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### HETEROCYCLIC SYNTHESSES WITH THIOLES AND NITRILES: SYNTHESIS OF SOME NEW PYRIMIDO[4',5':4,5] THIAZOLO[3,2-a], THIAZOLO[3,2-a] AND TRIAZOLO[3,2-a]- BENZIMIDAZOLE DERIVATIVES

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# HETEROCYCLIC SYNTHESSES WITH THIOLES AND NITRILES: SYNTHESIS OF SOME NEW PYRIMIDO[4',5':4,5] THIAZOLO[3,2-*a*], THIAZOLO[3,2-*a*] AND TRIAZOLO[3,2-*a*]-BENZIMIDAZOLE DERIVATIVES

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Treatment of 6-benzoylbenzimidazole-2-thiol (**1**) and 3-amino-6-benzoylthiazolo[3,2-*a*]-benzimidazole-2-carbonitrile (**3**) with various reagents under different conditions is reported to afford cyclized and uncyclized compounds of potential pharmacological interest.

**Keywords:** 4-chloroacetylantipyrine; 2-bromodimedone; hydrazoneyl chloride

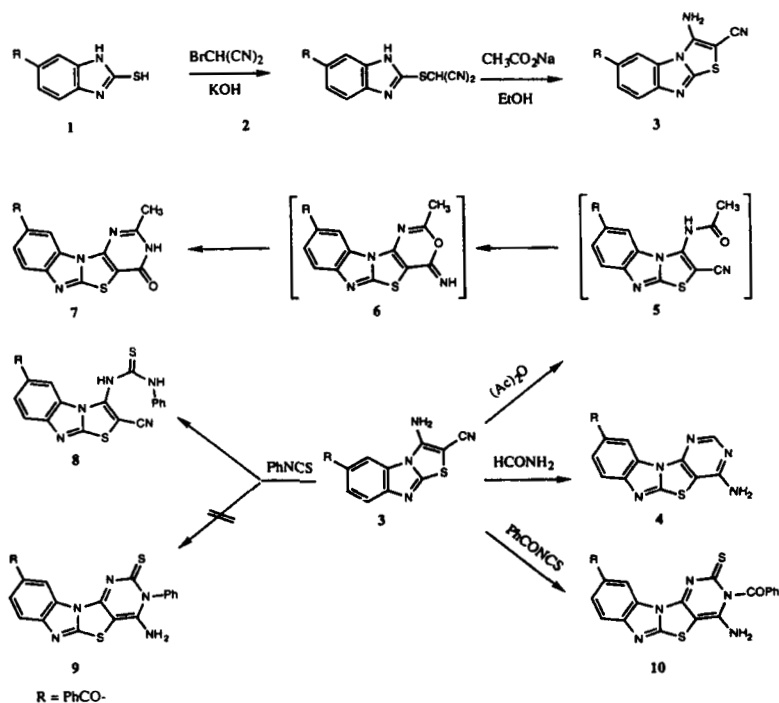
Thiazolo[3,2-*a*]benzimidazole derivatives exhibit antibacterial activity<sup>1</sup> and act as hypoglycemic agents.<sup>2</sup> Their biological properties include anti-tumor,<sup>3</sup> antiviral,<sup>4</sup> antitubercular,<sup>5</sup> and anticonvulsant activity.<sup>6</sup> They have also been employed as fungicidal,<sup>7</sup> insecticide,<sup>8</sup> photographic sensitizer<sup>9</sup> and as chromophoric units in cyanine dyes.<sup>10</sup>

The versatile benefits and connection with previous efforts directed towards the facile synthesis of heterocyclic ring system,<sup>11-13</sup> and interest in the reaction of benzimidazole derivatives prompted us to investigate the reaction of 6-benzoyl-2-thiolobenzimidazole (**1**) with bromomalononitrile. This reaction yielded compound **2**, which was refluxed with anhydrous sodium acetate in ethanol to afford 3-amino-6-benzoylthiazolo[3,2-*a*]benzimidazole-2-carbonitrile (**3**). Its structure was confirmed by elemental analysis and spectral data. IR showed absorption bands at  $\nu$  3395,

