text{CHCH}_3$ )\*; 16.2 ( $\text{CHCH}_3$ ); 38.8 ( $\text{CHCH}_3$ ); 39.0 ( $\text{CHCH}_3$ )\*<sup>2</sup>; 42.8 ( $\text{ArCH}_2$ ); 43.4 ( $\text{ArCH}_2$ )\*; 53.4 ( $\text{ArOCH}_3$ )\*; 55.7 ( $\text{ArOCH}_3$ ); 61.9 (C-8)\*; 62.3 (C-8); 64.0 ( $\text{NCH}_2\text{N}$ )\*; 61.1 ( $\text{NCH}_2\text{N}$ ); 64.5 ( $\text{NCH}_2\text{N}$ ); 64.6 ( $\text{NCH}_2\text{N}$ )\*; 73.6 ( $\text{SCH}_2\text{N}$ ); 73.7 ( $\text{SCH}_2\text{N}$ )\*; 101.1 ( $\text{OCH}_2\text{O}$ ); 108.5; 109.6 (C-9); 109.8 (C-9)\*; 114.6 ( $\text{C}\equiv\text{N}$ ); 114.7 ( $\text{C}\equiv\text{N}$ )\*; 119.6 (C Ar)\*; 119.9 (C Ar); 120.7 (C Ar)\*; 120.9 (C Ar); 121.5 (C Ar); 122.3 (C Ar); 122.5 (C Ar)\*; 134.9 (C Ar); 135.0 (C Ar)\*; 139.1 (C Ar); 143.9 (C Ar); 145.7 (C Ar); 151.7 (C-9a)\*; 151.9 (C-9a); 154.3 (C Ar)\*; 154.6 (C Ar)\*<sup>2</sup>.

**8-[2-(1,3-Benzodioxol-5-yl)-1-methylethyl]-3,7-di(4-methylphenyl)-3,4,7,8-tetrahydro-2H,6H-pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitrile (1g).** White amorphous powder.  $^1\text{H}$  NMR spectrum (400 MHz),  $\delta$ , ppm ( $J$ , Hz) (ratio of main and minor diastereomers 10:1): 0.86 (2.73H, d,  $^3J = 6.4$ ,  $\text{CHCH}_3$ ); 0.95 (0.27H, d,  $^3J = 6.7$ ,  $\text{CHCH}_3$ )\*; 1.93-2.03 (1H, m,  $\text{CHCH}_3$ )\*<sup>2</sup>; 2.18-2.24 (1H, m,  $\text{ArCH}_2\text{CH}$ )\*<sup>2</sup>; 2.28 (3H, s,  $\text{ArCH}_3$ )\*<sup>2</sup>; 2.30 (3H, s,  $\text{ArCH}_3$ )\*<sup>2</sup>; 2.96-3.04 (1H, m,  $\text{ArCH}_2\text{CH}$ )\*<sup>2</sup>; 3.47 (0.09H, d,  $^3J = 7.7$ , H-8)\*; 3.52 (0.91H, d,  $^3J = 8.1$ , H-8); 4.46 (1H, d,  $^2J = 12.9$ )\*<sup>2</sup> and 4.54 (1H, d,  $^2J = 12.9$ ,  $\text{NCH}_2\text{N}$ )\*<sup>2</sup>; 4.77-4.79 (2H, m,  $\text{NCH}_2\text{N}$ )\*<sup>2</sup>; 4.96 (1H, d,  $^2J = 12.6$ )\*<sup>2</sup> and 5.19 (1H, d,  $^2J = 12.6$ ,  $\text{NCH}_2\text{S}$ )\*<sup>2</sup>; 5.92 (0.18H, s,  $\text{OCH}_2\text{O}$ )\*; 5.93 (1.82H, s,  $\text{OCH}_2\text{O}$ ); 6.57

(1H, d,  $^3J = 7.6$ , H Ar)\*<sup>2</sup>; 6.64-6.68 (2H, m, H Ar)\*<sup>2</sup>; 6.76 (2H, d,  $^3J = 8.0$ , H Ar)\*<sup>2</sup>; 6.90-6.95 (4H, m, H Ar)\*<sup>2</sup>; 6.99 (2H, d,  $^3J = 8.4$ , H Ar)\*<sup>2</sup>.

