

# Synthesis and Antifungal Activity of Some New Benzimidazole Derivatives

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Synthesis and antifungal evaluation of 5-ethoxycarbonyl-2-(substituted-benzyl or phenoxyethyl)benzimidazole are reported. Structures of the compounds were elucidated with IR-,  $^1\text{H}$ -NMR-,  $^{13}\text{C}$ -NMR-, mass-spectra and elemental analysis. Preliminary results show that none of the synthesized benzimidazole derivatives has antifungal activity at the concentration of 100  $\mu\text{g/ml}$  against *Candida parapsilosis*, *Candida stellatoidea*, and *Candida pseudotropicalis*.

Synthese und antimykotische Wirkung einiger neuer Benzimidazol-Derivate

Synthese und antimykotische Prüfung einiger 5-Ethoxycarbonyl-2-(subst. benzyl oder phenoxyethyl)benzimidazole werden beschrieben. Die Strukturen wurden durch IR-,  $^1\text{H}$ -NMR,  $^{13}\text{C}$ -NMR und Massenspektren sowie Elementaranalysen gesichert. Erste Resultate zeigen, daß diese Benzimidazol-Derivate keine Aktivität gegen *Candida albicans*, *C. parapsilosis*, *C. stellatoidea* und *C. pseudotropicalis* in der Konzentration 100  $\mu\text{g/ml}$  haben.

Since the benzimidazole ring was shown to inhibit the growth of certain yeasts<sup>1)</sup> many benzimidazole compounds have been synthesized<sup>2,3)</sup>. However, as resistance to benzimidazoles is developed within a short period of time, the use of benzimidazoles as agricultural fungicides was given up recently<sup>4)</sup>. The imidazoles, 1,2,4-triazoles, and pyrimidines which were prepared by chemical variations of benzimidazoles are most important antimycotica and agricultural fungicides, respectively<sup>3)</sup>.

In the present paper, the synthesis and antifungal activity of new simple benzimidazole derivatives are described.

## Results and Discussion

The synthesis of the benzimidazole compounds involved two steps: 1) preparation of mono amide derivatives by reaction of ethyl 3,4-diaminobenzoate with chlorides of substituted phenyl or phenoxyacetic acids; 2) preparation of the final products, by dehydration of the intermediates 1-8, with anhydrous  $\text{ZnCl}_2$  and dry HCl gas. Only compound 9 was obtained by Phillips' method<sup>5)</sup> in very low yield. -  $^{13}\text{C}$ -NMR shift assignments of compounds 1a-8a and 9 are given in Tab. 1.

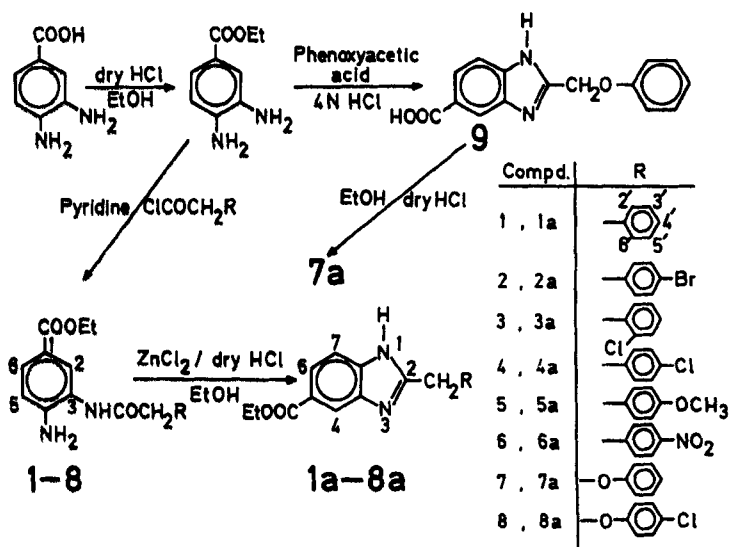
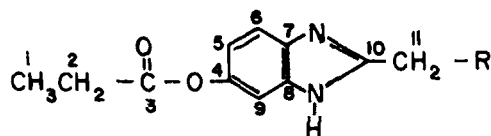


