

# Ruthenium-Catalyzed Asymmetric N-Acyl Nitrene Transfer Reaction: Imidation of Sulfide

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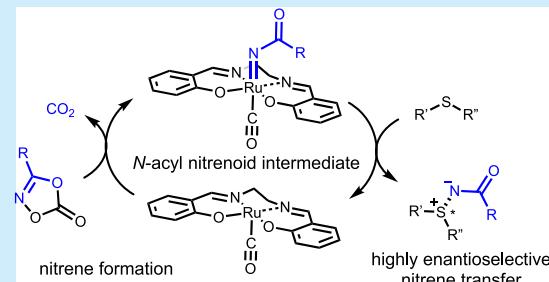
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**ABSTRACT:** The asymmetric nitrene transfer reaction is a useful and strong tool for the construction of nitrogen functional groups such as *N*-sulfonyl amide and carbamic ester in a highly enantioselective manner. On the other hand, there is a substantial limitation in this field: the transfer of *N*-acyl amide via the corresponding nitrene intermediates is still difficult because *N*-acyl nitrenes undergo undesired nitrene dimerization or Curtius rearrangement. Herein, we achieved highly enantioselective imidation of sulfides via catalytic *N*-acyl nitrene transfer with (OC)ruthenium–salen complex **2b** as the catalyst and 3-substituted 1,4,2-dioxazol-5-ones **1** as the nitrene source. Complex **2b** can decompose dioxazolones **1** to the desired *N*-acyl nitrene intermediates without any activation via heating or UV irradiation, or transfer generating nitrene intermediates to the sulfur atom of sulfides with good to excellent enantioselectivities ( $\leq 98\%$  ee) without diazene and isocyanate contamination.

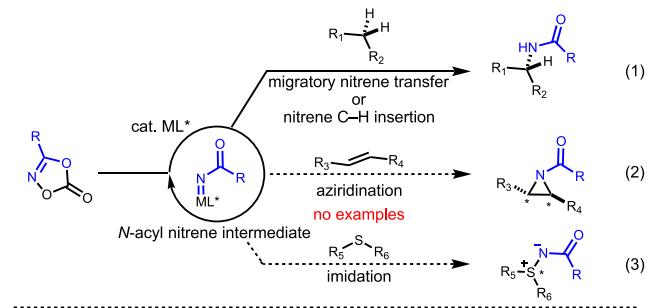
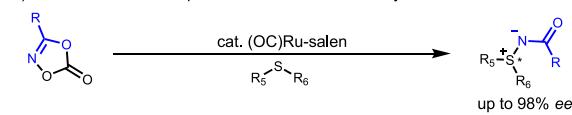


Nitrene transfer reactions such as imidation of heteroatoms,<sup>1,2</sup> aziridination of olefins,<sup>3</sup> and amination of C—H bonds<sup>4</sup> are useful and powerful tools for the synthesis of nitrogen-containing organic compounds, which are ubiquitous in biologically active compounds such as alkaloids and amides. Thus, these transformations have attracted continuing and growing interest, leading to the development of various powerful and useful methodologies in this field.<sup>5</sup> Today, the introduction of a nitrogen functional group can be readily performed in a highly site-selective, chemoselective, and enantioselective manner by using the appropriate catalyst and nitrene source. However, there is a substantial limitation in this field: the transfer of *N*-acyl amides via the corresponding nitrene intermediates is still difficult. Unfortunately, *N*-acyl nitrene species often undergo Curtius rearrangement and produce the corresponding isocyanates or related products under UV irradiation or thermal conditions.<sup>6</sup> Although catalytic *N*-acyl nitrene transfer reactions using azides have been reported, the product is contaminated with diazene derivatives as coproducts.<sup>7</sup> On the other hand, the reaction using metal(nitride) complexes as stoichiometric nitrene source affords the desired product in a chemo- and stereoselective manner.<sup>8</sup> These results indicate that *N*-acyl metal(nitrene), which can be obtained catalytically, would give the desired *N*-acyl-amidated products without the formation of isocyanates and diazenes as byproducts. Recently, Bolm and co-workers demonstrated the high utility of 3-substituted 1,2,4-dioxazol-5-ones **1** in the imidation of sulfides.<sup>9,10</sup> Thus, the imidation of sulfides and sulfoxides using dioxazolones **1** as the nitrene source proceeds successfully with complete chemoselectivity. Subsequently, Chang et al. reported a catalytic C—

