

Phenanthrene Synthesis via Chromium-Catalyzed Annulation of 2-Biaryl Grignard Reagents and Alkynes

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S Supporting Information

ABSTRACT: A chromium/2,2'-bipyridine-catalyzed annulation reaction of 2-biarylmagnesium reagents with alkynes is reported. The reaction is applicable to a variety of aryl- and/or alkyl-substituted internal alkynes as well as 2-biaryl and related Grignard reagents, thus affording phenanthrene derivatives in moderate to good yields. The reaction proceeds at the expense of excess alkyne as a hydrogen acceptor and thus does not need



an external oxidant. Deuterium-labeling experiments shed light on the reaction mechanism, which likely involves multiple intramolecular C-H activation processes on chromium.

henanthrene represents one of the simplest structural elements in materials science based on polycyclic aromatic

Scheme 1. Annulation of 2-Functionalized Biaryl and Alkyne to Phenanthrene via C-H Activation

(a) Annulation of 2-functionalized biaryl and alkyne to phenanthrene







hydrocarbons (PAHs).¹ The selective synthesis of substituted phenanthrenes would be not only useful for the preparation of phenanthrene-based materials² but also promising, when combined with other C-C coupling methods such as the Scholl reaction, for the bottom-up construction of extended polyaromatic systems.³ Among various approaches to phenanthrene synthesis, transition-metal-catalyzed annulation of a 2-

Table 1. Optimization of Reaction Conditions^a



^aReaction conditions: 1a (prepared from 2-bromobiphenyl and Mg/ LiCl; 0.2 mmol in THF), 2a (0.4 mmol), CrCl₂ (10 mol %), ligand (20 mol %), THF, 80 °C, 18 h. ^bDetermined by GC using *n*-tridecane as an internal standard. Isolated yield is shown in parentheses. ^cNorbornene (0.4 mmol) was added. ^d2,3-Dichlorobutane (0.5 mmol) was added. ^eThe reaction was performed at room temperature.

functionalized biaryl and an alkyne via C-H activation is attractive in terms of atom and step economy (Scheme 1a).^{4–6} Since the seminal works of Heck and Larock on the palladiumcatalyzed annulation of 2-iodobiaryl,^{4a,b} annulation reactions employing different biaryl substrates have been achieved under palladium,^{4c} iron,^{4d} iridium,^{4e} or rhodium^{4f} catalysis. Among them, Nakamura's iron-catalyzed annulation of 2-biaryl Grignard reagents using 1,2-dichloroisobutane as an oxidant

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Scheme 2. Scope of Alkynes



"Bphen was used instead of bpy. ^bPhenyl(cyclopropyl)acetylene was used as the reactant, and an inseparable mixture of **3al** and **3al**' was obtained.

is notable for the low cost of the catalyst and the mild reaction temperature.^{4d} This reaction, however, requires 2 equiv of the Grignard reagent, which can be more precious than the alkyne, because 1 equiv is sacrificially consumed as a hydrogen acceptor. Here, we report on a chromium-catalyzed annulation reaction of a 2-biaryl Grignard reagent and an alkyne to form a phenanthrene derivative. In contrast to iron catalysis, the present reaction proceeds at the expense of excess alkyne as a hydrogen acceptor.

The present study was prompted by our recent finding of a chromium-catalyzed addition reaction of an aryl Grignard reagent to a dialkylalkyne to afford an *ortho*-alkenylarylmagnesium species via a 1,4-chromium migration (Scheme 1b).^{7–10} During this study, we observed a small amount of 1,2,3,4-tetraalkylnaphthalene, which appeared to be formed via dehydrogenative annulation of the styrenylchromium or *ortho*-alkenylarylchromium species with another molecule of the alkyne via $C(sp^2)$ -H activation. Building on this observation, and in light of recent progress in chromium catalysis,^{11–13} we envisioned that the 2-biaryl Grignard reagent could undergo annulation with an alkyne to form a phenanthrene derivative.





^{*a*}Obtained as a mixture with the defluorinated product (i.e., **3aa**; 11%).

At the outset, we explored the reaction between the 2biphenylmagnesium bromide–lithium chloride complex (1a), prepared from 2-bromobiphenyl and Mg·LiCl,¹⁴ and diphenylacetylene (2a, 2 equiv) (Table 1). In the presence of $CrCl_2$ (10 mol %) in THF at 80 °C, the reaction afforded 9,10diphenylphenanthrene (3aa) in 19% GC yield (entry 1). Upon ligand screening, 2,2'-bipyridine (bpy) was found to dramatically improve the yield of 3aa to 80% (78% isolated yield; entry 2). 1,10-Phenanthroline and bathophenanthroline (bphen) also promoted the reaction in somewhat lower yields (entries 3 and 4), while terpyridine and diphosphine ligands such as dppe were much less effective (entries 5 and 6).

GCMS analysis of the $CrCl_2$ /bpy-catalyzed reaction indicated the formation of (*E*)-stilbene (4) and 1,2,3triphenylnapthalene (5) as byproducts. While their origin was probed later (vide infra), 4 and 5 appeared to be formed by reduction of 2a and dimerization of 2a, respectively. The consumption of 2a could not be suppressed by the addition of norbornene as a hydrogen acceptor (entry 7). Unlike the ironcatalyzed annulation,^{4d} a dichloroalkane oxidant was only detrimental (entry 8), causing substantial homocoupling of 1a. Note that the present reaction took place even at room temperature, albeit with lower conversion and yield (entry 9).

