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Practical Syntheses of Functionalized 1-Oxo-L,2,3,4-Tetrahydro-β-carboline-3carboxylic Acid Esters

Laurent Jeannin ^a , Janos Sapi ^a , Elka Vassileva ^a , Pierre Renard ^b & Jean-Yves Laronze ^a

^a Laboratoire de Chimie Thérapeutique, associé au CNRS: "Isolement, Structure, Transformations et Synthèse de Substances Naturelles", Faculté dc Pharmacie Université de Reims Champagne-Ardenne, F-51096, Reims, France

^b ADIR et Cie, 1, rue Carle Hébert, F-92415, Courbevoie, France Published online: 21 Aug 2006.

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PRACTICAL SYNTHESES OF FUNCTIONALIZED 1-OXO-1,2,3,4-TETRAHYDRO-β-CARBOLINE-3-CARBOXYLIC ACID ESTERS

Laurent Jeannin^a, Janos Sapi^a, Elka Vassileva^a, Pierre Renard^b, and Jean-Yves Laronze^{a*}

^aLaboratoire de Chimie Thérapeutique, associé au CNRS : "Isolement, Structure, Transformations et Synthèse de Substances Naturelles" Faculté de Pharmacie, Université de Reims Champagne-Ardenne F-51096 Reims, France

^b ADIR et Cie, 1, rue Carle Hébert F-92415 Courbevoie, France

Abstract: Three methods have been proposed for the preparation of functionalized 1-oxo-1,2,3,4-tetrahydro- β -carboline-3-carboxylates (3) from which the "acylazide formation-Curtius rearrangement-acid catalyzed ring closure" sequence starting from hemi-ester hemi-acids (8) seemed to be quite general.

1,2,3,4-Tetrahydro- β -carboline-3-carboxylic acid derivatives can be considered as constrained analogues of the aromatic amino-acid tryptophan. With regard to these properties we were interested in the preparation of some nonpeptide antagonists of the vaso-active peptide angiotensin II (AG II). Among these compounds we found 1 to be a specific AT₂ receptor antagonist to AG II (IC₅₀=0.9 μ Mol), possessing a therapeutic profile¹ related to that of Parke-Davis' tetrahydro-1*H*-imidazo[4,5-*c*]pyridine derivative 2.²

^{*}To whom correspondence should be addressed.



In the course of research aimed at studying SAR we decided to introduce a carbonyl function at C-1 of the tetrahydro- β -carbonyl core maintaining the possibility for other pharmacomodulations at N-2, C-4, C-6 and N-9 positions. Here we disclose the synthetic aspects of this programme, developed for the preparation of the unknown 1-0x0-1,2,3,4-tetrahydro- β -carboline-3-carboxylic acid derivatives 3, possible precursors of conformationally more restricted analogues of 1. From synthetic point of view two of the three proposed pathways were inspired by previous results of the literature using tryptamines and an activated form of carbon dioxide.



path A: i: triphosgene, Et₃N, CH₃CN; ii: (1) 30% HBr-AcOH, reflux; or (2) HCl(bubbling)-CH₃CN, reflux. path B: iii: Ph₃PCb, Et₃N, CH₂Cb; iv: (1) CO₂, CH₃CN, 90°C, 3 bar, 3 days; or (2) CO(OMe)₂, CH₃CN, sealed tube, 110°C, 3 days.

path C: v: DPPA, Proton Sponge[®], CH₃CN; ii: (1) 30% HBr-AcOH, reflux; or (2) HCl(bubbling)-CH₃CN, reflux; or (3) 30% HBr-AcOH, toluene, reflux.