Table 1:  $^{13}\text{C}$ -chemical shift assignment of compounds 1a-9

C number Comp. number	1	2	3	4	5	6	7	8	9	10	11	R
1 a	14.2	32.3	165.2	133.7	126	114.3	135	133	115.4	155	60.3	
2 a	13.9	32.3	165.3	135	127	114.4	135	132	115.7	154	61.35	
3 a	14.5	31.4	165.5	136.6	127.2	114.7	135.4	132.5	115.9	154.8	61.3	
4 a	13.5	31.7	165	133.7	127	114.3	136.8	132.8	115.4	154.2	61.5	
5 a	14.2	31.6	165	135	126.09	114.5	135	132	115.4	155.7	61.3	
6 a	14.3	33	162	135.6	126.9	115.1	135.6	132.9	116.9	152.2	61.8	
7 a	14.3	33	165	132.8	126.6	114.7	135.7	132.7	115.2	154.7	62.2	
8 a	14.6	28.8	169	133.8	126.0	115.6	135.7	130.1	114.1	155.9	62.7	
9	-	-	166.4	130	125.1	115.2	130.1	130	115.3	153.3	64	

Antifungal activities of compounds 1a-8a, and 9 against *Candida parapsilosis*, *C. stellatoidea*, *C. albicans*, and *C. pseudotropicalis* were studied by liquid dilution in the test-tube (micrograms per ml)<sup>6</sup>. Since the compounds were insufficiently soluble in water, all compounds were dissolved in DMSO (1 mg/ml) and then final concentrations were adjusted to 100 µg/ml, 50 µg/ml, 25 µg/ml by using liquid medium and in the presence of the pertinent fungus suspension. Final inculum of microorganism was realized as  $10^{-5}$ - $10^{-6}$  cells/ml. Higher doses were not used because of the influence of DMSO. None of the compounds was active even at 100 µg/ml doses against the fungi mentioned above.

## Experimental Part

Melting points: Büchi SMP-20 capillary m.p. apparatus, uncorrected. - CHN-microanalyses: Hewlett-Packard 185 C,H,N analyser. -  $^1\text{H}$ -NMR (80,13 MHz) and  $^{13}\text{C}$ -NMR (20,1 MHz) spectra: Bruker AC-80 spectrometer,  $d_6$ -DMSO, TMS as internal standard,  $\delta$  (ppm) scale. - Mass spectra: Varian MAT CH 7. - IR (KBr) SP-1100 Pye-Unicam. All N-C=O-bands at  $1655$ - $1665\text{ cm}^{-1}$ , all O-C=O-bands at  $1730$ - $1735\text{ cm}^{-1}$ , if not stated otherwise.

Ethyl 3,4-diaminobenzoate<sup>7</sup> was prepared in our laboratory; 3,4-diaminobenzoic acid from Merck, Fluka; phenylacetic acid derivatives from Aldrich. Sabouraud Dextrose Broth (Oxoid) liquid media was used for the antifungal activity.

## Method A

The related phenoxyacetic acids or phenylacetic acids (7.35 mmol) were refluxed in benzene (5 ml) with  $\text{SOCl}_2$  (2.5 ml) for 60 min at  $80^\circ\text{C}$ . Then solvent and excess of  $\text{SOCl}_2$  were evaporated and ethyl 3,4-diaminobenzoate (7.5 mmol in 15 ml of benzene) and 7.35 mmol of pyridine were added. The mixture was refluxed for 3 h. Removal of the solvent gave a residue which was crystallized from  $\text{CHCl}_3/\text{EtOH}$  (2:8) to give compounds 1-8. Melting points and yields of these intermediates are given below.

