

## Acylguanidine inhibitors of $\beta$ -secretase: Optimization of the pyrrole ring substituents extending into the $S_1$ and $S_3$ substrate binding pockets

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**Abstract**—Proteolytic cleavage of amyloid precursor protein by  $\beta$ -secretase (BACE-1) and  $\gamma$ -secretase leads to formation of  $\beta$ -amyloid (A $\beta$ ) a key component of amyloid plaques, which are considered the hallmark of Alzheimer's disease. Small molecule inhibitors of BACE-1 may reduce levels of A $\beta$  and thus have therapeutic potential for treating Alzheimer's disease. We recently reported the identification of a novel small molecule BACE-1 inhibitor *N*-[2-(2,5-diphenyl-pyrrol-1-yl)-acetyl]guanidine (**3.a.1**). We report here the initial hit-to-lead optimization of this hit and the SAR around the aryl groups occupying the  $S_1$  and  $S_2$  pockets leading to sub-micromolar BACE-1 inhibitors.

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Alzheimer's disease (AD) is a progressive neurodegenerative disease that is the leading cause of dementia.<sup>1</sup> Varying estimates suggest that from 5% to 50% of early onset AD cases result from a variety of genetic mutations, meaning the remainder are sporadic with the risk increasing with age.<sup>2</sup>  $\beta$ -Amyloid plaques are one of the key hallmarks of AD and are formed by aggregation of amyloid fibrils, which in turn are formed from the neurotoxic amyloid  $\beta$ -peptide ( $\beta$ -amyloid, A $\beta_{40,42}$ ). Proteolytic cleavage of amyloid precursor protein (APP) by  $\beta$ -secretase (also referred to as  $\beta$ -site APP cleaving enzyme-1, BACE-1, Asp-2, and memapsin-2) generates the membrane-bound  $\beta$ -C-terminal fragment ( $\beta$ CTF, also referred to as C99) which in turn is cleaved by

$\gamma$ -secretase to generate A $\beta$ .<sup>3–5</sup> Thus, BACE-1 inhibitors may decrease levels of A $\beta$  and have therapeutic benefit in the treatment of AD.<sup>6,7</sup> A high throughput screening campaign identified *N*-[2-(2,5-diphenyl-pyrrol-1-yl)-acetyl]guanidine **3.a.1** (Ar<sup>1</sup> & Ar<sup>2</sup> = Ph, Table 1) as an inhibitor of BACE-1 with low micromolar activity (BACE-1 IC<sub>50</sub> = 3.7  $\mu$ M, K<sub>d</sub> = 2.8  $\mu$ M).<sup>8</sup> Compound **3.a.1** inhibited A $\beta$  formation in a cellular assay with an IC<sub>50</sub> of 8.9  $\mu$ M and consistent with a BACE-1 mediated inhibition mechanism caused a dose dependent reduction of  $\beta$ CTF and A $\beta$  levels in a radiolabeled immunoprecipitation assay without affecting  $\alpha$ -secretase amyloid precursor protein ( $\alpha$ -sAPP) levels.<sup>8</sup> Compound **3.a.1** was selective for BACE-1 over cathepsin D (IC<sub>50</sub> = 60  $\mu$ M), pepsin (IC<sub>50</sub> > 50  $\mu$ M), and BACE-2 (IC<sub>50</sub> > 50  $\mu$ M).

An X-ray structure of **3.a.1** with BACE-1 was solved and revealed key binding interactions between the

**Keywords:** BACE-1;  $\beta$ -Secretase; Inhibitor; Acylguanidine.

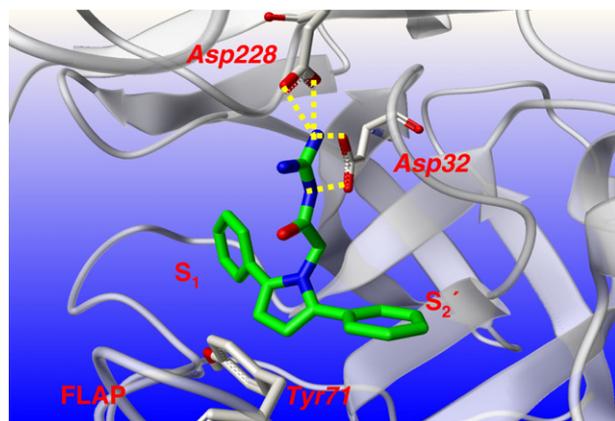
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**Table 1.** BACE-1 inhibitory activities of compounds **3(a–l)**, (**1–13**)