**8-[2-(1,3-Benzodioxol-5-yl)-1-methylethyl]-3,7-dibenzyl-3,4,7,8-tetrahydro-2H,6H-pyrimido[4,3-*b*]-[1,3,5]thiadiazine-9-carbonitrile (1h).** Yellow tar, glassifies on prolonged standing. According to HPLC-MS and <sup>1</sup>H NMR data the content of the main material is about 73%. <sup>1</sup>H NMR spectrum (400 MHz),  $\delta$ , ppm (*J*, Hz) (ratio of main and minor diastereomers 10:9): 0.72 (1.41H, d,  $^3J = 6.3$ , CHCH<sub>3</sub>)\*; 0.81 (1.59H, d,  $^3J = 6.8$ , CHCH<sub>3</sub>); 1.90-2.12 (2H, m, CHCH<sub>3</sub>, ArCHHCH)\*<sup>2</sup>; 2.93-2.97 (0.47H, m, ArCHHCH)\*; 3.08-3.12 (0.53H, m, ArCHHCH); 3.33-3.39 (4H, m, 2CH<sub>2</sub>Ph)\*<sup>2</sup>; 3.43-4.87 (7H, m, H-8, 2NCH<sub>2</sub>N, NCH<sub>2</sub>S)\*<sup>2</sup>; 5.92-5.97 (2H, m, OCH<sub>2</sub>O)\*<sup>2</sup>; 6.53-7.42 (13H, m, H Ar)\*<sup>2</sup>. Mass spectrum (ESI), *m/z* (*I*<sub>rel</sub>, %): 525 [M+H]<sup>+</sup> (37), 406 [M+H-PhCH<sub>2</sub>N=CH<sub>2</sub>]<sup>+</sup> (100), 120 [PhCH<sub>2</sub>NH=CH<sub>2</sub>]<sup>+</sup> (16).

**3,7-Di(3-methylphenyl)-8-(4-methylphenyl)-3,4,7,8-tetrahydro-2H,6H-pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitrile (1i).** Pale-yellow amorphous powder. <sup>1</sup>H NMR spectrum (400 MHz),  $\delta$ , ppm (*J*, Hz): 2.30 (6H, br. s, 2 CH<sub>3</sub>); 2.36 (3H, s, CH<sub>3</sub>); 4.14 (1H, d,  $^2J = 13.0$ ) and 4.63 (1H, d,  $^2J = 13.0$ , NCH<sub>2</sub>N); 4.81 (1H, d,  $^2J = 13.0$ ) and 4.97 (1H, d,  $^2J = 13.0$ , NCH<sub>2</sub>N); 5.13 (1H, s, H-8); 5.08 (1H, d,  $^2J = 12.9$ ) and 5.27 (1H, d,  $^2J = 12.9$ , SCH<sub>2</sub>N); 6.69-7.28 (12H, m, H Ar).

**3,7-Dibenzyl-8-(4-methylphenyl)-3,4,7,8-tetrahydro-2H,6H-pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitrile (1j).** Beige amorphous powder. <sup>1</sup>H NMR spectrum (400 MHz),  $\delta$ , ppm (*J*, Hz): 2.30 (3H, s, CH<sub>3</sub>); 3.49 (1H, d,  $^2J = 12.3$ ) and 3.70 (1H, d,  $^2J = 12.3$ , NCH<sub>2</sub>N); 3.90 (2H, br. s, PhCH<sub>2</sub>); 4.00 (1H, d,  $^2J = 12.4$ ) and 4.11 (1H, d,  $^2J = 12.4$ , NCH<sub>2</sub>N); 4.19 (2H, br. s, PhCH<sub>2</sub>); 4.23 (1H, s, H-8); 4.51 (1H, d,  $^2J = 12.1$ ) and 4.74 (1H, d,  $^2J = 12.1$ , SCH<sub>2</sub>N); 7.09-7.45 (14H, m, H Ar).