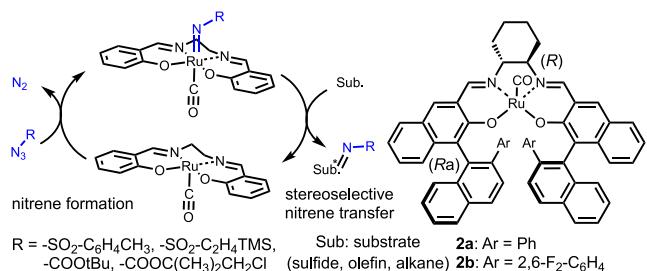
H amination using **1** as the nitrogen source.<sup>11</sup> In this reaction, a cationic (pentamethylcyclopentadienyl)rhodium complex induces the decomposition of **1** into the desired nitrene intermediate without any activation method such as heating or photoirradiation, and *N*-acyl amide groups are transferred to C—H bonds in a chemoselective and site-selective manner. After these findings had been published, the C—H amidation reaction based on the transition metal/dioxazolone **1** system has developed considerably, including asymmetric versions.<sup>12–14</sup>

However, the enantioselective *N*-acyl nitrene transfer reaction is still limited to the C—H amination (Scheme 1a).<sup>15</sup> This limitation could be overcome if the *N*-acyl nitrene group could be transferred to heteroatoms and/or olefins with chemo- and stereoselectivity, which would boost the applicability of nitrene transfer reactions in organic synthesis. Motivated by this, we were interested in the development of the catalytic asymmetric *N*-acyl nitrene transfer reactions. Herein, we describe highly chemo- and enantioselective *N*-acyl imidation of sulfides catalyzed by (OC)ruthenium–salen complexes **2** using dioxazolone **1** as the nitrene source (Scheme 1b).

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**Scheme 1. Asymmetric N-Acyl Nitrene Transfer Reactions**a) Catalytic asymmetric *N*-acyl nitrene transfer reaction using dioxazolonesb) This works: first examples of enantioselective *N*-acyl imidation of sulfides

Recently, we reported that (OC)ruthenium–salen complexes **2** are efficient catalysts in asymmetric nitrene transfer reactions such as imidation of sulfides, aziridination of olefins, and C–H aminations, using *N*-sulfonyl and *N*-alkoxycarbonyl azides as nitrene sources (Figure 1).<sup>5c,16–18</sup> These ruthenium-catalyzed nitrene transfer reactions using azides could be expected to prompt further development of *N*-acyl nitrene transfer reactions.



**Figure 1.** Cross-selective oxidative coupling between two different arenols.

We conducted the *N*-acyl imidation of methyl phenyl sulfide **3a** using ruthenium–salen complex **2a** as the catalyst (Table 1). No reaction was observed in the presence of *N*-benzoyl azide (entry 1). Fortunately, 3-phenyl-1,4,2-dioxazol-5-one **1a** gave the desired sulfide **4aa** with 84% ee in quantitative yield without any activation such as heating or photoirradiation (entry 2). Diazene and isocyanate derivatives were not observed under those conditions. Encouraged by these results, we conducted the *N*-acyl imidation of methyl phenyl sulfide **3a** using 3-phenyl-1,4,2-dioxazol-5-one **1a** as the nitrene precursor with a series of ruthenium complexes **2** as catalysts (Table 1, entries 2 and 3, and Table S1<sup>19</sup>). Complex **2b** showed the best enantioselectivity (95% ee) in tetrahydrofuran (THF) as the solvent. The imidation with the complex **2b**/dioxazolone **1a** system in nonpolar and less polar solvents proceeded with good to high enantioselectivities (entries 3–8). On the other hand, the reaction in methanol was sluggish (entry 9). Acetonitrile could be used as the solvent, albeit with reduced yield and enantioselectivity (entry 10). It was also possible to reduce the catalyst loading to 0.2 mol % in THF without affecting the chemical yield or the enantioselectivity (entries 11 and 12).

