

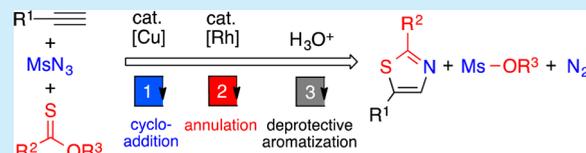
Facile Synthesis of 2,5-Disubstituted Thiazoles from Terminal Alkynes, Sulfonyl Azides, and Thionoesters

Tomoya Miura,* Yuuta Funakoshi, Yoshikazu Fujimoto, Junki Nakahashi, and Masahiro Murakami*

Department of Synthetic Chemistry and Biological Chemistry, Kyoto University, Katsura, Kyoto 615-8510, Japan

S Supporting Information

ABSTRACT: A sequential procedure for the synthesis of 2,5-disubstituted thiazoles from terminal alkynes, sulfonyl azides, and thionoesters is reported. A copper(I)-catalyzed 1,3-dipolar cycloaddition of terminal alkynes with sulfonyl azides affords 1-sulfonyl-1,2,3-triazoles, which then react with thionoesters in the presence of a rhodium(II) catalyst. The resulting 3-sulfonyl-4-thiazolines subsequently aromatize into the corresponding 2,5-disubstituted thiazoles by elimination of the sulfonyl group.



A thiazole ring represents a privileged structural motif often found in natural products and pharmaceutically active substances.¹ Moreover, thiazoles multisubstituted with aryl groups are core components constituting materials of interesting optical² and electronic properties.³ The condensation reaction of α -halocarbonyl compounds with thioamides, named Hantzsch thiazole synthesis, is the most widely used and reliable procedure for their preparation.⁴ However, 2,5-disubstituted thiazoles are less accessible with the Hantzsch thiazole synthesis⁵ because chemically labile α -haloaldehydes are required for their synthesis. An alternative method to synthesize those substituted thiazoles is given by palladium-catalyzed C–H arylation reactions of thiazoles with aryl halides, in which an aryl substituent is installed onto a preformed thiazole core.⁶ On the other hand, the synthetic methods to directly construct thiazole skeletons possessing 2,5-disubstituents from easily available simple starting substances are still limited.^{7,8} We now report a facile method to synthesize 2,5-disubstituted thiazoles from terminal alkynes, sulfonyl azides, and thionoesters (Figure 1). The transformation

electrophilic carbene carbon and a nucleophilic imino nitrogen in the molecule. They act as the 1,3-dipoles in a formal sense in the reactions with a variety of dipolarophiles, which include alkynes,¹¹ allenes,¹² nitriles,¹³ aldehydes and imines,¹⁴ isocyanates and isothiocyanates,¹⁵ and indoles,¹⁶ affording the corresponding [3 + 2] cycloadducts. In the present study, we examined if thionoesters could serve as the suitable dipolarophiles.¹⁷ First, we prepared *O*-methyl benzothioate (**2a**) from methyl benzoate and the Lawesson's reagent according to a literature procedure.¹⁸ Then, 4-phenyl-1-tosyl-1,2,3-triazole (**1a**, 1.0 equiv) was treated with **2a** (1.5 equiv) in the presence of (^tBuCO₂)₄Rh₂ (2.0 mol %) and 4 Å molecular sieves (MS)¹⁹ in chloroform (2 mL) at 70 °C (eq 1). The triazole **1a** was

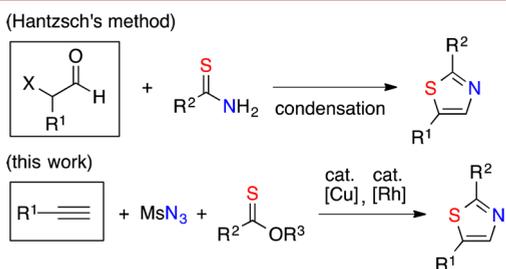
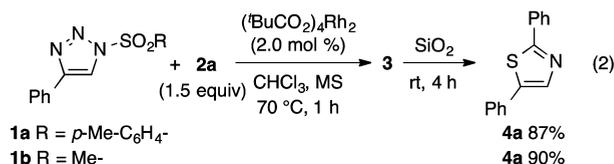
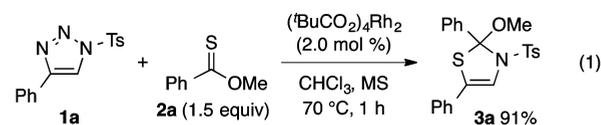


Figure 1. Construction of 2,5-disubstituted thiazoles.

consists of two catalytic reactions; a copper(I)-catalyzed 1,3-dipolar cycloaddition of terminal alkynes with sulfonyl azides⁹ and a rhodium(II)-catalyzed reaction of the resulting 1-sulfonyl-1,2,3-triazoles with thionoesters.