Ar <sup>1</sup>	Ar <sup>2</sup>	a	b	c	d	e	f	g	h	i	j	k	l
	Inhibition (%) at 10 μM/IC <sub>50</sub> (μM)	Ph	2-Cl-Ph	2-MeO-Ph	3-F-Ph	3-Cl-Ph	3-MeO-Ph	3-CN-Ph	4-F-Ph	4-MeO-Ph	4-CF3-Ph	1-Naphthyl	3-Benzothienyl
<b>1</b>	Ph	<b>3.7</b>		51%	65%		78%	34%	66%		63%	56%	
<b>2</b>	2-Cl-Ph	<b>3</b>	56%	62%	69%		39%	52%	<b>2</b>	85%	81%	35%	40%
<b>3</b>	3-Cl-Ph	69%	58%	51%	56%		37%	38%	66%	59%	75%	34%	28%
<b>4</b>	3-Br-Ph	68%	69%	46%	47%			41%	63%	66%	64%	25%	9%
<b>5</b>	3-Me-Ph	<b>4.6</b>	70%	56%	63%		57%	22%	72%	41%	71%	37%	38%
<b>6</b>	4-Me-Ph	75%	43%	48%	70%		62%	3%	65%	39%	66%	61%	
<b>7</b>	4-MeO-Ph	53%		32%	58%		61%	23%	53%	12%	43%	9%	16%
<b>8</b>	2,5-di-Cl-Ph	3.5	48%	78%	68%		65%	60%	<b>2</b>	83%	92%	51%	
<b>9</b>	4-PhO-Ph	<b>1.3</b>	<b>1.1</b>	<b>1.3</b>	<b>1.4</b>	<b>1.3</b>	<b>1.5</b>	<b>0.6</b>	<b>1.4</b>	<b>0.5</b>	<b>0.6</b>		<b>0.7</b>
<b>10</b>	4-(4-Ac-PhO)-Ph	<b>1.0</b>	<b>0.7</b>	<b>1.6</b>	<b>1.2</b>		<b>1.3</b>	<b>0.8</b>	<b>1.3</b>	<b>0.6</b>	<b>0.9</b>		
<b>11</b>	4-BnO-Ph	64%	84%		58%		79%	41%	47%		25%		7%
<b>12</b>	2-Naph	56%	70%	53%	50%		64%	39%	50%	44%	72%	29%	40%
<b>13</b>	6-Me-2-Naph	60%	84%	53%	61%		79%	46%	58%	17%	49%		

acylguanidine moiety and the two catalytic aspartic acids, Asp32 and Asp228 (Fig. 1).<sup>8–10</sup> The pyrrole ring points toward the flap region, making a  $\pi$ -edge stacking interaction with Tyr71 and stabilizing the flap in an ‘open’ conformation relative to published peptidomimetic co-structures.<sup>11</sup> The two phenyl groups on the pyrrole ring extend into the S<sub>1</sub> and S<sub>2</sub> substrate binding pockets. The *para*-position of the P<sub>1</sub> phenyl group projects directly toward the unoccupied S<sub>3</sub> pocket, indicating an opportunity to add substituents to the P<sub>1</sub> phenyl extending into the S<sub>3</sub> pocket and thereby potentially increasing binding affinity. This manuscript describes the initial hit-to-lead optimization of this series of BACE-1 inhibitors.

Initial SAR investigation of this lead began with the synthesis and assay of a 12 × 13 combinatorial library. The selection of target molecules was carried out interac-

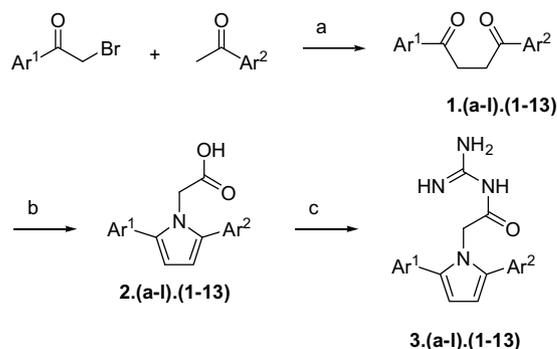


**Figure 1.** Complex of **3.a.1** with BACE-1 showing hydrogen bonding interactions between the acylguanidine moiety and Asp32 and Asp228, the  $\pi$ -edge stacking interaction between the pyrrole ring and Tyr71 and the two phenyl groups projecting into S<sub>1</sub> and S<sub>2</sub>.<sup>10</sup>

tively with molecular modelers and was based on the anticipated ability of different substituent groups to be accommodated into the S<sub>1</sub>–S<sub>3</sub> and/or S<sub>2</sub> pockets. Analogs were synthesized from  $\alpha$ -bromomethyl ketones and methyl ketones as starting materials as shown in Scheme 1.

