ELSEVIER

Contents lists available at SciVerse ScienceDirect

Tetrahedron

journal homepage: www.elsevier.com/locate/tet



$Ca(OH)_2$ -mediated efficient synthesis of 2-amino-5-hydroxy-4*H*-chromene derivatives with various substituents

Srinivasa Rao Kolla, Yong Rok Lee*

School of Chemical Engineering and Technology, Yeungnam University, Gyeongsan 712-749, Republic of Korea

ARTICLE INFO

Article history: Received 1 August 2011 Received in revised form 27 August 2011 Accepted 29 August 2011 Available online 3 September 2011

Keywords: Resorcinols 2-Benzylidenemalononitriles 2-Amino-5-hydroxy-4H-chromene Calcium hydroxide

ABSTRACT

A variety of novel 2-amino-5-hydroxy-4*H*-chromene derivatives with various substituents on the 4*H*-chromene ring were efficiently synthesized by one-pot reactions of substituted resorcinols and various 2-benzylidenemalononitriles in the presence of calcium hydroxide in methanol at room temperature. This simple method provided 2-amino-5-hydroxy-4*H*-chromenes with high yields under mild reaction conditions.

© 2011 Published by Elsevier Ltd.

1. Introduction

2-Amino-4*H*-chromenes are important heterocyclic compounds¹ with a number of biological and pharmacological properties including anti-microbial,² anti-viral,^{3,4} anti-proliferative,⁵ anti-tumor,⁶ and anti-cancer⁷ activities. They have also been widely employed as cosmetics, pigments,⁸ and potent biodegradable agrochemicals.⁹ In particular, derivatives, such as HA 14-1 and MX58151, with proapoptotic activity, are being developed as anticancer agents.^{10,11}

2-Amino-4*H*-chromenes are generally synthesized by reactions of phenol or resorcinol with malononitrile and arylaldehydes (Scheme 1). Basic alumina, ¹² piperidine, ¹³ morpholine, ¹⁴ cetyltrimethylammonium chloride, ¹⁵ [bmim]OH, ¹⁶ K₂CO₃/microwave, ¹⁷ DBU/microwave, ¹⁸ and chitosan ¹⁹ have been used for these reactions as catalytic and stoichiometric reagents. Interestingly, reactions of resorcinol with malononitrile and arylaldehydes afforded 2-amino-7-hydroxy-4*H*-chromene derivatives as sole products without requiring isolation of expected 2-amino-5-hydroxy-4*H*-chromene derivatives. In these cases, resorcinol reacted with malononitrile and arylaldehydes at the position 6 rather than at the position 2 under above described conditions and further cyclized to give 2-amino-7-hydroxy-4*H*-chromenes regioselectively, probably due to steric hindrance between two hydroxyl groups in a *meta*-disubstituted compound.

$$R = H, OH$$

Scheme 1.

Although several syntheses of 2-amino-4*H*-chromenes from phenol or resorcinol have been described, reactions employing substituted resorcinols have not previously been reported. This work reports reactions between resorcinols with various substituents and 2-benzylidenemalononitriles in the presence of several bases (Scheme 2). A variety of 2-amino-5-hydroxy-4*H*-chromene derivatives with various substituents on the 4*H*-chromene ring were easily and efficiently synthesized.

$$\begin{array}{c} \text{OH} \\ \text{R}_1 \\ \\ \text{R}_2 \end{array} + \text{R}_3 \xrightarrow{\text{U}} \begin{array}{c} \text{CN} \\ \text{CN} \end{array} \begin{array}{c} \text{bases} \\ \text{R}_1 \\ \\ \text{NH}_2 \end{array} \begin{array}{c} \text{NH}_2 \\ \text{N} \end{array}$$

Scheme 2.

2. Results and discussion

To give a variety of 2-amino-5-hydroxy-4*H*-chromene derivatives, reactions of readily available resorcinols with substituents of

^{*} Corresponding author. Tel.: $+82\ 53\ 810\ 2529$; fax: $+82\ 53\ 810\ 4631$; e-mail address: yrlee@yu.ac.kr (Y.R. Lee).