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3259  $\text{cm}^{-1}$  characteristic for ( $\text{NH}_2$ ), a sharp cyano absorption band at  $\nu$  2200 and at  $\nu$  1654  $\text{cm}^{-1}$  due to (CO) vibration. The  $^1\text{H-NMR}$  spectrum revealed a multiplet at  $\delta$  7.33–8.21 attributed to  $\text{NH}_2$  and the aromatic protons. Mass spectra revealed the molecular ion peak  $m/z = 318$  (48%) compatible with the molecular formula  $\text{C}_{17}\text{H}_{10}\text{N}_4\text{OS}$ .

The reaction of compound **3** with boiling formamide gave 9-benzoylpyrimido[4',5':4,5]thiazolo[3,2-*a*]benzimidazole-4-amine (**4**). Its structure is based on elemental analysis and spectral data. Compound **3** was refluxed with acetic anhydride to afford compound **7**,  $^{14}$  via *N*-acetylation to yield **5** which apparently cyclized into **6** which then in turn rearranged to **7**. The IR spectrum of compound **7** showed absorption bands at 3349, 2926, 1669 and 1654  $\text{cm}^{-1}$  assignable to  $\text{NH}$ ,  $\text{CH}_3$ ,  $2\text{CO}$  group respectively. The  $^1\text{H-NMR}$  spectrum revealed a singlet at  $\delta$  2.85 corresponding to (3 H) of  $\text{CH}_3$  besides a multiplet at  $\delta$  7.40–8.80 ppm, (9 H) attributed to the aromatic and  $\text{NH}$  protons.

