

Isolation of Bicyclo[3.2.1]oct-3-ene-2,8-dione Products (Formal 5 + 2 Cycloadducts) from Reactions of Styrenes with 2-Alkoxy-1,4-benzoquinones

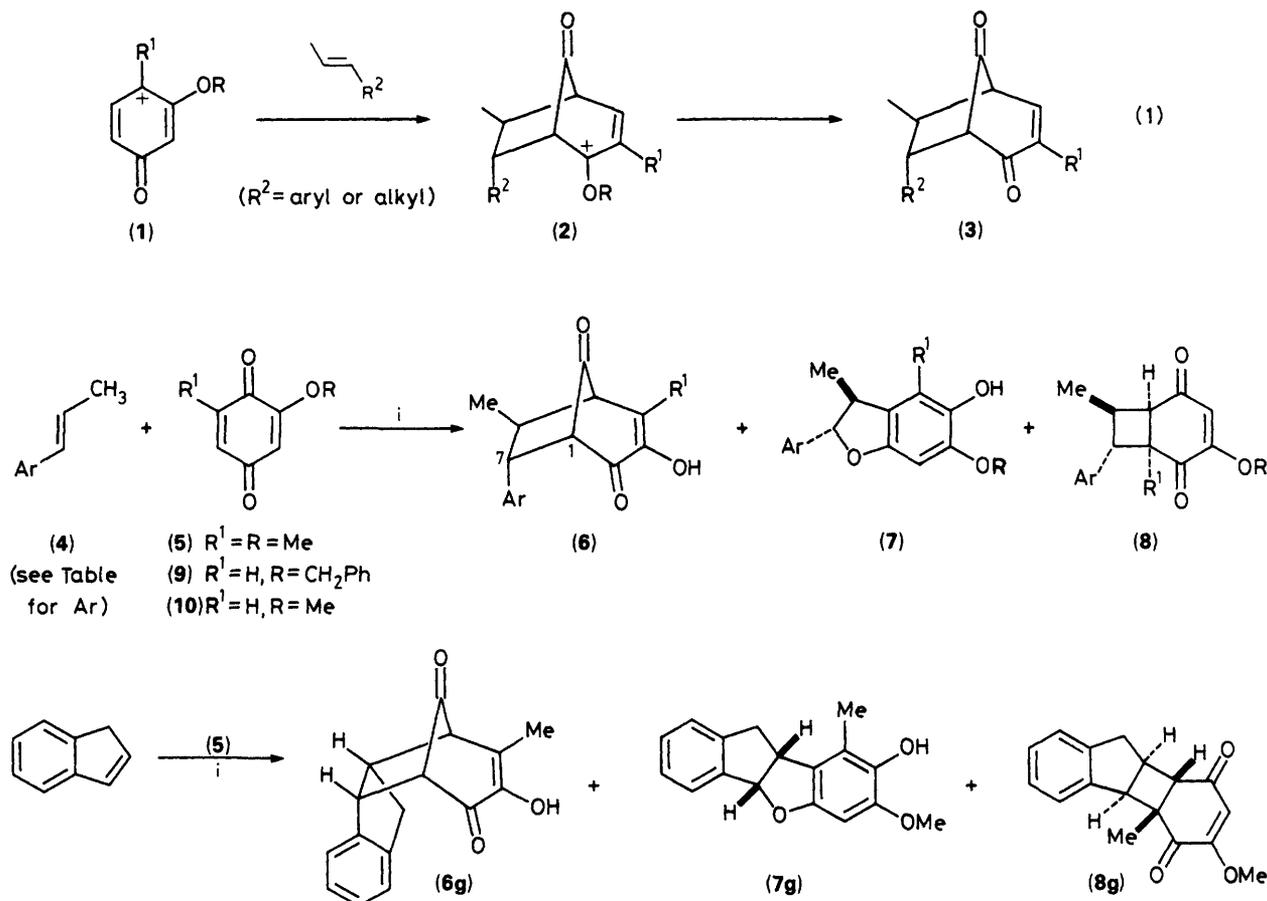
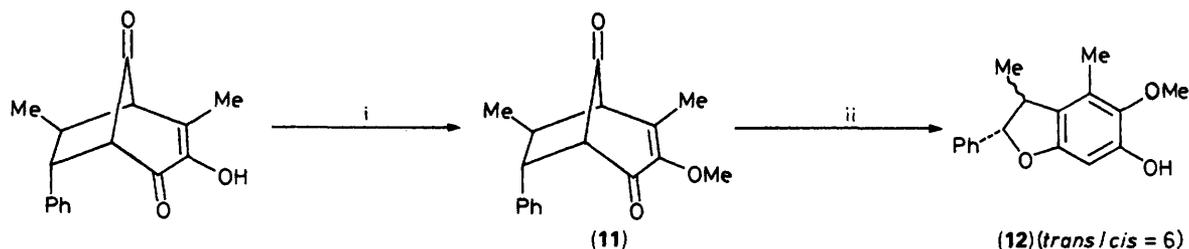
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Titanium(IV) catalysed reactions of 2-alkoxy-1,4-benzoquinones and various *trans*- β -methylstyrenes stereoselectively produce 7-aryl-3-hydroxy-6-methylbicyclo[3.2.1]oct-3-ene-2,8-diones in moderate yield.