Entry	Starting material ^a	R ¹	R ²	R ³	R ⁴	Method	Product b (ratio)	Yield ^C (%)
1	(S)-4a	Н	Me	Н	н	A(1)	(S)-3a	81
2	4b	Bn	Me	н	Н	A(1)	3b	48
3	4c	CHEt ₂	Me	н	Н	A(2)	3c	54
4	4d	(CH ₂) ₂ CO ₂ Me	Me	н	Н	A(2)	3d	17
5	4e	Н	Me	н	OMe	A(1)	3e	65
6	4f(erythro)	Н	Et	Ph	н	A(2)	3f (c)	61
7	4f(threo)	Н	Et	Ph	Н	A(2)	3f (<i>tr</i>)	58
8	(S)- 4a	н	Me	н	Н	B(1)	(S)-3a	45
9	(S)-4a	н	Me	н	Н	B(2)	(S)- 3a	56
10	8f	н	Et	Ph	Н	C(2)	3f (c:tr, 1:1)	45
11	8g	Н	Et	Bn	Н	C(1)	3g (c:tr, 1:1)	40
12	8h	н	Ēt	CHEt ₂	Н	C(3)	3h (c:tr, 1:2)	62
13	8i	н	Et	C6H11	Н	C(3)	3i (c:tr, 1:2)	51
14	8j	Н	Εt	Me	Br	C(3)	3j (c:tr, 1:1)	68
15	8k	Н	Et	CH2OCbz	Н	<u>C(2)</u>	3k (c:tr, 1:1)	26

Table 1. Preparation of some 1-oxo-1,2,3,4-tetrahydro- β -carboline-3-carboxylic acid ester derivatives (3)

^a Optically active (entries 1,8,9), racemates (entries 2-5), diastereomeric [1:1, (for entries 10,11, 14,15); 1:2 (for entries 12,13)] racemates. ^b relative configuration: $CO_2R^2/R^3 c$: *cis*; *tr* : *trans.* ^c isolated yield (calculated from 4 or 8) for fully characterized products.

Thus, path A employed a procedure reported for the preparation of some 1-oxo-tetrahydro- β -carboline alkaloids.³ Treatment of L-tryptophan methyl ester 4a with triphosgene in the presence of triethylamine led to isocyanate 5 (non isolated) which was then cyclized into 3a in strongly acidic medium (30 % HBr in acetic acid, or CH₃CN saturated with HCl). Formation of symmetrical urea, resulting from attack of intermediates 5 and (or) 6 by a second amine molecule was avoided by the use of a slight excess of triethylamine. Inspite of these drastic conditions chirality at C-3 survived, evidenced by chiral lanthanide shift measurements. This otherwise costly reaction offered a convenient method for the transformation of amines, including secondary ones (entries 2-5) into target molecules except for 4d where a retro-Michael cleavage of the N-propionic ester chain could be accounted for the low yield.

The second method (path B), involved a thermal ring closure of isocyanates obtained from iminophosphoranes and carbonyl compounds. This sequence had been developed by Molina and co-workers for the synthesis of the fully aromatic isoquinoline, γ - and β -carboline derivatives.⁴ Iminophosphorane 7a was prepared from L-tryptophan methyl ester 4a with Ph₃PCl₂ in the presence of triethylamine. Exposure of 7a to carbon dioxide or dimethyl carbonate in a sealed tube at 90-110°C led to 3a in moderate yield. Low yield formation of iminophosphorane from β -substituted tryptophans like 4f led us to give up this method.

The third approach (path C) which appeared quite general was based on a tandem "acylazide formation-Curtius rearrangement-acid catalyzed ring closure" sequence, starting from the appropriate acid.⁵ Hemi-acid hemi-esters **8**, precursors of the corresponding β -substituted tryptophan esters **4**,⁶ were directly transformed (*via* isocyanates **5**) into 1-oxo-tetrahydro- β -carbolines **3** by using diphenylphosphorylazide (DPPA) and Proton Sponge[®] according to Yamada's protocol,⁷ followed by strong acid treatment. Overall yields (from **8**) ranged between 40 and 68 %, except for **3k**, which was sensitive to successive basic and acid conditions. It is interesting to note that, in all cases, no epimerization was observed and diastereomeric ratios (entries 12, 13) were conserved under the Proton Sponge[®]-mediated isocyanate formation. Furthermore, functionalization of the lactam nitrogen following this pathway required an additional alkylation step.

In conclusion, tandem "acylazide formation-Curtius rearrangement-acid catalyzed ring closure" sequence proved to be efficient even for multigram synthesis of polyfunctionalized oxo-tetrahydro- β -carboline derivatives, whose detailed biological investigations will be disclosed elsewhere.

