

Available online at www.sciencedirect.com



Bioorganic & Medicinal Chemistry Letters

Bioorganic & Medicinal Chemistry Letters 17 (2007) 3595-3598

Design, synthesis, and structure–activity relationship of carbamate-tethered aryl propanoic acids as novel PPAR α/γ dual agonists

Nam-Jung Kim,^a Kwang-Ok Lee,^a Bon-Woong Koo,^a Funan Li,^a Ja-Kyung Yoo,^b Hyun-Ju Park,^b Kyung-Hoon Min,^b Joong In Lim,^c Mi Kyung Kim,^c Jin-Kwan Kim^c and Young-Ger Suh^{a,*}

^aCollege of Pharmacy, Seoul National University, San 56-1, Sillim-dong, Gwanak-gu, Seoul 151-742, Republic of Korea ^bCollege of Pharmacy, Sungkyunkwan University, Suwon-si, Gyeonggi-do 440-746, Republic of Korea ^cDong-a Pharm.Co., Ltd, 47-5, Sanggal-dong, Giheung-gu, Yongin-si, Gyeonggi-do 449-905, Republic of Korea

> Received 28 February 2007; revised 16 April 2007; accepted 18 April 2007 Available online 25 April 2007

Abstract—We have developed a new class of PPAR α/γ dual agonists, which show excellent agonistic activity in PPAR α/γ transactivation assay. In particular, (*R*)-9d was identified as a potent PPAR α/γ dual agonist with EC₅₀s of 0.377 μ M in PPAR α and 0.136 μ M in PPAR γ , respectively. Interestingly, the structure–activity relationship revealed that the stereochemistry of the identified PPAR α/γ dual agonists significantly affects their agonistic activities in PPAR α than in PPAR γ . © 2007 Elsevier Ltd. All rights reserved.

As members of the nuclear hormone receptor super family of ligand-activated transcription factors, the Peroxisome proliferator-activated receptors (PPARs). transduce a wide variety of signals to regulate glucose and lipid homeostasis, sensitivity to insulin, cell proliferation/differentiation, and innate and adaptive immune responses. Up to date, three isotypes, which are designated PPARa, PPARy, and PPARo, have been identified and cloned.¹ Recently, the rapid progress in functional analysis of these receptors prompted the scientific and pharmaceutical attention to PPARs due to their biological importance. In particular, the PPAR agonists have intensively been studied as a therapeutic target of type 2 diabetes. As rosiglitazone is a known PPAR γ agonist as one of the representative insulin sensitizing thiazolidinediones (Fig. 1),² activation of PPAR γ induces adipocyte differentiation, lipid uptake, and insulin sensitization, while PPARa regulates lipid homeostasis via control of fatty acid catabolism.³ The PPARα agonists such as fibrate analogs have been used to treat dyslipidemia by lowering the free triglyceride

Keywords: Type 2 diabetes; PPARs; PPARa/y dual agonists.

* Corresponding author. Tel.: +82 2 880 7875; fax: +82 2 888 0649; e-mail: ygsuh@snu.ac.kr



Figure 1. PPAR γ agonist (rosiglitazone) and PPAR α/γ dual agonist (muraglitazar).

(TG) plasma concentration.^{4,5} Considering the significantly increased risk of complications related to lipid catabolism among the patients with type 2 diabetes, it has been postulated that PPAR α/γ dual agonists, such as muraglitazar, might provide outstanding therapeutic effects for the treatment of type 2 diabetes and dyslipidemia (Fig. 1).⁶ We herein report identification of the

⁰⁹⁶⁰⁻⁸⁹⁴X/\$ - see front matter @ 2007 Elsevier Ltd. All rights reserved. doi:10.1016/j.bmcl.2007.04.057

novel and potent PPAR α/γ dual agonists as well as their structure–activity relationship.

In order to efficiently develop the novel PPAR α/γ dual agonists, we initially attempted virtual screening of our chemical library for PPAR γ (PDB number: 1K74) and then identified the compound 1 with weak PPAR γ agonistic activity through a cell based transactivation assay of the virtual hits. In addition, replacement of benzoic acid moiety of 1 with β -aryl α -(S)-ethoxy propanoic acid, a well-known moiety of high binding affinity to PPARs,⁷ has led us to discover the lead compound 9a (Fig. 2) as a potent PPAR α/γ dual agonist. Thus, we commenced our studies on development of novel PPAR α/γ dual agonists based on the structure of **9a** as well as their structure-activity relationship. The novel series of 9a consisting of hydrogen bonding donor/ acceptor (A), linker (B), and hydrophobic part (C), 8 which are the typical functional regions of the known PPAR α/γ dual agonists, were designed and synthesized for the detail investigation of each region.

