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Total synthesis of Eudistomins Y_1-Y_6

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ABSTRACT

The first total synthesis of Eudistomins Y_1 – Y_6 , brominated phenolic β -carboline marine metabolites with a unique benzoyl moiety at C1, have been prepared in three steps, utilizing MAOS, in overall yields ranging from 6% to 25%.

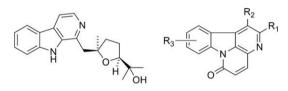
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β-Carboline alkaloids are a prevalent class of biologically active natural products **1–4** with a wide range of structural (Fig. 1) and pharmacological (cytotoxic, antiviral, antimicrobial, etc.) diversity.^{1–11} Of these, the eudistomins represent an ever-expanding sub-class isolated from marine tunicates of the *Eudistoma* genus.^{12–21} Since the initial discovery by Rinehart in 1987 of Eudistomins A–Q (**4**),¹² additional members Eudistomins R–W have been reported.^{12–21}

In 2008, seven new β -carboline-based metabolites, coined Eudistomins Y_1-Y_7 (Fig. 2) were isolated from a tunicate of the genus *Eudistoma* off the coast of Korea by Kang and co-workers. ²² These new metabolites differ from all previously reported Eudistomins A–W by the presence of a benzoyl group at C1. Preliminary biological evaluation demonstrated that Eudistomin Y_6 (10) had a moderate antibacterial activity against Gram-positive bacteria (*Staphylococcus epidermis* and *Bacillus subtilis*, MICs of 12.5 and 25 µg/mL, respectively) without cytotoxicity in an MTT assay at 100 µM. ²² However, no synthetic efforts toward these novel metabolites have been reported to date.

We, and others, have developed expedited synthetic routes to access β -carboline alkaloids, and our laboratory has also synthesized unnatural analogs with unique and unexpected biological activities.²³ Based on the unique structures of Eudistomins Y_1-Y_7 (5–11), the initial biological activity, and the potential for diversity-oriented synthesis once an expedient synthetic route was in place, we initiated a total synthesis campaign targeting 5–11.

Our retrosynthetic analysis is shown in Scheme 1. We envisioned the direct precursor of Eudistomins Y_1-Y_7 (**5–11**) to be an appropriately functionalized 1-benzyl-4,9-dihydro-3*H*-pyrido[3,4-b]indole **12**, that could be oxidized to deliver **5–11**.²⁴ Tricycle **12** would be accessed through a Bischler–Napieralski reaction²⁵ with intermediate **13**, which could be prepared by a coupling reaction



1: (-)-Isocyclocapitelline

2: Canthine Alklaoids

3: 3,10-Dibromomfascaplysin

4: Eudistomins A-Q

Figure 1. Representative β -carboline alkaloids **1–4**.

 5: Eudistomin Y_1 $R_1 = H$ $R_2 = H$ $R_3 = H$ $R_4 = H$

 6: Eudistomin Y_2 $R_1 = Br$ $R_2 = H$ $R_3 = H$ $R_4 = H$

 7: Eudistomin Y_3 $R_1 = H$ $R_2 = H$ $R_3 = Br$ $R_4 = H$

 8: Eudistomin Y_4 $R_1 = Br$ $R_2 = H$ $R_3 = Br$ $R_4 = H$

 9: Eudistomin Y_5 $R_1 = H$ $R_2 = H$ $R_3 = Br$ $R_4 = Br$

 10: Eudistomin Y_6 $R_1 = Br$ $R_2 = H$ $R_3 = Br$ $R_4 = Br$

 11: Eudistomin Y_7 $R_1 = H$ $R_2 = Br$ $R_3 = Br$ $R_4 = Br$

Figure 2. Structures of Eudistomins Y_1 – Y_7 (5–11).

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$$\begin{array}{c} R_4 & \text{OH} \\ R_3 & \text{OH} \\ R_1 & \text{R}_3 \\ R_2 & \text{Bischler-Napieralski} \\ \end{array}$$

Scheme 1. Retrosynthetic analysis of Eudistomins $Y_1 - Y_7$ (6–11).

between the appropriately functionalized *p*-hydroxyphenylacetic acid **14** and (1*H*-indol-3-yl)ethanamine **15**.

