



## Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/gpss20>

### 2-(4-Naphtho[1,2-d]furan-2-yl-1,3-thiazol-2-yl)-ethanenitrile in Heterocyclic Synthesis

Abdou O. Abdelhamid<sup>a</sup> & Mohamed A. M. Alkhodshi<sup>a</sup>

<sup>a</sup> Department of Chemistry, Cairo University, Giza, Egypt

Published online: 01 Feb 2007.

To cite this article: Abdou O. Abdelhamid & Mohamed A. M. Alkhodshi (2005) 2-(4-Naphtho[1,2-d]furan-2-yl-1,3-thiazol-2-yl)-ethanenitrile in Heterocyclic Synthesis, *Phosphorus, Sulfur, and Silicon and the Related Elements*, 180:7, 1647-1656, DOI: [10.1080/10426500590885066](http://dx.doi.org/10.1080/10426500590885066)

To link to this article: <http://dx.doi.org/10.1080/10426500590885066>

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or

indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at <http://www.tandfonline.com/page/terms-and-conditions>

## 2-(4-Naphtho[1,2-*d*]furan-2-yl-1,3-thiazol-2-yl)-ethanenitrile in Heterocyclic Synthesis

Abdou O. Abdelhamid

Mohamed A. M. Alkhodshi

Department of Chemistry, Cairo University, Giza, Egypt

*Coumarin, thiazole, and 2,3-dihydro-1,3,4-thiadiazole derivatives were synthesized from 2-(4-naphtho[1,2-*d*]furan-2-yl-1,3-thiazol-2-yl)ethanenitrile and different reagents. Structures of the new compounds were elucidated on the basis of elemental analysis, spectral data, and alternative methods of synthesis whenever possible.*

**Keywords** 2,3-Dihydro-1,3,4-thiadiazole; coumarin; hydrazonoyl halides; thiazole

### INTRODUCTION

Recently, a large number of thiazole derivatives have been found to exhibit pharmacological activity.<sup>1,2</sup> They also were used as an anthelmintic,<sup>3</sup> fungicidal,<sup>4</sup> and antifungal activity, inhibiting in vivo the growth of *Xanthomonas oryzae*,<sup>5</sup> an ingredient of herbicides.<sup>6</sup> As an extension of our study<sup>7–11</sup>, we report herein the synthesis of different thiadiazoles, coumarines, and pyrazolo[1,5-*a*]pyrimidines containing thiazole moiety.