Recently, 1-sulfonyl-1,2,3-triazoles have received much attention as the precursors of α -imino metal carbene complexes.¹⁰ The generated carbene complex possesses an



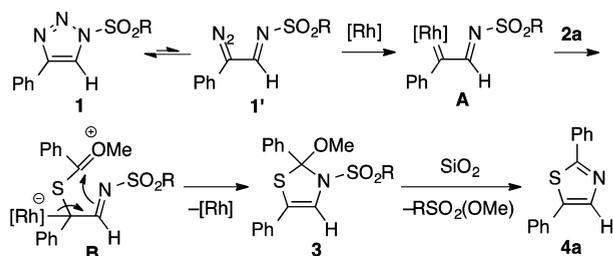
consumed in 1 h, and after chromatographic purification using modified silica gel, 2,5-diphenyl-2-methoxy-3-tosyl-4-thiazoline (**3a**) was obtained in 91% isolated yield. The structure of **3a** was confirmed by its single-crystal X-ray analysis. The thiazoline **3a** was labile under acidic conditions to readily aromatize by elimination of a sulfonyl group. Thus, when acidic silica gel was directly added to the reaction mixture containing **3a**, elimination of methyl 4-methylbenzenesulfonate occurred to afford 2,5-diphenylthiazole (**4a**) in 87% yield based on **1a** (eq 2). 1-Mesy-

Received: April 2, 2015

substituted triazole **1b** also reacted well with **2a** to give a slightly better yield of **4a** (90%).

A plausible mechanism for the production of the thiazole **4a** from 1-sulfonyl-1,2,3-triazole **1** and *O*-methyl benzothioate (**2a**) is depicted in Scheme 1. Initially, a reversible ring-chain

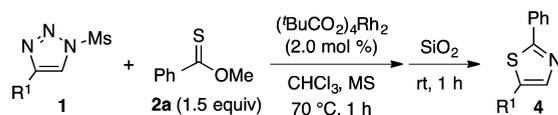
Scheme 1. Proposed Mechanism for the Formation of Thiazole 4a



tautomerization of **1** generates α -diazoimine **1'**, which reacts with rhodium(II) to afford α -imino rhodium carbene complex **A** with extrusion of molecular nitrogen. The sulfur of **2a** attacks the electrophilic carbene center of **A** to furnish zwitterionic intermediate **B**.²⁰ The anionic rhodium releases an electron pair, which induces the addition of the imino nitrogen to the carbon of the oxonium ion, forming a five-membered ring. The resulting 4-thiazoline **3** readily aromatizes upon treatment with acidic silica gel by elimination of methyl sulfonate to give the thiazole **4a**.

A variety of triazoles **1** were subjected to the sequential reaction with *O*-methyl benzothioate (**2a**) (Table 1). Triazoles

Table 1. Rh(II)-Catalyzed Reaction of Various Triazoles **1** with **2a**^a



entry	triazole 1		product 4	yield (%) ^b
	R ¹			
1	<i>p</i> -Me-C ₆ H ₄ -	1c	4b	96
2	<i>p</i> -MeO-C ₆ H ₄ -	1d	4c	84
3	<i>p</i> -CF ₃ -C ₆ H ₄ -	1e	4d	90
4	<i>p</i> -Br-C ₆ H ₄ -	1f	4e	96
5	<i>o</i> -Br-C ₆ H ₄ -	1g	4f	91
6	<i>p</i> -I-C ₆ H ₄ -	1h	4g	87
7	<i>p</i> -(pin)B-C ₆ H ₄ -	1i	4h	67
8	3-Thienyl-	1j	4i	98
9	1-Cyclohexenyl-	1k	4j	87
10	<i>n</i> -Propyl-	1l	4k	49 ^c