Our synthetic efforts initially focused on Eudistomin Y_1 (**5**), the simplest member of this class (Scheme 2). A standard EDCI/HOBt coupling reaction between (1*H*-indol-3-yl)ethanamine **16** and *p*-hydroxyphenylacetic acid **17** gave **18** in 79% yield. Classical Bischler–Napieralski conditions proved sluggish, so we developed microwave-assisted conditions (POCl₃, toluene, 120 °C, 30 min) which smoothly delivered **19**. Multiple oxidation conditions were also explored, including standard hv/O₂, but good results were ultimately achieved with MnO₂ under another microwave-assisted protocol to produce Eudistomin Y_1 (**5**) in 31% yield for the two steps with crude **19**. Thus, a rapid three-step, two-pot sequence was designed and optimized to access the Eudistomin Y_1 – Y_7 (**5**–**11**) scaffold in 25% overall yield.²⁶

To access the brominated congeners, Eudistomins Y_2-Y_7 (**6–11**), we were pleased to find that all of the requisite starting materials were readily accessible, except the 2-(6-bromo-1*H*-indo-3-yl)eth-anamine required to synthesize **11**, which was readily prepared.

Similarly, Eudistomin Y_2 (**6**) was prepared by a standard EDCI/HOBt coupling reaction between 2-(5-bromo-1*H*-indol-3-yl)ethanamine **19** and *p*-hydroxyphenylacetic acid **17** gave **20** in 90% yield (Scheme 3). Our microwave-assisted Bischler–Napieralski conditions provided **21**, which carried forward crude into a microwave-assisted MnO₂ protocol to deliver Eudistomin Y_2 (**6**) in 20% yield for the two steps and 18% overall.

Eudistomin Y₃ (**7**), containing a bromine on the benzoyl moiety, was prepared by a standard EDCI/HOBt coupling reaction between

Scheme 2. Total synthesis of Eudistomin Y_1 (5).

Scheme 3. Total synthesis of Eudistomin Y2 (6).

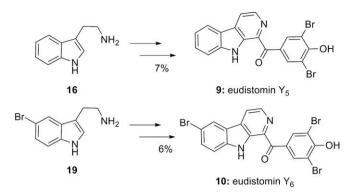
Scheme 4. Total synthesis of Eudistomin Y₃ (7).

(1H-indol-3-yl)ethanamine **16** and 3-bromo-4-hydroxyhydroxyphenylacetic acid **22** to afford **23** in 98% yield (Scheme 4). Application of a now standard microwave-assisted Bischler-Napieralski protocol provided crude **24**, followed by our microwave-assisted MnO₂ protocol to deliver Eudistomin Y₃ (**7**) in 15% yield.

Eudistomin Y_4 (**8**) possess bromines on both the (1*H*-indol-3-yl)ethanamine component **14** and the *p*-hydroxyphenylacetic acid **15**. Fortunately, our standard three-step, two-pot sequence proved to work with equivalent efficiency (Scheme 5). In the event, 2-(5-bromo-1*H*-indol-3-yl)ethanamine **19** was coupled to 3-bromo-4-hydroxyhydroxyphenylacetic acid **22** employing EDCI/HOBt conditions to provide **25**. Two successive microwave-assisted reactions (Bischler–Napieralski and MnO₂ oxidation) delivered Eudistomin Y_4 (**8**) in 15% overall yield.

Following the protocols outlined in Schemes 1–4, Eudistomins Y_5 and Y_6 (**9** and **10**) were synthesized in three steps with overall yields of 7% and 6%, respectively (Scheme 6). While the amide coupling steps proceeded in high yields, the polybrominated sub-

Scheme 5. Total synthesis of Eudistomin Y₄ (8).



Scheme 6. Total synthesis of Eudistomin Y_5 (9) and Y_6 (10).

strates performed poorly in the microwave-assisted Bischler-Napieralski and MnO₂ oxidation reactions.

In every case, the ¹H and ¹³C NMR spectra of the synthetic Eudistomins Y₁-Y₆ matched that reported for the natural products (5-**10**).²⁷ Attempts to prepare Eudistomin Y₇ failed employing this methodology, due perhaps to stereoelectronic effects and or solubility issues of the polybrominated scaffold. Overall yields for the three-step, two-pot process were low, but not unexpected based on the electronics of the system with the carbonyl moiety at C1. For the Bischler-Napieralski and MnO₂ oxidation steps, thermal conditions failed entirely. Only MAOS provided the desired Eudistomin scaffold, but in modest to poor yields.

