

Natural Product Chemistry. Part 121 [1]. Synthesis of Dicoumarinyl Ethers with the Structures Proposed for Fatagarine and Oreojasmine

Johannes Reisch^{a,*}, Anura Wickramasinghe^a [2], and Vijaya Kumar^b

^a Institut für Pharmazeutische Chemie, Westfälische Wilhelms-Universität,
D-4400 Münster, Federal Republic of Germany

^b Department of Chemistry, University of Peradeniya, Peradeniya, Sri Lanka

(Received 3 March 1988. Accepted 21 March 1988)

Synthesis of the 7,8'-dicoumarinyl ethers: 7-methoxy-7',8-oxydicoumarin and 6,7-dimethoxy-7',8-oxydicoumarin, established that the structures of fatagarine and oreojasmine for which these two structures have been proposed, have to be revised. Synthesis of 7-methoxy-5,7'-oxydicoumarin and 8-methoxy-7,7'-oxydicoumarin exclude the possibility of these dicoumarinyl ether structures for fatagarine.

(*Keywords:* Dicoumarinyl ethers; 6,7-Dimethoxy-7',8-oxydicoumarin; Fatagarine; 7-Methoxy-5,7'-oxydicoumarin; 7-Methoxy-7',8-oxydicoumarin; 8-Methoxy-7,7'-oxydicoumarin; Oreojasmine; *Ruta oreojasme*)

Naturstoffchemie. 121. Mitt.: Synthese von Dicumarinylethern mit den für das Fatagarin und das Oreojasmin vorgeschlagenen Strukturen

Durch die Synthese der 7,8-Dicumarinylether: 7-Methoxy-7',8-oxydicoumarin und 6,7-Dimethoxy-7',8-oxydicoumarin, ließ sich nachweisen, daß die für das Fatagarin und Oreojasmin vorgeschlagenen Strukturen revidiert werden müssen. Die Darstellung des 8-Methoxy-7,7'-oxydicoumarin und 7-Methoxy-5,7'-oxydicoumarin schließen die Möglichkeit dieser Strukturen für das Fatagarin aus.

Introduction

Although several dicoumarinyl ethers have been found to occur in nature [3–5], the isolation of fatagarine and oreojasmine from the fruits of *Ruta oreojasme* (Rutaceae) constituted the first report of the natural occurrence of a 7,8'-dicoumarinyl ether [6]. In an attempt to confirm the

structures of these dicoumarinyl ethers, we have synthesized 7-methoxy-7',8-oxydicoumarin (**1 a**) and 6,7-dimethoxy-7',8-oxydicoumarin (**1 b**) and have been able to show that the structures proposed for fatagarine and oreojasmine [6] require revision. The physical and spectral data reported for fatagarine [6] were also found to be not in agreement with those of 7-methoxy-5,7'-oxydicoumarin (**2**) or 8-methoxy-7,7'-oxydicoumarin (**3**), which were synthesized as alternative possible structures for fatagarine.