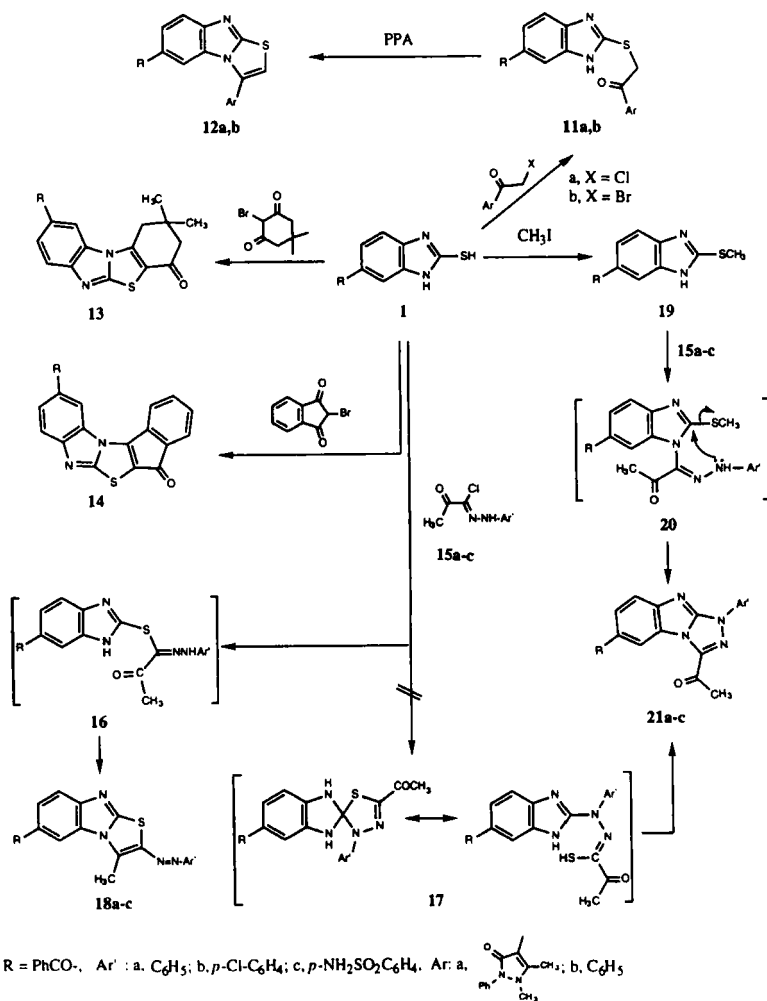


SCHEME 1

Compound **3** reacted with phenyl isothiocyanate to afford a product whose IR spectrum showed a cyano absorption band at  $\nu$  2198  $\text{cm}^{-1}$ . The anticipated structure of **9** was therefore excluded and **8** was assigned as being a thiourea derivative. The reaction of **3** with benzoyl isothiocyanate led to the cyclized product **10**. The IR spectrum of **10** did not show any absorption bands that can be attributed to the presence of the cyano group. The cyclization in the second case may be enhanced by the presence of a carbonyl group and may be inhibited in the first case due to stereochemical aspects.

Compound **1** underwent cyclocondensation when refluxed with 4-chloroacetylantipyrine<sup>15</sup> and phenacyl bromide in absolute ethanol to yield the ketones **11a, b** which on treatment with PPA, underwent a cyclo-dehydration, to give a single compound (TLC) in each case, 7-benzoyl-1,3-thiazolo[3,2-*a*]benzimidazole derivatives **12a, b**. Structures **11** and **12** were established based on elemental analyses and spectroscopic studies. Thus, the mass spectrum of **12a** revealed a molecular ion peak  $m/z = 464$  (16%) corresponding to the molecular formula  $\text{C}_{27}\text{H}_{20}\text{N}_4\text{O}_2\text{S}$ . Its  $^1\text{H-NMR}$  spectrum showed a singlet at  $\delta$  2.31 and 3.43 for the antipyrine  $\text{CH}_3$  and  $\text{NCH}_3$ , respectively and a multiplet at  $\delta$  6.8–7.84 attributed to the SCH protons and the aromatic protons. Similarly, compound **1** reacted with 2-bromodimedone<sup>16</sup> in absolute ethanol to afford 4-benzoyl-7,9-dihydro-8,8-dimethylbenzo[3',2':4,5]thiazolo[3,2-*a*]benzimidazole-10(H)-one (**13**) in a yield of 60%. Also, compound **1** reacted with indan-1,3-dione in the presence of *N*-bromosuccinimide to furnish compound **14** in a yield of 55%. Analytical and spectral data are consistent with structure **13** and **14** (Scheme 2, Tables I and II).

When compound **1** reacted with the hydrazonyl chloride derivatives **15a-c**<sup>17</sup> in ethanolic sodium ethoxide solution products identified as 2-arylozo-7-benzoyl-1-methyl-1,3-thiazolo[3,2-*a*]benzimidazole **18a-c** or 1-acetyl-3H-3-aryl-7-benzoyl-1,2,4-triazolo-[4,3-*a*]benzimidazole **21a-c** were formed. Structure **21** resulted *via* elimination of  $\text{H}_2\text{S}$  from the corresponding compound **17** which was obtained by a 1,3-dipolar cycloaddition of nitrilimine to the double bond of the imidazole ring. The formation of compound **18** is explained by a stepwise reaction involving substitution to give the acyclic hydrazone **16**. Cyclization of the latter is completed by elimination of water. The structures **18a-c** were deduced from their ele-



SCHEME 2

mental analysis and spectral evidence. The IR spectrum of these compounds showed the absence of a NH absorption band and its mass spectra revealed the molecular ion peak compatible with the assigned structure. The  $^1\text{H-NMR}$  spectrum revealed besides the expected signals the absence of a singlet corresponding to the NH group (Table II).