Thus, the first total synthesis of Eudistomins Y_1-Y_6 (5–10) has been completed, requiring only three synthetic steps in a twopot process and with overall yields ranging from 6% to 25%. We are currently evaluating 5-10 against a large panel of discrete molecular targets in radioligand binding assays,²⁸ and we are in the process of initiating a diversity-oriented synthesis campaign to synthesize libraries of unnatural analogs.²⁹ These efforts are underway and will be reported in due course.

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References and notes

- 1. Trujillo, J. I.; Meyers, M. J.; Anderson, D. R.; Hegde, S.; Mahoney, M. W.; Vernier, W. F.; Buchler, I. P.; Wu, K. K.; Yang, S.; Hartmann, S. J.; Reitz, D. B. Bioorg. Med. Chem. Lett. 2007, 17, 4657.
- Winkler, J. D.; Londregan, A. T.; Ragains, J. R.; Hamann, M. T. Org. Lett. 2006, 8,
- Costa, E. V.: Pinheiro, M. L. B.: Xavier, C. M.: Silva, I. R. A.: Amaral, A. C. F.: Souza, A. D. L.; Barison, A.; Campos, F. R.; Ferreira, A. G.; Machado, G. M. C.; Leon, L. L. P. J. Nat. Prod. 2006, 69, 292.
- 4. Phuong, N. M.; Van Sung, T.; Porzel, A.; Schmidt, J.; Merzweiler, K.; Adam, G. Phytochemistry 1999, 52, 1725.
- Hartung, J.; Drees, S.; Geiss, B.; Schmidt, P. Synlett 2003, 223.
- Potts, K. T.; Mattingly, G. S. J. Org. Chem. 1968, 33, 3985.
- Gribble, G. W.; Barden, T. C.; Johnson, D. A. Tetrahedron 1988, 44, 3195.
- Lipinska, T. Tetrahedron Lett. 2002, 43, 9565.
- Segraves, N. L.; Robinson, S. J.; Garcia, D.; Said, S. A.; Fu, X.; Schmitz, F. J.; Pietraszkiewicz, H.; Valeriote, F. A.; Crews, P. J. Nat. Prod. **2004**, *67*, 783.
- Zhidkov, M. E.; Baranova, O. V.; Balaneva, N. N.; Fedorov, S. N.; Radchenko, O. S.; Dubovitskii, S. V. Tetrahedron Lett. 2007, 48, 7998.
- Ohmoto, T.; Koike, K.. In The Alkaloids; Brossi, A., Ed.; Academic Press: New York, 1989; Vol. 36, pp 135-170.
- Rinehart, K. L.; Kobayashi, J.; Harbour, G. C.; Hughes, R. G., Jr.; Mizsak, S. A.; Scahill, T. A. J. Am. Chem. Soc. 1984, 106, 1524.
- Kobayashi, J.; Harbour, G. C.; Gilmore, J.; Rinehart, K. L., Jr. J. Am. Chem. Soc. **1984**, 106, 1526.

