

2,7-Diborylanthracene as a Useful Building Block for Extended π -Conjugated AromaticsRyota Ozawa,¹ Kenji Yoza,² and Kenji Kobayashi*¹¹Department of Chemistry, Faculty of Science, Shizuoka University, 836 Ohya, Suruga-ku, Shizuoka 422-8529²Bruker axs, 3-9-B Moriya, Kanagawa-ku, Yokohama, Kanagawa 221-0022

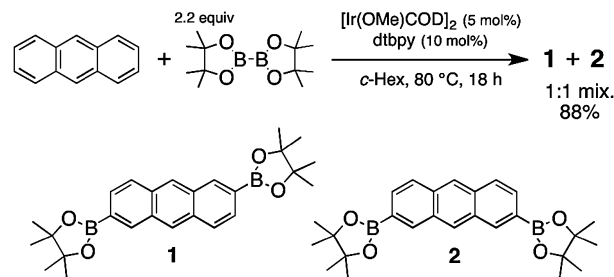
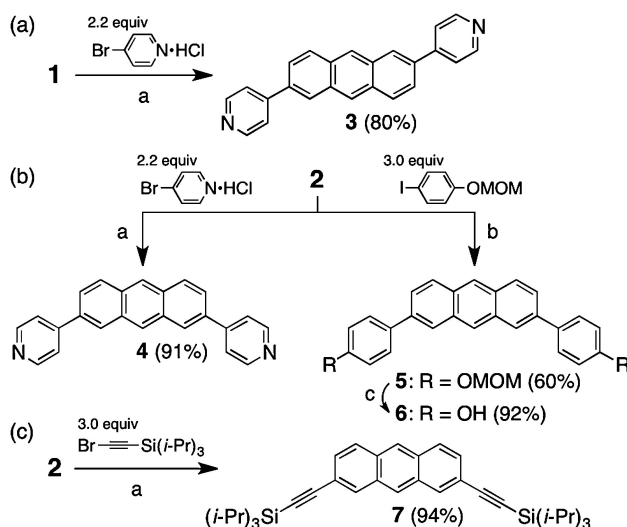
(Received May 11, 2011; CL-110401; E-mail: skkobay@ipc.shizuoka.ac.jp)

Ir-catalyzed direct diborylation of anthracene produced a 1:1 mixture of 2,6- and 2,7-diborylanthracenes, which could be separated by recrystallization. The Suzuki–Miyaura cross-coupling using 2,7-diborylanthracene gave extended π -conjugated 2,7-disubstituted anthracene derivatives as building blocks for macrocycles. A 1:1 mixture of 2,7-di(4-pyridyl)anthracene and [Pd(dppp)(OTf)₂] self-assembled into a [3 + 3] macrocycle.

Ir-catalyzed direct borylation of aromatic compounds using bis(pinacolato)diboron is a very useful reaction for the synthesis of arylboronic acid pinacol esters,¹ which are reagents for the Suzuki–Miyaura cross-coupling reaction.² Marder et al. first observed the synthesis of 2,6- and 2,7-diborylnaphthalenes.³ Recently, using this reaction, we have reported on the selective synthesis of 2,8- and 2,9-diboryltetracenes, which are very useful building blocks for the regiospecific synthesis of extended π -conjugated tetracene derivatives of organic semiconductors used in organic field-effect transistors.⁴

It is well known that the *meta*-phenylene unit and its derivatives are useful building blocks for the construction of extended π -conjugated macrocycles⁵ and self-assembled macrocycles.⁶ An extended version of the *meta*-phenylene unit is the 2,7-anthracene unit.^{7,8} Anthracene derivatives are materials for use in optoelectronics.⁹ In addition, 2,7-dibromoanthracene should be useful as a building block for the synthesis of extended π -conjugated anthracene macrocycles. However, the synthesis of 2,7-dibromoanthracene is difficult, because it requires a six-step synthetic procedure from anthrone and only gives a 9% yield in total.¹⁰ Our work was concerned with the efficient synthesis of 2,7-difunctionalized anthracenes. Herein, we report on: (1) the Ir-catalyzed direct diborylation of anthracene to give a 1:1 mixture of 2,6- and 2,7-diborylanthracenes **1** and **2**, (2) the Suzuki–Miyaura cross-coupling using **2** to give extended π -conjugated 2,7-disubstituted anthracene derivatives, and (3) the self-assembly of 2,7-dipyridylanthracene and [Pd(dppp)(OTf)₂] into a [3 + 3] macrocycle as an application of this reaction.