**3,7,8-Tri(4-methylphenyl)-3,4,7,8-tetrahydro-2H,6H-pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitrile (1k).** White finely crystalline powder. <sup>1</sup>H NMR spectrum (400 MHz),  $\delta$ , ppm (*J*, Hz): 2.25 (3H, s, CH<sub>3</sub>); 2.28 (3H, s, CH<sub>3</sub>); 2.30 (3H, s, CH<sub>3</sub>); 4.28 (1H, d,  $^2J = 13.0$ ) and 4.53 (1H, d,  $^2J = 13.0$ , NCH<sub>2</sub>N); 4.87 (1H, d,  $^2J = 12.7$ ) and 4.94 (1H, d,  $^2J = 12.7$ , NCH<sub>2</sub>N); 5.12 (1H, s, H-8); 5.10 (1H, d,  $^2J = 12.5$ ) and 5.32 (1H, d,  $^2J = 12.5$ , SCH<sub>2</sub>N); 7.00-7.20 (12H, m, H Ar).

**8-(2-Chlorophenyl)-3,7-di(4-methylphenyl)-3,4,7,8-tetrahydro-2H,6H-pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitrile (1l).** White finely crystalline powder. <sup>1</sup>H NMR spectrum (400 MHz),  $\delta$ , ppm (*J*, Hz): 2.27 (3H, s, CH<sub>3</sub>); 2.31 (3H, s, CH<sub>3</sub>); 4.20 (1H, d,  $^2J = 13.0$ ) and 4.49 (1H, d,  $^2J = 13.0$ , NCH<sub>2</sub>N); 4.68 (1H, d,  $^2J = 13.1$ ) and 4.94 (1H, d,  $^2J = 13.1$ , NCH<sub>2</sub>N); 5.12 (1H, d,  $^2J = 12.6$ ) and 5.23 (1H, d,  $^2J = 12.6$ , SCH<sub>2</sub>N); 5.28 (1H, s, H-8); 6.80-7.20 (8H, m, H Ar); 7.25-7.45 (4H, m, 2-ClC<sub>6</sub>H<sub>4</sub>).

**8-(4-Dimethylaminophenyl)-3,7-di(4-methylphenyl)-3,4,7,8-tetrahydro-2H,6H-pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitrile (1m).** Bright-yellow crystals. <sup>1</sup>H NMR spectrum (400 MHz),  $\delta$ , ppm (*J*, Hz): 2.31 (3H, s, ArCH<sub>3</sub>); 2.32 (3H, s, ArCH<sub>3</sub>); 2.95 (6H, s, (CH<sub>3</sub>)<sub>2</sub>N); 4.15 (1H, d,  $^2J = 12.6$ ) and 4.46 (1H, d,  $^2J = 12.6$ , NCH<sub>2</sub>N); 4.77-4.78 (2H, m, NCH<sub>2</sub>N); 4.94 (1H, s, H-8); 5.01 (1H, d,  $^2J = 12.5$ ) and 5.21 (1H, d,  $^2J = 12.5$ , SCH<sub>2</sub>N); 6.64 (2H, d,  $^3J = 8.1$ , H Ar); 6.90-7.02 (8H, m, 2MeC<sub>6</sub>H<sub>4</sub>); 7.09 (2H, d,  $^3J = 8.1$ , H Ar).

**3,7-Di(4-methylphenyl)-8-(4-nitrophenyl)-3,4,7,8-tetrahydro-2H,6H-pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitrile (1n).** Bright-yellow crystals. <sup>1</sup>H NMR spectrum (500 MHz),  $\delta$ , ppm (*J*, Hz): 2.27 (3H, s, CH<sub>3</sub>); 2.28 (3H, s, CH<sub>3</sub>); 4.03 (1H, d,  $^2J = 13.2$ ) and 4.68 (1H, d,  $^2J = 13.2$ , NCH<sub>2</sub>N); 4.89 (1H, d,  $^2J = 12.7$ ) and 4.95 (1H, d,  $^2J = 12.7$ , NCH<sub>2</sub>N); 5.15 (1H, d,  $^2J = 12.2$ ) and 5.31 (1H, d,  $^2J = 12.2$ , SCH<sub>2</sub>N); 5.37 (1H, s, H-8); 7.00-7.05 (8H, m, 2MeC<sub>6</sub>H<sub>4</sub>); 7.60 (2H, d,  $^3J = 8.1$ , H Ar); 8.25 (2H, d,  $^3J = 8.1$ , H Ar).