TABLE I Physical and analytical data of the newly compounds

Comp No.	MP°C Solvent	Yield %	MF M.wt.	Analysis% (Calcd./Found)			
				C	H	N	S
<b>1</b>	260	80	C <sub>14</sub> H <sub>10</sub> N <sub>2</sub> O <sub>4</sub> S	66.12	3.96	11.01	12.60
	EtOH		(254.312)	66.20	3.90	11.30	12.80
<b>2</b>	172	65	C <sub>17</sub> H <sub>10</sub> N <sub>4</sub> OS	64.14	3.16	17.59	10.07
	EtOH		(318.34)	64.20	3.20	17.60	10.07
<b>3</b>	245	80	C <sub>17</sub> H <sub>10</sub> N <sub>4</sub> OS	64.14	3.16	17.59	10.07
	DMF		(318.34)	64.20	3.20	17.60	10.80
<b>4</b>	> 300	71	C <sub>18</sub> H <sub>11</sub> N <sub>5</sub> OS	62.59	3.20	20.27	9.28
	EtOH		(345.378)	62.80	3.30	20.20	10.00
<b>7</b>	> 300	48	C <sub>19</sub> H <sub>12</sub> N <sub>4</sub> O <sub>2</sub> S	63.32	3.35	15.54	8.89
	AcOH		(360.39)	63.60	3.40	15.90	8.80
<b>8</b>	219	75	C <sub>24</sub> H <sub>15</sub> N <sub>5</sub> OS <sub>2</sub>	63.55	3.31	15.44	14.12
	EtOH		(453.548)	63.40	3.30	15.90	14.50
<b>10</b>	308	78	C <sub>25</sub> H <sub>15</sub> N <sub>5</sub> O <sub>2</sub> S <sub>2</sub>	62.35	3.13	14.45	13.31
	EtOH		(481.546)	62.60	3.10	14.70	13.70
<b>11a</b>	221	72	C <sub>27</sub> H <sub>22</sub> N <sub>4</sub> O <sub>3</sub> S	67.20	4.59	11.61	6.64
	EtOH		(482.558)	67.30	4.80	11.20	6.80
<b>11b</b>	195	76	C <sub>22</sub> H <sub>16</sub> N <sub>2</sub> O <sub>2</sub> S	70.94	4.33	7.52	8.60
	EtOH		(372.44)	70.50	4.30	7.50	8.70
<b>12a</b>	292	40	C <sub>27</sub> H <sub>20</sub> N <sub>4</sub> O <sub>2</sub> S	69.80	4.33	12.06	6.90
	EtOH/H <sub>2</sub> O		(464.547)	69.30	4.40	12.10	6.40
<b>12b</b>	263	76	C <sub>22</sub> H <sub>14</sub> N <sub>2</sub> OS	74.55	3.98	7.90	9.04
	EtOH		(354.425)	74.50	4.00	8.10	9.10
<b>13</b>	206	68	C <sub>22</sub> H <sub>18</sub> N <sub>2</sub> O <sub>2</sub> S	70.54	4.84	7.47	8.56
	EtOH		(374.59)	70.70	4.80	7.50	8.50
<b>14</b>	> 300	70	C <sub>23</sub> H <sub>12</sub> N <sub>2</sub> O <sub>2</sub> S	72.61	3.17	7.36	8.42
	DMF		(380.42)	73.00	3.10	7.40	8.30
<b>18a</b>	(190–5)	63	C <sub>23</sub> H <sub>16</sub> N <sub>4</sub> OS	69.67	4.06	14.13	8.08

Comp No.	MP°C Solvent	Yield %	MF M.wt.	Analysis% (Calcd./Found)			
				C	H	N	S
	EtOH		(396.47)	69.60	4.10	14.20	8.20
<b>18b</b>	240	52	C <sub>23</sub> H <sub>15</sub> N <sub>4</sub> OSCl	64.10	3.50	13.00	7.44
	EtOH		(430.96)	64.00	3.50	13.10	7.80
<b>18c</b>	245	42	C <sub>23</sub> H <sub>17</sub> N <sub>5</sub> O <sub>3</sub> S <sub>2</sub>	58.09	3.60	14.72	13.48
	EtOH		(475.55)	58.10	3.60	14.70	13.49
<b>21a</b>	235	68	C <sub>23</sub> H <sub>16</sub> N <sub>4</sub> O <sub>2</sub>	72.62	4.24	14.72	
	DMF		(380.40)	72.80	4.30	14.90	
<b>21b</b>	200	62	C <sub>23</sub> H <sub>15</sub> N <sub>4</sub> O <sub>2</sub> Cl	66.58	3.64	13.56	
	EtOH		(414.89)	66.60	3.70	13.60	
<b>21c</b>	291	42	C <sub>23</sub> H <sub>17</sub> N <sub>5</sub> O <sub>4</sub> S	60.12	3.72	15.24	6.97
	EtOH/DMF		(459.48)	60.30	3.80	15.40	6.60

