# Multicomponent Synthesis of 2-Alkylthio-6-amino-3,5-dicyano-1,4-dihydropyridine-4-spirocycloalkanes. Molecular and Crystal Structure of 6-Amino-2-(2-methylbenzylthio)-3,5-dicyano-1,4dihydropyridine-4-spirocyclopentane

# V. D. Dyachenko, V. N. Nesterov, and I. V. Dyachenko

Taras Shevchenko Lugansk National University, ul. Oboronnaya 2, Lugansk, 91011 Ukraine e-mail: dvd lug@online.lg.ua

#### Received April 27, 2010

**Abstract**—2-Alkylthio-6-amino-3,5-dicyano-1,4-dihydropyridine-4-spirocycloalkanes were synthesized via the reaction of cycloalkylidene malononitriles with cyanothioacetamide and alkyl halides. The structure of 6-amino-2-(2-methylbenzylthio)-3,5-dicyano-1,4-dihydropyridine-4-spirocyclopentane was determined by the X-ray diffraction analysis.

**DOI:** 10.1134/S1070363211040232

Spiro-substituted nitrogen-containing heterocycles can be used to treat certain pathologies of the central nervous system [1–3], rheumatoid arthritis, and atherosclerosis [4]. We have previously developed methods for the synthesis of functionally substituted 1,4-dihydropyridine-4-spirocycloalkanes based on the reaction of cycloalkylidene(cyano)thioacetamide with malononitrile [5] or cyanothioacetamide [6], the condensation of cyclohexylidene(cyano)thioacetamide with 1-morpholino-1-cyclopentene [7], and the reaction of cyclohexylidene(cyano)thioacetamide with Meldrum's acid [8].

In this work we developed a new method of synthesis of these compounds via a multicomponent condensation of cycloalkylidene malononitriles I with cyanothioacetamide II and alkyl halides III in DMF in the presence of equimolar amount of *N*-methylmorpholine. The reaction products are 2-alkylthio-6-amino-3,5-dicyano-1,4-dihydropyridine-4-spirocyclo-alkanes IVa–IVe identical to those obtained previously by the alkylation of *N*-methylmorpholinium 6-amino-3,5-dicyano-1,4-dihydropyridine-4-spirocycloalkane-2-thiolate with alkyl halides [5, 6].

The reaction scheme includes apparently the alkylation of cyanothioacetamide **II** with alkyl halides **III** to form enaminonitriles **V**. The latter are alkylated

with cycloalkylidene malononitriles I by Stork method [9] to afford adducts VI, followed by the intramolecular cyclization to give iminopyridines VII, which are stabilized as the corresponding amino derivatives IVa–IVe.

The unambiguous structure of compound **IVa** was established by the X-ray diffraction analysis. Figure 1 shows a general view of the molecule, the bond lengths and angles are given in Tables 1 and 2.

In the investigated molecule the conformation of the hydrogenated heterocyclic ring can be described as flattened *boat*. The atoms  $N^1$  and  $C^4$  are displaced out of the ring plane (the plane involves the other four atoms with an accuracy of 0.003 Å) by -0.138 and -0.339 Å, respectively. The heterocycle has the same conformation in the substituted 1,4-dihydropyridine molecules [5, 10, 11], which contain bulky substituents or spirocyclohexane ring in the position 4. The cyclopentane substituent has an envelope conformation: the  $C^9$  atom is displaced out of the plane of the other atoms by 0.599 Å. The dihedral angle between the plane and the *boat* bottom of the heterocycle is 94.3°. Such mutual arrangement of the rings leads to a minimum non-valence intramolecular steric contacts between the hydrogen atoms in cyclopentane moiety and cyano groups. Indeed, there is only one non-valence



General view of molecule IVa.

contact  $H^{111}$ ... $C^{12}$  (2.37 Å) in the molecule, which is less than the sum of the van der Waals radii of H and C atoms [12]. Probably, in the almost planar fragments  $N^4-C^6=C^5-C^{12}=N^3$  of the heterocycle the conjugation leads to some shortening of the single bonds and lengthening of the double bonds in compari-son with the standard values [13] noted earlier [5]. The figure shows that the planar aryl substituent is located slightly over the heterocycle plane (torsion angles  $C^3C^2S^1C^{13} -$ 93.2°,  $C^2S^1C^{13}C^{14} - 54.3°$ , and  $S^1C^{13}C^{14}C^{15} - 73.6°$ .