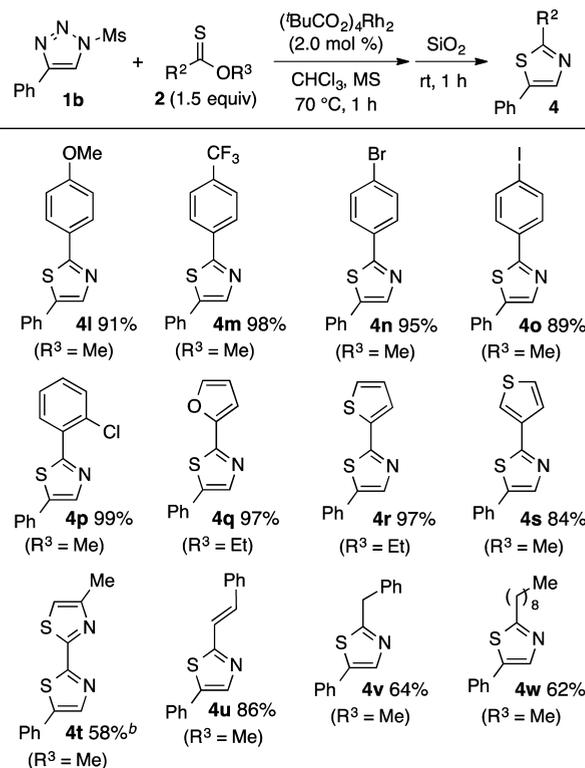
^aOn a 0.20 mmol scale. ^bIsolated yield. ^cTs-substituted triazole **1l** and **2a** (5.0 equiv) were used.

1c–i, possessing an aryl group at the 4-position, afforded the corresponding thiazoles **4b–h** in yields ranging from 67% to 96% (entries 1–7). Notably, the aryl group tolerates a halogen atom and a boron group as the substituents, although the produced halo- and boron-substituted thiazoles are difficult to synthesize with the previously reported palladium-catalyzed C–H arylation reactions of triazoles.⁶ Whereas 1-cyclohexenyl-substituted triazole **1k** successfully participated in the sequential reaction

(entry 9), *n*-propyl-substituted triazole **1l** afforded the product **4k** in 49% yield, probably due to a 1,2-hydride shift occurring with the rhodium carbene complex (entry 10).²¹

A diverse array of thionoesters was also readily prepared from the corresponding carboxylic esters by the reaction with the Lawesson's reagent, and they were reacted with the triazole **1b** (Scheme 2). 2-Aryl-5-phenylthiazoles **4l–t** were obtained in

Scheme 2. Rh(II)-Catalyzed Reaction of **1b** with Various *O*-Alkyl Thioates **2**^a



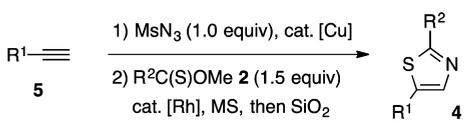
^aOn a 0.20 mmol scale. ^bTsOH·H₂O was used instead of SiO₂.

yields ranging from 58% to 99%. Styryl-substituted thionoester was also effectively converted into the product **4u** in 86% yield. In the case of alkyl-substituted thionoesters, the aromatization process from intermediate 4-thiazolines was slightly more sluggish to afford the products **4v** and **4w** in moderate yields.

The synthetic usefulness of the present reaction was demonstrated by its successful integration into a one-pot procedure which directly started from readily available terminal alkynes **5** (Table 2). First, a solution of **5** (1.0 equiv), mesyl azide (1.0 equiv), and CuTC (5.0 mol %) in chloroform was stirred for 8 h at room temperature, generating 1-mesyl-substituted triazoles **1**. Second, thionoesters **2** (1.5 equiv) and (tBuCO₂)₄Rh₂ (2.0 mol %) were added to the same vessel, which was then heated at 70 °C for 1 h. Finally, acidic silica gel was added to the reaction mixture to promote deprotective aromatization. The corresponding thiazoles **4** were obtained in overall yields ranging from 80% to 85%. Thus, the rhodium-catalyzed annulation reaction in the second step was barely interrupted by the copper catalyst employed in the first step. This one-pot procedure saves a significant amount of time and solvents for a workup/purification procedure.²²

The one-pot procedure was applied to steroidal substrate **6a** and δ -tocopherol-derived substrate **6b** (eqs 3 and 4). The

