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# Novel Asymmetric Desymmetrization of *meso-1*, 2-Diols *via* Diastereoselective β-Elimination of Chiral α-Arylsulfinyl Acetals

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Abstract: Enantioselective desymmetrization of *meso*-1,2-diols was accomplished *via* diastereoselective C—O bond fission of acetals derived from various optically active  $\beta$ -ketosulfoxides with high diastereoselectivities. The substituent at the  $\gamma$ -position of the  $\beta$ -ketosulfoxide plays an important role in this cleavage. Copyright © 1996 Elsevier Science Ltd

Differentiation of enantiotopic groups in achiral bifunctional molecules with  $\sigma$ -symmetry has been widely used as a strategy for creating optically active molecules.<sup>10</sup> Furthermore, each diastereomer produced in this reaction can often be converted into either enantiomer by an appropriate manipulation, thereby enabling the enantioconvergent synthesis of a desired product, the so-called *meso* trick.<sup>20</sup> Asymmetrization of  $\sigma$ -symmetrical diols has recently received considerable attention as a fascinating methodology for synthesizing useful chiral building blocks for various natural products, and great effort has been devoted to the development of efficient methods.<sup>3-7</sup> While the induction of chirality for  $\sigma$ -symmetrical diols by enzymatic reactions is well known,<sup>30</sup> the use of chemical methods is less well established.<sup>4-7</sup> Since the application of the enzymatic method is limited due to the high specificity of substrates, the development of efficient chemical methods is very important.

In a previous work, we developed a chemical asymmetric desymmetrization of prochiral 1,3-diols via acetalization with an intramolecular chiral  $\beta$ -ketosulfoxide moiety followed by diastereoselective acetal fission to afford chiral mono-protected 1,3-diols.<sup>46, 46+)</sup> Based on this result, we planned to investigate the potency of  $\beta$ -ketosulfoxides as a tool for differentiating enantiotopic groups in  $\sigma$ -symmetrical diols, as shown in Scheme 1.

In this paper, we report a novel asymmetric desymmetrization of *meso*-1,2-diols with chiral  $\beta$ -ketosulfoxide as a chiral auxiliary.





#### **RESULTS AND DISCUSSION**

Enantiomerically pure (R)- $\beta$ -ketosulfoxides 1-3<sup>8</sup>) were acetalized with *cis*-1,2-bis(trimethylsiloxy)cyclopentane or *cis*-1,2-bis(trimethylsiloxy)cyclohexane in the presence of trimethylsilyl trifluoromethanesulfonate (TMSOTf) catalyst<sup>9</sup>) to give a separable diastereomeric mixture of 4a-7a and 4b-7b (Scheme 2, Table 1).<sup>10,11</sup>



Scheme 2
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Table 1. Acetalization of β-Ketosulfoxides 1–3.							
Diol	Sulfoxide	R	Product	Yield (%) <sup>a)</sup>			
<u>(n)</u>				<u>a</u>	b		
1	1	Me	4	54	26		
1	2	Ph	5	0	74		
1	3	Bzl	6	24	· 48		
2	3	Bzl	7	24	48		

a) Isolated yield.

On treatment with one equivalent of lithium diisopropylamide (LDA), the acetal 4a gave the alcohols 8a-c in low yield along with a significant amount of 4a and 4b. This problem was overcome by increasing the

amount of base; i.e., the reaction was completed when six equivalents of base were used. Cleavage A gave a mixture of  $\alpha$ ,  $\beta$ - and  $\beta$ ,  $\gamma$ -unsaturated sulfoxides. Bond cleavage A to 8a and 8b predominates over B to 8c. The diastereometric isomer 4b showed better selectivity than 4a. Selectivity in 1,2-dimethoxyethane (DME) was better than that in tetrahydrofuran (THF) (Scheme 3, Table 2).



Scheme 3

Substrate	Conditions (equiv.)	Yield (%) _	Cleavage Ratio *)	
			Α	В
4a	LDA (6), THF, -78 °C	95	74	26
	LDA (6), DME , -78 °C	92	81	19
4b	LDA (6), THF, -78 °C	81	82	18
	LDA (6), DME, -78 °C	83	89	11

Table 2. Diastereoselective Acetal Cleavage of 4a and 4b.

a) The ratio was determined by 200 MHz <sup>1</sup>H-NMR spectroscopy.

The olefin geometry of **8b** was determined by the observation of an NOE between the vinylic proton and the  $\alpha$ -methine proton of vinylic ether on the cyclopentane ring.<sup>12)</sup> The absolute configuration of these alcohols **8a–c** was determined by Mosher's method<sup>13)</sup> after conversion into the  $\alpha$ -methoxy- $\alpha$ -(trifluoromethyl)phenylacetic acid (MTPA) ester **9**<sup>14)</sup> in 3 steps (MTPA esterification, hydrolysis of the vinylic ether, and acetylation), as shown in Scheme 4.



Reagents: a) (+)-MTPACI, Et<sub>3</sub>N, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C. b) (-)-MTPACI, Et<sub>3</sub>N, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C. c) IN HCl, acetone, room temp. d) Ac<sub>2</sub>O, pyridine, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C. Scheme 4

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On the other hand, acetal 5b (R=Ph) did not provide any cleaved products, but rather isomerized to 5a in 98% yield on treatment with six equivalents of LDA in THF at -78 °C. The resulting 5a did not react under the In contrast to the phenyl derivatives 5a and 5b, the benzyl derivatives 6a and 6b easily same conditions. afforded the chiral alcohols 10a and 10b, each as a mixture of (E/Z)-isomers under the same conditions. Like 4a and 4b, the diastereometric isomers 6a and 6b both preferentially gave the products arising from cleavage A. In addition, the selectivity was generally higher than that of 4a and 4b. In this case, three equivalents of base was enough to complete the reaction without decreasing the yield. Acetals 7a and 7b derived from cis-1,2cyclohexanediol also selectively gave alcohols 11a and 11b with S-configuration. Acetals 6b and 7b with an endo-benzyl group showed better selectivity than the corresponding diastereomeric isomers 6a and 7a, respectively. The selectivity of 7a was somewhat low compared with that of the corresponding cyclopentanediol derivative 6a. For both 6 and 7, fairly good selectivity was observed when N, N, N', N'. tetramethylethylenediamine (TMEDA) was added (Schemes 5 and 6, Table 3). The stereochemistry of the olefin in 10 and 11 was determined by an NOE experiments.<sup>12)</sup> The absolute configurations of the products were confirmed by their transformation into the known compounds.<sup>15,16)</sup>







Scheme 6

Substrate	Conditions (equiv.)	Yield (%)	Cleavage Ratio *)	
			Α	В
	LDA (3), THF,78 °C	92	86	14
	LDA (3), HMPA <sup>b)</sup> (3), THF, -78 °C	91	66	34
6a	LDA (3), TMEDA (3), THF, -78 °C	92	84	16
	LDA (3), DME, -78 °C	92	90	10
	LiTMP <sup>c)</sup> (3), THF, –78 °C	90	74	26
	LDA (3), THF, -78 °C	91	90	10
	LDA (3), HMPA (3), THF, -78 °C	90	86	14
6b	LDA (3), TMEDA (3), THF, -78 °C	92	94	6 <sup>d)</sup>
	LDA (3), DME, -78 °C	92	48	52
	LiTMP (3), THF, -78 °C	93	87	13
7a	LDA (5), THF, -78 °C	95	46	54
	LDA (5), TMEDA (5), THF, -78 °C	88	63	37
7b	LDA (5), THF, -78 °C	95	94	6 <sup>e)</sup>
	LDA (5), TMEDA (5), THF, -78 °C	90	92	8

Table 3. Diastereoselective Acetal Cleavage of 6 and 7.

a) The ratio was determined by 200 or 270 MHz <sup>1</sup>H-NMR spectroscopy.

b) HMPA = hexamethylphosphoric triamide. c) LiTMP = lithium 2,2,6,6-tetramethylpiperidide.

d) (Z)-10a : (E)-10a : (Z)-10b : (E)-10b = 88 : 6 : 2 : 4.

e) (Z)-11a : (E)-11a : (E)-11b : (Z)-11c : (E)-11c = 76 : 4 : 14 : 3 : 3.

The different reactivities of the  $\alpha$ -sulfingl acetals can be explained as follows (Scheme 7). As shown in Scheme 7(a), for 5, the initially formed alkoxide anion 12 would rapidly recyclize to a stable anion 13. On the other hand, for 4, 6 and 7, further deprotonation occurred in the anion 14 (R=H or Ph, n=1 or 2) bearing a hydrogen at the  $\gamma$ -position of the sulfinyl group to give a dianion 15, in which recyclization would be prevented by electronic repulsion between the anions, as shown in Scheme 7(b). Since the proton at the  $\gamma$ -position of the sulfinyl group in 6a and 6b is more acidic than that in 4a and 4b, less base is needed to generate dianion 15 from 6a and 6b, compared to that from 4a and 4b.



Scheme 7

The diastereoselectivity in this acetal fission can be explained as follows (Scheme 8). The *p*-tolyl group would be *anti* to a bulky bicyclic ring to minimize steric repulsion. For isomers with an *exo-p*-tolylsulfinylmethyl group 4b, 6b, and 7b, the conformer III would predominate over the conformer IV since the partially positive sulfur atom benefits from so-called attractive *gauche* interaction<sup>17</sup> with the partially negative acetal oxygens in III. In contrast, conformer IV is destabilized by an unfavorable *gauche* interaction between the sulfinyl group and the bulky substituent R. Consequently, cleavage A would be more favorable than cleavage B because of diastereoselective deprotonation of the *pro-S* proton (a proton *gauche* to the sulfinyl oxygen), <sup>16</sup> followed by *anti* elimination. For 4a, 6a, and 7a, the conformer I would be more favorable than II since the steric repulsion between the cyclopentane ring and the sulfinylmethyl group in II is greater than the *gauche* interaction between the sulfinyl group and R in I. As a result, cleavage A predominates over cleavage B. In the case of **6a** (R=Bzl, n=1), the steric repulsion between the cyclopentane ring in **7a** (R=Bzl, n=2). Therefore, the diastereoselectivity of **6a** is somewhat higher than that of **7a**.







In summary, we have developed a novel method for desymmetrizing  $\sigma$ -symmetrical diols *via* acetal formation with  $\beta$ -ketosulfoxides followed by base-promoted diastereoselective acetal cleavage. In these transformations,  $\beta$ -ketosulfoxides serve as a chiral auxiliary which gives rise to excellent diastereoselectivity. An important facet of this study is that both the diastereometric isomers give the same product preferentially, although acetalization gives a diasetereometric mixture. This transformation may be useful in the synthesis of natural products. Applications are under investigation and will be reported in due course.

#### **EXPERIMENTAL**

General. Melting points are uncorrected. Optical rotations were measured using a JASCO DIP-360 digital polarimeter. IR spectra were measured with a Hitachi 260-10 IR spectrometer or a Horiba FT-210 IR spectrometer. <sup>1</sup>H-NMR spectra were measured with a Varian VXR-200 spectrometer (200 MHz), a JEOL JNM-EX270 spectrometer (270 MHz), or a JEOL JNM-GX500 spectrometer (500 MHz). <sup>13</sup>C-NMR spectra were measured with a Varian VXR-200 spectrometer (500 MHz). <sup>13</sup>C-NMR spectra were measured with a Varian VXR-200 spectrometer (50.3 MHz) or a JEOL JNM-EX270 spectrometer (67.8 MHz). Mass spectra were taken with a Shimadzu QP-1000 mass spectrometer and a JEOL JMS-D300 mass spectrometer. HPLC analyses were performed using a Waters 6000A pump, A Waters  $\mu$ -PORASIL (3.9 mm× 30 cm)column, a Waters RCM 25×10 column and a Soma S-310 UV detector (at 254 nm). Unless otherwise stated, all reactions were performed with anhydrous solvents and the extract was dried over anhydrous MgSO<sub>4</sub> before evaporation. Merck Kieselgel 60 was used as an adsorbent for column chromatography. Merck Kieselgel 60 PF<sub>214</sub> was used for preparative TLC.