electron-withdrawing groups were investigated. The reaction of 2,4dihydroxy acetophenone (1a) with 2-benzylidenemalononitrile 2a under several bases was first attempted based on reported reactions of resorcinol with malononitrile and benzaldehyde in the presence of several bases (Table 1). 1 equiv of triethylamine, piperidine, morpholine, ethylenediamine, or DBU as organic bases for 12 h in methanol at room temperature did not result in the desired products (entries 1–5). Using KOH as an inorganic base resulted in no products (entry 6). Ca(OH)₂ in several solvents resulted in the desired product **3a**. Reaction using 20 mo% of Ca(OH)₂ in methanol at room temperature for 20 h produced 3a in 55% yield (entry 7). Interestingly, despite low solubility of Ca(OH)2 in alcohol, 1 equiv of Ca(OH)2 in methanol at room temperature for 3 h increased the yield of 3a to 80% (entry 8). It has been reported by other group that Ca(OH)₂ is an important metal reagent and a useful promoter in the aldol-type reaction of phenolic enolates of resorcinols with benzaldehyde in methanol.²⁰ Other polar solvents, EtOH and H₂O, produced **3a** in 61 and 51% yield, respectively (entries 9 and 10). Importantly, in these reactions, only 3a was formed and the expected 2-amino-7-hydroxy-4H-chromene as a regioisomer was not produced. The structure of compound **3a** was determined by ¹H NMR analysis. The spectrum of **3a** showed a characteristic peak of methine on the chromene ring at δ 4.65 as a singlet and a methyl peak of an acetyl group appeared at δ 2.58 as a singlet. The compound was further identified by AB coupling constants of aromatic protons on the 4H-chromene ring at δ 7.90 (J=8.7 Hz) and δ 6.72 (J=8.7 Hz). The ¹³C NMR spectrum showed a characteristic tertiary carbon peak with a methine proton at δ 36.1 and a carbonyl carbon peak at δ 204.4.

reaction quickly produced 2-amino-5-hydroxy-4*H*-chromene derivatives **3b**—**k** with various substituents on the benzene ring. Unsubstituted resorcinol **1f** produced **3l** in 90% yield (entry 11).

The formation of product **3a** can be explained by comparing with previously described mechanism as shown in Scheme 3.²¹ The regioselectivity of **3a** may be determined by the deactivation of the *o*-hydroxy group of **1a** by intramolecular hydrogen bonding to neighboring carbonyl group. The more reactive proton of the 4-hydroxy group of **1a** was extracted by calcium hydroxide to form an intermediate **4**, which reacted with 2-benzylidenemalononitrile (**2a**) to give **5**. Tautomerism of the intermediate **5** gave intermediate **6**, which was deprotonated by Ca(OH)₂ to produce **7**. Intramolecular nucleophilic addition of phenoxide to the cyano group of **7**, followed by protonation and isomerization produced **3a**.

In summary, a one-pot, two-component condensation reaction of various resorcinols and substituted 2-benzylidenemalononitriles was successfully accomplished to give 2-amino-5-hydroxy-4*H*-chromene derivatives. This methodology has advantages of mild reaction conditions, simple experimentation, and high yields.

3. Experimental

3.1. General

All experiments were carried out in the methanol. Pre-coated silica gel plates (Merck, Art. 5554) with a fluorescent indicator were used for analytical TLC. Flash column chromatography was performed using silica gel 9385 (Merck). ¹H and ¹³C NMR spectra

Table 1Reaction of 2,4-dihydroxy acetophenone with 2-benzylidenemalononitrile under several bases

Entry	Base	Solvent	Time (h)	Yield (%)	Entry	Base	Solvent	Time (h)	Yield (%)
1		МеОН	12	0	6	KOH (1 equiv)	МеОН	12	0
2	N H (1 eq)	МеОН	12	0	7	Ca(OH) ₂ (0.2 equiv)	МеОН	20	55
3	$\binom{O}{N}$ (1 eq)	МеОН	12	0	8	Ca(OH) ₂ (1 equiv)	МеОН	3	80
4	H ₂ N NH ₂ (1 eq)	МеОН	12	0	9	Ca(OH) ₂ (1 equiv)	EtOH	4	61
5	DBU (1 equiv)	MeOH	12	0	10	Ca(OH) ₂ (1 equiv)	H ₂ O	6	51

To explore the generality and scope of this methodology, additional reactions of various substituted resorcinols and 2-benzylidenemalononitriles were attempted under the optimized conditions (Table 2). Reactions of 2,4-dihydroxy acetophenone (1a) and 2-benzylidenemalononitriles **2b–e**, with electron-withdrawing groups on the benzene ring, at room temperature for 2–4 h produced **3b–e** in 81–87% yield (entries 1–4). Electron-donating group on the 2-benzylidenemalononitriles resulted in **3f** and **3g** in 74 and 73% yield, respectively, after reaction for 10–12 h at room temperature (entries 5 and 6). Other resorcinols, **1b** and **1c** produced **3h** and **3i** in 82 and 79%, respectively (entries 7 and 8). Resorcinols with ester groups on the benzene ring produced the expected cycloadducts **3j** and **3k** in 84 and 81% yield, respectively (entries 9 and 10). Each

were recorded on a Varian Model VNS spectrometer in DMSO- d_6 and CD₃OD. IR spectra were recorded on a Jasco FTIR 5300 spectrophotometer. HRMS spectra were recorded at the Korea Basic Science Institute on a Jeol JMS-700 spectrometer.

