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Regioselective reactions of *N*¹-alkyl-*N*²-(4-nitrophenyl)ethanediamide and acetylenic esters in the presence of *tert*-butyl isocyanide

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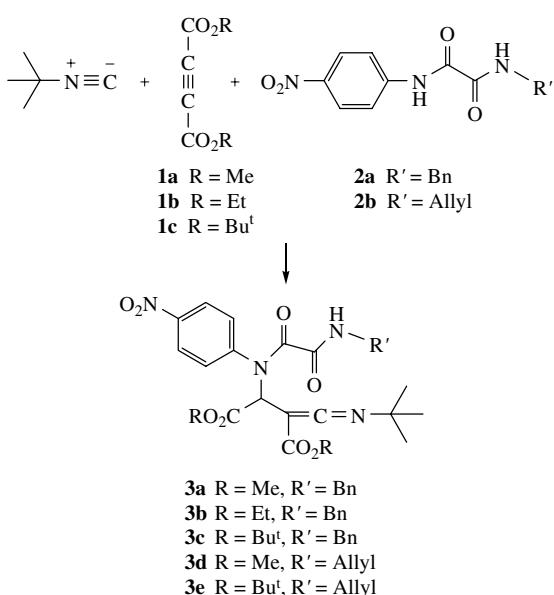
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The regioselective reactions of *N*¹-alkyl-*N*²-(4-nitrophenyl)ethanediamide with dialkyl acetylenedicarboxylates in the presence of *tert*-butyl isocyanide lead to dialkyl 2-{[2-(alkylamino)-2-oxoacetyl]-4-nitroanilino}-3-[(*tert*-butylimino)methylene]succinates in good yields.

Multicomponent reactions are of interest to combinatorial chemistry, for example, versatile isocyanide-based Ugi and Passerini reactions.^{1–4} In recent years, the synthetic applications of multi-functional heteroallenes have been widely investigated.^{5,6} In spite of extensive developments in the chemistry of modified ketenes and isocyanates,^{7,8} little attention has been paid to the synthesis of ketenimines.^{8–10}

Ketenimines are transient intermediates in many interconversions, especially, in elimination–addition processes and in the formation of heterocyclic ring systems.^{11–14} We report a regioselective one-pot synthesis of highly functionalised ketenimines **3**. Thus, the reaction of dialkyl acetylenedicarboxylates **1** with *N*¹-alkyl-*N*²-(4-nitrophenyl)ethanediamide **2** in the presence of *tert*-butyl isocyanide leads to dialkyl 2-{[2-(alkylamino)-2-oxo-



Scheme 1

acetyl]-4-nitroanilino}-3-[*(tert*-butylimino)methylene]succinates **3a–e** in good yields (Scheme 1).[†]

The reaction of *tert*-butyl isocyanide with electron-deficient acetylenic esters **1** in the presence of NH acids **2** proceeded in dichloromethane at room temperature and was complete within 24 h. The IR, ¹H NMR and ¹³C NMR spectra of the crude products clearly indicated the formation of stable ketenimines **3** (Scheme 1).

[†] The ¹H and ¹³C NMR spectra were measured with a Bruker DRX-500 AVANCE instrument in CDCl₃ as a solvent at 500.1 and 125.7 MHz, respectively. Mass spectra were recorded on a Finnigan-Matt 8430 mass spectrometer operating at an ionization potential of 70 eV.

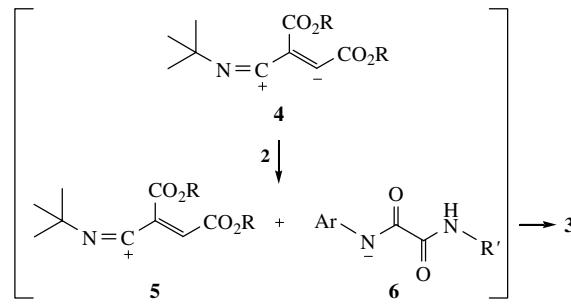
*Preparation of N¹-benzyl-N²-(4-nitrophenyl)ethanediamide **2a**:* benzylamine (1.07 g, 10 mmol) was added to a stirred solution of ethyl 2-(4-nitroanilino)-2-oxoacetate (2.38 g, 10 mmol) in CH₂Cl₂ (20 ml). The reaction mixture was stirred for 4 h. The product precipitated as a white powder, which was filtered off and washed with Et₂O. Product **2** was obtained as a white powder; yield 2.93 g (98%), mp 169–170 °C.