**Table 1. Optimization of Imidation Using 3-Phenyl-1,4,2-dioxazolone as a Nitrene Source<sup>a</sup>**

entry	catalyst	solvent	yield (%) <sup>b</sup>	ee (%) <sup>c</sup>
1 <sup>d</sup>	<b>2a</b>	THF	NR <sup>e</sup>	—
2	<b>2a</b>	THF	>99	84
3	<b>2b</b>	THF	>99	95
4	<b>2b</b>	1,4-dioxane	>99	90
5	<b>2b</b>	Et <sub>2</sub> O	73	81
6	<b>2b</b>	AcOEt	>99	93
7	<b>2b</b>	toluene	>99	90
8	<b>2b</b>	CH <sub>2</sub> Cl <sub>2</sub>	>99	94
9	<b>2b</b>	MeOH	trace	—
10	<b>2b</b>	CH <sub>3</sub> CN	46	76
11 <sup>f</sup>	<b>2b</b>	THF	>99	95
12 <sup>g</sup>	<b>2b</b>	THF	88	97

<sup>a</sup>Reactions were carried out using **1a** (0.13 mmol) and **3a** (0.1 mmol) with ruthenium complex **2** (2 mol %) at 25 °C for 24 h, unless otherwise specified. <sup>b</sup>Isolated yield, based on sulfide **3**. <sup>c</sup>Determined by HPLC with a chiral stationary phase. <sup>d</sup>*N*-Benzoyl azide (0.13 mmol) was used as the nitrene source instead of **1a**. <sup>e</sup>No reaction was observed. <sup>f</sup>Carried out with **2b** (1 mol %) and 4 Å molecular sieves (20 mg) for 18 h. <sup>g</sup>Carried out on a 1.0 mmol scale with **2b** (0.2 mol %) and 4 Å molecular sieves (20 mg) for 24 h.

With the optimized conditions in hand, we conducted the asymmetric imidation using 3-substituted 1,2,4-dioxazolones **1** as nitrene precursors with methyl phenyl sulfide in the presence of complex **2b** as the catalyst (Table 2). The reaction of *para*- and *meta*-substituted 3-phenyl-1,2,4-dioxazolones **1b–k** gave better yields than the *ortho*-substituted derivatives **1l–o**, and high enantioselectivities were obtained in all cases (90–98% ee) (entries 1–14). In addition, 3-alkyl- and 3-alkenyl-1,2,4-dioxazolones **1p–u** could be also successfully applied as the nitrene source. 3-Methyl-, 3-pentyl-, and 3-(2-propyl)-dioxazolones **1t** gave desired products **4** with good enantioselectivity and moderate to high yields (entries 15–17, respectively). However, no reaction occurred when using dioxazolone **1s** as the nitrene source (entry 18). Furthermore, 3-(*E*)-propenyldioxazolones **1t** also produced the corresponding *N*-acyl sulfimides **4ta** with good enantioselectivities (entry 19).

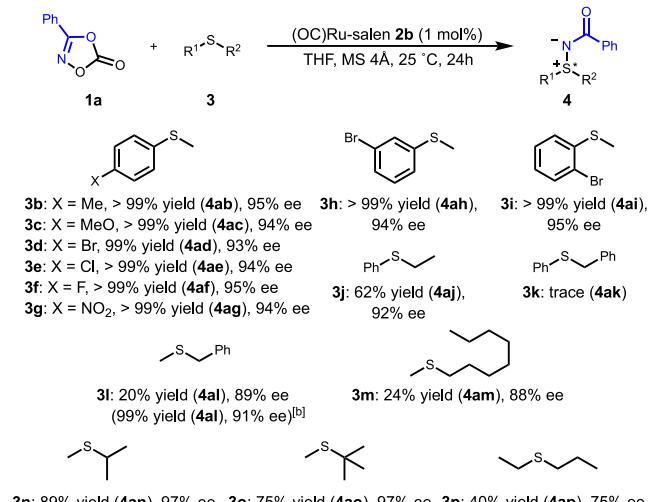
We further examined the asymmetric sulfimidation of various sulfides **3** using the complex **2b**/3-phenyldioxazolone **1a** system (Scheme 2). In general, aryl methyl sulfides **3b–g** gave the desired products **4** with high enantioselectivities and high yields, irrespective of the electronic nature and position of the substituents. Ethyl phenyl sulfide **3j** and benzyl phenyl sulfide **3k** also underwent the *N*-acyl imidation with moderate stereoselectivities. Meanwhile, benzyl methyl sulfide **3l** gave an 89% ee, albeit sluggish. Fortunately, the chemical yield of **4al** was improved to 99% by using a 4 mol % catalyst loading. Dialkyl sulfides **3m–p** also gave the products with good to high enantioselectivities, and sterically hindered dialkyl sulfides such as methyl 2-propyl sulfide **3n** and methyl *tert*-butyl sulfide **3o** afforded yields better than those of the less hindered sulfides **3m** and **3p**. Ethyl propyl sulfide **3p** also gave the desired sulfimide **4ap** with acceptable enantioselectivity (75% ee).