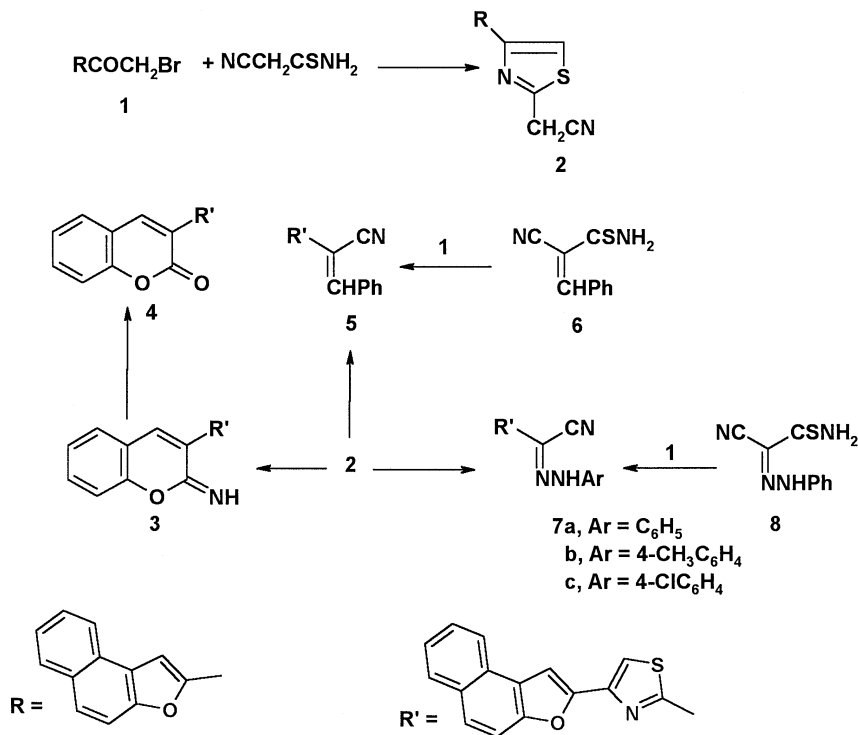
### RESULTS AND DISCUSSION

2-Bromo-1-naphtho[1,2-*d*]furan-2-ylethan-1-one (**1**) reacted with cyanothioacetamide in boiling ethanol under reflux and afforded 2-(4-naphtho[1,2-*d*]furan-2-yl-1,3-thiazol-2-yl)ethanenitrile (**2**). Structure **2** was confirmed on the basis of elemental analysis, spectral data, and chemical transformation. <sup>1</sup>H NMR ( $\delta$  ppm) spectrum showed signals at  $\delta$  = 4.12 (s, 2H, CH<sub>2</sub>CN), 7.51–8.19 (m, 8H, thiazole C-5, and ArH's). Its IR (cm<sup>-1</sup>) revealed bands at 3101 (C–H), 2252 (CN) and 1620 (C=C).

Received June 17, 2004; accepted August 12, 2004.

Address correspondence to Abdou O. Abdelhamid, Department of Chemistry, Faculty of Science, Cairo University, Giza 12316, Egypt. E-mail: Abdou.abdelhamid@yahoo.com

Compound **2** reacted with each of salicylaldehyde and benzaldehyde to give 3-(4-naphtho[1,2-*d*]furan-2-yl-1,3-thiazol-2-yl)-2H-chromen-2-imine (**3**) and (2*E*)-2-(4-naphtho[1,2-*d*]furan-2-yl(1,3-thiazol-2-yl))-3-phenylprop-2-enenitrile (**5**), respectively (Scheme 1).



### SCHEME 1

Structures of **3** and **5** were established on the basis of elemental analysis, spectral data, and chemical transformation (or alternative route synthesis).  $^1\text{H}$  NMR ( $\delta$  ppm) spectrum of **3** showed signals at  $\delta = 6.68\text{--}7.41$  (m, 13H, ArH's) and 8.42 (s, 1H). Its IR ( $\text{cm}^{-1}$ ) spectrum revealed bands at 3124 (NH), 3057 (CH), and 1599 ( $\text{C}=\text{C}$ ). More evidence on structure **3** came from its conversion to 3-(4-naphtho[1,2-*d*]furan-2-yl-1,3-thiazole-2-yl)-2H-chromen-2-one (**4**) by hydrochloric acid. IR spectrum of **4** revealed bands at 3055 (CH), 1712 (CO), and no bands between 3150–3500 due the absence of NH group.  $^1\text{H}$  NMR ( $\delta$  ppm) spectrum of **5** showed one signal at  $\delta = 7.00\text{--}7.41$  (m, ArH's). Its IR ( $\text{cm}^{-1}$ ) spectrum revealed bands at 3057 (CH), 2214 (CN), and 1604 ( $\text{C}=\text{C}$ ). Thus, compound 2-(aminothioxomethyl)-3-phenylprop-2-enenitrile (**6**) reacted with naphtho[1,2-*d*]furan-2-ylethan-1-one (**1**) in boiling ethanol

to give a product identical in all respects (mp., mixed mp., and spectra) with **5**.

Treatment of **2** with arenediazonium chloride in ethanolic sodium acetate solution at 0°C afforded (2E)-3-aza-3-(arylamino)-2-naphtho[1,2-*d*]furan-2-yl(1,3-thiazol-2-yl))-2-enenitrile **7a-c**, respectively. Structure **7** was confirmed on the basis of elemental analysis, spectral data, and chemical transformation. Thus, <sup>1</sup>H NMR spectrum of **7b** showed signals at  $\delta$  = 2.37 (s, 3H), 7.26–8.20 (m, 12H), and 14.05 (s, br., 1H). Its IR revealed bands at 3349 (NH) and 2221 (CN). Its <sup>13</sup>C NMR showed signals at 24, 102, 111, 115, 116, 119, 121, 124, 127, 128, 129, 130, 140, 143, 153, 155 ppm. Also, treatment of the appropriate (2E)-2-aminothioxomethyl)-3aza)-3-(arylamino)prop-2-enenitrile **8a-c** with 2-Bromo-1-naphtho[1,2-*d*]furan-2-ylethan-1-one (**1**) in boiling acetic acid under reflux afford products identical in all respects (mp., mixed mp., and spectra) with **7a-c**.