On the other hand, 6-benzoyl-2-methylsulfonylbenzimidazole (**19**) reacted with hydrazonyl chloride in absolute ethanol in the presence of TEA to afford **21a-c**. These latter products resulted *via* elimination of methanthiole from the corresponding acyclic adduct **20**. The structures of **21a-c** were inferred from their elemental analyses and spectral data. Thus, their <sup>1</sup>H-NMR spectra revealed besides the expected signals, no signals due to either (SCH<sub>3</sub>) or (NH) protons (Tables I and II).

## EXPERIMENTAL

Melting points were recorded on a Gallenkamp melting point apparatus and are uncorrected. IR spectra were recorded as KBr pellets with a FTIR-8201 PC spectrophotometer (Shimadzu). <sup>1</sup>H NMR spectra were obtained on a Varian Gemini 200 MHz spectrometer in DMSO-*d*<sub>6</sub> or CDCl<sub>3</sub> as solvent and TMS as an internal reference. Mass spectra were performed on a Shimadzu GCMS-QP 1000 Ex at 70 ev. Microanalysis were performed by the Microanalytical center University of Cairo. 4-Chloroacetylantipyrine<sup>15</sup> and hydrazonyl chloride derivatives **15a-c**<sup>17</sup> were prepared according to reported procedures.



TABLE II Spectral data of the newly compounds

Compd. No.	IR ( $\nu$ $\text{cm}^{-1}$ ) Selected Bands	$^1\text{H}$ NMR ( $\delta$ )	MS $m/z$ (%)
<b>1</b>	3410 (NH), 1654 (CO)		254 (100), 177 (60), 149 (30)
<b>2</b>	3421 (NH), 2191, 2142 (2 CN), 1654 (CO)		318 (48), 241 (46), 255 (100), 177 (95), 105 (68), 77 (79)
<b>3</b>	3395, 3359 ( $\text{NH}_2$ ), 2200 (CN), 1654 (CO)	7.33–8.21 (m, 10H, Ar-H + $\text{NH}_2$ )	
<b>4</b>	3330–3280 ( $\text{NH}_2$ ), 1666 (CO)	7.10–7.89 (m, 10 H, Ar-H + $\text{NH}_2$ ), 8.2 (s, 1H, N=CH-N-)	346 (17), 318 (42), 290 (13), 254 (35), 177 (100)
<b>7</b>	3349 (NH), 2926 ( $\text{CH}_3$ ), 1669, 1954 (2CO)	2.85 (s, 3H, $\text{CH}_3$ ), 7.4–8.8 (m, 9H, Ar-H + N-H)	360 (18), 242 (40), 310 (10), 177 (85), 105 (100), 77 (50)
<b>8</b>	3400, 3250 (2NH), 2198 (CN), 1656 (CO)	6.56 (s, 1H, NH), 6.81 (s, 1H, NH), 7.01–8.10 (m, 13H, Ar-H)	
<b>10</b>	3392, 3139 ( $\text{NH}_2$ ), 1715, 1656 (2CO)	6.92–7.90 (m, 13H, Ar-H + $\text{NH}_2$ )	481 (13), 366 (2.6), 254 (100), 178 (8.5), 144 (18.5)
<b>11a</b>	3392 (NH), 2986 (-CH-aliph), 1679 (CO), 1650 (CO)		483 (19), 482 (16.7), 255 (92), 215 (58), 177 (100)
<b>11b</b>	3330 (NH), 1690 (CO), 1653 (CO)		
<b>12a</b>	2980 (-CH-aliph), 1669, 1653 (2CO)	2.31 (s, 3H, $\text{CH}_3$ ), 3.43 (s, 3H, -N- $\text{CH}_3$ ), 6.8–7.84 (m, 14H, Ar-H + S-CH=)	464 (16), 414 (40), 254 (100), 228 (10), 177 (51)
<b>12b</b>	1654 (CO), 1630 (C=N)		

