

Regio- and Stereoselective Hydrosilylation of 1,3-Enynes Catalyzed by Palladium

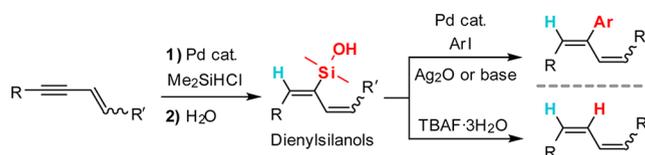
Hui Zhou and Christina Moberg*

Department of Chemistry, Organic Chemistry, School of Chemical Science and Engineering, KTH Royal Institute of Technology, SE 100 44 Stockholm, Sweden

kimo@kth.se

Received January 16, 2013

ABSTRACT



In the presence of Pd(0) and a phosphine, hydrosilylation of 1,3-enynes with Me_2SiHCl proceeds to yield dienyloxy silanols with the silicon function added to the internal alkyne carbon atom and with (*E*)-configuration of newly formed olefinic bond. The silanols isolated after hydrolysis of the primarily obtained products serve as precursors to conjugated dienes with different substitution patterns.

Hydrometalations transform alkynes to reactive vinyl-metal or vinylmetalloid species, which are important building blocks in organic synthesis.¹ Hydrosilylations of alkynes are of particular interest since the product vinylsilanes have low toxicity, are easy to handle, and are tolerant to a wide range of reaction conditions. Vinylsilanes with a silicon–heteroatom bond are attractive reagents, which, for example, can participate as reactive nucleophilic partners in palladium-catalyzed cross-coupling reactions² and be subjected to Tamao–Fleming oxidations.³ The stereo- and regiochemistry of the hydrosilane addition is a function of the substrate, the catalyst, which most commonly is a platinum, ruthenium, or cationic rhodium complex, and to some extent the reaction conditions.⁴ Whereas terminal alkynes under appropriate conditions react with high stereo- and site selectivity, either in a syn or an anti fashion,

reactions with internal alkynes often result in mixtures of products, unless a directing group⁵ is present in the substrate.

Hydrometalated 1,3-enynes are valuable precursors for stereodefined 1,3-dienes. We are, however, aware of only a few examples of hydrosilylations of this type of compounds. Regio- and stereoselective platinum-catalyzed *cis*⁶ and ruthenium-catalyzed *trans*⁷ hydrosilylations have been used in natural product syntheses. Enynes with a terminal triple bond have been shown to yield dienes with the silicon function in terminal position,⁸ whereas an internal dienyne gave a mixture of regioisomers.⁹ 1,3-Enynes equipped with a propargylic alcohol function were shown to be subject to site-selective *trans* addition of triethylsilane in the presence of a ruthenium complex, but enynes lacking alcohol function were resistant to the reaction.¹⁰ The most extensively

(1) (a) Munslow, I. J. In *Modern Reduction Methods*; Andersson, P. G., Munslow, I. J., Eds.; Wiley-VCH: Weinheim, 2008; pp 363–385. (b) Trost, B. M.; Ball, Z. T. *Synthesis* **2005**, 853–887.

(2) (a) Hiyama, T. *J. Organomet. Chem.* **2002**, *653*, 58–61. (b) Denmark, S. E.; Regens, C. S. *Acc. Chem. Res.* **2008**, *41*, 1486–1499. (c) Nakao, Y.; Hiyama, T. *Chem. Soc. Rev.* **2011**, *40*, 4893–4901. (d) Chang, W.-T. T.; Smith, R. C.; Regens, C. S.; Bailey, A. D.; Werner, N. S.; Denmark, S. E. *Org. React.* **2011**, *75*, 213–745.

(3) (a) Fleming, I.; Henning, R.; Plaut, H. *J. Chem. Soc., Chem. Commun.* **1984**, 29–31. (b) Tamao, K.; Ishida, N.; Tanaka, T.; Kumada, M. *Organometallics* **1983**, *2*, 1694–1696.

(4) Lim, D. S. W.; Anderson, E. A. *Synthesis* **2012**, *44*, 983–1010.

(5) (a) Trost, B. M.; Ball, Z. T.; Laemmerhold, K. M. *J. Am. Chem. Soc.* **2005**, *127*, 10028–10038. (b) Kawasaki, Y.; Igawa, K.; Tomooka, K. *J. Am. Chem. Soc.* **2011**, *133*, 20712–20715.