Various 3,4-dialkoxy- or 3-alkoxy-4-alkyl-cyclohexa-2,5-dien-1-one-4-yl cations (**1**), generated from acid catalysed solvolysis of *p*-quinone monoketals¹ or *p*-quinol ethers² or by electrochemical oxidation of 3,4-dialkoxyphenols,³ react with alkenes to yield bicyclo[3.2.1]octenyl cations (**2**) perhaps *via* a

concerted, thermally allowed 5 + 2 ($4\pi + 2\pi$) cycloaddition (equation 1).⁴ Dealkylation of (**2**) then produces the isolable bicyclo[3.2.1]oct-3-ene-2,8-dione adducts (**3**). These reactions are intermolecular variants of the remarkable perezone to pipitzol rearrangement⁵ and have been used in elegant,

Scheme 1. Reagents and conditions: i, $\text{TiCl}_4/\text{Ti}(\text{OPri})_4$, CH_2Cl_2 , -78°C .Scheme 2. Reagents and conditions: i, K_2CO_3 (2 equiv.), MeI (20 equiv.), acetone, room temp., 12 h, 81%; ii, H_2SO_4 (1 equiv.)/HOAc (10 equiv.), CH_2Cl_2 , 24 h, 48% at 60% conversion.

efficient syntheses of neolignans,¹⁻³ sesquiterpenes,^{1,3} and tropolones.⁶ To our knowledge, the formation of formal 5 + 2 cycloadducts has not been routinely observed in reactions of 1,4-benzoquinones directly with alkenes under thermal or Lewis acid catalysed conditions, in spite of their prominent use in synthesis.^{4,7} We now report the isolation of bicyclo[3.2.1]octenedione products from titanium(IV) catalysed reactions of styrenes with 2-alkoxy-1,4-benzoquinones,⁸ which extends the generality of formal 5 + 2 cycloaddition processes of pentadienyl cations^{4c} to include reactions of Ti^{IV} -1,4-benzoquinone complexes with styrenes.