- 14. Kinzer, K. F.; Cardellina, J. H., II Tetrahedron Lett. 1987, 28, 925.
- Schupp, P.; Poehner, T.; Edrada, R.; Ebel, R.; Berg, A.; Wray, V.; Proksch, P. J. Nat. Prod. 2003, 66, 272,
- Kobayashi, J.; Nakamura, H.; Ohizumi, Y.; Hirata, Y. Tetrahedron Lett. 1986, 27, 1191
- Kobayashi, J.; Cheng, J. F.; Ohta, T.; Nozoe, S.; Ohizumi, Y.; Sasaki, T. J. Org. Chem. 1990, 55, 3666
- Murata, O.; Shigemori, H.; Ishibashi, M.; Sugama, K.; Hayashi, K.; Kobayashi, J. Tetrahedron Lett. 1991, 32, 3539.
- Adesanya, S. A.; Chbani, M.; Pais, M.; Debitus, C. J. Nat. Prod. 1992, 55, 525.
- 20 Kang, H.; Fenical, W. Nat. Prod. Lett. 1996, 9, 7
- 21. Van Wagoner, R. M.; Jompa, J.; Tahir, A.; Ireland, C. M. J. Nat. Prod. 1999, 62, 794.
- Wang, W.; Nam, S.-J.; Lee, B.-C.; Kang, H. *J. Nat. Prod.* **2008**, *71*, 163.
 (a) Lindsley, C. W.; Wisnoski, D. D.; Wang, Y.; Leister, W. H.; Zhao, Z. *Tetrahedron Lett.* **2003**, *44*, 4495; (b) Lindsley, C. W.; Bogusky, M. J.; Leister, W. H.; McClain, R. T.; Robinson, R. G.; Barnett, S. F.; Defeo-Jones, D.; Ross, C. W.; Hartman, G. D. Tetrahedron Lett. 2005, 46, 2779.
- (a) Radchenko, O. S.; Novikov, V. L.; Elyakov, G. B. Tetrahedron Lett. 1997, 38, 5339; (b) Garcia, M. D.; Wilson, J. A.; Emmerson, D. P. G.; Jenkins, P. R.; Mahale, S.; Chaudhuri, B. Org. Biomol. Chem. 2006, 4, 4478.
- (a) Wolfe, J.P. Bischler-Napieralski Reaction in Name Reactions in Heterocyclic Chemistry 2005, p 376; (b) Pal, B.; Jaisankar, P.; Giri, V. S. Synth. Commun. 2003, 33, 2339.
- experimental. Tryptamine 16 (1 g, hydroxyphenylacetic acid 17 (0.94 g, 6.2 mmol), and N-hydroxybenzotriazole (1.76 g, 13.0 mmol) were added to a 250 ml round-bottomed flask, and dissolved in 9:1 DMF/DIEA (50 ml). 1-Ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride (3.6 g, 18.6 mmol) was then added and the reaction was allowed to stir overnight. Once complete the reaction was quenched with 1 N HCl and extracted three times with DCM (100 ml). The organic layer was dried with MgSO₄, filtered, and concentrated. The crude residue was purified by column chromatography to yield coupled product 18 (1.39 g, 4.7 mmol) in 76% yield.

Coupled product 18 (0.25 g, 0.850 mmol) was added to a 5 ml microwave vial and dissolved in toluene (3 ml). POCl₃ (0.79 ml, 8.50 mmol) was then added all at once and the reaction vessel capped, and heated to 120 °C for 30 min. Once complete the toluene was removed and the reaction quenched with satd NaHCO₃ and extracted three times with DCM (100 ml). The organic layer was dried with MgSO4, filtered and concentrated to obtain 21 which was used without further purification.

Crude 21 (100 mg, 0.362 mmol), was added to a 20 ml MW vial and partially dissolved in DCE (10 ml). MnO₂ (315 mg, 3.62 mmol) was then added all at once and the reaction vessel capped and heated to 160 °C for 60 min. Once complete the reaction was vacuum filtered, concentrated, and purified by preparative HPLC to obtain **5** (32.4 mg, 0.112 mmol) in 31% yield.

NMR (¹H and ¹³C), Hi-RES MS data for Y₁-Y₆. Eudistomins Y₁: ¹H NMR (400 MHz, DMSO- d_6) δ 10.39 (br s, 1H), 8.51 (d, J = 5.2 Hz, 1H), 8.41 (d, J = 5.2 Hz, 1H), 8.31 (d, J = 7.6 Hz, 1H), 8.24 (dt, J = 8.8, 2.8 Hz, 2H), 7.77 (d, J = 8.4 Hz, 1H), 7.59 (t, J = 7.2 Hz, 1H), 7.30 (t, J = 7.2 Hz, 1H), 6.92 (dt, J = 8.8, 2.4 Hz, 2H). ¹³C NMR (150 MHz, DMSO- d_6) δ 191.6, 161.9, 141.6, 137.3, 136.7, 135.6, 133.7, 130.8, 128.9, 128.2, 121.8, 120.1, 120.0, 118.2, 114.9, 112.9. HRMS (Q-TOF): m/z calcd for $C_{18}H_{13}N_2O_2$ [M+H⁺]: 289.0977, found: 289.0973.

