New cytotoxic selenoderivatives of guaianolides

Paolo BARBETTI¹, Giuseppe FARDELLA¹, Ione CHIAPPINI¹, Vito SCARCIA² and Ariella FURLANI CANDIANI²

¹Institute of Pharmaceutical Chemistry and Pharmaceutical Technique, University of Perugia, Perugia, and ²Institute of Pharmacology and Pharmacognosy, University of Trieste, Italy

(Received January 19, 1988, accepted December 19, 1988)

Summary — A series of 13-phenylselenoderivatives of natural and semisynthetic guaianolides were prepared and their cytotoxicity tested *in vitro* against KB cell cultures. Generally the presence of the 13-Se-phenyl group led to an increased bioactivity (ID_{50}) supporting the hypothesis that it may act as a "masked" α -methylene- γ -lactone group.

Résumé — **Nouveaux dérivés séléniés cytotoxiques des guaianolides.** Une série de 13 nouveaux dérivés séléniés des guaianolides naturels et semisynthétiques a été préparée et testée in vitro contre des cultures cellulaires KB. Généralement, la présence des sélénodérivés a mené à une augmentation de la bioactivité (ID₅₀). Les résultats obtenus confirment l'hypothèse que les produits séléniés peuvent agir biologiquement comme un groupe α -méthylène γ -lactone « masqué ».

selenoderivatives / guaianolides / sesquiterpene lactones / cytotoxic activity

Introduction

Naturally occurring unsaturated γ - and δ -lactones have attracted considerable interest in recent years because of their cytotoxic, anti-inflammatory, antibacterial, antihyperlipidemic and anti-allergic activity [1, 3].

In particular α -methylene- γ -lactone sesquiterpenes of natural, semisynthetic and synthetic origin can be considered as members of a large class of alkylating agents employed in cancer chemotherapy, and have been used as antitumour drugs in some cases [4-7].

It is well known that the alkylating nature, and hence the potential tumour-inhibiting ability of these sesquiterpene γ -lactones derives from a Michaelis-type interaction of the α , β -unsaturated lactone moiety with the nucleophile cellular components (sulphydryl enzymes) [8].

It is also widely recognized that the usefulness of most sesquiterpene lactones has been limited because of their toxicity [8]. Unfortunately, the biologically active moiety is also very reactive and hence may give rise to indiscriminate reactions towards various biological nucleophiles. Several attempts have been made to achieve lower toxicity in these compounds by improving the hydrophilicity *via* attachment to an appropriate carrier (carbohydrate or steroid moieties) and by derivatisation of α -methylene- γ butyrolactone (sulphonate esters, silyl enol esters and others) with the aim of exploring the role of the exocyclic methylene which has a good leaving group [3, 8].

Such modification may give rise to irreversible alkylations through the replacement of the leaving group with the biological nucleophile and hence might enhance the selectivity of the α , β -unsaturated exomethylene lactone moiety [3] (see Scheme 1).



Scheme 1.

As an extensive of this hypothesis the synthesis of appropriate precursors that might act as "masked" α -methylene- δ -lactones has recently been proposed, *i.e.* a number of selenyl- δ -lactones with different lipophilic moieties; their cytotoxic activity has been screened and correlated with the hypothesis of the *in situ* generation of corresponding α -methylene lactones, probably *via* the formation of selenoxides [3]. In actual fact, the intracellular oxidation of the phenylseleno group to selenoxide seems to be particularly plausible tumour cells.

It has been supposed that cancer cells are more sensitive than normal cells to H_2O_2 -mediated cytotoxicity because they lack sufficient catalase and peroxidases for enzymatic defence [9].

In addition, the high rate of pericycle-syn-elimination of β -H phenylselenoxide and therefore its nature as a very good leaving group, is well known in Se-organic chemistry [10].

This paper deals with the preparation of a new series of

300

13-phenylselenoderivatives (VII-XI) of some natural and semisynthetic guaianolides (I-IV and VI); the *in vitro* cytotoxic activity of all the compounds (X-XI) against KB cell cultures is reported and some conclusions are drawn. In this case in fact, the 13-phenylselenoderivatives might act as precursors of the α,β -exomethylene- γ -lactone group undergoing Se-oxidation and retro-elimination inside the cell (see Scheme 2).



Scheme 2.

Consequently a member of the 13-phenylseleno-guaianolide family, acting as an endocellular source of its exomethylene- δ -lactone parent compound, might behave as a potential procytotoxic agent. Some interesting results on the relationship between bioactivity and chemical feature are discussed.







