DDQ-mediated Direct Intramolecular-Dehydrogenative-Coupling (IDC): Expeditious Approach to the Tetracyclic Core of Ergot Alkaloids[‡]

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An efficient route to 2-oxindoles bearing an all-carbon quaternary center at the pseudobenzylic position has been developed via a DDQ-mediated Intramolecular-Dehydrogenative-Coupling (IDC). The methodology involves a one-pot C-alkylation of β -*N*-arylamido esters (7) concomitant with dehydrogenative-coupling in the presence of stoichiometric amount of DDQ. A tentative mechanistic route has been proposed for the oxidative coupling. The methodology provides a two-step entry to the ergoline structure of ergot alkaloids.

Direct C–H functionalization through oxidative coupling between two C–H bonds¹ offers a unique opportunity to access molecules of great synthetic interest from readily available starting materials.² Classically, this method involves one or two metal catalysts together with an oxidizing agent which can serve as a hydrogen acceptor.³ Despite several advantages associated with these strategies, few challenges such as highly efficient and selective dehydrogenative-coupling utilizing two different hydrocarbons as reagents under metal-free conditions still remains to be addressed. Toward this end, on the basis of the wellestablished studies on various oxidations using DDQ,⁴ Li and co-workers demonstrated a Cross-Dehydrogenative-Coupling $(CDC)^{5a}$ of benzyl ethers (1a) with unmodified ketones (1b) (Scheme 1) utilizing two C(sp³)-H bonds.⁵





2-Oxindoles containing an all carbon quaternary center at the pseudobenzylic position constitute a common structural motif in many biologically active natural products and have become an attractive and challenging synthetic target.⁶ Recently, Kündig,⁷ Taylor,⁸ and we^{9a} have developed

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⁽¹⁾ For reviews describing oxidative C-C formation, see: (a) Yeung, C. S.; Dong, V. M. *Chem. Rev.* **2011**, *111*, 1215. (b) Cho, S. H.; Kim, J. Y.; Kwak, J.; Chang, S. *Chem. Soc. Rev.* **2011**, 5068. (c) McGlacken, G. P.; Bateman, L. M. *Chem. Soc. Rev.* **2009**, 2447.

⁽²⁾ For reviews on the ideal chemical synthesis, see: (a) Newhouse, T.; Baran, P. S.; Hoffmann, R. W. *Chem. Soc. Rev.* **2009**, 3010. (b) Gaich, T.; Baran, P. S. *J. Org. Chem.* **2010**, 75, 4657. (c) Hendrickson, J. B. *J. Am. Chem. Soc.* **1975**, 97, 5784.

^{(3) (}a) Dyker, G., Ed. Handbook of C-H Transformations: Applications in Organic Synthesis, Volume 2; Wiley: New York, 2005. (b) Ashenhurst, J. A. Chem. Soc. Rev. 2010, 540. (c) Le Bras, J.; Muzart, J. Chem. Rev. 2011, 111, 1170. (d) Liu, C.; Zhang, H.; Shi, W.; Lei, A. Chem. Rev. 2011, 111, 1780.

⁽d) Liu, C.; Zhang, H.; Shi, W.; Lei, A. Chem. Rev. 2011, 111, 1780.
(4) (a) Xu, Y. -C.; Roy, C.; Lebeau, E. Tetrahedron Lett. 1993, 34, 8189. (b) Xu, Y. -C.; Lebeau, E.; Attardo, G.; Myers, P. L.; Gillard, J. W. J. Org. Chem. 1994, 59, 4868.

^{(5) (}a) For metal-free CDC, see: Zhang, Y.; Li, C.-J. J. Am. Chem. Soc. 2006, 128, 4242. (b) Zhang, Y.; Li, C.-J. Angew. Chem., Int. Ed. 2006, 45, 1949. (c) Li, C.-J. Acc. Chem. Res. 2009, 42, 335. (d) Yoo, W.-J.; Li, C.-J. Top. Curr. Chem. 2010, 292, 281.