Intermolecular hydrogen bonds  $N^4 - H^{42} \cdots N^3 (-1 - x, 1 - y, -z) [N^4 \cdots N^3 3.003(6), N^4 - H^{42} 0.85(4), H^{42} \cdots N^3 2.20(4) Å, angle N^4 - H^{42} \cdots N^3 157(2)^{\circ}] and N^1 - H^1 \cdots N^2$ 

Table 1. Some bond lengths (d, A) in molecule IVa

Bond	d	Bond	d
$N^1-C^2$	1.373(5)	C <sup>4</sup> –C <sup>8</sup>	1.556(5)
$C^2 - S^1$	1.768(4)	C <sup>4</sup> -C <sup>11</sup>	1.557(5)
$S^1 - C^{13}$	1.843(5)	C <sup>5</sup> -C <sup>6</sup>	1.364(5)
$C^{2}-C^{3}$	1.352(5)	C <sup>5</sup> -C <sup>12</sup>	1.415(5)
$C^{3}-C^{7}$	1.427(5)	C <sup>12</sup> –N <sup>3</sup>	1.139(6)
$C^{7}-N^{2}$	1.149(5)	C <sup>6</sup> -N <sup>4</sup>	1.349(5)
$C^{3}-C^{4}$	1.516(5)	C <sup>6</sup> -N <sup>1</sup>	1.373(5)
$C^{4}-C^{5}$	1.524(5)		

(-1 + x, 0.5 - y, -0.5 + z) [N<sup>1</sup>...N<sup>2</sup> 3.003(6), N<sup>1</sup>-H<sup>1</sup> 0.73(4), H<sup>1</sup>...N<sup>2</sup> 2.30(4) Å, angle N<sup>1</sup>-H<sup>1</sup>...N<sup>2</sup> 162(2)°] join the molecules in the crystal into the infinite layers.

Further alkylation of compounds IVb-IVd with allyl bromide IIIe and benzyl chloride IIIb, respectively, in DMF in the presence of 10% aqueous solution of KOH proceeds regioselectively at the amino group and results in 6-allylamino-2-benzylthio-3.5-dicvano-1.4-dihvdropvridine-4-spirocvclohexane 6-benzylamino-2-(4-bromophenylcarbamoyl-VIII. methylthio)-3,5-dicyano-1,4-dihydropyridine-4-spirocyclohexane IX, and 6-benzylamino-2-methylthio-3,5dicvano-1.4-dihvdropyridine-4-spirocyclohexane X (method a). Compounds VIII-X can be obtained in one-pot procedure by the successive treatment of the reaction mixture with the corresponding alkyl halide IIIb-IIIe in an alkaline medium without isolating thioethers **IV** (method *b*).

The acylation of compounds **IVb**, **IVc** with chloroacetyl chloride or acetyl bromide in the boiling toluene proceeds regioselectively exclusively at the amino group to form the corresponding *N*-acyl derivative **XI** or **XII**, respectively.

The structure of the synthesized compounds IVa– IVe, VIII–X was confirmed by the physicochemical and spectral data. A characteristic feature of the <sup>1</sup>H NMR spectra of the compounds VIII and IX is the presence of the signals of nonequivalent protons of SCH<sub>2</sub>-group as two doublets at  $\delta$  4.08 and 4.41 ppm (<sup>2</sup>J 14.0 Hz) (VIII) and at  $\delta$  3.81 and 4.18 ppm (<sup>2</sup>J 15.5 Hz) (IX). The splitting of the methylene proton signals indicates the absence of free rotation of substituents at the sulfur atom.

Table 2. Some bond angles ( $\omega$ , deg) in molecule IVa

Angle	ω	Angle	ω
$C^{2}S^{1}C^{13}$	99.4(2)	C5C4C11	112.3(3)
$C^2N^1C^6$	121.0(3)	$C^{8}C^{4}C^{11}$	102.9(3)
$C^{3}C^{2}N^{1}$	120.2(3)	$C^{6}C^{5}C^{12}$	116.5(3)
$C^3C^2S^1$	123.8(3)	$C^6C^5C^4$	123.2(3)
$N^1C^2S^1$	115.9(3)	$C^{12}C^5C^4$	120.2(3)
$C^2C^3C^7$	118.3(3)	$N^4C^6C^5$	125.6(3)
$C^2C^3C^4$	122.9(3)	$N^4C^6N^1$	115.2(3)
$C^7 C^3 C^4$	118.7(3)	$C^5C^6N^1$	119.2(3)
$C^{3}C^{4}C^{5}$	106.2(3)	$N^2C^7C^3$	177.1(4)
$C^{3}C^{4}C^{8}$	109.8(3)	$N^{3}C^{12}C^{5}$	178.7(5)
$C^5C^4C^8$	113.7(3)	$C^{14}C^{13}S^{1}$	113.0(3)
$C^{3}C^{4}C^{11}$	112.1(3)		