EXPERIMENTAL

Melting points were determined on a Reichert melting point apparatus and are uncorrected. IR spectra (v, cm⁻¹) were recorded on a BOMEM FTIR apparatus and a COSMIC interferometer; UV spectra were recorded on a Varian 634 spectrophotometer; ¹H- and ¹³C-NMR spectra were measured on a Bruker AC 300 apparatus at 300 MHz and 75 MHz, respectively. Mass spectra (E = 70 eV) were obtained on a JEOL JMS D-300 spectrometer; Kieselgel 60 PGF₂₅₄ (Merck

 N° 7749) was used for thin layer chromatography and Kieselgel 60 (Merck N° 9385) for flash chromatography.

Iminophosphorane of methyl L-tryptophanate (7a)

To an ice-cold CH₂Cl₂ (50 ml) solution of dichlorotriphenylphosphorane (12.5 mmol) was added dropwise within 0.5-1 h a cold CH₂Cl₂ (50 ml) solution of methyl L-tryptophanate (4a) (9 mmol) and triethylamine (12.5 mmol). At the end of the addition the precipitate was filtered off, the solvent was evaporated and the residue was purified by flash chromatography (elution: CH₂Cl₂ : 3-8 % MeOH) to give 7a in 92 % yield. 7a: mp 198-200°C. $[\alpha]_{D}$ = -81° (*c* 1, MeOH). UV (MeOH) 215, 265, 270, 281 nm. IR (KBr) v 3140, 1740, 1445 cm⁻¹. ¹H NMR (CDCl₃) δ 10.85 (1H, s), 7.44-7.67 (17H, m), 7.08 (2H, m), 6.81 (1H, t, J=7 Hz), 3.61-3.72 (2H, m), 3.58 (1H, s), 3.21 (1H, dd, J=13.5, 3 Hz). ¹³C NMR (CDCl₃) δ 172.3, 136.6, 134.3, 133.6, 133.5, 129.4, 129.2, 126.4, 126.3, 121.5, 120.9, 120.1, 118.5, 117.4, 112.4, 109.1, 56.6, 52.1, 26.5. MS *m*/*z* (%) 478 (M⁺, 0.2), 419 (4), 348 (52), 277 (100).

General procedures for the preparation of 1-oxo-1,2,3,4-tetrahydro- β carboline-3-carboxylic acid esters (3)

path A: To a solution of amine (4a-f) (5.0 mmol) in freshly distillated acetonitrile (30-40 ml), triethylamine (5.5 mmol) and then triphosgene (3.3 mmol) were added. Vigorous stirring was maintained for 0.5-1 h under nitrogen, then (1) 30% HBr in AcOH (1.2 ml) was added and the reaction mixture was refluxed for 1.5 h, or (2) gaseous HCl was bubbled (2-5 min.) and the reaction mixture was heated for 2-2.5 h. After evaporation of the solvent the residue was dissolved in CH₂Cl₂ (30 ml), washed successively with water, 10% K₂CO₃, dried over MgSO₄ and evaporated to dryness. The residue was purified by circular or flash chromatography (eluting: CH₂Cl₂ : acetone 4:0->3:1).

<u>path B</u>: (1) In an autoclave, to a solution of iminophosphorane (7a) (1.0 mmol) in acetonitrile (30 ml) dry ice (10-15 mmol) was added and then the reaction mixture was heated at 90°C for 3 days. (2) A solution of 7a (1 mmol) and dimethylcarbonate (6 mmol) in acetonitrile (30 ml) was heated in a sealed tube at 110°C for 3 days. After evaporation of the solvent the residue was alkalinized (10% NaHCO3), extracted with CH₂Cl₂ (3x10 ml), the combined organic layers were dried over MgSO4, filtered and evaporated to dryness. Purification of the solid residue by circular chromatography (elution: CH₂Cl₂: 5% THF) afforded (S)-3a in 45-56 % yield.

