Accepted Manuscript

Synthesis of 2-Substituted Indoles by Iridium (III)-Catalyzed C-H Functionalization of *N*-phenylpyridin-2-amines

Lei Zhang, Junyu Chen, Jinkang Chen, Licheng Jin, Xiangyun Zheng, Xinpeng Jiang, Chuanming Yu

PII:	S0040-4039(19)30235-7		
DOI:	https://doi.org/10.1016/j.tetlet.2019.03.027		
Reference:	TETL 50668		
To appear in:	Tetrahedron Letters		
Received Date:	1 January 2019		
Revised Date:	8 March 2019		
Accepted Date:	11 March 2019		



Please cite this article as: Zhang, L., Chen, J., Chen, J., Jin, L., Zheng, X., Jiang, X., Yu, C., Synthesis of 2-Substituted Indoles by Iridium (III)-Catalyzed C-H Functionalization of *N*-phenylpyridin-2-amines, *Tetrahedron Letters* (2019), doi: https://doi.org/10.1016/j.tetlet.2019.03.027

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.



Tetrahedron Letters

journal homepage: www.elsevier.com

Synthesis of 2-Substituted Indoles by Iridium (III)-Catalyzed C-H Functionalization of *N*-phenylpyridin-2-amines

Lei Zhang, Junyu Chen, Jinkang Chen, Licheng Jin, Xiangyun Zheng, Xinpeng Jiang*, and Chuanming Yu*

College of pharmaceutical sciences, Zhejiang University of Technology, Hangzhou 310014, P. R. of China

ARTICLE INFO

ABSTRACT

Article history: Received Received in revised form Accepted Available online A highly regioselective synthesis of 2-substituted indoles was realized through Ir(III)-catalyzed C–H functionalization of *N*-phenylpyridin-2-amines followed by the reaction with sulfoxonium ylides and intramolecular cyclization under mild conditions. The reaction completed with broad range of substrate scopes and gave various 2-substituted indoles in up to 98% yields.

2018 Elsevier Ltd. All rights reserved.

Keywords: Iridium Sulfoxonium ylides C-H functionalization Indole

Introduction

Indole is a privileged scaffold with numerous applications in both life sciences and material sciences [1]. In particular, 2-arylindoles [2] and 2-aliphatic substituted indoles [3] are intriguing structural motifs owing to their presence in a series of marketed drugs (Figure 1). Consequently, tremendous efforts have been devoted to constructing 2-substituted indole scaffolds. Recently, Rh- [4], Pd- [5], Ru- [6], Ni- [7] and other transition metals [8] catalyzed C–H functionalization provided a fundamental step for the preparation of various 2-substituted indoles.



Fig.1 Selected drugs containing 2-substituted indoles

In 2013, Saá's group [9] reported a Rh(III)-catalyzed synthesis of 2-methylindoles with stoichiometric copper(II) as oxidant (Scheme 1a). Shortly after, Jana's group [10] developed a palladium-catalyzed synthesis of 2-arylindole



Scheme 1 Synthesis of 2-substituted indoles via annulative C–H activation.

(Scheme 1b). More recently, Kim and Cui applied diazo compounds and vinyl azides to synthesize 2-substituted indoles (Scheme 1c and 1d) [11]. Despite the significant advance made in this field, the previous methods still suffer from limited

substrate scopes or the need for potential dangerous diazo compounds. Therefore, the development of a versatile and safe method is highly desirable. Li [12] and others [13] found that sulfoxonium ylides could worked as a safer carbene precursors in various C–H activation catalyzed by Rh(III) [14], Ru(II) [15], and Co(III) [16]. Based on our previous work [17], herein, we wish to report an Ir(III)-catalyzed synthesis of 2-substituted indoles from *N*-phenylpyridin-2-amines and sulfoxonium ylides via C–H functionalization and cyclization.