<i>Compd. No.</i>	<i>IR (<math>\nu</math> cm<sup>-1</sup>) Selected Bands</i>	<i><sup>1</sup>H NMR (<math>\delta</math>)</i>	<i>MS <math>m/z</math> (%)</i>
<b>13</b>	2982 (-CH-aliph), 1675 (CO), 1653 (CO)	1.03 (s, 3H, CH <sub>3</sub> ), 1.09 (s, 3H, CH <sub>3</sub> ), 2.20 (s, 2H, CH <sub>2</sub> ), 2.40 (s, 2H, CH <sub>2</sub> ), 7.20–7.98 (m, 8H, Ar-H)	374 (8), 360 (100), 310 (35), 263 (54.5), 105 (27), 77 (37.5)
<b>14</b>	1670 (CO), 1650 (CO)		380 (9), 367 (26), 366 (100), 289 (44), 77 (25)
<b>18a</b>	1654 (CO), 1635 (C=N)		396 (11.7), 313 (77), 177 (100), 116 (14), 77 (25)
<b>18b</b>	1649 (CO), 1635 (C=N)		
<b>18c</b>	3230 (SO <sub>2</sub> NH <sub>2</sub> ), 1654 (CO)	2.13 (s, 3H, CH <sub>3</sub> ), 7.4–8.2 (m, 4H, Ar-H+NH <sub>2</sub> )	
<b>21a</b>	1698, 1654 (2CO), 1637 (C=N)	2.7 (s, 3H, CH <sub>3</sub> ), 7.2–8.1 (m, 13H, Ar-H)	
<b>21b</b>	1693, 1651 (2CO), 1630 (C=N)	2.79 (s, 3H, CH <sub>3</sub> ), 7.3–8.30 (m, 12H, Ar-H)	416 (20), 415 (15.7), 414 (58.5), 337 (100), 232 (18), 77 (50)
<b>21c</b>	1701, 1650 (2CO), 1630 C=N		

**(6-Benzoylbenzimidazol-2-ylsulphamyl)malononitrile (2)**

To an aqueous solution of 6-benzoylbenzimidazol-2-thiol (1) (2.54 g, 0.01 mol) and KOH (0.56 g; 0.01 mol), an ethanolic solution of bromomalononitrile (1.45 g, 0.01 mol) was added dropwise with stirring for 2 h at room temperature. The resulting precipitate was collected by filtration and crystallized from ethanol.

**3-Amino-6-benzoylthiazolo[3,2-*a*]benzimidazole-2-carbonitrile (3)**

Compound **2** (3.18 g; 0.01 mol) was refluxed in absolute ethanol in the presence of anhydrous sodium acetate (1.64 g; 0.02 mol) for 5 h. The resulting precipitate was separated by filtration, washed several times with water and recrystallized from DMF.

**9-Benzoylpyrimido[4',5':4,5]thiazolo[3,2-*a*]benzimidazole-4-amine (4)**

Compound **3** (1.59 g; 0.005 mol) was refluxed in formamide (10 ml) at 150°C for 5 h. After cooling, the reaction mixture was poured onto cold water and the solid formed was collected and recrystallized from ethanol.

**9-Benzoyl-2-methylpyrimido[4',5':4,5]thiazolo[3,2-*a*]benzimidazol-4(3H)-one (7)**

Compound **3** (1.59g; 0.005 mol) was refluxed in a mixture of Ac<sub>2</sub>O/AcOH (1:1) (15 ml) for 6 h, after cooling the obtained solid was collected and recrystallized from acetic acid.

***N*-Phenyl-*N'*-(6-benzoyl-2-cyanothiazolo[3,2-*a*]benzimidazol-3-yl)thiourea (8)**

To a solution of **3** (1.59 g; 0.005 mol) in dry acetone (20 ml) was added phenyl isothiocyanate (0.68 g; 0.005 mol), and the reaction mixture was refluxed for 2 h. The resulting precipitate was filtered off and recrystallized from ethanol.