Thus, addition of *trans*- β -methylstyrenes (4a-d) to a mixture of $\text{TiCl}_4/\text{Ti}(\text{OPri})_4$ and 2-methoxy-6-methyl-1,4-benzoquinone (5), in dichloromethane at -78°C produces, after

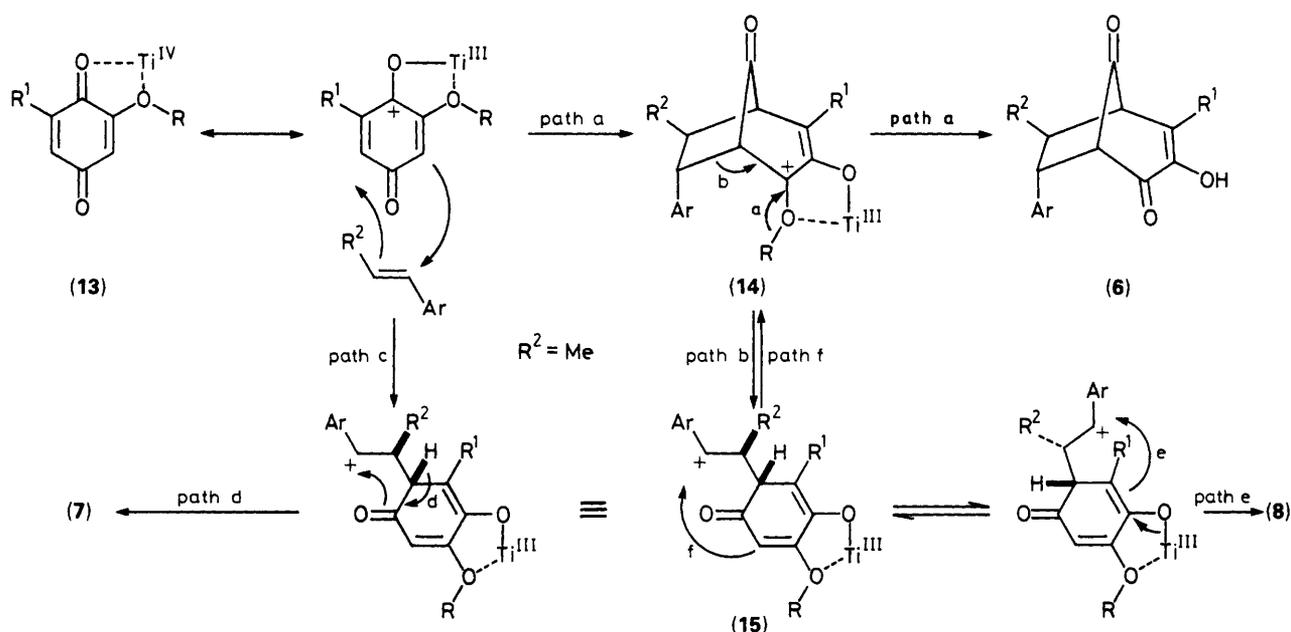
aqueous bicarbonate quench, extraction and silica gel chromatography, the bicyclo[3.2.1]octenediones (6), dihydrobenzofurans (7), and bicyclo[4.2.0]octenediones (8) (Scheme 1 and Table 1).[†] In a similar manner, 2-benzyloxy-1,4-benzoquinone (9) and styrene (4a) afford (6e) and (7e), and 2-methoxy-1,4-benzoquinone (10) reacts with (4b) to give (6f), (7f), and (8f). Indene reacts with quinone (5) to give (6g), (7g), and (8g). Adduct (6a) does not give dihydrobenzofuran or bicyclo[4.2.0]octenedione products related to (7) or (8) upon exposure to protic [H_2SO_4 , *p*-TsOH (Ts = $\text{OSO}_2\text{C}_6\text{H}_4\text{Me}$), $\text{CF}_3\text{CO}_2\text{H}$, room temp. to 80°C] or Lewis

[†] All new compounds exhibited satisfactory spectra (^1H and ^{13}C n.m.r., i.r., mass and u.v.), elemental analysis, and/or exact mass.

Table 1. Titanium(IV) catalysed reactions of 2-alkoxy-1,4-benzoquinones with styrenes.