The compounds **9a–h**, **10**, **11** were conveniently synthesized as outlined in Scheme 1. The key intermediate **6**



Figure 2. Conversion of hit 1 to lead compound 9a.

was synthesized from the commercially available 4hydroxybenzaldehyde 2. Sequential O-benzylation of phenol and Wittig olefination of aldehyde, followed by hydrogenation of the resulting olefin, provided 3, which was readily converted to the acid 4 by benzylation and ester hydrolysis. The racemic 4 was transformed into the optically pure ester 5 via a sequence of (R)-2-phenylglycinol coupling, separation of the resulting two diastereomers, and amide hydrolysis, followed by ethyl ester formation. The absolute configuration of the ester 5 was confirmed by comparison of its optical rotation with that of the known compound.⁹ Triflation of 5, Sonogashira coupling of the resulting triflate 6 with the racemic or optically pure propargyl alcohol 7,¹⁰ and then hydrogenation of the triple bond afforded the corresponding optically active or racemic intermediate 8 in optically pure form or as a mixture of two diastereomers, respectively. The analogs 9a-h, 11 were prepared by coupling of the corresponding intermediate 8 with the appropriate isothiocyanate or isocyanate, followed by ester hydrolysis, while the acid 10 was prepared by direct hydrolysis of the hydroxyl ester 8. Our design and synthesis of the analogs of 9a was focused on the thiocarbamate analogs rather than the carbamate because our preliminary studies revealed that the thiocarbamates are much more active than the carbamates for PPAR α/γ as shown in the activities of 9a and 11 of Table 1. The optically active thiocarbamates could also be prepared by employing (S)- or (R)-7 by the same procedure. The analog 14 was synthesized from propargyl alcohol by analogy to 9a-h as depicted in Scheme 2. The thiocarbamate 18 was synthesized from commercially available racemic 1-phenyl-1, 2-ethanediol (15). Selective tosylation of the primary alcohol,¹¹ TBS protection of the remaining secondary alcohol, and then alkylation



Scheme 1. Reagents and conditions: (a) benzylchloride, K_2CO_3 , DMF, 80 °C, 93%; (b) (1,2-diethoxy-2-oxyethyl)triphenylphosphoniumchloride, tetramethylguanidine, CH_2Cl_2 , 0 °C to rt, 93%; (c) H_2 , Pd/C, MeOH, 95%; (d) benzylchloride, K_2CO_3 , DMF, 120 °C, 75%; (e) LiOH·H₂O, MeOH/THF/H₂O, 94%; (f) EDCI, HOBT, DIEA, CH₂Cl₂, then (*R*)-phenylglycinol, 0 °C to rt, 30% (desired diastereomer); (g) concd-H₂SO₄, H₂O/1,4-dioxane, reflux; (h) TMSCl, EtOH, 45% for two steps; (i) Tf₂O, Et₃N, CH₂Cl₂, 0 °C to rt, 92%; (j) (*S*)- or (*R*)-7 for **9a**, **9c**, and **9d**, and **7** for **9b**, **9e–h**, **10**, **11**, Pd(PPh₃)₂Cl₂, CuI, Et₃N, TBAI, DMF, 50 °C, 83%; (k) H₂, Pd/C, MeOH, 100%; (l) NaH, alkyl or arylisothiocyanate, THF; (m) BF₃·OEt₂, benzylisocyanate, toluene; (n) LiOH·H₂O, MeOH/THF/H₂O.

Table 1. In vitro activities of the synthesized compounds in cell-based GAL4-PPAR α/γ transactivation assay

Compound ^a	Human PPARα		Human PPARγ	
	EC ₅₀ (μM)	Efficacy ^b (%)	EC ₅₀ (μM)	Efficacy ^b (%)
Rosiglitazone	3.46	56	0.03	100
Gemfibrozil	193.30	100	147.80	79
(R)-9a	0.26	101	0.35	113
(S)-9a	0.07	234	0.58	159
9b	5.25	154	4.58	204
(R)-9c	2.77	112	1.05	190
(S)-9c	0.88	120	0.49	177
(R)-9d	0.38	71	0.14	197
(<i>S</i>)-9d	3.27	66	0.16	187
9e	20.40	130	5.39	182
9f	3.94	128	3.48	152
9g	7.26	104	3.60	118
9h	10.70	117	1.06	145
10	1.51	108	0.30	119
11	6.05	164	1.42	160
14	1.47	118	1.40	117
18	6.21	136	7.17	138

^a (*R*) or (*S*)-9a, 9c and 9d as well as 14 are optically active, while 9b, 9eh, 10, 11, and 18 are 1:1 diastereomeric mixtures.

^b The relative maximum efficacy to the percentage of the standards.

of the intermediate **5** with the resulting tosylate **16** gave the ether **17** after desilylation. Reaction of **17** with benzylisothiocyanate and ester hydrolysis provided **18**.