Chemistry

Compounds II-IV were prepared according to the known procedures using grosheimin I and cynaropicrin XII as starting material [11]. Compound V was obtained by treating I with NaOEt in anhydrous EtOH.

Compound VI was obtained by alkaline hydrolysis of XII [11]. Selenoderivatives VII-XI were prepared by the known procedure [3], appropriately modified by reacting the corresponding guaianolides with diphenyldiselenide in the presence of NaBH₄ in anhydrous EtOH (see Scheme 3).



Scheme 3.

Discussion

The *in vitro* cytostatic activity of compounds I-XI has been evaluated against KB cell cultures and is expressed as ID_{50} (see Table III).

The ID_{50} values are expressed as molar concentrations since in our opinion they are more interesting than the ID_{50} values in micrograms per millilitre of MEM, considering that there are notable differences among the molecular weights of the compounds.

All substances appeared to be very active, showing ID_{50} values much lower than the standard criteria for a significant level of activity [7]; in particular, compound **VIII** showed an uncommon cytotoxic property.

From the above results some interesting observations can be drawn about the structure-activity relationship of guaianolides I-VI, and the related 11-H-13-phenylseleno-derivatives VII-XI (see Tables III and IV).

It can be noted that the acetylation of the C₆-OH group of grosheimin I increases the cytotoxicity \approx 7-fold (M) in compound III, probably because of enhancement of the lipophilicity and membrane permeability. On the other hand, higher positive effect in bioactivity – a factor of about 11 – is caused by the reduction in the C₃-carbonyl of I to a more polar C₃-OH group in compound II.

In addition, comparison of the acetylated compound IV with the parent non-acetylated II shows that the acetylation of both C₃-OH and C₈-OH groups leads to a less active compound (a factor increase of about 4). Considering that the acetylation of C₈-OH should increase the bioactivity (compound I/III), one can hypothesize that the C₃-OH free group would play a positive pharmacodynamic role in this class of substances. The lowest cytotoxic activity of compound V may be justified by considering the presence in C₁₃ of a not good-leaving ethoxyl group.

The value of $ID_{50}=0.84 \ \mu g/ml$ for compound **VI** may be compared with $ID_{50}=3.5-5.5 \ \mu g/ml$ reported in the literature for cynaropicrin **XII** [13]. In this case one may hypothesize that the hydrophilic property of the side-chain