(6) Gallenkamp, D.; Fürstner, A. *J. Am. Chem. Soc.* **2011**, *133*, 9232–9235.

(7) Micoine, K.; Fürstner, A. *J. Am. Chem. Soc.* **2010**, *132*, 14064–14066.

(8) Bergueiro, J.; Montenegro, J.; Cambeiro, F.; Saá, C.; López, S. *Chem.—Eur. J.* **2012**, *18*, 4401–4410.

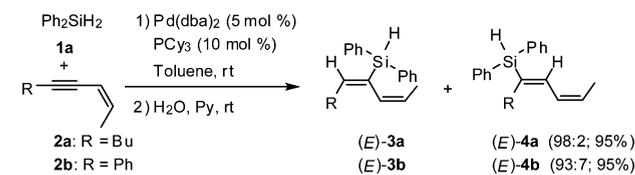
(9) Bergueiro, J.; Montenegro, J.; Cambeiro, F.; Saá, C.; López, S. *Chem.—Eur. J.* **2012**, *18*, 14100–14107.

(10) (a) Trost, B. M.; Ball, Z. T.; Jöge, T. *Angew. Chem., Int. Ed.* **2003**, *42*, 3415–3418. (b) Trost, B. M.; Ball, Z. T. *J. Am. Chem. Soc.* **2005**, *127*, 17644–17655.

studied substrates are silicon-substituted butenynes;¹¹ other examples are rare.¹² The desired dienylsilanes have instead commonly been prepared by indirect methods, such as silylation of dienyilmagnesium compounds,¹³ ring-opening of silylated sulfolenes,¹⁴ electrophilic addition to bisilylated 2-butyne,¹⁵ Horner–Emmons reaction of the anion of silylated 2-propenylphosphonate,¹⁶ and Suzuki coupling of an α -iodo vinylsilane.¹⁷

Intrigued by the lack of more general studies, we decided to investigate hydrosilylations of variously substituted 1,3-enynes, thereby getting access to less substituted 1,3-dienes than obtained via our previous silaborations of the same substrates.¹⁸ Addition of diphenylsilane to (*Z*)-non-2-en-4-yne (**2a**) and (*Z*)-5-phenyl-2-en-4-yne (**2b**) in the presence of Pd(dba)₂ and PCy₃ was first attempted.¹⁹ We were pleased to note that addition to the alkyne function occurred with high regioselectivity to yield the products of pure *cis*-addition (Scheme 1), as evidenced by NOE (see the Supporting Information). In the absence of phosphine ligand, low yields of products were obtained (7% from **2a**).

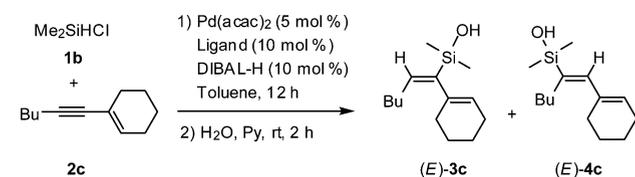
Scheme 1. Palladium-Catalyzed Addition of Diphenylsilane to 1,3-Enynes



After these preliminary attempts, we turned our attention to heterosubstituted silanes, which are crucial as nucleophilic partners in palladium-catalyzed cross-coupling reactions.²⁰ The conditions which proved to be successful for additions described in Scheme 1 unfortunately resulted in no or unsatisfactory yields of products when diphenylsilane was exchanged for chlorodimethylsilane, even at elevated temperature (80 °C, see the Supporting Information,

Table S1). However, improved results were observed in the presence of Pd(0) catalysts obtained by in situ reduction of Pd(II). Thus, Pd(acac)₂ was reduced with DIBAL-H in the presence of a phosphine prior to the addition of enyne and silane. After completed reaction, water and pyridine were added, which resulted in hydrolysis to the desired silanol. In order to find the most suitable reaction conditions, various ligands were screened in the reaction with enyne **2c** (Table 1, entries 1–5). Use of triphenylphosphine and diphenylcyclohexylphosphine resulted in poor yields, most likely due to inefficient oxidative addition of the silane to Pd(0) (Table 1, entries 1 and 2). Highest site selectivity, in combination with a high yield, was observed with electron-rich PEt₃ (Table 1, entry 4). A temperature of 80 °C was required, as lower temperature resulted in lower yields (Table 1, entries 6–8).

