

Letter

Rhodium(III)-Catalyzed Oxidative Annulation of 2,2'-Bipyridine N-Oxides with Alkynes via Dual C–H Bond Activation

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Supporting Information

ABSTRACT: Rh(III)-catalyzed switchable annulation of 2,2'-bipyridine Noxides with internal alkynes via dual C–H bond activation has been developed. Tuning the reaction conditions enabled the reaction pathway to be switched between rollover and nonrollover annulation, delivering 5,6-disubstituted-1,10phenanthrolines and 5,6,7,8-tetrasubstituted-1-(pyridin-2-yl)isoquinoline 2oxides in high yields, respectively. The procedures feature excellent



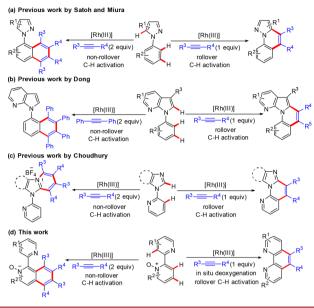
regioselectivity, broad substrate scope, and high tolerance of functional groups. The synthetic utilities of these obtained products were demonstrated in the catalytic reactions.

he past decade has witnessed the tremendous advances of transition-metal-catalyzed direct conversion of inert C-H bonds into C-C and C-X bonds for the synthesis and functionalization of various heterocyclic compounds.¹ It is not surprising that the wide application of 2,2'-bipyridines in coordination chemistry, transition-metal catalysis, supramolecular chemistry, and material science has encouraged the synthetic chemists to explore the feasibility of direct functionalization of these useful compounds under transitionmetal catalysis via direct pyridyl C-H bond activation. However, attempts at direct C-H functionalization of 2,2'bipyridines has been met with limited success, mainly because of their low electrophilicity and the catalyst poisoning by the strong coordination of N atoms.³ In 2009, Miura and coworkers first reported the regioselective direct alkenylation of 2,2'-bipyridines with terminal silvlacetylenes under Rh(I) catalysis.^{3a} Later, Chang et al. described the hydroarylation of alkenes and alkynes with 2,2'-bipyridines in the presence of a Rh(NHC) catalytic system.^{3b,c} It is believed that the introduction of the NHC ligand allows the facile decomplexation of one of the N atoms from N,N-chelate complex of 2,2'-bipyridines, thereby facilitating rollover cyclometalation that results in pyridyl C-H activation. Recent studies of the stoichiometric reaction of 2,2'-bipyridine N-oxide with a dimethylplatinum(II) complex by Puddephatt and co-workers revealed a rollover cyclometalation of the N-oxide ring with pyridine as a directing group, and no coordination of the Noxide oxygen was observed.^{4a} Therefore, it is natural to expect that 2,2'-bipyridine N-oxides would be more suitable substrates for direct C-H functionalization under transition-metal catalysis. Unfortunately, the chemistry about the direct functionalization of 2,2'-bipyridine N-oxides has been less investigated, with only one report known in the literature. Tzschucke and co-workers reported the Pd(II)-catalyzed direct halogenation of 2,2'-bipyridine N-oxides via rollover cyclometalation, and the halogenated products could be easily deoxygenated. $^{\rm 4b}$

Recently, Rh(III)-catalyzed C-H bond annulation with alkynes has evolved into a powerful tool for the preparation of highly substituted polycyclic aromatic compounds.⁵⁻⁷ In particular, several novel fused polycyclic (hetero)arenes could be readily synthesized by annulation reactions of (hetero)arenes with alkynes under Rh(III) catalysis via double C-H activation involving chelation-assisted ortho C-H activation and subsequent rollover or nonrollover C-H activation.^{6,} Notably, the reaction pathway between rollover and nonrollover annulation can be switched by varying the reaction conditions, thereby delivering different types of products from the same substrates. For example, Miura and Satoh et al. found that Rh(III)-catalyzed rollover annulation of 1-phenyl-1Hpyrazoles with 1 equiv of alkynes could give 4,5-disubstituted pyrazolo[1,5-a]quinolones (Scheme 1a, right), while the nonrollover annulation of 1-phenyl-1H-pyrazoles with 2 equiv of alkynes afforded 1-(5,6,7,8-tetrasubstituted naphthalen-1-yl)-1H-pyrazoles as the products under slightly modified conditions (Scheme 1a, left).^{6a} In a similar way, Dong and co-workers described the selective synthesis of different 7azaindole derivatives in the annulation of 7-azaindoles with alkynes (Scheme 1b).^{6b} More recently, Choudhury et al. demonstrated the rollover and nonrollover annulations of imidazoles with alkynes (Scheme 1c).^{6h} Inspired by these elegant studies, we envisioned that tuning the reaction conditions may allow the structurally similar 2,2'-bipyridine N-oxides to undergo chelation-assisted rollover or nonrollover annulation with alkynes to give novel 2,2'-bipyridine derivatives that cannot be readily prepared using known methods. Following our continued interest in catalytic direct C-H functionalization of heteroarenes,⁸ herein, we demonstrate that