Intramolecular-Dehydrogenative-Coupling (IDC) reactions by utilizing the C(sp²)-H bond with a C(sp³)-H bond. Herein, we envisioned a DDQ (2,3-dichloro-5,6-dicyanobenzoquinone) mediated direct C–H functionalization of β -*N*arylamido ester (**3a**) to carry out an IDC of a C(sp²)-H bond ortho to the *N*-alkylanilides with an internal C(sp³)-H bond of active methine (**3a**) without using any metal based oxidant.

Employing DDQ-mediated strategy of oxidative coupling,¹⁰ we set forth to investigate the possibility of an IDC to transform β -*N*-arylamido ester of the type **3a** into **4a** (Scheme 1). Mechanistically, a single electron transfer (SET) from the anion (prepared *in situ* from **3a** by the treatment of KO'Bu) to DDQ may generate a K⁺.DDQ radical (**6a**) and a radical **5a** (Scheme 2), which could in turn form an intermediate aryl radical **5b**. The latter, following another SET to **6a** could form K⁺.DDQ anion (**6b**) and an intermediate aryl carbocation **5c**, which can be stabilized by amide nitrogen (see **5d**). Eventually, abstraction of a α -hydrogen from the iminium intermediate **5d** in the presence of **6b** with resultant rearomatization leads to the desired oxidative coupling product (**4a**) and K⁺ of *p*-quinol derivative **6c**.

Scheme 2. Proposed Mechanism for DDQ-mediated IDC



Initially, we embarked on our studies taking C-methyl β -*N*-phenylamido methylester **3a** as substrate and charged it with 1.2 equiv of KO'Bu at rt followed by treatment with 1.1 equiv DDQ in DMSO to afford product **4a** in 89% yield in 30 min. Similarly, in case of other oxidants such as iodosobenzene diacetate (PIDA) and [bis(trifluoro-acetoxy) iodo]benzene (PIFA) which could also follow a SET mechanism, we found that the reaction worked equally efficient to afford **4a** in 80% and 86% yields,

respectively [see, Supporting Information (SI) for details].^{9b} No product was formed in the absence of DDQ, or PIDA, or PIFA. Since, KO'Bu is also known to promote alkylation of **7a** (Scheme 3),^{9a} we thought to carry out a simultaneous C-alkylation of β -N-arylamido ester (**7a**) using KO'Bu and methyl iodide accompanied with a DDQ-mediated dehydrogenative-coupling in a one-pot protocol (Scheme 3).

Scheme 3. One-pot C-alkylation and IDC Mediated by DDQ



To establish a standard reaction protocol, we selected β -*N*-arylamido ester (**7a**) and methyl iodide as model substrates. Exhaustive optimization studies [see, SI for details] revealed that methylation of **7a** could be done in presence of 1.2 equiv of KO'Bu and 1.1 equiv of methyl iodide, concomitant with oxidative coupling using 1.2 equiv of KO'Bu and 1.1 equiv of DDQ to afford the desired product in 85% yield (Scheme 3).

In order to explore the synthetic viability of this oxidative coupling, we then extended it to various β -*N*-arylamido esters 7 [see, SI for details] and alkyl halides. As shown in Figure 1, a wide range of 2-oxindoles (**4a**-**n**) could be obtained in good yields. Gratifyingly, β -*N*-arylamido esters (7) with different substituents including allyl, methallyl, dimethylallyl, and geranyl, tolerate the standard reaction conditions well to afford a variety of 2-oxindoles **4q**-**z** (Figure 1) in good yields. This prompted us to examine the versatility of our approach by extending it to substrates like **8a**, which in turn provides an entry to spiro-fused 2-oxindoles **9** in 72% yield. The essence of spiro-fused 2-oxindole is evident from a range of natural products including coerulescine (**10a**) and horsfiline (**10b**) (Scheme 4).¹¹

Under optimized conditions, a few more substrates have also been examined as shown in Figure 2. We observed that compounds **7a** and **11a** were not suitable for IDC and simply led to the decomposition, probably indicating the involvement of tertiary radical in the oxidative coupling process. Unprotected amides such as **8b** and **11b** were also leading to multiple spots on TLC. We speculate that this might be due to the involvement of competitive reactions of nitrogen and carbon centered radical species.¹²