Styrene	Quinone	TiCl ₄ : Ti(OPr ⁱ) ₄ [equiv. Ti ⁺ /equiv. quinone = 1.0]	Products (% yield)
a; Ar = Ph			(6a) R ¹ = Me, Ar = Ph
b; Ar = 2-MePh			(6b) R ¹ = Me, Ar = 2-MePh
c; Ar = 4-MePh			(6c) R ¹ = Me, Ar = 4-MePh
d; Ar = 4-ClPh			(6d) R ¹ = Me, Ar = 4-ClPh
			(6e) R ¹ = H, Ar = Ph
			(6f) R ¹ = H, Ar = 2-MePh
			(7a) or (8a) R ¹ = R = Me, Ar = Ph
			(7b) or (8b) R ¹ = R = Me, Ar = 2-MePh
			(7c) or (8c) R ¹ = R = Me, Ar = 4-MePh
			(7d) or (8d) R ¹ = R = Me, Ar = 4-ClPh
			(7e) or (8e) R ¹ = H, R = CH ₂ Ph, Ar = Ph
			(7f) or (8f) R ¹ = H, R = Me, Ar = 2-MePh
(4a)	(5)	3:1	(6a) (41)
(4b)	(5)	2:1	(6b) (44)
(4c)	(5)	2:1	(6c) (21)
(4d)	(5)	3:1	(6d) (27) ^a
(4a)	(9)	3:1	(6e) (15)
(4b)	(10)	3:1	(6f) (13–22)
Indene	(5)	3:1	(6g) (32)
			(7a) (10)
			(7b) (2)
			(7c) (10)
			(7d) (2) ^a
			(7e) (11)
			(7f) (4–10)
			(7g) (3)
			(8a) (17)
			(8b) (32)
			(8c) (37)
			(8d) (8) ^a
			—
			(8f) (44–60)
			(8g) (22)

^a In this experiment, 33% of quinone (5) was recovered; the yields reported are not based on recovered starting material.

**Scheme 3**

acids [TiCl₄, Me₃SiOSO₂CF₃, Bu₂Si(OSO₂CF₃)₂, BF₃·Et₂O, –78 to 80 °C]. Interestingly, however, methylation of (6a) to (11) followed by treatment with H₂SO₄ does produce dihydrobenzofuran (12) (Scheme 2)⁹ and the acid catalysed rearrangements of (8) to (7) have been previously reported.⁸

Compounds (7c) and (8c) have been previously characterized⁸ and the structure of (6a) was established from spectral data.^{2,6,9} An *endo* orientation of the aryl substituents in (6a–g) is consistent with an observed 7–8 Hz coupling constant between H-1 and H-7. The stereochemistry of the cyclobutane adducts (8) is assigned from ¹H–¹H nuclear Overhauser enhancement (n.o.e.) experiments⁸ at 300 or 500 MHz where all of the ¹H resonances are distinct and readily assigned by decoupling experiments. Structures of the other products were established by spectral comparison to (6a), (7c), or (8c).

Compounds (6a–g) may result from an allowed 5 + 2 (4π + 2π) cycloaddition of the pentadienyl cation moiety of the quinone–Ti^{IV} complex (13) with the styrenes to give cation

(14) (Scheme 3, path a)⁹ which is then dealkylated by chloride ion in the reaction mixture or upon aqueous quench. Bicyclic cation (14) may also be an intermediate in the formation of the dihydrobenzofurans or the bicyclo[4.2.0]octenediones (7) or (8), respectively, *via* fragmentation to (15) (Scheme 3, path b).⁸ Alternatively, (14) may be formed in competition with cation (15) (path c) which leads to (7) and (8) *via* paths d and e, respectively, or (14) may be formed from (15) *via* path f. In any event, the isolation of bicyclo[3.2.1]octenediones (6), reported herein, does indicate that products of formal 5 + 2 cycloaddition between *p*-benzoquinones and styrenes can occur under Ti^{IV} catalysed conditions. We are continuing to explore the mechanism, generality and synthetic utility of these reactions.

‡ The failure of (6a) to give products related to (7) or (8) does not rule out (14) as an intermediate in the formation of (7) or (8) since the complex formed from (6a) and an acid is not the same as (14) or as the complex formed on protonolysis of (11).

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