**3,7-Di(4-methylphenyl)-8-(3-nitrophenyl)-3,4,7,8-tetrahydro-2H,6H-pyrimido[4,3-*b*][1,3,5]thiadiazine-9-carbonitrile (1o).** Bright-yellow crystals. <sup>1</sup>H NMR spectrum (400 MHz),  $\delta$ , ppm (*J*, Hz): 2.26 (3H, s, CH<sub>3</sub>); 2.27 (3H, s, CH<sub>3</sub>); 4.05 (1H, d,  $^2J = 12.9$ ) and 4.68 (1H, d,  $^2J = 12.9$ , NCH<sub>2</sub>N); 4.87 (1H, d,  $^2J = 13.3$ ) and 4.94 (1H, d,  $^2J = 13.3$ , NCH<sub>2</sub>N); 5.14 (1H, d,  $^2J = 12.5$ ) and 5.32 (1H, d,  $^2J = 12.5$ , SCH<sub>2</sub>N); 5.37 (1H, s, H-8);

\* Signals of the minor diastereomers.

\*<sup>2</sup> Overlapping of the signals of the main and minor diastereomers.

6.97 (2H, d,  $^3J = 7.9$ , H Ar); 7.03 (2H, d,  $^3J = 7.9$ , H Ar); 7.04-7.08 (4H, m, H Ar); 7.71 (1H, dd,  $^3J = 7.5$ ,  $^3J = 7.9$ , H-5'); 7.80 (1H, d,  $^3J = 7.5$ , H-6'); 8.18 (1H, s, H-2'); 8.22 (2H, d,  $^3J = 7.9$ , H-4').

**3,7-Di(4-methylphenyl)-3,4,7,8-tetrahydro-2H,6H-pyrimido[4,3-b][1,3,5]thiadiazine-9-carbonitrile (1p).** A. Obtained by condensation of acetaldehyde, thioamide **5**, *p*-toluidine, and HCHO. A mixture of cyanothioacetamide (**5**) (0.5 g, 5.0 mmol), acetaldehyde (0.5 ml, 8.9 mmol) and *N*-methylmorpholine (3 drops) was stirred in EtOH (15-20 ml) for 30 min at 20°C. *p*-Toluidine (1.1 g, 10.3 mmol) and 37% formalin free from paraformaldehyde (3.0 ml, 40.0 mmol) were added. The mixture was refluxed for 2-3 min and then stirred at 20°C. The precipitate formed after 24 h was filtered off, washed with EtOH, and immediately recrystallized from a mixture of acetone and DMF (1:1). Compound **1p** (200 mg, 11%) was obtained as a light-yellow, finely crystalline powder.

B. Obtained from thioamide **5**, formalin, and toluidine. A mixture of the cyanothioacetamide (**5**) (1.00 g, 10.0 mmol), *p*-toluidine (1.38 g, 12.9 mmol), and 37% formalin (3.0 ml, 40.0 mmol) was refluxed for 2-3 min in EtOH (30 ml) with stirring. After 24 h the precipitate was filtered off and recrystallized from a mixture of ethanol and DMF (1:1). Yield 480 mg (13%).  $^1\text{H}$  NMR spectrum (400 MHz),  $\delta$ , ppm ( $J$ , Hz): 2.25 (3H, s,  $\text{CH}_3$ ); 2.27 (3H, s,  $\text{CH}_3$ ); 3.96 (2H, br. s, 8- $\text{CH}_2$ ); 4.65 (2H, br. s) and 4.78 (2H, br. s, 4,6- $\text{CH}_2$ ); 5.08 (2H, br. s,  $\text{SCH}_2\text{N}$ ); 6.87-6.91 (4H, m, 2H-3',5'); 6.98 (4H, d,  $^3J = 7.6$ , 2H-2',6'). Mass spectrum (ESI),  $m/z$  ( $I_{\text{rel}}$ , %): 364  $[\text{M}+\text{H}]^+$  (100), 245  $[\text{M}+\text{H}-\text{ArN}=\text{CH}_2]^+$  (76), 120  $[\text{ArNH}=\text{CH}_2]^+$  (81).

