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Access to Z-Selective 1,3-Enynes via Ni-Catalyzed Intermolecular Cross-Alkylalkynylation of Terminal Alkynes

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 Cite This: Org. Lett. 2021, 23, 5186–5191
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 ABSTRACT: Access to 1,3-conjugated enynes with defined stereo Image: Conjugated enynes with defined stereo Image: Conjugated enynes with defined stereo

selectivity is highly desirable and challenging. Herein, we report a facile synthesis of stereodefined 1,3-conjugated enynes via Nicatalyzed intermolecular cross-alkylalkynylation of alkynes with unsaturated carbonyl compounds and alkynes or alkynyl silicates. The operational simple protocol proceeds at room temperature and

Pot a Ni-Ni-R¹ + $R^2 \xrightarrow{O} R^3$ + $R^4 \xrightarrow{room temp.}$ $R^3 = H, alkyl$ X = H or silicate • 40 examples • exclusive Z-selectivity I mild conditions and broad scope

tolerates a wide range of functional groups, providing an attractive alternative to carbonyl-tethered trisubstituted conjugated 1,3enynes from easily accessible starting materials.

onjugated 1,3-envnes are widely found substructures in natural products, pharmaceuticals, and functional materials.¹ They also serve as precursors for diverse functional groups and versatile intermediates to build molecular complexity.² To date, significant advances have been achieved in the synthesis of 1,3-enynes,³ such as the Wittig reaction or Horner-Wadsworth-Emmons (HWE) reaction,⁴ transition-metalcatalyzed cross-couplings of alkynes with alkenes,⁵ crossdimerization of alkynes,⁶ dehydration of propargylic alcohols, and others.⁸ However, the existing methods suffer from several drawbacks, such as harsh reaction conditions, inconvenience of accessing advanced starting materials, difficulty to control the regio- or/and stereoselectivity, and the use of expensive and toxic precious metal catalysts. To this regard, the direct synthesis of 1,3-envnes from readily available and stable starting materials in a stereodefined manner is challenging and highly desirable. Transition-metal-catalyzed carbo-alkynylation of alkynes has emerged as one of the promising ways to construct conjugated 1,3-enynes. Jiang and co-workers developed a rhodium-/copper-catalyzed 1,1-alkylalkynylation of terminal alkynes with α -diazoketones via a dediazotized carbene C-H insertion process to access 1,3-envnes in an Eselectivity (Scheme 1a).9 Lv group reported a copper-catalyzed 1,1-alkylalkynylation of alkynes with α -haloacetamides to give E-1,3-enynes (Scheme 1a).¹⁰ Zhu and co-workers reported a tandem trans-carbohalogenation/Sonogashira coupling reaction, furnishing substituted 1,3-envnes with E-selectivity (Scheme 1b).¹¹ Unfortunately, the above-mentioned methods are limited to the homodimerization of alkynes to give 1,3enynes with two identical substituents. Moreover, only E-1,3enynes were accessible. On the other hand, nickel-catalyzed intermolecular dicarbofunctionalizations of alkynes have become one of the most straightforward and effective ways to synthesize multisubstituted alkenes.¹² However, few examples utilizing this strategy to construct 1,3-envnes have been developed, probably due to the competitive oligomeriza-

Scheme 1. Strategies to Access Stereodefined 1,3-Enynes by Difunctionalization of Alkynes



 Received:
 May 23, 2021

 Published:
 June 14, 2021



Letter

tion of alkynes. Recently, Koh and Zhao¹³ reported a nickelcatalyzed three-component reductive alkylalkynylation of terminal alkynes with alkyl precursors and alkynyl bromides to deliver E-1,3-envnes (Scheme 1c). The Ikeda group reported a three-component reaction of nickel-catalyzed cross-coupling of alkynes, enones with organostannane, organozinc, or organoaluminum reagents, providing access to 1,3-enynes in Z-selectivity.¹⁴ However, the methods required employing a stoichiometric amount of toxic and sensitive reagents, such as DIBAL-H, organometallic reagents (organotin/zinc/aluminum), and TMSCl, under harsh conditions, resulting in limited scope with narrow functional group tolerance (Scheme 1d). As our interest continued in earthabundant metal-catalyzed selective transformations,¹⁵ we envisioned building stereodefined 1,3-envnes via intermolecular oxidative cyclometalation of alkynes and enones, followed by a selective cross-coupling reaction with an easily available and user-friendly alkynyl precursor. Herein, we reported the regio- and stereoselective synthesis of 1,3-enynes by Nicatalyzed intermolecular cross-coupling of alkynes with enones and alkynyl silicates/alkynes at room temperature (Scheme 1e), enabling the access to 1,3-envnes with exclusive Zconfiguration under mild conditions with user-friendly reagents.