$^1\text{H}$ -NMR spectra of 1-8: 1.30-1.40 (t, 3H,  $J = 7\text{ Hz}$ ,  $\text{CH}_2\text{CH}_3$ ), 4.25-4.30 (q, 2H,  $J = 7\text{ Hz}$ ,  $\text{CH}_2\text{CH}_3$ ), 3.65-4.75 (s, 2H,  $\text{COCH}_2$  and  $\text{OCH}_3$  protons of compound 5), 6.80-8.1 (aromatic protons), 8.1-8.2 (s, 1H,  $\text{NHCO}$ ), 9.70 (s, 2H,  $\text{NH}_2$ ). The integral values support the proton number.

## Method B

To a mixture of 2 mmol of comp. 1-8 and absol. EtOH (20 ml) were added 11 mmol freshly prepared anhydrous  $\text{ZnCl}_2$ ; after dissolving all  $\text{ZnCl}_2$ , dry HCl gas (approximately 1.5 g) was passed through the clear

Comp. No	1	2	3	4	5	6	7	8
m.p. °C	170	201	203	195	165	248	168	213
Yield (%)	27.4	51.2	31.8	47.2	45.6	31.5	48.3	57.3

solution which was refluxed until the starting materials were used up (at least 3 h). Then, ethanol was evaporated, dilute  $\text{NH}_3$  solution was added, and the mixture was extracted with  $\text{CHCl}_3$ . The chloroform solution was washed with water 3 times, dried ( $\text{Na}_2\text{SO}_4$ ) and evaporated.

The oily residue was dissolved in EtOH, 0.02 ml HCl acid (25 %) were added and the mixture was stirred vigorously. After addition of ether, the HCl salts of the benzimidazole derivatives **1a-8a** precipitated.

#### 2-Phenylmethyl-5-[1H]benzimidazole carboxylic acid ethyl ester HCl (**1a**)

**1a** was obtained from compound **1** as white crystals, m.p. 233°C, yield 47.3 %. -  $^1\text{H-NMR}$ : 1.40 (t,  $J = 7$  Hz, 3H,  $\text{CH}_2\text{CH}_3$ ), 4.40 (q,  $J = 7$  Hz, 2H,  $\text{CH}_2\text{CH}_3$ ), 4.60 (s, 2H,  $\text{CH}_2$ ), 7.25-7.6 (m, 5H, phenyl), 7.9 (dd,  $J_{7,6} = 8$  Hz,  $J_{7,4} = 0.7$  Hz, 1H, H-7), 8.1 (dd,  $J_{6,7} = 8$  Hz,  $J_{4,6} = 1.4$  Hz, 2H, H-4,6), 8.30 (s, 1H,  $\text{N}^1\text{-H}$ ), 8.75 (broad s,  $\text{N}^+\text{-H}$ ). - MS (70 eV)  $m/z$ : 280 (100 %,  $\text{M}^+$ ), 251 (13), 235 (50), 207 (13), 91 (8). -  $\text{C}_{17}\text{H}_{17}\text{O}_2\text{N}_2\text{Cl}$  (316.5) Calcd. C 64.5 H 5.37 N 8.85 Found C 64.8 H 5.10 N 9.10.

