



STRUCTURE ACTIVITY RELATIONSHIPS OF TETRAHYDROCANNABINOL ANALOGUES ON HUMAN CANNABINOID RECEPTORS.

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Abstract: A series of Δ^8 -tetrahydrocannabinol (THC) and biphenylic derivatives were prepared and their binding affinity for both human cannabinoid receptors hCB₁ and hCB₂ evaluated.

Δ^9 -Tetrahydrocannabinol, an active component of marijuana, has been used for thousands of years in the treatment of a variety of ailments.¹ It possesses a wide range of physiological effects such as analgetic, appetite stimulant, anti-inflammatory, anti-convulsive, anti-emetic, immunosuppressive and intraocular pressure lowering.^{1,2} In spite of this therapeutic interest, its use has been limited by its psychotropic effects.² The search for analogues of medicinal value without the psychotropic effects has therefore received much attention.³

With the recent discovery of a second peripheral cannabinoid receptor,⁴ it has been proposed that the psychotropic effects of cannabinoids may be mediated by the receptor expressed in the brain (CB₁), while some of the other beneficial properties may be associated with the peripheral receptor (CB₂). Given the structural differences between CB₁ and CB₂, which have both been cloned and expressed,^{4,5} it should be possible to prepare selective hCB₂ ligands that might be valuable in the treatment of certain diseases. We now report our results on the SAR of some THC analogues.

All compounds⁶ were evaluated in binding studies using the displacement of radiolabelled [³H] (-) CP-55940⁷ to determine the ligand potencies on hCB₂ and hCB₁. The compounds in this study were Δ^8 -THC analogues containing a dimethylheptyl (DMH) chain, known to demonstrate increased potency⁸ relative to the pentyl chain of Δ^9 -THC in the classical CB₁ linked pharmacological assays as well as at the CB₁ receptor.

We paid special attention to the critical issue of the optical purity of cannabinoid ligands.⁹ This turns out to be very important with some extremely potent ligands. Indeed, if such a ligand is present as a contaminant in its enantiomer, the K_i value obtained will be erroneous and directly linked to the optical purity of the batch under study. We were able to synthesize starting material and products at >97% optical purity using Brown's optical enrichment procedure on (+) and (-)-pinene.¹⁰

Table 1 shows the effect of chirality on receptor binding affinity of two pairs of enantiomers of THC analogues prepared according to reported procedure from (+) and (-)-pinene.¹¹ In both cases, the (-) enantiomer exhibited stronger binding than the (+) enantiomer for both the hCB₁ and hCB₂ receptor by at least 45 fold, consistent with previous reports. Unfortunately, the desired selectivity for hCB₂ over hCB₁ was found to be poor in both the (-) and in the (+) series.

A phenolic oxygen and a lipophilic chain are believed to be necessary for classical cannabinoid activity.¹² Table 2 shows the effect of modifications at the C-1 position on the affinity (K_i) at both receptors while keeping the DMH chain unaltered. For instance, transformation of the phenol **1** to the methyl ether **5**¹³ resulted in a 40

fold loss in potency on hCB₂. On the other hand this loss in potency resulted in an increase in selectivity to 793 fold by virtue of the very weak binding on the hCB₁ receptor, as reported by Mechoulam *et al.*¹⁴ An important effect on hCB₁ was also observed when the methoxy group of **5** was replaced by a hydrogen atom to give **7**.¹⁵ The binding affinity increased from 15.9 μ M to 0.25 μ M, still 300 fold less potent than phenol **1**. Such a large effect on affinity caused by a small structural change may indicate that a region of the hCB₁ receptor was involved in interactions such as hydrogen bonding and/or steric effects with the ligand. Interestingly, the K_i of hCB₂ was not affected by this change. The phosphate **6**¹⁶ was not very active, particularly on hCB₁. A K_i value of 0.44 μ M was obtained on CB₂ and >20 μ M for CB₁. In this particular case, polarity and steric factors certainly played an important role. We concluded from the results of Table 2 that simple modification of the phenol to a methoxy group gave a less potent but selective CB₂ cannabinoid.

Table 1: Binding affinity of THC analogues in the natural (-) and the unnatural (+) series.

(-)-1 R= Me (+)-2
(-)-3 R= CHO (+)-4

	K _i (hCB ₂ , nM) ^a	K _i (hCB ₁ , nM) ^a	hCB ₁ /hCB ₂
1-	0.49 ± 0.18	0.83 ± 0.13	1.7
2-	28.8 ± 6.4	38.8 ± 6.1	1.3
3-	2.61 ± 0.36	2.24 ± 0.05	0.9
4-	118.6 ± 30.8	97.0 ± 6.1	0.8

a) Values are mean ± S.E.M. or individual determination.

