Contents lists available at ScienceDirect

Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet

Synthetic studies of incednine: synthesis of C1–C13 pentaenoic acid segment

Takashi Ohtani, Hiroshi Kanda, Kensuke Misawa, Yoshifumi Urakawa, Kazunobu Toshima*

Department of Applied Chemistry, Faculty of Science and Technology, Keio University, 3-14-1 Hiyoshi, Kohoku-ku, Yokohama 223-8522, Japan

ARTICLE INFO

ABSTRACT

Article history: Received 26 January 2009 Revised 26 February 2009 Accepted 27 February 2009 Available online 4 March 2009 Stereoselective synthesis of the C1–C13 pentaenoic acid segment (**4**) of the novel antibiotic incednine (**1**) is described. © 2009 Elsevier Ltd. All rights reserved.

Keywords: Incednine Antibiotic Apoptosis Pentaene Horner–Wadsworth–Emmons olefination

In 2008, Imoto and co-workers reported the isolation of a novel antibiotic, incednine (1), from Streptomyces sp. Although determination of the configuration at the C-23 stereocenter was supported by computer modeling studies and the stereochemistry was not totally clear, they also disclosed the structural elucidation of **1**.¹ This novel natural product exhibited significant inhibitory activity against the anti-apoptotic oncoproteins Bcl-2 and Bcl-xL, with a mode of action distinct from those of other inhibitors which inhibit the binding capacity of Bcl-xL to pro-apoptotic protein Bax. These proteins are overexpressed in many cancers, resulting in the expansion of transformed populations and advancement of the multidrug-resistant stage.²⁻⁴ Therefore, **1** is expected to be a leading compound in the development of novel anti-tumor drugs, and it is also expected to be a useful tool for further study of the Bcl-2 and Bcl-xL functions. The identification of its target protein could provide a new insight into the anti-apoptotic mechanism of Bcl-2 family proteins.

Structurally, **1** has been shown to contain several unique features. It contains α -methoxy- α , β -unsaturated amide structure and independent conjugated pentaene and tetraene systems in the 24-membered macrolactam core. Furthermore, the macrolactam core is coupled with two amino sugars by β -glycosidic bonds. Because of its important biological activity and novel molecular architecture, **1** has been deemed as a prime target for total synthesis. Herein we report the stereoselective synthesis of the C1–C13 pentaenoic acid segment **4**.

In our strategy for the total synthesis of **1**, as shown in Figure 1, the C-11 glycosidic bond is constructed in the last stage of the syn-

thesis. The aglycon **2** is prepared from the pentaenoic acid segment **4**, which is prepared from ethylene glycol (**6**), and the tetraene segment **3**. The pentaene structure in **4** is constructed by various olefination reactions, and the creation of the C-10 and C-11 stereocenters is achieved by Sharpless asymmetric epoxidation⁵ of the allylic alcohol **5** (Fig. 1).

The synthesis of pentaenoic acid segment **4**, corresponding to the C1-C13 of 1, is summarized in Schemes 1 and 3. We first synthesized the known epoxide 9, referring to the procedure of Shimizu and Nakata.⁶ Ethylene glycol (**6**) was protected as a mono-PMB ether using PMBCl and KOH at 130 °C in 92% yield.⁷ The resulting primary alcohol 7 was then converted into the α,β -unsaturated ester **8** by a one-pot Swern oxidation-Wittig reaction,⁸ using Ph₃PC(Me)CO₂Et, in 99% yield with E selectivity. After reduction of 8 to the allylic alcohol 5 in 99% yield using DIBAL-H, 5 was treated with Ti(O-*i*Pr)₄, (-)-DIPT, and TBHP at -78 °C (Sharpless asymmetric epoxidation) to furnish epoxide **9** in 78% yield with 90% ee. At this stage, we examined the regioselective introduction of a hydroxyl group function to epoxide 9. After many attempts, we finally found that applying Honda's conditions⁹ using Me₄NB- $H(OAc)_3$ gave the desired diol **10** in high yield (92%). The terminal hydroxyl group of 10 was protected as a TBS ether in 94% yield, and the acetyl group was then removed using NaOMe to give diol 11 in 78% yield. It was found under these conditions that migration of the silyl group produced **11** and **12** in a ratio of 78:21. Fortunately, however, the undesired silvl ether 12 could easily be converted into the desired 11 by treatment with NaOMe in MeOH. Protection of the 1,2-diol in 11 as a cyclic acetal isopropylidene group, using Me₂C(OMe)₂ and CSA, followed by deprotection of the TBS group with TBAF and Swern oxidation of the resulting primary alcohol produced aldehyde 13 in 88% overall yield. Subsequent Wittig