Compound **2** reacted with phenyl isothiocyanate in *N,N*-dimethylformamide containing potassium hydroxide to give non-isolable product which converted to a product by hydrochloric acid. The resulting product seems to be one of the four isomers **9a-d** (cf. Scheme 2). According to M.O. calculation, using Hyper Chem. AM1 semi-empirical Method, the total energy proved that the most stable isomer formulated was 2-(4-naphtho[1,2-*d*]furan-2-yl(1,3-thiazol-2-yl))-3-(phenylamino)-3-thioxopropanenitrile (**9B**). Also, structure **9** was elucidated by microanalytical, spectral data, and chemical transformation. Thus, treatment of **9** with iodomethane afforded (2E)-3-methylthio-2-(4-naphtho[1,2-*d*]furan-2-yl(1,3-thiazol-2-yl))-3-(phenylamino)prop-2-enenitrile (**10**) (cf. Scheme 3).

IR spectrum of **10** revealed characteristic bands at 3386 (NH) and 2198 (CN). Its <sup>1</sup>H NMR showed signals at  $\delta$  = 2.27 (s, 3H), 7.31–8.10 (m, 13H), and 12.53 (s, br., 1H). <sup>13</sup>C NMR showed signals at  $\delta$  = 16, 71, 102, 111, 115, 116, 118, 122, 127, 129, 130, 143, 144, 150, 153, 155, and 167 ppm. Also, compound **10** reacted with hydrazine hydrate in boiling ethanol under reflux gave [5-Amino-4-(4-naphtho[1,2-*d*]furan-2-yl)(1,3-thiazol-2-yl)]pyrazol-3-yl]phenyl amine (**11**). Structure **11** was elucidated by elemental analysis and spectral data (cf. Experimental).

On the other hand, the appropriate hydrazonoyl halides **12a-h** reacted with **9** in ethanolic triethylamine at room temperature to give 2,3-dihydro-1,3,4-thiadiazole derivatives **15a-h**, respectively (Scheme 3). IR spectra of **15a-h** revealed bands near 2190 (CN) and 1720–1680 (CO). <sup>1</sup>H NMR spectrum of **15a** showed signals at  $\delta$  = 4.15 (s, 3H) and 7.52–8.31 (m, 13H) ppm. Its <sup>13</sup>C NMR spectrum showed signals at  $\delta$  = 14, 61, 89, 115, 119, 118, 121, 122, 127, 128, 129, 130, 146, 149, 150, 153, 154, 155, 162, and 163 ppm.



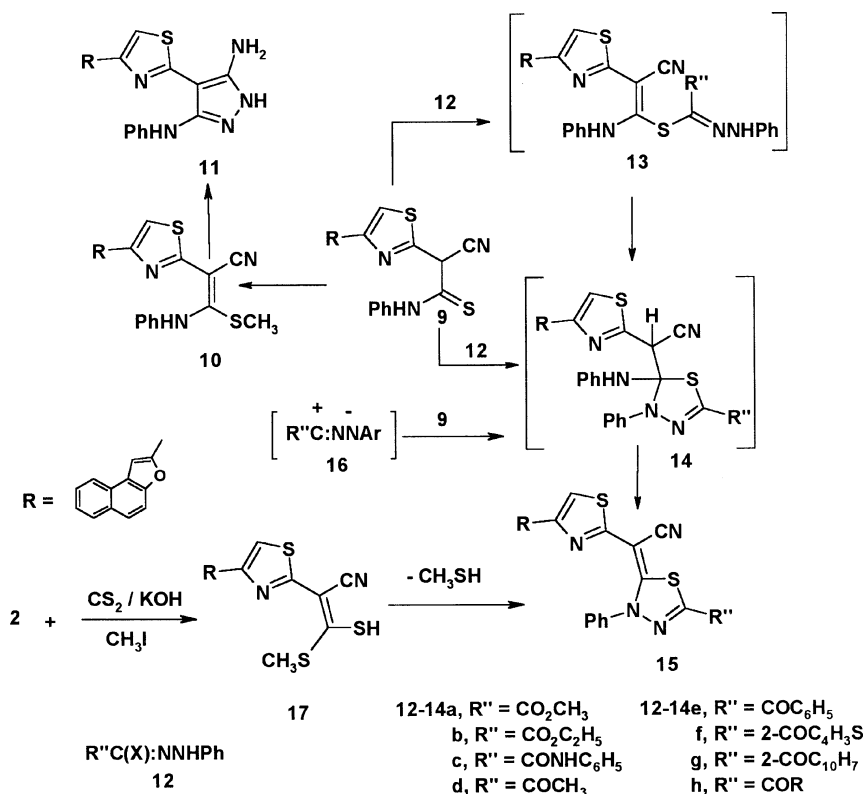
Downloaded by [Baskent Universitesi] at 23:36 19 December 2014