3.2. General procedure for the preparation of 2-amino-3-cyano-6-acetoxy-5-hydroxy-4*H*-chromene derivatives 3a-l

A mixture of various resorcinols (1.0 mmol), substituted 2-benzylidenemalononitrile (1.5 mmol), and calcium hydroxide (1.0 mmol) in MeOH (5 mL) was stirred at room temperature until the completion of the reaction, as indicated by TLC (ethyl acetate/*n*-hexane). The reaction mixture was filtered through sintered funnel

Table 2Additional reactions for the synthesis of 2-amino-5-hydroxy-4*H*-chromene derivatives in the presence of Ca(OH)₂

Entry	resorcinol	malononitrile	Time (h)	Product	Yield (%)
1		O_2N CN CN 2b	2	O OH 3b Br	82
2		Br CN CN 2c	4	CN O OH 3c CN	87
3	OOH	CI CN 2d	2	O OH 3d F	81
4	1a OH	F CN CN 2e	3	ONH ₂ OOH 3e	85
5		CN 2f	10	O OH 3f OCH ₃	74
6		$_{\mathrm{H_{3}CO}}$ CN $_{\mathrm{CN}}$ 2g	12	O OH 3g CN	73
7	O OH 1b OH O OH	CN CN 2a	5	O OH CN 3h	82
8	O OH 1c OH O OH	CN CN 2a	5	O NH ₂ O OH 3i CN NH ₂	79
9	1d OH	CN CN 2a	5	O OH CN 3j	84
10	1e OH	CN CN 2a	3	O NH ₂ O OH 3k CN NH ₂	81
11	oH OH	CN CN 2a	5	HO NH ₂	90

Scheme 3.

and the solid was washed with ethyl acetate (10 mL) to remove excess 2-benzylidenemalononitrile. The solid containing product was dissolved in THF (30 mL) and filtered to remove remaining Ca(OH)₂. The THF layer was evaporated under reduced pressure and finally recrystallization with ethanol give pure white solid.

3.2.1. 6-Acetyl-2-amino-5-hydroxy-4-phenyl-4H-chromene-3-carbonitrile (**3a**). A mixture of 2,4-dihydroxy acetophenone **1a** (152 mg, 1.0 mmol), 2-benzylidenemalononitrile **2a** (231 mg, 1.5 mmol), and Ca(OH)₂ (74 mg, 1.0 mmol) in 5 mL of methanol was stirred at room temperature for 3 h. After completion of the reaction, recrystallization with ethanol gave product **3a** (245 mg, 80%) as a white solid: charred at 290 °C; ¹H NMR (300 MHz, DMSO- d_6) δ 12.95 (1H, s), 7.90 (1H, d, J=8.7 Hz), 7.30–7.25 (2H, m), 7.20–7.11 (3H, m), 7.08 (2H, s,), 6.72 (1H, d, J=8.7 Hz), 4.65 (1H, s), 2.58 (3H, s); ¹³C NMR (75 MHz, DMSO- d_6) δ 204.4, 160.1, 159.4, 153.8, 144.7, 131.8, 128.4, 127.1, 126.6, 120.0, 115.9, 111.7, 107.3, 57.2, 36.1, 26.6; IR (KBr)

3438, 3317, 3200, 2881, 2195, 1660, 1414, 1251, 1076, 824 cm $^{-1}$; HRMS m/z (M $^+$) calcd for $C_{18}H_{14}N_2O_3$: 306.1004. Found: 306.1008.