(*Rs*)-3-Phenyl-1-(*p*-tolylsulfinyl)propan-2-one (3). A solution of methyl *p*-tolyl sulfoxide (1.28 g, 8.30 mmol) in THF (20 ml) was added to a stirred solution [prepared from n-BuLi (1.6 M in hexane; 11.4 ml, 18.3 mmol) and diisopropylamine (2.44 ml, 17.4 mmol) in THF (30 ml)] at -78 °C and the whole was stirred at -78 °C for 30 min. Methyl phenylacetate (1.31 ml, 9.13 mmol) was added to the mixture and the whole was stirred at -78 °C for 30 min. The reaction was quenched with saturated NH<sub>4</sub>Cl. The organic layer was separated and the aqueous layer was extracted with AcOEt. The combined organic layers were washed with water and brine, and then dried. The solvent was evaporated and the residue was chromatographed on silica gel with AcOEt-hexane (1 : 2) to give 3 (1.99 g, 88 %) as colorless crystals, mp 107-108 °C (AcOEt-hexane).  $[\alpha]_{D}^{27}$  +161.45 (c = 1.14, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 2.42 (s, 3H, Ar-CH<sub>3</sub>), 3.72 (d, 1H, *J* = 15.8 Hz, CH<sub>2</sub>SO or CH<sub>2</sub>Ph), 3.85 (d, 1H, *J* = 15.7 Hz, CH<sub>2</sub>SO or CH<sub>2</sub>Ph), 7.12 (d, 2H, *J* = 7.3 Hz, Ar-H), 7.25-7.34 (m, 5H, Ar-H), 7.51 (d, 2H, *J* = 8.5 Hz, Ar-H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 21.4, 51.8, 66.8, 124.0, 127.3, 128.8, 129.5. 130.1, 132.4, 139.4, 142.2, 199.1. IR (KBr): 3020, 1720, 1610, 1510, 1045 cm<sup>-1</sup>. MS *m/z* (%): 272 (M\*, 100), 133 (100). *Anal.* Calcd for C<sub>16</sub>H<sub>16</sub>O<sub>2</sub>S: C, 70.56; H, 5.92; S, 11.77. Found : C, 70.24; H, 6.01; S, 11.70.

(*Rs*)-3-*exo*-Methyl-3-*endo*-(*p*-tolylsulfinyl)methyl-2, 4-dioxa-*cis*-bicyclo[3.3.0]octane and (*Rs*)-3-*endo*-Methyl-3-*exo*-(*p*-tolylsulfinyl)methyl-2, 4-dioxa-*cis*-bicyclo[3.3.0]octane (4a and 4b). TMSOTf (0.187 ml, 0.969 mmol) was added to a stirred solution of  $\beta$ -ketosulfoxide 1 (1.90 g, 9.69 mmol) and *cis*-1,2-bis(trimethylsiloxy)cyclopentane (2.87 g, 11.6 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (95 ml), and the whole was stirred at 25 °C for 12 h. After the addition of saturated NaHCO<sub>3</sub>, the whole was extracted with AcOEt. The extract was washed with water and brine, and then dried. The solvent was evaporated and the residue was chromatographed on silica gel with AcOEt-hexane (1 : 1) to afford 4a (1.46 g, 54 %) as a colorless oil and 4b (710 mg, 26 %) as colorless crystals. 4a:  $[\alpha]_D^{26} + 101.9$  (c = 1.00, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.43 (s, 3H, CCH<sub>3</sub>), 1.43–1.63 (m, 3H), 1.90–2.07 (m, 3H), 2.41 (s, 3H, Ar–CH<sub>3</sub>), 3.13 (d, 1H, *J* = 14.1 Hz, CH<sub>2</sub>CHO), 7.31 (d, 2H, *J* = 8.4 Hz, Ar–H), 7.58 (d, 2H, *J* = 8.4 Hz, Ar–H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 21.3, 22.7, 23.3, 33.1, 33.1, 68.8, 81.4, 81.6, 107.3, 124.0, 129.8, 141.0, 142.3. IR (CHCl<sub>3</sub>): 3000, 2940, 1040 cm<sup>-1</sup>. MS *m/z* (%): 280 (M<sup>+</sup>, 0.4), 127 (100). *Anal.* Calcd for C<sub>15</sub>H<sub>20</sub>O<sub>3</sub>S: C, 64.26 ; H, 7.19 ; S, 11.44. Found : C, 64.23 ; H, 7.25 ; S, 11.28. 4b: mp 107–108 °C (AcOEt-hexane). [ $\alpha]_D^{27}$  +132.3 (c = 1.05, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.37–1.46 (m, 2H), 1.55–1.62 (m, 1H), 1.64 (s, 3H, CC<u>H</u><sub>3</sub>), 1.68–1.79 (m, 1H), 1.90 (d, 1H, J = 6.0 Hz), 1.93 (d, 1H, J = 6.0 Hz), 2.42 (s, 3H, Ar–C<u>H</u><sub>3</sub>), 2.95 (d, 1H, J = 13.7 Hz, C<u>H</u><sub>2</sub>SO), 3.07 (d, 1H, J = 13.7 Hz, C<u>H</u><sub>2</sub>SO), 4.58 (t, 1H, J = 6.0 Hz, CH<sub>2</sub>C<u>H</u>O), 4.68 (d, 1H, J = 6.0 Hz, CH<sub>2</sub>C<u>H</u>O), 7.32 (d, 2H, J = 8.6 Hz, Ar–<u>H</u>), 7.55 (d, 2H, J = 8.6 Hz, Ar–<u>H</u>). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 21.4, 22.2, 25.2, 33.3, 33.4, 66.5, 81.7, 81.8, 107.4, 124.0, 130.0, 141.5, 141.7. IR (CHCl<sub>3</sub>): 3000, 2975, 1600, 1040 cm<sup>-1</sup>. MS *m/z* (%): 280 (M<sup>+</sup>, 0.2), 127 (100). *Anal.* Calcd for C<sub>15</sub>H<sub>20</sub>O<sub>3</sub>S: C, 64.26 ; H, 7.19 ; S, 11.44. Found : C, 63.94 ; H, 7.17 ; S, 11.24.

## (Rs)-3-endo-Phenyl-3-exo-(p-tolylsulfinyl)methyl-2,4-dioxa-cis-bicyclo[3.3.0]octane (5b).

TMSOTf (12 µl, 0.062 mmol) was added to a solution of  $\beta$ -ketosulfoxide 2 (159 mg, 0.615 mmol) and *cis*-1,2-bis(trimethylsiloxy)cyclopentane (182 mg, 0.739 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 ml) with stirring at 25 °C and the mixture was stirred at 25 °C for 36 h. After the addition of saturated NaHCO<sub>3</sub>, the whole was extracted with AcOEt. The extract was washed with water and brine, and then dried. The solvent was evaporated and the residue was chromatographed on silica gel with AcOEt-hexane (1 : 1) to give 5b (156 mg, 74 %) as colorless crystals, mp 82–83 °C (AcOEt-hexane). [ $\alpha$ ]<sub>D</sub><sup>30</sup> +91.9 (c = 1.13, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>) &: 1.23–1.35 (m, 1H), 1.38–1.54 (m, 3H), 1.84–1.96 (m, 2H), 2.38 (s, 3H, Ar–CH<sub>3</sub>), 3.23 (d, 1H, *J* = 14.1 Hz, CH<sub>2</sub>SO), 3.39 (d, 1H, *J* = 14.1 Hz, CH<sub>2</sub>SO), 4.94 (t, 1H, *J* = 5.6 Hz, CH<sub>2</sub>CHO), 5.04 (d, 1H, *J* = 5.6 Hz, CH<sub>2</sub>CHO), 7.26 (d, 2H, *J* = 7.7 Hz, Ar–H), 7.31–7.37 (m, 3H, Ar–H), 7.45 (d, 2H, *J* = 8.1 Hz, Ar–H), 7.54–7.58 (m, 2H, Ar–H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>) &: 21.2, 22.4, 33.2, 33.3, 70.5, 83.0, 83.3, 108.9, 123.8, 125.3, 128.1, 128.3, 129.8, 141.0, 141.4, 141.9. IR (CHCl<sub>3</sub>): 2976, 1600, 1495, 1040 cm<sup>-1</sup>. MS *m/z* (%): 342 (M<sup>\*</sup>, 0.2), 127 (100). *Anal.* Calcd for C<sub>20</sub>H<sub>12</sub>O<sub>3</sub>S: C, 70.15 ; H, 6.47 ; S, 9.36. Found : C, 69.95 ; H, 6.48 ; S, 9.11.

## (Rs)-3-exo-Phenyl-3-endo-(p-tolylsulfinyl)methyl-2,4-dioxa-cis-bicyclo[3.3.0]octane (5a).

A solution of **5b** (17.2 mg, 0.0503 mmol) in THF (1 ml) was added to a stirred LDA solution [prepared from *n*-BuLi (1.6 M in hexane; 0.189 ml, 0.302 mmol) and diisopropylamine (42 µl, 0.30 mmol) in THF (5 ml)] at -78 °C and the whole was stirred at -78 °C for 30 min. The reaction was quenched with saturated NH<sub>4</sub>Cl. The organic layer was separated and the aqueous layer was extracted with AcOEt. The combined organic layers were washed with brine, and then dried. The solvent was evaporated and the residue was chromatographed on silica gel with AcOEt-hexane (1 : 2) to give **5a** (16.9 mg, 98 %) as colorless crystals, mp 70-72 °C (AcOEthexane).  $[\alpha]_D^{10} + 106.1$  (c = 1.34, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.42-1.54 (m, 2H), 1.64-1.70 (m, 1H), 2.09-2.14 (m, 1H), 2.16-2.27 (m, 2H), 2.39 (s, 3H, Ar-CH<sub>3</sub>), 3.28 (d, 1H, *J* = 14.3 Hz, CH<sub>2</sub>SO), 3.35 (d, 1H, *J* = 14.3 Hz, CH<sub>2</sub>SO), 4.45-4.49 (m, 2H, CHOC), 7.27-7.36 (m, 5H, Ar-H), 7.49 (d, 2H, *J* = 7.3 Hz, Ar-H), 7.57 (d, 2H, *J* = 8.1 Hz, Ar-H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 21.2, 23.1, 32.9, 33.0, 69.7, 81.8, 81.8, 107.6, 123.8, 125.7, 128.4, 128.5, 129.7, 140.5, 140.8, 142.5. IR (CHCl<sub>3</sub>): 3020, 2980, 1603, 1501, 1042 cm<sup>-1</sup>. MS *m/z* (%): 342 (M<sup>+</sup>, 0.3), 189 (100). *Anal.* Calcd for C<sub>20</sub>H<sub>22</sub>O<sub>3</sub>S: C, 70.15 ; H, 6.47 ; S, 9.36. Found : C, 70.06 ; H, 6.46 ; S, 9.30.