I, n = 1 (a), n = 2 (b); III, IV, Hgl = Br, Z = 2-CH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>, n = 1 (a), Hgl = Cl, Z = Ph, n = 2 (b), Hgl = I, Z = H, n = 2 (c), Hlg = Cl, Z = 4-BrC<sub>6</sub>H<sub>4</sub>NHCO, n = 2 (d), Hgl = Br,  $Z = CH=CH_2$ , n = 1 (e).

Introduction of 3-methylpyrazol-5-one **XIII** into the reaction with cyclohexylidene malononitrile **Ib** in ethanol in the presence of *N*-methylmorpholine led to the formation of a previously unknown 6'-amino-3'methyl-2'*H*-spiro[cyclohexane-1,4'-pyran[2,3-*c*]pyrazolo]-5'-carbonitrile **XIV** as a result of the Michael's reaction, which proceeds with the cyclization of the corresponding adduct **XV**. Note that in the <sup>1</sup>H NMR spectrum of compound **XIV** the assignment of the NH-proton signal of pyrazole ring as belonging to the 2*H*-isomer, but not to the 1*H*-isomer, was carried out taking into account the X-ray diffraction data for 4-alkyl-substituted 2H-pyrazolo[2,3-c]pyrans [14].

## **EXPERIMENTAL**

Crystals of the compound IVa are monoclinic, at -120°C: a 8.344(4), b 18.220(7), c 11.282 (4) Å, β 90.90(3)°, V 1715(1) Å<sup>3</sup>,  $d_{calc}$  1.303 g cm<sup>-3</sup>, Z 4, space group P21/n. Cell parameters and intensities of 3184 independent reflections were measured on a four-circle automatic diffractometer Syntex P21 ( $\lambda MoK_{\alpha}$ , graphite monochromator,  $\theta/2\theta$ -scanning,  $\theta_{max} 26^{\circ}$ ). The structure was solved by the direct method, which revealed all non-hydrogen atoms, and was refined in full-matrix anisotropic approximation for the non-hydrogen atoms using 2473 reflections with  $I > 3\sigma$ . All hydrogen atoms were identified by the difference Fourier synthesis and refined isotropically. The final values of the divergence factors: R 0.076, R<sub>W</sub> 0.2139 (S 1.08). All calculations were performed by the program SHELXTL PLUS [15] (version PC).

The <sup>1</sup>H NMR spectra were recorded on a Gemini-200 (199.975 MHz) (**IVa–IVe**, **XI**, **XII**), Bruker AM-300 (300.13 MHz) (**XIV**), and Bruker DR-500 spectrometers (500.13 MHz) (**VIII–X**) in DMSO- $d_6$ , internal reference TMS. The IR spectra were recorded on an IKS-40 instrument in mineral oil. The GLC spectra were recorded on instruments Grommass GC/MC-Hewlett-Packard 5890/5972 (column HP-5 MS, 70 eV) in methylene chloride solution (**VIII**, **IX**) and Kratos MS-890 (70 eV) with the direct injection of the substance into the ion source (**XIV**). The melting points were determined on a Koeffler block. The reaction progress was monitored by TLC (Silufol UV-254, acetone–hexane, 3:5, detecting with iodine vapor and UV irradiation).

General procedure for the synthesis of 2alkylthio-6-amino-3.5-dicyano-1.4-dihydropyridine-4-spirocycloalkanes (IVa-IVe). To a stirred solution of 1 g (10 mmol) of cyanothioacetamide II in 15 ml of DMF were added 10 mmol of alkyl halide III and 1.1 ml (10 mmol) of N-methylmorpholine. After stirring for 5 min, to the reaction mixture was added 10 mmol of cycloalkylidene malononitrile I. The mixture was stirred for 1 h and then kept for 24 h at 20°C. Then the mixture was diluted with an equal volume of water. The precipitate formed was filtered off, washed with water, ethanol, and hexane. Compounds IVa-IVd were crystallized from EtOH. The <sup>1</sup>H NMR spectra, melting points, and chromatographic data were identical to those reported previously [5, 6]. Yields 63 (IVa), 74 (IVb), 69 (IVc), 79 (IVd) and 71% (IVe).