To test the feasibility of the reaction, we set out to evaluate the reaction conditions using phenylacetylene (1a), pent-3-en-2-one (2a, mixture of isomers), and trimethylsilylacetylene silicate (3a) as the prototype substrates. After extensive optimization of the reaction parameters, we defined the use of NiI₂ (10 mol %) as catalyst, zinc (30 mol %) as reductant, and methanol (5 μ L) as additive in DMF (0.1 M) at room temperature as the standard conditions, affording the threecomponent cross-coupling product 4,5-enynone 4a in 89% isolated yield (Table 1, entry 1). The reaction proceeded smoothly in the absence of methanol to give 4a in 81% yield (Table 1, entry 2). Alternating the amount of methanol decreased the efficiency of the reaction, delivering 4a in 83% and 90% yields, respectively (Table 1, entries 3 and 4). No desired product was obtained without zinc (Table 1, entry 5). Decreasing the amount of zinc led to lower yield of 4a (Table 1, entry 6). Comparable yield of 4a was obtained with higher loading of zinc (Table 1, entry 7). The use of Mn instead of Zn delivered 4a in 11% yield (Table 1, entry 8). The selection of nickel precatalyst is also crucial to the reaction (Table 1, entries 9-13). The reaction proceeded smoothly in DMA, giving the desired product 4a in 75% yield (Table 1, entry 14). Other solvents, such as tetrahydrofuran, acetonitrile, and dioxane, were not suitable for this sequential three-component transformation (Table 1, entries 15-17). Alternating the concentration and stoichiometry of 3a delivered inferior results (Table 1, entries 18-20).¹⁶ Conducting the reaction at higher concentration led to a heterogeneous mixture, which significantly decreased the yield of 4a (Table 1, entry 18). Diluting the concentration of the reaction led to slower reaction rate, delivering 4a in 39% yield (Table 1, entry 19).

With the optimized conditions in hand, we turned to evaluate the scope of this reaction (Figure 1). The reaction tolerated a wide range of functional groups with different substitution patterns for alkynes, enones, and alkynyl silicates. First, the scope of the terminal alkyne was tested. Electrondonating- and electron-withdrawing-substituted aryl alkynes were all good substrates for this transformation. Aryl terminal alkynes with *para-, meta-,* or *ortho*-substitution patterns could

Table 1. Evaluation of the Reaction Conditions^a

Ph t 1a	Me ² TMS O + Si(OMe) ₃ Nil ₂ (10 mol%) MeOH (5 μL) DMF (0.1 M), rt 18 h 'standard conditions'	Ph H 4a
entry	variation from "standard conditions"	yield of 4a
1	none	94% (89%) ^b
2	no MeOH	81%
3	3 μ L of MeOH	83%
4	8 μ L of MeOH	90%
5	no Zn	0%
6	10 mol % of Zn	27%
7	50 mol % of Zn	91%
8	Mn instead of Zn	11%
9	no NiI ₂	0%
10	NiBr ₂ instead of NiI ₂	23%
11	NiBr ₂ ·glyme instead of NiI ₂	26%
12	NiCl ₂ instead of NiI ₂	20%
13	$Ni(OAc)_2$ instead of NiI_2	0%
14	DMA instead of DMF	75%
15	THF instead of DMF	2%
16	MeCN instead of DMF	trace
17	dioxane instead of DMF	0%
18	DMF (0.2 M)	11%
19	DMF (0.05 M)	39%
20	3a (2 equiv)	77%

^{*a*}The reaction was carried out using 1a (0.15 mmol), 2a (0.1 mmol), and 3a (0.3 mmol) under indicated conditions. Yield was determined by GC using *n*-dodecane as the internal standard. ^{*b*}Isolated yield after flash chromatography. TMS = trimethylsilyl.

be converted to corresponding ketone-tethered Z-1,3-envnes in moderate to good yields (4b-4o). It is noteworthy that alkynyl arylhalides were compatible in this reaction with halides intact (4f, 4g, 4j, 4m, and 4n), leaving halides as a chemical handle for further elaboration. Multisubstituted aryl and fused aromatic alkynes were successfully transformed to corresponding Z-1,3-enynes (4p and 4q) in 68% and 76% vields, respectively. Heteroaryl alkynes were also good subtrates for this reaction, delivering heteroaromatic-containing Z-1,3-enynes (4r and 4s) in 62% and 68% yields. The Zconfiguration of the product was confirmed by the NOE of 4p. Unfortunately, the use of aliphatic alkynes failed to deliver the desired envne products. Next, the scope of unsaturated alkenes was examined. Acyclic 2,3-enones with different substitution patterns were all well-tolerated under the reaction conditions, giving the desired products (5a-5d) in 75%-85% yields. Five-, six-, and seven-membered cyclic 2,3-enones were good substrates in the reaction, furnishing cyclic ketone tethered conjugated Z-1,3-envnes (5e-5g) in 55-70% yields. Then, the scope of alkynyl silicates was investigated. Aliphatic alkyne, silyl alkyne, and aryl alkyne derived silicates could be coupled with terminal alkynes and enones to deliver diverse conjugated 2,3-envnes with exclusive Z-selectivity in 60%-94% yields (6a-6g). Notably, this protocol provides direct access to 1,3enynes with similar substituents at 2- and 4-positions (6e-6g), which are difficult to access using other methods.