Results and Discussion

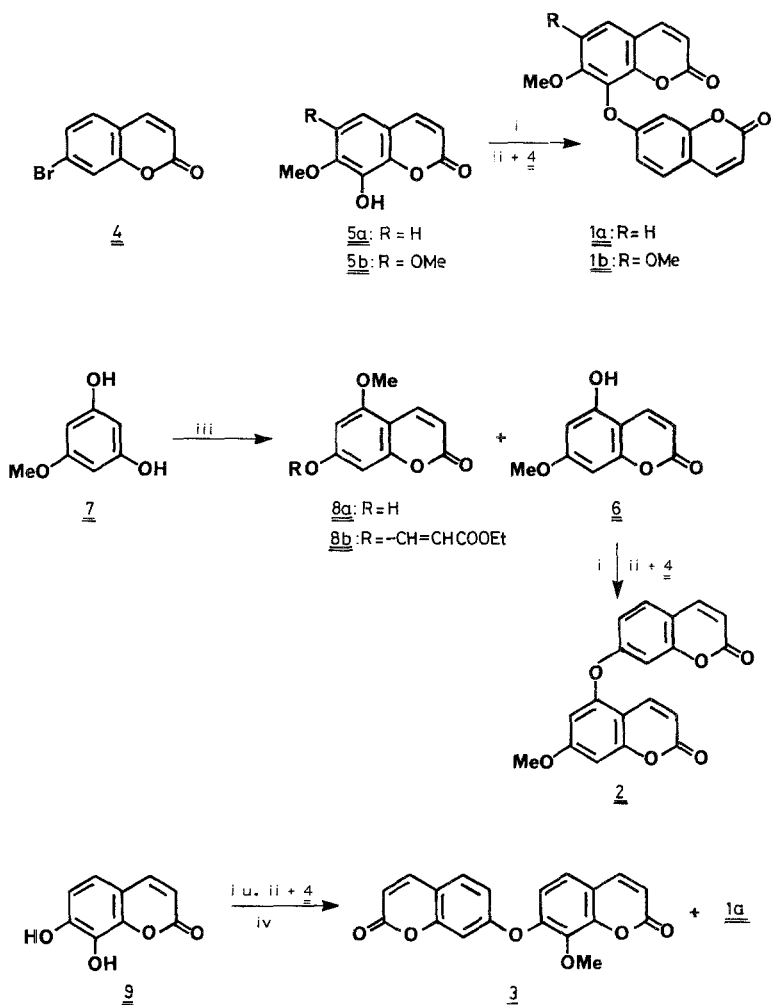
7-Methoxy-7',8-oxydicoumarin (**1 a**) was synthesized via *Ullmann* condensation [7] of 7-bromocoumarin (**4**) with the sodium salt of 8-hydroxy-7-methoxycoumarin (**5 a**). 7-Bromocoumarin (**4**) and the hydroxycoumarin **5 a** were obtained via *Pechmann* condensation [8] of malic acid with 3-bromophenol and 3-methoxycatechol, respectively. The sodium salt of 8-hydroxy-7-methoxycoumarin (**5 a**) was condensed with bromocoumarin **4** under *Ullmann* conditions [9] to give 7-methoxy-7',8-oxydicoumarin (**1 a**) in 25% yield. The product, m.p. 264°, differed in its spectral and physical properties from those reported for fatagarine, m.p. 233–234° [6] (Table 1).

6,7-Dimethoxy-7',8-oxydicoumarin (**1 b**) was prepared in 18% yield in a similar manner via *Ullmann* condensation of 7-bromocoumarin (**4**) with the sodium salt of 6,7-dimethoxy-8-hydroxycoumarin (**5 b**). The product, m.p. 201° differed in its spectral properties from those reported [6] for oreojasmine, m.p. 238–239° (Table 1).

Fatagarine and oreojasmine could therefore not be 7,8'-dicoumarinyl ethers and their structures have to be revised. With a view to propose an alternative structure for fatagarine, the synthesis of the 5,7'-dicoumarinylether, 7-methoxy-5,7'-oxydicoumarin (**2**) and 7,7'-dicoumarinylether, 8-methoxy-7,7'-oxydicoumarin (**3**) were attempted by the *Ullmann* condensation of the sodium salts of 5-hydroxy-7-methoxycoumarin (**6**) and 7,8-dihydroxy-coumarin (**9**), respectively, with 7-bromocoumarin (**4**).

5-Hydroxy-7-methoxycoumarin (**6**) was prepared by the reaction of ethyl propynoate and 5-methoxyresorcinol (**7**) [10]. The major product of the reaction (40%) was 7-hydroxy-5-methoxy-coumarin (**8 a**), the desired hydroxycoumarin **6** (7%) and the ether **8 b** (5%) being formed as minor products. *Ullmann* condensation of the sodium salt of the latter with 7-bromocoumarin (**4**) gave 7-methoxy-5,7'-oxydicoumarin (**2**) in 25% yield. This product, m.p. 196–197°, differed in its spectral properties from those reported for fatagarine [6] (see Table 1).