3.2.2. 6-Acetyl-2-amino-5-hydroxy-4-(4-nitrophenyl)-4H-chromene-3-carbonitrile (**3b**). A mixture of 2,4-dihydroxy acetophenone **1a** (152 mg, 1.0 mmol), 2-(4-nitrobenzylidene) malononitrile **2b** (300 mg, 1.5 mmol), and Ca(OH)₂ (74 mg, 1.0 mmol) in 5 mL of methanol was stirred at room temperature for 2 h. After completion of the reaction, recrystallization with ethanol gave product **3b** (288 mg, 82%) as a solid: mp 246–248 °C; ¹H NMR (300 MHz, DMSO- d_6) δ 12.96 (1H, s), 8.16 (2H, d, J=8.7 Hz), 7.94 (1H, d, J=9.0 Hz), 7.41 (2H, d, J=8.7 Hz), 7.21 (2H, s), 6.75 (1H, d, J=9.0 Hz), 4.86 (1H, s), 2.58 (3H, s); ¹³C NMR (75 MHz, DMSO- d_6) δ 204.6, 160.1, 159.5, 153.7, 152.1, 146.3, 132.4, 128.6, 123.8, 119.7, 116.0, 110.3, 107.5, 55.9, 36.1, 26.6; IR (KBr) 3439, 3337, 3209, 2935, 2193, 1654, 1520, 1409, 1348, 1253, 820 cm⁻¹; HRMS m/z (M⁺) calcd for C₁₈H₁₃N₃O₅: 351.0855. Found: 351.0856.

3.2.3. 6-Acetyl-2-amino-4-(4-bromophenyl)-5-hydroxy-4H-chromene-3-carbonitrile (**3c**). A mixture of 2,4-dihydroxy acetophenone **1a** (152 mg, 1.0 mmol), 2-(4-bromobenzylidene) malononitrile **2c** (350 mg, 1.5 mmol), and Ca(OH)₂ (74 mg, 1.0 mmol) in 5 mL of methanol was stirred at room temperature for 4 h. After completion of the reaction, recrystallization with ethanol gave product **3c** (335 mg, 87%) as a solid: mp 283–285 °C; 1 H NMR (300 MHz, DMSO- 1 6) 3 6 12.94 (1H, s), 7.87 (1H, d, 1 8.7 Hz), 7.45 (2H, d, 1 8.1 Hz), 7.08 (2H, s), 7.07 (2H, d, 1 8.1 Hz), 6.66 (1H, d, 1 8.7 Hz), 4.64 (1H, s), 2.55 (3H, s); 1 8 NMR (150 MHz, DMSO- 1 6) 3 6 204.6, 160.5, 159.7, 153.9, 144.4, 132.3, 131.5, 129.6, 120.0, 119.9, 116.3, 111.4, 107.5, 56.9, 35.9, 26.9; IR (KBr) 3442, 3327, 3205, 2967, 2197, 1661, 1485, 1410, 1253, 1075, 820 cm⁻¹; HRMS 1 8 (M⁺) calcd for 1 9 calcd for 1 9 calcd for C₁₈H₁₃BrN₂O₃: 384.0110. Found: 384.0113.

3.2.4. 6-Acetyl-2-amino-4-(4-chlorophenyl)-5-hydroxy-4H-chromene-3-carbonitrile (**3d**). A mixture of 2,4-dihydroxy acetophenone **1a** (152 mg, 1.0 mmol), 2-(4-chlorobenzylidene) malononitrile **2d** (282 mg, 1.5 mmol), and Ca(OH)₂ (74 mg, 1.0 mmol) in 5 mL of methanol was stirred at room temperature for 2 h. After completion of the reaction, recrystallization with ethanol gave product **3d** (276 mg, 81%) as a solid: mp 288–290 °C; ¹H NMR (300 MHz, DMSO- d_6) δ 12.95 (1H, s), 7.89 (H, d, J=8.7 Hz), 7.32 (2H, d, J=8.1 Hz), 7.13 (2H, d, J=8.1 Hz), 7.12 (2H, s), 6.69 (1H, d, J=8.7 Hz), 4.66 (1H, s), 2.56 (3H, s); ¹³C NMR (75 MHz, DMSO- d_6) δ 204.4, 160.1, 159.4, 153.7, 143.7, 132.0, 131.2, 129.0, 128.4, 119.8, 115.9, 111.1, 107.4, 56.7, 35.6, 26.5; IR (KBr) 3440, 3328, 3210, 2928, 2200, 1662, 1487, 1412, 1251, 1080, 821 cm⁻¹; HRMS m/z (M⁺) calcd for $C_{18}H_{13}$ ClN₂O₃: 340.0615. Found: 340.0616.

