## 102. Photo-induced Molecular Transformations

Part 1231)

One-Step Synthesis of 1*H*-Benz[f]indole-4,9-diones by a New Regioselective Photoaddition of 2-Amino-1,4-naphthoquinone with Various Alkenes and Its Application to One-Step Synthesis of Kinamycin Skeleton

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Dedicated to Prof. Kurt Schaffner on the occasion of his 60th birthday

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2,3-Dihydro-1*H*-benz[*f*]indole-4,9-diones are formed in one-step in 45-82% yields by an unprecedented [2+3]-type regioselective photoaddition of 2-amino-1,4-naphthoquinone with various electron-rich alkenes and the [2+3] adducts derived from aminonaphthoquinone with vinyl ethers and vinyl acetate to give 1*H*-benz[*f*]indole-4,9-diones including a benzindole-dione with a kinamycin skeleton in 33-72% yields. A probable pathway leading to the formation of the dihydroindole-dione involving air oxidation of an intermediary hydroquinone is proposed.

In [2], we have reported a one-step formation of 2,3-dihydronaphtho[2,3-b] furan-4,9-diones in high yields by a new [2+3]-type regioselective photoaddition of 2-hydroxy-1,4-

<sup>&</sup>lt;sup>1</sup>) Part 122: [1].

naphthoquinones with a variety of alkenes and its application to a two-step synthesis of natural quinone, maturinone.

Here, we wish to report on a new one-step synthesis of 2,3-dihydro-1H-benz[f]indole-4,9-diones and 1H-benz[f]indole-4,9-diones by a new [2+3]-type regioselective photo-addition of 2-amino-1,4-naphthoquinone (1) with a variety of cyclic and acyclic alkenes, as outlined in *Scheme 1*.

1*H*-Benz[f]indole-4,9-diones comprise an important group of heterocyclic quinones to which several physiologically active quinones, such as kinamycins [3], belong. Methods so far reported for the synthesis of this class of compounds, however, are not necessarily simple and require several reaction steps [4].

Typically, a solution of 1 [5] (85 mg, 0.49 mmol) and isobutene (2a; 0.55 g, 9.8 mmol) in benzene (70 ml) is irradiated through a *Pyrex*-filter with a 500-W high-pressure Hg arc in a  $N_2$  atmosphere for 1 h at room temperature. The usual workup and purification by preparative TLC exclusively give 2,3-dihydro-2,2-dimethyl-1*H*-benz[*f*]indole-4,9-dione (3a; 91 mg, 82%).

The photoaddition of naphthoquinone 1 with other alkenes, such as 2-methylbut-2-ene (2b), and  $\alpha$ -methylstyrene (2c), also took place regioselectively to give 2,3-dihydro-2,2,3-trimethyl-1H-benz[f]indole-4,9-dione (3b) and 2,3-dihydro-2-methyl-2-phenyl-1H-benz[f]indole-4,9-dione (3c), respectively. Photoaddition with 2c resulted in an accompanying formation of the  $2\pi + 2\pi$  adduct 5 in 23% yield.

The photoaddition of naphthoquinone 1 with vinyl ethers and vinyl acetate, such as ethyl vinyl ether (2d), 2-methoxypropene 2e, and vinyl acetate (2f), on the other hand, gave 1*H*-benz[f]indole-4,9-dione 4d-f, respectively, [4b] in 33–72% total yields under the above-mentioned conditions and the reaction scale outlined in *Scheme 1*. The results are summarized in the *Table*.

Alkene <sup>a</sup> )	Irradiation time [h]	Product <sup>b</sup> )	M.p. [°C]	Yield <sup>c</sup> ) [%]
2a	1	3a	200 (dec.)	82
2b	12	3b	205-207	66
2c	2	3c <sup>d</sup> )	162-165	45
2d	1.5	4d	297-299	33
2e	3	<b>4</b> e <sup>e</sup> )	300	72
2f	4.5	4e	(dec.)	47
2g	2.5	4g	290 (dec.)	68

Table. Results of Photoadditions of 2-Amino-1,4-naphthoquinone (1) with Alkenes

- a) The molar ratio: alkene/2-amino-1,4-naphthoquinone 1: 20.
- b) Satisfactory analytical and spectral results were obtained for all the products.
- c) Total yield.
- d)  $2\pi + 2\pi$  adduct 5 (23%) is an accompanying product.
- e) [3b]: 304-305 °C (dec.)

These benzindole-diones are formed by a spontaneous elimination of an alcohol or AcOH from the initial adducts, 2,3-dihydro-1*H*-benz[f]indole-4,9-diones **3d**-f, in the course of the reaction or separation by preparative TLC. A similar photoaddition of naphthoquinone 1 with 1-methoxycyclohexene (2g) gave 7,8,9,10-tetrahydrobenzo[b]carbazole-5,11-dione (4g), a framework of kinamycin [3], in one-step in 68% yield (Scheme 2).

## Scheme 2

No photoaddition took place with electron-deficient olefins, such as methyl methacrylate, or with *N*-substituted 2-aminonaphthoquinones, such as commercially available 2-(phenylamino)-1,4-naphthoquinone and 2-(benzylamino)-1,4-naphthoquinone [6].

The probable overall reaction path of the presented photoaddition leading to 2,3-dihydro-1H-benz[f]indole-4,9-diones is outlined in Scheme 3. The initial stage in this photochemical addition can be explained within the framework of an accepted model of  $2\pi + 2\pi$  photochemical additions (for reviews, see [7]). Irradiation of naphthoquinone 1 in benzene may well generate tautomeric excited triplet A and A'. The excited tautomer A may then form preferentially an exciplex with alkene to give a biradical  $\mathbf{B}r$  or an ionic intermediate Bi generated by an electron transfer. It seems likely that the biradicals Brhave a appreciable polar character or are ionic intermediates such as Bi. The regioselectivity found in the present addition is a strong indication of the involvement of a more stabilized biradical or ionic intermediate, such as Br and Bi, in the formation of 2,3-dihydro-1*H*-benz[f]indole-4,9-diones 3a-f. An intramolecular cyclization of the intermediate gives a hydroquinone C' analogous to the formation of 2,3-dihydronaphtho[2,3-b]furan-4,9-diones by the photoaddition of 2-hydroxy-1,4-naphthoguinones with various alkenes [2]. We were, however, not successful in isolating the corresponding diacetate C' by acetylating the hydroquinone corresponding to 3a immediately after the photoadditions. 2,3-Dihydro-1H-benz[f]indole-4,9-diones 3a-f are formed by rapid air oxidation of the hydroquinone during the workup, since we found that a yellow-colored hydroquinone corresponding to 3a, prepared by a reduction of quinone 3a by catalytic hydrogenation with Pd/C as a catalyst, turned rapidly into a purple-colored quinone 3a as soon as hydroquinone was exposed to air. A by-product 5 may be formed by a  $2\pi + 2\pi$  addition from an excited enol form A' through intermediate D and E, followed by hydrolysis of the resulting imino-ketone E.

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