7,8-Dihydroxycoumarin (daphnetin; **9**) was obtained by *Pechmann* condensation of pyrogallol. *Ullmann* condensation of the sodium salt of



i: NaOMe, MeOH; ii: CuCl, Pyridine, 120° C, 18 h; iii: CH=CHCOOEt, ZnCl₂, 100° C, 1 h;
 iv: MeI, Aceton, K₂CO₃, 60° C, 0.5 h.

daphnetin with 7-bromocoumarin (**4**) gave a mixture of oxydicoumarins. Methylation of this mixture yielded 7-methoxy-7',8'-oxydicoumarin (**1a**) and 8-methoxy-7',7'-oxydicoumarin (**3**). The latter, m.p. 191–192°, too differed in its spectral properties from those of fatagarin [6] (see Table 1).

The structures of fatagarine and oreojasmine therefore need to be revised as they cannot have the structures **1a** and **1b** proposed for them [6].

Table 1. Chemical shifts (δ_H) (60 MHz) in CDCl_3

Compound	H-3, H-3' (d, $J = 10$ Hz)	H-4, H-4' (d, $J = 10$ Hz)	OMe (s)	Others
Fatagarine [4]	6.33	7.47, 7.66	3.88	H-5, H-5' (d, $J = 8.5$ Hz) 7.37 H-6', H-8' (m) 6.75–7.10
Oxydicoumarin (1a)	6.25	7.61, 7.65	3.88	H-6, H-6' (dd, $J = 8.5, 2.5$ Hz) 6.93 H-8' (d, $J = 2.5$ Hz) 6.60
Oxydicoumarin (2)	6.23, 6.37	7.72, 7.88	3.87	H-5' (d, $J = 8.5$ Hz) 7.51 H-6 (d, $J = 2.5$ Hz) 6.42 H-6', H-8 (m) 7.10–6.85 H-8' (d, $J = 2.5$ Hz) 6.68
Oxydicoumarin (3)	6.33, 6.43	7.71, 7.74	4.00	H-5' (d, $J = 8.5$ Hz) 7.48 H-5 (d, $J = 8.5$ Hz) 7.29 H-6, H-6', H-8' (m) 6.85–7.10
Oreojasmine [4]	6.34	7.48, 7.67	3.93 3.97	H-5' (d, $J = 8.5$ Hz) 7.32 H-5, H-6', H-8' (m) 6.80–7.10
Oxydicoumarin (1b)	6.25, 6.33	7.67	3.92 3.96	H-5 (s) 6.88 H-6' (dd, $J = 8.5, 2.5$ Hz) 6.96 H-8' (d, $J = 2.5$ Hz) 6.71

Experimental

Melting points (uncorr.): *Kofler*-hot stage microscope; IR (KBr): Pye Unicam SP 3-200; UV (*MeOH*): Carl Zeiss DMR 21; ^1H NMR (CDCl_3): Varian T 60, Bruker WM 300; ^{13}C NMR (CDCl_3): Bruker WM 300; MS (70 eV): MAT 44 S; TLC: Silicagel 60 F₂₅₄ (Merck); Column Chromatography: Silicagel 60 (Merck), grain size 0.063–0.2 mm; 3-Bromophenol (Janssen Chimica), 3-Methoxycatechol (Janssen Chimica), 6,7-Dimethoxy-8-hydroxycoumarin (Carl Roth K.G.) and 5-Methoxyresorcinol (Aldrich Chem. Co.) were used without further purification. Identities of compounds were established by mmp, UV and ^1H NMR comparisons, unless otherwise stated.

7-Bromocoumarin (4)

A mixture of 3-bromophenol (17.3 g, 0.1 mol), malic acid (13.4 g, 0.1 mol) and conc. H_2SO_4 (45 ml) was heated at 70 °C and then gradually to 120 °C during 4–5 h until the evolution of CO_2 stopped. The cooled reaction mixture was poured on to crushed ice and extracted with ether. Work up followed by chromatography gave 7-bromocoumarin (4) (2.12 g, 9.5%), m.p. 123 °C (lit. [8] m.p. 123.5 °C) and 5-bromocoumarin (0.26 g, 1.2%), m.p. 96 °C (lit. [8] m.p. 96 °C).