The subset of 2-oxindoles constitutes a common structural motif in many biologically active alkaloids and therefore has gained significant attention from synthetic community. A wide range of hexahydropyrroloindolines

⁽⁶⁾ For reviews on 2-oxindoles, see: (a) Millemaggi, A.; Taylor, R. J. K. *Eur. J. Org. Chem.* **2010**, 4527. For a review on asymmetric catalytic oxindole syntheses, see: (b) Zhou, F.; Liu, Y. L.; Zhou, J. A. *Adv. Synth. Catal.* **2010**, *352*, 1381. (c) For synthesis of 2-oxindoles via a Friedel-Crafts alkylations from our group, see: Ghosh, S.; Kinthada, L. K.; Bhunia, S.; Bisai, A. *Chem. Commun.* **2012**, 10132.

⁽⁷⁾ For oxidative coupling using 2.2 equiv of CuCl₂, see: (a) Jia, Y. X.; Kündig, E. P. *Angew. Chem., Int. Ed.* **2009**, *48*, 1636. (b) Dey, C.; Kündig, E. P. *Chem. Commun.* **2012**, 3064. (c) Dey, C.; Kündig, E. P. *Chem. Commun.* **2012**, 3064.

⁽⁸⁾ For oxidative coupling using 1.0 equiv of Cu(OAc)₂.H₂O, see: (a) Perry, A.; Taylor, R. J. K. *Chem. Commun.* **2009**, 3249. (b) Franckevicius, V.; Cuthbertson, J. D.; Pickworth, M.; Pugh, D. S.; Taylor, R. J. K. *Org. Lett.* **2011**, *13*, 4264. For coupling using 5–10 mol% of Cu(OAc)₂.H₂O, see: (c) Klein, J. E. M. N.; Perry, A.; Pugh, D. S.; Taylor, R. J. K. *Org. Lett.* **2010**, *12*, 3446. (d) Moody, C. L.; Franckevicius, V.; Drouhin, P.; Klein, J. E. M. N.; Taylor, R. J. K. *tetrahedron. Lett.* **2012**, *53*, 1897.

^{(9) (}a) For I₂-mediated IDC, see: Ghosh, S.; De, S.; Kakde, B. N.; Bhunia, S.; Adhikary, A.; Bisai, A. *Org. Lett.* **2012**, *14*, 5864. (b) We sincerely thank the reviewers for their valuable suggestions to check IDC using oxidants which could also follow a SET.

^{(10) (}a) Buckle, D. R. *Encyclopaedia of Reagent for Organic Synthesis*;
Paquette, L. A., Ed.; John Wiley & Sons: Chichester, UK, 1995; Vol. 3, p 1699. (b)
Patir, S.; Ertürk, E. *J. Org. Chem.* 2011, *76*, 335. (c) Ying, B.-P.; Trogden,
B. G.; Kohlman, D. T.; Liang, S. X.; Xu, Y.-C. Org. Lett. 2004, *6*, 1523.

^{(11) (}a) Deppermann, N.; Thomanek, H.; Prenzel, A. H. G. P.; Maison, W. J. Org. Chem. 2010, 75, 5994. (b) Trost, B. M.; Brennan, M. K. Org. Lett. 2006, 8, 2027.

⁽¹²⁾ Richter, J. M.; Whitefield, B. W.; Maimone, T. J.; Lin, D. W.; Castroviejo, M. P.; Baran, P. S. J. Am. Chem. Soc. 2007, 129, 12857.

⁽¹³⁾ Wang, Y.-H.; Long, C.-L.; Yang, F.-M.; Wang, X.; Sun, Q.-Y.; Wang, H.-S.; Shi, Y.-N.; Tang, G.-H. J. Nat. Prod. **2009**, 72, 1151.