**Table 2. Scope and Limitation of 3-Substituted 1,4,2-Dioxazolone 1<sup>a</sup>**

entry	R	product	yield (%) <sup>b</sup>	ee (%) <sup>c</sup>
1	4-MeC <sub>6</sub> H <sub>4</sub> (1b)	4ba	>99	96
2	4-MeOC <sub>6</sub> H <sub>4</sub> (1c)	4ca	99	98
3	4-BrC <sub>6</sub> H <sub>4</sub> (1d)	4da	99	95
4	4-ClC <sub>6</sub> H <sub>4</sub> (1e)	4ea	93	94
5	4-FC <sub>6</sub> H <sub>4</sub> (1f)	4fa	>99	93
6	4-CF <sub>3</sub> C <sub>6</sub> H <sub>4</sub> (1g)	4ga	51	91
7	3-MeC <sub>6</sub> H <sub>4</sub> (1h)	4ha	>99	97
8	3-BrC <sub>6</sub> H <sub>4</sub> (1i)	4ia	>99	92
9	3-ClC <sub>6</sub> H <sub>4</sub> (1j)	4ja	>99	92
10	3-FC <sub>6</sub> H <sub>4</sub> (1k)	4ka	98	93
11	2-MeC <sub>6</sub> H <sub>4</sub> (1l)	4la	46	90
12	2-BrC <sub>6</sub> H <sub>4</sub> (1m)	4ma	75	94
13	2-ClC <sub>6</sub> H <sub>4</sub> (1n)	4na	86	94
14	2-FC <sub>6</sub> H <sub>4</sub> (1o)	4oa	>99	95
15	Me (1p)	4pa	91	94 (R) <sup>d</sup>
16	n-pentyl (1q)	4qa	>99	93
17	iPr (1r)	4ra	52	96
18	tBu (1s)	4sa	NR	—
19	(E)-Ph=C-C- (1t)	4ta	94	94

<sup>a</sup>Unless otherwise specified, all reactions were carried out using 1 (0.13 mmol) and 3a (0.1 mmol) with complex 2b (1 mol %) and 4 Å molecular sieves (20 mg) in THF at 25 °C. <sup>b</sup>Isolated yield, based on methyl phenyl sulfide 3a. <sup>c</sup>Determined by HPLC with a chiral stationary phase. <sup>d</sup>Determined by chiroptical comparison after 4pa was converted into sulfoximine 6pa.<sup>20</sup>

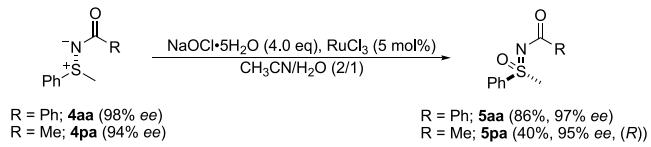
**Scheme 2. Scope and Limitation of Sulfide 3<sup>a</sup>**



<sup>a</sup>Unless otherwise specified, all reactions were carried out using 1a (0.11 mmol) and 3 (0.1 mmol) with complex 2b (1 mol %) at 25 °C in the presence of 4 Å molecular sieves. <sup>b</sup>Carried out with 2b (4 mol %).

N-Acy l sulfimides 4aa and 4pa could be oxidized to sulfoximines 5aa and 5pa, respectively, in good yields without a decreasing enantiomeric ratio (Scheme 3).<sup>21</sup>

**Scheme 3. Oxidation of Sulfimides 4 to Sulfoximines 5**



In conclusion, highly enantioselective N-acyl imidation of sulfide 3 was achieved by using dioxazolones 1 as the nitrene source in the presence of (OC)ruthenium–salen complex 2b as the catalyst. It was thought that ruthenium complex 2b decomposes various dioxazolones 1 to the corresponding N-acyl ruthenium(nitrene) intermediates without any activation such as heating and photoirradiation and produces the desired sulfimides 4 in an excellent enantio- and chemoselective manner. Fortunately, during these studies, undesired nitrene dimerization and Curtius rearrangement were not observed. Moreover, the reaction proceeded with a high yield even at a 0.2 mol % catalyst loading to give the corresponding products. Obtained sulfimides 4 could be transformed into the corresponding sulfoximines 5 with no erosion of the enantiomeric excess.

## ASSOCIATED CONTENT

### Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.0c01373>.

Optimization, experimental procedure, characterization data, HPLC conditions, and NMR spectra (PDF)

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### Notes

The authors declare no competing financial interest.

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