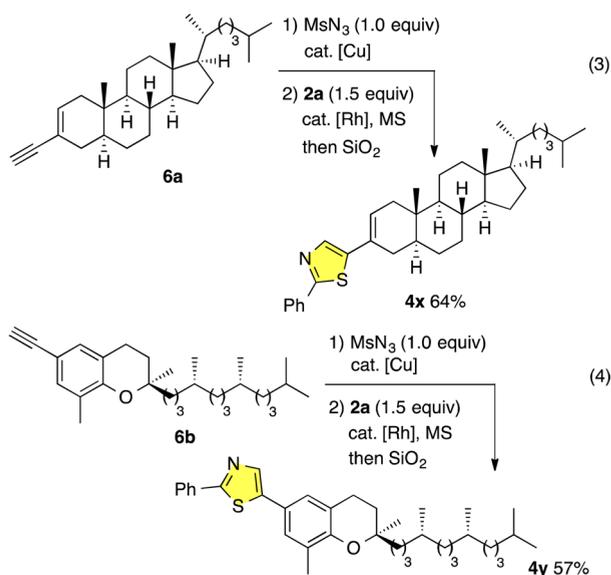
Table 2. One-Pot Synthesis of Thiazoles 4 Starting from Terminal Alkynes^a



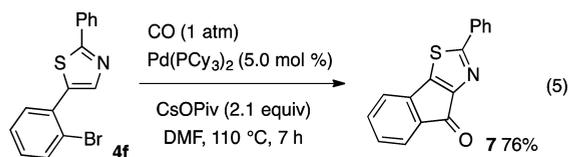
entry	R ¹	5	R ²	4	yield (%) ^b
1	Ph-	5a	Ph-	4a	80
2	<i>p</i> -MeO-C ₆ H ₄ -	5b	Ph-	4c	82
3	3-Thienyl-	5c	Ph-	4i	82
4	Ph-	5a	<i>p</i> -MeO-C ₆ H ₄ -	4l	85
5	Ph-	5a	<i>p</i> -Br-C ₆ H ₄ -	4n	83

^aOn a 0.20 mmol scale. ^bIsolated yield.

corresponding thiazolyl-substituted derivatives **4x** and **4y** were successfully obtained in 64% and 57% overall yields based on the starting terminal alkynes **6a** and **6b**.

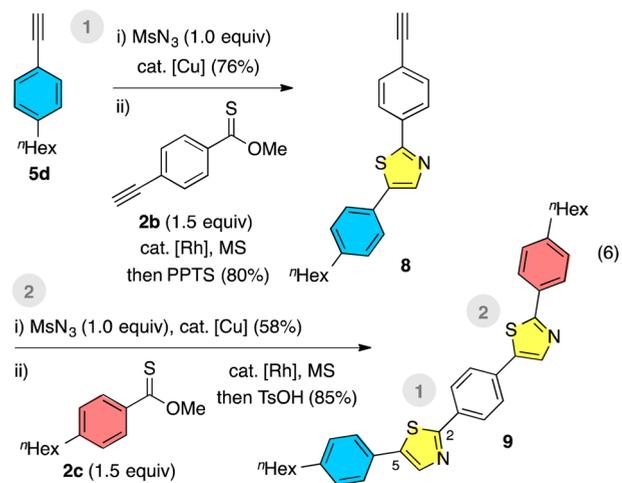


Derivatization of the obtained thiazole was exemplified in eq 5. When the palladium(0)-catalyzed conditions developed by

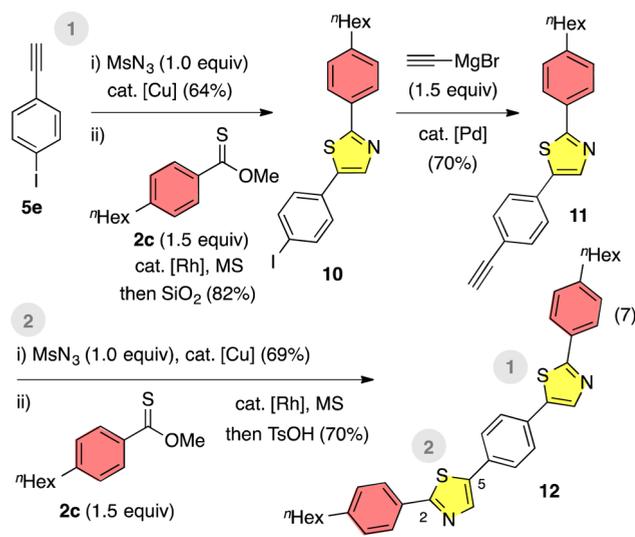


Larock²³ were applied to *o*-bromophenyl-substituted thiazole **4f**, the carbonylative cyclization reaction took place to give 4*H*-indeno[2,1-*d*]thiazole-4-one **7**.

