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Synthesis and Absolute Configuration of Optically Active *E*-1-Alkoxymethoxybut-2-enyl(tri-n-butyl)stannanes: Stereoselective Reactions with Aldehydes

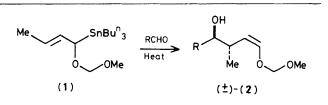
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(1R)- and (1S)-1-[(-)-menthoxymethoxy]-*E*-but-2-enyl(tri-n-butyl)stannanes (5) and (6), whose configurations at C(1) were assigned by correlation with (2R)- and (2S)-pentan-2-ol, react stereoselectively on heating with benzaldehyde to give (3S,4S)- and (3R,4R)-4-hydroxy-3-methyl-*cis*-1,2-enol ethers (11) and (13), respectively.

(\pm)-*E*-1-Methoxymethoxybut-2-enyl(tri-n-butyl)stannane (1) reacts stereoselectively with aldehydes on heating to give (\pm)-*anti*-4-hydroxy-3-methyl-*cis*-1,2-enol ethers (2).^{1†} We report here the preparation and characterization of optically active 1-[(-)-menthoxymethoxy]-*E*-but-2-enyl(tri-n-butyl)-stannanes (5) and (6) together with details of their reactions with aldehydes.

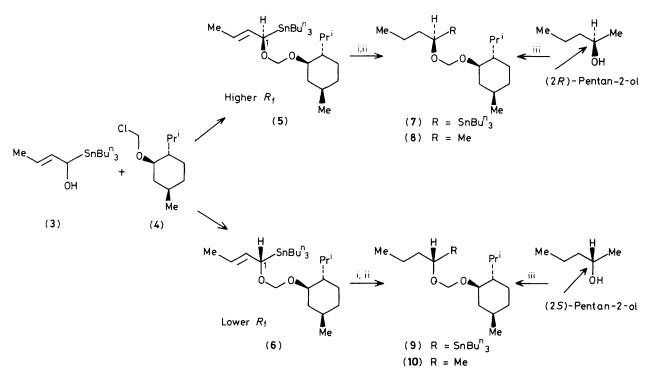
Chloromethyl (-)-menthyl ether (4)³ was prepared by bubbling anhydrous HCl through a solution of (-)-menthol and trioxane in CH₂Cl₂ in the presence of anhydrous MgSO₄ (b.p. 82—84 °C at 0.8 mm Hg, 54%), and was added to a mixture of racemic stannol (3)¹ and di-isopropylethylamine in CH₂Cl₂ to give, after 2 h at 0 °C and 2 h at 20 °C, a mixture of the diastereoisomeric but-2-enylstannanes (5) and (6), isol-



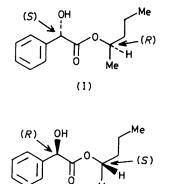
ated as a mixture by flash chromatography (59%). Careful short-column chromatography (eluted with 40/60 petroleumbenzene, 5:1) gave the (1*R*)-diastereoisomer (5), $[\alpha]_D^{20}$ +12.1° (CHCl₃), followed by the (1*S*)-diastereoisomer (6), $[\alpha]_D^{20}$ -91.1° (CHCl₃), which could be distinguished by t.l.c. and by high field (300 MHz) ¹H n.m.r. spectroscopy.‡

[†] For the *syn-anti* nomenclature see the discussion by S. Masamune, ref. 2.

[‡] New compounds were fully characterised spectroscopically and by combustion analysis whenever possible.

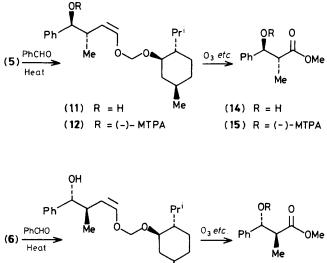


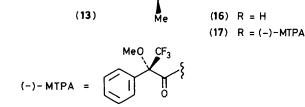
Scheme 1. Reagents: i, TsNHNH₂, NaOAc, EtOH; ii, BuⁿLi, tetrahydrofuran, -78 °C, 5 min, then Me₂SO₄, -78 °C, 1³/₄ h; iii, (4), Prⁱ₂NEt, 0 °C, 2 h, then 20 °C, 2 h.