3.2.5. 6-Acetyl-2-amino-4-(4-fluorophenyl)-5-hydroxy-4H-chromene-3-carbonitrile (**3e**). A mixture of 2,4-dihydroxy acetophenone **1a** (152 mg, 1.0 mmol), 2-(4-fluorobenzylidene) malononitrile **2e** (260 mg, 1.5 mmol), and Ca(OH)₂ (74 mg, 1.0 mmol) in 5 mL of methanol was stirred at room temperature for 3 h. After completion of the reaction, recrystallization with ethanol gave product **3e** (275 mg, 85%) as a solid: mp 275–277 °C; 1 H NMR (300 MHz, DMSO- 4 G) 5 12.97 (1H, s), 7.91 (1H, d, 4 J=8.7 Hz), 7.19–7.07 (6H, m), 6.71 (1H, d, 4 J=8.7 Hz), 4.68 (1H, s), 2.58 (3H, s); 13 C NMR (75 MHz, DMSO- 4 G) 5 204.4, 162.5, 160.2, 159.4, 153.7, 141.0, 131.9, 129.0 (d, 4 J_{CF}=8.3 Hz), 119.9, 116.0, 115.1 (d, 4 J_{CF}=21.4 Hz), 111.5, 107.3, 57.1, 35.4, 26.6; IR (KBr) 3431, 3331, 3214, 2878, 2198, 1663, 1505, 1414, 1247, 1078, 818 cm⁻¹; HRMS 4 J(M⁺) calcd for C₁₈H₁₃FN₂O₃: 324.0910. Found: 324.0908.

3.2.6. 6-Acetyl-2-amino-5-hydroxy-4-p-tolyl-4H-chromene-3-carbonitrile (**3f**). A mixture of 2,4-dihydroxy acetophenone **1a** (152 mg, 1.0 mmol), 2-(4-methylbenzylidene) malononitrile **2f** (252 mg, 1.5 mmol), and Ca(OH)₂ (74 mg, 1.0 mmol) in 5 mL of methanol was stirred at room temperature for 10 h. After completion of the reaction, recrystallization with ethanol gave product **3f**

(237 mg, 74%) as a solid: mp 280–282 °C; 1 H NMR (300 MHz, DMSO- d_{6}) δ 12.89 (1H, s), 7.80 (1H, d, J=8.7 Hz), 7.02–6.94 (6H, m), 6.63 (1H, d, J=8.7 Hz), 4.54 (1H, s), 2.50 (3H, s), 2.16 (3H, s); 13 C NMR (75 MHz, DMSO- d_{6}) δ 204.4, 160.1, 159.4, 153.8, 141.8, 135.8, 131.7, 128.9, 127.0, 120.0, 115.9, 111.9, 107.3, 57.4, 35.7, 26.5, 20.5; IR (KBr) 3627, 3427, 3330, 3213, 2955, 2194, 1653, 1415, 1251, 1129, 1069, 814 cm $^{-1}$; HRMS m/z (M $^{+}$) calcd for $C_{19}H_{16}N_{2}O_{3}$: 320.1161. Found: 320.1164.

3.2.7. 6-Acetyl-2-amino-5-hydroxy-4-(4-methoxyphenyl)-4H-chromene-3-carbonitrile (**3g**). A mixture of 2,4-dihydroxy acetophenone **1a** (152 mg, 1.0 mmol), 2-(4-methoxybenzylidene) malononitrile **2g** (276 mg, 1.5 mmol), and Ca(OH)₂ (74 mg, 1.0 mmol) in 5 mL of methanol was stirred at room temperature for 12 h. After completion of the reaction, recrystallization with ethanol gave product **3g** (245 mg, 73%) as a solid: mp 248–250 °C; ¹H NMR (300 MHz, DMSO- d_6) δ 12.96 (1H, s), 7.88 (1H, d, J=8.7 Hz), 7.06–7.03 (4H, m), 6.83 (2H, d, J=8.4 Hz), 6.70 (1H, d, J=8.7 Hz), 4.60 (1H, s), 3.70 (3H, s), 2.57 (3H, s); ¹³C NMR (75 MHz, DMSO- d_6) δ 204.4, 160.1, 159.4, 158.0, 153.7, 136.8, 131.7, 128.2, 120.1, 115.9, 113.7, 112.1, 107.3, 57.5, 55.0, 30.4, 26.5; IR (KBr) 3433, 3324, 3203, 2950, 2194, 1655, 1509, 1410, 1252, 1068, 823 cm⁻¹; HRMS m/z (M⁺) calcd for $C_{19}H_{16}N_2O_4$: 336.1110. Found: 336.1111.