(*Rs*)-3-*exo*-Benzyl-3-*endo*-(*p*-tolylsulfinyl)methyl-2,4-dioxa-*cis*-bicyclo[3.3.0]octane and (*Rs*)-3-*endo*-3-Benzyl-3-*exo*-(*p*-tolylsulfinyl)methyl-2,4-dioxa-*cis*-bicyclo[3.3.0]octane (6a and 6b). TMSOTF (46  $\mu$ l, 0.24 mmol) was added to a solution of  $\beta$ -ketosulfoxide 3 (650 mg, 2.39 mmol) and *cis*-1,2-bis(trimethylsiloxy)cyclopentane (708 mg, 2.87 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (70 ml) with stirring at 25 °C and the mixture was stirred at 25 °C for 60 h. After the addition of saturated NaHCO<sub>3</sub>, the whole was extracted with AcOEt. The extract was washed with water and brine, and then dried. The solvent was evaporated and the residue was chromatographed on silica gel with AcOEt–hexane (1 : 1) to afford 6a (373 mg, 24 %) as a colorless

oil and **6b** (746 mg, 48 %) as colorless crystals. **6a**:  $[\alpha]_{D}^{25}$  +90.2 (c = 1.12, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.42–1.53 (m, 2H), 1.54–1.64 (m, 1H), 1.90–2.02 (m, 2H), 2.04–2.10 (m, 1H), 2.39 (s, 3H, Ar-CH<sub>3</sub>), 3.01 (d, 1H, *J* = 13.7 Hz, CH<sub>2</sub>SO), 3.02 (s, 2H, Ar-CH<sub>2</sub>), 3.17 (d, 1H, *J* = 13.7 Hz, CH<sub>2</sub>SO), 4.63 (t, 1H, *J* = 5.2 Hz, CHOC), 4.65 (t, 1H, *J* = 5.2Hz, CHOC), 7.20–7.30 (m, 7H, Ar–H), 7.50 (d, 2H, *J* = 8.6 Hz, Ar–H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 21.4, 21.9, 33.2, 33.2, 44.0, 65.0, 81.8, 81.9, 108.3, 124.0, 126.6, 127.7, 129.9, 131.3, 135.1, 141.4, 141.6. IR (CHCl<sub>3</sub>): 2955, 2925, 1605, 1498, 1033 cm<sup>-1</sup>. MS *m/z* (%): 356 (M<sup>+</sup>, 0.6), 264 (100). *Anal.* Calcd for C<sub>21</sub>H<sub>24</sub>O<sub>3</sub>S: C, 70.76 ; H, 6.79 ; S, 8.99. Found : C, 70.46 ; H, 6.83 ; S, 8.75. **6b**: mp 118–119 °C (AcOEt–hexane).  $[\alpha]_{D}^{22}$  +109.3 (c = 1.00, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.91–1.05 (m, 1H), 1.25–1.38 (m, 3H), 1.72–1.81 (m, 2H), 2.40 (s, 3H, Ar–CH<sub>3</sub>), 2.89 (d, 1H, *J* = 14.1 Hz, CH<sub>2</sub>SO), 3.08 (d, 1H, *J* = 14.1 Hz, CH<sub>2</sub>SO), 3.19 (d, 1H, *J* = 14.1 Hz, Ar–CH<sub>3</sub>), 3.24 (d, 1H, *J* = 14.1 Hz, Ar–CH<sub>2</sub>), 4.59 (t. 1H, *J* = 5.3 Hz, CHOC), 7.20–7.35 (m, 7H, Ar–H), 7.49 (d, 2H, *J* = 8.1 Hz, Ar–H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 21.3, 22.9, 33.3, 33.3, 43.2, 66.9, 82.4, 82.5, 109.5, 123.9, 126.8, 128.2, 129.8, 130.4, 135.7, 140.9, 142.3. IR (CHCl<sub>3</sub>): 2920, 1600, 1025 cm<sup>-1</sup>. MS *m/z* (%): 356 (M<sup>+</sup>, 0.3), 264 (100). *Anal.* Calcd for C<sub>21</sub>H<sub>24</sub>O<sub>3</sub>S: C, 70.76 ; H, 6.79 ; S, 8.99. Found : C, 70.70 ; H, 6.77 ; S, 8.94.

(Rs)-8-exo-Benzyl-8-endo-(p-tolylsulfinyl)methyl-7,9-dioxa-cis-bicyclo[4.3.0]nonane and (Rs)-8-endo-Benzyl-8-exo-(p-tolylsulfinyl)methyl-7,9-dioxa-cis-bicyclo[4.3.0]nonane (7a and 7b). TMSOTf (14  $\mu$ l, 0.073 mmol) was added to a solution of  $\beta$ -ketosulfoxide 3 (200 mg, 0.734 mmol) and cis-1,2-bis(trimethylsiloxy)cyclohexane (500 mg, 1.92 mmol) in CH<sub>2</sub>Cl, (30 ml) with stirring at 25 °C and the mixture was stirred at 25 °C for 140 h. After the addition of saturated NaHCO<sub>1</sub>, the whole was extracted with AcOEt. The extract was washed with water and brine, and then dried. The solvent was evaporated and the residue was chromatographed on silica gel with AcOEt-hexane (2 : 5) to afford 7a (65.0 mg, 24 %) as a colorless oil and 7b (130 mg, 48 %) as colorless crystals. 7a:  $[\alpha]_{D}^{29}$  +71.1 (c = 0.50, CHCl<sub>1</sub>). <sup>1</sup>H-NMR (CDCl<sub>1</sub>) δ: 1.21-1.33 (m, 2H), 1.48-1.60 (m, 2H), 1.70-1.84 (m, 4H), 2.40 (s, 3H, Ar-CH<sub>1</sub>), 2.99 (d, 1H, J = 13.7 Hz, C<u>H</u>,SO), 3.14 (s, 2H, Ar-C<u>H</u>,), 3.26 (d, 1H, J = 13.7 Hz, C<u>H</u>,SO), 4.06–4.12 (m, 2H, C<u>H</u>OC), 7.20–7.33 (m, 5H, Ar-<u>H</u>), 7.32 (d, 2H, J = 7.7 Hz, Ar-<u>H</u>), 7.51 (d, 2H, J = 7.7 Hz, Ar-<u>H</u>). <sup>13</sup>C-NMR (CDCl<sub>3</sub>) δ: 20.3, 20.3, 21.4, 27.9, 28.0, 44.5, 67.9, 74.6, 74.7, 107.5, 124.1, 126.7, 128.2, 129.8, 130.9, 135.9, 141.2, 142.1. IR (KBr): 2925, 2860, 1730, 1495, 1454, 1261, 1228, 1128, 1090, 1003 cm<sup>-1</sup>. MS *m/z* (%): 370 (M<sup>+</sup>, 11.4), 279 (100). High MS Calcd for C<sub>22</sub>H<sub>26</sub>O<sub>3</sub>S-O: 354.1651. Found: 354.1650. 7b: mp 125–126 °C (AcOEt).  $[\alpha]_{D}^{25}$  +104.3 (c = 0.51, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.20–1.28 (m, 2H), 1.40– 1.47 (m, 2H), 1.54–1.61 (m, 2H), 1.64–1.72 (m, 2H), 2.39 (s, 3H, Ar-CH<sub>3</sub>), 2.91 (d, 1H, J = 13.7 Hz,  $CH_{3}SO$ , 3.11 (s, 2H, Ar- $CH_{2}$ ), 3.28 (d, 1H, J = 13.7 Hz,  $CH_{3}SO$ ), 4.15 (dd, 1H, J = 6.0, 5.2 Hz, CHOC), 4.19 (dd, 1H, J = 6.0, 5.2 Hz, CHOC), 7.22–7.31 (m, 5H, Ar–H), 7.38 (d, 2H, J = 6.8 Hz, Ar–H), 7.45 (d, 2H, J = 7.7 Hz, Ar-<u>H</u>). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 20.4, 20.5, ,21.4, 27.7, 27.9, 46.0, 65.7, 74.3, 74.3, 107.0, 124.0, 126.8, 128.0, 129.8, 131.4, 135.5, 141.2, 141.9. IR (KBr): 2927, 2854, 1733, 1495, 1456, 1259, 1093, 1039 cm<sup>-1</sup>. MS m/z (%): 370 (M<sup>+</sup>, 16.2), 217 (100). Anal. Calcd for C<sub>2</sub>, H<sub>26</sub>O<sub>3</sub>S: C, 71.32; H, 7.07; S, 8.66. Found : C, 71.10 ; H, 7.04 ; S, 8.61.

(*Rs*)-2-[(1*R*, 2*S*)-2-Hydroxycyclopentyloxy]-3-(*p*-tolylsulfinyl)propene (8a), (*E*, *Rs*)-2-[(1*R*, 2*S*)-2-Hydroxycyclopentyloxy]-1-(*p*-tolylsulfinyl)propene (8b), and (*Rs*)-2-[(1*S*, 2*R*)-2-Hydroxycyclopentyloxy]-3-(*p*-tolylsulfinyl)propene (8c). Representative Procedure in Table 2: A solution of 4a (52.1 mg, 0.186 mmol) in DME (1 ml) was added to a stirred LDA solution [prepared from *n*-BuLi (1.6 M in hexane; 0.697 ml, 1.12 mmol) and diisopropylamine (0.156 ml, 1.12 mmol) in DME (4 ml)] at -78 °C and the mixture was stirred at -78 °C for 30 min. The reaction was quenched with saturated NH<sub>4</sub>Cl.

The organic layer was separated and the aqueous layer was extracted with AcOEt. The extract was washed with brine, and then dried. The solvent was evaporated and the residue was purified by preparative TLC with AcOEt-hexane (1:1) to afford 8a (31.4 mg, 60 %), 8b (7.6 mg, 15 %) and 8c (8.9 mg, 17 %) each as a colorless oil. 8a:  $[\alpha]_{D}^{30}$  +169.5 (c = 1.09, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.47-1.62 (m, 1H), 1.70-1.93 (m, 13.3 Hz, CH,SO), 3.81 (d, 1H, J = 2.6 Hz, C=CH,), 4.16 (d, 1H, J = 2.6 Hz, C=CH,), 4.18–4.26 (m, 1H, CHOH), 4.18-4.26 (m, 1H, CHOC), 7.32 (d, 2H, J = 8.6 Hz, Ar-H), 7.49 (d, 2H, J = 8.6 Hz, Ar-H). <sup>13</sup>C-NMR (CDCl,) 5: 19.3, 21.3, 27.5, 30.6, 61.5, 71.1, 79.6, 89.6, 124.2, 129.7, 139.3, 141.5, 150.5. IR (CHCl<sub>1</sub>): 3200-3400, 3000, 1630, 1290, 1035 cm<sup>-1</sup>. MS m/z (%): 280 (M<sup>+</sup>, 0.2), 140 (100). Anal. Calcd for  $C_{1,4}H_{20}O_{3}S \cdot 1/3H_{20}C$ , 62.91 ; H, 7,27 ; S, 11.19. Found : C, 62.94 ; H, 7.20 ; S, 11.26. 8b:  $[\alpha]_{0}^{29}$ +58.2 (c = 0.85, CHCl<sub>1</sub>). <sup>1</sup>H-NMR (CDCl<sub>1</sub>)  $\delta$ : 1.48 (m, 6H), 2.33 (s, 3H, CC<u>H<sub>1</sub></u>), 2.41 (s, 3H, Ar-C<u>H<sub>1</sub></u>), 4.02-4.13 (m, 1H, CHOH), 4.24-4.28 (m, 1H, CHOC), 5.44 (s, 1H, C=CH), 7.30 (d, 2H, J = 8.6 Hz, Ar-H), 7.48 (d, 2H, J = 8.6 Hz, Ar-H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 18.3, 19.6, 21.3, 27.7, 30.8, 72.7, 80.0, 109.2, 124.0, 129.8, 140.6, 142.6, 165.1. IR (CHCl<sub>3</sub>): 3150–3550, 2990, 1600, 1385, 1305, 1085, 1026 cm<sup>-1</sup>. MS m/z (%): 280 (M<sup>+</sup>, 0.5), 127 (100). High MS Calcd for C<sub>15</sub>H<sub>20</sub>O<sub>3</sub>S: 280.1133. Found: 280.1133. 8c:  $[\alpha]_{0}^{30}$  +201.9 (c = 1.20, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.47–1.57 (m, 1H), 1.67–1.92 (m, 5H), 2.42 (s, 3H, Ar-CH<sub>1</sub>), 2.80 (bs, 1H, OH), 3.48 (d, 1H, J = 12.8 Hz, CH<sub>2</sub>SO), 3.56 (d, 1H, J = 12.8 Hz, CH<sub>2</sub>SO), 4.12-4.17 (m, 1H, C<u>H</u>OH), 4.20 (d, 1H, J = 3.1 Hz,  $C = CH_2$ ), 4.22–4.27 (m, 1H, C<u>H</u>OC), 4.24 (d, 1H, J = 3.1 Hz, C=C $\underline{H}_{2}$ ), 7.32 (d, 2H, J = 8.2 Hz, Ar– $\underline{H}$ ), 7.55 (d, 2H, J = 8.2 Hz, Ar– $\underline{H}$ ). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 19.5, 21.4, 27.6, 30.9, 63.0, 72.0, 79.6, 89.5, 124.2, 129.8, 140.5, 141.7, 151.5. IR (CHCl<sub>3</sub>): 3200-3600, 2950, 1630, 1290, 1039 cm<sup>-1</sup>. MS m/z (%): 280 (M<sup>+</sup>, 0.5), 140 (100). Anal. Calcd for C<sub>15</sub>H<sub>20</sub>O<sub>3</sub>S · 1/4H<sub>2</sub>O: C, 62.91 ; H, 7,27; S, 11.19. Found : C, 62.94; H, 7.20; S, 11.26.