#### 2-(4-Bromophenyl)methyl-5-[1H]benzimidazole carboxylic acid ethyl ester HCl (**2a**)

**2a** was obtained from **2** as white needles, m.p. 255°C, yield 43 %. -  $^1\text{H-NMR}$ : 1.40 (t,  $J = 7$  Hz, 3H,  $\text{CH}_2\text{CH}_3$ ), 4.40 (q,  $J = 7$  Hz, 2H,  $\text{CH}_2\text{CH}_3$ ), 4.60 (s, 2H,  $\text{CH}_2$ ), 7.55 (s, 4H, H-2',3',5',6'), 7.8 (dd,  $J_{7,6} = 8$  Hz,  $J_{7,4} = 0.7$  Hz, 1H, H-7), 8.1 (dd,  $J_{6,7} = 8$  Hz,  $J_{4,6} = 1.4$  Hz, 2H, H-4,6), 8.25 (s, 1H,  $\text{N}^1\text{-H}$ ), 9.70 (broad s,  $\text{N}^+\text{-H}$ ). - MS (70 eV)  $m/z$ : 360 (100 %,  $\text{M}^+$ ), 358 (100,  $\text{M}^+$ ), 345 (12), 343 (12), 331 (16), 329 (16), 315 (56), 313 (56), 250 (12), 207 (28), 172 (36), 170 (36), 118 (44). -  $\text{C}_{17}\text{H}_{16}\text{O}_2\text{N}_2\text{BrCl}$  (395.5) Calcd. C 51.6 H 4.05 N 7.08 Found C 51.5 H 4.05 N 7.50.

#### 2-(2-Chlorophenyl)methyl-5-[1H]benzimidazole carboxylic acid ethyl ester HCl (**3a**)

**3a** was obtained from **3** as white needles, m.p. 216°C, yield 38 %. -  $^1\text{H-NMR}$ : 1.40 (t,  $J = 7$  Hz, 3H,  $\text{CH}_2\text{CH}_3$ ), 4.40 (q,  $J = 7$  Hz, 2H,  $\text{CH}_2\text{CH}_3$ ), 4.6 (s, 2H,  $\text{CH}_2$ ), 7.25-7.7 (m, 4H, H-3',4',5',6'), 7.8 (dd,  $J_{7,6} = 9$  Hz,  $J_{7,4} = 0.7$  Hz, 1H, H-7), 8.1 (dd,  $J_{6,7} = 9$  Hz,  $J_{4,6} = 1.45$  Hz, 2H, H-4,6), 8.30 (s, 1H,  $\text{N}^1\text{-H}$ ), 6.75 (s,  $\text{N}^+\text{-H}$ ). - MS (70 eV)  $m/z$ : 316 (4 %,  $\text{M}^+$ ), 314 (12,  $\text{M}^+$ ), 280 (100), 272 (5), 270 (16), 265 (12), 251 (32), 206 (30), 127 (5), 125 (15), 118 (17), 91 (8).  $\text{C}_{17}\text{H}_{16}\text{O}_2\text{N}_2\text{Cl}_2$  (351) Calcd. C 58.1 H 4.56 N 7.98 Found C 58.0 H 4.63 N 8.25.

#### 2-(4-Chlorophenyl)methyl-5-[1H]benzimidazole carboxylic acid ethyl ester HCl (**4a**)

**4a** was obtained from **4** as white crystals, m.p. 252°C, yield 44.4 %. -  $^1\text{H-NMR}$ : 1.40 (t,  $J = 7$  Hz, 3H,  $\text{CH}_2\text{CH}_3$ ), 4.40 (q,  $J = 7$  Hz, 2H,  $\text{CH}_2\text{CH}_3$ ), 4.6 (s, 2H,  $\text{CH}_2$ ), 7.45 and 7.6 (2d,  $J_{2',3'} = 12$  Hz,  $\text{A}_2\text{X}_2$ , 4H, H-2',3',5',6'), 7.8 (d,  $J_{7,6} = 8$  Hz, 1H, H-7), 8.1 (dd,  $J_{6,7} = 8$  Hz,  $J_{4,6} = 1.45$  Hz, 2H, H-4,6), 8.30 (s, 1H, H-7), 10.4 (broad s,  $\text{N}^+\text{-H}$ ). - MS (70 eV)  $m/z$ : 316 (33 %,  $\text{M}^+$ ), 314 (100,  $\text{M}^+$ ), 301 (3), 299 (10), 287 (5), 285 (15), 271 (17), 269 (52), 250 (9), 243 (4), 241 (11.5), 128 (10), 126 (30), 118 (10), 91 (9). -  $\text{C}_{17}\text{H}_{16}\text{O}_2\text{N}_2\text{Cl}_2$  (351) Calcd. C 58.1 H 4.56 N 7.98 Found C 57.9 H 4.63 N 8.20.