6-Allylamino-2-benzylthio-3,5-dicyano-1,4-dihydropyridine-4-spirocyclohexane (VIII). a. To a stirred

solution of 3.4 g (10 mmol) of substituted 1,4-dihydropyridine-4-spirocyclohexane IVb in 15 ml of DMF was added in succession 5.6 ml (10 mmol) of 10% aqueous KOH solution and 0.85 ml (10 mmol) of allyl bromide IIIe. The mixture was stirred for 1 h and kept for 24 h. Then the reaction mixture was diluted with an equal volume of water. The precipitate formed was filtered off and washed with water, ethanol, and hexane. Yield 2.7 g (71%), colorless needles, mp 144-146°C (AcOH). IR spectrum, v, cm<sup>-1</sup>: 3316 (NH), 2194 (C=N). <sup>1</sup>H NMR spectrum, δ, ppm: 1.05–1.22 m (2H, CH<sub>2</sub>), 1.38–1.66 m (6H, 3CH<sub>2</sub>), 1.83–2.04 m (2H, CH<sub>2</sub>), 2.27–2.43 m (2H, NCH<sub>2</sub>), 4.08 d and 4.41 d (1H, SCH<sub>2</sub>, <sup>2</sup>J 14.0 Hz), 5.47 d (1H, =CH<sub>2</sub>, <sup>3</sup>J<sub>trans</sub> 17.0 Hz), 4.41 d (1H, =CH<sub>2</sub>,  ${}^{3}J_{cis}$  10.0 Hz), 5.47–5.54 m (1H, CH=), 7.19 t (2H, Ph,  ${}^{3}J$  7.5 Hz), 7.25 t (1H, Ph,  ${}^{3}J$ 8.0 Hz), 7.31 d (2H, Ph, <sup>3</sup>J 7.5 Hz), 7.94 br.s (1H, NH), 8.8 br.s (1H, NH). Mass spectrum (EI, 70 eV), m/z ( $I_{\rm rel}$ , %): 377 (100) [M + 1]<sup>+</sup>. Found, %: C 70.09; H 6.31; N 14.75. C<sub>22</sub>H<sub>24</sub>N<sub>4</sub>S. Calculated, %: C 70.18; H 6.43; N 14.88.

6-Benzylamino-2-(4-bromophenylcarbomoylmethylthio)-3,5-dicyano-1,4-dihydropyridine-4-spirocyclohexane (IX) was obtained similarly from thioether IVd and benzyl chloride IIIb. Yield 4.17 g (76%), yellow powder, mp 140-142°C (AcOH). IR spectrum, v, cm<sup>-1</sup>: 3328 (NH), 2184 (C≡N), 1670 (CONH). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 1.02–1.12 m (4H, 2CH<sub>2</sub>), 1.63–1.84 m (4H, 2CH<sub>2</sub>), 2.13–2.25 m (2H, CH<sub>2</sub>), 2.91 d and 3.02 d (1H, NCH<sub>2</sub>, <sup>2</sup>J 14.1 Hz), 3.81 d and 4.18 d (1H, SCH<sub>2</sub>, <sup>2</sup>J 13.5 Hz), 7.03 t (2H, Ph, <sup>3</sup>J 7.0 Hz), 7.11 t (1H, Ph, <sup>3</sup>J 6.5 Hz), 7.19 d (2H, Ph,  ${}^{3}J7.0$  Hz), 7.48 d and 7.58 d (2H, C<sub>6</sub>H<sub>4</sub>,  ${}^{3}J8.0$  Hz), 7.99 br.s (1H, NH), 8.46 br.s (1H, NH), 10.19 br.s (1H, NHCO). Mass spectrum (EI, 70 eV), m/z ( $I_{rel}$ , %): 549  $(100) [M + 1]^+$ . Found, %: C 58.97; H 4.68; N 12.71. C<sub>27</sub>H<sub>26</sub>BrN<sub>5</sub>OS. Calculated, %: C 59.12; H 4.78; N 12.77.