	Ч	H,	H	Н₄	H,	Ц	H,	H	H ₉	\mathbf{H}_{ll}	H ₁₃	H	H ₁₅ (0Ac 21	HArom. 3	H Arom.	$\mathbf{H}_{1'}$ $\mathbf{H}_{2'}$	Solvents	
Grosheimin I	3.19 d(8.(0) 2.55 m 0) 2.55 m	1	2.35 m	2.25 m	d(10.0 4.00 d(10.0))) 3.07 m	3.92 ш	2.88 d(6.0)	1	6.31 d(2.0)	a,b 5.11s 5.87 e	1.17 d(8.0) -			÷		CDCl [§]	
11- <i>H</i> ,-13-Se-phenyl grosheimin VII	3.00 m	2.60 d(14 2.60 d(14 2.50 m	- (0)	2.16m ⁴	d(9.0) 1. 2.25 d(9.0) d(9.0)	4.01 t(9.0)	, 2.50 m	3.66 m	2.32 m 2.67 d(6.0) d(12.0) 2.01 d(12.0)	3.24* m)	3.27 d(10.0)	4.80 s 4.80 s	1.14 d(9.5) -	- 7.	58 d(7.8)	.23 m	1	DMSO-d ⁸	
11-H,13-4thoxy grosheimin V	d(8.1 3.15 d(8.1 d(2.0	0) 1) 2.51 m 3)	· 1	2.30 m	2.45 m	d(9.5) 4.05 d(9.05	2.20 m	3.70 m	2.85 m 2.30 m	2.85 m	4.10 d(9.0) 4.10 d(9.0) 3.55 d(9.0)	5.10s 4.80s	1.24 (9.0)	I	I		3.65 m 1.27 t(9	.0) CDCI ₃	
3-Dyhydro grosheimin II	1	J	3.85 ш	1	I	d(10) 4.08 d(10)	1	3.60 m	1	I	6.08 d(9.0) 6.08 d(3) 6.12 d(3)	4.98 brs 4 90 hrs	1.12 d(7.5) -	1	,		I	cDCt	
3.Dihydro, 11- <i>H</i> ,13.Se-phenyl grosheimin VIII	1.94 m	2.14 d(5) d(5) d(5)	5) (0) 3.75 m	1.86 m	1.75 m	d(10.5 3.97 d(10.5	5) d(10.5) 2.33 d(10.5) 3) d(10.5)) 3.59 m	2.68 d(11.9 d(4.0)) 2.97 d(10.5) d(3.7)	^{3.53} d(13.2) ^{3.53} d(3.7) ^{3.30} d(13.2)	5.03 s 4.98 s	- 1.21 d(6.5)	- 7.	56 d(8.0) $~\tilde{\imath}$	/.28 m	1	CDCI3	
8-Acetyl- grosheimin III	1	2.60 m	I	I	I	4.18 t(9)	5.00 m	1		I	6.40 d(13.7) 6.40 d(3)	5.12 s	1.20 d(8) 2	2.20s -	'	I	I	CDCI ⁵	
8-Acetyl, 11-H,13-Se-phenyl-	I	I	I	ł	ł	d(9) 4.00 400	1	- 1 1 1 1 1	I	ł	3.43 d(3) 3.78 d(3)	4.80 s	1.22 d(5) 2	2.0s 7.r	50 m 0	/.28 m	ı I	cDCI	
Brosteinuu LA 3-Dihydro- 3,8-diacetyl prosheimin IV	I	I	4.60 m	ļ	ł	(9) 3.93 d(9)	I	4,40 m	I	ı	5.42 d(3) 5.42 d(3)	4.70 brs	1.10 d(7.5) ²	2.17s - 2.16s -	,	1	I	cDCI	
3-Dihydro-3,8- diacetyl,11-H,13, Se-phenyl	ŧ	I	4.72 m	I	1	(9) 3.98 d(9)	I	- 4.72 m	I	I	3.35-3.00 m	5.06s 5.02s	1.12 d(6) 1	2.07 s 7.: 1.92 s	55 m 2	1.28 ш	I	CDCI ^b	
grosneumin A 8-Deacyl cyanopicrin VI	I	1	4.53 m	ł	ł	d(9) 4.18 d(9)	I	3.96 m	ı	I	6.25 d(3) 6 15 d(3)	5.15s 5 Mis	5.47 s	1		I	1	cDCI	
8-Deacyl-11- <i>H</i> ,13- Se-phenyl cynaropicrin XI	1	I	d(9) 4.66 d(9)	ļ	I	4.07 d(9) d(9)	t	3.68 m	I	I	3.42 d(3)	5.00s 4.92s	5.25 brs	- 7.:	55 d(7.5) 7	7.20 m	I	CD30D	

Table I. 1H NMR data.

Coupling constants in Hz are between parentheses. ^aSpectra recorded at 400 MHz. ^bSpectra recorded at 90 MHz. *Collapsing to d, J=10.0 Hz by irr. at δ 2.5. **Collapsing to q, J=9.5 Hz by irr. at δ 2.25.

