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Tetrahydrofluorenones with conformationally restricted side chains as selective estrogen receptor beta ligands

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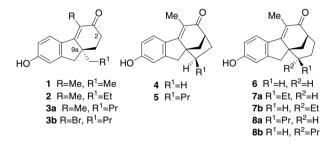
Abstract—A series of 2–9a bridged tetrahydrofluorenone derivatives were prepared which exhibited significant binding affinity for ER β and were highly selective.

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The estrogen receptor (ER) is a member of the superfamily of nuclear hormone receptors which act as ligand activated transcriptional factors.¹ The estrogen receptor was first cloned in 1986 and, at the time, was thought to be the exclusive receptor mediating physiological responses to estradiol.^{2,3} In 1996 a second estrogen receptor was discovered and identified as ER β ,^{4,5} the original cloned receptor being named ER α . It was found that ER β is widely expressed in a variety of tissues, including lung, prostate, ovarian granulosa cells, and brain, but is not the dominant ER expressed in the uterus or breast.^{6,7} The discovery of ER β and its differential tissue expression stimulated efforts both to define the relative physiological roles of ER α and ER β and to identify subtype selective ligands.

Both estrogen receptors (human) show substantial homology in the DNA binding domain (96%) and, to a lesser extent (58%), in the ligand binding domain (LBD).⁵ The ligand binding pocket differs in only two amino acids; the ER β binding pocket has a Met336 replacing a Leu384 in ER α and an Ile373 replacing a Met421.⁸ Given the subtle differences in the LBD it is not surprising that estradiol binds equally well to both receptors.

Several groups have disclosed efforts to design ERβ subtype selective ligands working from a variety of diverse structural platforms including tetrahydrochrysenes,⁹ 6*H*-benzo[*c*]chromen-6-ones,¹⁰ androstenediols,¹¹ diarylpropionitriles,¹² benzimidazoles,¹³ arylbenzothiophenes,¹⁴ arylbenzoxazines,¹⁵ triazines,¹⁶ biphenyls,^{17,18} aryl diphenolic azoles,¹⁹ and 2-phenyl-benzofurans.²⁰



Our effort in this area has focused on the tetrahydrofluorenone platform. We recently disclosed a new class of tetrahydrofluorenone compounds as potent ER β selective agonists (e.g. 1–3).²¹ Within this series the 9a alkyl substituent of these compounds was shown to be a key determinant of binding potency and selectivity. In addition, the 9a-(S) configuration of this substituent was found to be essential for good activity. In this report, we have explored the effect of conformationally restricting this key binding element by incorporating it into a constrained ring system. The derivatives prepared include the simple ethylene and propylene bridged compounds **4** and **6**. These can be viewed as conformation-

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ally restricted extensions of compounds 1 and 2. Building upon these two derivatives, pendant alkyl chains were incorporated in analogs 5, 7a, and 8a. The comparison of compounds 4 and 6 to the crystallographically determined conformation of 3b in complex with hER β^{21} (Fig. 1) indicated that the pendant alkyl chains would have to be attached exo to the bridged ring platform in order to map the substituents to the butyl of 3b.

The synthetic routes to racemic bridged analogs 4 and 6 are described in Scheme 1. With the exception of the first step, the routes to the two analogs are similar. The route to 4 started with the reductive alkylation of 5-methoxy indanone 9 with commercially available glycoaldehyde dimer to give the 2-alkylated indanone 10a. The route to 6 began from the alkylation of 9 with 3-bromotriethy-1silvloxypropane²² to give **10b**. Robinson annulation conditions employing ethyl vinyl ketone (EVK) with 10a and 10b, followed by cyclization under acidic conditions, and deacetylation of the resulting acetates gave the alcohols 11a and 11b. Mesylation of 11a and 11b, followed by their conversion to iodides and cyclization with LDA gave the products 12a and 12b. The final step involved demethylation of the aromatic methyl ethers and was accomplished with aluminum chloride and ethanethiol to give phenols 4 and 6.

As mentioned above, the preparation of the chiral ethylene bridged compound **5** bearing a pendant propyl chain was guided by the need to establish the two contiguous stereochemical centers. Thus, the stereochemistry of the pendant propyl chain was established at the start with the alkylation of (*S*)-3-propylbutyrolactone²³ with 2-methoxy benzylbromide to give **13** (see Scheme 2). Internal Friedel–Crafts alkylation of **13** in polyphosphoric acid gave indanone **14** in low yield. Michael addition of EVK in the Robinson annulation sequence gave diastereomers **15a** (21%) and **15b** (57%). The desired diastereomer **15a**²⁴ was processed following chemistry previously described to give **16**. Demethylation of **16** afforded the desired product **5**.