Results and Discussion

We initiated our studies with N-phenylpyridin-2-amine 1a and sulfoxonium ylide 2a as model substrates in the presence of [IrCp*Cl₂]₂ (5 mol %), AgSbF₆ (20 mol %) and PivOH (1 eq) in DCE under Ar at 100 °C for 12 h. We were delighted to obtain the desired product 4a in 44% yield and byproduct 4aa in 33% yield, which was formed by the reaction of 4a and the second sulfoxonium ylide (Table 1, entry 1). It was found that silver salts had a significant influence on the reaction. To our delight, AgOAc gave the best result and increased the yield of 4a to 75% (entries 2-5). Screening of solvents revealed that DCE is the preferred solvent in this transformation (entries 5-8). Moreover, acetic acid was a better additive than PivOH, afforded the product 4a in 81% yield (entry 9). Increasing the amount of 2a could not improve the yield due to the increased amount of byproduct 4aa (entry 10). Control experiments confirmed that poor yields of 4a were obtained when additive was omitted (entry 11). Furthermore, when the directing group replaced by -COMe or -CONMe₂ the reaction was complicated (entry 12 and 13). Therefore, the optimal reaction conditions were established as 1a (0.4 mmol), 2a (0.48 mmol, 1.2 equiv), [IrCp*Cl₂]₂ (5 mol %), AgOAc (20 mol %), and HOAc (0.4 mmol, 1 eq) in DCE (3 mL) stirred at 100 °C for 12 h under Ar protection.

Table 1

Optimization of reaction conditions^a

C H. DG	+	[Cp*IrCl ₂] ₂ (5 Ag-salt, Pivi		nd DG
1a	2a	12 h	4a	4aa
Entry	[Ag]	solvent	yield of 4a(%) ^b	yield of 4aa (%) ^b
1	AgSbF ₆	DCE	44	33
2	AgOTf	DCE	44	9
3	AgBF ₄	DCE	44	37
4	AgNTf ₂	DCE	19	38
5	AgOAc	DCE	75	10
6	AgOAc	toluene	66	-
7	AgOAc	TFE	33	38
8	AgOAc	THF	31	9
9°	AgOAc	DCE	81	-
10^{d}	AgSbF ₆	DCE	34	65
11 ^e	AgOAc	DCE	45	-
$12^{\rm f}$	AgOAc	DCE	-	-
13 ^g	AgOAc	DCE	-	-

^a Reaction Conditions: **1a** (0.4 mmol), **2a** (0.48 mmol, 1.2 eq), $[Cp*IrCl_2]_2$: (5 mol %), [Ag]: (20 mol %), PivOH: (1 eq), 12 h, 100 °C, Ar. DCE = 1,2-dichloroethane, PivOH = pivalic acid, TFE = 2,2,2-trifluoroethanol, DG = 2-pyridinyl.

^b Isolated yield.

^cHOAc as an additive.

 $^{\rm f}$ DG = -COMe

 g DG = -CONMe₂

With the optimized reaction conditions in hands, we investigated the reaction of various sulfoxonium ylides (Table 2). The reaction proceeded smoothly to afford indoles in good yields with excellent functional group compatibility.

Sulfoxonium ylides bearing electron-donating or withdrawing groups at *para*-position of the aromatic ring displayed high reactivity, providing the desired products **3a–3h** in 65–95% yields. Similarly, *meta-* and *ortho-* substituted sulfoxonium ylides with methyl, methoxy, chloro, bromo, and flouro groups could also afforded desired products **3i-3o** in 77-88% yields. Moreover, disubstituted benzoyl sulfoxonium ylides were well tolerated in this transformation, providing the corresponding products **3p** and **3q** in 87% and 80% yields. Furthermore, this reaction was not limited to phenyl substrates, sulfoxonium ylides containing a thiophene ring or naphthalene gave **3r** and **3s** in 73% and 87% yields respectively. Comparably, sulfoxonium ylides with alkyl substituents showed better reactivity to deliver the corresponding products **3t-3w** in 56%-98% yields.

Table 2

Scope of substituted sulfoxonium ylides^a.



^a Reaction Conditions: **1a** (0.4 mmol), **2** (0.48 mmol, 1.2 eq), [Cp*IrCl₂]₂(5 mol %), AgOAc (20 mol %), HOAc (1 eq), 12 h, 100 °C, Ar.

After exploration of sulfoxonium ylides, we investigated the scope and limitation of N-phenylpyridin-2-amines (Table 3). N-Phenyl-2-aminopyridines bearing electron-donating and withdrawing groups at para position of benzene ring all reacted smoothly with sulfoxonium ylides with high efficiency and gave corresponding 2-arylindoles 4b-4e in 66-85% yields. Accordingly, good to high isolated yields obtained for meta-substituted substrates 4f-4h indicated full tolerance of various substituents. In sharp contrast, ortho-methylaniline were ineffective substrates in this reaction, giving 4i and 4j in 32% and 22% yields. Luckily, the yield of 4j could improved to 66% by prolonging the reaction time to 24 h. Moreover, disubstituted aniline also well tolerated and gave coupling product 4k in 67% yield. To demonstrate the synthetic utility reaction of this protocol, а scale-up of *N*-phenylpyridin-2-amine **1a** and **2a** was carried out under the optimized conditions, giving corresponding product 4a in 82% yield (Scheme 2).

