A Nonmetal Approach to α-Heterofunctionalized Carbonyl Derivatives by Formal Reductive X–H Insertion**

Eric J. Miller, Wei Zhao, Jonathan D. Herr, and Alexander T. Radosevich*

Synthetic methods leading to the formation of α -heterofunctionalized carbonyl derivatives are valuable for the preparation of important molecular targets.^[1] Among the varied synthetic approaches to these compounds,^[2] the direct functionalization of an X-H bond by insertion of a carbene equivalent has proven expedient and versatile.^[3-5] In practice, the required carbene synthons may be accessed by metalcatalyzed decomposition of α-diazo compounds.^[6] Despite the utility of this approach, its appeal is tempered by the fact that only a modest number of diazo compounds are commercially available, a circumstance that is due in large part to hazards associated with preparation, isolation, and storage of these compounds.^[7] Consequently, alternative strategies for the direct X-H functionalization transformation that retain the power and simplicity of the diazo decomposition approach, but broaden access to starting materials, would be desirable.

In considering alternative carbene synthetic equivalents,^[8] we recognized that α -diazo carbonyl compounds are routinely prepared from the corresponding α -keto carbonyl derivative.^[9] Consequently, we considered the possibility that this class of readily accessible and bench-stable compounds might be utilized directly en route to carbene-like reactivity, thereby obviating the intermediacy of unstable diazo compounds. We envisioned that exposure of α -keto ester 3 to a suitable oxygen-atom acceptor would furnish a dipolar structure of the type 4, which by analogy to α -diazo ester resonance contributor 2 might serve as a synthetic equivalent of a carbenoid.^[10] In a practical sense, intermediates of the type 4 may be accessed by the well-known Kukhtin-Ramirez redox condensation of 1.2-dicarbonyl compounds with trivalent phosphorus derivatives ($E = PR_3$; Scheme 1).^[11] Indeed, the intermediacy of these adducts in various epoxide- and cyclopropane-forming procedures^[12,13] lends support for their formal carbene equivalency, and isolated evidence suggests a further potential for X-H functionalization reactivity.^[14] Herein, we realize this potential in an operationally simple, metal-free method for the preparation of α -heterofunctionalized carbonyl derivatives by direct reductive X-H functionalization employing readily available and bench-stable coupling components.

Supporting information for this article is available on the WWW under http://dx.doi.org/10.1002/anie.201205604.

 $\begin{array}{c} R \\ \searrow \\ N_{2} \\ 1 \\ \alpha \text{-Diazo Ester} \\ \downarrow \\ \alpha \text{-Diazo Ester} \\ 0 \\ \alpha \text{-Diazo Ester} \\ 0 \\ \alpha \text{-Diazo Ester} \\ 0 \\ \alpha \text{-} \\$

Scheme 1. Synthetic routes to access carbene synthons.

An investigation was initiated using the direct reductive coupling of methyl benzoylformate (5) and p-cresol (6) mediated by oxygen-atom acceptors as a demonstration reaction [Eq. (1)]. In accord with the hypothesis outlined



above, we have found the addition of tris(dimethylamino)phosphorus to a solution of **5** and **6** results in direct reductive O–H functionalization, leading to the formation of the α phenoxy ester **7** in good yield.^[15] Neither phosphines (PPh₃) nor phosphites (P(OMe)₃) are similarly reactive under otherwise identical conditions,^[16] an observation that may be attributed to inferior P=O bond enthalpies.^[17]

In terms of scope, the coupling method tolerates a range of electronically and sterically diverse α -keto ester substrates (Table 1). Both electron-rich (9) and electron-deficient (10) benzoylformate derivatives serve as viable substrates for coupling, and the series of *p*-halogenated *O*-aryl mandelate derivatives (11–13) may be prepared. Aryl substitution of the ketone is not strictly required for efficient reaction; alkyl-substituted keto ester substrates yield the expected corresponding α -phenoxy ester product in good yield. For example, methyl (14), primary alkyl (15), and secondary alkyl substituents (16) are all well-tolerated.