′ Me (II)

Absolute configurations at C(1) were assigned to stannanes (5) and (6) by correlation with (2*R*)- and (2*S*)-pentan-2-ol as shown in Scheme 1. Thus di-imide reduction of the higher R_f stannane, followed by transmetallation (BuⁿLi, -78 °C, 5 min) and methylation (Me₂SO₄, -78 °C, 1³/₄ h), a procedure known to involve retention of configuration,⁴ gave the (2*R*)-pentan-2-ol derivative (8) which was also prepared from authentic (2*R*)-pentan-2-ol.⁵§ Similarly the lower R_f stannane was converted into the (2*S*)-pentan-2-ol derivative (10) which was also prepared from (2*S*)-pentan-2-ol. The pentan-2-ol

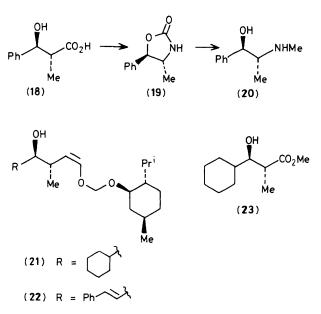




Scheme 2

derivatives (8) and (10) were clearly distinguished by high field (300 MHz) ¹H n.m.r. spectroscopy, and enabled configurational assignments to be made as shown.

^{§ (±)-}Pentan-2-ol was resolved following the procedure of J. K. Whitesell (ref. 6). Using (S)-mandelic acid, the ester (I) of (2R)-pentan-2-ol crystallized out which gave (2R)-pentan-2-ol, $[\alpha]_D^{20}$ – 13.5° (CHCl₃) (lit.⁵ – 16.1°), on hydrolysis. Ester (II) was obtained from (R)-mandelic acid.

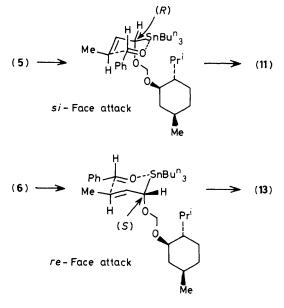


The menthoxymethoxystannanes (5) and (6) were then heated separately with an excess of benzaldehyde (130 °C, 15 h) under an argon atmosphere. The (1*R*)-isomer (5) gave the (3*S*,4*S*)-4-hydroxy-3-methyl-*cis*-1,2-enolether (11), whereas the (1*S*)-isomer (6) gave the (3*R*, 4*R*)-enol ether (13), both isolated in 70–80% yields after flash chromatography. The diastereoisomers (11) and (13) could be distinguished by ¹H n.m.r. spectroscopy, and examination of the crude reaction mixtures showed that no appreciable crossover had occurred, only the (3*S*,4*S*)-isomer (11) being obtained from stannane (5), and only the (3*R*,4*R*)-isomer (13) being obtained from stannane (6); see Scheme 2.

Ozonolysis followed by a dimethyl sulphide work up, oxidation (Ag₂O), and esterification (CH₂N₂) of the enol ethers (11) and (13), gave the enantiomeric hydroxy-esters (14) and (16), respectively (40% overall). The enantiomeric excess of each of these esters exceeded 90% as measured by optical rotation,⁷ and by conversion into their (-)- α -methoxy- α -(trifluoromethyl)phenylacetate [(-)-MTPA] derivatives (15) and (17).⁸ Prior conversion of the enol ether (11) into its (-)-MTPA ester (12) followed by ozonolysis, oxidation, and esterification gave ester (15) with an enantiomeric excess of >98% so showing that the small amount of racemization observed earlier had occurred during the ozonolysis or subsequent steps perhaps *via* reversible aldol equilibration.

Absolute configurations were assigned to the hydroxyesters by correlation with pseudoephedrine. Thus the (–)hydroxy-acid, readily available by cinchonidine resolution ⁷ of the (\pm)-acid,⁹ was found to have the absolute configuration shown in formula (**18**) since Schmidt rearrangement [(PhO)₂P(O)N₃] and reduction (LiAlH₄) gave (–)pseudoephedrine (**20**) whose absolute configuration is known.¹⁰ Esterification of the (–)-acid (**18**) with diazomethane gave the (–)-ester (**14**).

The selective transformations of benzaldehyde into *anti*-4hydroxy-3-methyl-*cis*-1,2-enol ethers (11) and (13) using the optically active α -alkoxybut-2-enylstannanes (5) and (6), are consistent with the cyclic transition states shown in Scheme 3. The marked preference for the α -alkoxy group to adopt an axial position in each of these transition states^{1,11} ensures that the (1*R*)- and (1*S*)-diastereoisomers (5) and (6) react selectively with the *si*- and *re*-faces of the benzaldehyde carbonyl group as shown.





Similar discrimination was observed with other aldehydes. Thus the (1*R*)-stannane (**5**) was heated with an excess of cyclohexanecarboxaldehyde and cinnamaldehyde to give adducts (**21**) (80%) and (**22**) (68%). The stereochemistry of these products was assigned by analogy with the benzaldehyde series, and was confirmed for the cyclohexanecarboxaldehyde adduct (**21**) by ozonolysis, oxidation, and esterification, which gave the (–)-hydroxy ester (**23**), $[\alpha]_D^{20}$ –8.1° (CHCl₃), the absolute configuration of which has been assigned by Meyers.¹²

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