The 1,4-diarylbutane-1,4-diones **1(a–l)**, (**1–13**) were prepared in a one-step cross-coupling reaction under the action of ZnCl<sub>2</sub> · *t*-BuOH · Et<sub>2</sub>NH using the procedure of Kulinkovich.<sup>12</sup> Reactions were carried out over 3–5 days at ambient temperature and gave modest yields of the diketones (36–50%). The crude diketones were coupled with glycine in an acid catalyzed condensation and subsequent CDI mediated coupling of the resultant pyrrole acetic acids with guanidine hydrochloride gave the target acylguanidines.

All compounds were assayed for BACE-1 inhibition in a fluorescence resonance energy transfer (FRET) assay<sup>8</sup> at a concentration of 10 μM, and IC<sub>50</sub>'s were determined

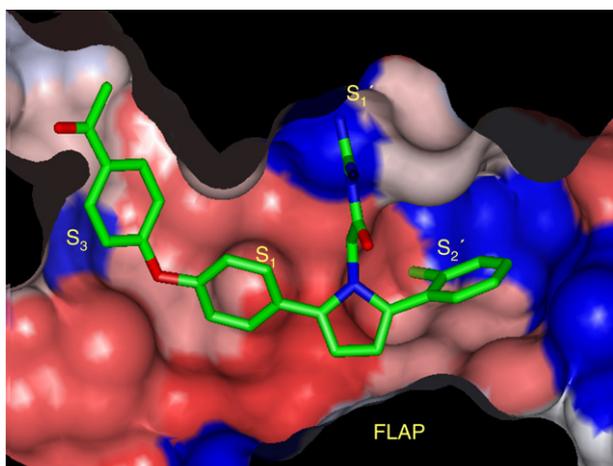


**Scheme 1.** Reagents and conditions: (a) Et<sub>2</sub>NH, *t*-BuOH, ZnCl<sub>2</sub>, toluene, rt, 2–5 days; (b) glycine, *p*-TSA, EtOH, 80 °C, 3 days; (c) i—CDI, DMF, rt, 1 h; ii—guanidine·HCl, Et<sub>3</sub>N, rt, 5 h.

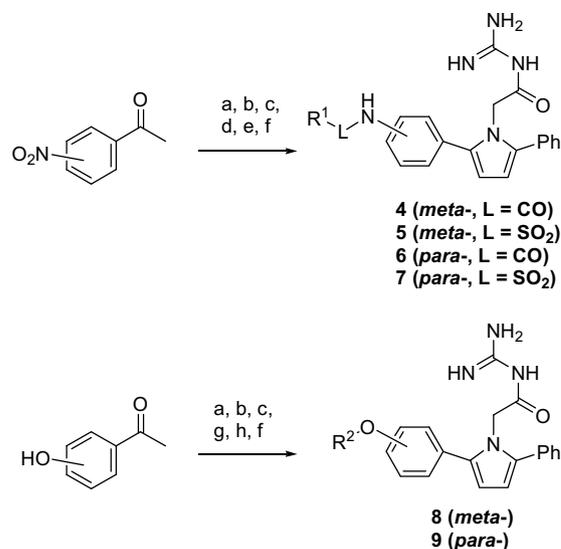
for compounds showing > 80% inhibition (Table 1). For acyl guanidines with smaller phenyl ring substituents (Table 1, rows 1–8), BACE-1 inhibition activity was not appreciably higher than for **3.a.1** (3.7  $\mu\text{M}$ , 66% inhibition @10  $\mu\text{M}$ ). Because of the pseudo-symmetry of molecules with smaller phenyl ring substituents, the phenyl rings may be accommodated in either the  $S_1$  and  $S_2'$  pockets making interpretation of the SAR difficult. Improved potency was seen in all examples with the 4-phenoxyphenyl and 4-(4-acylphenoxy)phenyl substituent groups (Table 1, rows 9 and 10). These functional groups were included in the library design with the expectation that the phenoxy group on the  $P_1$  phenyl would extend into the  $S_3$  pocket. *para*-Benzyloxyphenyl analogs were poorly tolerated indicating this group may be too large to be accommodated (Table 1, row 11). Similarly the bicyclic ring systems were poorly tolerated (Table 1, row 12 and 13, and columns k and l). When  $\text{Ar}^1$  is 4-(PhO)Ph or 4-(AcPhO)-Ph, the arylethers are locked into the  $S_1$ – $S_3$  pocket facilitating interpretation of the  $\text{Ar}^2$  SAR. Somewhat surprisingly, no clear preference for *ortho*, *meta*, or *para*-substitution of the  $P_2'$  phenyl was observed in this library (see Table 1, rows 9 and 10).