^d **2a** (0.88 mmol, 2.2 eq). ^e No additive

Table 3

Scopes of substituted N-phenylpyridin-2-amines^a



^a Reaction Conditions: **1** (0.4 mmol), **2a** (0.48 mmol, 1.2 eq), [Cp*IrCl₂]₂ (5 mol %), AgOAc (20 mol %), HOAc (1 eq), 12 h, 100 °C, Ar. ^b24 h.



Scheme 2 Gram-scale reaction.



Scheme 3 Mechanistic studies.

For gaining insight into the reaction mechanism, the kinetic isotope effect (KIE) has been measured [18]. The KIE value of intermolecular competition ($k_H/k_D = 1.2$) and parallel reactions ($k_H/k_D = 1.1$) indicated that C–H activation is possibly not the rate-determine step in this transformation (scheme 3a and 3b). Moreover, a competition experiment has been performed using substrates bearing different electronic properties [15d]. The reaction of **1b/1d** afforded the corresponding products **4b/4d** in 1.88:1 ratio based on HPLC analysis, indicating that an electron-donating group tended to favor the reaction (Scheme 3c).

Based on the precedent literature reports [19], a plausible mechanistic pathway was given in Scheme 4. Initially, a pyridinyl directing group on **1a** coordinated to a Ir(III) catalyst,

subsequent C–H cleavage generated a six-membered iridacycle intermediate I. Then the coordination of the sulfoxonium ylide **2a** reacted with I and followed by the elimination of DMSO to produce a metal-carbenoid intermediate II. The iridium (III) intermediate III was generated by migratory insertion of the Ir-aryl bond into the activated carbine, which then underwent protonation to generate intermediate IV and Cp*Ir(III) species for the next catalytic cycle. Finally, intermediate IV underwent intramolecular nucleophilic cyclization and dehydration to furnish the final product indole **4a**.



Scheme 4. Proposed Reaction Mechanism

In conclusion, we disclosed a Cp*Ir(III)-catalyzed highly efficient coupling reaction of anilines with sulfoxonium ylides by using pyridinyl as a directing group. The reaction provides a concise synthetic strategy to 2-substituted indoles with good to excellent yields under mild conditions. The further investigations on its synthetic applications are currently underway in our laboratory.

Acknowledgements

This work was supported by the National Natural Science Foundation of China (NSFC) (Grant No. 21676252 and 21506191).

Supplementary data

Supplementary data associated with this article can be found, in the online version, at https://

References

- (a) Bandini, M.; Eichholzer, A. Angew. Chem., Int. Ed. 48 (2009) 9608-9644;
- (b) Dolle, R. E.; Bourdonnec, B. L.; Worm, K.; Morales, G. A.; Thomas, C. J.; Zhang, W. J. Comb. Chem. 12 (2010) 765-806;
 (c) Zhang, Z.-Z.; Liu, B.; Wang, C.-Y.; Shi, B.-F. Org. Lett. 17 (2015) 4094-4097;
 (d) Liu C. L. Z. Theorem V. Liu M. Chem. Domain (2010)
- (d) Li, L.; Chen, Z.; Zhang, X.; Jia, Y. Chem. Rev. 118 (2018) 3752-3832.
- [2] (a) Lloyd, D. G.; Buenemann, C. L.; Todorov, N. P.; Manallack, D. T.; Dean, P. M. J. Med. Chem. 47 (2004) 493-496;
 (b) Li, D.; Yang, J.; Ma, H.; Sun, C.; Feng, R. J. Cell. Biochem. 119 (2018) 9899-9909.