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Scheme 1. Rh(III)-Catalyzed Rollover and Nonrollover Annulation of Heteroarenes with Alkynes via Dual C–H Activation



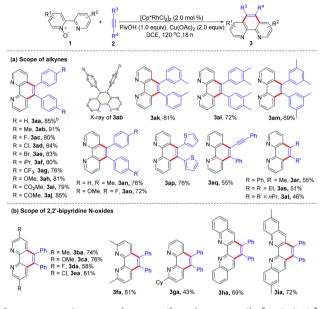
switchable rollover and nonrollover annulation of 2,2′bipyridine N-oxides with internal alkynes to access 5,6disubstituted-1,10-phenanthrolines and 5,6,7,8-tetrasubstituted-1-(pyridin-2-yl)isoquinoline 2-oxides, respectively, can be achieved by tuning the reaction conditions in the Rh(III) catalytic systems (Scheme 1d).

We commenced our studies by investigating the reaction of 2,2'-bipyridine N-oxide (1a) with 1,2-diphenylethyne (2a) (see the Supporting Information (SI)). We first attempted the reaction of 1a (0.5 mmol) and 2a (1.0 mmol) with $[Cp*RhCl_2]_2$ as the catalyst precursor and $Cu(OAc)_2$ as the oxidant in DCE at 120 °C. To our surprise, the reaction only gave a mixture of 5,6-diphenyl-1,10-phenanthroline (3aa) and 2,2'-bipyridine in 29% and 20% yields, respectively, and the expected product, 5,6-diphenyl-1,10-phenanthroline 1-oxide (3aa') was not observed (see Table S1, entry 1, in the SI). This result suggests that in situ deoxygenation might occur during the process. The yield of 3aa could be increased to 45% upon the addition of HOAc as the additive (Table S1, entry 2). Subsequent screening of additives revealed that PivOH turned out to be optimal and provided the product 3aa in 90% yield (Table S1, entry 5). It appears that deoxygenation of 1a was suppressed in the presence of acid, as no 2,2'-bipyridine was observed in these reactions (Table S1, entries 2-7). Further optimization showed that the yield of 3aa dropped significantly when replacing DCE with other solvents (Table S1, entries 8-21), and other oxidants were inferior to $Cu(OAc)_2$ (Table S1, entries 22-30). The reaction did not occur when omitting the oxidant, thereby ruling out the possibility that the N-oxide moiety acts as an internal oxidant in the reaction process (Table S1, entry 43).⁹ No reaction proceeded in the absence of [Cp*RhCl₂]₂, suggesting the indispensability of a rhodium catalyst for this reaction (Table S1, entry 44). Notably, neither a dried solvent, nor an inert atmosphere is required in this reaction. Thus, the optimal reaction conditions were set as $[Cp*RhCl_2]_2$ (2.0 mol %), PivOH (1.0 equiv), and Cu(OAc)_2 (2.0 equiv) in DCE at 120 $^{\circ}$ C for 18 h.

Under the optimized reaction conditions, we then examined the coupling reaction of 1a with various internal alkynes (see Scheme 2a). Symmetric diarylacetylenes 2b-2j bearing differ-

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Scheme 2. Annulation of 2,2'-Bipyridine N-Oxides with 1 equiv of Internal Alkynes^{*a*}



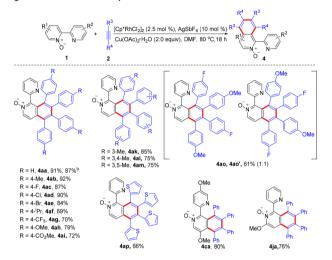
"Reaction conditions: 1 (0.2 mmol), 2 (0.22 mmol), $[Cp*RhCl_2]_2$ (2.0 mol%), $Cu(OAc)_2$ (2.0 equiv), PivOH (1.0 equiv), DCE (3.0 mL), 120 °C, 18 h. ^bIsolated yield in a gram scale reaction.

ent para-substituents on the aryl ring all smoothly underwent a reaction with 1a to afford the corresponding products (3ab-3aj) in 71%–91% yields; the electronic nature of these substituents has little effect on the reaction's efficiency. The structure of 3ab was confirmed by single-crystal X-ray diffraction (CCDC 1563903). Bis(3-substituted phenyl) acetylenes 2k-2m also exhibited good reactivity, delivering the target products (3ak-3am) in good yields. Similarly, unsymmetrical diarylacetylenes (2n,2o) proceeded well, allowing for the isolation of 3an and 3ao in 76% and 72% yields, respectively. 1,2-Di(thiophen-2-yl)ethyne (2p) has been proven to be a viable coupling partner to generate 3ap in 76% yield. The reaction also tolerated the use of 1,4-diphenylbutadivne (2q), 1-phenyl-1-propyne (2r), and dialkylacetylenes (2s)and 2t). To showcase the scalability of the current transformation, a gram-scale reaction of 1a and 2a was performed to give 3aa in 85% yield under the standard reaction conditions.