3.2.8. 2-Amino-5-hydroxy-4-phenyl-6-propionyl-4H-chromene-3-carbonitrile (**3h**). A mixture of 1-(2,4-dihydroxyphenyl)propan-1-one **1b** (166 mg, 1.0 mmol), 2-benzylidenemalononitrile **2a** (231 mg, 1.5 mmol), and Ca(OH)₂ (74 mg, 1.0 mmol) in 5 mL of methanol was stirred at room temperature for 5 h. After completion of the reaction, recrystallization with ethanol gave product **3h** (263 mg, 82%) as a solid: mp 268–270 °C; 1 H NMR (300 MHz, DMSO- 4 G) 5 12.99 (1H, s), 7.92 (1H, d, 4 J=9.0 Hz), 7.30–7.25 (2H, m), 7.20–7.11 (3H, m), 7.07 (2H, s), 6.71 (1H, d, 4 J=9.0 Hz), 4.65 (1H, s), 3.05 (2H, q, 4 J=7.2 Hz), 1.05 (3H, t, 4 J=7.2 Hz); 4 C NMR (75 MHz, DMSO- 4 G) 5 206.7, 160.0, 159.5, 153.7, 144.7, 130.8, 128.4, 127.1, 126.6, 120.0, 115.4, 111.8, 107.3, 57.2, 36.1, 30.9, 7.9; IR (KBr) 3430, 3331, 3217, 2980, 2194, 1651, 1409, 1260, 1081, 835 cm $^{-1}$; HRMS 4 M/z (M $^+$) calcd for C 4 9H $_1$ 6N $_2$ O3: 320.1161. Found: 320.1161.

3.2.9. 2-Amino-6-benzoyl-5-hydroxy-4-phenyl-4H-chromene-3-carbonitrile (3i). A mixture of 2,4-dihydroxybenzophenone 1c (214 mg, 1.0 mmol), 2-benzylidenemalononitrile 2a (231 mg, 1.5 mmol), and Ca(OH)₂ (74 mg, 1.0 mmol) in 5 mL of methanol was stirred at room temperature for 5 h. After completion of the reaction, recrystallization with ethanol gave product 3i (291 mg, 79%) as a solid: mp 280–282 °C; 1 H NMR (300 MHz, DMSO- 4 6) δ 12.40 (1H, s), 7.65–7.62 (3H, m), 7.56–7.54 (2H, m), 7.50 (1H, d, 4 8.4 Hz), 7.31–7.28 (2H, m), 7.23–7.17 (3H, m), 7.11 (2H, s), 6.73 (1H, d, 4 8.4 Hz), 4.73 (1H, s); 4 8 NMR (75 MHz, DMSO- 4 6) δ 199.6, 178.5, 160.4, 159.5, 153.8, 144.7, 137.1, 133.4, 132.1, 128.8, 128.4, 127.2, 126.7, 120.0, 115.9, 112.4, 107.6, 57.2, 36.2; IR (KBr) 3431, 3323, 2187, 1650, 1599, 1404, 1342, 1254, 1081, 832, 705 cm⁻¹; HRMS 4 8 (M⁺) calcd for C₂₃H₁₆N₂O₃: 368.1161. Found: 368.1157.

3.2.10. Methyl 2-amino-3-cyano-5-hydroxy-4-phenyl-4H-chromene-6-carboxylate (3j). A mixture of methyl 2,4-dihydroxybenzoate 1d (168 mg, 1.0 mmol), 2-benzylidenemalononitrile 2a (231 mg, 1.5 mmol), and Ca(OH)₂ (74 mg, 1.0 mmol) in 5 mL of methanol was stirred at room temperature for 5 h. After completion of the reaction, recrystallization with ethanol gave product 3j (271 mg, 84%) as a solid: mp 275–277 °C; 1 H NMR (300 MHz, DMSO- d_6) δ 10.99 (1H, s), 7.73 (1H, d, J=9.0 Hz), 7.28–7.23 (2H, m), 7.19–7.08 (3H, m), 7.04 (2H, s), 6.69 (1H, d, J=9.0 Hz), 4.64 (1H, s), 3.82 (3H, s); 13 C NMR (75 MHz. DMSO- d_6) δ 169.2, 159.6, 158.7, 153.6, 144.8, 129.7, 128.4, 127.1, 126.6, 120.1, 111.9, 108.4, 107.7, 57.2, 52.6, 36.3; IR (KBr) 3428, 3332, 3216, 3071, 2957, 2875, 2197, 1664, 1408, 1338, 1262, 1075, 782 cm $^{-1}$; HRMS m/z (M^+) calcd for $C_{18}H_{14}N_{2}O_4$: 322.0954. Found: 322.0958.