[3] (a) Bonelli, J. Int. J. Clin. Pharmacol., Ther. Toxicol. 22 (1984) 189-193;

(b) Arisawa, M.; Kasaya, Y.; Obata, T.; Sasaki, T.; Nakamura, T.;Araki, T.; Yamamoto, K.; Sasaki, A.; Yamano, A.; Ito, M.; Abe, H.; Ito, Y.; Shuto, S. J. Med. Chem. 55 (2012) 8152-8163;

- (c) Mugnaini, C.; Rabbito, A.; Brizzi, A.; Palombi, N.; Petrosino, S.; Verde, R.; Di Marzo, V.; Ligresti, A.; Corelli, F. Eur. J. Med. Chem. 161 (2019) 239-251.
- [4] (a) Shen, M.; Leslie, B. E.; Driver, T. G. Angew. Chem. Int. Ed. Engl. 47 (2008) 5056-5059;
 - (b) Stuart, D. R.; Alsabeh, P.; Kuhn, M.; Fagnou, K. J. Am. Chem. Soc. 132 (2010) 18326-18339;
 - (c) Nguyen, Q.; Sun, K.; Driver, T. G. J. Am. Chem. Soc. 134 (2012) 7262-7265.
- [5] (a) Tan, Y.; Hartwig, J. F. J. Am. Chem. Soc. 132 (2010) 3676-3677;
- (b) Shi, Z.; Glorius, F. Angew. Chem., Int. Ed. 51 (2012) 9220-9222.[6] (a) Ackermann, L.; Lygin, A. V. Org. Lett. 14 (2012) 764-767;
- (b) Manna, M. K.; Bhunia, S. K.; Jana, R. Chem. Commun. 53 (2017) 6906-6909;

(c) Manna, M. K.; Bairy, G.; Jana, R. J. Org. Chem. 83 (2018) 8390-8400;

(d) Nareddy, P.; Jordan, F.; Szostak, M. Org. Lett. 20 (2018) 341-344.

- [7] (a) Song, W.; Ackermann, L. Chem. Commun. 49 (2013) 6638-6640;
 (b) Zhao, T.-T.; Xu, W.-H.; Zheng, Z.-J.; Xu, P.-F.; Wei, H. J. Am. Chem. Soc. 140 (2018) 586-589.
- [8] (a) Bernini, R.; Fabrizi, G.; Sferrazza, A.; Cacchi, S. Angew. Chem., Int. Ed. 48 (2009) 8078-8081;
 - (b) Wang, Y.; Ye, L.; Zhang, L. Chem. Commun. 47 (2011) 7815-7817;

(c) Zhu, X.; Su, J.-H.; Du, C.; Wang, Z.-L.; Ren, C.-J.; Niu, J.-L.; Song, M.-P. Org. Lett. 19 (2017) 596-599.

- [9] Cajaraville, A.; López, S.; Varela, J. A.; Saá, C. Org. Lett. 15 (2013) 4576-4579.
- [10] Manna, M. K.; Hossian, A.; Jana, R. Org. Lett. 17 (2015) 672-675.
- [11] (a) Mishra, N. K.; Choi, M.; Jo, H.; Oh, Y.; Sharma, S.; Han, S. H.; Jeong, T.; Han, S.; Lee, S.-Y.; Kim, I. S. Chem. Commun. 51 (2015) 17229-17232;

(b) Jie, L.; Wang, L.; Xiong, D.; Yang, Z.; Zhao, D.; Cui, X. J. Org. Chem. 83 (2018) 10974-10984.

[12] (a) Xu, Y.; Yang, X.; Zhou, X.; Kong, L.; Li, X. Org. Lett. 19 (2017) 4307-4310;

(b) Xu, Y.; Zhou, X.; Zheng, G.; Li, X. Org. Lett. 19 (2017) 5256-5259;

(c) Xu, Y.; Zheng, G.; Yang, X.; Li, X. Chem. Commun. 54 (2018) 670-673;

(d) Hu, P.; Zhang, Y.; Liu, B.; Li, X. Org. Chem. Front. 5 (2018) 3263-3266.

[13] (a) Barday, M.; Janot, C.; Halcovitch, N. R.; Muir, J.; Aissa, C. Angew. Chem., Int. Ed. 56 (2017) 13117-13121; (b) Oh, H.; Han, S.; Pandey, A. K.; Han, S. H.; Mishra, N. K.; Kim, S.; Chun, R.; Kim, H. S.; Park, J.; Kim, I. S. J. Org. Chem. 83 (2018) 4070-4077;

(c) Halskov, K. S.; Witten, M. R.; Hoang, G. L.; Mercado, B. Q.; Ellman, J. A. Org. Lett. 20 (2018) 2464-2467.