#### 2-(4-Methoxyphenyl)methyl-5-[1H]benzimidazole carboxylic acid ethyl ester HCl (**5a**)

**5a** was obtained from **5** as white needles, m.p. 218°C, yield 47.6 %. -  $^1\text{H-NMR}$ : 1.40 (t,  $J = 7$  Hz, 3H,  $\text{CH}_2\text{CH}_3$ ), 3.7 (s, 3H,  $\text{OCH}_3$ ), 4.40 (q,  $J = 7$

Hz, 2H,  $\text{CH}_2\text{CH}_3$ ), 4.50 (s, 2H,  $\text{CH}_2$ ), 6.9 and 7.45 (2d,  $J_{2',3'}$  and  $J_{5',6'} = 9$  Hz,  $\text{A}_2\text{X}_2$ , 4H, H-2',3',5',6'), 7.8 (d,  $J_{7,6} = 8$  Hz, 1H, H-7), 8.1 (dd,  $J_{6,7} = 8$  Hz,  $J_{4,6} = 1.45$  Hz, 2H, H-4,6), 8.30 (s, 1H,  $\text{N}^1\text{-H}$ ). - MS (70 eV)  $m/z$ : 310 (100 %,  $\text{M}^+$ ), 295 (29), 265 (29), 221 (8), 118 (10). -  $\text{C}_{18}\text{H}_{19}\text{O}_3\text{N}_2\text{Cl}$  (346.5) Calcd. C 62.34 H 5.48 N 8.08 Found C 62.2 H 5.90 N 8.15

#### 2-(4-Nitrophenyl)methyl-5-[1H]benzimidazole carboxylic acid ethyl ester HCl (**6a**)

**6a** was obtained from **6** as white crystals, m.p. 269°C, yield 36 %. -  $^1\text{H-NMR}$ : 1.40 (t,  $J = 7$  Hz, 3H,  $\text{CH}_2\text{CH}_3$ ), 4.40 (q,  $J = 7$  Hz, 2H,  $\text{CH}_2\text{CH}_3$ ), 4.70 (s, 2H,  $\text{CH}_2$ ), 7.25 (broad s,  $\text{N}^+\text{-H}$ ), 7.7 and 7.9 (2d,  $J_{2',3'}$  and  $J_{5',6'} = 3.5$  Hz,  $\text{A}_2\text{X}_2$ , 4H, H-2',3',5',6'), 8.05 (dd,  $J_{7,6} = 8$  Hz,  $J_{4,7} = 0.8$  Hz, 1H, H-7), 8.15-8.35 (m, 3H, H-4,6 and  $\text{N}^1\text{-H}$ ). - MS (70 eV)  $m/z$ : 325 (100 %,  $\text{M}^+$ ), 310 (12), 296 (13), 280 (78), 234 (12), 207 (10). -  $\text{C}_{17}\text{H}_{16}\text{O}_4\text{N}_3\text{Cl}$  (361.5) Calcd. C 56.4 H 4.43 N 11.6 Found C 56.1 H 4.38 N 11.3