Compound	C1	C_2	c3	C₄	Č	ບຶ	c,	రి	° C	C10	C ₁₁	C ₁₂	C _{I3}		C ₁₅	Č l	2, C	Ac	C. Arom.	Solvent
Grosheimin I	40.40* 45.88 d	43.53** 47.08 t	218.67 218.13 s	47.23 39.29 d	49.80* 48.52 d	83.26 81.99 d	51.11 49.94 d	73.20 71.71 d	49.16** 1 42.36 1 t s	[45.36] [44.36]	138.73 1 137.72 1 s s	170.29 169.6	124.52 1 123.17 1 1	14.44 1 13.81 1 q	5.02 4.25				- 131.24 129.08	Pyr DMSO
11-H,13-Se-phenyl grosheimin VII	46.43* d	48.62** t	221.66 s	38.89 d	46.56* d	83.34 d	49.89≠ d	73.66 d	43.14** 1 t s	(45.15 5	50.14≠ 1 d s	176.59 2	27.69 1 t	13.42 1 q	3.85 -	1	I		d d 128.81 126.41 d d	OSMG
11-H,15-Ethoxy grosheimin V 3-Dyhidro grosheimin II	4/.09* d d d	4/.0/** t 38.97** t	211.11 s 77.52# d	39.49 d 42.92 d	47.11* d d	83.67 d d f	50.67 d 53.56 d	74.44 d 72.78# d	43.72** { 43.72** { 1	[46.05 /	49.05 1 d 5 139.76 1 s s	176.73 (s 1 171.39 1 171.39 1	58.09# 1 121.79 1 121.79 1	13.76 1 9 14.17 1 9	- 4.46 6 - t 8.33 - t	6.44* 15 - 9 -			1 1	DMSO Pyr
3.Dihydro,11,- <i>H</i> , 13-Se-phenyl grosheimin VIII	48.21* d	46.59** t	78.37 d	38.50 d	47.11* d	81.36 d	50.48 d	75.21 d	42.62** 1 t s	(43.80	64.51 1 d s	171.30 2	27.92 1 t	15.29 1 9	8.06	I	- JI	9.45	133.221 /C.121 d d 127.49 d	CDCI ₃
8-Acetyl- grosheimin III	43.20* d	46.85** t	218.23 s	40.55 d	46.41* d	82.23 d	51.08 d	74.33 d	43.04**] t s	142.41 1	135.98 1 s s	171.10	124.58 1	16.68 1 9	4.86 -	1	°, a , s	1.04	1	CDCI ₃
8-Acetyl,11- <i>H</i> , 13-Se-phenyl grosheimin IX	43.47* d	47.08** t	218.26 s	39.45 d	46.83* d	83.26 d	50.96 d	75.84 d	43.47** 1 t s	142.76 4	d s	175.36 2	28.12 1	15.99 I. 9	4.20 -		≍°८¢;	9.06	133.37 129.24 d d 127.53 d	cDCI3
3-Dihydro,3,8- diacetyl gros- heimin IV	43.67* d	49.77** t	79.95≠ t	35.57 d	43.54* d	80.54 d	51. <i>57</i> d	74.40≠ d	40.36** 1 t s	[41.50]	137.35 1 s s	1 169.7 1 s	122.40 1	17.06 1 q	8.29 -	I .	- 0 7 0	0.00 109.09 s 1.08 22.05 q	1	CDCI ₅
3.Dihydro-3,8-dia- cetyl,11-H,13.Se- phenyl grosheimin X S.Deacyl cynaro- picrin VI	43.74* d 44.07* d	42.85** t 39.94** t	79.52≠ d 73.46 d	35.34 d 154.82 s	47.47* d 50.92* d	81.08 d 79.94 d	50.75• d 51.46 d	76.17# d 72.00 d	42.56** 1 t s 45.12** 1 t s	[42.02 5 [44.64]	50.82• 1 d s 140.19 1 s s	175.34 2 1 1 170.43 1 1	27.93 1 t (22.56 1 t	16.42 1 q 16.24 1 t	7.86 - 11.02 -		¥°, ₽	9.66 0.33	133.63 129.16 d d 127.50 d -	CDCI ₃ Pyr
8-Deacyl,11-H,13- Se-phenyl cynaro- picrin XI	46.60* d	39.94 t	75.36≠ d	133.42 s	48.96* c	81.29 d	50.83 d	73.59≠ . d	44.32** 1 t s	145.09 5	52.82 1 d s	178.97 2	28.64 1	15.89 1 t	10.05 -)	I		132.15 129.93 d d 127.68 d	CD;OD

*, **, \neq , • The assignments of these signals may be reversed.

302

Table II. 13C NMR data.

Table III. Cytostatic activity against KB cell growth.

Compounds ^a	Drug conc. (μ g/ml MEM)	% growth inhib.±SE	Significance level (p)	ID ₅₀ values ^b (M)
Guaianolides grosheimin I	0.10 1.00 2.50 5.00 10.00	1.66±0.87 25.27±0.61 83.57±0.75 cytolysis cytolysis	<0.05 <0.01 <0.01 	5.6×10-6
3-Dihydro grosheimin II	0.10 0.50 1.00 2.50 10.00	48.81±0.89 53.45±0.70 60.15±1.22 68.06±0.57 cytolysis	<0.01 <0.01 <0.01 <0.01	4.9×10 ⁻⁷
8-Acetyl grosheimin III	0.10 1.00 2.50 5.00 10.00	39.02±0.51 65.75±0.94 88.47±0.33 cytolysis cytolysis	<0.01 <0.01 <0.01 	7.9×10 ⁻⁷
3-dihydro-3,8-diacetyl grosheimin IV	0.10 0.50 1.00 2.50 10.00	1.40±0.25 43.09±1.08 52.69±1.75 93.80±0.66 cytolysis	ns <0.01 <0.01 <0.01	2.0×10 ⁻⁶
11-H-13-OEt grosheimin V	0.10 1.00 2.50 5.00 10.00	$\begin{array}{c} 1.42 \pm 0.72 \\ 15.69 \pm 0.65 \\ 59.05 \pm 0.66 \\ 80.80 \pm 0.50 \\ 95.65 \pm 0.49 \end{array}$	ns <0.01 <0.01 <0.01 <0.01	7.8×10 ⁻⁶
8-Deacyl cynaropicin VI	$\begin{array}{c} 0.10 \\ 0.50 \\ 1.00 \\ 1.25 \\ 2.50 \\ 10.00 \end{array}$	0.52±0.25 41.99±0.67 52.50±1.69 62.34±0.76 cytolysis cytolysis	ns <0.01 <0.01 <0.01 	3.2×10 ⁻⁶
11-H-13-Se-Phenyl derivatives				
VII	0.10 1.00 2.50 5.00 10.00	54.15 ± 0.75 66.58 ± 0.60 84.35 ± 0.48 89.11 ± 0.50 99.27 ± 0.47		2.1×10 ⁻⁷
VIII	0,10 1.00 2.50 5.00 10.00	59.82±0.65 78.73±0.86 85.16±0.50 90.12±0.46 97.00±0.49		4.7×10 ⁻⁸
IX	0.10 1.00 1.25 2.50 5.00	8.21±2.31 8.15±2.12 19.08±1.16 61.90±1.64 cytolysis	<0.05 <0.05 <0.01 <0.01	4.6×10 ⁻⁶
X	0.10 1.00 1.25 2.50 5.00	0.57±2.72 1.11±2.30 66.77±0.58 74.37±0.94 98.55±0.36 cytolysis	n.s <0.01 <0.01 <0.01	1.2×10 ⁻⁶
XI	0.10 1.00 2.50 5.00 10.00	49.85±1.19 55.01±1.45 91.94±0.36 cytolysis cytolysis	<0.01 <0.01 <0.01 	3.6×10-7