Table 2: Binding affinity of THC analogues bearing various groups at C-1.

1- R= OH 5- R= OCH₃
6- R= OP(O)(OC₂H₅)₂ 7- R= H

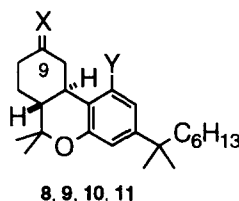
	K _i (hCB ₂ , nM) ^a	K _i (hCB ₁ , nM) ^a	hCB ₁ /hCB ₂
1-	0.49 ± 0.18	0.83 ± 0.13	1.7
5-	20.0 ± 12.4	15850 ± 2960	793
6-	441.7 ± 74.8	>20 000, >5000	>45
7-	20.8 ± 11.2	249.7 ± 31.0	12

a) Values are mean ± S.E.M. or individual determination.

Nabilone **8**^{12,17} and related compounds were prepared and the binding data are shown in Table 3. Nabilone differs structurally from Δ^9 -THC and **1** by the presence of a carbonyl group at C-9. Compound **8** was very potent, but not selective with a $K_i \approx 2$ nM for both receptors. In fact, **8** is very similar to **3** with respect to binding affinity on both receptors, in spite of the carbonyl being moved one carbon atom away. The methylene analogue **10**¹³ had similar potency for both receptors. On the other hand, transformation of the phenolic function of **10** to a methyl ether (**11**)¹³ resulted in a loss of potency on hCB₁, leading to a high degree of selectivity (>1000), of the same order of magnitude as observed in the case of **5**. Compounds **5** and **11** actually only differ by the position of the double bond on the A ring. We concluded from this, that the position of the double bond was not critical to activity at hCB₂ in this series.

The methylation of nabilone to the ether **9**¹⁷ resulted in a small increase in selectivity, but in a 72 fold loss in binding affinity for hCB₂ and a 284 fold loss on hCB₁. In the series of cannabinoids of Table 3, compound **11** stands out as a potent and selective hCB₂ binder.

Table 3: Binding affinity of THC analogues in the nabilone and exo-methylene series.



	X	Y	K_i (hCB ₂ , nM) ^a	K_i (hCB ₁ , nM) ^a	hCB ₁ /hCB ₂
8-	O	OH	1.84 ± 0.42	2.19 ± 0.89	1.2
9-	O	OCH ₃	132.2 ± 44.3	621 ± 215	4.7
10-	CH ₂	OH	0.58 ± 0.30	1.82 ± 0.11	3.1
11-	CH ₂	OCH ₃	19.4 ± 3.8	> 20 000, > 20 000	>1000

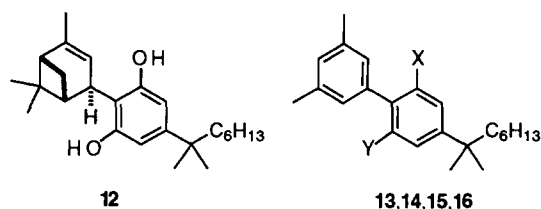
a) Values are mean \pm S.E.M. or individual determination.

The radiolabelled CP-55940 used to determine the ligand potencies, is devoid of the rigid tricyclic ring system¹⁸ common to the THC family, and we decided to test some non-tricyclic analogues. Compound **12**¹⁹ used to prepare the tricyclic ring system, was found to be particularly potent at hCB₂ with a K_i of 41 nM (9 fold selective, Table 4). Replacement of the terpenyl moiety of **12** by a planar template that better approximates the A ring, such as a 3,5-dimethylphenyl group, greatly simplifies the molecule as well as the synthesis of potential analogues. This change resulted in stronger binding of **13** *vis-à-vis* **12** to both receptors and the net result was an improved selectivity for hCB₂. In order to evaluate the effectiveness of the phenol as hydrogen bonding donor, **13** was converted to dimethyl ether **16**. This modification led to a significant loss of the binding affinity on hCB₂ (ca. 200 fold), with a K_i of 433nM. Considering the structural simplicity of **16**, such binding is still

noteworthy. For the hCB₁ receptor, this modification resulted in an inactive compound.

The selectivity was lost by replacement of one OH group by a hydrogen atom (**14**). Mainly due to improved CB₁ binding. Removal of both OH groups yielded hydrocarbon **15**, which was essentially inactive at both receptors.

Table 4: Binding affinity of bicyclic and biphenylic analogues.