8-Hydroxy-7-methoxycoumarin (5a)

A mixture of 3-methoxycatechol (3.4 g, 27 mmol), malic acid (2.5 g, 19 mmol) and conc. H_2SO_4 (6.2 ml) was reacted as above. Chromatography followed by crystallisation from *MeOH*/hexane gave 8-hydroxy-7-methoxycoumarin (1.6 g, 30%), m.p. 173–174 °C (lit. [11] m.p. 173–175 °C).

7-Methoxy-7',8-oxydicoumarin (1a)

7-Bromocoumarin (262 mg, 1.16 mmol) and the sodium salt of 8-hydroxy-7-methoxycoumarin (252 mg, 1.16 mmol) were refluxed in pyridine (2 ml) with CuCl (18 mg, 0.18 mmol) for 18 h under N_2 . Acidification (2*N* HCl) followed by extraction with hexane and then, dichloromethane gave two extracts, which on chromatography gave respectively 7-bromocoumarin (170 mg) and on crystallisation from *MeOH*, 7-methoxy-7',8-oxydicoumarin (35 mg, 25%), m.p. 264 °C.

IR: 1730 (C=O), 1615 (C=C, arom.), 1100, 1110 (C—O), 840 (C—H, arom.). UV (*MeOH*): λ_{max} nm ($\log \epsilon$) = 205 (4.64), 241.5 sh. (3.85), 291 sh. (4.31), 316 (4.47). ^1H -NMR: See Table 1. ^{13}C -NMR: δ (ppm) = 56.5 (O—CH₃), 102.4 (C-8'), 108.7 (C-6), 113.2 (C-6'), 113.6, 113.7, 113.9 (C-3,3',4a,4'a), 125.5 (C-5), 129.0 (C-5'), 129.2 (C-8a), 143.4, 143.5 (C-4, 4'), 147.9 (C-8), 155.2 (C-7), 155.4 (C-8'a), 160.0 (C-7'), 160.8, 161.2 (C-2, 2'). MS: m/e (%) = 336 (M^+ , 90), 308 (M^+ -CO, 20), 293 (308-CH₃, 10), 265 (293-CO, 14), 89 (C₇H₅⁺, 100). Anal. calc. for C₁₉H₁₂O₆: C 67.84, H 3.60. Found: C 67.80, H 3.64%.

6,7-Dimethoxy-7',8-oxydicoumarin (1b)

The above reaction when repeated using the sodium salt of 6,7-dimethoxy-8-hydroxycoumarin (286 mg, 1.16 mmol) gave recovered 7-bromocoumarin (120 mg) and 6,7-dimethoxy-7',8-oxydicoumarin as colourless crystals from *MeOH* (41 mg, 18%), m.p. 201 °C.

IR: 1710, 1730 (C=O), 1615 (C=C, arom.), 1140 (C—O), 840 (C—H, arom.). UV (*MeOH*): λ_{max} nm ($\log \epsilon$) = 206 (4.64), 227 sh. (4.37), 248 sh. (3.70), 291 (4.25), 320 (4.27), 340 sh. (4.10). ^1H -NMR: See Table 1. ^{13}C -NMR: δ (ppm)

= 56.5, 61.6 ($2 \times \text{O}-\text{CH}_3$), 103.0 (C-8'), 106.2 (C-6'), 113.0 (C-5), 114.2 (C-4a, 4'a), 114.6 (C-7), 115.8 (C-3, 3'), 129.1 (C-5'), 134.3 (C-6), 142.9, 143.1 (C-4, 4'), 145.9 (C-8a), 150.2 (C-8), 155.5 (C-8'a), 159.7 (C-7'), 160.7 (C-2, 2'). MS: m/e (%) = 366 (M^+ , 100), 338 ($M^+-\text{CO}$, 8), 89 (C_7H_5^+ , 26). Anal. calc. for $\text{C}_{20}\text{H}_{14}\text{O}_7$: C 65.56, H 3.85. Found: C 65.47, H 4.01%.