Next, we tested the reactions of alkylalkynylation of alkynes with enones using terminal alkynes as both alkene and alkyne equivalents (Figure 2). Transition-metal-catalyzed dimerization of terminal alkynes provided efficient and atom-economic synthetic routes to 1,3-enynes.⁶ However, the regio- and

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Figure 1. Scope for the stereodefined synthesis of 1,3-enynes from alkynyl silicates. The reaction was run on a 0.2 mmol scale. For detailed reaction conditions, see Table 1 (entry 1). Isolated yield after flash chromatography is shown. TBS = tert-butyldimethylsilyl, TES = triethylsilyl.

stereoselectivity control remains challenging, and the homodimerization of terminal alkynes may lead to different regioisomers. It is noteworthy that 2 equiv of alkynes underwent sequential coupling with 1 equiv of enones to give ketone-tethered conjugated 1,3-envnes with good functional group tolerance. Terminal aryl alkynes with electrondonating or electron-withdrawing substituents were all tolerated in this transformation, furnishing a wide range of conjugated 1,3-envnes with Z-selectivity in moderate to good yields (7a-7j). The Z-configuration of the product was further confirmed by the NOE of 7b and X-ray diffraction analysis of 7l. Cyclic enones with different ring size were all compatible under the reaction conditions, furnishing the desired 1,3-enyne products in synthetic useful yields (7k and 7l). Conjugated enals with diverse α - or β -substitution patterns could be converted to the corresponding aldehydes with conjugated Zenynes (7m and 7n) in 55% and 61% yields, respectively.

To shed light on the mechanism of this reaction, we set up a series of control experiments to probe the reaction mechanism (Figure 3). First, deuterium-labeling experiments for three-

component reaction were investigated. Reactions of an alkyne and an enone with an alkynyl silicate in the presence of deuterated methanol under otherwise standard conditions were conducted (Figures 3a and b). The reactions proceeded smoothly to deliver the desired 1,3-dienynes (8 and 9) in 68% and 77% yields, respectively. The reaction of a deuterated alkyne, an enone, and an alkynyl silicate in the presence of methanol was conducted to give 10 in 62% yield (Figure 3c). Deuterium was found at the α -position of the carbonyl group in the presence of CD₃OD, while no deuterium was detected at the α -position of the carbonyl group when CH₃OH was used as the additive. Deuterium was detected on alkene when deuterated alkyne was used, while no deuterium was involved on alkenes using regular alkynes. Next, a deuterium-labeling experiment for a two-component reaction was conducted (Figure 3d). The reaction of a deuterated alkyne with an enone delivered the desired product 11 in 88% yield with 95% deuterium incorporation on the alkene moiety and partial deuterium incorporation at the α -proton of the carbonyl group. These results indicated the proton on an alkene moiety





Figure 2. Scope for the stereodefined synthesis of 1,3-enynes from terminal alkynes. The reaction was conducted using alkyne 1 (0.8 mmol) and enone 2 (0.2 mmol) under the indicated conditions.

exclusively originated from the terminal alkyne, and the α -proton of the carbonyl group was exchangeable with the surroundings.

Based on the mechanistic results and the literature,¹⁴ a plausible mechanism of this reaction is depicted in Figure 3e. First, Ni(0) species from Ni(II) by reduction initiated metallacyclic intermediate A via cross-oxidative cyclometalation between an alkyne and an enone. C-Bound intermediate A could equilibriate with O-bound intermediate B. In the presence of a proton source (methanol or terminal alkyne), intermediate C was formed via protonation, which could further undergo transmetalation with an alkynyl silcate or a teminal alkyne to give intermediate **D**. In the case of an alkynyl silicate, the formation of a pentavalent silicate intermediate might be essential to transfer the alkynyl group to nickel. In the case of a terminal alkyne, a deprotonated acetylide might directly attack a Ni(II) center. D underwent reductive elimination to form the final product along with the Ni(0)intermediate to complete the catalytic cycle.

In summary, an efficient synthesis of conjugated 1,3-enynes via Ni-catalyzed intermolecular cross-coupling reaction between terminal alkynes, α,β -unsaturated carbonyl compounds, and alkynes or alkynyl silicates was reported. The reaction features intermolecular regio- and stereoselective alkylalkynylation of alkyne from easily available and user-friendly starting materials, providing a highly flexible method for introduction of trisubstituted 1,3-enynes to the β -C of carbonyls in exclusive *Z*-selectivity from easily available starting materials.



Figure 3. Mechanistic study and proposed mechanism for the reaction.

ASSOCIATED CONTENT Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.orglett.1c01728.

General procedures, characterization of new compounds, and copies of NMR spectra (PDF)

Accession Codes

CCDC 2086578 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

We sincerely acknowledge NSFC (21801126 and 21971101), Guangdong Basic and Applied Basic Research Foundation (2019A1515011976), The Pearl River Talent Recruitment Program (2019QN01Y261), and Guangdong Provincial Key Laboratory of Catalysis (No. 2020B121201002) for financial support. We acknowledge the assistance of SUSTech Core Research Facilities. We thank Dr. Xiaoyong Chang (SUSTech) for X-ray crystallographic analysis of 7l.

DEDICATION

This paper is dedicated to the memory of Prof. Ei-ichi Negishi.

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(16) For more details on condition optimization, see the Supporting Information.

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