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moisture) with benzaldehydes (5). The structure of the ethylene derivatives 6 thus obtained were established by microanalyses, physical, and spectral data.

The arsorane 4b seems to be more reactive towards benzaldehydes than the phosphorane 4a as can be seen from the better yields of 6 obtained using 4b. This higher reactivity of 4b may be attributed to the lower d_{π} - p_{π} resonance of 4b as compared to 4a.

When the intermediates 4a, b are generated using sodium methoxide in methanol they do not undergo the reaction (cf. Ref. 5). However, when two equivalents of 4-methylbenz-aldehyde are added to the solution containing 4a or 4b and the mixture is stirred for 3-4h, 4-methyl-4'-(4-methyl-cinnamoyl)-stilbene (7) is obtained as a thick yellow precipitate by simultaneous aldol condensation and carbonyl ole-fination.

4-Acetylbenzyltriphenylphosphonium Bromide (3a):

A solution of triphenylphosphine (2a; 5.24 g, 0.02 mol) in benzene (30 ml) is added dropwise, with stirring, to a solution of 4-bromomethylacetophenone (1; 4.26 g, 0.02 mol) in benzene (30 ml) under a nitrogen atmosphere. The resultant mixture is refluxed for 4h and then allowed to stand overnight. The salt 3a is collected by filtration, dried, and recrystallized twice from chloroform/hexane; yield: 7.4 g (78 %); colorless crystals, m.p. 247-249°.

C₂₇H₂₄BrOP calc. C 68.21 H 5.05 (475.4) found 68.24 5.07

I.R. (KBr): $v_{\text{max}} = 1725 \,\text{cm}^{-1}$ (C=O).

¹H-N.M.R. (CDCl₃): δ = 2.64 (s, 3H, CO-CH₃); 5.75 (d, 2H, P-CH₂, J_{PH} = 15 Hz); 7.30-8.20 ppm (m, 19 H_{arom}).

4-Acetylbenzyltriphenylarsonium Bromide (3b):

A solution of triphenylarsine (2b; 12.2 g, 0.04 mol) and 4-bromomethylacetophenone (1; 8.52 g, 0.04 mol) in benzene (60 ml) is refluxed for 3 days. Excess solvent is then evaporated on a steam bath and ethyl acetate (30 ml) is added to the residue (\sim 15 ml). The precipitated salt 3b is isolated by filtration and recrystallized twice from chloroform/benzene to give a microcrystalline product; yield: 12.6 g (60 %); m.p. 167-168°.

C₂₇H₂₄AsBrO calc. C 62.42 H 6.42 (519.3) found 62.48 6.37

1.R. (KBr): $v_{\text{max}} = 1725 \,\text{cm}^{-1}$ (C=O).

¹H-N.M.R. (CDCl₃): δ = 2.35 (s, 3H, CO-CH₃); 6.30 (s, 2H, As-CH₂); 7.13-8.42 ppm (m, 19 H_{arom}).

Preparation of 4-Acetylstilbenes (6); General Procedure:

A suspension of phosphorane 4a or arsorane 4b is prepared by adding sodium hydride (96 mg, 4 mmol) to the phosphonium salt 3a (1.9 g, 4 mmol) or arsonium salt 3b (2.08 g, 4 mmol), respectively, in benzene (100 ml) under a nitrogen atmosphere. To this suspension, the benzaldehyde 5 (4 mmol) is added and the mixture is stirred for 6-8 h at room temperature. The solid product is filtered off and the filtrate is evaporated on a steam bath. The

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The literature¹ reveals that Friedel-Crafts acetylation of stilbene gives 4,4'-diacetylstilbene; in no case was a monoacetylated stilbene obtained. The synthesis of substituted monoacetylstilbenes could not be achieved by other conventional methods adopted for the preparation of stilbenes. Prompted by these reports and following our interest in the reactivity alkylidenearsoranes^{2, 3, 4} we wish to report a convenient route for the synthesis of substituted 4-acetylstilbenes by two different routes using 4-acetylbenzylidenetriphenylphosphoranes and the corresponding arsoranes.