^aThe compounds were previously dissolved in dimethyl sulfoxide. ^bThe results are expressed as molar concentrations at which cells showed a 50% growth inhibition (ID_{50}) .

Table IV. R-values for the couples of related compounds: guaianolides / Se-derivative.

Couple	R	
I / VII II / VIII III / IX IV / X VI / XI	26.7 10.4 0.17 1.7 8.9	
ID., guaianoli	de (M)	

$$R = \frac{5000}{ID_{s0} \text{ Se-derivative (M)}}$$

methacryl ester in C_8 can prevent the transport of XII into the cell.

The major activity of compound VI with respect to grosheimin I could be related both to the reduction of the C_3 carbonyl group in the molecule and probably also to the presence of a further exomethylene non-saturation in C_4 .

The presence of the Se-phenyl group in compounds VII-XÎ generally leads to a notable enhancement of cytoxicity with respect to the parent substances (see Rvalues in Table IV).

The highest "Se-effect" in increasing cytotoxicity is shown by the pair of compounds grosheimin I/Se-derivative VII (R=26.7), and the same trend can be observed grosheimin II / Se-derivative 3-dihvdro VIII for (R=10.4), for 8-deacyl-cynaropicrin VI / Se-derivative XI (R=8.9) and also for the products IV / X (R=1.7). The pair of compounds 8-acetyl grosheimin III / Se-derivative **IX** is inconsistent with the general trend, showing a most surprising inversion of the R-ratio (R=0.17). Further investigation will be needed to elucidate the significance of this particular case.

It is also of interest to note that compounds II, VII, and **XI** showed very low ID_{50} values (0.1 μ g/ml) and that 11-H-13-phenylseleno-3-dihydro grosheimin VIII (ID_{50} = $0.02 \,\mu g/ml = 4.7 \times 10^{-8}$ M) must be considered one of the most cytotoxic substances in the family of the sesquiterpene lactones and derivatives reported in the literature [7]

With the sole exclusion of the pair of compounds III / IX, the in vitro pharmacological data obtained permit us to state for the moment that the presence of the Se-phenyl group in C₁₃ leads to more cytotoxic compounds, and lends support to the hypothesis outlined above, thus making it worthwhile and of interest to continue our research along these lines.

The in vitro assay obviously cannot be conclusive regarding antitumour effectiveness of our compounds, and so the most *in vitro* active products should be submitted for *in* vivo testing against murine experimental tumours.

Experimental protocols

Chemistry

Melting points were determined on a Kofler hot stage instrument and are uncorrected. ¹H NMR were recorded on Varian EP-390 90-MHz and Bruker Spectrospin 400-MHz instruments using TMS as the internal standard (chemical shifts in δ values). ¹³C NMR spectra were determined using a Bruker WP80SY spectrometer. Mass spectra were obtained with an AEI MS 902 spectrometer.

For column chromatography, Merck silicagel 60 (230-400 mesh) was used. Reactions were monitored by TLC using Merck DC-Alufolien Kieselgel 60 F_{254} . The microanalyses were performed by the Microanalytical Laboratory of the Institute of Organic Chemistry, Pharmacy Faculty, Perugia.