Table 1. Screening of Reaction Conditions for Hydrosilylation of **2c**



entry	temp (°C)	ligand	yield ^a (%)	3c : 4c ^b
1	80	PPh ₃	10	87:13
2	80	PPh ₂ Cy	15	84:16
3	80	PCy ₃	92	84:16
4	80	PEt ₃	90	93:7
5	80	SPhos	61	75:25
6	20	PEt ₃	14	93:7
7	40	PEt ₃	50	93:7
8	60	PEt ₃	72	93:7

^a Isolated yield. ^b Determined by ¹H NMR analysis of crude reaction mixture.

The high preference for formation of the product with the silicon function at the internal position is probably a result of electronic factors and can be explained by the preference for addition of palladium to the allylic position, assuming that the reaction proceeds via the Chalk–Harrod mechanism, which involves oxidative addition²¹ of the silane to the zerovalent metal center followed by insertion of the olefin into the M–H bond.²² Equally high site selectivity was observed in platinum-catalyzed hydrosilylations of conjugated ynones.²³

(21) The formation of a palladium hydride by oxidative addition of silanes has been demonstrated; see: Boyle, R. C.; Mague, J. T.; Fink, M. J. *J. Am. Chem. Soc.* **2003**, *125*, 3228–3229.

(22) Chalk, A. J.; Harrod, J. F. *J. Am. Chem. Soc.* **1965**, *87*, 16–21.

(23) (a) Rooke, D. A.; Ferreira, E. M. *J. Am. Chem. Soc.* **2010**, *132*, 11926–11928. (b) Rooke, D. A.; Ferreira, E. M. *Angew. Chem., Int. Ed.* **2012**, *51*, 3225–3230. (c) Sumida, Y.; Kato, T.; Yoshida, S.; Hosoya, T. *Org. Lett.* **2012**, *14*, 1552–1555.

(24) A decrease in catalyst loading from 5 to 1 mol % resulted in a decreased yield (70% instead of 90% for the reaction in entry 1) under otherwise identical conditions.

(11) (a) Kusumoto, T.; Hiyama, T. *Chem. Lett.* **1985**, 1405–1408. (b) Kusumoto, T.; Ando, K.; Hiyama, T. *Bull. Chem. Soc. Jpn.* **1992**, *65*, 1280–1290. (c) Maruyama, Y.; Yoshiuchi, K.; Ozawa, F.; Wakatsuki, Y. *Chem. Lett.* **1997**, 623–624. (d) Maruyama, Y.; Yoshiuchi, K.; Ozawa, F. *J. Organomet. Chem.* **2000**, *609*, 130–136.

(12) Lacombe, F.; Radkowski, K.; Seidel, G.; Fürstner, A. *Tetrahedron* **2004**, *60*, 7315–7324.

(13) (a) Kawakami, Y.; Ichitani, M.; Kunisada, H.; Yuki, Y. *Polym. J.* **1996**, *28*, 513–519. (b) Sando, F.; Murata, J.; Endo, T. *J. Polym. Sci., Part A: Polym. Chem.* **1998**, *36*, 2037–2042. (c) Junker, C. S.; Welker, M. E.; Day, C. S. *J. Org. Chem.* **2010**, *75*, 8155–8165.

(14) Chou, T.-s.; Tso, H.-H.; Tao, Y.-T.; Lin, L. C. *J. Org. Chem.* **1987**, *52*, 244–246.

(15) Pornet, J. *J. Organomet. Chem.* **1988**, *340*, 273–282.

(16) Christov, V. C. *Phosphorus, Sulfur* **1998**, *132*, 79–83.

(17) Frank, S. A.; Chen, H.; Kunz, R. K.; Schnaderbeck, M. J.; Roush, W. R. *Org. Lett.* **2000**, *2*, 2691–2694.

(18) Zhou, H.; Moberg, C. *J. Am. Chem. Soc.* **2012**, *134*, 15992–15999.

(19) Pd₂(dba)₃·CHCl₃/PCy₃ has previously been used for the hydrosilylation of alkynes; see: Motoda, D.; Shinokubo, H.; Oshima, K. *Synlett* **2002**, 1529–1531.