Our method consists of the reaction of 4-bromomethylacetophenone (1) with an equimolar amount of triphenylphosphine (2a) or triphenylarsine (2b) in boiling benzene, conversion of the resultant phosphonium (3a) or arsonium salt (3b) to the phosphorane (4a) or arsorane (4b) by treatment with base, and reaction of 4a, b (which cannot be isolated because of their instability and sensitivity towards air and

$$H_3C-C$$
 $CH=M(C_6H_5)_3$
 A_3C-C
 $CH=CH-C$
 $CH=C$
 C

Table, 4-Acetylstilbenes (6) prepared

X	prep- ared from	Yield [%]	Recrystal- lization solvent	m.p.ª	Molecular formula ^b	I.R. (KBr) $v_{\text{max}} \left[\text{cm}^{-1} \right]$	¹ H-N.M.R. (CDCl ₃) δ [ppm]
Н	3a	42	benzene/hexane	160–161°	C ₁₆ H ₁₄ O	1661°; 1600°;	
	3b	56	benzene/hexane	158-159°	(222.3)	938e	
4-NO ₂	3a	70	benzene/hexane	187-189°	$C_{16}H_{13}NO_{3}$	1681°; 1613 ^d ;	2.63 (s, 3H, COCH ₃);
	3b	85	benzene/hexane	188-190°	(267.3)	960°	7.16-7.36 (q, 2H _{otefin}); 7.53-8.33 (m, 8H _{arom})
4-N(CH ₃) ₂	3a	58	benzene/PE	228-230°	$C_{18}H_{19}NO$	1676°; 1608 ^d ;	2.60 (s, 3H, CO-CH ₃);
	3 b	70	benzene/PE	231–233°	(265.3)	950°	3.05 [s, 6H, N(CH ₃) ₂]; 6.75-7.20 (q, 2H _{olefin}); 7.32-8.15 (m, 8H _{arom})
3,4-O-CH ₂ -O-	3a	54	benzene/hexane	153-154°	$C_{17}H_{14}O_3$	1672°; 1605°;	2.60 (s, 3H, CO-CH ₃);
	3 b	76	benzene/hexane	154-156°	(266.3)	955°	6.02 (s, 2 H, O—CH ₂ —O); 6.85-7.28 (q, 2 H _{olefin}); 7.38-8.20 (m, 7 H _{arom})

^a Melting points were determined using a Gallenkamp apparatus and are uncorrected.

residue is chromatographed over neutral alumina using benzene/petroleum ether $(40-60^\circ)$ (1:1) as eluent. The product is recrystallized from the solvent given in the Table.

4-Methyl-4'-(4-methylcinnamoyl)-stilbene (7):

A solution of phosphorane 4a or arsorane 4b is prepared by adding sodium methoxide (4 mmol) to a solution of the phosphonium salt 3a (0.95 g, 2 mmol) or arsonium salt 3b (1.04 g, 2 mmol), respectively, in methanol (50 ml) under a nitrogen atmosphere. To this solution, 4-methylbenzaldehyde (0.48 g, 4 mmol) is added and the mixture is stirred for 2h at room temperature. The resultant yellow solid is isolated by filtration, washed with water, dried, and recrystallized from chloroform/hexane (1:1); yield from 3a: 0.34 g (50%); m.p. $221-223^{\circ}$; yield from 3b: 0.40 g (60%); m.p. $218-220^{\circ}$.

C₂₅H₂₂O calc. C 88.75 H 6.50 (338.4) found 88.69 6.43

M.S.: m/e = 338 (M⁺, 12%), 337 (75), 336 (100), 323 (85), 308 (25), 221 (54), 177 (58), 161 (66), 145 (54), 117 (40), 105 (20), 91 (27).

I.R. (KBr): $v_{\text{max}} = 1667 \text{ cm}^{-1}$ (C=O).

U.V. (CHCl₃): $\lambda_{max} = 356 \text{ nm}$.

¹H-N.M.R. (CDCl₃): $\delta = 2.05$ (s, 6H, 2CH₃); 6.66–7.15 (m, 4H_{olefin}); 7.20–7.71 ppm (m, 12H_{arom}).

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^b The microanalyses were in good accord with the calculated values: C, ± 0.07 ; H, ± 0.07 .

[°] C=0.

 $^{^{}d}$ C=C.

^e Out-of-plane deformations of the H-atom of the trans-olefinic system.

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