Angew. Chem. Int. Ed. 2012, 51, 1-6

© 2012 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

🕏 WILEY 盾

^[*] E. J. Miller, W. Zhao, J. D. Herr, Prof. Dr. A. T. Radosevich Department of Chemistry, Pennsylvania State University University Park, PA 16802 (USA) E-mail: radosevich@psu.edu

^[**] We thank the Pennsylvania State University for financial support of this research.



Table 1: Scope of $\alpha\text{-keto}$ ester substrates for reductive O–H functionalization. $^{[a]}$



[a] Yields of isolated product.

With respect to the protic oxygen coupling partner, the method also proves to be quite versatile (Table 2). The suite of cresol regioisomers all react smoothly to provide the corresponding α -aryloxy esters in uniformly high yields (17–19). Modulation of the electronic character of the phenol derivative has only a modest influence on the reaction outcome; the reactivity of electron-rich phenols (4-methoxy-phenol) is exemplary (20), whereas yields are modestly diminished by electron-withdrawing substitution (22). The method is not limited to phenolic O–H functionalization;

Table 2: Scope of nucleophiles for reductive O-H functionalization.[a]





www.angewandte.org

2 **rr** © 2012 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

Angew. Chem. Int. Ed. 2012, 51, 1-6

These are not the final page numbers!

indeed, carboxylic acids undergo smooth coupling to give α acetoxylated esters (**23**, **24**) in good yields. The method can also be extended to aliphatic alcohols, although the reaction with these nucleophiles appears somewhat less general. For instance, propargyl alcohol and trifluoroethanol both undergo functionalization under standard conditions to give the anticipated products **25** and **26**, respectively. However, only trace amounts compound **27** can be observed when ethanol is employed as a coupling partner.^[18,19] This observed reactivity for aliphatic alcohols correlates with pK_{a} ,^[20] suggesting a degree of O–H bond heterolysis in the coupling event (see below).

The coupling method can be successfully applied to the direct X–H functionalization of other heteroatomic nucleophiles (Table 3). Sulfonyl-protected amine derivatives



[a] Yields of isolated product.

undergo smooth coupling to give protected phenylglycine (28) and alanine (29) derivatives in good yield. This method thus permits the direct synthesis of diverse *N*-substituted α -amino ester products by reaction of *N*-aryl (30) and *N*-alkyl (31–33) sulfonamides. Protic *N*-heterocyclic compounds can also be employed in the coupling; for example, phthalimide (34), imidazole (35), and pyrazole (36) all lead to the coupling products.

Mechanistically, we believe the reductive X–H functionalization transformation is initiated by Kukhtin–Ramirez condensation of tris(dimethylamino)phosphorus with the α keto ester substrate (Scheme 2).^[11] Addition of the aminophosphine reagent to the carbonyl oxygen of the keto ester substrate could give a 1,3-dipolar intermediate (**37**), which would experience stabilization as oxyphosphonium enolate resonance structure **37**' or dioxaphospholene valence isomer **38**; features governing partitioning of the type **37–38** have been investigated.^[21] Subsequent X–H functionalization could then evolve along two distinct mechanistic pathways. Loss of tris(dimethylamino)phosphine oxide could reveal



Scheme 2. Possible mechanisms for X-H functionalization.

a free carbene intermediate (**39**) in solution (Scheme 2, pathway A),^[21] the capture of which by X–H bond insertion would lead to product **41**. Alternatively, adducts **37–38** may be intercepted by proton transfer from the pronucleophile (X–H) to give alkoxyphosphonium intermediate **40** (pathway B), which could then undergo nucleophilic displacement to furnish product.^[22–24]

In principle, the foregoing mechanistic possibilities may be differentiated by stereochemical experiments employing homochiral phosphorus reagents. In this analysis, use of a homochiral phosphorus reagent would desymmetrize key intermediates **37–38**. In the event a stepwise polar mechanism were operative, facially selective protonation to give stereodefined alkoxyphosphonium **40** could be envisioned, and subsequent stereospecific displacement of the phosphoric triamide leaving group would yield stereodefined product **41**. By contrast, if a free carbene were generated in accord with pathway A, a stereoselective X–H functionalization would not be expected.