**1,3,5-Tri(4-methylphenyl)-1,3,5-perhydrotriazine (7).** 3-(2-Chlorophenyl)-2-cyanoacrylamide (**6a**) (0.564 g, 2.73 mmol), *p*-toluidine (0.658 g, 6.14 mmol), and 37% HCHO (2.5 ml, 33.30 mmol) were refluxed in DMF (4 ml) for 3-4 min and then held for 72 h at 20°C. Yield 205 mg (28%). Colorless needle like crystals; mp 125°C (mp 127.9°C [25], 126-127°C [27]),  $R_f = 0.81$  (acetone-hexane, 1:1).  $^1\text{H}$  NMR spectrum (500 MHz),  $\delta$ , ppm ( $J$ , Hz): 2.21 (9H, s, 3 $\text{CH}_3$ ); 4.77 (6H, br. s, 3 $\text{CH}_2$ ); 7.05 (12H, dd,  $^3J = 8.1$ ,  $^3J = 8.1$ , H Ar). Found, %: C 80.56; H 7.65; N 11.79.  $\text{C}_{24}\text{H}_{27}\text{N}_3$ . Calculated, %: C 80.63; H 7.61; N 11.75.

**Multicomponent Condensation of Furfural, Cyanoacetamide, Formalin, and *p*-Toluidine.** A mixture of cyanoacetamide (0.5 g, 6.0 mmol), furfural (0.5 ml, 6.0 mmol), and morpholine (1 drop) was stirred in EtOH (20-25 ml) for 30 min at 20°C. *p*-Toluidine (0.71 g, 6.6 mmol) and 37% formalin free from an admixture of paraformaldehyde (2.0 ml, 26.6 mmol) were added. The mixture was refluxed for 2-3 min, filtered through a plaited filter, and held for 48 h at 20°C. The precipitate formed was filtered off and washed with EtOH to give the sandy-gray product (0.34 g). According to  $^1\text{H}$  NMR data (400 MHz) this product is the (*E*)-2-cyano-3-(2-furyl)prop-2-enamide (**6b**) (35% yield) with an admixture of about 2% of the perhydrotriazine **7**.  $^1\text{H}$  NMR spectrum (400 MHz),  $\delta$ , ppm ( $J$ , Hz): 6.70 (1H, dd,  $^3J = 3.4$ ,  $^3J = 1.6$ , H-4 furyl); 7.33 (1H, br. d,  $^3J = 3.4$ , H-3 furyl); 7.47 (1H, br. s) and 7.56 (1H, br. s,  $\text{CONH}_2$ ); 7.92 (1H, d,  $^3J = 1.6$ , H-5 furyl); 7.96 (1H, s,  $-\text{CH}=\text{}$ ). Proton signals of the perhydrotriazine **7**: 2.23 (9H, s, 3 $\text{CH}_3$ ); 4.72 (6H, br. s, 3 $\text{CH}_2$ ); 6.90 (12H, dd,  $^3J = 8.1$ ,  $^3J = 8.1$ , H Ar).

## REFERENCES

1. V. V. Dotsenko, S. G. Krivokolysko, and V. P. Litvinov, *Izv. Akad. Nauk, Ser. Khim.*, 134 (2012).
2. C. J. Moody in: A. R. Katritzky and C. W. Rees (editors), *Comprehensive Heterocyclic Chemistry*, Vol. 3, Pergamon Press, Oxford (1984), p. 1039.
3. R. K. Smalley in: A. J. Boulton (editor), *Comprehensive Heterocyclic Chemistry II*, Vol. 6, Elsevier, Oxford (1996), p. 783.
4. N. Shobana and P. Farid, in: A. R. Katritzky, C. A. Ramsden, E. F. V. Scriven, and R. J. K. Taylor (editors), *Comprehensive Heterocyclic Chemistry III*, Vol. 9, Elsevier, Oxford (2008), p. 457.
5. V. V. Dotsenko, S. G. Krivokolysko, and V. P. Litvinov, *Monatsh. Chem.*, **139**, 657 (2008).
6. Z. Y. Wang, H. X. Shi, and H. J. Shi, *Synth. Commun.*, **31**, 2841 (2001).
7. Z. Y. Wang, T. P. You, H. J. Shi, and H. X. Shi, *Molecules*, **1**, 89 (1996).