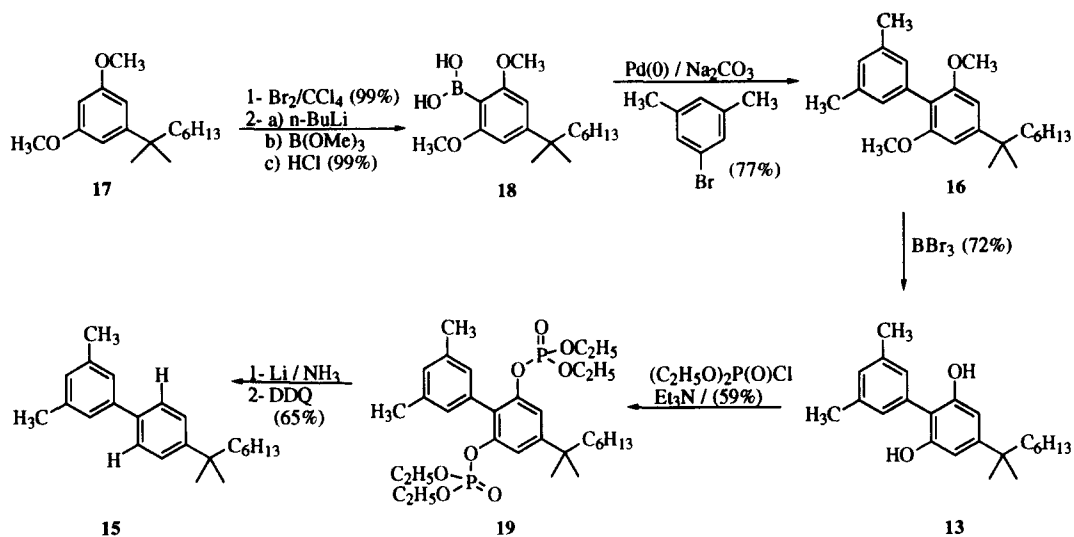
	X	Y	K _i (hCB ₂ , nM) ^a	K _i (hCB ₁ , nM) ^a	hCB ₁ /hCB ₂
12-	-	-	41.2 ± 14.1	350.3 ± 41.4	9.0
13-	OH	OH	2.00 ± 1.08	79.1 ± 12.5	40
14-	OH	H	4.14 ± 2.51	12.5 ± 1.8	3.0
15-	H	H	2138 ± 1317	> 30 000, > 30 000	>14
16-	OCH ₃	OCH ₃	433 ± 203	> 20 000, > 20 000	>46

a) Values are mean ± S.E.M. or individual determination.

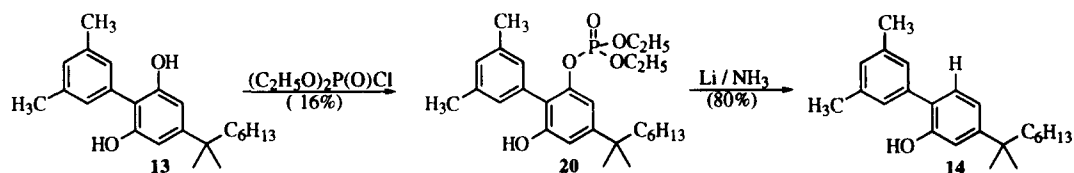
Compounds (**13-16**) were prepared according to Schemes 1 and 2. The bis-ether **17** was brominated²⁰ and converted to the boronic acid **18** via a metal-halogen exchange reaction with *n*-butyllithium. A Suzuki coupling reaction with 5-bromo-*m*-xylene gave **16**²¹ in an overall yield of 75%. Demethylation^{11a} of the bis-ether was performed with BBr₃ at 0° C to furnish **13** in 72% yield. Compound **15**^{11a} was obtained by the formation of diphosphonate **19** followed by reduction with Li/NH₃ and reoxidation of the aromatic ring with DDQ (because of partial overreduction) in 38% yield for the 3 steps. Finally, compound **14** was prepared in a manner similar to that used for the synthesis of **15**, by formation of monophosphonate **20** in 16% yield and reduction with Li/NH₃ in 80% yield.

In conclusion, we have shown that in the naturally occurring Δ⁸-THC stereochemical series, analogues such as **5** and **11** are very selective for the hCB₂ receptor regardless of the position of the exo or endo double bond. It was preferable not to have the free OH group at the C-1 position to obtain hCB₂ selective compounds. When the hydroxyl group (**1**) was replaced by a methoxy group (**5**), significant binding was retained at the hCB₂ receptor while the selectivity toward hCB₂ increased to 1000 fold. We have demonstrated, with non-cannabinoid type structures like **12-14** that the tricyclic moiety is not necessary to reach nM affinity, and that such compounds may have high selectivity.

Scheme 1



Scheme 2



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