As depicted in Equation (2), we have found that reductive coupling of methyl benzoylformate and benzoic acid deriv-



atives employing homochiral diazaphospholidine (R,R)-**42**^[25] is indeed stereoselective. The reductive coupling gives *O*-acyl mandelate derivative (*S*)-**43** with good yield and excellent stereoselectivity.^[26]

In light of the observed stereoselectivity, we regard as remote the possibility that genuine carbene intermediates are formed (pathway A) under our conditions; instead, carbenoid reactivity evinced by a stepwise, polar mechanism (pathway B) would appear to be most consistent with the observed outcome. This mechanism, in which proton transfer precedes α -C–X bond formation, distinguishes our metal-free method from X–H bond functionalization by metal carbenoids, where proton transfer follows α -C–X bond formation.^[27] The complementarity in electronic character implied by this mechanistic distinction (species **37–38** are nucleophilic, whereas metal carbenoids are electrophilic^[28]), suggests the potential for differential application of these two classes of reactive intermediates (Scheme 3). Specifically, N–H func-



Scheme 3. Intermolecular N–H functionalization versus intramolecular cyclopropanation.

tionalization prevails when a solution of allyl benzoylformate (44) and *N*-phenyl sulfonamide 46 are exposed to tris(dimethylamino)phosphorus under our optimized conditions; product 47 is selectively produced in 82 % yield, and cyclopropane 48 is not observed. By contrast, only 48 is produced from the addition of rhodium acetate to a methylene chloride solution of diazo substrate 45 and sulfonamide 46, indicating that intramolecular olefin trapping effectively competes with intermolecular N–H insertion for the electrophilic metal carbenoid.

In conclusion, we have described a mild and direct method the coupling of α -keto esters and protic oxygen and nitrogen pronucleophiles for the synthesis of valuable α -heterofunctionalized carbonyl derivatives. The reaction effects a direct reductive X–H functionalization by employing nonmetal reagents and substrates that are readily available and benchstable. Consequently, neither prefunctionalization of the coupling partners nor the intermediacy of highly reactive diazo compounds is required. Additional applications of the reactive system composed of α -dicarbonyl compounds and oxygen-atom acceptors for the controlled access to nucleophilic carbene synthons are currently under investigation.

© 2012 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

einheim www.angewandte.org These are not the final page numbers!



Experimental Section

Representative procedure: In a dried 25 mL round bottom flask, methyl benzoylformate (164 mg, 1.0 mmol) and *p*-cresol (113 mg, 1.05 mmol) were dissolved in dry tetrahydrofuran (10 mL, 0.1M) and the solution was cooled to -78 °C. Tris(dimethylamino)phosphine (0.20 mL, 1.05 mmol) was added dropwise by a syringe. Upon complete addition of the aminophosphine, the cooling bath was removed and the solution was allowed to warm to ambient temperature over the course of 1 h. Once the starting material was consumed (monitored by TLC), the reaction mixture was concentrated to a residue and redissolved in 50 mL of ethyl acetate. The organic layer was washed sequentially with 10% NaOH and brine, then dried (Na₂SO₄) and concentrated. The product was isolated in 88% yield (225 mg) after purification by column chromatography (silica gel, 1:10 EtOAc/hexanes as eluent).