3.2.11. Ethyl 2-amino-3-cyano-5-hydroxy-7-methyl-4-phenyl-4H-chromene-6-carboxylate (**3k**). A mixture of ethyl 2,4-dihydroxy-6-

methylbenzoate **1e** (196 mg, 1.0 mmol), 2-benzylidenemalononitrile **2a** (231 mg, 1.5 mmol), and Ca(OH)₂ (74 mg, 1.0 mmol) in 5 mL of methanol was stirred at room temperature for 3 h. After completion of the reaction, recrystallization with ethanol gave product **3k** (284 mg, 81%) as a solid: mp 268–270 °C; 1 H NMR (300 MHz, DMSO- 4 6) δ 11.09 (1H, s), 7.29–7.24 (2H, m), 7.20–7.09 (3H, m), 6.98 (2H, s), 6.55 (1H, s), 4.62 (1H, s), 4.28 (2H, q, 1 5–7.2 Hz), 2.40 (3H, s), 1.28 (3H, t, 1 5–7.2 Hz); 13 C NMR (150 MHz, DMSO- 4 6) δ 169.7, 159.9, 157.9, 151.8, 145.3, 140.3, 128.6, 127.3, 126.7, 120.4, 111.7, 110.2, 110.1, 61.7, 57.6, 36.6, 22.4, 14.1; IR (KBr) 3427, 3325, 3213, 2978, 2198, 1661, 1404, 1273, 1172, 1022, 818 cm $^{-1}$; HRMS 1 7 (M $^{+}$ 1) calcd for 2 80 calcd for 2 90 calcd for 2 91 calcd for 2 90 calcd for 2 90 calcd for 2 90 calcd for 2 91 calcd for 2 92 calcd for 2 91 calcd for 2 91 calcd for 2 93 calcd for 2 93 calcd for 2 93 calcd for 2 94 calcd for 2 94 calcd for 2 94 calcd for 2 96 cal

3.2.12. 2-Amino-7-hydroxy-4-phenyl-4H-chromene-3-carbonitrile (31). A mixture of resorcinol 1f (110 mg, 1.0 mmol), 2-benzylidene-malononitrile 2a (231 mg, 1.5 mmol), and Ca(OH)₂ (74 mg, 1.0 mmol) in 5 mL of methanol was stirred at room temperature for 5 h. After completion of the reaction, recrystallization with ethanol gave product 31 (238 mg, 90%) as a solid: mp 233–235 °C; ¹H NMR (300 MHz, DMSO- d_6) δ 9.68 (1H, s), 7.32–7.27 (2H, m), 7.22–7.15 (3H, m), 6.85 (2H, s), 6.80 (1H, d, J=8.4 Hz), 6.47 (1H, d, J=8.4 Hz), 6.41 (s, 1H), 4.61 (1H, s); ¹³C NMR (75 MHz, CD₃OD) δ 162.4, 158.6, 150.8, 147.5, 131.3, 129.7, 128.9, 128.0, 122.1, 115.5, 113.6, 103.7, 58.6, 42.1; IR (KBr) 3494, 3426, 3336, 2189, 1645, 1506, 1406, 1155, 1050, 853 cm⁻¹; HRMS m/z (M⁺) calcd for C₁₆H₁₂N₂O₂: 264.0899. Found: 264.0898.

Acknowledgements

This work was supported by grant No. RTI04-01-04 from the Regional Technology Innovation Program of the Ministry of Knowledge Economy (MKE).

References and notes

- See for instance: (a) Makarem, S.; Mohammadi, A. A.; Fakhari, A. R. Tetrahedron Lett. 2008, 49, 7194; (b) Elinson, M. N.; Dorofeev, A. S.; Miloserdov, F. M.; Ilovaisky, A. I.; Feducovich, S. K.; Belyakov, P. A.; Nikishin, G. I. Adv. Synth. Catal. 2008, 350, 591 and references cited therein.
- Khafagy, M. M.; Abd El-Wahab, A. H. F.; Eid, F. A.; El-Agody, A. M. Farmaco 2002, 57, 715