[14] (a) Xing, Y.-Y.; Liu, J.-B.; Sun, C.-Z.; Huang, F.; Chen, D.-Z. J. Org. Chem. 83 (2018) 4545-4553;

(b) Wu, X.; Xiong, H.; Sun, S.; Cheng, J. Org. Lett. 20 (2018) 1396-1399;

(c) Hu, P.; Zhang, Y.; Xu, Y.; Yang, S.; Liu, B.; Li, X. Org. Lett. 20 (2018) 2160-2163;

(d) You, C.; Pi, C.; Wu, Y.; Cui, X. Adv. Synth. Catal. 360 (2018) 4068-4072;

(e) Zhou, C.; Fang, F.; Cheng, Y.; Li, Y.; Liu, H.; Zhou, Y. Adv. Synth. Catal. 360 (2018) 2546-2551.

[15] (a) Liang, Y.-F.; Yang, L.; Rogge, T.; Ackermann, L. Chem. - Eur. J. 24 (2018) 16548-16552;

(b) Neuhaus, J. D.; Bauer, A.; Pinto, A.; Maulide, N. Angew. Chem., Int. Ed. 57 (2018) 16215-16218;

(c) Shi, X.; Wang, R.; Zeng, X.; Zhang, Y.; Hu, H.; Xie, C.; Wang, M. Adv. Synth. Catal. 360 (2018) 4049-4053;

(d) Xie, H.; Lan, J.; Gui, J.; Chen, F.; Jiang, H.; Zeng, W. Adv. Synth. Catal. 360 (2018) 3534-3543.

- [16] Ji, S.; Yan, K.; Li, B.; Wang, B. Org. Lett. 20 (2018) 5981-5984.
- [17] (a) Jiang, X.; Chen, J.; Zhu, W.; Cheng, K.; Liu, Y.; Su, W.-K.; Yu, C. J. Org. Chem. 82 (2017) 10665-10672;
 (b) Jiang, X.; Wang, X.; Huang, X.; Li, G.; Yu, C. J. Saudi Chem. Soc. 21 (2017) 587-592;

(c) Zhang, L.; Zheng, X.; Chen, J.; Cheng, K.; Jin, L.; Jiang, X.; Yu, C. Org. Chem. Front. 5 (2018) 2969-2973;

(d) Jiang, X.; Li, G.; Yu, C. Tetrahedron Lett. 59 (2018) 1506-1510; (e) Jiang, X.; Zheng, C.; Lei, L.; Lin, K.; Yu, C. Eur. J. Org. Chem.

(e) Jung, X., Zheng, C., Lei, E., Ehi, K., Tu, C. Eur. J. Org. Chem 2018 (2018) 1437-142;

- (f) Jiang, X.; Zhu, B.; Lin, K.; Wang, G.; Su, W.-k.; Yu, C. Org. Biomol. Chem. 17 (2019) 2199-2203.
- [18] (a) Wang, Q.; Wang, F.; Yang, X.; Zhou, X.; Li, X. Org. Lett. 18 (2016) 6144-6147;
 (a) Wang, Q.; Wang, F.; Yang, X.; Zhou, X.; Li, X. Org. Lett. 18 (2016) 6144-6147;

(b) Wang, F.; Jin, L.; Kong, L.; Li, X. Org. Lett. 19 (2017) 1812-1815.

[19] (a) Mishra, N. K.; Choi, M.; Jo, H.; Oh, Y.; Sharma, S.; Han, S. H.; Jeong, T.; Han, S.; Lee, S.-Y.; Kim, I. S. Chem. Commun. 51 (2015) 17229-17232;

(b) Tang, G.-D.; Pan, C.-L.; Li, X. Org. Chem. Front. 3 (2016) 87-90;
(c) Karmakar, U.; Das, D.; Samanta, R. Eur. J. Org. Chem. 2017 (2017) 2780-2788;

(d) Zheng, G.; Tian, M.; Xu, Y.; Chen, X.; Li, X. Org. Chem. Front. 5 (2018) 998-1002;

(e) Zhou, X.; Xia, J.; Zheng, G.; Kong, L.; Li, X. Angew. Chem., Int. Ed. 57 (2018) 6681-6685.

- **1.** Ir(III)-catalyzed C–H activation synthesize 2-substituted indoles efficiently.
- 2. Various functional groups tolerated and gave 2-substituted indoles in good yields.
- 3. KIE experiments were conducted to probe the possible mechanism. Acception

3

Graphical Abstract

To create your abstract, type over the instructions in the template box below. Fonts or abstract dimensions should not be changed or altered.

2

CCE

4