The in vitro activities of the synthesized analogs are summarized in Table 1.¹² Generally, the aryl thiocarbamates (**9a**, **9c**, and **9d**) exhibited higher agonistic activity than the alkyl thiocarbamates (**9e–h**) in both PPAR α and PPAR γ transactivation assays. In particular, the analog **9a** possessing benzyl thiocarbamate moiety displayed potent activities for PPAR α/γ regardless of the stereochemistry. However, it decreased as the chain length of C-region decreased (**9b**) or increased (**9c**). Interestingly, the similar trend was observed for the alkyl thiocarbamates and the propyl turned out to be a linker of optimal length. It is noteworthy that (*R*)-**9d** possessing the bulkiest substituent in the lipophilic moiety (C-region) showed the most potent activity for PPAR γ with EC₅₀ of 0.14 μ M. These results might imply that the bulky substituents in the lipophilic region are beneficial for the high PPAR γ activation and the similar results were observed for the alkyl thiocarbamates as the analog 9h with cyclohexylmethyl substituent exhibited the most potent activity for PPAR γ . Our preliminary work revealed that the thiocarbamates are much more active than the carbamates for PPAR α/γ , as it was confirmed by the activities of 9a (EC₅₀: 0.26-0.58 µM) and 11 (EC₅₀: 1.42–6.50 µM). Replacement of the carbon linker with an ether linkage decreased the activity as shown in the activities of 9a and 18. Importantly, the agonistic activity of the thiocarbamate analogs sensitively changes according to the stereochemistry at the benzylic carbon, particularly for PPAR α . The (S)-isomer of 9a was more potent than the corresponding (R)-isomer for PPAR α , while both exhibited the similar activity for PPAR γ . The (S)-isomer of **9c** is slightly more potent than the (R)-isomer of **9c** in both PPAR α and γ . In case of the compound **9d** showing the most potent activity for PPAR γ , the (R)-isomer of 9d was 10-fold more potent than the (S)-isomer for PPARα. The docking studies with Surflex-DockTM,¹³ shown in Fig. 3, provided a rationale in part for the stereochemistry/substituent-dependent selectivity for the PPAR isotypes. The benzylthiocarbamoyl moiety of (S)-9a seems to fit well to the lipophilic region of the active site of PPAR α , while the corresponding moiety of (R)-9a orients toward the hydrophilic region instead of the lipophilic region, which is enforced by the stereochemistry. In addition, the calculated binding energies revealed the lower binding energy for (S)-9a compared to that for (R)-9a. However, *p-tert*-butyl benzyl group of 9d appears to be too bulky to fit to the lipophilic pocket of PPAR α in both (R)- and (S)-isomers and subsequently the bulky lipophilic moiety to orient toward the spacious hydrophilic pocket. As the results, the activities of (R)-9a, (S)-9d, and (R)-9d seem to be reduced by the inadequate orientations of their lipophilic moiety. Removal of the phenyl moiety of the B-region of 9a significantly drops the activity for both PPAR α and PPAR γ as shown in the activities of 14. This would be understood in terms of formation of the favorable conformation for the high activity, which is induced



Scheme 2. Reagents and conditions: (a) 6, $Pd(PPh_3)_2Cl_2$, CuI, Et₃N, TBAI, DMF, 60 °C, 45%; (b) H₂, Pd/C, MeOH, 76 %; (c) NaH, benzylisothiocyanate, THF, 0 °C to rt; (d) LiOH·H₂O, MeOH/THF/H₂O, 65% for two steps; (e) Bu₂SnO, *p*-toluenesulfonylchloride, Et₃N, CH₂Cl₂, 0 °C to rt, 94%; (f) TBSCl, imidazole, DMAP, DMF, 65 °C, 85%; (g) 5, K₂CO₃, DMF, 130 °C, (h) 2 N-HCl, THF, 50 °C, 40% for two steps; (i) NaH, benzylisothiocyanate, THF; (j) LiOH·H₂O, MeOH/THF/ H₂O, 60% for two steps.



Compound	(S)-9a	(R)-9a	<i>(S)</i> -9d	(R)-9d
Energy*	-8.86	-8.69	-7.94	-8.66

Figure 3. Docking model of (S)-9a (white), (R)-9a (magenta), (S)-9d (cyan), (R)-9d (orange), and the co-crystallized ligand as a reference molecule (yellow) with PPAR α (PDB No. 1K7L). LP, lipophilicity; dotted line, hydrogen bonding. *Note.* *Log value (kcal/mol).

by the phenyl substituent. On the other hand, elimination of the thiocarbamate moiety of the thiocarbamate **9a** significantly reduced the activity for PPAR α as shown in the activity of the alcohol **10** while the activity for PPAR γ remained. This result obviously supports that the thiocarbamate moiety is essential for the high activity of our series for PPAR α .