General procedure for the preparation of dimethyl 2-{[2-(benzylamino)-2-oxoacetyl]-4-nitroanilino}-3-[(tert*-butylimino)methylene]succinate **3a**:* to a stirred solution of 0.60 g N¹-benzyl-N²-(4-nitrophenyl)ethanediamide (2 mmol) and 0.28 g dimethyl acetylenedicarboxylate (2 mmol) in 6 ml CH₂Cl₂, was added dropwise at 0 °C for 10 min 0.45 g *tert*-butyl isocyanide (2 mmol) in 2 ml CH₂Cl₂. The reaction mixture was then allowed to warm up to room temperature and stand for 24 h. The solvent was removed under reduced pressure and the product was extracted with *n*-hexane-EtOAc (5:1) and crystallised at –20 °C. Pale yellow crystals; yield: 0.79 g (75%), mp 116–118 °C. ¹H NMR (CDCl₃, Me₄Si) δ_H: 1.36 (s, 9H, CMe₃), 3.69 and 3.80 (2s, 6H, 2OMe), 4.32 (d, 2H, NCH₂, ³J_{HH} 5 Hz), 5.58 (s, 1H, CH), 7.19–7.36 (m, 5H, Ph), 7.48 (br. s, 1H, NH), 7.54 and 8.22 (2d, 4H, C₆H₄, ³J_{HH} 9 Hz). ¹³C NMR (CDCl₃, Me₄Si) δ_C: 30.1 (CMe₃), 43.5 (NCH₂), 61.7 (N-CMe₃), 62.4 (CH), 124.2, 127.8, 127.9, 128.8 and 129.0 (9CH), 136.7, 146.9 and 147.1 (3C), 158.7, 161.1, 161.5, 168.7 and 169.7 (C=C=N and 4C=O). IR (KBr, ν/cm^{−1}): 3300 (NH), 2022 (C=C=N), 1738 and 1653 (C=O). MS, m/z (%): 524 (M⁺, 3), 299 (25), 91 (100), 65 (25). Found (%): C, 59.5; H, 5.4; N, 10.7. Calc. for C₂₆H₂₈N₄O₈ (524.5) (%): C, 59.54; H, 5.38; N, 10.68.

For **3b**: pale yellow crystals; yield 0.94 g (85%), mp 125–127 °C. ¹H NMR (CDCl₃, Me₄Si) δ_H: 1.28 and 1.36 (2t, 6H, 2Me, ³J_{HH} 7 Hz), 1.40 (s, 9H, CMe₃), 4.19 and 4.29 (2q, 4H, 2OCH₂, ³J_{HH} 7 Hz), 4.32 (d, 2H, NCH₂, ³J_{HH} 5 Hz), 5.52 (s, 1H, CH), 7.22–7.35 (m, 5H, Ph), 7.49 (br. s, 1H, NH), 7.59 and 8.26 (2d, 4H, C₆H₄, ³J_{HH} 9 Hz). ¹³C NMR (CDCl₃, Me₄Si) δ_C: 14.6 and 14.8 (2Me), 30.5 (CMe₃), 43.8 (NCH₂), 60.1 and 62.2 (C=C=N and N-CMe₃), 60.9 and 62.5 (2OCH₂), 62.6 (CH), 124.5, 128.2, 128.3, 129.1 and 129.4 (9CH), 137.4, 147.4 and 147.5 (3C), 159.2, 162.1, 162.8, 168.4 and 169.6 (C=C=N and 4C=O). IR (KBr, ν_{max}/cm^{−1}): 3310 (NH), 2020 (C=C=N), 1730 and 1675 (C=O). MS, m/z (%): 552 (M⁺, 2), 299 (28), 91 (100). Found (%): C, 60.9; H, 5.8; N, 10.1. Calc. for C₂₈H₃₂N₄O₈ (552.6) (%): C, 60.86; H, 5.84; N, 10.14.

The ¹H NMR spectrum of **3a** (CDCl₃) showed signals for *tert*-butyl (δ 1.36 ppm), methoxy (δ 3.69 and 3.80 ppm), methylene (δ 4.32 ppm, d, ³J_{HH} 5 Hz), methine (δ 5.58 ppm) and NH (δ 7.48 ppm) protons together with aromatic protons at δ 7.19–8.22 ppm. The ¹³C NMR spectrum of **3a** exhibited 20 sharp signals in agreement with the proposed structure. The sp²-hybridised carbon atom of the ketenimine residue appears at δ 59.1 ppm, as a result of strong electron delocalization. The structural assignments of compounds **3a–e** made on the basis of their NMR spectra were supported by their IR spectra. These compounds show strong absorption bands at about 2020 cm^{−1}.

Mechanistically, it is conceivable that the reaction involves the initial formation of 1:1 zwitterionic intermediate **4** between *tert*-butyl isocyanide and the acetylenic ester.^{2,4} The protonation of **4** by diamide **2** and the subsequent attack of the resulting nucleophile on positively charged ion **5** afforded ketenimine **3**. In this reaction, the stronger NH acid of **2** acts as a source of protons (Scheme 2).