Figure 1. Substrate scope of DDQ-mediated IDC. The standard condition: reactions were carried out on a 1.0 mmol of **7** with 1.1 mmol of alkyl halide in the presence of 1.2 mmol of KO'Bu in 3 mL of DMSO at 25 °C for alkylations and 1.1 mmol of DDQ in presence of 1.2 mmol of KO'Bu under heating at 100 °C for oxidative couplings.

Scheme 4. Synthesis of Spiro 2-Oxindole by IDC





Figure 2. Substrate scope using different β -arylamidoesters.

(Figure 3) sharing a substituted prenyl functionality at C(3a), such as selaginellic acid (**12a**), *N*-selaginelloyl-L-phenylalanine (**12b**), neoselaginellic acid (**13a**), and *N*-neoselaginelloyl-L-phenylalanine (**13b**),¹³ and reverse-prenyl functionality at C(3a),^{14,15} as well as geranyl functionality at C(3a)¹⁵ which exhibit a broad spectrum of biological

activities, drew our interest to exploit the potential utility of 2-oxindole derivatives as precursors.

In order to achieve our target aiming synthetic approaches to these alkaloids, we further expanded the substrate scope. However, we found difficulties in onepot alkylation followed by IDC when the reaction was conducted with iodide 15 as alkylating agent (Figure 3). Thus, it was decided to carry out the IDC with C-alkylated starting materials $\pm 16a$ -c. Notably, we could synthesize IDC products $\pm 17a - b$ and ± 18 in 62–64% yields from $\pm 16a - c$ (Figure 3).¹⁶ We believe that these compounds have potential for the regioselective and enantioselective Tsuji-Trost decarboxylative prenylation/reverse-prenylation, geranylation/reverse-geranylation in the presence of suitable chiral Pd(0) complexes.¹⁷ Especially compound $\pm 17b$ could be a potential intermediate for the total synthesis of 12a-b and 13a-b via enantioselective decarboxylative prenvlation reaction.



Figure 3. Potential utility of 2-oxindoles.

In addition, fascinated by their striking polycyclic molecular architectures and wide spectrum of physiological activities, we targeted to synthesize tetracyclic ergoline structure (20) of the ergot alkaloids (Figure 4), such as lysergic acid (19a), lysergol (19b), ergometrine (19c) and ergocristam (19d).¹⁸ We reasoned that the tetracyclic core

^{(14) (}a) Williams, R. M.; Sanz-Cervera, J. F.; Sancenón, F.; Marco, J. A.; Halligan, K. J. Am. Chem. Soc. 1998, 120, 1090. (b) Finefield, J. M.; Frisvad, J. C.; Sherman, D. H.; Williams, R. M. J. Nat. Prod. 2012, 75, 812 and references therein.

^{(15) (}a) Peters, L.; König, G. M.; Terlau, H.; Wright, A. D. J. Nat. Prod. 2002, 65, 1633. (b) Okada, M.; Sato, I.; Cho, S. J.; Dubnau, D.; Sakagami, Y. Tetrahedron 2006, 62, 8907. (c) Rochfort, S. J.; Moore, S.; Craft, C.; Martin, N. H.; Van Wagoner, R. M.; Wright, J. L. C. J. Nat. Prod. 2009, 72, 1773.

⁽¹⁶⁾ One-pot alkylation of β -*N*-arylamido esters 7 with **15** in presence of KO'Bu were found inefficient. Therefore, IDC was carried out using C-alkylated ±16a-c.

⁽¹⁷⁾ For Pd-catalyzed asymmetric prenylations and geranylations, using carbonates as external electrophiles, see: Trost, B. M.; Malhotra, S.; Chan, W. H. J. Am. Chem. Soc. **2011**, *133*, 7328.

^{(18) (}a) Inuki, S.; Iwata, A.; Oishi, S.; Fujii, N.; Ohno, H. J. Org. Chem. 2011, 76, 2072. For a discussion on structure diversity, and biosynthetic gene of ergot alkaloids, see: (b) Wallwey, C.; Li, S. -M. Chem. Soc. Rev. 2011, 496.