(1S,2R)-2-Acetoxycyclopentyl (R)- $\alpha$ -Methoxy- $\alpha$ -(trifluoromethyl)phenylacetate (9a). From 8a: Triethylamine (64 µl, 0.46 mmol) and dimethylaminopyridine (DMAP) (18.7 mg, 0.153 mmol) were added to a solution of 8a (85.6 mg, 0.306 mmol) in CH,Cl, (4 ml) at 0 °C and the whole was stirred at 0 °C for 10 min. (+)-MTPACl (86 µl, 0.46 mmol) was added to the stirred mixture at 0 ℃ and the mixture was stirred at 0 ℃ for 30 min. After the addition of saturated NaHCO<sub>3</sub>, the whole was extracted with CHCl<sub>3</sub>. The extract was washed with water and brine, and then dried. The solvent was evaporated and the residue was chromatographed on silica gel with AcOEt-hexane (1:2) to give (+)-MTPA ester (147 mg, 97 %) as a colorless oil.  $[\alpha]_{n}^{24}$  +78.4 (c = 1.53, CHCl<sub>1</sub>). <sup>1</sup>H-NMR (CDCl<sub>1</sub>)  $\delta$ : 1.50–1.68 (m, 2H), 1.76–2.03 (m, 4H), 2.41 (s, 3H, Ar-CH<sub>3</sub>), 3.23 (d, 1H, J = 12.8 Hz, CH<sub>3</sub>SO), 3.34 (d, 1H, J = 12.8 Hz, CH<sub>3</sub>SO), 3.55 (s, 3H, OCH<sub>3</sub>), 4.00 (d, 1H, J = 2.6 Hz,  $C = CH_2$ ), 4.05 (d, 1H, J = 2.6 Hz,  $C = CH_2$ ), 4.38 (dd, 1H, J = 11.1, 6.0 Hz, CHOC), 5.32 (dd, 1H, J = 11.1, 6.0 Hz, CHOC), 7.25-7.30 (m, 2H, Ar-H), 7.37-7.44 (m, 5H, Ar-H), 7.52-7.56 (m, 2H, Ar-H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>) δ: 19.1, 21.4, 27.5, 28.1, 55.3, 63.1, 76.4, 76.8, 89.1, 121.1, 124.3, 125.4, 127.5, 128.3, 129.5, 129.6, 132.0, 140.2, 141.5, 151.0, 165.9. IR (CHCl<sub>3</sub>): 3000, 2950, 1760, 1645, 1305, 1185, 1033 cm<sup>-1</sup>. MS m/z (%): 496 (M<sup>+</sup>, 35.9), 189 (100). Anal. Calcd for C<sub>2</sub>, H<sub>22</sub>F<sub>3</sub>O<sub>5</sub>S: C, 60.47 ; H, 5.48 Found : C, 60.20; H, 5.63. IN HCl (0.03 ml) was added to a solution of the (+)-MTPA ester of 8a (78.2 mg, 0.158 mmol) in acetone (5 ml) at 25 °C and the mixture was stirred at 25 °C for 30 min. After the addition of saturated NaHCO<sub>1</sub>, acetone was evaporated and the residue was extracted with AcOEt. The extract was washed with brine and dried. The solvent was evaporated and the resulting alcohol was used in the next step without further purification. Acetic anhydride (45 µl, 0.47 mmol) was added to a solution of the crude alcohol in pyridine (2ml) at 25 °C and the mixture was stirred at 25 °C for 10 h. After the addition of 1N HCl, the whole was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The extract was washed with brine and dried. The solvent was evaporated and the

residue was purified by preparative TLC with AcOEt-hexane (1 : 5) to give **9a**<sup>14</sup> (50.1 mg, 88 %) as a colorless oil.  $[\alpha]_{D}^{24}$  +42.8 (c = 1.27, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.61–1.70 (m, 1H), 1.71–1.80 (m, 1H), 1.84– 1.94 (m, 2H), 1.90 (s, 3H, OCOCH<sub>3</sub>), 1.99–2.10 (m, 2H), 3.55 (s, 3H, OCH<sub>3</sub>), 5.13 (td, 1H, *J* = 7.7, 4.3 Hz, CHOAc), 5.44 (dd, 1H, *J* = 9.4, 4.3 Hz, CHO(+)-MTPA), 7.38–7.42 (m, 3H, Ar–H), 7.53–7.56 (m, 2H, Ar– H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 19.0, 20.7, 27.7, 28.5, 55.3, 74.1, 76.3, 121.2, 125.4, 127.4, 128.4, 129.6, 132.2, 165.7, 170.4. IR (CHCl<sub>3</sub>): 2990, 1745, 1605, 1172, 1045 cm<sup>-1</sup>. MS *m/z* (%): 360 (M<sup>+</sup>, 0.6), 189 (100). High MS Calcd for C<sub>1,2</sub>H<sub>19</sub>F<sub>3</sub>O<sub>5</sub>: 360.1182. Found: 360.1179.

**From 8b:** By a similar procedure to that described for (+)-MTPA esterification of **8a**, **8b** (16.6 mg, 0.0593 mmol) was converted into the (+)-MTPA ester (27.0 mg, 92 %) as colorless crystals, mp 95–96°C (AcOEthexane).  $[\alpha]_D^{25}$  +6.7 (c = 1.00, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.56–1.71 (m, 2H), 1.81–2.06 (m, 4H), 2.05 (s, 3H, CCH<sub>3</sub>), 2.40 (s, 3H, Ar–CH<sub>3</sub>), 3.56 (s, 3H, OCH<sub>3</sub>), 4.53 (dd, 1H, J = 10.3, 6.0 Hz, CHOC), 5.36 (s, 1H, C=CH), 5.34–5.39 (m, 1H, CHOC), 7.30 (d, 2H, J = 8.6 Hz, Ar–H), 7.41–7.44 (m, 3H, Ar–H), 7.47 (d, 2H, J = 8.6 Hz, Ar–H), 7.53–7.56 (m, 2H, Ar–H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 17.8, 19.1, 21.2, 27.6, 28.0, 55.3, 76.0, 77.2, 108.8, 121.0, 123.9, 125.3, 127.3, 128.3, 129.6, 129.7, 132.0, 140.4, 142.7, 164.7, 165.6. IR (CHCl<sub>3</sub>): 3000, 1750, 1600, 1100–1300 cm<sup>-1</sup>. MS m/z (%): 496 (M<sup>+</sup>, 5.7), 480 (100). Anal. Calcd for C<sub>25</sub>H<sub>27</sub>F<sub>3</sub>O<sub>5</sub>S: C, 60.47 ; H, 5.48. Found : C, 60.37 ; H, 5.48. By a similar procedure to that described for the preparation of **9a**, the (+)-MTPA ester of **8b** (109 mg, 0.220 mmol) was converted into **9a**<sup>140</sup> (71.3 mg, 90 %) as a colorless oil.

(1S,2R)-2-Acetoxycyclopentyl (S)- $\alpha$ -Methoxy- $\alpha$ -(trifluoromethyl)phenylacetate (9b). By a similar procedure to that described for (+)-MTPA esterification of 8a, 8a (51.0 mg, 0.182 mmol) was converted into the (-)-MTPA ester (84.9 mg, 94 %) as a colorless oil.  $[\alpha]_D^{28}$  +42.6 (c = 1.50, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 1.52–1.71 (m, 2H), 1.74–1.86 (m, 2H), 1.88–2.02 (m, 2H), 2.40 (s, 3H, Ar–CH<sub>3</sub>), 3.30 (d, 1H, J = 12.8 Hz, CH,SO), 3.51 (d, 1H, J = 12.8 Hz, CH,SO), 3.58 (s, 3H, OCH,), 4.09 (d, 1H, J = 2.6 Hz,  $C=CH_2$ , 4.12 (d, 1H, J = 2.6 Hz,  $C=CH_2$ ), 4.36–4.40 (m, 1H, CHOC), 5.34 (dd, 1H, J = 9.4, 4.3 Hz, CHOC), 7.26-7.30 (m, 2H, Ar-H), 7.38-7.44 (m, 5H, Ar-H), 7.56-7.60 (m, 2H, Ar-H). <sup>13</sup>C-NMR  $(CDCl_1)$   $\delta$ : 19.0, 21.4, 27.3, 28.1, 55.4, 63.5, 76.5, 77.4, 89.4, 121.2, 124.3, 125.4, 127.5, 128.3, 129.6, 129.6, 132.2, 140.2, 141.6, 151.2, 165.8. IR (CHCl<sub>3</sub>): 3000, 1750, 1639, 1295, 1179, 1042 cm<sup>-1</sup>. MS m/z (%): 496 (M<sup>+</sup>, 4.7), 189 (100). Anal. Calcd for  $C_{25}H_{27}F_3O_5S$ : C, 60.47; H, 5.48. Found : C, 60.48; H, 5.55. By a similar procedure to that described for the preparation of 9a, the (-)-MTPA ester of 8a (81.1 mg, 0.164 mmol) was converted into 9b (53.0 mg, 90 %) as a colorless oil.  $[\alpha]_{D}^{22}$  -9.0 (c = 1.82, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>) & 1.60-1.68 (m, 1H), 1.74-1.91 (m, 3H), 1.96 (s, 3H, OCOCH<sub>3</sub>), 1.97-2.09 (m, 2H), 3.58 (s, 3H, OCH<sub>3</sub>), 5.17 (td, 1H, J = 7.7, 4.3 Hz, CHOAc), 5.44 (dd, 1H, J = 8.6, 4.3 Hz, CHO(+)-MTPA), 7.38-7.42 (m, 3H, Ar-<u>H</u>), 7.56-7.59 (m, 2H, Ar-<u>H</u>). <sup>13</sup>C-NMR (CDCl<sub>3</sub>) δ: 19.0, 20.7, 27.6, 28.4, 55.3, 74.2, 76.5, 121.2, 125.4, 127.4, 128.3, 129.6, 132.2, 165.8, 170.3. IR (CHCl<sub>3</sub>): 2990, 2955, 1745, 1169, 1045 cm<sup>-1</sup>. MS m/z (%): 360 (M<sup>+</sup>, 0.6), 189 (100). High MS Calcd for C<sub>1.7</sub>H<sub>1.9</sub>F<sub>1.0</sub>c; 360.1185. Found: 360.1185.

### (1R, 2S)-2-Acetoxycyclopentyl (S)-a-Methoxy-a-(trifluoromethyl)phenylacetate (ent.-9a).

By a similar procedure to that described for (+)-MTPA esterification of **8a**, **8c** (62.7 mg, 0.126 mmol) was converted into the (-)-MTPA ester (101 mg, 91 %) as a colorless oil.  $[\alpha]_{D}^{26}$  +92.2 (c = 1.25, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.58–1.67 (m, 1H), 1.78–2.08 (m, 5H), 2.41 (s, 3H, Ar–CH<sub>3</sub>), 3.04 (d, 1H, *J* = 12.8 Hz, CH<sub>2</sub>SO), 3.34 (d, 1H, *J* = 12.8 Hz, CH<sub>2</sub>SO), 3.55 (s, 3H, OCH<sub>3</sub>), 4.14 (d, 1H, *J* = 2.6 Hz, C=CH<sub>2</sub>), 4.16 (d, 1H, *J* = 2.6 Hz, C=CH<sub>2</sub>), 4.47 (dd, 1H, *J* = 10.3, 6.0 Hz, CHOC), 5.33 (dd, 1H, *J* = 10.3, 6.0 Hz, CHOC), 7.26–7.28 (m, 2H, Ar–H), 7.34–7.39 (m, 3H, Ar–H), 7.41–7.43 (m, 2H, Ar–H), 7.54–7.56 (m, 2H, Ar–H).

<sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 19.0, 21.3, ,27.7, 28.3, 55.3, 64.3, 76.4, 76.7, 89.5, 121.1, 124.1, 125.3, 127.5, 128.3 129.5, 129.7, 132.1, 140.7, 141.5, 151.0, 165.9. IR (CHCl<sub>3</sub>): 3002, 1750, 1639, 1299, 1178, 1050 cm<sup>-1</sup>. MS *m/z* (%): 496 (M<sup>+</sup>, 7.5), 189 (100). *Anal.* Calcd for C<sub>25</sub>H<sub>27</sub>F<sub>3</sub>O<sub>5</sub>S · 1/4H<sub>2</sub>O: C, 59.93 ; H, 5.53. Found : C, 60.00 ; H, 5.57. By a similar procedure to that described for the preparation of **9a**, the (-)-MTPA ester of **8c** (155 mg, 0.312 mmol) was converted into **ent.-9a** (99 mg, 88 %) as a colorless oil.  $[\alpha]_D^{24}$  -46.0 (c = 1.11 CHCl<sub>3</sub>). The spectroscopic data (<sup>1</sup>H-NMR, <sup>13</sup>C-NMR, IR, and MS) was consistent with data for **9a**.

(1*R*, 2*S*)-2-Acetoxycyclopentyl (*R*)-α-Methoxy-α-(trifluoromethyl)phenylacetate (ent.-9b). By a similar procedure to that described for (+)-MTPA esterification of 8a, 8c (35.4 mg, 0.126 mmol) was converted into the (+)-MTPA ester (59.0 mg, 94%) as a colorless oil.  $[α]_D^{25}$  +117.6 (c = 1.65, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 1.56–1.66 (m, 1H), 1.77–1.88 (m, 3H), 1.92–2.05 (m, 2H), 2.41 (s, 3H, Ar–CH<sub>3</sub>), 3.24 (d, 1H, *J* = 12.8 Hz, CH<sub>2</sub>SO), 3.46 (d, 1H, *J* = 12.8 Hz, CH<sub>2</sub>SO), 3.58 (s, 3H, OCH<sub>3</sub>), 4.20 (s, 1H, *J* = 3.5 Hz, C=CH<sub>2</sub>), 4.44–4.48 (m, 1H, CH<sub>2</sub>CHO), 5.34–5.38 (m, 1H, CH2CHO), 7.29 (d, 2H, *J* = 8.6 Hz, Ar–H), 7.36–7.40 (m, 3H, Ar–H), 7.47 (d, 2H, *J* = 8.6 Hz, Ar–H), 7.55–7.58 (m, 2H, Ar–H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>) δ: 19.0, 21.4, 27.5, 28.3, 55.4, 64.3, 76.5, 77.3, 89.8, 121.2, 124.1, 125.4, 127.5, 128.3, 129.6, 129.8, 132.1, 140.6, 141.6, 151.1, 165.9. IR (CHCl<sub>3</sub>): 2980, 1740, 1625, 1285, 1162, 1038 cm<sup>-1</sup>. MS *m/z* (%): 496(M<sup>+</sup>, 7.5), 189 (100). *Anal.* Calcd for C<sub>25</sub>H<sub>27</sub>F<sub>3</sub>O<sub>5</sub>S · 1/4H<sub>2</sub>O: C, 59.93 ; H, 5.53. Found : C, 59.90 ; H, 5.53. By a similar procedure to that described for the preparation of the 9a, the (+)-MTPA ester of 8c (139 mg, 0.281 mmol) was converted into ent.-9b <sup>14</sup>) (91 mg, 90%) as a colorless oil. [α]<sub>D</sub><sup>23</sup> +8.9 (c = 1.07, CHCl<sub>3</sub>). The spectroscopic data (<sup>1</sup>H-NMR, <sup>13</sup>C-NMR, IR, and MS) was consistent with data for 9b.

(1R, 2S)- and (1S, 2R)-2-[(R)- $\alpha$ -Methoxy- $\alpha$ -(trifluoromethyl)phenylacetoxy]cyclopentyl (Rs)-(p-Tolylsulfinyl)acetate (16a and 16b). Representative Procedure in Table 3: TMEDA (0.382 ml, 2.53 mmol) was added to a stirred LDA solution [prerared from n-BuLi (1.6 M in hexane; 1.58 ml, 2.53 mmol) and diisopropylamine (0.354 ml, 2.53 mmol) in THF (30 ml)] at  $-78 \,^{\circ}$  C and the mixture was stirred at  $-78 \,^{\circ}$  C for 5 min. A solution of 6b (300 mg, 0.843 mmol) in THF (10 ml) was added dropwise to the stirred mixture at -78 °C and the whole was stirred at -78 °C for 30 min. The reaction was quenched with saturated NH<sub>4</sub>Ct. The organic layer was separated and the aqueous layer was extracted with AcOEt. The combined organic layers were washed with water and brine, and then dried. The solvent was evaporated and the residue was chromatographed on silica gel with AcOEt-hexane (2:1) to give a mixture of 10a and 10b (277 mg, 92 %). The diastereometric mixture was used in the next step without further purification. Triethylamine (0.163 m). 1.17mmol) and DMAP (47.5 mg, 0.389 mmol) were added to a solution of the mixture of 10a and 10b (277 mg, 0.778 mmol) in CH<sub>2</sub>Cl<sub>2</sub>(10 ml) at 0 °C and the whole was stirred at 0 °C for 10 min. (+)-MTPACl (0.218 ml, 1.17 mmol) was added to the stirred mixture at 0 °C and the mixture was stirred at 0 °C for 30 min. After the addition of saturated NaHCO<sub>3</sub>, the whole was extracted with CHCl<sub>3</sub>. The extract was washed with water and brine, and then dried. The solvent was evaporated and the residue was chromatographed on silica gel with AcOEt-hexane (1:2) to give (+)-MTPA ester (436 mg, 98 %). The crude product was used in the next step without further purification. A stream of ozone was bubbled through the solution of (+)-MTPA ester (436 mg, 0.762 mmol) in MeOH (8 ml) at -78 °C until a pale blue color developed. Nitrogen was allowed to bubble through the solution to remove excess ozone. Dimethyl sulfide (0.60 ml, 7.6 mmol) was added to the solution. The whole was allowed to warm to 25 °C and stirred for 1 h. The solvent was evaporated and the residue was chromatographed on silica gel with AcOEt-hexane (1:1) to give a mixture of 16a and 16b (372 mg, 98 %). The mixture was separated by HPLC [AcOEt : hexane = 2 : 5, flow rate = 4 ml/min, tR = 39.7 min (16a), tR =

45.8 min (16b)] to afford 16a (350 mg, 92 %) and 16b (22.0 mg, 6%) each as a colorless oil. 16a:  $[\alpha]_{D}^{28}$ -20.9 (c = 1.05, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.58–1.70 (m, 2H), 1.79–2.10 (m, 4H), 2.42 (s, 3H, Ar-CH<sub>3</sub>), 3.40 (d, 1H, *J* = 13.7 Hz, CH<sub>2</sub>SO), 3.53 (brs, 3H, OCH<sub>3</sub>), 3.63 (d, 1H, *J* = 13.7 Hz, CH<sub>2</sub>SO), 5.11 (td, 1H, *J* = 6.8, 4.3 Hz, CHOC), 5.36 (dd, 1H, *J* = 10.3, 4.3 Hz, CHOC), 7.32 (d, 2H, *J* = 8.1 Hz, Ar-H), 7.38–7.42 (m, 3H, Ar-H), 7.50–7.55 (m, 4H, Ar-H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 19.0, 21.4, 27.7, 28.3, 55.3, 61.2, 75.4, 76.1, 121.1, 124.2, 125.3, 127.4, 128.4, 129.7, 130.0, 132.1, 139.8, 142.4, 163.9, 165.8. IR (CHCl<sub>3</sub>): 3005, 2950, 1737, 1585, 1031 cm<sup>-1</sup>. MS *m*/z (%): 498 (M<sup>+</sup>, 5.5), 139 (100). *Anal.* Calcd for C<sub>24</sub>H<sub>25</sub>F<sub>3</sub>O<sub>6</sub>S: C, 57.82 ; H, 5.05. Found : C, 57.76 ; H, 5.19. **16b**:  $[\alpha]_{D}^{30}$  +124.4 (c = 1.19, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.58–1.76 (m, 2H), 1.80–1.92 (m, 2H), 1.95–2.10 (m, 2H), 2.43 (s, 3H, Ar-CH<sub>3</sub>), 3.39 (d, 1H, *J* = 13.9 Hz, CH<sub>2</sub>SO), 3.51 (brs, 3H, OCH<sub>3</sub>), 3.60 (d, 1H, *J* = 13.9 Hz, CH<sub>2</sub>SO), 5.15 (td, 1H, *J* = 6.8, 4.3 Hz, CHOC), 5.36 (dd, 1H, *J* = 9.6, 4.3 Hz, CHOC), 7.33 (d, 2H, *J* = 8.2 Hz, Ar-H), 7.35–7.40 (m, 3H, Ar-H), 7.47–7.51 (m, 2H, Ar-H), 7.53 (d, 2H, *J* = 8.2 Hz, Ar-H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 18.9, 21.3, 164.2, 165.6. IR (CHCl<sub>3</sub>): 3040, 2980, 1745, 1600, 1040 cm<sup>-1</sup>. MS *m*/z (%): 498 (M<sup>+</sup>, 7.6), 139 (100).

(1R, 2S)- and (1S, 2R)-2-[(R)- $\alpha$ -Methoxy- $\alpha$ -(trifluoromethyl)phenylacetoxy]cyclopentyl (Rs)-(p-Tolylsulfinyl)acetate (16a and 16b). A solution of 6b (23.2 mg, 0.0651 mmol) in THF (1 ml) was added dropwise to a stirred LDA solution [prerared from n-BuLi (1.6 M in hexane; 122 µl, 0.195 mmol) and diisopropylamine (27.3  $\mu$ l, 0.195 mmol) in THF (3 ml)] at -78 % and the mixture was stirred at -78 % for 30 The reaction was quenched with saturated NH<sub>4</sub>Cl. The organic layer was separated and the aqueous min. layer was extracted with AcOEt. The combined organic layers were washed with water and brine, and then dried. The solvent was evaporated and the residue was purified by preparative TLC with AcOEt-hexane (1:1) to give a mixture of (Z)-10a and (Z)-10b [18.6 mg, 80%, (Z)-10a : (Z)-10b = 87 : 13] and (E)-10a and (E)-10b [2.5 mg, 11%, (E)-10a : (E)-10b = 83 : 17]. By a similar procedure to that described for the preparation of a mixture of 16a and 16b from a mixture of (E/Z)-10a and (E/Z)-10b, a mixture of (Z)-10a and (Z)-10b [7.4 mg, 0.021 mmol, (E)-10a: (E)-10b = 87: 13] was converted into a mixture of 16a and 16b, which was purified by preparative TLC with AcOEt-hexane (1:1) [9.3 mg, 90 % in 2 steps, 16a: 16b = 81:19]. By a similar procedure to that described for the preparation of a mixture of 16a and 16b from a mixture of (E/Z)-10a and (E/Z)-10b, a mixture of (E)-10a and (E)-10b [2.4 mg, 0.0070 mmol, (E)-10a : (E)-10b = 83 : 17] was converted into a mixture of 16a and 16b [3.3 mg, 94 % in 2 steps, 16a : 16b = 86 : 14].

## (1S, 2R)-2-Acetoxycyclopentyl (R)- $\alpha$ -Methoxy- $\alpha$ -(trifluoromethyl)phenylacetate (9a).

From 16a: An excess of Raney Ni (W2) was added to a solution of 16a (14.2 mg, 0.0285 mmol) in EtOH (1 ml) at 25  $^{\circ}$ C and stirred at 25  $^{\circ}$ C for 5 min, then filtered. The solvent was evaporated and the residue was chromatographed on silica gel with AcOEt-hexane (1 : 2) to give 9a<sup>10</sup> (10.2 mg, quant.) as a colorless oil.