Scheme 2

The three-component reaction of N¹-alkyl-N²-(4-nitrophenyl)-ethanediamide with dialkyl acetylenedicarboxylates in the presence of *tert*-butyl isocyanide provides a simple, one-pot and regioselective synthesis of polyfunctionalised ketenimines of potential synthetic interest.

For **3c**: pale yellow crystals; yield 1.10 g (90%), mp 140–142 °C. ¹H NMR (CDCl₃, Me₄Si) δ_H: 1.36, 1.40 and 1.48 (3s, 27H, 3CMe₃), 4.30 (d, 2H, NCH₂, ³J_{HH} 5 Hz), 5.48 (s, 1H, CH), 7.21–7.38 (m, 5H, Ph), 7.52 (br. s, 1H, NH), 7.56 and 8.22 (2d, 4H, C₆H₄, ³J_{HH} 9 Hz). ¹³C NMR (CDCl₃, Me₄Si) δ_C: 28.0, 28.3 and 30.0 (3CMe₃), 43.5 (NCH₂), 61.4 and 62.1 (C=C=N and N-CMe₃), 61.9 (CH), 80.9 and 83.2 (2OCMe₃), 124.4, 128.1, 128.3, 129.2 and 130.0 (9CH), 137.3, 147.1 and 147.5 (3C), 159.5, 161.9, 162.7, 167.9 and 169.1 (C=C=N and 4C=O). IR (KBr, ν_{max}/cm^{−1}): 3290 (NH), 2023 (C=C=N), 1725, 1681 and 1652 (C=O). MS, m/z (%): 608 (M⁺, 2), 299 (20), 91 (100), 57 (90). Found (%): C, 63.1; H, 6.6; N, 9.2. Calc. for C₃₂H₄₀N₄O₈ (608.7) (%): C, 63.14; H, 6.62; N, 9.20.

For **3d**: pale yellow crystals; yield 0.87 g (92%), mp 112–114 °C. ¹H NMR (CDCl₃, Me₄Si) δ_H: 1.35 (s, 9H, CMe₃), 3.70 and 3.76 (2s, 6H, 2OMe), 3.82 (m, 2H, NCH₂), 5.15 (m, 2H, =CH₂), 5.57 (s, 1H, CH), 5.75 (m, 1H, =CH), 7.33 (br. s, 1H, NH), 7.52 and 8.21 (2d, 4H, C₆H₄, ³J_{HH} 9 Hz). ¹³C NMR (CDCl₃, Me₄Si) δ_C: 30.1 (CMe₃), 41.7 (NCH₂), 51.9 and 52.9 (2OMe), 59.0 and 61.7 (C=C=N and N-CMe₃), 62.4 (CH), 117.2 (=CH₂), 124.1 (2CH), 129.0 (CH), 132.7 (2CH), 147.0 and 147.6 (2C), 158.5, 161.3, 163.3, 168.7 and 169.6 (C=C=N and 4C=O). IR (KBr, ν_{max}/cm^{−1}): 3335 (NH), 2022 (C=C=N), 1734, 1679 and 1654 (C=O). MS, m/z (%): 474 (M⁺, 3), 249 (100), 57 (20), 41 (30). Found (%): C, 55.7; H, 5.5; N, 11.8. Calc. for C₂₂H₂₆N₄O₈ (474.5) (%): C, 55.69; H, 5.52; N, 11.81;

For **3e**: pale yellow crystals; yield 1.05 g (94%), mp 146–148 °C. ¹H NMR (CDCl₃, Me₄Si) δ_H: 1.46, 1.47 and 1.54 (3s, 27H, 3CMe₃), 3.80 (m, 2H, NCH₂), 5.20 (m, 2H, =CH₂), 5.34 (s, 1H, CH), 5.53 (m, 1H, =CH), 7.39 (br. s, 1H, NH), 7.58 and 8.25 (2d, 4H, C₆H₄, ³J_{HH} 9 Hz). ¹³C NMR (CDCl₃, Me₄Si) δ_C: 28.4, 28.7 and 30.5 (3CMe₃), 42.1 (NCH₂), 61.7 and 62.2 (C=C=N and N-CMe₃), 62.2 (CH), 81.0 and 83.3 (2OCMe₃), 117.5 (=CH₂), 124.3 (2CH), 129.9 (CH), 133.2 (2CH), 147.2 and 147.4 (2C), 159.2, 161.5, 161.9, 167.6 and 168.9 (C=C=N and 4C=O). IR (KBr, ν_{max}/cm^{−1}): 3330 (NH), 2020 (C=C=N), 1730, 1680, and 1650 (C=O). MS, m/z (%): 558 (M⁺, 2), 333 (25), 57 (100), 41 (35). Found (%): C, 60.2; H, 6.8; N, 10.0. Calc. for C₂₈H₃₈N₄O₈ (558.6) (%): C, 60.20; H, 6.86; N, 10.03.

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