Method A: 11-H-13-Se-phenyl grosheimin **VII** 312 mg (1 mmol of diphenyldiselenide and 76 mg (2 mmol) of NaBH₄ were dissolved under stirring in 20 ml of anhydrous EtOH at room temperature and in N₂ atmosphere.

As soon as the yellow solution changed to decoloration 262 mg (1 mmol) of grosheimin I was added and, after few minutes, a white powder

began to precipitate. The mixture was stirred for a further 15 min, then the precipitate was filtered in vacuo, washed with cool EtOH and turned to unitary on TLC. The solution was poured into 50 ml of HCl-acidulated water and then extracted with $CHCl_3$. The chloroform extract was washed first with NaHCO₃, then with NaCl aqueous solutions and dried over anhydrous Na₂SO₄. The evaporation of the solvent in vacuo yielded a crude yellow oily residue which was purified by means of silicagel flash-CC (eluent: CHCl₃ CHCl₃ / MeOH=95 / 5). The TLC-pure fractions of 11-H-13-Se-

CHCl₃ CHCl₃ / MeOH=95/5). The TLC-pure fractions of 11-H-13-Sephenyl grosheimin **VII** obtained proved to be TLC and spectroscopically identical to the predicted, spontaneously precipitated one. Yield = 338 mg (0.8 mmol); mp=237-240°C (from EtOH). Anal. for C₂₁H₂₄O₄Se: calcd.=C 74.09; H 7.11; O 18.80; found=C 74.11; H 7.09; O 18.80. IR (ν_{max} , cm⁻¹, KBr): 3500-3300 (-OH), 2980, 1770 (γ -butyrolactone), 1740 (5-membered ring ketone), 1620, 1580, 1440, 1400, 910 (aromatic and olefine unsaturations), 1360, 1300, 1200, 1170 (C-O and C-Se-C). ¹H NMR (δ , ppm. DMSO-4, 400 1170 (C-O and C-Se-C). ¹H NMR (δ , ppm, DMSO-d₆, 400 MHz)=see Table I; ¹³C NMR (δ , ppm, DMSO-d₆, 22.5 MHz)=see Table II; MS (m/z, I%)=422.2 (7.0)M⁺ ⁸²Se, 420.6 (42.2)M⁺ ⁸⁰Se, 418.8 (22.5)M⁺ ⁷⁸Se, 158.0 (11.3), 93.7 (11.3), 91.6 (25.3).

3-Dihydro-11-H-13-Se-phenyl grosheimin VIII

The reaction of Ph-selenization of 3-dihydro grosheimin II was carried out by the same procedure and stoichiometric ratios as reported in method A. Starting from 312 mg (1 mmol) of diphenyldiselenide, 76 mg (2 mmol) of NaBH₄ and 264 mg (1 mmol) of **III**, 340 mg (0.8 mmol) of pure **VIII** were obtained. mp=148-149°C (from ethyl ether).

Anal. for $C_{21}H_{26}O_4Se$: calcd. = C 73.66; H 7.65; O 18.69; found=C 73.35; H 7.78; O 18.87. IR (ν_{max} , cm⁻¹, CHCl₃): 3600 and 3400 (-OH), 3000-2860, 1770 (γ -butyrolactone), 1640, 1580, 1480-1440, 910 (olefine and aromatic unsaturations), 1350, 1290, 1165, 1140, 1070, 990 (C-O and C-Se-C). ¹H NMR (δ_c , ppm, CDCl₃, 400 MHz)=see Table I; ¹³C NMR (δ_c , ppm, CDCl₃, 22.5 MHz)=see Table II. MS (m/z, 1%)=424.3 (3.7) M⁺ ⁸²Se, 422.2 (22.9) M⁺ ⁸⁰Se, 420.4 (12.5) M⁺ ⁷⁸Se, 192.4 (14.6), 159.0 (20.8), 158.0 (12.5), 157 (10.4), 105.5 (10.4), 91.6 (25.0) (25.0).

3-Dihydro-3,8-diacetyl-11-H-13-Se-phenyl grosheimin X

358 mg (1 mmol) of 3-dihydro-3-8-diacetyl grosheimin IV were subjected to the reaction of selenization according the same procedure of method A. About 300 mg (0.59 mmol) of pure compound X were obtained as an amorphous powder.

as an antorphous powder. Anal. for $C_{25}H_{30}O_6Se$: calcd. =C 70.40; H 7.09; O 22.51; found =C 70.80; H 6.98; O 22.22. IR (ν_{max} , cm⁻¹, CHCl₃): 2940, 2990, 1770 (γ -butyrolactone), 1730 (C=O), 1645, 1570, 900 (olefine and aromatic unsaturations), 1370, 1250-X1150 and 1100-1000 (C-O and C-Se-C). ¹H NMR (δ , ppm, CDCl₃, 90 MHz)=see Table I; ¹³C NMR (δ , ppm, CDCl₃, 22.5 MHz)=see Table II.