#### 2-(Phenoxy)methyl-5-[1H]benzimidazole carboxylic acid ethyl ester HCl (**7a**)

**7a** was obtained from **7** and esterification of compound **9**, by using dry HCl and abs. EtOH, as white powder, m.p. 242°C, yield 44 %. -  $^1\text{H-NMR}$ : 1.40 (t,  $J = 7$  Hz, 3H,  $\text{CH}_2\text{CH}_3$ ), 4.40 (q,  $J = 7$  Hz, 2H,  $\text{CH}_2\text{CH}_3$ ), 5.70 (s, 2H,  $\text{CH}_2$ ), 6.9-7.5 (m, 5H, H-phenyl), 7.9 (d,  $J_{7,6} = 8$  Hz, 1H, H-7), 8.1 (dd,  $J_{6,7} = 8$  Hz,  $J_{4,6} = 1.45$  Hz, 2H, H-4,6), 8.30 (broad s,  $\text{N}^+\text{-H}$ ), 8.4 (s, 1H,  $\text{N}^1\text{-H}$ ). - MS (70 eV)  $m/z$ : 296 (7 %,  $\text{M}^+$ ), 203 (100), 189 (14), 174 (29), 158 (13), 131 (29), 65 (10). -  $\text{C}_{17}\text{H}_{17}\text{O}_3\text{N}_2\text{Cl}$  (332.5) Calcd. C 61.35 H 5.11 N 8.42 Found C 61.0 H 5.05 N 8.10.

#### 2-(4-Chlorophenoxy)methyl-5-[1H]benzimidazole carboxylic acid ethyl ester HCl (**8a**)

**8a** was obtained from **8** as white powder, m.p. 267°C, yield 41 %. -  $^1\text{H-NMR}$ : 1.40 (t,  $J = 7$  Hz, 3H,  $\text{CH}_2\text{CH}_3$ ), 4.40 (q,  $J = 7$  Hz, 2H,  $\text{CH}_2\text{CH}_3$ ), 5.70 (s, 2H,  $\text{CH}_2$ ), 6.25 (broad s,  $\text{N}^+\text{-H}$ ), 7.2 and 7.4 (2d,  $J_{2',3'}$  and  $J_{5',6'} = 10$  Hz,  $\text{A}_2\text{X}_2$ , 4H, H-2',3',5',6'), 7.8 (d,  $J_{7,6} = 8$  Hz, 1H, H-7), 8.1 (dd,  $J_{6,7} = 8$  Hz,  $J_{4,6} = 1.45$  Hz, 2H, H-4,6), 8.30 (s, 1H,  $\text{N}^1\text{-H}$ ). - MS (70 eV)  $m/z$ : 332 (3 %,  $\text{M}^+$ ), 330 (10,  $\text{M}^+$ ), 287 (1.5), 285 (5), 203 (100), 175 (41), 158 (11), 132 (30). -  $\text{C}_{17}\text{H}_{16}\text{O}_3\text{N}_2\text{Cl}_2$  (367). - Calcd. C 55.6 H 4.36 N 7.63 Found C 55.9 H 4.72 N 7.67.

#### 2-(Phenoxy)methyl-5-[1H]benzimidazole carboxylic acid (**9**)

Ethyl 3,4-diaminobenzoate (0.905 g, 5 mmol), phenoxyacetic acid (0.76 g, 5 mmol) and 4N HCl (25 ml) were heated under reflux for 4 h. After cooling, the mixture was adjusted to pH 7 with 10 % NaOH. The precipitated solid was collected, and washed with water. After drying, the crude product was purified on a column of silicagel (0.06-0.20 mm diameter) with  $\text{CHCl}_3$  : isopropanol (8:2), m.p. 135°C, Yield 19.8, IR  $\text{cm}^{-1}$ : 1690 (COOH),  $^1\text{H-NMR}$ : 5.40 (s, 2H,  $\text{CH}_2$ ), 6.9-7.5 (m, 5H, H-phenyl), 7.6 (d,  $J_{7,6} = 8$  Hz, 1H, H-7), 7.9 (dd,  $J_{6,7} = 8$  Hz,  $J_{4,6} = 1.43$  Hz, 2H, H-4,6), 8.30 (s, 1H,  $\text{N}^1\text{-H}$ ). - MS (70 eV)  $m/z$ : 268 (14 %,  $\text{M}^+$ ), 174 (100), 130 (10). -  $\text{C}_{15}\text{H}_{12}\text{O}_3\text{N}_2$  (268) Calcd. C 67.16 H 4.48 N 10.45 Found C 67.5 H 4.18 N 10.75

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