⁽¹⁹⁾ For synthesis of 3-monosubstituted-2-oxindoles from IDC product of the type ±4, see: Pugh, D. S.; Klein, J. E. M. N.; Perry, A.; Taylor, R. J. K. *Synlett* **2010**, 934.

similar to that of tetracyclic 2-oxindoles **20** could serve as an advanced intermediate¹⁹ to access various structures of ergot alkaloids with a unified strategy. Compound **20** in turn could be accessed from 2-oxindole **21** via a direct biaryl coupling.

Remarkably, 2-bromo-pseudo benzylbromides also appeared to be good candidates for an electrophilic partner in a one-pot C-alkylation. A variety of 2-oxindoles (**21a** and **23a**–c),²⁰ having suitable bromo functionality for biaryl-coupling, could be easily accessed in high yields thus highlighting the usefulness of our methodology (Figure 4).



Figure 4. Further utility of 2-oxindoles.

With a synthetically viable route to the 2-oxindoles in hand, we were inclined to synthesize the advanced intermediate 20 featuring a pyridine annulated tetracyclic 2-oxindole. All efforts to optimize the biaryl-coupling of 23a as a model substrate to afford 24a using KO^tBu promoted homolytic aromatic substitution (HAS)²¹ led to multitude of products. However, on subjecting 23a to a Pd(0)-catalyzed direct arylations,²² we found that the direct arylation product 24a can be achieved in 82% yield (Scheme 5). Under optimized conditions, 23b and 21a also afforded the expected biaryl-coupling products without event. The X-ray crystal structures of 24b and 20a unambiguously proved the formation of ergoline type skeletons. Interestingly, we found that if the reaction was carried out at 170 °C, compound $\pm 23c$ afforded directly to the olefin 25 in 53% unoptimized yield in one-pot sequence (Scheme 6), which represents an unprecedented one-pot Pd(0)-catalyzed direct biaryl-coupling, activation of p-methoxybenzyl (PMB) ester followed by a decarboxylative depalladation

reaction presumably through the intermediacy of ± 26 and 27a-b.





Scheme 6. One-pot Synthesis of Tetracyclic 2-Oxindole 25



In summary, we report an efficient DDQ-mediated intramolecular-dehydrogenative-coupling (IDC) for the synthesis of a variety of 2-oxindoles bearing an all-carbon quaternary stereocenter at the pseudobenzylic position. The strategy involves a facile one-pot C-alkylation concomitant with oxidative coupling in the presence of stoichiometric DDQ. This study not only offers a vital method for the oxidative C–C bond construction but also clearly demonstrates the potential of the modular 2-oxindole scaffolds. Further exploration of this strategy as well as its application in the synthesis of complex alkaloids is currently under active investigation in our laboratory.

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Supporting Information Available. General experimental procedures and characterization of all new compounds, including CIF files of compounds **24b** and **20a**. This material is available free of charge via the Internet at http://pubs.acs.org.

⁽²⁰⁾ For use of 5-bromo-2-methoxy-6-picolylbromide **22**, see: Bisai, V.; Sarpong, R. *Org. Lett.* **2010**, *12*, 2551.

^{(21) (}a) Studer, A.; Bossart, M. In *Radicals in Organic Synthesis*; Renaud, P.; Sibi, M. P., Eds.; Wiley-VCH; Weinheim, 2001; Vol. 2, Chapter 1.4, pp 62–80. For reviews of homolytic aromatic substitution HAS with aryl radicals, see: (b) Bowman, W. R.; Storey, J. M. D. *Chem. Soc. Rev.* **2007**, 1803. (c) De, S.; Ghosh, S.; Bhunia, S.; Sheikh, J. A.; Bisai, A. *Org. Lett.* **2012**, *14*, 4466.

⁽²²⁾ For reviews on transition-metal catalyzed direct arylation, see: (a) Zeni, G.; Larock, R. C. *Chem. Rev.* **2006**, *106*, 4644. (b) Alberico, D.; Scott, M. E.; Lautens, M. *Chem. Rev.* **2007**, *107*, 174. (c) Cacchi, S.; Fabrizi, G. *Chem. Rev.* **2011**, *111*, PR215 and references therein.

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