Received: July 14, 2012 Published online: ■■■, ■■■

Keywords: carbenoids \cdot insertion \cdot ketones \cdot phosphorus \cdot synthetic methods

- a) G. M. Coppola, H. F. Schuster, α-Hydroxy Acids in Enantioselective Synthesis, Wiley-VCH, Weinheim, 1997; b) G. M. Coppola, H. F. Schuster, Asymmetric Synthesis: Construction of Chiral Molecules Using Amino Acids, Wiley-Interscience, New York, 1987.
- [2] For alternative syntheses of α-heterofunctionalized carbonyl derivatives, see: Reduction/hydrogenation of α-keto and α-imino carbonyl derivatives and their tautomers: a) W.S. Knowles, Angew. Chem. 2002, 114, 2096; Angew. Chem. Int. Ed. 2002, 41, 1998; b) R. Noyori, Angew. Chem. 2002, 114, 2108; Angew. Chem. Int. Ed. 2002, 41, 2008; Electrophilic functionalization of enols and enamines: c) S. Mukherjee, J. W. Yang, S. Hoffmann, B. List, Chem. Rev. 2007, 107, 5471; d) B.-C. Chen, P. Zhou, F.A. Davis, E. Ciganek, Org. React. 2003, 62, 1; Nucleophilic addition to (imino)glyoxylates: e) D.A. Evans, S. W. Tregay, C. S. Burgey, N. A. Paras, T. Vojkovsky, J. Am. Chem. Soc. 2000, 122, 7936; f) K. Mikami, M. Terada, T. Nakai, J. Am. Chem. Soc. 1990, 112, 3949; g) S. M. Weinreb, Top. Curr. Chem. 1997, 190, 131; h) S. Kobayashi, Y. Mori, J. S. Fossey, M. M. Salter, Chem. Rev. 2011, 111, 2626.
- [3] For examples of racemic carbenoid X-H insertions, see: a) P. Yates, J. Am. Chem. Soc. 1952, 74, 5376; b) M. P. Moyer, P. L. Feldman, H. Rapoport, J. Org. Chem. 1985, 50, 5223.
- [4] For enantioselective O-H insertions, see: a) T. C. Maier, G. C. Fu, J. Am. Chem. Soc. 2006, 128, 4594; b) C. Chen, S.-F. Zhu, B. Liu, L.-X. Wang, Q.-L. Zhou, J. Am. Chem. Soc. 2007, 129, 12616; c) S.-F. Zhu, C. Chen, Y. Cai, Q.-L. Zhou, Angew. Chem. 2008, 120, 946; Angew. Chem. Int. Ed. 2008, 47, 932.
- [5] For enantioselective N-H insertion, see: a) S. Bachmann, D. Fielenbach, K. A. Jørgensen, Org. Biomol. Chem. 2004, 2, 3044;
 b) B. Liu, S.-F. Zhu, W. Zhang, C. Chen, Q.-L. Zhou, J. Am. Chem. Soc. 2007, 129, 5834; c) E. C. Lee, G. C. Fu, J. Am. Chem. Soc. 2007, 129, 12066; d) B. Xu, S.-F. Zhu, X.-L. Xie, J.-J. Shen, Q.-L. Zhou, Angew. Chem. 2011, 123, 11685; Angew. Chem. Int. Ed. 2011, 50, 11483.
- [6] M. P. Doyle, M. A. McKervey, T. Ye, Modern Catalytic Methods for Organic Synthesis with Diazo Compounds: From Cyclopropanes to Ylides, Wiley-Interscience, New York, 1998.
- [7] a) M. Regitz, G. Maas, *Diazo Compounds—Properties and Synthesis*, Academic Press, Orlando, **1986**; b) M. Regitz in *The Chemistry of Diazonium and Diazo Groups*, *Vol. 2* (Ed.: S. Patai), Wiley, New York, **1978**, p. 659.