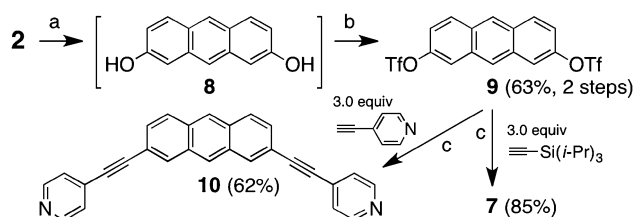
The reaction of anthracene with 2.2 equiv of bis(pinacolato)diboron in the presence of [Ir(OMe)(COD)]₂ (5 mol %) and 4,4'-di-*tert*-butyl-2,2'-bipyridine (dtbpy, 10 mol %) in cyclohexane at 80 °C for 18 h under Ar in the dark gave a 1:1 mixture of 2,6- and 2,7-bis[(pinacolato)boryl]anthracenes **1** and **2** in 88% yield (Scheme 1). Recrystallization of an initial 1:1 mixture of **1** and **2** followed by repeated recrystallizations of the **2**-rich mixture from hot toluene gave pure **2** (41% yield in total from anthracene). Similarly, repeated recrystallizations of the **1**-rich mixture from hot benzene yielded pure **1** (39% yield in total from anthracene).¹¹ The NMR spectra of both products revealed the positions of the diborylation of anthracene. In the ¹H NMR spectra, **1** and **2** showed two and three singlet aromatic

Scheme 1. Synthesis of diborylanthracenes **1** and **2**.

Scheme 2. Synthesis of **3**–**7**. (a) Method A: [Pd(PPh₃)₄] (4 mol %), Na₂CO₃ (10 equiv), toluene–EtOH–H₂O (4:2:1 for **3** and **4** or 10:5:1 for **7**), 80 °C, 42 h for **3** and **4** and 13 h for **7**. (b) Method B: [Pd(OAc)₂] (4 mol %), SPhos (4 mol %), K₃PO₄ (4 equiv), THF–H₂O (9:1), 60 °C, 42 h. (c) 2 M HCl–*i*-PrOH–THF (1:2:2), 40 °C, 22 h.

signals, respectively, and in the ¹³C NMR spectra, **1** and **2** exhibited six and seven aromatic signals, respectively (the carbon atom attached to the boron atom was not observed).¹¹

The 2,6- and 2,7-diborylanthracenes **1** and **2** are very useful building blocks for the regiospecific synthesis of extended π -conjugated anthracenes. The Suzuki–Miyaura cross-coupling reaction of **1** and **2** with 2.2 equiv of 4-bromopyridine hydrochloride catalyzed by [Pd(PPh₃)₄] (4 mol %) in the presence of Na₂CO₃ (10 equiv) in toluene–EtOH–H₂O at 80 °C (Method A) for 42 h gave 2,6- and 2,7-di(4-pyridyl)anthracenes **3** and **4** in 80% and 91% yields, respectively (Schemes 2a and 2b). The reaction of **2** with 4-bromopyridine catalyzed by Pd(OAc)₂ (4 mol %), SPhos (4 mol %),¹² and K₃PO₄ (4 equiv) in



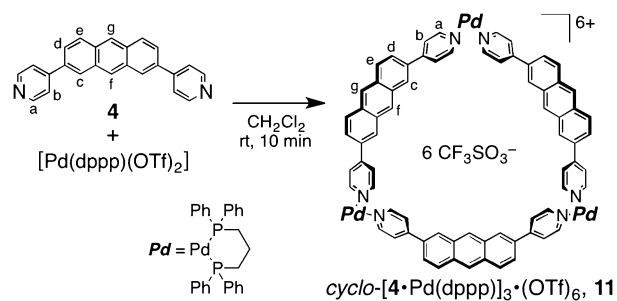
Scheme 3. Synthesis of 7–10. (a) Oxone (4 equiv), THF–acetone–H₂O (10:2:1), rt, 3 h. (b) Tf₂O (3 equiv), Et₃N (4 equiv), CH₂Cl₂, 0 °C to rt, 18 h. (c) [PdCl₂(PPh₃)₂] (8 mol %), CuI (8 mol %), PPh₃ (16 mol %), toluene–Et₃N (1:1), rt, 2 h to 60 °C, 2 h to 80 °C, 12 h.