5-Hydroxy-7-methoxycoumarin (6)

A stirred mixture of 5-methoxyresorcinol (2.24 g, 16 mmol), ethyl propynoate (2.35 g, 24 mmol) and freshly fused ZnCl_2 (2.18 g, 16 mmol) were heated on a steam bath for 1 h. The cooled reaction mixture was acidified (5% aqueous. HCl), and filtered. The pale brown residue gave on crystallisation from MeOH and from chromatography of the mother liquors, 7-hydroxy-5-methoxycoumarin (1.22 g, 40%), m.p. 251–252 °C (lit. [12] m.p. 244–245 °C). Chromatography of the mother liquors also gave, on crystallisation from ethyl acetate, 5-hydroxy-7-methoxycoumarin (215 mg, 7%), m.p. 238–239 °C (lit. [13] m.p. 224–228 °C) and the less polar ether **7b** (230 mg, 5%), m.p. 158–159 °C.

Less polar ether **7b**: IR: 1720 (C=O), 1650 (C=C), 1620 (C=C, arom.), 1200, 1160 (C—O), 830, 810 (C—H, arom., C=C—H). UV (MeOH): λ_{max} nm ($\log \epsilon$) = 203 (4.32), 221 sh. (4.21), 258 (4.08), 316 (4.08). $^1\text{H-NMR}$ (60 MHz): δ (ppm) = 8.15 (d, J = 9.6 Hz; 1 H, H-4), 7.05 (d, J = 6.8 Hz; 1 H, C=CH-*o-ph*), 6.67, 6.57 (2d, J = 2 Hz; 2 H, H-6,7), 6.27 (d, J = 9.6 Hz; 1 H, H-3), 5.34 (d, J = 6.8 Hz; 1 H, C=CH-*o-Ac*), 4.32 (q, J = 6.6 Hz; 2 H, —CH₂), 1.34 (t, J = 6.6 Hz; 3 H, —CH₃). MS: m/e (%) = 290 (M^+ , 100), 245 ($M^+-\text{OC}_2\text{H}_5$, 38), 217 (245-CO, 90), 189 (217-CO, 42).

7-Methoxy-5,7'-oxydicoumarin (2)

7-Bromocoumarin (128 mg, 0.57 mmol), the sodium salt of 5-hydroxy-7-methoxycoumarin (123 mg, 0.57 mmol), and CuCl (12 mg, 0.12 mmol) in pyridine (2 ml) when reacted by the method described above for oxydicoumarins gave recovered 7-bromocoumarin (40 mg) and 7-methoxy-5,7'-oxydicoumarin, colourless crystals from MeOH (32 mg, 25%), m.p. 196–197 °C.

IR: 1730 (C=O), 1612 (C=C, arom.), 1125, 1150 (C—O), 840 (C—H, arom.). UV (MeOH): λ_{max} nm ($\log \epsilon$) = 207 (4.56), 242 sh. (3.93), 253 sh. (3.82), 291 sh. (4.22), 319 (4.42). $^1\text{H-NMR}$: See Table 1. $^{13}\text{C-NMR}$: δ (ppm) = 56.1 (O—CH₃), 97.2 (C-6), 102.6 (C-8'), 105.9 (C-4a), 106.5 (C-8), 112.9 (C-6), 115.0, 115.2, 115.3 (C-3, 3', 4'a), 129.5 (C-5'), 137.7 (C-4), 142.8 (C-4'), 153.1 (C-7'), 155.5 (C-8'a), 156.8 (C-8a), 159.3 (C-7), 160.2 (C-5), 160.6 (C-2'), 163.3 (C-2). MS: m/e (%) = 336 (M^+ , 100), 308 ($M^+-\text{CO}$, 52), 293 (308-CH₃, 30), 265 (293-CO, 24), 237 (265-CO, 10), 209 (237-CO, 12), 181 (209-CO, 10), 89 (C_7H_5^+ , 22). Anal. calc. for $\text{C}_{19}\text{H}_{12}\text{O}_6$: C 67.84, H 3.60. Found: C 67.61, H 3.51%.

7,8-Dihydroxycoumarin (Daphnetin) (9)

A mixture of pyrogallol (12.6 g, 0.1 mol), malic acid (13.4 g, 0.1 mol) and conc. H_2SO_4 (45 ml) was reacted as earlier. Chromatography followed by crystallisation from methanol gave 7,8-dihydroxycoumarin (3.55 g, 20%) m.p. 255–256 °C (lit. [14] m.p. 253–255 °C).