We next investigated the scope of 2,2'-bipyridine N-oxides (Scheme 2b). We found that 5,5'- and 6,6'-disubstituted 2,2'-bipyridine N-oxides (1b-1f) reacted smoothly with 2a to provide the corresponding products (3ba-3fa) in 58%-81% yields. In the case of 6-cyclohexyl-[2,2'-bipyridine] 1-oxide (1g), the product 3ga was isolated only in 43% yield, probably because of the steric effect. Delightfully, 2,2'-biquinoline 1-oxides (1h,1i) could also engage in the reaction to deliver the expected products (3ha, 3ia) in good yields.

During the course of optimization of the reaction conditions for the production of 3aa, we serendipitously found that the reaction of 1a and 2a in DCE at 100 °C with $[Cp*RhCl_2]_2$ as the catalyst precursor and Cu(OAc)₂ as the oxidant gave 5,6,7,8-tetraphenyl-1-(pyridin-2-yl)isoquinoline 2-oxide (4aa) (23% yield, 18 h) as the solo product, in the absence of any additive. This observation prompted us to further explore this transformation. After extensive optimization, we found that the combination of $[Cp*RhCl_2]_2$, AgSbF₆ and Cu(OAc)₂·H₂O could work effectively to catalyze the reaction of 2,2'-bipyridine N-oxide with 2 equiv of internal alkynes in DMF at 80 °C to give the corresponding 5,6,7,8-tetrasubstituted-1-(pyridin-2-yl)isoquinoline 2-oxides (4aa-4ai, 4al, and 4am) in good to excellent yields with good tolerance of functional groups in both coupling partners (see Scheme 3). We noted that

Scheme 3. Annulation of 2,2'-Bipyridine N-Oxides with 2 equiv of Internal Alkynes^a

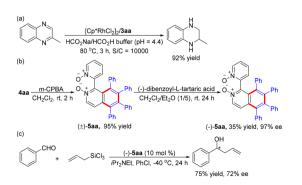


^{*a*}Reaction conditions: $[Cp*RhCl_2]_2$ (2.5 mol %), AgSbF₆ (10 mol %), 1 (0.2 mmol), 2 (0.5 mmol), Cu(OAc)₂·H₂O (2.0 equiv), DMF (3.0 mL), 80 °C, 18 h. ^{*b*}Isolated yield in a gram-scale reaction.

unsymmetrical alkyne 20 was also applicable, but a mixture of two inseparable regioisomers was isolated in an \sim 1:1 ratio (4ao and 4ao'). When 1,2-di(thiophen-2-yl)ethyne (2p) was used, 4ap was obtained in 66% yield. Reactions of 1c and 1j with 2a proceeded smoothly to produce 4ca and 4ja in 80% and 76% yields, respectively. Further study indicated that 1a could be transformed to 4aa in a gram-scale with 87% yield, demonstrating the scalability of this transformation.

To showcase the utility of these annulation products, we first attempted the application of **3aa** as the ligand in an Rh(III)-catalyzed transfer hydrogenation of quinoxaline in water, and the desired 1,2,3,4-tetraquinoxaline was obtained in 92% yield (Scheme 4a).¹⁰ Considering the potential of 2,2'-bipyridine N,N'-dioxides as promising chiral organocatalysts,^{3b} (-)-**5aa**

Scheme 4. Synthetic Application of 2,2'-Bipyridine N-Oxide Derivatives

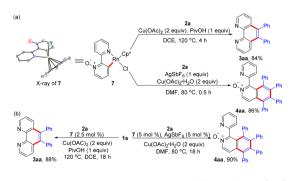


was then synthesized via N-oxidation of 4aa with *m*-CPBA to give racemic 5aa and subsequent optical resolution of racemic 5aa with (–)-dibenzoyl-L-tartaric acid (Scheme 4b). With (–)-5aa (10 mol%) as the catalyst, the allylation of benzaldehyde with trichloroallylsilane delivered the target product 1-phenylbut-3-en-1-ol in 75% yield and 72% ee (Scheme 4c).^{3b,7c} In addition, products 4 could be readily deoxygenated by PCl₃, and the measurement of absorption and emission spectra of these obtained 2,2'-bipyridines were performed (see the SI). Notably, 5,6,7,8-tetrakis(4-chlorophen-yl)-1-(pyridin-2-yl)isoquinoline (6ad) exhibited strong fluorescence emission at a broad range of 360–475 nm.