The terminal alkyne-based thiazole synthesis was further extended to an iterative procedure for the synthesis of linear oligomeric arylene compounds. For example, the ethynyl-substituted benzothioate **2b** presents a useful building block for the iterative procedure (eq 6). The first thiazole formation from **5d** and **2b** was carried out in a stepwise manner to furnish ter(arylene) **8** possessing a terminal ethynyl group. Next, the terminal ethynyl group of the ter(arene) **8** was utilized for the second thiazole formation with 4-hexylbenzothioate **2c** to produce the quinque(arylene) **9** consisting of two thiazole and three benzene rings.

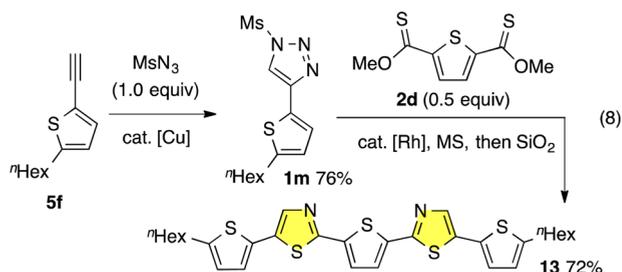


Quinque(arylene) **12** having a different array of two thiazole and three benzene rings was also synthesized (eq 7). Initially, 5-

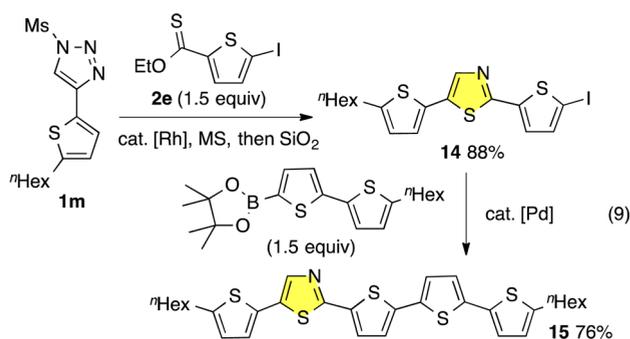


(4-iodophenyl)thiazole **10** was prepared from **5e** and **2c**, and then a terminal ethynyl group was introduced on the phenyl ring by a palladium-catalyzed coupling reaction,²⁴ forming **11**. The second thiazole formation was carried out using **11** and **2c** to produce the quinque(arylene) **12**.

Quinque(thiophene/thiazole) oligomers could be efficiently synthesized based on the present thiazole synthesis. The triazole **1m** was prepared from 2-ethynylthiophene **5f** and mesyl azide. Then, thiophene-2,5-bis(carbothioate) **2d** (0.1 mmol) was reacted with **1m** (0.2 mmol). Double annulation took place to afford symmetrical quinque(thiophene/thiazole) oligomer **13** in 72% yield (eq 8).



Unsymmetrical quinque(thiophene/thiazole) oligomer **15** was constructed from the same triazole **1m** (eq 9). Initially,



the thiazole formation from **1m** and 5-iodothiophene-2-carbothioate **2e** was carried out to furnish iodo-substituted ter(thiophene/thiazole) oligomer **14**. Then, boryl-substituted bi(thiophene) was reacted with **14** in the presence of a palladium catalyst to produce **15** in 76% yield.²⁵

In summary, we have demonstrated that thionoesters can act as the dipolarophiles toward α -imino rhodium(II) carbene complexes and developed a useful method for the synthesis of 2,5-disubstituted thiazoles starting from terminal alkynes. This procedure was successfully applied to late-stage transformation of biorelated derivatives and highly selective synthesis of oligomeric arylene compounds.

ASSOCIATED CONTENT

Supporting Information

Experimental procedures, spectral data for the new compounds, and details of the X-ray analysis (CIF). The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.5b00960.