^{*} Corresponding author. Tel./fax: +81 45 566 1576.

E-mail address: toshima@applc.keio.ac.jp (K. Toshima).

^{0040-4039/\$ -} see front matter \odot 2009 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2009.02.203



Figure 1. Retrosynthetic analysis of incednine (1).



Scheme 1. Reagents and conditions: (a) PMBCl, KOH, 130 °C, 3 h (Dean–Stark), then 35 °C, 14 h, 92%; (b) (COCl)₂, DMSO, Et₃N, CH₂Cl₂, -78 °C to rt, 1 h, then Ph₃PC(Me)CO₂Et, rt, 30 min, 99%; (c) DIBAL-H, PhMe, -78 °C, 20 min, 99%; (d) Ti(O*i*-Pr)₄, (-)-DIPT, TBHP/decane, MS4A, CH₂Cl₂, -20 °C, 65 h, 78% (90% ee); (e) Me₄NBH(OAc)₃, PhMe, 60 °C, 13 h, 92%; (f) TBSCl, imid., DMF, 0 °C to rt, 3 h, 94%; (g) NaOMe, MeOH, -20 °C, 24 h, 78%; (h) NaOMe, MeOH, 0 °C to rt, 75%; (i) Me₂C(OMe)₂, CSA, acetone, 0 °C, 2 h, 98%; (j) TBAF, THF, rt, 5 h, 97%; (k) (COCl)₂, DMSO, Et₃N, CH₂Cl₂, -78 °C to -20 °C, 1.5 h, 93%; (l) **14**, PhMe, 60 °C, 21 h, 97% (*E*/*Z* = 79/21); (m) PPTS, MeOH, rt to 40 °C, 4 d, 64%; (n) TESOTF, 2,6-lutidine, CH₂Cl₂, 0 °C, 20 min, 98%.

reaction of **13** with **14**¹⁰ introduced the diene structure, producing **15** in 97% yield with E/Z = 79/21. Although the E/Z isomers were inseparable at this stage, we were able to separate them at the following stage. Compound **15** was converted to the tetraene segment **16** in three steps (1: DIBAL-H reduction (93%); 2: One-pot MnO₂ oxidation-Wittig reaction using Ph₃PC(Me)CHO (74%); and 3: Horner–Wadsworth–Emmons reaction using $(i-PrO)_2P(O)CH(O-Me)CO_2Me$ (49%)). Unfortunately, all attempts to selectively remove the PMB group in **16** failed due to instability under acidic and oxidative conditions. It was also predicted that deprotection of the isopropylidene group at the C10 and C11 positions in the latter step would prove to be a difficult problem in the synthesis of **1**. Therefore, at this stage, the cyclic acetal of **15** was removed using PPTS in MeOH, and the resulting diol **17** was protected with TES groups using TESOTf and 2,6-lutidine to give **18** in 63% overall yield.

Fortunately, we found that optical resolution of diol **17** using the chiral resolving reagent (*R*)-3a-allyl-3,3a,4,5-tetrahydro-2*H*cyclopenta[*b*]furan (CPF)¹¹ developed by Nemoto et al. was effective (Scheme 2). Subjecting **17** to CPF in the presence of catalytic amounts of PPTS in PhMe led to the corresponding acetals in 99% yield as a mixture of diastereomers, **19** (94%) and **19'** (5%), which could easily be separated by flash column chromatography (CH₂Cl₂/EtOAc = 10/1 (ΔR_f = 0.089)). Removal of CPF from **19** using PTSA in MeOH gave (2*S*,3*R*)-**17** as a single enantiomer in 99% yield.