Euclistomis Y₂: ¹H NMR (400 MHz, DMSO- d_6) 7, 10th 1. 263-0373. Euclistomis Y₂: ¹H NMR (400 MHz, DMSO- d_6) 8 10.42 (br s, 1H), 8.53 (d, J = 5.2 Hz, 1H), 8.45 (d, J = 4.8 Hz, 1H), 8.23 (d, J = 8.8 Hz, 2H), 7.72 (s, 2H), 6.91 (d, J = 8.8 Hz, 2H).). ¹³C NMR (150 MHz, DMSO- d_6) δ 191.1, 161.9, 140.2, 137.7, 137.1, 135.7, 133.7, 131.2, 129.7, 128.0, 124.4, 122.0, 118.7, 114.8, 112.0. HRMS (Q-TOF): m/z calcd for $C_{18}H_{12}N_2O_2Br$ [M+H⁺]: 367.0082, found: 367.0081.

Eudistomins Y₃: 1 H NMR (400 MHz, DMSO- d_{6}) δ 11.30 (s, 1H), 8.56 (m, 2H), 8.44 (d, J = 5.2 Hz, 1H), 8.31 (d, J = 8.0 Hz, 1H), 8.33 (m, 1H), 7.78 (d, J = 8.4 Hz, 1Hz)121.8, 120.1, 120.0, 118.7, 115.6, 112.9, 108.8. HRMS (Q-TOF): m/z calc for $C_{18}H_{12}N_2O_2Br$ [M+H *]: 367.0082; found: 367.0082.

Eudistomins Y₄: ¹H NMR (400 MHz, DMSO- d_6) δ 10.38 (br s, 1H), 8.51 (d, J = 4.8 Hz, 1H), 8.42 (d, J = 4.8 Hz, 1H), 8.31 (d, J = 8.0 Hz, 1H), 8.24 (d, J = 6.8 Hz, (d, J = 7.2 Hz, 2H), 7.77 (d, J = 8.4 Hz, 1H), 7.59 (t, J = 8.0 Hz, 1H), 7.30 (t, J = 8.0 Hz, 1H), 6.92 (d, J = 7.2 Hz, 2H). 13 C NMR (150 MHz, DMSO- d_6) δ 190.6, 160.1, 140.6, 137.6, 137.4, 136.7, 133.7, 136.3, 131.7, 130.0, 128.9, 124.7, 121.8, 120.1, 115.9, 114.9, 112.9, 108.4. HRMS (Q-TOF): m/z calcd for $C_{18}H_{11}N_2O_2Br_2$ [M+H⁺]: 444.9817; found: 444.9817

Eudistomins Y₅: 1 H NMR (400 MHz, DMSO- d_{6}) δ 10.15 (br s, 1H), 8.53 (m, 3H), 8.43 (m, 1H), 8.30 (d, J = 7.6 Hz, 1H), 7.80 (d, J = 9.2 Hz, 1H), 7.58 (t, J = 6.8 Hz, 1H), 7.28 (t, J = 8.0 Hz, 1H). 13 C NMR (150 MHz, DMSO- d_6) δ 190.1, 157.8, 144.7, 138.8, 136.7, 135.9, 135.4, 135.1, 132.6, 130.9, 126.6, 121.4, 120.3, 120.1, 119.8, 113.7, 112.2, 110.7. HRMS (Q-TOF): m/z calcd for $C_{18}H_{10}N_2O_2NaBr_2$ [M+Na⁺]: 466.9007, found: 466.9009.

Eudistomins Y_6 : 1H NMR (400 MHz, DMSO- d_6) δ 8.62 (br s, 1H), 8.58 (d, J = 4.8 Hz, 1H), 8.52 (m, 3H), 7.75 (m, 2H). ¹³C NMR (150 MHz, DMSO- d_6) δ 189.2, 156.0, 140.8, 137.9, 136.4, 135.8, 135.2, 131.9, 130.6, 128.9, 126.6, 125.0, 122.5, 120.9, 120.0, 115.5, 112.7, 111.4. HRMS (Q-TOF): m/z calcd for $C_{18}H_{10}N_2O_2Br_3$ [M+H⁺]: 522.8292; found: 522.8287.

- Kennedy, J. P.; Brogan, J. T.; Lindsley, C. W. J. Nat. Prod. 2008, 71, 1783
- (a) Kennedy, J. P.; Conn, P. J.; Lindsley, C. W. Bioorg. Med. Chem. Lett. 2009, 19, 3204; (b) Lewis, J. A.; Daniels, N. R.; Lindsley, C. W. Org. Lett. 2008, 10, 4545.