<u>path C</u>: To a solution of hemi-acid hemi-ester (**8f-k**) (2.5 mmol) in dry acetonitrile (20 ml) were added successively Proton Sponge[®] (1.05 eq.) and DPPA (1.05 eq.), the reaction mixture was heated under argon for a night. After cooling the solution was (1) treated with 30% HBr in AcOH (1.5 ml), or (2) saturated with gaseous HCl (HCl concentration was maintained by a low-rate bubbling), or (3) after the evaporation of acetonitrile the residue was dissolved in toluene (30 ml) and 30% HBr in AcOH (1.5 ml) was added and the obtained mixture was heated under reflux for 1-2 h. After evaporation of the solvent the residue was dissolved in EtOAc/CHCl3 (6:1) (40 ml), washed successively with water and 5% NaHCO3 solution, the combined organic layers were dried (MgSO4), the solvent was removed under reduced pressure and the crude product (**3f-k**) was purified by circular chromatography (elution: hexane : EtOAc, 5:0 ->1:1; or CH₂Cl₂ with 5->15% acetone).

(S)-3a: mp 178-179°C. $[\alpha]_D$ = +106.4° (*c* 1.02, MeOH). UV (MeOH) 224, 303 nm. IR (KBr) v 3358, 3226, 1736, 1662 cm⁻¹. ¹H NMR (CDCl₃) δ 10.35 (1H, s), 7.59-7.14 (4H, m), 6.55 (1H, s), 4.59 (1H, m), 3.80 (3H, s), 3.50-3.28 (2H, m). ¹³C NMR (CDCl₃) δ 171.4, 162.6, 138.0, 125.9, 125.3, 124.9, 120.2, 120.1, 117.5, 112.9, 55.0, 52.8, 24.0. MS *m*/*z*(%) 244 (M⁺, 41), 185 (100). Anal. calcd. for C_{13H12N2O3} (244.25): C, 63.91; H, 4.95; N, 11.47. Found: C, 63.54; H, 4.67; N, 11.05.

3b: mp 184-185°C. UV (MeOH) 237, 301 nm. IR (KBr) v 3242, 1743, 1640 cm⁻¹. ¹H NMR (CDCl₃) δ 10.73 (1H, s), 7.55 (1H, d, J=8 Hz), 7.42-7.07 (8H, m), 5.83 (1H, d, J=16 Hz), 4.35 (1H, dd, J=7.5, 2 Hz), 4.03 (1H, d, J=16 Hz), 3.62 (3H,s), 3.53 (1H, dd, J=17, 2 Hz), 3.31 (1H, dd, J=17, 7.5 Hz). ¹³C NMR (CDCl₃) δ 171.9, 161.6, 137.9, 137.3, 128.7, 127.9, 127.6, 126.3, 125.0, 124.9, 120.0, 119.8, 115.5, 112.8, 59.3, 52.7, 49.1, 24.3. MS *m*/*z* (%) 334 (M⁺, 73), 275 (83), 243 (27), 232 (8), 220 (15). Anal. calcd. for C₂₀H₁₈N₂O₃ (334.36): C, 71.84; H, 5.43; N, 8.38. Found: C, 71.79; H, 5.16; N, 8.33.

3c: mp 164-166°C. UV (MeOH) 222, 303 nm. IR (KBr) v 3229, 1744, 1643 cm⁻¹. ¹H NMR (CDCl₃) δ 10.78 (1H, s), 7.58-7.08 (4H, m), 4.43 (1H, dd, J=14, 7.5 Hz), 4.40 (1H, d, J=7.2 Hz), 3.63 (3H, s), 3.59 (1H, d, J=17 Hz), 3.43 (1H, dd, J=17, 7.2 Hz), 2.64 (1H, dd, J=14, 7.2 Hz), 1.72 (1H, m), 1.47 (4H, m), 0.98 (6H, t, J=7.5 Hz). MS *m*/*z*(%) 328 (M⁺, 78), 269 (93), 257 (66), 229 (42). HREIMS Calcd. for C19H24N2O3: 328.17869, found: 328.17749.