A co-crystal structure of compound **3.b.10** (BACE-1  $\text{IC}_{50}$  = 0.7  $\mu\text{M}$ ), with BACE-1 (Fig. 2) shows the acyl-guanidine forming key H-bonding interactions with the two aspartic acids and the pyrrole ring  $\pi$ -stacking with the Tyr71 of the flap region.<sup>10</sup> The 4-(4-acetylphenoxy)-phenyl group occupies the  $S_1$  and  $S_3$  substrate binding pockets as predicted by our models and the 2-chloro group on the  $P_2'$  phenyl is oriented toward the back of the  $S_2'$  pocket, although this substituent does not appear to contribute significantly to potency (compare **3.a.10** and **3.b.10**, Table 1).

Because of the significant increase in potency observed for acyl guanidines with 4-phenoxy substituents on the  $P_1$  phenyl, additional analogs were designed to further investigate substituents projecting into the  $S_3$  pocket. *Meta* and *para*-acylamino-phenyl and sulfonylamino-phenyl (**4–7**), and *meta* and *para*-alkoxyphenyl substitu-



**Figure 2.** Complex of **3.b.10** with BACE-1. Polar/charged residues are shown in blue and lipophilic residues are shown in red.<sup>10</sup>



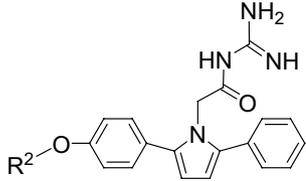
**Scheme 2.** Reagents and conditions: (a) Et<sub>2</sub>NH, *t*-BuOH, ZnCl<sub>2</sub>, toluene, rt, 2–5 days; (b) glycine, AcOH, 120 °C, 5 h; (c) TMSCl, MeOH, 70 °C, 2 h; (d) 10% Pd/C, MeOH–THF, 1 h; (e) R<sup>1</sup>COCl or R<sup>1</sup>SO<sub>2</sub>Cl, Et<sub>3</sub>N, DCM, 16 h; (f) guanidine, DMSO, rt, 2 h; (g) BBr<sub>3</sub>, DCM, –78 °C, then rt, 5 h; (h) R<sup>2</sup>Br, K<sub>2</sub>CO<sub>3</sub>, DMF, 65 °C, 16 h.

ents (**8–9**) were prepared by modifications of the general route outlined in Scheme 1 (see Scheme 2).

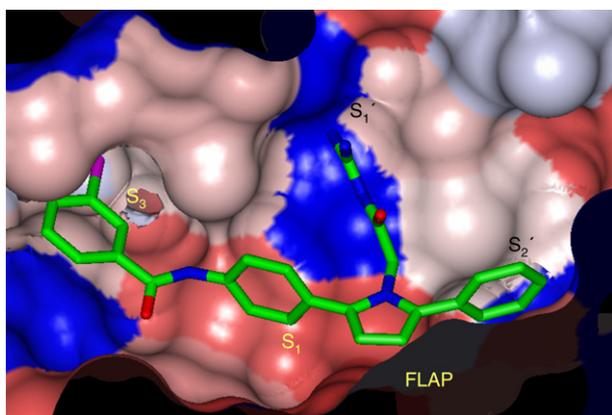
The amides and sulfonamides linked to the  $P_1$  phenyl at the *meta*-position (**4** and **5**) all had weak activity (data not shown), while the *para*-alkyl, heteroaryl, and arylamides (Table 2, **6a–j**) generally had low  $\mu\text{M}$  activity with **6g** being the most potent analog ( $\text{IC}_{50}$  = 0.6  $\mu\text{M}$ ). The corresponding *para*-sulfonamides (**7**) had much weaker activity. Similarly, the set of *meta*-alkoxyphenyl analogs (**8**) had low activity (data not shown) while the *para*-alkoxyphenyl analogs (Table 3, **9a–g**) were surprisingly