- [8] a) Closs and Moss described "intermediates which exhibit reactions qualitatively similar to those of carbenes without necessarily being free divalent carbon species" as carbenoids. G. L. Closs, R. A. Moss, J. Am. Chem. Soc. 1964, 86, 4042; b) IUPAC describes carbenoids as "complexed carbene-like entities that display the reactivity characteristics of carbenes". G. P. Moss, P. A. S. Smith, D. Tavernier, Pure Appl. Chem. 1995, 67, 1307; for example, see iodonium ylides: c) P. Müller, Acc. Chem. Res. 2004, 37, 243; a-halo organometallics: d) G. Köbrich, A. Akhtar, F. Ansari, W. E. Breckoff, H. Büttner, W. Drischel, R. H. Fischer, K. Flory, H. Fröhlich, W. Goyert, H. Heinemann, I. Hornke, H. R. Merkle, H. Trapp, W. Zündorf, Angew. Chem. 1967, 79, 15; Angew. Chem. Int. Ed. Engl. 1967, 6, 41; sulfonium ylides: e) M. Gandelman, B. Rybtchinski, N. Ashkenazi, R. M. Gauvin, D. Milstein, J. Am. Chem. Soc. 2001, 123, 5372; sulfoxonium ylides: f) J. E. Baldwin, R. M. Adlington, C. R. A. Godfrey, D. W. Gollins, J. G. Vaughan, J. Chem. Soc. Chem. Commun. 1993, 1434; g) I. K. Mangion, I. K. Nwamba, M. Shevlin, M. A. Huffman, Org. Lett. 2009, 11, 3566; Tosylhydrazone salts: h) J. R. Fulton, V. K. Aggarwal, J. de Vicente, Eur. J. Org. Chem. 2005, 1479; i) J. Barluenga, M. Tomás-Gamasa, F. Aznar, C. Valdés, Angew. Chem. 2010, 122, 5113; Angew. Chem. Int. Ed. 2010, 49, 4993; j) J. Barluenga, C. Valdés, Angew. Chem. 2011, 123, 7626; Angew. Chem. Int. Ed. 2011, 50, 7486; k) Z. Shao, H. Zhang, Chem. Soc. Rev. 2012, 41, 560.
- [9] G. Maas, Angew. Chem. 2009, 121, 8332; Angew. Chem. Int. Ed. 2009, 48, 8186.
- [10] The generation of carbene intermediates by phosphorus(III)mediated atom transfer is implicated in the Corey–Winter olefination; see: a) E. J. Corey, R. A. E. Winter, J. Am. Chem. Soc. 1963, 85, 2677; b) E. Block, Org. React. 1984, 30, 457; c) D. Horton, C. G. Tindall, Jr., J. Org. Chem. 1970, 35, 3558.
- [11] F. H. Osman, F. A. El-Samahy, *Chem. Rev.* 2002, *102*, 629, and references therein.
- [12] a) V. Mark, J. Am. Chem. Soc. 1963, 85, 1884; b) M. S. Newman,
 S. Blum, J. Am. Chem. Soc. 1964, 86, 5598; c) F. Ramirez, A. S.
 Gulati, C. P. Smith, J. Org. Chem. 1968, 33, 13; d) X. Liu, J. G.
 Verkade, J. Org. Chem. 2000, 65, 4560.
- [13] a) E. Corre, A. Foucaud, J. Chem. Soc. D 1971, 570; b) H. Fauduet, R. Burgada, Synthesis 1980, 642.
- [14] R. Burgada, H. Fauduet, Nouv. J. Chim. 1980, 4, 113.
- [15] Formation of hexamethylphosphoramide as a by-product is confirmed by a ³¹P NMR spectrum of the crude reaction mixture. Under these conditions, the reductive X-H functionalization apparently outcompetes amide-to-alkoxide ligand exchange at phosphorus(III). E. E. Nifantiev, M. K. Grachev, S. Y. Burmistrov, *Chem. Rev.* **2000**, *100*, 3755.
- [16] See Supporting Information for full experimental details.
- [17] The following P=O bond dissociation enthalpies have been reported: (Et₂N)₃P=O 156, (EtO)₃P=O 151, Ph₃P=O 127 kcal mol⁻¹. S. B. Hartley, W. S. Holmes, J. K. Jacques, M. F. Mole, J. C. McCoubrey, *Q. Rev.* **1963**, *17*, 204.
- [18] Epoxide products arising from reductive dimerization of substrate (see Ref. [12]) are formed instead under these conditions.
- [19] When ethanol is employed as bulk solvent, O-H functionalization is observed but is attended by concomitant transesterification.