3d: mp 151-153°C. UV (MeOH) 222, 303 nm. IR (KBr) v 3215, 1740, 1641, cm⁻¹. ¹H NMR (CDCl₃) δ 10.48 (1H, s), 7.55 (1H, d, J=9 Hz), 7.48 (1H, d, J=9 Hz),

7.31 (1H, t, J=9 Hz), 7.11 (1H, t, J=9 Hz), 4.72 (1H, dd, J=15, 7 Hz), 4.35 (1H, dt, J= 14, 5 Hz), 3.72 (3H, s), 3.63 (3H, s), 3.55 (1H, d, J=1.5 Hz), 3.41 (2H, m), 2.95-2.75 (2H, m). ¹³C NMR (CDCl₃) δ 173.0, 172.4, 161.7, 137.8, 126.6, 125.0, 124.9, 120.1, 115.9, 112.6, 62.6, 52.8, 51.7, 44.0, 33.6, 24.4. MS *m*/*z* (%) 330 (M⁺, 65), 299 (13), 271 (86), 239 (56). HREIMS Calcd. for C₁₇H₁₈N₂O₅: 330.12157, found: 330.12143.

3e: mp 215-216°C. UV (MeOH) 222, 296, 304 nm. IR (KBr) v 3343, 3241, 1745, 1672 cm⁻¹. ¹H NMR (CDCl₃, CD₃OD) δ 7.35 (1H, d, J=8 Hz), 7.05-6.91 (2H, m), 4.52 (1H, dd, J=9, 5.5 Hz), 3.85 (3H, s), 3.81 (3H, s), 3.40 (1H, dd, J=16, 5.5 Hz), 3.31 (1H, dd, J=16, 9 Hz). ¹³C NMR (CDCl₃, DMSO-d₆) δ 171.7, 161.6, 153.8, 132.8, 126.6, 124.7, 115.7, 115.5, 113.4, 100.0, 55.2, 54.3, 52.3, 23.6. MS *m*/*z* (%) 274 (M⁺, 73), 257 (21), 215 (100), 200 (29). HREIMS Calcd. for C14H14N2O4: 274.09536, found: 274.09510.

3f (*cis*): amorphous. UV (MeOH) 218, 302 nm. IR (KBr) v 3236, 1736, 1666 cm⁻¹. ¹H NMR (CDCl₃) δ 10.81 (1H, s), 7.50-6.98 (9H, m), 6.72 (1H, d, J=4 Hz), 5.01 (1H, d, J=3.5 Hz), 4.47 (1H, dd, J=4, 3.5 Hz), 4.11 (2H, q, J=7 Hz), 1.12 (3H, t, J=7 Hz). ¹³C NMR (CDCl₃) δ 171.2, 162.7, 140.2, 138.1, 128.6, 127.9, 127.4, 125.9, 125.3, 124.8, 120.6, 120.4, 119.4, 112.8, 63.0, 62.0, 41.0, 13.9. MS *m*/*z*(%) 334 (M⁺·, 45), 261 (100), 233 (52). Anal. calcd. for C₂₀H₁₈N₂O₃ (334.39): C, 71.83; H, 5.43; N, 8.38. Found: C, 71.56; H, 5.32; N, 8.10.

3f (*trans*): amorphous. UV (MeOH) 218, 302 nm. IR (KBr) v 3403, 1732, 1660 cm⁻¹. ¹H NMR (CDCl₃) δ 9.83 (1H, s), 7.51-7.05 (9H, m), 6.23 (1H, s), 5.08 (1H, d, J=5.8 Hz), 4.81 (1H, d, J=5.8 Hz), 4.12 (2H, q, J=7 Hz), 1.18 (3H, t, J=7 Hz). ¹³C NMR (CDCl₃, CD₃OD) δ 166.5, 161.7, 137.9, 136.6, 128.2, 127.6, 125.6, 125.4, 124.0, 121.7, 120.3, 120.1, 112.5, 61.8, 59.7, 40.6, 13.6. MS *m/z*(%) 334 (M⁺, 44), 261 (63), 233 (100). Anal. calcd. for C₂₀H₁₈N₂O₃ (334.39): C, 71.83; H, 5.43; N, 8.38. Found: C, 71.52; H, 5.03; N, 8.15.