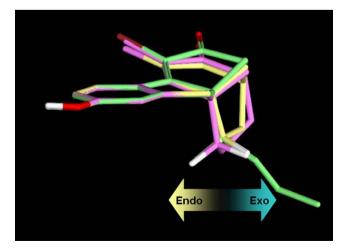
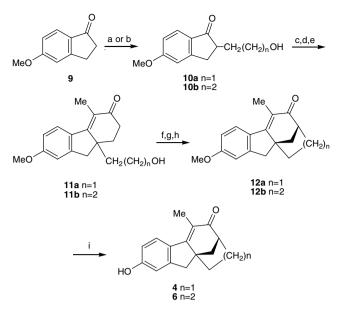
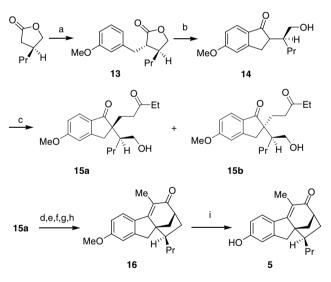


Figure 1. Superposition of the non-bridged 3b (green) as determined in a crystallographic complex with hER β and modeled bridged compounds 4 (yellow) and 6 (purple).



Scheme 1. Reagents and conditions: (a) glycoaldehyde dimer, NaOMe, H₂, 10% Pd/C, MeOH, rt; (b) 1-bromo-3-triethylsilyloxypropane, NaH, DMF, rt; (c) EVK, NaOMe, MeOH, rt; (d) 6 N HCl, HOAc, 80 °C; (e) NaOMe, MeOH, rt; (f) MsCl, Et₃N, CH₂Cl₂, 0 °C; (g) NaI, acetone, reflux; (h) LDA, THF, -78 °C to rt; (i) AlCl₃, EtSH, CH₂Cl₂, rt.



Scheme 2. Reagents and conditions: (a) LDA, 3-methoxybenzyl bromide, THF, -78 °C; (b) PPA, 100 °C; (c) EVK, NaOMe, MeOH, 60 °C; (d) 6 N HCl, HOAc, 80 °C; (e) 6 N HCl, MeOH, rt; (f) MsCl, Et₃N, CH₂Cl₂, 0 °C; (g) NaI, acetone, reflux; (h) LDA, THF, -78 °C to rt; (i) AlCl₃, EtSH, CH₂Cl₂, rt.

The relative stereochemical assignment of **5** was made on the basis of ¹H NMR²⁵ utilizing COSY and NOESY experiments (see Fig. 2 for key NOESY interactions).²⁶

The racemic propylene analogs **7a**, **7b**, **8a**, and **8b** were prepared as described in Scheme 3 following chemistry similar to that previously described. Michael addition of the appropriate acrylate to indanone **9** gave the alkylated indanones **16a** and **16b**. Reduction of the esters

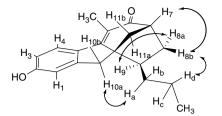
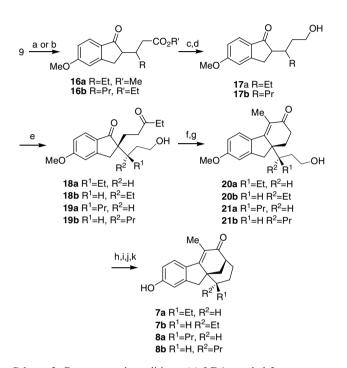


Figure 2. Key ¹H NOESY interactions observed.



Scheme 3. Reagents and conditions: (a) LDA, methyl 2-pentenoate, THF, -78 °C; (b) LDA, ethyl 2-hexenoate, THF, -78 °C; (c) LAH, THF, 0 °C; (d) Jones reagent, acetone, -78 °C; (e) EVK, NaOMe, MeOH, 60 °C; (f) 6 N HCl, HOAc, 80 °C; (g) 6 N HCl, MeOH, rt; (h) MsCl, Et₃N, CH₂Cl₂, 0 °C; (i) NaI, acetone, reflux; (j) LDA, THF, -78 °C; (k) AlCl₃, EtSH, CH₂Cl₂, rt.

16a and 16b also resulted in the reduction of the indanone ketone to give diols which were reoxidized to provide the requisite indanones 17a and 17b. Robinson annulation conditions using EVK with indanone 17a gave diastereomers 18a (31%) and 18b (44%). Similar reaction of 17b gave diastereomers 19a (21%) and 19b (52%). Cyclization of the alcohols 18a, 18b, 19a, and 19b gave the tetrahydrofluorenone acetates which were deacetylated to give the alcohols 20a, 20b, 21a, and 21b, respectively. Following established chemistry these intermediates were converted to 7a and 7b, and 8a and 8b.

The ER binding results of analogs prepared are shown in Table 1. Comparison of the ethylene bridged analog 4 to the unbridged ethyl analog 1 shows at least equivalent or slightly better ER β binding with an improvement in selectivity. Comparison of the propylene bridged analog 6 to its unbridged equivalent 2 shows equal ER β binding and selectivity. Both the ethylene 4 and propylene 6 bridged analogs have similar ER β binding, although 4 shows better selectivity.

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Compound	Human ERα IC ₅₀ (nM)	Human ERβ IC ₅₀ (nM)	Selectivity ERα/ERβ
1	1210	28	43
2	455	14	33
3a ^a	567	19	30
4	793	12	66
5 ^a	97	1	97
6	414	11	37
7a	146	5	29
7b	310	45	7
8a	84	5	16
8b	605	141	4

^a Chiral, all others racemates.