AUTHOR INFORMATION

Corresponding Authors

*E-mail: tmiura@sbchem.kyoto-u.ac.jp.

*E-mail: murakami@sbchem.kyoto-u.ac.jp.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

We thank Dr. Y. Nagata (Kyoto University) for his kind help in the X-ray analysis. This work was supported by MEXT (Scientific Research (B)) and JST (ACT-C). Y. Funakoshi thanks the JSPS for Yong Scientists for a Research Fellowship.

REFERENCES

- (1) For reviews, see: (a) Lewis, J. R. *Nat. Prod. Rep.* **2000**, *17*, 57. (b) Jin, Z. *Nat. Prod. Rep.* **2003**, *20*, 584. (c) Davy, D.; Serra, G. *Mar. Drugs* **2010**, *8*, 2755. (d) Baumann, M.; Baxendale, I. R.; Ley, S. V.; Nikbin, N. *Beilstein J. Org. Chem.* **2011**, *7*, 442. For selected examples of the total synthesis of thiazole alkaloids, see: (e) Müller, H. M.; Delgado, O.; Bach, T. *Angew. Chem., Int. Ed.* **2007**, *46*, 4771. (f) Schotes, C.; Ostrovskiy, D.; Senger, J.; Schmidt-kunz, K.; Jung, M.; Breit, B. *Chem.—Eur. J.* **2014**, *20*, 2164.
- (2) (a) Mori, A.; Sugie, A. *Bull. Chem. Soc. Jpn.* **2008**, *81*, 548. (b) Murai, T.; Hori, F.; Maruyama, T. *Org. Lett.* **2011**, *13*, 1718. (c) Tao, T.; Ma, B.-B.; Peng, Y.-X.; Wang, X.-X.; Huang, W.; You, X.-Z. *J. Org. Chem.* **2013**, *78*, 8669.
- (3) Ando, S.; Murakami, R.; Nishida, J.; Tada, H.; Inoue, Y.; Tokito, S.; Yamashita, Y. *J. Am. Chem. Soc.* **2005**, *127*, 14996.

(4) For reviews, see: (a) Riego, E.; Hernández, D.; Albericio, F.; Álvarez, M. *Synthesis* **2005**, 1907. (b) Kempson, J. In *Name Reactions in Heterocyclic Chemistry II*; Li, J. J., Ed.; Wiley: Hoboken, 2011; Chapter 5.4, pp 299–308.

(5) (a) Aitken, K. M.; Aitken, R. A. *Tetrahedron* **2008**, *64*, 4384. (b) St. Denis, J. D.; Zajdlík, A.; Tan, J.; Trinchera, P.; Lee, C. F.; He, Z.; Adachi, S.; Sudan, A. K. *J. Am. Chem. Soc.* **2014**, *136*, 17669.

(6) For recent papers, see: (a) Mori, A.; Sekiguchi, A.; Masui, K.; Shimada, T.; Horie, M.; Osakada, K.; Kawamoto, M.; Ikeda, T. *J. Am. Chem. Soc.* **2003**, *125*, 1700. (b) Roger, J.; Pozgan, F.; Doucet, H. *J. Org. Chem.* **2009**, *74*, 1179. (c) Shibahara, F.; Yamauchi, T.; Yamaguchi, E.; Murai, T. *J. Org. Chem.* **2012**, *77*, 8815. (d) Liu, X.-W.; Shi, J.-L.; Yan, J.-X.; Wei, J.-B.; Peng, K.; Dai, L.; Li, C.-G.; Wang, B.-Q.; Shi, Z.-J. *Org. Lett.* **2013**, *15*, 5774. (e) Tani, S.; Uehara, T. N.; Yamaguchi, J.; Itami, K. *Chem. Sci.* **2014**, *5*, 123 and references cited therein.

(7) Sheldrake, P. W.; Matteucci, M.; McDonald, E. *Synlett* **2006**, 460.

(8) For the synthesis of 2,4-disubstituted thiazoles from terminal alkynes and thioamides using a gold catalyst, see: Wu, G.; Zheng, R.; Nelson, J.; Zhang, L. *Adv. Synth. Catal.* **2014**, *356*, 1229.

(9) Raushel, J.; Fokin, V. V. *Org. Lett.* **2010**, *12*, 4952.