8-Methoxy-7,7'-oxydicoumarin (3)

7-Bromocoumarin (0.562 g, 2.5 mmol), the sodium salt of 7',8-dihydroxycoumarin (0.502 g, 2.5 mmol), and CuCl (18 mg, 0.18 mmol) in pyridine (5 ml),

when reacted, by the method described above for oxydicoumarins, gave recovered 7-bromocoumarin (0.146 g) and a mixture of hydroxy dicoumarinyldethers (0.165 g, 28%). This mixture in dry acetone (10 ml) was refluxed with CH_3I (0.35 g)/anh. K_2CO_3 (0.25 g) for 0.5 h. The product, on chromatographic separation gave 7-methoxy-7',8'-oxydicoumarin (**1a**) (90 mg, 52%) and 8-methoxy-7,7'-oxydicoumarin (**3**) (60 mg, 35%) as colourless crystals from *MeOH*, m.p. 191–192 °C.

IR: 1 720 (C=O), 1 610 (C=C, arom.), 1 135 (C—O), 850 (C—H, arom.). UV (*MeOH*): λ_{max} nm ($\log \epsilon$) = 209 (4.74), 240 sh. (4.12), 248 sh. (4.03), 294.5 sh. (4.50), 320 (4.59). $^1\text{H-NMR}$: See Table 1. $^{13}\text{C-NMR}$: δ (ppm) = 61.8 (O—CH₃), 104.9 (C-8'), 113.8 (C-6), 114.6 (C-6, 4a), 115.8 (C-3'), 117.4 (C-4'a), 117.7 (C-3), 122.8 (C-5), 129.2 (C-5'), 139.7 (C-8), 143.0, 143.2 (C-4, 4'), 148.6 (C-8a), 150.0 (C-7), 155.5 (C-8'a), 159.6 (C-7'), 160.4, 160.5 (C-2, 2'). MS: m/e (%) = 336 (M^+ , 100), 308 (M^+ -CO, 10), 305 (M^+ -OCH₃, 21), 265 (308-COCH₃, 15), 89 (C_7H_5^+ , 50). Anal. calc. for $\text{C}_{19}\text{H}_{12}\text{O}_6$: C 67.84, H 3.60. Found: C 67.84, H 3.57%.

Acknowledgements

We are grateful to the "Deutsche Forschungs-Gemeinschaft" for financial support and to the "Deutscher Akademischer Austauschdienst" for the award of a fellowship to A. W.

References

- [1] For part 120 see: *Adesina SK, Olatunji OA, Bergenthal D, Reisch J* (1988) *Pharmazie* 43: 221
- [2] A Part of the PhD dissertation Wickramasinghe A (Peradeniya [Sri Lanka]/Münster) (in preparation)
- [3] *Viswanathan N, Balakrishnan V* (1974) *Ind J Chem* 12: 450
- [4] *Reisch J, Novak I, Szendrei K, Minker E* (1968) *Planta Med* 16: 372
- [5] *Majumder PL, Sengupta GC, Dinda BN, Chatterjee A* (1974) *Phytochemistry* 13: 1929
- [6] *Gonzalez AG, Reyes RE, Arencibia JB* (1975) *An Quim* 71: 842
- [7] *Moroz AA, Swartenberg MS* (1974) *Russ Chem Rev* 43: 679
- [8] *Reisch J, Rosenthal BHW* (1987) *Monatsh Chem* 118: 871
- [9] *Williams AL, Kinney RE, Bridger RF* (1967) *J Org Chem* 32: 2501
- [10] *Kaufman KD, Kelly RC* (1965) *J Heterocyclic Chem* 2: 91
- [11] *Murray RDH, Jorge ZD* (1984) *Tetrahedron* 40: 5229
- [12] *Tanino H, Inoue S* (1969) *Chem Pharm Bull* 17: 1071
- [13] *Murray RDH, Ballantyne MM, Hogg TC, McCabe PH* (1975) *Tetrahedron* 31: 2960
- [14] *Dean FM, Costa AMBSRCS, Harborne JB, Smith DM* (1978) *Phytochemistry* 17: 505