(15, 2R)-2-Acetoxycyclopentyl (S)- $\alpha$ -Methoxy- $\alpha$ -(trifluoromethyl)phenylacetate (ent.-9b). From 16b: By a similar procedure to that described for the preparation of 9a from 16a, 16b (18.2 mg, 0.0365 mmol) was converted into ent.-9b<sup>14</sup> (13.1 mg, quant.) as a colorless oil.

(E,Rs)- and (Z,Rs)-2-[(1R,2S)-2-Hydroxycyclohexyloxy]-1-phenyl-3-(p-tolylsulfinyl)]propene [(E)- and (Z)-11a], (E,Rs)-2-[(1R,2S)-2-Hydroxycyclohexyloxy]-3-phenyl-1-(p-tolylsulfinyl)]propene [(E)-11b], and (E,Rs)- and (Z,Rs)-2-[(1S,2R)-2-Hydroxycyclohexyl-

oxy]-1-phenyl-3-(p-tolylsulfinyl)]propene [(E)- and (Z)-11c]. Representative Procedure inTable 3: A solution of acetal 7b (100 mg, 0.270 mmol) in THF (2ml) was added to a stirred LDA solution [prepared from n-BuLi (1.6 M in hexane; 0.844 ml, 1.35 mmol) and diisopropylamine (0.188 ml, 1.35 mmol) in THF (8 ml)] at −78 ℃ and the mixture was stirred at −78 ℃ for 30 min. The reaction was quenched with saturated NH<sub>4</sub>Cl. The organic layer was separated and the aqueous layer was extracted with AcOEt. The combined organic layers were washed with brine and dried. The solvent was evaporated and the residue was chromatographed on silica gel with AcOEt-hexane (1:2) to give a mixture of (Z)-11a and (Z)-11c [75.1 mg, 75 %, (Z)-11a : (Z)-11c = 96 : 4], (E)-11a (3.8 mg, 4 %), (E)-11b (13.3 mg, 13 %), and (E)-11c (2.9 mg, 3%) each as colorless crystals. A mixture of (Z)-11a and (Z)-11c:  $^{1}$ H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.19–1.33 (m, 2H, CH<sub>2</sub>), 1.41–1.67 (m, 4H, CH<sub>2</sub>), 1.70–1.83 (m, 2H, CH<sub>2</sub>), 2.39 (s,  $3 \times 4/100$ H, Ar–CH<sub>3</sub>), 2.41 (s,  $3 \times 4/100$ H, Ar–CH<sub>3</sub>), 3.41 (s,  $3 \times 4/100$ H, 3.41 (s,  $3 \times 4$ 96/100H, Ar-CH<sub>1</sub>), 3.53 (d,  $1 \times 96/100$ H, J = 13.7 Hz, CH<sub>2</sub>SO), 3.73 (d,  $1 \times 4/100$ H, J = 13.7 Hz, CH<sub>2</sub>SO), 3.66–3.75 (m, 1H, CHOH), 3.79 (d,  $1 \times 4/100$ H, J = 13.7 Hz, CH,SO), 3.97 (d,  $1 \times 96/100$ H, J = 13.7 Hz, CH,SO), 4.11-4.19 (m, 1H, CHOC), 5.40 (s,  $1 \times 4/100$ H, C=CH), 5.60 (s,  $1 \times 96/100$ H, C=CH), 7.19 (d, 1  $\times$  96/100H, J = 7.7 Hz, Ar-H), 7.20 (d, 1 $\times$  4/100H, J = 7.7 Hz, Ar-H), 7.27-7.34 (m, 4H, Ar-H), 7.43(d,  $2 \times 4/100$ H, J = 7.7 Hz, Ar-<u>H</u>), 7.49 (d,  $2 \times 96/100$ H, J = 7.7 Hz, Ar-<u>H</u>), 7.55 (d,  $2 \times 4/100$ H, J = 8.5 Hz, Ar-<u>H</u>), 7.57 (d,  $2 \times 96/100$ H, J = 8.5 Hz, Ar-<u>H</u>). IR (KBr): 3388, 2937, 2862, 1727, 1641, 1597, 1493, 1446, 1363, 1170, 1083, 1039 cm<sup>-1</sup>. MS m/z (%): 370 (M<sup>+</sup>, 18.7), 91 (100). (E)-11a: mp 103-105 ℃ (AcOEt-hexane).  $[\alpha]_{D}^{26}$  -45.8 (c = 0.58, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.30–1.44 (m, 2H), 1.56–1.78 (m, 4H), 1.82–1.89 (m, 1H), 2.01–2.09 (m, 1H), 2.38 (s, 3H, Ar-CH<sub>3</sub>), 3.73 (d, 1H, J = 13.3 Hz, CH<sub>3</sub>SO), 3.93 (d, 1H, J = 13.3 Hz, CH,SO), 3.94–3.99 (m, 1H, CHOH), 4.32–4.37 (m, 1H, CHOC), 5.92 (s, 1H, C=CHPh, 6.92 (d, 2H, J = 6.9 Hz, Ar-H), 7.11–7.20 (m, 5H, Ar-H), 7.38 (d, 2H, J = 8.5 Hz, Ar-H). <sup>13</sup>C-NMR (CDCl<sub>1</sub>) δ: 21.4, 21.6, 22.1, 26.7, 30.5, 57.6, 67.6, 76.9, 107.3, 124.2, 126.1, 128.1, 128.8, 129.8, 135.5, 139.4, 141.6, 146.6. IR (KBr): 3375, 2927, 2854, 1734, 1641, 1495, 1446, 1400, 1240, 1180 1140, 1084, 1041 cm<sup>-1</sup>. MS m/z (%): 370 (M<sup>+</sup>, 1.7), 133 (100). Anal. Calcd for C<sub>22</sub>H<sub>26</sub>O<sub>3</sub>S: C, 71.32 ; H, 7.07; S, 8.66. Found : C, 71.12; H, 7.06; S, 8.63. (E)-11b: mp 109–110  $^{\circ}$ C (AcOEt-hexane).  $[\alpha]_{D}^{27}$  -67.5  $(c = 0.58, CHCl_3)$ . <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.15–1.80 (m, 8H), 2.39 (s, 3H, Ar-CH<sub>3</sub>), 3.73–3.78 (m, 1H, CHOH), 3.95 (d, 1H, J = 14.1 Hz, CH<sub>2</sub>SO), 4.08 (d, 1H, J = 14.1 Hz, CH<sub>2</sub>SO), 4.13–4.18 (m, 1H, CHOC), 5.52 (s, 1H, C=C<u>H</u>), 7.28–7.39 (m, 7H, Ar–<u>H</u>), 7.43 (d, 2H, J = 8.6 Hz, Ar–<u>H</u>). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 20.8, 21.3, 21.8, 25.7, 30.2, 38.3, 68.1, 77.7, 109.3, 124.1, 127.2, 128.8, 128.9, 129.9, 136.9, 140.6, 142.5, 166.0. IR (KBr): 3394, 3346, 2937, 2860, 1734, 1593, 1495, 1450, 1313, 1219, 1163, 1082, 1032 cm<sup>-1</sup>. MS m/z (%): 370 (M<sup>+</sup>, 25.7), 224 (100). Anal. Calcd for C<sub>22</sub>H<sub>26</sub>O<sub>3</sub>S: C, 71.32 ; H, 7.07 ; S, 8.66. Found : C, 71.08; H, 7.03; S, 8.57. (E)-11c: mp 107–108 °C (AcOEt–hexane).  $[\alpha]_{D}^{25}$  +270.2 (c = 0.69, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 1.34–1.42 (m, 2H), 1.58–1.75 (m, 4H), 1.80–1.89 (m, 1H), 2.03–2.12 (m, 1H), 2.40 (s, 3H, Ar-CH<sub>4</sub>), 3.64 (d, 1H, J = 12.8 Hz, CH<sub>2</sub>SO), 3.74 (d, 1H, J = 12.8 Hz, CH<sub>2</sub>SO), 3.85–3.94 (m, 1H, 1H, 2H) and 2H) CHOH), 4.39–4.12 (m, 1H, CHOC), 6.03 (s, 1H, C=CHPh), 7.20 (d, 2H, J = 8.6 Hz, Ar–H), 7.22–7.29 (m, 5H, Ar-H), 7.50 (d, 2H, J = 8.6 Hz, Ar-H). <sup>13</sup>C-NMR (CDCl<sub>1</sub>) & 21.4, 21.6, 22.0, 26.9, 30.5, 59.8, 68.3, 77.5, 107.1, 123.9, 126.4, 128.3, 128.8, 129.9, 135.8, 140.7, 141.7, 148.4. IR (KBr): 3425, 2929, 2854, 1735, 1639, 1495, 1446, 1400, 1243, 1182, 1128, 1083, 1041 cm<sup>-1</sup>. MS *m/z* (%): 370 (M\*, 25.1), 91 (100). Anal. Calcd for C<sub>22</sub>H<sub>26</sub>O<sub>3</sub>S: C, 71.32; H, 7.07; S, 8.66. Found : C, 71.11; H, 7.04; S, 8.59.

# (1R, 2S)-2-Methoxyethoxymethoxycyclohexyl (R)- $\alpha$ -Methoxy- $\alpha$ -(trifluoromethyl)phenyl-

acetate (17a). From (E)-11a: MEMCl (43  $\mu$ l, 0.38 mmol) was added to a mixture of alcohol (E)-11a (70.0 mg, 0.189 mmol) and disopropylethylamine (99  $\mu$ l, 0.57 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (1 ml) at 25 °C and the mixture was stirred at 25 °C for 36 h. After the addition of saturated NaHCO<sub>3</sub>, the whole was extracted with Et<sub>2</sub>O.

The extract was washed with saturated NH<sub>4</sub>Cl and brine, and then dried. The solvent was evaporated and the residue was chromatographed on silica gel with AcOEt-hexane (2:1) to give MEM ether (75.0 mg, 87 %) as a colorless oil.  $[\alpha]_{D}^{26}$  +25.5 (c = 0.99, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.35–1.46 (m, 2H), 1.60–1.76 (m, 4H), 1.91–2.07 (m, 2H), 2.38 (s, 3H, Ar–CH<sub>4</sub>), 3.37 (s, 3H, OCH<sub>4</sub>), 3.57 (t, 2H, J = 4.7 Hz, CH<sub>2</sub>OCH<sub>4</sub>), 3.64 (d, 1H, J = 12.8 Hz, CH,SO), 3.74–3.79 (m, 1H, OCH,CH,OCH<sub>3</sub>), 3.80–3.85 (m, 1H, OCH,CH,OCH<sub>3</sub>), 3.91– 3.94 (m, 1H, CHOC), 3.95 (d, 1H, J = 12.8 Hz, CH,SO), 4.36–4.40 (m, 1H, CHOC), 4.88 (s, 2H, OCH,O), 5.90 (s, 1H, C=CH), 7.05 (d, 2H, J = 7.3 Hz, Ar-H), 7.14 (t, 1H, J = 7.3 Hz, Ar-H), 7.21 (t, 4H, J = 8.1 Hz, Ar-<u>H</u>), 7.46 (d, 2H, J = 8.1 Hz, Ar-<u>H</u>). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 21.4, 21.8, 22.0, 26.5, 28.4, 59.0, 60.5, 66.9 71.8, 74.8, 75.7, 94.6, 107.0, 124.2, 126.1, 128.1, 128.9, 129.7, 135.6, 140.9, 141.5, 147.6. IR (KBr): 2939, 1639, 1492, 1446, 1398, 1240, 1184, 1130, 1047 cm<sup>-1</sup>. MS m/z (%): 458 (M<sup>+</sup>, 11.3), 89 (100). Anal. Calcd for C<sub>36</sub>H<sub>14</sub>O<sub>5</sub>S: C, 67.43 ; H, 7.51 ; S, 6.92. Found : C, 67.46 ; H, 7.59 ; S, 6.52. IN HCl (0.03 ml) was added to a solution of MEM ether (14.5 mg, 0.0316 mmol) in acetone (1 ml) at 25 °C and the mixture was stirred at 25 °C for 15 min. After the addition of saturated NaHCO<sub>3</sub>, acetone was evaporated and the residue was extracted with AcOEt. The extract was washed with brine and dried. The solvent was evaporated and the residue was diluted with  $CH_2Cl_2$  (2 ml). DMAP (19.3 mg, 0.158 mmol) was added to the solution with stirring at 0 % and the whole was stirred at 0 % for 10 min. (+)-MTPACI (18  $\mu$ I, 0.095 mmol) was added to the stirred mixture at 0 °C and the whole was stirred at 25 °C for 30 min. After the addition of saturated NaHCO<sub>3</sub>, the whole was extracted with Et, O. The extract was washed with saturated NH<sub>4</sub>Cl and brine, and then dried. The solvent was evaporated and the residue was chromatographed on silica gel with AcOEt-hexane (2:1) to give 17a (10.6 mg, 80 % in 2 steps) as a colorless oil.