8. Z. Y. Wang, T. P. You, H. J. Shi, and H. X. Shi, *Gaodeng Xuexiao Huaxue Xuebao*, **18**, 550 (1997); *Chem. Abstr.*, **127**, 95265t (1997).
9. H. X. Shi, H. J. Shi, and Z. Y. Wang, *Youji Huaxue*, **20**, 344 (2000); *Chem. Abstr.*, **133**, 120280c (2000).
10. Z. A. Hozein, A. A. O. Sarhan, H. A. H. El-Sherief, and A. M. Mahmoud, *Z. Naturforsch., B: J. Chem. Sci.*, **52**, 1401 (1997); *Chem. Abstr.*, **128**, 88906v (1998).
11. H. J. Shi, H. X. Shi, and Z. Y. Wang, *J. Heterocycl. Chem.*, **38**, 929 (2001).
12. H. J. Shi, Z. Y. Wang, and H. X. Shi, *Chimia*, **51**, 529 (1997); *Ref. Zh. Khim.*, 13Zh272 (1998).
13. H. J. Shi, Z. Y. Wang, and H. X. Shi, *Synth. Commun.*, **29**, 2027 (1999).
14. T. B. Sarfraz, S. A. Husain, N. Murtaza, and B. S. Siddiqui, *Pak. J. Sci. Ind. Res.*, **43**, 334 (2000).
15. Z. A. Hozein, *J. Chem. Res. Synop.*, 99 (2000).
16. K. A. Frolov, V. V. Dotsenko, S. G. Krivokolysko, and V. P. Litvinov, *Izv. Akad. Nauk, Ser. Khim.*, 2158 (2005).
17. L. D. S. Yadav, A. Vaish, and S. Sharma, *J. Agric. Food Chem.*, **42**, 811 (1994).
18. A. A. O. Sarhan, S. H. Abdel-Hafez, H. El-Sherief, and T. Aboel-Fadl, *Synth. Commun.*, **36**, 987 (2006).
19. V. V. Dotsenko, S. G. Krivokolysko, and V. P. Litvinov, *Monatsh. Chem.*, **139**, 271 (2008).
20. V. V. Dotsenko, S. G. Krivokolysko, E. B. Rusanov, and V. P. Litvinov, *Izv. Akad. Nauk, Ser. Khim.*, 1384 (2007).
21. V. V. Dotsenko, S. G. Krivokolysko, A. N. Chernega, and V. P. Litvinov, *Dokl. Akad. Nauk.*, **389**, 763 (2003).
22. V. V. Dotsenko, K. A. Frolov, S. G. Krivokolysko, A. N. Chernega, and V. P. Litvinov, *Monatsh. Chem.*, **137**, 1089 (2006).
23. V. V. Dotsenko, S. G. Krivokolysko, and V. P. Litvinov, *Izv. Akad. Nauk, Ser. Khim.*, 1420 (2007).
24. V. V. Dotsenko, K. A. Frolov, S. G. Krivokolysko, and V. P. Litvinov, *Izv. Akad. Nauk, Ser. Khim.*, 1436 (2009).
25. J. G. Miller and E. C. Wagner, *J. Am. Chem. Soc.*, **54**, 3698 (1932).
26. J. S. A. Brunskill, A. De, and D. F. Ewing, *J. Chem. Soc., Perkin Trans. I*, 629 (1978).
27. A. Rivera, O. L. Torres, J. D. Leitón, M. S. Morales-Ríos, and P. Joseph-Nathan, *Synth. Commun.*, **32**, 1407 (2002).