**6-Benzylamino-2-methylthio-3,5-dicyano-1,4-di-hydropyridine-4-spirocyclohexane (X)** was obtained similarly from thioether **IVc** and benzyl chloride **III**. Yield 2.35 g (67%), yellow crystals, mp 227–228°C (EtOH). IR spectrum, v, cm<sup>-1</sup>: 3322 (NH), 2199 (C≡N). <sup>1</sup>H NMR spectrum, δ, ppm: 1.14–1.38 m (3H, cyclohexane), 1.51–1.73 m (5H, cyclohexane), 1.99–2.17 m (2H, cyclohexane), 2.45 s (3H, CH<sub>3</sub>), 2.84 d and 3.04 d (1H, NCH<sub>2</sub>, <sup>2</sup>J 16.0 Hz), 7.12 d (2H, Ph, <sup>3</sup>J 8.0 Hz), 7.29–7.33 m (3H, Ph), 7.38 br.s (1H, NH), 8.45 br.s (1H, NH). Found, %: C 68.44; H 6.29; N 15.80. C<sub>20</sub>H<sub>22</sub>N<sub>4</sub>S. Calculated, %: C 68.54; H 6.35; N 15.99.

Synthesis of compound VIII. b. To a stirred solution of 1 g (10 mmol) of cyanothioacetamide II in 15 ml of DMF was added 1.15 ml (10 mmol) of benzyl chloride IIIb and 1.1 ml (10 mmol) of N-methylmorpholine. The reaction mixture was stirred for 5 min. Then 1.5 g (10 mmol) of cyclohexylidene malononitrile Ib was added, and the mixture was stirred for 1 h and kept for 24 h at 20°C. Then to the stirred mixture was added 5.6 ml (10 mmol) of 10% aqueous KOH solution and 0.85 ml (10 mmol) of allyl bromide IIIe. The reaction mixture was stirred for 1 h and kept for 24 h. Then the mixture was stirred for 1 h and kept for 24 h. Then the mixture was diluted with an equal volume of water. The precipitate formed was filtered off and washed with water, ethanol, and hexane. Yield 2.9 g (77%).

Synthesis of compound IX by the method b is similar using the appropriate alkyl halides IIIb and IIId. Yield 4.6 g (84%).

Synthesis of compound **X** by the method b is similar using the appropriate alkyl halides **IIIb** and **IIIc**. Yield 2.4 g (69%).

*N*-(2-Benzylthio-3,5-dicyano-1,4-dihydropyridine-4-spirocyclohexan-6-yl)-2-chloroacetamide (XI). A mixture of 3.4 g (10 mmol) of compound IVb and 0.8 ml (10 mmol) of chloroacetyl chloride in 20 ml of toluene was refluxed for 1 h and kept for 24 h. The formed colorless crystals were filtered off and washed with hexane. Yield 2.8 g (68%), mp 121–122°C. IR spectrum, v, cm<sup>-1</sup>: 3312 (NH), 2195 (C≡N), 1671 (CONH). <sup>1</sup>H NMR spectrum, δ, ppm: 1.12–1.92 m (10H, 5CH<sub>2</sub>), 4.19 s (2H, SCH<sub>2</sub>), 4.22 s (2H, ClCH<sub>2</sub>), 7.14–7.43 m (5H, Ph), 10.5 br.s (1H, NH), 10.56 br.s (1H, NHCO). Found, %: C 60.95; H 5.02; N 13.44. C<sub>21</sub>H<sub>21</sub>ClN<sub>4</sub>OS. Calculated, %: C 61.08; H 5.13; N 13.57.

*N*-(2-Methylthio-3,5-dicyano-1,4-dihydropyridine-4-spirocyclohexan-6-yl)acetamide (XII) was obtained similarly from thioether IVc and acetyl bromide. Yield 2.2 g (73%), colorless crystals, mp 159–161°C. IR spectrum, v, cm<sup>-1</sup>: 3334 (NH), 2288 (C=N), 1677 (CONH). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 1.22–1.79 m (10H, 5CH<sub>2</sub>), 2.05 s (3H, CH<sub>3</sub>CO), 2.48 s (3H, SCH<sub>3</sub>), 10.18 br.s (1H, NH), 10.34 br.s (1H, NHCO). Found, %: C 59.42; H 5.89; N 18.41. C<sub>15</sub>H<sub>18</sub>N<sub>4</sub>OS. Calculated, %: C 59.58; H 6.00; N 18.53.