8-Acetyl-11-H-13-Se-Phenyl grosheimin IX

304 mg (1 mmol) of 8-acetyl grosheimin III were selenized according to the same procedure as method A and yielded 210 mg (0.69 mmol) of

pure compound IX as an amorphous powder. Anal. for $C_{23}H_{26}O_5$ Se: calcd. = C 72.23; H 6.85; O 20.92; found=C 72.05; H 6.95; O 21.0. IR (ν_{max} , cm⁻¹, CHCl₃): 2920, 1775 (γ -butyrolactone), 1740–1730 (C=O groups), 1640, 1580, 1450, 910 (aromatic and olefine unsaturations), 1370, 1290, 1165, 1115, 1080, 1030 (C=O and C-Se-C). ¹H NMR (δ, ppm, CDCl₃, 90 MHz)=see Table I; ¹³C NMR (δ, ppm, CDCl₃, 22.5 MHz)=see Table II.

8-deacyl-11-H-13-Se-phenyl cynaropicrin XI

262 mg (1 mmol) of 8-deacyl cynaropicrin VI were selenized as reported above and yielded 335 mg (0.8 mmol) of pure compound XI. mp=148-50 °C (from benzene).

Anal. for $C_{21}H_{24}O_4$ Se: calcd. = C 74.09; H 7.11; O 18.80; found = C 74.01; H 7.28; O 18.71. IR (ν_{max} , cm⁻¹, KBr)=3400 (-OH), 2900, 1760, (γ -butyrolactone), 1580, 1440, 1400, 900, 740 (aromatic and olefine unsaturations), 1340, 1300, 1200, 1090 (C-O and C-Se-C). ¹H NMR (δ , ppm, CD₃OD, 90 MHz)=see Table I; ¹³C NMR (δ , ppm, CD₃OD, 22.5 MHz)=see Table II.

Method B: 11-H-13-ethoxy grosheimin V

262 mg (1 mmol) of grosheimin I were added to a solution of NaOEt in 2 ml of anhydrous EtOH under stirring at room temperature and in N₂ atmosphere. After 15 min. of reaction, the solution was poured into 50 ml of HCl-acidulated water then extracted with CHCl₃. The chloroform extract was washed first with NaHCO₃, then with NaCl aqueous solutions and then dried over anhydrous Na_2SO_4 . The evaporation of the solvent in vacuo yielded a crude amorphous residue wich was purified by means of repeated silicagel flash-CC (eluent: CHCl₃ CHCl₃ / MeOH= 95/5). The TLC-pure fractions obtained were recovered as a white amorphous powder.

Anal. for $C_{17}H_{24}O_5$: calcd. = C 66.22; H 7.84; O 25.94; found = C 66.32; H 7.90; O 25.78. IR (ν_{max} , cm⁻¹, CHCl₃): 3350 (-OH), 2880, 1760 (γ -butyrolactone), 1740, (5-membered ring ketone), 1660, 1595, 1515, 890 (olefine unsaturation), 1150, 1080, 970 (C-O). ¹H NMR (δ , ppm, CDCl₃, 400 MHz)=see Table I; ¹³C NMR (δ , ppm, DMSO-d₆, 22.5 MHz)=coc Table II. MS (m (τ , 19(λ)=208, 8 (8 11) M+ 264, 7 (13, 51) MHz)=see Table II. MS (m/z, 1%)=308.8 (8.11) M+, 264.7 (13.51) M+-OEt, 228.9 (34.23), 211.9 (27.3), 174.1 (27.9), 138.1 (26.1), 135.6 (83.8), 134.6 (100), 59.6 (39.64), 43.3 (24.3).

Pharmacology

An established cell line of human epidermoid carcinoma of the mouth (KB cells) was employed to test the cytostatic activity of the new compounds by the method of Geran *et al.* [12]. The cells were grown in 25 cm^2 tissue culture flasks with Eagle's minimal essential medium (MEM) [14] supplemented with 10% newborn calf serum, 10% non-essential amino acids and glutamine 2 mM, and buffered with TES (Ntris[hydroxymethyl] methyl-2-aminoethane sulphonic acid 3, mM), BES (N, N, bis[2-hydroxyethyl]²-2-aminoethane sulphonic acid, 3 mM), HEPES (N-2-hydroxyethylpiperazine-N'-2-ethane sulphonic acid, 3 mM) and Tricine (N-tris[hydroxymethyl]methylglycine, 3 mM) [15].