**From (E)-11b:** By a similar procedure to that described for the preparation of the MEM ether from (E)-11a, (E)-11b (35.2 mg, 0.0951 mmol) was converted into MEM ether (38.5 mg, 88 %) as a colorless oil.  $[\alpha]_{D}^{24}$  -41.1 (c = 1.32, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.10–1.39 (m, 4H), 1.48–1.86 (m, 4H), 2.39 (s, 3H, Ar-C<u>H<sub>3</sub></u>) 3.37 (s, 3H, OC<u>H<sub>3</sub></u>), 3.48–3.55 (m, 2H, C<u>H<sub>2</sub>OCH<sub>3</sub></u>), 3.62–3.66 (m, 2H, OC<u>H<sub>2</sub>CH<sub>2</sub>OCH<sub>3</sub></u>), 3.72–3.77 (m, 1H, C<u>HOC</u>), 3.85 (d, 1H, J = 13.7 Hz, Ar-C<u>H<sub>2</sub></u>), 4.19 (d, 1H, J = 13.7 Hz, Ar-C<u>H<sub>2</sub></u>), 4.23–4.28 (m, 1H, C<u>HOC</u>), 4.65 (d, 1H, J = 6.7 Hz, OC<u>H<sub>2</sub>O</u>), 4.70 (d, 1H, J = 6.7 Hz, OC<u>H<sub>2</sub>O</u>), 5.44 (s, 1H, C<u>H</u>=C), 7.23–7.40 (m, 9H, Ar-<u>H</u>). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 21.3, 21.3, 21.8, 26.0, 28.0, 38.3, 59.0, 66.9, 71.7, 74.4, 76.2, 94.3, 108.3, 124.2, 126.8, 128.5, 128.9, 129.8, 137.2, 140.4, 142.7, 166.7. IR (KBr): 2937, 1595, 1495, 1454, 1313, 1220, 1161, 1107, 1082, 1041 cm<sup>-1</sup>. MS m/z (%): 458 (M<sup>\*</sup>, 3.8), 89 (100). Anal. Calcd for C<sub>26</sub>H<sub>34</sub>O<sub>5</sub>S: C, 67.43; H, 7.51; S, 6.92. Found : C, 67.59; H, 7.61; S, 6.67. By a similar procedure to that described for the preparation of 17a from MEM ether of (E)-11a, MEM ether of (E)-11b (22.0 mg, 0.048 mmol) was converted into 17a (16.5 mg, 82 % in 2 steps) as a colorless oil.  $[\alpha]_D^{20} - 1.0$  (c = 0.73, CHCl<sub>3</sub>). <sup>1</sup>H-NMR spectroscopic property of 17a was consistent with the authentic data written in reference 5b.

#### (15, 2R)-2-Methoxyethoxymethoxycyclohexyl (R)-a-Methoxy-a-(trifluoromethyl)phenyl-

acetate (17b). By a similar procedure to that described for the preparation of the MEM ether from (*E*)-11a, (*E*)-11c (20.0 mg, 0.0540 mmol) was converted into MEM ether (22.0 mg, 89 %) as a colorless oil.  $[\alpha]_{D}^{24}$ +39.6 (c = 0.70, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.35–1.46 (m, 2H), 1.51–1.80 (m, 4H), 1.91–2.01 (m, 1H), 2.10–2.20 (m, 1H), 2.39 (s, 3H, Ar-CH<sub>3</sub>), 3.38 (s, 3H, OCH<sub>3</sub>), 3.50 (d, 1H, *J* = 12.8 Hz, CH<sub>2</sub>SO), 3.55– 3.58 (m, 2H, CH<sub>2</sub>OCH<sub>3</sub>), 3.74–3.83 (m, 2H, OCH<sub>2</sub>CH<sub>2</sub>OCH<sub>3</sub>), 3.88–3.94 (m, 1H, CHOC), 4.03(d, 1H, *J* = 12.8 Hz, CH<sub>2</sub>SO), 4.45–4.49 (m, 1H, CHOC), 4.86 (d, 2H, *J* = 1.3 Hz, OCH<sub>2</sub>O), 5.99 (s, 1H, C=CH), 7.14– 7.18 (m, 2H, Ar-H), 7.22–7.28 (m, 5H, Ar-H), 7.54 (d, 2H, *J* = 8.2 Hz, Ar-H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 21.2, 21.4, 22.5, 26.5, 27.8, 59.0, 60.5, 66.9, 71.7, 75.1, 75.2, 94.2, 107.5, 124.2, 126.1, 128.1, 129.0, 129.7, 135.7, 141.0, 141.4, 147.3. IR (KBr): 2937, 1732, 1641, 1492, 1446, 1398, 1240, 1182, 1086, 1045 cm<sup>-1</sup>. MS m/z (%): 458 (M<sup>+</sup>, 3.6), 89 (100). Anal. Calcd for C<sub>26</sub>H<sub>34</sub>O<sub>5</sub>S: C, 67.43; H, 7.51; S, 6.92. Found : C, 67.56; H, 7.58; S, 6.65. By a similar procedure to that described for the preparation of **17a** from MEM ether of (*E*)-**11a**, MEM ether of (*E*)-**11c** (14.5 mg, 0.0316 mmol) was converted into known **17b** (11.4 mg, 86 %) as a colorless oil.  $[\alpha]_D^{2^4}$  +29.4 (c = 0.25, CHCl<sub>3</sub>). <sup>1</sup>H-NMR spectroscopic poperty of **17b** was consistent with the authentic data written in reference 5b.

(1R,2S)-2-Acetoxycyclohexyl (Rs)-(p-Tolylsulfinyl)acetate (18a). DMAP (50.2 mg, 0.411 mmol) were added to a solution of (E)-11a (38.0 mg, 0.103 mmol) in CH,Cl, (3 ml) at 0 °C and the mixture was stirred at 0 % for 10 min. Acetyl chloride (15 µl, 0.21 mmol) was added to the stirred mixture at 0 % and the mixture was stirred at 0 °C for 30 min. After the addition of saturated NaHCO<sub>3</sub>, the whole was extracted with Et,O. The extract was washed with saturated NH<sub>4</sub>Cl and brine, and then dried. The solvent was evaporated and the residue was chromatographed on silica gel with AcOEt-hexane (1:1) to give the acetate (40.0 mg, 94 %) as a colorless oil.  $[\alpha]_{0}^{33}$  +29.7 (c = 1.43, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.40–1.52 (m, 2H), 1.65–1.80 (m, 4H), 1.92-2.12 (m, 2H), 2.12 (s, 3H, OCH<sub>3</sub>), 2.39 (s, 3H, Ar-CH<sub>3</sub>), 3.62 (d, 1H, J = 12.9 Hz, CH<sub>3</sub>SO), 3.93 (d, 1H, J = 12.9 Hz, CH,SO), 4.46–4.51 (m, 1H, CHOC), 4.99–5.04 (m, 1H, CHOC), 5.91 (s, 1H, C=C<u>H</u>), 7.07 (d, 2H, J = 7.7 Hz, Ar-<u>H</u>), 7.13-7.25 (m, 5H, Ar-<u>H</u>), 7.47 (d, 2H, J = 8.6 Hz, Ar-<u>H</u>). <sup>13</sup>C-NMR (CDCl.) & 21.0, 21.3, 21.4, 22.4, 27.1, 27.5, 60.6, 71.9, 73.7, 107.4, 124.2, 126.2, 128.1, 128.9, 129.7, 135.4, 141.0, 141.5, 147.8, 170.8. IR (KBr): 2941, 1732, 1641, 1493, 1446, 1373, 1238, 1184, 1132, 1086, 1051 cm<sup>-1</sup>. MS m/z (%): 396 (M<sup>\*</sup>-O, 0.7), 133 (100). High MS Calcd for C<sub>14</sub>H<sub>28</sub>O<sub>4</sub>S: 412.1706. Found: 412.1704. A stream of ozone was bubbled through the solution of the acetate (66.0 mg, 0.160 mmol) in MeOH (5 ml) at -78 °C until a pale blue color developed. Nitrogen was allowed to bubble through the solution to remove excess ozone. Dimethyl sulfide (118 µl, 1.60 mmol) was added to the mixture. The solution was allowed to warm to 25 °C and stirred for 1h. The solvent was evaporated and the residue was chromatographed on silica gel with AcOEt-hexane (1 : 1) to give 18a (50.0 mg, 92 %) as a colorless oil.  $[\alpha]_{0}^{32}$ +94.9 (c = 0.90, CHCl<sub>1</sub>). <sup>1</sup>H-NMR (CDCl<sub>1</sub>)  $\delta$ : 1.38–1.46 (m, 2H), 1.57–1.70 (m, 4H), 1.78–1.86 (m, 2H), 2.02 (s, 3H, OCH<sub>1</sub>), 2.42 (s, 3H, Ar-CH<sub>1</sub>), 3.66 (d, 1H, J = 13.7 Hz, CH<sub>2</sub>SO), 3.82 (d, 1H, J = 13.7 Hz, CH,SO), 4.95–4.99 (m, 1H, CHOC), 5.06–5.10 (m, 1H, CHOC), 7.34 (d, 2H, J = 8.5 Hz, Ar–H), 7.59 (d, 2H, J = 8.5 Hz, Ar–<u>H</u>). <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 21.1, 21.4, 21.4, 21.6, 27.4, 27.5, 61.9, 70.7, 72.9, 124.2, 130.1, 140.1, 142.3, 164.3, 170.4. IR (KBr): 2943, 2866, 1732, 1495, 1450, 1369, 1267, 1236, 1120, 1086 1051 cm<sup>-1</sup>. MS m/z (%): 338 (M<sup>+</sup>, 35.7), 139 (100). High MS Calcd for C<sub>17</sub>H<sub>27</sub>O<sub>5</sub>S: 338.1186. Found: 338.1179.