(10) For reviews, see: (a) Gulevich, A. V.; Gevorgyan, V. *Angew. Chem., Int. Ed.* **2013**, *52*, 1371. (b) Davies, H. M. L.; Alford, J. S. *Chem. Soc. Rev.* **2014**, *43*, 5151. (c) Anbarasan, P.; Yadagiri, D.; Rajasekar, S. *Synthesis* **2014**, *46*, 3004.

(11) (a) Miura, T.; Yamauchi, M.; Murakami, M. *Chem. Commun.* **2009**, 1470. (b) Chattopadhyay, B.; Gevorgyan, V. *Org. Lett.* **2011**, *13*, 3746. (c) Shi, Y.; Gevorgyan, V. *Org. Lett.* **2013**, *15*, 5394.

(12) (a) Schultz, E. E.; Sarpong, R. *J. Am. Chem. Soc.* **2013**, *135*, 4696. (b) Miura, T.; Hiraga, K.; Biyajima, T.; Nakamuro, T.; Murakami, M. *Org. Lett.* **2013**, *15*, 3298.

(13) Horneff, T.; Chuprakov, S.; Chernyak, N.; Gevorgyan, V.; Fokin, V. V. *J. Am. Chem. Soc.* **2008**, *130*, 14972.

(14) Zibinsky, M.; Fokin, V. V. *Angew. Chem., Int. Ed.* **2013**, *52*, 1507.

(15) Chuprakov, S.; Kwok, S. W.; Fokin, V. V. *J. Am. Chem. Soc.* **2013**, *135*, 4652.

(16) Spangler, J. E.; Davies, H. M. L. *J. Am. Chem. Soc.* **2013**, *135*, 6802.

(17) For [4 + 2] cycloaddition reactions of thiocarbonyl compounds with 1,3-dienes, see: (a) Weinreb, S. M.; Staib, R. R. *Tetrahedron* **1982**, *38*, 3087. (b) Timoshenko, V. M.; Siry, S. A.; Rozhenko, A. B.; Shermolovich, Y. G. *J. Fluorine Chem.* **2010**, *131*, 172.

(18) Scheibye, S.; Kristensen, J.; Lawesson, S.-O. *Tetrahedron* **1979**, *35*, 1339.

(19) A trace amount of hydration product was formed in the absence of MS, even if we used freshly distilled chloroform. For rhodium-catalyzed hydration of triazoles, see: Miura, T.; Biyajima, T.; Fujii, T.; Murakami, M. *J. Am. Chem. Soc.* **2012**, *134*, 194.

(20) For the reaction of thionoesters and thioamides with rhodium(II) carbene complexes, see: (a) Takano, S.; Tomita, S.; Takahashi, M.; Ogasawara, K. *Synthesis* **1987**, 1116. (b) Honda, T.; Ishige, H.; Araki, J.; Akimoto, S.; Hirayama, K.; Tsubuki, M. *Tetrahedron* **1992**, *48*, 79. (c) Shi, B.; Blake, A. J.; Lewis, W.; Campbell, I. B.; Judkins, B. D.; Moody, C. J. *J. Org. Chem.* **2010**, *75*, 152.

(21) (a) Miura, T.; Funakoshi, Y.; Morimoto, M.; Biyajima, T.; Murakami, M. *J. Am. Chem. Soc.* **2012**, *134*, 17440. (b) Selander, N.; Worrell, B. T.; Fokin, V. V. *Angew. Chem., Int. Ed.* **2012**, *51*, 13054.

(22) For reviews on sequential multistep catalytic processes, see: (a) Ambrosini, L. M.; Lambert, T. H. *ChemCatChem* **2010**, *2*, 1373. (b) Rueping, M.; Koenigs, R. M.; Atodiresi, I. *Chem.—Eur. J.* **2010**, *16*, 9350. (c) Sadig, J. E. R.; Willis, M. C. *Synthesis* **2011**, 1.

(23) Campo, M. A.; Larock, R. C. *Org. Lett.* **2000**, *2*, 3675.

(24) Negishi, E.; Kotori, M.; Xu, C. *J. Org. Chem.* **1997**, *62*, 8957.

(25) Ashizawa, M.; Niimura, T.; Yu, Y.; Tsuboi, K.; Matsumoto, H.; Yamada, R.; Kawachi, S.; Tanioka, A.; Mori, T. *Tetrahedron* **2012**, *68*, 2790.