3g (*cis*, *trans* 1:1 mixture): amorphous. UV (MeOH) 224, 303 nm. IR (KBr) v 3375, 3242, 1734, 1662 cm⁻¹. ¹H NMR (CDCl₃) δ 10.48 (10.28) (1H, s), 7.52-6.43 (9H, m), 4.78 (1H, d, J=4.5 Hz), 4.28-3.95 (2H, m), 3.86 (1H, m), 3.21-2.81 (3H, m), 2.35 (1H, s, br), 1.33, (1.05) (3H, t, J=8 Hz). ¹³C NMR (CDCl₃) δ 172.0 (169.3), 162.7 (161.5), 138.7 (137.6), 137.6 (137.5), (129.6) 129.2, 128.7 (128.2), 126.7 (126.4), (125.7) 124.8, 125.2 (125.0), (122.8) 121.2, (120.5) 120.4, (120.4) 120.2, 116.1 (113.6), (112.7) 112.5, (62.1) 61.8, (59.4) 58.3, 40.2 (36.4), 38.0 (37.1), (14.2) 13.9. Signals in brackets () belong to the same diastereomer. MS *m/z* (%) 348 (M⁺⁺, 18), 257 (88), 213 (42), 185 (100). Anal. calcd. for C21H20N2O3 (348.39): C, 72.39; H, 5.79; N, 8.04. Found: C, 72.07; H, 6.01; N, 7.88.

3h (*cis, trans* 1:2 mixture): amorphous. UV (MeOH) 224, 303 nm. IR (KBr) v 3233, 1738, 1665 cm⁻¹. ¹H NMR (CDCl₃) δ 11.1 (10.8) (1H, s), 7.50 (7.55) (1H, d, J=9 Hz), 7.38 (7.45) (1H, d, J=9 Hz), 7.21-7.08 (2H, m), 6.51 (7.05) (1H, s), 4.85 (4.35) (1H, d, J=5.5 (10) Hz), 4.01 (4.0) (2H, q, J=7 Hz), 3.85 (3.81) (1H, d, J=5.5 (10) Hz), 2.61 (1.72) (1H, m), 1.48 (1.05) (3H, t, J=7 Hz), 1.45-1.15 (4H, m), 1.25,1.11, (0.99, 0.81) (6H, t, J=9 Hz). ¹³C NMR (CDCl₃) δ (172.7) 169.8, (163.2) 161.8, 137.9 (137.7), 127.1 (125.6), (125.6) 125.5, 124.9 (124.8), 120.8 (120.4), 120.2 (120.0), 120.2 (120.1), 113.0 (112.8), 61.8 (61.6), 60.1 (56.6), (45.4) 44.5, (36.8) 36.2, 24.3 (22.8), 23.2 (22.3), 14.1 (13.8), 12.1 (11.8), 10.8 (10.8). Signals in brackets () belong to the *trans* diastereomer. MS *m/z* (%) 328 M⁺, 17), 257 (55), 213 (36), 185 (100). HREIMS Calcd. for C19H24N2O3: 328.17869, found: 328.17749.

3i (*cis*, *trans* 1:2 mixture): amorphous. UV (MeOH) 225, 304 nm. IR (KBr) v 3230, 1734, 1665 cm⁻¹. ¹H NMR (CDCl₃) δ 10.80 (10.45) (1H, s), 7.61 (1H, d, J=8 Hz), 7.45 (7.51) (1H, d, J=8 Hz), 7.31-7.05 (2H, m), (6.62) 6.40 (1H, s), 4.45 (4.81) (1H, d, J=4 (6) Hz), 4.01 (4.42) (2H, q, J=7 Hz), 3.52 (1H, m), 1.95-1.52 (11H, m), 1.38 (1.09) (3H, t, J=7 Hz). Signals in brackets () belong to the *trans* diastereomer. ¹³C NMR [**3i** (*trans*)] (CDCl₃) δ 169.8, 161.6, 137.5, 137.4, 126.8, 125.0, 121.4, 120.3, 112.8, 62.0, 59.8, 40.9, 40.4, 33.2, 28.7, 26.8, 26.3, 26.1, 25.6, 14.2. MS *m/z* (%) 340 (M⁺, 37), 325 (28), 257 (52), 251 (31), 213 (33). HREIMS Calcd. for C₂₀H₂₄N₂O₃: 340.17869, found: 340.17715.