In summary, we have identified the novel and potent PPAR α/γ dual agonists and established SAR of the carbamate-tethered propanoic acids through the computational and synthetic chemistry. The stereochemistry/ substituent-dependent activation of the particular isotype of PPARs by our carbamate series was elucidated on the basis of the binding mode of the ligands in the active site. Further work for the development of the therapeutically useful PPAR α/γ dual agonists based on our current results is in good progress.

Acknowledgments

This research work was supported by the grant from R& D Center for Antidiabetic Drugs, by Ministry of Health & Welfare, Republic of Korea (A020600), and in part from Center for Bioactive Molecular Hybrids, Yonsei University.

References and notes

- 1. Dreyer, C.; Krey, G.; Keller, H.; Givel, F.; Helftenbein, G.; Wahli, W. Cell **1992**, 68, 879.
- Lehmann, J. M.; Moore, L. B.; Smith-Oliver, T. A.; Wilkinson, W. O.; Willson, T. M.; Kliewer, S. A. J. Biol. Chem. 1995, 270, 12953.
- 3. Staels, B.; Auwerx, J. Curr. Pharm. Des. 1997, 3, 1.
- 4. Chong, P. H.; Bachenheimer, B. S. Drugs 2000, 60, 55.
- 5. Milionis, H. J.; Elisaf, M. S.; Mikhailidis, D. P. Curr. Med. Res. Opin. 2000, 16, 21.
- Devasthale, P. V.; Chen, S.; Jeon, Y.; Qu, F.; Shao, C.; Wang, W.; Zhang, H.; Farrelly, D.; Golla, R.; Grover, G.; Harrity, T.; Ma, Z.; Moore, L.; Ren, J.; Seethala, R.; Cheng, L.; Sleph, P.; Sun, W.; Tieman, A.; Wetterau, J. R.; Doweyko, A.; Chandrasena, G.; Chang, S. Y.; Humphreys, W. G.; Sasseville, V. G.; Biller, S. A.; Ryono, D. E.; Selan, F.; Hariharan, N.; Cheng, P. T. W. J. Med. Chem. 2005, 48, 2248.
- Buckle, D. R.; Cantello, B. C. C.; Cawthorne, M. A.; Coyle, P. J.; Dean, D. K.; Faller, A.; Haigh, D.; Hindley, R. M.; Jefcott, L. J.; Lister, C. A.; Pinto, I. L.; Rami, H. K.; Smith, S. A. *Bioorg. Med. Chem. Lett.* **1996**, *6*, 2127.
- Wei, Z.; Petukhov, P. A.; Bizik, F.; Teixeira, J. C.; Mercola, M.; Volpe, E. A.; Glazer, R. I.; Willson, T. M.; Kozikowski, A. P. J. Am. Chem. Soc. 2004, 126, 16714.
- Ebdrup, S.; Pettersson, I.; Rasmussen, H. B.; Deussen, H.-J.; Jensen, A. F.; Mortensen, S. B.; Fleckner, J.; Pridal, L.; Nygaard, L.; Sauerberg, P. J. Med. Chem. 2003, 46, 1306.
- Powell, N. A.; Rychnovsky, S. D. Tetrahedron Lett. 1996, 37, 7901.
- Martinelli, M. J.; Vaidyanathan, R.; Pawlak, J. M.; Nayyar, N. K.; Dhokte, U. P.; Doeke, C. W.; Zollars, L. M. H.; Moher, E. D.; Khau, V. V.; Košmrlj, B. J. Am. Chem. Soc. 2002, 124, 3578.
- 12. In vitro transactivation assays: the ligand binding domains (LBD) of the human PPAR α/γ receptors were fused to the DNA binding domain (DBD) of the yeast transcription factor GAL4. CV-1 cells were transiently transfected with an expression vector for the respective PPAR chimera along with a reporter construct containing five copies of the GAL4DNA binding site and pRL-TK as a control vector (Promega). The test compounds were dissolved in DMSO and diluted 1:1000 in media. Cells were treated with the compounds at seven concentrations ranging from 0.03 to 100 μ M for 24 h followed by dual luciferase assay using Dual-Glo luciferase reagent (promega). EC₅₀ values were calculated by nonlinear regression using SigmaPlot 4.0 (SPSS).
- 13. The X-ray structures of PPARα and PPARγ complexed with the ligand were taken from the RCSB Protein Data Bank (PDB code: 1K7L for PPARα/1K74 for PPARγ). The protein was prepared for docking, using the protein preparation and refinement utility provided by Surflex-Dock™ suite in SYBYL 7.0. Water molecules of the crystal structure were removed from the complexes, and the hydrogen atoms were added computationally at appropriate positions. Calculations for the docking studies between the prepared PPARα protein and the selected compounds were performed using SYBYL 7.0 software.