Enantiomerically pure **18**, obtained as described above, was treated with DIBAL-H to reduce the ester function, and the resulting allylic alcohol was oxidized with MnO_2 and then treated with



Scheme 2. Reagents and conditions: (a) CPF, PPTS, PhMe, rt, 1.5 h, 99%; (b) p-TsOH, MeOH, rt, 2 h, 99%.



Scheme 3. Reagents and conditions: (a) DIBAL-H, PhMe, -78 °C, 30 min, 95%; (b) MnO₂, PhMe, 40 °C, 5 h, then Ph₃PC(Me)CO₂Et, 40 °C, 16 h, 95%; (c) DDQ, CH₂Cl₂/pH 7.2 phosphate buffer (1/1), 0 °C to rt, 19 h, 72%; d) Dess–Martin periodinane, pyr., CH₂Cl₂, 0 °C to rt, 17 h, 92%; (e) (Ph₃P*CH₂I)I⁻, NaHMDS, HMPA, THF, -98 °C, 1 h, 67%; (f) DIBAL-H, PhMe, -78 °C, 10 min, 98%; (g) MnO₂, CH₂Cl₂, 40 °C, 2 h, 77%; (h) (MeO)₂P(O)CH(OMe)CO₂Me, KHMDS, 18-crown-6 ether, THF, 0 °C to rt, 1 h, 82%; (i) 1.0 M KOH aq, 1,4-dioxane, 0 °C to 30 °C, 2 h, 66%.

Ph₃P=C(Me)CO₂Et to furnish the α , β -unsaturated ester **20** in 90% overall yield. At this stage we again constructed an enol ether structure **21** in three steps (1:DIBAL-H reduction (78%); 2:MnO₂ oxidation (89%):3. Horner-Wadsworth-Emmons reaction using (MeO)₂P(O)CH(OMe)CO₂Me) (54%)). Unfortunately, deprotection of PMB group of 21 was again unsuccessful. Therefore, we next attempted to introduce a vinyl iodide structure to the α , β -unsaturated ester 20 in advance. It was found, fortunately, that removal of the PMB group in 20 with DDQ proceeded very smoothly, and Dess-Martin oxidation of the resulting alcohol gave aldehyde 22 in 66% overall yield. Wittig reaction of **22** with $(Ph_3P^+CH_2I)I^-$ in the presence of NaHMDS and HMPA in THF at -98 °C led to vinyl iodide **23** in 67% yield with Z selectivity. The α , β -unsaturated ester in 23 was reduced by DIBAL-H, and the resulting allylic alcohol was oxidized by MnO₂ to furnish aldehyde 24 in 75% overall yield. Finally. Horner-Wadsworth-Emmons olefination of 24 using (MeO)₂P(O)CH(OMe)CO₂Me in the presence of KHMDS and 18crown-6 ether¹² in THF, followed by hydrolysis using KOH, gave the C1–C13 pentaenoic acid segment 4^{13} in 54% overall yield.

In conclusion, we achieved a stereoselective synthesis of pentaenoic acid segment **4**, which is a key segment in the synthesis of incednine (**1**).

Acknowledgments

We sincerely thank Professor M. Imoto and Dr. Y. Futamura of Keio University, and Dr. Y. Takahashi of the Microbial Chemistry Research Center, for providing us with very useful information on the chemical and physical properties of incednine. We also thank ZEON Corporation for providing the chiral resolving reagent, ALBO-V, a CPF reagent. This research was supported in part by the 21st Century COE Program 'Keio Life-Conjugated Chemistry' and the High-Tech Research Center Project for Private Universities: Matching Fund Subsidy, 2006–2011, from the Ministry of Education, Culture, Sports, Science, and Technology of Japan (MEXT).