(20) (a) Hirabayashi, K.; Kawashima, J.; Nishihara, Y.; Mori, A.; Hiyama, T. *Org. Lett.* **1999**, *1*, 299–301. (b) Denmark, S. E.; Wehrli, D. *Org. Lett.* **1999**, *1*, 565–568.

Table 2. Palladium-Catalyzed Hydrosilylation of Various 1,3-Enynes to Silanols with Dimethylchlorosilane

Me_2SiHCl (1b) + $\text{R}^1\text{C}\equiv\text{C}-\text{C}(\text{R}^2)=\text{C}(\text{R}^3)-\text{R}^4$ (2)

 1) $\text{Pd}(\text{acac})_2$ (5 mol %), PEt_3 (10 mol %), DIBAL-H (10 mol %)

 2) H_2O , Py , rt, 2 h

 Toluene, 80 °C, 12 h

Products: (E)-3 and (E)-4

entry	enynes, 2	product, 3	yield (%) ^a (3:4) ^b
1			90 (97:3) ^{c,d}
2			94 (99:1)
3			90 (80:20)
4			92 (93:7)
5			96 (95:5)
6			92 (96:4)
7			96 (91:9)
8			90 (95:5)
9		complex mixture	-
10		-	NR ^e
11			50 (8:92)
12		complex mixture	-
13		complex mixture	-
14			95 (99:1)

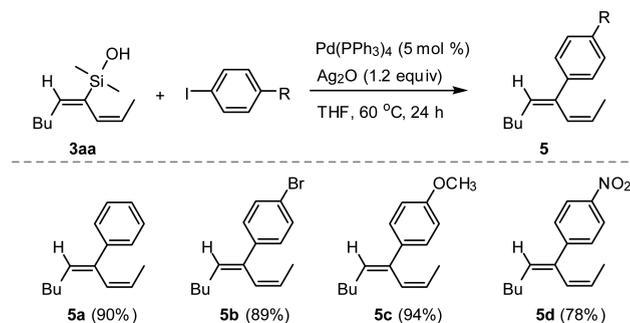
^a Isolated yield. ^b Determined by ¹H NMR analysis of crude reaction mixture. ^c Replacement of $\text{Pd}(\text{acac})_2$ by $\text{Pd}(\text{OAc})_2$ gave identical results. ^d 0.15 mmol enyne was used. From a reaction using 1.5 mmol of enyne, 94% product was isolated. ^e No reaction.

The optimized conditions were applied to hydrosilylations of a variety of enynes (Table 2).²⁴ From most reactions, high

yields of cis-addition products with the silyl function at the internal position, and with retained configuration of the original olefinic bond, were obtained (Table 2, entries 1–8). A substrate with a terminal olefin was unreactive (Table 2, entry 10), and enynes with a silicon substituent or a heteroatom in propargylic position resulted in complex mixtures (Table 2, entries 9, 12–13). However, enyne **2o**, with a heteroatom in a position more remote from the alkyne function, was smoothly transformed to the desired product (Table 2, entry 14). Reaction of an enyne with terminal alkyne function resulted in a moderate yield of an adduct with opposite site selectivity (Table 2, entry 11); this mode of addition is that commonly observed with terminal alkynes. No allene derivatives were observed in any of the reaction mixtures.

The vinylsilanols obtained proved to serve as versatile nucleophiles in palladium-catalyzed cross-couplings.^{2b} As expected, reaction with iodoarenes in the presence of $\text{Pd}(\text{PPh}_3)_4$ mediated by Ag_2O ²⁵ gave high yields of coupling products (Scheme 2).

Scheme 2. Ag_2O -Mediated Cross-Coupling



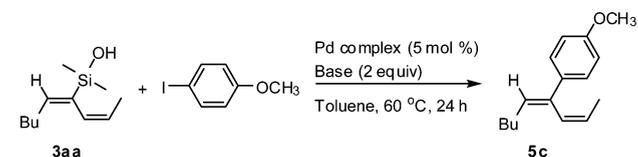
In our previous studies, we showed that the type of product obtained from palladium-catalyzed cross-coupling of 1,1-disubstituted 2-dienylsilanols was dependent on the activator used.¹⁶ Whereas Ag_2O -mediated coupling afforded 1,3-dienes, like in the above examples, coupling performed in the presence of base gave 1,2-dienes. We were therefore surprised to note that, under the same conditions, the present substrates resulted in clean conversion to 1,3-dienes (Table 3). The use of bases with sodium as counterion resulted in no or low yields (Table 3, entries 1, 3–5), even under microwave conditions (Table 3, entry 2), while those with potassium as counterion gave higher yields (Table 3, entries 6–9).²⁶

Finally, treatment of vinylsilane **3** with tetrabutylammonium fluoride resulted in desilylation (Scheme 3). The

(25) Weibel, J.-M.; Blanc, A.; Pale, P. *Chem. Rev.* **2008**, *108*, 3149–3173.