For the in vitro cytostatic assay the cells in a logarithmic growth phase, which were re-fed 24 h before testing, were used. The cells were treated for 5 min at 37°C with 0.05% 1:250 trypsin solution and then suspended in MEM to obtain a concentration of 10⁵ cells/ml. One millilitre was seeded in each Leighton tube. The samples were allowed to incubate for 24 h at 37°C. After this interval, the viable cells were attached to the bottom of the tubes. The tubes were regrouped at random and baseline was evaluated in 5 of these by counting the cells detached by trypsin solution [16]. The culture medium of the other Leighton tubes was discarded and the cells were fed with 4 ml MEM (control tubes) and with 4 ml MEM containing the compounds to be tested (treated tubes). The compounds were dissolved immediately before use in sterile dimethylsulfoxide (DMSO). Further dilutions were performed with the growth medium to the desired drug concentration. The final solvent concentra-

tion in MEM (0.5% in every tube) was previously tested by us and did not show cytotoxic effect. At least 5 concentration levels were used for each compound and each concentration value was tested in triplicate. The incubation was carried out at 37°C for 72 h, the time interval in which exponential growth occurs. Cell growth was estimated by counting the viable cells as previously described [16]. The cytostatic activity was evaluated as a percentage of cell growth inhibition in the treated tubes with respect to the controls on the basis of the formula: $100 - [(T-B)/(C-B) \times 100]$, where B was the baseline and T and C were the number of viable cells, respectively, in the treated and the con-trol tubes after 72 h of incubation. The significance of these results was evalutated by use of the t test.

The inhibition values were plotted against $\log D$, where D is the drug concentration in micrograms per millilitre of MEM. From these curves the ID_{50} values were obtained, where ID_{50} is the concentration at which the cells showed 50% growth inhibition in relation to the control values. The ID_{50} values were expressed as molar concentrations, setting the activity threshold at 10^{-4} M, since that appears to be a fairly realistic cutoff point for most compounds [17].

Acknowledgments

his work was supported by the Ministero Pubblica Istruzione of Italy.

References

- 1 Barbetti P., Fardella G., Chiappini I. & Casinovi C.G. (1981) Ann.
- Ist. Sup. Sanità 17 (2), 255 2 Lee K.M., Huang H.C., Piantadori C., Pagano C. & Geissman T.A. (1982) Cancer Res. 537, 61
- Groutas W.C., Theodorakis M.C., Tomkins W.A.F., Herro G. & Gaynor T. (1984) J. Med. Chem. 548, 27 and references there in 3
- 4 Livingstone R.B. & Carter S.K. (1970) in: Single Agents in Cancer Chemotherapy. Plenum Press, New York
- 5 Becker F.F. (ed.) (1977) in: Cancer, a Comprehensive Treatise. Plenum Press, New York
- 6 Prett W. & Ruddon R.W. (1979) in: The Anticancer Drugs. Oxford
- Univ. Press, Oxford Petit G.R. & Gragg G.M. (1978) *in: Biosynthethic Products for Cancer Chemotherapy II.* Plenum Press, London
- Stang P.J. & Trepton W.L. (1981) J. Med. Chem. 468, 24
- Cederbaum A. & Wasserman L.R. (1980) Proc. Am. Assoc. Cancer Res. 21, 26
- 10 Clive D.L.J. (1978) Tetrahedron Rep. 34, 1049
- 11 Barbetti P., Fardella G., Chiappini I., Scarcia V. & Furlani Candiani A. (1985) Farmaco Ed. Sci. 10, 755
- 12 Geran R.I., Greenberg N.H., McDonald M.M., Schumacher A.M. & Abbott B.J. (1972) Cancer Chemother. Rep. 3, 1
- Ohno N., Hizai H., Yoshioka H., Dominguez H. & Mabry T.J. (1973) 13 Phytochemistry 2, 221 14 Eagle H. (1959) Science 130, 432
- 15 Eagle H. (1971) Science 174, 500
- 16 Craciunescu D.G., Doadrio A., Furlani A. & Scarcia V. (1982) Chem.-Biol. Interact. 42, 153
- Hakala M.T. & Rustum Y.M. (1979) in: Methods in Cancer Research 17 (V.T. De Vita Jr & H. Busch, eds), Academic Press, New York, vol. 16, p. 247