References and notes

- Futamura, Y.; Sawa, R.; Umezawa, Y.; Igarashi, M.; Nakamura, H.; Hasegawa, K.; Yamasaki, M.; Tashiro, E.; Takahashi, Y.; Akamatsu, Y.; Imoto, M. J. Am. Chem. Soc. 2008, 130, 1822.
- Tsujimoto, Y.; Finger, L. R.; Yunis, J.; Nowell, P. C.; Croce, C. M. Science 1984, 226, 1097.
- Reed, J. C.; Cuddy, M.; Slabiak, T.; Croce, C. M.; Nowell, P. C. Nature 1988, 336, 259.
- 4. Gross, A.; McDonnell, J. M.; Korsmeyer, S. J. Gene Dev. 1999, 13, 1899.
- 5. Katsuki, T.; Sharpless, K. B. J. Am. Chem. Soc. 1980, 102, 5974.
- Shimizu, T.; Kusaka, J.; Ishiyama, H.; Nakata, T. Tetrahedron Lett. 2003, 44, 4965.
 Chehade, K. A. H.; Kiegiel, K.; Isaacs, R. J.; Pickett, J. S.; Bowers, K. E.; Fierke, C.
- A.; Andres, D. A.; Spielmann, H. P. J. Am. Chem. Soc. 2002, 124, 8206.
 Labelle, M.; Morton, H. E.; Guindon, Y.; Springer, J. P. J. Am. Chem. Soc. 1988, 110,
- 4533. 9. Honda, T.; Mizutani, H. *Heterocycles* **1998**, 48, 1753.
- Buchta, A.; Andree, F. Chem. Ber. **1959**, 92, 3111.
- 1. (a) Nemoto, H. *Tetrahedron Lett.* **1994**, 35, 7785; (b) Nemoto, H.; Tsutsumi, H.;
- 11. (a) Netholo, R. *Perlandaron Lett.* **1994**, *55*, 7785, (b) Netholo, H., Sudsuhl, H., Yuzawa, S.; Peng, X.; Zhong, W.; Xie, J.; Miyoshi, N.; Suzuki, I.; Shibuya, M. *Tetrahedron Lett.* **2004**, *45*, 1667; (c) Zhong, W.; Xie, J.; Peng, X.; Kawamura, T.; Nemoto, H. *Tetrahedron Lett.* **2005**, *46*, 7451; (d) Nemoto, H.; Zhong, W.; Kawamura, T.; Kamiya, M.; Nakano, Y.; Sakamoto, K. Synlett **2007**, 2343.
- (a) Bottin-Strzalko, T.; Corset, J.; Froment, F.; Pouet, M.-J.; Seyden-Penne, J.; Simonnin, M.-P. J. Org. Chem. **1980**, 45, 1270; (b) Paterson, I.; McLeod, M. D. Tetrahedron Lett **1997**, 38, 4183.
- 13. Selected ¹H NMR (300 MHz, CDCl₃) (δ , SiMe₄; *J* Hz) data for **4**: δ 6.77 (1H, s, H3), 6.54 (1H, dd, *J* = 11.1 and 13.8 Hz, H6), 6.44 (1H, d, *J* = 13.8 Hz, H5), 6.35 (1H, dd, *J* = 10.5 and 11.1 Hz, H7), 6.33 (1H, d, *J* = 7.8 Hz, H13), 6.26 (1H, dd, *J* = 10.5 and 14.7 Hz, H8), 6.16 (1H, dd, *J* = 7.8 and 8.4 Hz, H12), 5.89 (1H, d, *J* = 14.7 Hz, H9), 4.12 (1H, d, *J* = 8.4 Hz, H11), 3.71 (3H, s, C₂-OMe), 2.13 (3H, s, C₄-Me), 1.36 (3H, s, C₁₀-Me), 0.94 and 0.93 (each 9H, t, *J* = 8.4 Hz, (*CH*₃CH₂)₃Si), 0.58 (12H, m, (CH₃CH₂)₃Si).