THF–H₂O at 60 °C (Method B) for 42 h produced **4** in 41% yield. On the other hand, the coupling reaction of **2** with 3 equiv of 1-iodo-4-(methoxymethoxy)benzene based on Methods A and B gave **5** in 38% and 60% yields, respectively (Scheme 2b). Compound **5** was hydrolyzed by HCl (aq) to produce 2,7-bis(4-hydroxyphenyl)anthracene (**6**) in 92% yield. The Suzuki–Miyaura cross-coupling reaction can be applied to 1-bromoalkynes,¹³ which is an alternative method to the Sonogashira cross-coupling reaction. Thus, the coupling reaction of **2** with 3 equiv of bromo(triisopropylsilyl)acetylene^{13b} using Method A gave 2,7-bis[(triisopropylsilyl)ethynyl]anthracene (**7**) in 94% yield (Scheme 2c).

The transformation of 2,7-diborylanthracene **2** into 2,7-bis-triflate **9** would further enhance the utility of **2** as a synthetic building block, because **9** can be used in the Sonogashira cross-coupling (Scheme 3). Thus, the reaction of **2** with Oxone[®] produced crude 2,7-dihydroxyanthracene (**8**),^{14a} which was treated with Tf₂O to give **9** (63% yield in two steps).^{14b} The Sonogashira cross-coupling reaction of **9** with 3 equiv of 4-ethynylpyridine and triisopropylsilylacetylene gave 2,7-bis-(4-pyridylethynyl)anthracene (**10**) and **7** in 62% and 85% yields, respectively.

2,7-Di(4-pyridyl)anthracene (**4**) serves as a building block for self-assembled macrocycles.^{6,15} A 1:1 mixture of **4** and square-planar [Pd(dppp)(OTf)₂] self-assembled into a [3 + 3] macrocycle, *cyclo*-[4•Pd(dppp)]₃•(OTf)₆ (**11**), via Pd–pyridyl coordination bonds (Scheme 4). The ¹H NMR spectrum of a 1:1 mixture of **4** and [Pd(dppp)(OTf)₂] in CD₂Cl₂ at room temperature after 10 min showed a highly symmetric new species, along with the disappearance of **4** and [Pd(dppp)(OTf)₂], suggesting the formation of **11** (Figure 1). The pyridyl α-proton was shifted downfield by 0.33 ppm relative to that of free **4**, indicative of Pd–pyridyl coordination bonds being formed. The UV–vis spectrum also supported the formation of a Pd–pyridyl bond, where the longest wavelength λ_{max} of this mixture in CH₂Cl₂ was red-shifted by 6 nm relative to that of free **4**.¹¹ A 2:1 mixture of **4** and [Pd(dppp)(OTf)₂] gave a mixture of **11**, free **4**, and unknown species (Figure 1d), indicating that **11** is not stable. Finally, the molecular structure of **11** was confirmed from single-crystal X-ray diffraction analysis, although the final *R* indices were poor because the triflate ions and inclusion solvents were highly disordered.^{11,16}

Single crystals of *cyclo*-[4•Pd(dppp)]₃•(OTf)₆ (**11**) suitable for X-ray diffraction analysis were obtained by slow evaporation of a CHCl₃ solution of **11**. Figure 2 shows the molecular structure of **11**, wherein there are two types of slightly different



Scheme 4. Self-assembly of **4** and [Pd(dppp)(OTf)₂] into **11**.

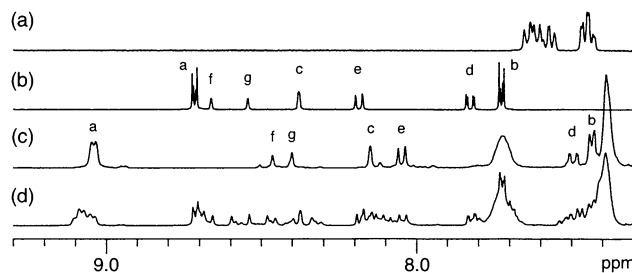


Figure 1. ¹H NMR spectra of (a) [Pd(dppp)(OTf)₂], (b) **4**, (c) a 1:1 mixture of **4** and [Pd(dppp)(OTf)₂] (formation of **11**), and (d) a 2:1 mixture of **4** and [Pd(dppp)(OTf)₂] in CD₂Cl₂ at rt.