**Table 2.** BACE-1 inhibitory activities of compounds **6** and **7**

	R <sup>1</sup>	L	% Inhibition at 10 $\mu\text{M}$ /IC <sub>50</sub> ( $\mu\text{M}$ )
<b>6a</b>	Me	–CO–	40%
<b>6b</b>	Et	–CO–	71%
<b>6c</b>	Cyclopropyl	–CO–	42%
<b>6d</b>	<i>t</i> -Bu	–CO–	79%
<b>6e</b>	cyclohexyl	–CO–	26%
<b>6f</b>	2,4-di-Cl-Ph	–CO–	<b>1.2</b>
<b>6g</b>	3-Br-Ph	–CO–	<b>0.6</b>
<b>6h</b>	3-MeO-Ph	–CO–	<b>1.8</b>
<b>6i</b>	4-Br-Ph	–CO–	<b>1.4</b>
<b>6j</b>	4-F-Bn	–CO–	54%
<b>7a</b>	Me	–SO <sub>2</sub> –	65%
<b>7b</b>	Et	–SO <sub>2</sub> –	75%
<b>7c</b>	<i>n</i> -Propyl	–SO <sub>2</sub> –	71%
<b>7d</b>	Bn	–SO <sub>2</sub> –	86%
<b>7e</b>	3-MeO-Ph	–SO <sub>2</sub> –	50%
<b>7f</b>	3-Br-Ph	–SO <sub>2</sub> –	65%
<b>7g</b>	3-Me-Ph	–SO <sub>2</sub> –	43%

**Table 3.** BACE-1 inhibitory activities of compounds **9**


	R <sup>2</sup>	% Inhibition at 10 μM/IC <sub>50</sub> (μM)
<b>9a</b>	Et	<b>1.6</b>
<b>9b</b>	allyl	<b>1.7</b>
<b>9c</b>	Propargyl	83%
<b>9d</b>	3-CN-Propyl	<b>0.9</b>
<b>9e</b>	Butyl	<b>1.8</b>
<b>9f</b>	4-CN-Bn	<b>1.9</b>
<b>9g</b>	Phenethyl	68%

**Figure 3.** Complex of **6g** with BACE-1. Polar/charged residues are shown in blue and lipophilic residues are shown in red.<sup>10</sup>

active, indicating that sub-μM activity is possible with a smaller alkyl group occupying the S<sub>3</sub> pocket (see **9d**).

The crystal structure of the complex of acylguanidine **6g** with BACE-1 was subsequently solved and revealed that the 3-bromobenzoyl substituent was tightly packed into the S<sub>3</sub> pocket (Fig. 3).<sup>10</sup> Accommodation of this rather large substituent into the pocket was only possible with twisting of the amide bond out of plane with respect to the P<sub>1</sub> group. Presumably this twisting of the amide bond is unfavorable leading to reduced potency. In contrast, a model of **9d** indicates that the cyanopropylether extends into the S<sub>3</sub> pocket with minimal strain achieving similar potency with greater atom efficiency.

In general, these inhibitors are selective for BACE-1 over cathepsin-D (usually >50-fold, although **6g** is only 4-fold) and pepsin (data not shown) and have limited selectivity (<2-fold) over BACE-2<sup>13</sup> (Table 3). Most sub-micromolar BACE-1 inhibitors had low micromolar activity in the cellular Aβ<sup>total</sup> lowering ELISA (Table 4).

In summary, in this initial hit-to-lead optimization effort we have achieved up to 6-fold increase in potency and demonstrated the ability to extend directly from the S<sub>1</sub> pocket into the S<sub>3</sub> pocket with groups attached via the *para*-position of the P<sub>1</sub> group and with a variety of link-

**Table 4.** IC<sub>50</sub> values for inhibition of BACE-1, BACE-2, cathepsin-D and cellular Aβ production by **3.b.10** and **6g**

	BACE-1 IC <sub>50</sub> , μM	BACE-2 IC <sub>50</sub> , μM	CathD IC <sub>50</sub> , μM	Cellular Aβ <sup>total</sup> ED <sub>50</sub> , μM
<b>3.b.10</b>	0.7	2.3	18.9	3.8
<b>6g</b>	0.6	0.5	2.1	5.6

ers. Future studies will focus on optimization of the interactions with S<sub>3</sub> while reducing the strain energy of the S<sub>1</sub>–S<sub>3</sub> linker, and also incorporation of substituents on the acylguanidine extending into the S<sub>1</sub>' pocket.

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