(26) A similar difference in reactivity was previously observed for coupling with aryl chlorides; see: Denmark, S. E.; Kallemeyn, J. M. *J. Am. Chem. Soc.* **2006**, *128*, 15958–15959.

(27) (a) Lu, G.-P.; Voigtritter, K. R.; Cai, C.; Lipshutz, B. H. *J. Org. Chem.* **2012**, *77*, 3700–3703. (b) Semba, K.; Fujihara, T.; Xu, T.; Terao, J.; Tsuji, Y. *Adv. Synth. Catal.* **2012**, *354*, 1542–1550.

Table 3. Base-Mediated Cross-Coupling

entry	Pd complex	base	yield (%) ^a
1	Pd(dba) ₂	NaH	NR ^b
2 ^c	Pd(dba) ₂	NaH	NR
3	Pd(dba) ₂	NaOt-Bu	NR
4	Pd(dba) ₂ /PCy ₃	NaH	NR
5	Pd(dba) ₂ /P(<i>t</i> -Bu) ₃	NaH	20
6	Pd(dba) ₂	KOt-Bu	88
7	Pd(dba) ₂	KOSiMe ₃	85
8	PEPPSI	KOt-Bu	90
9		KOt-Bu	90

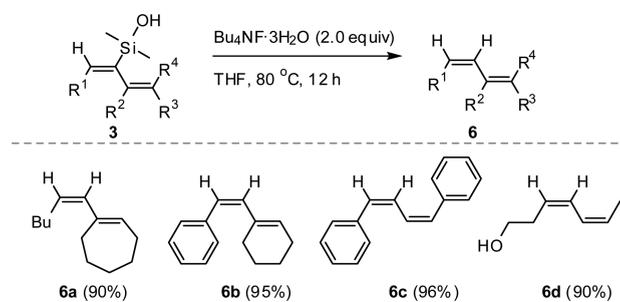
^a Isolated yield. ^b No reaction. ^c Reaction conditions: microwave, 100 °C, 2 h.

(*Z*)-configuration of the newly formed olefinic bond in **6b** was confirmed by comparison of its ¹H NMR spectrum with those previously published for the same compound²⁷ and the configuration of the remaining compounds by their ¹H coupling constants. The overall process constitutes a convenient method for the semihydrogenation of 1,3-enynes, occurring without *E*–*Z* isomerization and over-reduction, which are commonly encountered problems in this type of reductions.²⁸

In summary, we have demonstrated that dimethylchlorosilane undergoes palladium-catalyzed *cis*-addition

(28) For a recently reported selective process, see ref 23b.

(29) Negishi, E.-i.; Anastasia, L. *Chem. Rev.* **2003**, *103*, 1979–2017.

Scheme 3. Fluoride-Mediated Protodesilylation

to 1,4-disubstituted and 1,2,4-trisubstituted 1,3-enynes, conveniently accessible via Sonogashira coupling of terminal alkynes and vinyl halides,²⁹ to yield dienyloxy silanols with the silicon function at the internal position. Palladium-catalyzed coupling of the products with iodoarenes, using Ag₂O as well as base, resulted in stereo- and regiochemically defined 1,3-dienes, whereas desilylation with tetrabutylammonium fluoride resulted in overall *cis*-hydrogenation of the triple bond.

Acknowledgment. We thank Dr. Zoltan Szabo, Department of Chemistry, KTH, for making the NOE measurements. Financial support from the Carl Trygger Foundation for Scientific Research and the Swedish Research Council is gratefully acknowledged.

Supporting Information Available. Experimental procedures, compound characterization data, and spectra. This material is available free of charge via the Internet at <http://pubs.acs.org>.

The authors declare no competing financial interest.