To address the question of in situ deoxygenation in the formation of products **3**, several control experiments were conducted (see the SI). No reaction was observed when treating 2,2'-bipyridine with **2a** under the standard reaction conditions. The HRMS analysis of the reaction system of **1a** and **2a** under the standard conditions confirmed the generation of 5,6-diphenyl-1,10-phenanthroline 1-oxide (**3aa'**) during the reaction process. Furthermore, when **3aa'** was subjected to the standard reaction conditions, **3aa** was obtained in 95% yield. Further study indicated that this catalytic deoxygenation of **3aa'** proceeded equally well in the absence of Cu(OAc)₂ and PivOH, but the reaction did not occur without using [Cp*RhCl₂]₂. These results show that DCE could act both as a solvent and a reductant in this Rh(III)-catalyzed deoxygenation.

In order to gain more insight into the reaction mechanisms, additional experiments were performed. The reversibility of C-H activation was first examined (see the SI). The observed H/ D scrambling in the deuterium labeling experiments between 1a and CD₃CO₂D under the standard conditions in the absence of alkyne suggested that a reversible C-H activation step may be involved in the two transformations. However, no deuteration of 1a, 3aa, and 4aa was observed when performing the same reactions in the presence of 2a, revealing the possible involvement of an irreversible alkyne insertion.¹¹ The reaction of la with [Cp*RhCl₂]₂ led to the formation of a fivemembered cyclometalated rhodium(III) intermediate 7, the structure of which was confirmed by single-crystal X-ray diffraction (XRD) studies (CCDC 1563904). The coordination of pyridine N-oxide oxygen atom to Rh(III) center was not observed.¹² The stoichiometric reactions of 2a with 7 afforded the products 3aa and 4aa in high yields in shorter reaction times (Scheme 5a). Further studies showed that 7 could catalyze the annulation reactions of 1a with 2a to give the expected products 3aa and 4aa in essentially the same yields to those obtained under the standard reaction conditions (Scheme 5b). These results clearly indicate the potential intermediacy of

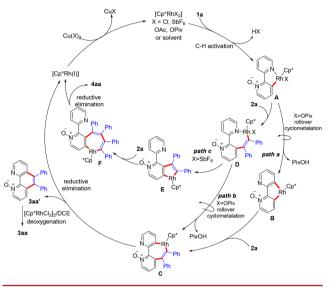
Scheme 5. Mechanistic Studies



the five-membered rhodacycle species in the two catalytic processes.

Based on these observations and the literature precedents, $^{5-7}$ possible catalytic cycles for both transformations are given in Scheme 6. First, cyclometalation of 1a generates a five-

Scheme 6. Plausible Mechanism



membered rhodacyclic intermediate A. In the presence of PivOH (X = OPiv), path a is favored to deliver a rhodacyclic intermediate B via a rollover cyclometalation, because of the strongly basic and coordinating nature of OPiv anion. Insertion of alkyne 2a gives a seven-membered rhodacycle C (path a), which undergoes reductive elimination to deliver product 3aa' and a Rh(I) species. The Rh(I) is reoxidized by $Cu(OAc)_2$ to regenerate the active Rh(III) catalyst for the next catalytic cycle, and Rh(III)-catalyzed deoxygenation of 3aa' affords the product 3aa. Alternatively, intermediate A possibly first reacts with 2a to afford a seven-membered rhodacycle D, and the presence of strongly basic and coordinating OPiv anion facilitates the following rollover cyclorhodation to produce C (path b). With weakly basic and less coordinating SbF_6 anion, the second intramolecular C-H activation of D occurs preferably to furnish a five-membered rhodacycle E (path c). After insertion of 2a once again and subsequent reductive elimination, product 4aa is obtained with the regeneration of the active Rh(III) catalyst.

In summary, we have achieved Rh(III)-catalyzed chelationassisted switchable annulation of 2,2'-bipyridine N-oxides with internal alkynes via double C–H activation. The neutral Rh(III) catalytic system favored rollover annulation to give 5,6disubstituted-1,10-phenanthrolines, whereas the cationic Rh-(III) catalytic system worked well for nonrollover annulation to deliver 5,6,7,8-tetrasubstituted-1-(pyridin-2-yl)isoquinoline 2oxides. The protocols have the advantages of high efficiency and functional group tolerance, excellent regioselectivity and broad substrate scope, enabling them to become attractive complements to the existing methods for the synthesis of useful 2,2'-bipyridine derivatives.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.or-glett.8b01434.

Detailed experimental procedures, characterization data, and copies of ¹H and ¹³C NMR spectra of products (PDF)

Accession Codes

CCDC 1563903 and 1563904 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, U.K.; fax: +44 1223 336033.

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Notes

The authors declare no competing financial interest.

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