Analog 5, which incorporates a pendant propyl chain in the ethylene bridge, shows significantly more potency in ER β binding and more selectivity than its parent 4. Moreover, compound 5 with its conformationally restricted side chain shows a greater than 10-fold enhancement in ER β potency and more than a doubling of selectivity relative to its unconstrained analog 3a.

The binding results shown in Table 1 for the pairs of compounds **7a**, **7b**, and **8a**, **8b** clearly confirm the modeling prediction of a preference for exo substitution on the bridge and show the significant effect that orientation of the pendant chain has on binding affinity towards both ER α and ER β . The more active propylene bridged analogs **7a** and **8a**, incorporating either a pendant ethyl or propyl chain in the exo orientation, showed equal ER β binding, although the selectivity of **7a** was approximately twice that of **8a**. Both of these analogs were less potent in ER β binding and significantly less selective than the ethylene analog **5** with a pendant propyl chain.

As has been described previously for the tetrahydrofluorenone lead class,²¹ we believe that the ER β selectivity for the bridged variant arises principally from two sources. The first is the planar nature of the tetrahydrofluorenone core which is maintained in the bridged system. This putative stabilizing interaction between the planar/aromatic surface of the tricyclic platform and Met336 of hER β is not possible with the analogous Leu384 in hER α (see Fig. 3). This feature appears to be common to other planar ER β selective molecules including some phytoestrogens such as genistein²⁸ or other synthetic molecules including some benzisoxazoles,¹⁹ for example.

The second selectivity determinant postulated for the tetrahydrofluorenone class is a favorable hydrophobic interaction of the 9a-alkyl substituent which protrudes orthogonally from the plane of the tricyclic core toward Ile373 in hER β as depicted in Figure 3. We speculate that Ile373 in hER β can nicely accommodate the presence of the alkyl substituent into space which is not available in hER α because the side chain of the analogous Met421 fills this space. Fixing the 9a-alkyl substituent into the proper orientation as in compound **5** may enhance this favorable hydrophobic

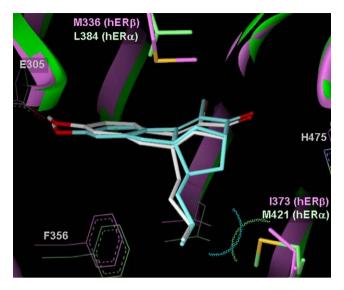


Figure 3. Superposition of the docked model of compound 5 (cyan) with the crystallographic complexes of compound 3b (white) with hER β (purple) and hER α (green) (pdb entry: 1ERE). Unless otherwise indicated, residue numbering is that of hER β .

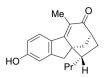
interaction and thus account for its increased ER β potency and selectivity relative to the corresponding unconstrained analog **3a**. The reduced potency and selectivity of the propylene bridged analog **8a** may be a consequence of its greater conformational flexibility and/or a less favorable orientation of the pendant propyl group.

In summary, we have designed a series of conformationally constrained 2–9a bridged tetrahydrofluorenones and incorporated pendant alkyl chains to optimize ER β binding affinity and selectivity. The ethylene bridged analog **5**, which conformationally constrains the butyl side chain of **3a**, exhibits improved ER β affinity by an order of magnitude while increasing subtype selectivity.

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- 24. Stereochemistry determined retrospectively from 5.
- 25. ¹H NMR (benzene- d_6): δ 0.65–0.72 (1H, m, H_b), 0.70 (3H, t, J = 7.1 Hz, Me), 0.80 (1H, m, H_c), 1.05 (1H, m, H_d), 1.21 (1H, m, H_a), 1.40 (1H, ddd, J = 4.8, 7.1, 13.3 Hz, H_{8b}), 1.48 (1H, dd, J = 4.3, 11.1 Hz, H_{11b}), 1.55–1.60 (1H, m, H₉), 1.59 (1H, d, J = 11.1 Hz, H_{11a}), 1.77 (1H, dd, J = 9.7, 13.3 Hz, H_{8a}), 2.18 (3H, s, Me), 2.49 (1H, d, J = 17.0 Hz, H_{10b}), 2.79 (1H, d, J = 17.0 Hz, H_{10a}), 3.00 (1H, dd, J = 4.3, 7.1 Hz, H₇), 6.54 (1H, d, J = 8.4 Hz, H₃), 6.59 (1H, s, H₁), 7.43 (1H, d, J = 8.4 Hz, H₄).
- 26. Compound **15b** was also converted by Scheme 2 to give the diastereomer shown below:



A ¹H NOESY experiment similar to that done for **5** showed the propyl function to be endo.

- 27. The IC50 values were generated in an estrogen receptor ligand binding assay. This scintillation proximity assay was conducted in NEN basic flash plates using tritiated estradiol and full length recombinant human ER α and ER β proteins, with incubation times of 3–22 h. Most compounds are single point determinations.
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