(15,2*R*)-2-Acetoxycyclohexyl (*Rs*)-(*p*-Tolylsulfinyl)acetate (18b). By a similar procedure to that described for the preparation of the acetate from (*E*)-11a, (*E*)-11c (60.2 mg, 0.163 mmol) was converted into the acetate (61.0 mg, 91 %) as a colorless oil.  $[\alpha]_D^{30}$  +52.9 (c = 0.65, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>) & 1.40–1.52 (m, 2H), 1.65–1.80 (m, 4H), 1.92–2.15 (m, 2H), 2.11 (s, 3H, OCH<sub>3</sub>), 2.39 (s, 3H, Ar–CH<sub>3</sub>), 3.55 (d, 1H, *J* = 12.8 Hz, CH<sub>2</sub>SO), 3.98 (d, 1H, *J* = 12.8 Hz, CH<sub>2</sub>SO), 4.42–4.47 (m, 1H, CHOC), 5.04–5.10 (m, 1H CHOC), 5.93 (s, 1H, C=CH), 7.07 (d, 2H, *J* = 6.9 Hz, Ar–H), 7.13–7.25 (m, 5H, Ar–H), 7.50 (d, 2H, *J* = 7.7 Hz, Ar–H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>) & 21.0, 21.3, 21.3, 22.3, 27.0, 27.2, 60.6, 72.0, 73.8, 107.6, 124.2, 126.2, 128.1, 128.9, 129.7, 135.3, 140.9, 141.5, 147.2, 170.6. IR (KBr): 2945, 2862, 1736, 1641, 1599, 1493, 1446, 1371, 1234, 1182, 1132, 1047 cm<sup>-1</sup>. MS *m/z* (%): 412 (M<sup>+</sup>, 0.7), 133 (100). High MS Calcd for C<sub>24</sub>H<sub>28</sub>O<sub>4</sub>S: 412.1706. Found: 412.1695. By a similar procedure to that described for the preparation of **18a** from the acetate of (*E*)-**11a**, the acetate of (*E*)-**11c** (33.9 mg, 0.0822 mmol) was converted into **18b** (25.0 mg, 90 %) as a colorless oil.  $[\alpha]_D^{31}$  +91.7 (c = 1.27, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (CDCl<sub>3</sub>) & 1.38–1.44 (m, 2H),

1.56–1.69 (m, 4H), 1.73–1.84 (m, 2H), 2.05 (s, 3H, OCH<sub>3</sub>), 2.42 (s, 3H, Ar–CH<sub>3</sub>), 3.66 (d, 1H, J = 13.7 Hz, CH<sub>2</sub>SO), 3.87 (d, 1H, J = 13.7 Hz, CH<sub>2</sub>SO), 4.95–4.99 (m, 1H, CHOC), 5.05–5.09 (m, 1H, CHOC), 7.34 (d, 2H, J = 8.5 Hz, Ar–H), 7.60 (d, 2H, J = 8.5 Hz, Ar–H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>) & 21.1, 21.1, 21.4, 21.8, 27.3, 27.7, 61.9, 70.8, 72.8, 124.2, 130.0, 140.1, 142.3, 164.1, 170.4. IR (KBr): 2941, 2866, 1736, 1495, 1450, 1371, 1269, 1236, 1120, 1084, 1053, 1018 cm<sup>-1</sup>. MS m/z (%): 338 (M<sup>+</sup>, 35.7), 139 (100). High MS Calcd for C<sub>17</sub>H<sub>22</sub>O<sub>5</sub>S: 338.1188. Found: 338.1188.

(1R,2S)- and (1S,2R)-2-Acetoxycyclohexyl (Rs)-(p-Tolylsulfinyl)acetate (18a and 18b). From a mixture of (Z)-11a and (Z)-11c: By a similar procedure to that described for the preparation of the acetate from (E)-11a, a mixture of (Z)-11a and (Z)-11c [99.8 mg, 0.270 mmol, (Z)-11a: (Z)-11c = 96 : 4] was converted into a mixture of the acetate (110 mg, 99 %, 96 : 4). Then, by a similar procedure to that described for the preparation of 18a from the acetate of (E)-11a, a mixture of the acetate (100 mg, 0.243 mmol) was converted into a mixture of 18a and 18b (75.0 mg, 91 %, 18a : 18b = 96 : 4).

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## **REFERENCES AND NOTES**

- For reviews of chemical asymmetric desymmetrization of σ-symmetrical compounds, see: Ward R. S. Chem. Soc. Rev. 1990, 19, 1–19; Maier M. Organic Synthesis Highlights II; Waldmann H. ed.; VCH: New York, 1995; pp. 203–222.
- 2. Nara M.; Terashima S.; Yamada S. Tetrahedron 1980, 36, 3161-3170 and 3171-3175.
- Reviews, (a) Theil F. Chem. Rev. 1995, 95, 2203-2227. (b) Santaniello E.; Ferraboschi P.; Grisenti P.; Manzocchi A. Chem. Rev. 1992, 92, 1071-1140. (c) Faber K.; Riva S. Synthesis 1992, 895-910 (d) Altenbach H.-J. Organic Synthesis Highlights; Mulzer J., Altenbach H.-J., Braun M., Krohn K., Reissig H.-U. eds.; VCH: New York, 1991; pp. 224-231. (e) Xie Z.-F. Tetrahedron: Asymmetry 1991, 2, 733-750. (f) Zhu L.-M.; Tedford M. C. Tetrahedron 1990, 46, 6587-6611. (g) Seebach D. Angew. Chem. Int. Ed. Engl. 1990, 29, 1320-1367. (h) Ohno M.; Otsuka M. Org. React. 1989, 37, 1-55.
- For differentiation based on acetal cleavage using sulfoxide as a chiral auxiliary, see: (a) Maezaki N.; Murakami M; Soejima M.; Tanaka T.; Imanishi T.; Iwata C. Chem. Pharm. Bull. in press. (b) Maezaki N.; Soejima M.; Sakamoto A.; Sakamoto I.; Mastumori Y.; Tanaka T.; Ishida T.; In Y.; Iwata C. Tetrahedron : Asymmetry 1996, 7, 29–32. (c) Maezaki N.; Soejima M.; Takeda M.; Sakamoto A.; Tanaka T.; Iwata C. J. Chem. Soc., Chem. Commun. 1994, 1345–1346. (d) Iwata C.; Maezaki N.; Hattori K.; Fujita M.; Moritani Y.; Takemoto Y.; Tanaka T.; Imanishi T. Chem. Pharm. Bull. 1993, 41, 339–345. (e) Iwata C.; Maezaki N.; Hattori K.; Fujita M.; Moritani Y.; Takemoto Y.; Tanaka T.; Imanishi T. Chem. Pharm. Bull. 1993, 41, 946–950. (f) Iwata C.; Maezaki N.; Murakami M.; Soejima M.; Tanaka T.; Imanishi T. J. Chem. Soc., Chem. Commun. 1992, 516–518. (g) Iwata C.; Fujita M.; Moritani Y.; Sugiyama K.; Hattori K.; Imanishi T. Tetrahedron Lett. 1987, 28, 3131–3134.

(h) Iwata C.; Fujita M.; Moritani Y.; Hattori K.; Imanishi T. Tetrahedron Lett. 1987, 28, 3135-3138.

- For differentiation based on acetal cleavage using trans-1,2-cyclohexanediol as a chiral auxiliary, see; (a) Sakai K.; Suemune H. Tetrahedron: Asymmetry 1993, 4, 2109–2118. (b) Suemune H.; Watanabe K.; Kato K.; Sakai K. Tetrahedron: Asymmetry 1993, 4, 1767–1770.
- For differentiation based on acetal cleavage using menthone as a chiral auxiliary, see: (a) Harada T.; Oku A. Synlett 1994, 95–104. (b) Harada T.; Wada I.; Oku A. J. Org. Chem. 1989, 54, 2599–2605. (c) Harada T.; Sakamoto K.; Ikemura Y.; Oku A. Tetrahedron Lett. 1988, 29, 3097–3100. (d) Harada T.; Hayashiya T.; Wada I.; Iwa-ake N.; Oku A. J. Am. Chem. Soc. 1987, 109, 527–532. (e) Harada T.; Wada I.; Oku A. Tetrahedron Lett. 1987, 28, 4181–4184.
- For direct differentiation of σ-symmetrical diols, see: (a) Ishihara K.; Kubota M.; Yamamoto H. Synlett 1994, 611-614. (b) Ikeda S.; Weinhouse M. I.; Janda K. D.; Lerner R. A. J. Am. Chem. Soc. 1991, 113, 7763-7764; Suzuki T.; Uozumi Y.; Shibasaki M. J. Chem. Soc., Chem. Commun. 1991, 1593-1595. (c) Appelt A.; Willis A. C.; Wild S. B. J. Chem. Soc., Chem. Commun. 1988, 938-940. (d) Mukaiyama T.; Tomioka I.; Shimizu M. Chem. Lett. 1984, 49-52. (e) Mukaiyama T.; Tanabe Y., Shimizu M. Chem. Lett. 1984, 401-404. (f) Ichikawa J.; Asami M.; Mukaiyama T. Chem. Lett. 1984. 949-952. (g) Ishii Y.; Osakada K.; Ikariya T.; Saburi M.; Yoshikawa S. Chem. Lett. 1982, 1179-1182. (h) Nara M.; Terashima S.; Yamada S. Tetrahedron 1980, 36, 3161-3170.
- β-Ketosulfoxides 1 and 2 were prepared as previously described by the reaction of the corresponding esters with enantiomerically pure (R)-methyl p-tolyl sulfoxide (Solladié, G. Synthesis 1981, 185–196.) in the presence of LDA. The optical purity was determined by comparison with the reported specific rotation (Banfi, L.; Colombo, L.; Gennari, C.; Annunziata, R.; Cozzi, F. Synthesis 1982, 829–831. Kunieda, N.; Nokami, J.; Kinoshita, M Chem. Lett. 1974, 369–372.).
- 9. Tsunoda T.; Suzuki M.; Noyori R. Tetrahedron Lett. 1980, 21, 1357-1358.
- 10. The stereochemistry of the acetals in 4-7a and 4-7b was determined based on nuclear Overhauser effect (NOE) experiments, in which marked NOE enhancements were observed between the angular methine protons on the bicyclic ring and the substituent R or the sulfinylmethyl group (Fig. 1).



Fig. 1. NOE Observaon of  $\alpha$ -Sulfinyl Actals.

11. Diasteteoselectivity of acetalization can be explained as follows. Acetalization of  $\beta$ -ketosulfoxides would proceed *via* the most stable oxonium intermediates (a or b), in which the smaller substituent (Rs) eclipses with the hydrogen (H\*) to minimize the steric interaction, therby affording mainly the acetals with the *endo*-larger group (RL). The higher selectivity for the acetal **5b** is due to the great difference in size between the phenyl group (RL) and the sulfinylmethyl group (Rs) (Fig.2).



Fig. 2. Plausible Reaction Mechanism for Acetalization.

12. The stereochemistry of the olefin in 8b, 10a-b, and 11a-c was determined by NOE experiment (Fig. 3).



Fig. 3. Geometry of the Olefin Moieties.

- 13. (a) Dale J. A.; Mosher H. S. J. Am. Chem. Soc. 1973, 95, 512-519. (b) Dale J. A.; Dull D. J.; Mosher H. S. J. Org. Chem. 1969, 34, 2543-2549.
- 14. Xie Z.-F.; Suemune H.; Nakamura I.; Sakai K. Chem. Pharm. Bull. 1987, 35, 4454-4459.
- 15. The stereochemistry of alcohols 10a and 10b was determined as follows. The diastereomeric isomers were separated into a mixture of (Z)-10a and (Z)-10b and a mixture of (E)-10a and (E)-10b. Since further separation was difficult at this stage, they were first converted into 16a and 16b by esterification and, ozonolysis and then separated. Products 16a and 16b were assumed to be derived from 10a and 10b, respectively, since the ratio of 10a to 10b was almost equal to that of 16a to 16b in each case. Desulfurization of 16a and 16b with Raney Ni (W2) afforded the known 9a and ent.-9b, respectively (Scheme 9).



then Me<sub>2</sub>S; c) Raney Ni (W2), EtOH, room temp.

Scheme 9

16. The absolute configurations of (E)-11a, (E)-11b, and (E)-11c were confirmed by their transformation into known compounds 17a or 17b<sup>sb</sup>. After protection of the alcohols as 2-methoxyethoxymethyl (MEM) ether, (E)-11a and (E)-11b were hydrolyzed with dilute hydrochloric acid followed by esterification with (+)-MTPACl to afford known 17a. (E)-11c was also converted into known 17b. Since separation of (Z)-11a and (Z)-11c was difficult, the mixture was converted into 18a and 18b via acetylation and ozonolysis without separation. Compounds (Z)-11a and (Z)-11c were considered to be derived from 18a and 18b, respectively, because the ratio of (Z)-11a to (Z)-11c (96:4) was the same as that of 18a to 18b (96:4). The absolute configurations of 18a and 18b were determined by <sup>1</sup>H-NMR spectroscopic analysis in comparison with those derived from (E)-11a and (E)-11c with known absolute configurations (Scheme 10).



#### Scheme 10

- Juaristi E. Introduction to Stereochemistry and Conformational Analysis; Wiley: New York, 1991; ch. 16; pp. 271-285.
- (a) Nakamura K.; Higaki M.; Adachi S.; Oka S.; Ohno A. J. Org. Chem. 1987, 52, 1414–1417. (b) Ogura K. Comprehensive Organic Synthesis Trost B. M.; Fleming I. eds.; Pergamon Press, 1991; vol. 1, pp. 505–539 and references cited therein. (c) Krief A. Comprehensive Organic Synthesis Trost B. M.; Fleming I. eds.; Pergamon Press, 1991; vol. 3, pp. 85–191.

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