- Smith, P. W.; Sollis, S. L.; Howes, P. D.; Cherry, P. C.; Starkey, I. D.; Cobley, K. N.; Weston, H.; Scicinski, J.; Merritt, A.; Whittington, A.; Wyatt, P.; Taylor, N.; Green, D.; Bethell, R.; Madar, S.; Fenton, R. J.; Morley, P. J.; Pateman, T.; Beresford, A. J. Med. Chem. 1998, 41, 787.
- 4. Martinez-Grau, A.; Marco, J. L. Bioorg. Med. Chem. Lett. 1997, 7, 3165.
- Dell, C. P.; Smith, C. W. European Patent Appl. EP 537949; Chem. Abstr. 1993, 119, 139102d.
- Mohr, S. J.; Chirigos, M. A.; Fuhrman, F. S.; Pryor, J. W. Cancer Res. 1975, 35, 3750.
 (a) Anderson, D. R.; Hegde, S.; Reinhard, E.; Gomez, L.; Vernier, W. F.; Lee, L.; Liu, S.; Sambandam, A.; Snider, P. A.; Masih, L. Bioorg. Med. Chem. Lett. 2005, 15, 1587; (b) Skommer, J.; Wlodkowic, D.; Matto, M.; Eray, M.; Pelkonen, J. Leukemia Res. 2006, 30, 322 and references cited therein; (c) Wang, J.-L.; Liu, D.; Zhang, Z.-J.; Shan, S.; Han, X.; Srinvasula, S. M.; Croce, C. M.; Alnemri, E. S.; Huang, Z. Proc. Natl. Acad. Sci. U.S.A. 2000, 97, 7124.
- Ellis, G. P. In *The Chemistry of Heterocyclic Compounds. Chromenes, Harmones, and Chromones*; Weissberger, A., Taylor, E. C., Eds.; John: New York, NY, 1977; Chapter II. pp 11–139.
- 9. Hafez, E. A.; Elnagdi, M. H.; Elagamey, A. A.; El-Taweel, F. A. M. Heterocycles 1987, 26, 903.
- 10. Doshi, J. M.; Tian, D.; Xing, C. J. Med. Chem. 2006, 49, 7731.
- (a) Kemnitzer, W.; Drewe, J.; Jiang, S.; Zhang, H.; Wang, Y.; Zhao, J.; Jia, S.; Herich, J.; Labreque, D.; Storer, R.; Meerovitch, K.; Bouffard, D.; Rej, R.; Denis, R.; Blais, C.; Lamothe, S.; Attardo, G.; Gourdeau, H.; Tseng, B.; Kasibhatla, S.; Cai, S. X. J. Med. Chem. 2004, 47, 6299; (b) Kemnitzer, W.; Drewe, J.; Jiang, S.; Zhang, H.; Wang, Y.; Zhao, J.; Crogan-Grundy, C.; Xu, L.; Lamothe, S.; Gourdeau, H.; Denis, R.; Tseng, B.; Kasibhatla, S.; Cai, S. X. J. Med. Chem. 2007, 50, 2858; (c) Kemnitzer, W.; Drewe, J.; Jiang, S.; Zhang, H.; Crogan-Grundy, C.; Labreque, D.; Bubenick, M.; Attardo, G.; Denis, R.; Lamothe, S.; Gourdeau, H.; Tseng, B.; Kasibhatla, S.; Cai, S. X. J. Med. Chem. 2008, 51, 417.
- Maggi, R.; Ballini, R.; Sartori, G.; Sartorio, R. Tetrahedron Lett. 2004, 45, 2297.
- Kemnitzer, W.; Kasibhatla, S.; Jiang, S.; Zhang, H.; Zhao, J.; Jia, S.; Xu, L.; Crogan-Grundy, C.; Denis, R.; Barriault, N.; Vaillancourt, L.; Charron, S.; Dodd, J.; Attardo, G.; Labrecque, D.; Lamothe, S.; Gourdeau, H.; Tseng, B.; Drewe, J.; Cai, S. X. Bioorg. Med. Chem. Lett. 2005, 15, 4745.
- Klokol, G. V.; Sharanina, L. G.; Nesterov, V. N.; Shklover, V. E.; Sharanin, Y. A.; Struchkov, Y. T. Zh. Org. Khim. 1987, 23, 412.
- Ballini, R.; Bosica, G.; Conforti, M. L.; Maggi, R.; Mazzacani, A.; Righi, P.; Sartori, G. Tetrahedron 2001, 57, 1395.
- 16. Gong, K.; Wang, H.-L.; Luo, J.; Liu, Z.-L. J. Heterocycl. Chem. 2009, 46, 1145.
- Kidwai, M.; Saxena, S.; Khalilur Rahman khan, M.; Thukral, S. S. Bioorg. Med. Chem. Lett. 2005, 15, 4295.
- 18. Raghuvanshi, D. S.; Singh, K. N. ARKIVOC 2010, x, 305.
- Al-Matar, H. M.; Khalil, K. D.; Meier, H.; Kolshorn, H.; Elnagdi, M. H. ARKIVOC 2008, xvi, 288.
- Saimoto, H.; Yoshida, K.; Murakami, T.; Morimoto, M.; Sashiwa, H.; Shigemasa, Y. J. Org. Chem. 1996, 61, 6768.
- Abd-El-Aziz, A. A.; El-Agrody, A. M.; Bedair, A. H.; Corkery, T. C.; Ata, A. Heterocycles 2004, 63, 1793.