3j (*cis*, *trans* 1:1 mixture): amorphous. UV (MeOH) 223, 304 nm. IR (KBr) v 3347, 1736, 1667 cm⁻¹. ¹H NMR (CDCl₃) δ 10.40 (10.15) (1H, s), 7.71 (1H, d, J=1.5 Hz), 7.41-7.23 (2H, m), 6.31 (6.25) (1H, s), 4.65 (4.15) (1H, d, J=5 (9) Hz), 4.38 (4.11) (2H, m), (3.82) 3.63 (1H, m), (1.48) 1.14 (3H, d, J=8 Hz), 1.37 (1.15) (3H, t, J=8 Hz). ¹³C NMR (CDCl₃) δ (171.7) 169.3, (162.0) 161.0, 136.6 (136.2), (128.3) 128.0, 126.1 (125.6), 125.4 (125.3), 123.2 (122.6), (122.4) 122.2, 114.5 (114.4), (114.3) 113.4, (113.7) 112.2, 62.1 (61.6), (59.2) 59.1, (30.0) 29.4, (20.1) (14.3), 14.1 (14.0). Signals in brackets () belong to the *trans* diastereomer. MS *m*/*z* (%) 352 (M⁺, 22), 350 (M⁺, 24), 308 (8), 306 (7), 279 (66), 277 (68), 262 (13), 260 (15). HREIMS Calcd. for C15H15N2O3Br: 350.02660 and 352.02456, found: 350.02663 and 352.02023.

3k (cis, trans 1:1 mixture): amorphous. UV (MeOH) 223, 304 nm. IR (KBr) v

3346, 3240, 1746, 1668 cm⁻¹. ¹H NMR (CDCl₃) δ 10.50 (10.27) (1H, s), 7.65 (7.68) (1H, d, J=8 Hz), 7.42 (7.45) (1H, d, J=8 Hz), 7.40-7.12 (7H, m), 6.55 (6.39) (1H, s, br), 5.17 (5.12) (2H, d, J=2 Hz), (4.68) 4.66 (1H, d, J=4.5 Hz), 4.31-4.02 (5H, m), 1.25 (1.24) (3H, t, J=7 Hz). Signals in brackets () belong to the same diastereomer. ¹³C NMR [3k (*trans*)] (CDCl₃) δ 169.0, 161.0, 154.6, 137.6, 134.9, 128.5, 128.4, 128.1, 126.5, 125.6, 124.2, 120.9, 120.0, 117.7, 113.0, 69.7, 64.5, 62.2, 56.8, 34.9, 13.6. MS *m*/*z* (%) 422 (M⁺, 3), 314 (2), 270 (45), 197 (100). HREIMS Calcd. for C₂₃H₂₂N₂O₆: 422.14825, found: 422.14779.

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REFERENCES

- Lévy, J.; Laronze, J.-Y.; Nagy, T.; Sapi, J.; Séraphin, D.; Renard, P.; Scalbert, E.; (ADIR et Cie) Fr. Demande N° 94.01234 (04.02.1994.)
- Blankley, C.J.; Hodges, J.C.; Klutchko, S.R.; Himmelsbach, R.J.; Chucholowski, A.; Connolly, C.J.; Neergaard, Van Nieuwenhze, M.S.; Sebastien A.; Quin III, J.; Essenburg, A.D.; Cohen, D.M. J. Med. Chem. 1991, 34, 3248.
- (a) Bracher, F.; Hildebrand, D. Liebigs Ann. 1992, 1315. (b) Bracher, F.; Hildebrand, D. Pharmazie, 1993, 48, 695.
- 4. For a recent review, see: Molina, P.; Vilaplana M.J. Synthesis, 1994, 1197.
- 5. Barrett, H.S.B.; Perkin W.H.Jr.; Robinson, R. J. Chem. Soc. 1930, 2942.
- Jeannin, L.; Nagy, T.; Vassileva, E.; Sapi, J.; Laronze, J.-Y. *Tetrahedron* Lett. 1995, 36, 2057.
- 7. (a) Shioiri T.; Ninomiya, K.; Yamada, S. J. Am. Chem. Soc. 1972, 94, 6203. (b) Ninomiya, K.; Shioiri, T.; Yamada, S. Tetrahedron, 1974, 30, 2151.

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