**6'-Amino-3'-methylthio-2'H-spiro[cyclohexane-1,4'-pyrano[2,3-c]pyrazol]-5'-carbonitrile (XIV)**. To a stirred solution of 1.5 g (10 mmol) of cyclohexylidene malononitrile **Ib** in 15 ml of ethanol was added 0.98 g (10 mmol) of 3-methylpyrazol-5-one **XII** and 3 drops of *N*-methylmorpholine. The reaction mixture was stirred for 1 h and kept for 24 h. The precipitate formed was filtered off and washed with ethanol and hexane. Yield 1.7 g (69%), colorless crystals, mp 135–137°C (EtOH). IR spectrum, v, cm<sup>-1</sup>: 3280, 3349, 3412 (NH), 2188 (C=N), 1648 ( $\delta$ NH<sub>2</sub>). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 1.15–2.14 m (10H, 5CH<sub>2</sub>), 2.31 s (3H, CH<sub>3</sub>), 6.34 br.s (2H, NH<sub>2</sub>), 11.88 br.s (1H, NH). Mass spectrum (EI, 70 eV), *m/z* (*I*<sub>rel</sub>, %): 245 (6) [*M* + 1]<sup>+</sup>, 244 (25) [*M*]<sup>+</sup>, 215 (9), 201 (100) [*M* – H<sub>2</sub>O – HCN]<sup>+</sup>, 188 (29), 146 (11), 123 (8), 111 (15), 91 (9), 78 (13), 66 (14), 55 (24), 39 (40). Found, %: C 63.81; H 6.48; N 22.87. C<sub>13</sub>H<sub>16</sub>N<sub>4</sub>O. Calculated, %: C 63.92; H 6.60; N 22.93.

## REFERENCES

- 1. Guillaumet, G., J. Pharm. Belg., 1994, vol. 49, no. 3, p. 216.
- 2. European Patent Appl. EP 546.358, 1992, C. A., 1994, vol. 120, no. 13, 164006d.
- 3. USA Patent no. 6686361, 2004, *Ref. Zh. Khim.*, 2004, 19O101P.
- 4. USA Patent no. 5962462, 1999, *Ref. Zh. Khim.*, 2000, 19091P.
- Dyachenko, V.D., Krivokolysko, S.G., Nesterov, V.N., Struchkov, Yu.T., and Litvinov, V.P., *Izv. Akad. Nauk, Ser. Khim.*, 1996, no. 10, p. 2535.
- 6. Dyachenko, V.D., Krivokolysko, S.G., and Litvinov, V.P., Izv. Akad. Nauk, Ser. Khim., 1997, no. 10, p. 1849.
- Dyachenko, A.D., Desenko, S.M., and Dyachenko, V.D., *Khim. Geterotsikl. Soedin.*, 2002, no. 6, p. 845.
- Dyachenko, A.D., Desenko, S.M., and Dyachenko, V.D., Vestn. Khar'kovsk. Univ., Khimiya, 2002, no. 549, no. 8(31), p. 51.
- Stork, G., Brizzolara, A., Landesman, H., Szmuszkovicz, E.J., and Terrel, R., J. Am. Chem. Soc., 1963, vol. 85, no. 2, p. 207.
- Nesterov, V.N., Shklover, V.E., Struchkov, Yu.T., Sharanin, Yu.A., Sherstopalov, A.M., and Rodinovskaya, L.A., *Acta Crystallogr. C*, 1985, vol. 41, p. 1191.
- Dyachenko, V.D., Krivokolysko, S.G., Nesterov, V.N., and Litvinov, V.P., *Khim. Geterotsikl. Soedin.*, 1996, p. 1243.
- 12. Rowland, R.S. and Taylor, R., J. Phys. Chem., 1996, vol. 100, p. 7384.
- Allen, F.H., Kennard, O., Watson, D.G., Brammer, L., Orpen, A.G., and Taylor, R., *J. Chem. Soc.*, *Perkin Trans. 2*, 1987, no. 12, p. S1.
- 14. Dyachenko, V.D. and Rusanov, E.B., *Khim. Geterotsikl.* Soedin., 2004, no. 2, p. 270.
- Robinson, W. and Sheldrick, G.M., in *Crystallographic* Computing – Techniques and New Technologies, Oxford: Oxford Univ. Press, 1988, p. 366.