$$\begin{array}{c} O \\ Ph \\ \hline CO_2 Me \end{array} \xrightarrow{1.05 \text{ equiv } P(NMe_2)_3} \\ \hline EtOH, -78 \ ^\circ C \text{ to } RT \\ 1 \text{ h} \\ \hline 57\% \\ \end{array} \xrightarrow{OEt} Ph \\ \hline CO_2 Et \\ \hline$$

[20] The following pK_a values have been determined in H₂O at 25 °C: pK_a (trifluoroethanol) 12.4, pK_a (propargyl alcohol) 13.6, pK_a -

www.angewandte.org

© 2012 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

These are not the final page numbers!

(ethanol) 15.9. See: P. Ballinger, F. A. Long, J. Am. Chem. Soc. 1960, 82, 795.

- [21] a) F. Ramirez, Pure Appl. Chem. 1964, 9, 337; b) F. Ramirez, Acc. Chem. Res. 1968, 1, 168.
- [22] The intermediacy of free carbenes from the deoxygenation of αdicarbonyl compounds with phosphorus(III) reagents has been proposed: A. Afonso, F. Hon, J. Weinstein, A. K. Ganguly, A. T. McPhail, J. Am. Chem. Soc. **1982**, 104, 6138.
- [23] For a phosphine-mediated reduction of electron-deficient ketones by a related mechanism, see: a) W. Zhang, M. Shi, *Chem. Commun.* 2006, 1218; b) M. Shi, X.-G. Liu, Y.-W. Guo, W. Zhang, *Tetrahedron* 2007, 63, 12731.
- [24] For related displacements of phosphine oxide from alkoxyphosphonium intermediates, see the Mitsunobu reaction: a) K. C. Kumara Swamy, N. N. Bhuvan Kumar, E. Balaraman, K. V. P.

Pavan Kumar, *Chem. Rev.* **2009**, *109*, 2551, and references therein. For Arbuzov and related reactions, see: b) B. R. Castro, *Org. React.* **1983**, *29*, 1.

- [25] Chiral diazaphospholidine 42 is prepared in one step on a gram scale by reaction of the appropriate chiral cyclohexanediamine derivative with P(NMe₂)₃. For related syntheses, see: A. Alexakis, S. Mutti, P. Mangeney, J. Org. Chem. 1992, 57, 1224.
- [26] Stereoselectivity is observed for phenol and sulphonamide nucleophiles, although the optical yields are diminished.
- [27] A polar mechanism is also believed to operate in metal carbenoid X-H insertion reactions: a) W. Kirmse, in Advances in Carbene Chemistry, Vol. 1 (Ed.: U. Brinker), JAI Press, Greenwich, CT, 1994, p. 1; b) Y. Liang, H. Zhou, Z.-X. Yu, J. Am. Chem. Soc. 2009, 131, 17783, and references therein.
- [28] M. P. Doyle, Acc. Chem. Res. 1986, 19, 348.



Carbenoids

Communications



E. J. Miller, W. Zhao, J. D. Herr, A. T. Radosevich* _____ IIII--IIII

A Nonmetal Approach to α-Heterofunctionalized Carbonyl Derivatives by Formal Reductive X-H Insertion



Keeping it organic: A direct synthesis of α -alkoxy and α -amino ester derivatives by direct reductive coupling of widely available, stable α -keto esters and protic pronucleophiles is described (see

scheme; X = OR, NR_2). The method serves as a convenient nonmetal alternative to X-H insertion by diazo decomposition.

6 www.angewandte.org

© 2012 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

Angew. Chem. Int. Ed. 2012, 51, 1-6

These are not the final page numbers!