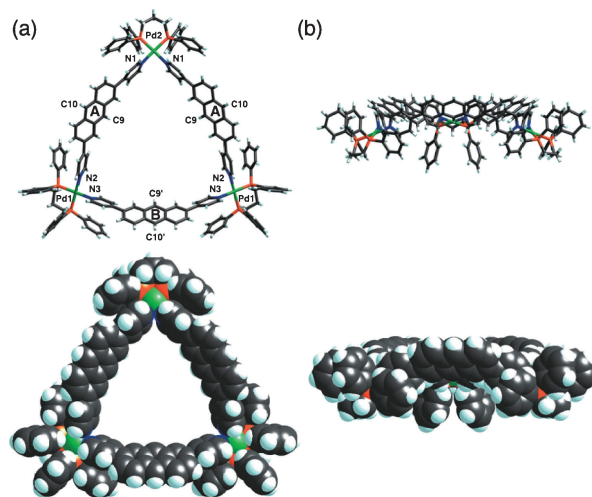


Figure 2. X-ray crystal structure of **11**: (a) front and (b) side views. Triflate ions are omitted for clarity.

conformations for both the **4** unit (ant-A and ant-B) and the [Pd(dppp)] unit (Pd1 and Pd2) to form a triangular structure. The lengths of sides of the triangular cavity are Pd1...Pd2 = 18.8 Å and Pd1...Pd1 = 18.4 Å. The distances between the anthracene rings are C10...C10 (between the ant-A rings) = 15.9 Å and C10...C10' (between the ant-A and ant-B rings) = 15.6 Å for the upper rim and C9...C9 = 11.8 Å and C9...C9' = 12.0 Å for the lower rim.

In summary, we have demonstrated the Ir-catalyzed direct diborylation of anthracene to form 2,6- and 2,7-diborylanthracenes **1** and **2**. The Suzuki–Miyaura cross-coupling using **2** gave extended π-conjugated 2,7-disubstituted anthracene derivatives as building blocks for macrocycles. A study on the synthesis of

an extended π -conjugated anthracene macrocycle based on **2**, **7**, and **9** is currently in progress.

This work was supported in part by the Asahi Glass Foundation.

This paper is in celebration of the 2010 Nobel Prize awarded to Professors Richard F. Heck, Akira Suzuki, and Ei-ichi Negishi.