**4-Amino-3,9-dibenzoylpyrimido[4',5':4,5]thiazolo[3,2-a]benzimidazol-2-thione (10)**

To a solution of (0.82 g; 0.005 mol) of benzoyl isothiocyanate (prepared in situ from benzoyl chloride and ammonium thiocyanate) in dry acetone was added **3** (1.59 g; 0.005 mol) and the reaction mixture was refluxed for 2 h. On cooling, a solid product was separated, filtered off, and recrystallized from ethanol.

**7-Benzoyl-1-(2',3'-dimethyl-5'-oxo-1'-phenylpyrazol-4'-yl)thiazolo[3,2-a]benzimidazole (12a): Prepared by a two step reaction****Step I**

A mixture of compound **1** (1.27 g; 0.005 mol) and 4-chloroacetyl antipyrine (1.33 g; 0.005 mol) in ethanol (50 ml) was refluxed for 6 h. The reaction mixture was cooled and neutralized by Na<sub>2</sub>CO<sub>3</sub> solution 5%. The obtained solid was collected and recrystallized from ethanol to give compound **11a**.

**Step II**

To polyphosphoric acid (10 ml) at room temperature, compound **11a** (2.4 g) was added. The reaction mixture was heated to 180°C and refluxed for 1 h. After the usual work up, the solid formed was separated, washed with water and recrystallized from ethanol/ H<sub>2</sub>O to give **12a**.

**7-Benzoyl-1-phenylthiazolo[3,2-a]benzimidazole (12b)**

It was prepared by the same manner as described in preparation of **12a** except that the phenacyl bromide was used instead of 4-chloroacetyl antipyrine.

**4-Benzoyl-7,9-dihydro-8,8-dimethylbenzo[3',2':4,5]thiazolo[3,2-a]benzimidazole-11(H)-one (13)**

A mixture of compound **1** (1.27 g; 0.005 mol) and 2-bromodimedone (1.09 g; 0.005 mol) in absolute ethanol (30 ml) was heated under reflux for 5 h. The reaction mixture was concentrated, cooled to room temperature

and neutralized by a 5% aqueous  $\text{Na}_2\text{CO}_3$  solution. The separated solid was recrystallized from ethanol and colorless needles were obtained.

#### **4-Benzoylindeno[3',2':4,5]thiazolo[3,2-a]benzimidazole-11(H)-one (14)**

A mixture of indan-1,3-dione (0.73 g; 0.005 mol), N-bromosuccinimide (1.78 g; 0.01 mol) in carbon tetrachloride (30 ml) was refluxed for about 2 h. The reaction mixture was filtered and the solvent distilled off. The residue was further refluxed with compound **1** (1.27 g; 0.005 mol) in absolute ethanol for 8 h. The solid which separated at the end of the refluxing period was filtered, dissolved in boiling water and neutralized with sodium carbonate solution to give a pale yellow precipitate which was filtered off and recrystallized from DMF.

#### **7-Benzoylthiazolo[3,2-a]benzimidazole derivatives (18a-c):**

##### ***General Procedure***

A mixture of compound **1** (1.27 g; 0.005 mol) and the appropriate hydrazoneyl chloride **15a-c** (0.005 mol) in ethanol/ sodium ethoxide solution was refluxed for 6 h. The solvent was evaporated and the residue was triturated with methanol. The solid obtained was collected and crystallized from ethanol.

#### **7-Benzoyltriazolo[4,3-a]benzimidazole derivatives (21a-c)**

To a solution of 2-methylsulfonylbenzimidazole (1.34 g; 0.005 mol) and the appropriate hydrazoneyl chloride **15a-c** (0.005 mol) in ethanol (30 ml) was added triethylamine (0.005 mol). The resulting solution was refluxed for 3 h. The solid formed was collected by filtration and crystallized from ethanol/ DMF. (see Table I, II).

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