With the $CrCl_2$ /bpy system, the reaction of **1a** with different alkynes was explored (Scheme 2). A variety of diarylalkynes participated in the annulation to afford 9,10-diarylphenanthrenes **3ab–3ah** in moderate to good yields. The reaction became somewhat sluggish with CF_3 groups on the *para*position (see **3ae** and **3af**). Di(2-thienyl)acetylene also afforded the desired product **3ai** in 49% yield. Aryl(alkyl)alkynes also proved to be competent substrates, for which better yields were achieved using bphen instead of bpy. Interestingly, the reaction of phenyl(cyclopropyl)acetylene was accompanied by a partial

Scheme 4. Deuterium-Labeling Experiments^a



^aThe yield was determined by GC for Scheme 5d. ¹H NMR integrations lower than that expected for nondeuterated compounds are indicated.

cleavage of the cyclopropyl ring, affording a small amount of 9phenyl-10-propenylphenanthrene **3al'** along with the expected product **3al**. Dialkylalkynes such as 4-octyne and 4-methylpent-2-yne also afforded the corresponding phenanthrene products **3ap** and **3aq**. Phenylacetylene failed to give the desired phenanthrene but underwent dimerization, as judged from GCMS analysis of the crude product.

Next, we subjected different 2-biaryl- and related Grignard reagents to the annulation with 2a (Scheme 3a). 4-Methoxyphenyl- and 4-fluorophenyl-substituted Grignard reagents afforded the products 3ba and 3ca, respectively, while the latter was accompanied by partial hydrodefluorination. 3-Tolyl-substituted Grignard reagent underwent regioselective annulation at the less hindered *ortho* position (3da). The steric hindrance of 2-tolyl- and 1-naphthyl-substituted Grignard reagents could be tolerated (3ea and 3fa). 2-Thienyl- and 2alkenyl-substituted Grignard reagents also afforded the desired products 3ga and 3ha in moderate to good yields. The reaction between unsymmetrical Grignard reagent 1d and 1-phenyl-1propyne (2j) afforded regioisomers 3dj and 3dj' in a ca. 1:1 ratio, demonstrating the lack of regioselectivity in the alkyne insertion process (Scheme 3b).

To gain insight into the reaction mechanism and the origin of the byproducts **4** and **5**, we performed a series of deuteriumScheme 5. Proposed Reaction Pathways



labeling experiments. First, the reaction between Grignard reagent **1b** and **2a**, when quenched with D_2O , afforded phenanthrene **3ba**, **4**, and **5**, with partial deuteration of the *ortho* and olefinic positions of **4** (Scheme 4a). Second, the reaction using pentadeuterated Grignard reagent $1a-d_5$ demonstrated partial deuterium transfer to the *ortho* and olefinic positions of **4** and the 4- and 5-positions of **5** (Scheme 4b). Third, the reaction between Grignard reagent **1d** and decadeuterated diphenylacetylene (**2a**- d_{10}) furnished the phenanthrene **3da**- d_{10} with slight deuteration of the less hindered bay region (Scheme 4c). Finally, the treatment of **1d** with a mixture of CrCl₂ (30 mol %) and bpy (60 mol %), upon quenching with D_2O , resulted in partial deuteration of the *ortho* positions of both the aryl rings (Scheme 4d).

To rationalize the lack of the regioselectivity in alkyne insertion (Scheme 3b) and the results of the deuterium-labeling experiments (Scheme 4), we propose reaction pathways (for 1a and 2a) shown in Scheme 5 (see the Supporting Information for further details). A 2-biarylchromium species A would undergo intramolecular C-H oxidative addition to give a metallafluorene hydride B. Migratory insertion of 2a into the Cr-H bond of B would generate an alkenyl-metallafluorene C, followed by isomerization to the *trans* isomer D presumably via a zwitterionic carbene intermediate.¹⁵ The intermediate D may undergo insertion of another molecule of 2a into the biaryl-Cr bond (G), followed by reductive elimination to afford 3aa along with a stilbenyl-Cr species H.6a Transmetalation between H and 1a would regenerate A and afford stilbenyl-Mg species (RMgX). As judged from the H/D scrambling results, D may also undergo sequential σ -bond metathesis via a metallaindene species E, causing migration of the chromium atom on the stilbenyl ligand. The resulting intermediate F would also lead to 3aa via alkyne insertion/ reductive elimination. The annulation of metallaindene species E or the stilbenyl-Cr species H with 2a would be responsible for the formation of 5. The role of the bipyridine ligand

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remains unclear, while we suspect that it facilitates the reductive elimination toward selective formation of **3aa**.

In summary, we have developed a chromium-catalyzed annulation reaction of 2-biarylmagnesium bromides and related Grignard reagents with internal alkynes to form phenanthrene derivatives. The reaction is mechanistically unique for the role of excess alkyne as a hydrogen acceptor and the involvement of multiple C–H activation processes. Further investigation into chromium-catalyzed C–H activation/C–C bond formation is underway.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.or-glett.7b03342.

Detailed experimental procedures and spectral data (PDF)

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Notes

The authors declare no competing financial interest.

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