References and Notes

- 1 a) T. Ishiyama, J. Takagi, K. Ishida, N. Miyaaura, N. R. Anastasi, J. F. Hartwig, *J. Am. Chem. Soc.* **2002**, *124*, 390. b) T. Ishiyama, Y. Nobuta, J. F. Hartwig, N. Miyaaura, *Chem. Commun.* **2003**, 2924. c) T. Ishiyama, *J. Synth. Org. Chem., Jpn.* **2005**, *63*, 440.
- 2 N. Miyaaura, A. Suzuki, *Chem. Rev.* **1995**, *95*, 2457.
- 3 D. N. Coventry, A. S. Batsanov, A. E. Goeta, J. A. K. Howard, T. B. Marder, R. N. Perutz, *Chem. Commun.* **2005**, 2172.
- 4 T. Kimoto, K. Tanaka, Y. Sakai, A. Ohno, K. Yoza, K. Kobayashi, *Org. Lett.* **2009**, *11*, 3658.
- 5 a) S. Lahiri, J. L. Thompson, J. S. Moore, *J. Am. Chem. Soc.* **2000**, *122*, 11315. b) Y. Tobe, N. Utsumi, K. Kawabata, A. Nagano, K. Adachi, S. Araki, M. Sonoda, K. Hirose, K. Naemura, *J. Am. Chem. Soc.* **2002**, *124*, 5350. c) W. Zhang, J. S. Moore, *Angew. Chem., Int. Ed.* **2006**, *45*, 4416, and references cited therein.
- 6 a) S. Leininger, B. Olenyuk, P. J. Stang, *Chem. Rev.* **2000**, *100*, 853. b) M. Fujita, M. Tominaga, A. Hori, B. Therrien, *Acc. Chem. Res.* **2005**, *38*, 369. c) T. Murase, S. Sato, M. Fujita, *Angew. Chem., Int. Ed.* **2007**, *46*, 5133.
- 7 For 2,7-naphthalene unit for macrocycles, see: a) Y. Takahira, H. Sugiura, M. Yamaguchi, *J. Org. Chem.* **2006**, *71*, 763. b) M. Yamaguchi, *J. Synth. Org. Chem., Jpn.* **2011**, *69*, 17.
- 8 For 9,10- and 1,8-anthracene units for macrocycles, see: a) K. Miyamoto, T. Iwanaga, S. Toyota, *Chem. Lett.* **2010**, *39*, 288. b) K. Miki, M. Fujita, Y. Inoue, Y. Senda, T. Kowada, K. Ohe, *J. Org. Chem.* **2010**, *75*, 3537. c) S. Toyota, *Chem. Lett.* **2011**, *40*, 12.
- 9 a) J. E. Anthony, *Chem. Rev.* **2006**, *106*, 5028. b) W. J. Yang, M. S. Seo, X. Q. Wang, S.-J. Jeon, B. R. Cho, *J. Fluoresc.* **2008**, *18*, 403. c) J. N. Moorthy, P. Venkatakrishnan, P. Natarajan, D.-F. Huang, T. J. Chow, *J. Am. Chem. Soc.* **2008**, *130*, 17320. d) A. Marrocchi, F. Silvestri, M. Seri, A. Facchetti, A. Taticchi, T. J. Marks, *Chem. Commun.* **2009**, 1380.
- 10 a) T. Kaneko, T. Makino, H. Miyaji, M. Teraguchi, T. Aoki, M. Miyasaka, H. Nishide, *J. Am. Chem. Soc.* **2003**, *125*, 3554. b) J. Yang, A. Dass, A.-M. M. Rawashdeh, C. Sotiriou-Leventis, M. J. Panzner, D. S. Tyson, J. D. Kinder, N. Leventis, *Chem. Mater.* **2004**, *16*, 3457. c) N. Vilà, Y.-W. Zhong, J. C. Henderson, H. D. Abruña, *Inorg. Chem.* **2010**, *49*, 796.
- 11 Supporting Information is available electronically on the CSJ-Journal Web site, <http://www.csj.jp/journals/chem-lett/index.html>.
- 12 SPhos: 2-Dicyclohexylphosphino-2',6'-dimethoxybiphenyl. T. E. Barder, S. D. Walker, J. R. Martinelli, S. L. Buchwald, *J. Am. Chem. Soc.* **2005**, *127*, 4685.
- 13 a) N. Miyaaura, K. Yamada, H. Sugimoto, A. Suzuki, *J. Am. Chem. Soc.* **1985**, *107*, 972. b) M. X.-W. Jiang, M. Rawat, W. D. Wulff, *J. Am. Chem. Soc.* **2004**, *126*, 5970. c) Y. Shi, X. Li, J. Liu, W. Jiang, L. Sun, *Tetrahedron Lett.* **2010**, *51*, 3626.
- 14 a) R. E. Maleczka, Jr., F. Shi, D. Holmes, M. R. Smith, III, *J. Am. Chem. Soc.* **2003**, *125*, 7792. b) I. Hisaki, S. Hiroto, K. S. Kim, S. B. Noh, D. Kim, H. Shinokubo, A. Osuka, *Angew. Chem., Int. Ed.* **2007**, *46*, 5125.
- 15 K. Kobayashi, Y. Yamada, M. Yamanaka, Y. Sei, K. Yamaguchi, *J. Am. Chem. Soc.* **2004**, *126*, 13896.
- 16 Crystallographic data reported in this manuscript have been deposited with Cambridge Crystallographic Data Centre as supplementary publication no. CCDC-824622. Copies of the data can be obtained free of charge via <http://www.ccdc.cam.ac.uk/conts/retrieving.html> (or from the Cambridge Crystallographic Data Centre, 12, Union Road, Cambridge, CB2 1EZ, U.K.; fax: +44 1223 336033; or deposit@ccdc.cam.ac.uk).