Downloaded by [Baskent Universitesi] at 23:36 19 December 2014

Downloaded by [Baskent Universitesi] at 23:36 19 December 2014

## Downloaded by [Baskent Universitesi] at 23:36 19 December 2014

Downloaded by [Baskent Universitesi] at 23:36 19 December 2014



SCHEME 3

FT-IR 8201 PC spectrophotometer.  $^1\text{H}$  NMR spectra were recorded in  $\text{CDCl}_3$  and  $(\text{CD}_3)_2\text{SO}$  solutions on a Varian Gemini 300 MHz spectrometer and chemical shifts are expressed in  $\delta$  units using TMS as an internal reference. Mass spectra were recorded on a GC-MS QP1000 EX Shimadzu. Elemental analysis were carried out at the Microanalytical Center of the Cairo University. Hydrazonoyl halides were obtained as previously reported.<sup>12-18</sup>

### SYNTHESIS OF 2-(4-NAPHTHO[1,2-*d*]FURAN-2-yl)-1,3-THIAZOL-2-yl)ETHANENITRILE (2)

A mixture of 2-Bromo-1-naphtho[1,2-*d*]furan-2-ylethan-1-one (**1**),<sup>19</sup> (2.9 g, 0.01 mmol) and cyanothioacetamide (1.0 g, 0.01 mmol) in ethanol (25 mL) was refluxed for 2 h. The reaction mixture was poured onto

ice-cold water (50 mL) and drops of conc. ammonium hydroxide (25%) were added. The resulting solid was collected and washed with water, and recrystallized from ethanol to give thiazolylacetonitrile **2** (Table I).

**TABLE I Characterization Data of the Newly Synthesized Compounds**

Corp. no.	Mp., °C solvent	Color yield%	Mol. formula mol. wt.	Calcd./found%			
				C	H	N	S
<b>2</b>	175–78	Pale Brown	C <sub>17</sub> H <sub>10</sub> N <sub>2</sub> OS	70.33	3.47	9.69	11.04
	EtOH	69	290.35	70.45	3.30	9.80	11.15
<b>3</b>	228–30	Yellow	C <sub>24</sub> H <sub>14</sub> N <sub>2</sub> O <sub>2</sub> S	73.08	3.58	7.10	8.13
	DMF	64	394.46	72.85	3.80	7.00	8.25
<b>4</b>	268–70	Yellowish gray	C <sub>24</sub> H <sub>13</sub> NO <sub>3</sub> S	72.90	3.54	12.14	8.11
	AcOH	61	395.44	73.10	3.40	12.00	7.95
<b>5</b>	209–210	Yellow	C <sub>24</sub> H <sub>14</sub> N <sub>2</sub> OS	76.17	3.73	7.40	8.47
	DMF	52	378.46	76.25	3.65	7.20	8.65
<b>7a</b>	232–34	Brown	C <sub>23</sub> H <sub>14</sub> N <sub>4</sub> OS	70.03	3.58	14.20	8.13
	Dioxane	40	394.46	69.80	3.68	13.95	8.25
<b>7b</b>	230–32	Brown	C <sub>24</sub> H <sub>16</sub> N <sub>4</sub> OS	70.57	3.95	13.72	7.85
	DMF	41	408.49	70.70	4.15	13.94	7.71
<b>7c</b>	265–67	Brown	C <sub>23</sub> H <sub>13</sub> ClN <sub>4</sub> OS	64.41	3.06	13.06	7.48
	Dioxane	39	428.90	64.50	2.90	12.90	7.70
<b>9</b>	220–22	Brown	C <sub>24</sub> H <sub>15</sub> N <sub>3</sub> OS <sub>2</sub>	67.74	3.55	9.87	15.07
	Dioxane	67	425.53	67.50	3.70	9.70	14.80
<b>10</b>	207–209	Yellowish gray	C <sub>25</sub> H <sub>17</sub> N <sub>3</sub> OS <sub>2</sub>	68.31	3.90	9.56	14.59
	DMF	58	439.56	68.20	3.80	9.72	14.82
<b>11</b>	244–46	Brown	C <sub>24</sub> H <sub>17</sub> N <sub>5</sub> OS	68.07	4.05	16.54	7.57
	AcOH	72	423.50	67.90	4.10	16.45	7.40
<b>15a</b>	275–77	Yellow	C <sub>27</sub> H <sub>16</sub> N <sub>4</sub> O <sub>3</sub> S <sub>2</sub>	63.77	3.17	11.02	12.61
	DMF	60	508.58	63.92	3.10	10.97	12.42
<b>15b</b>	302–304	Yellowish brown	C <sub>28</sub> H <sub>18</sub> N <sub>4</sub> O <sub>3</sub> S <sub>2</sub>	64.35	3.47	10.72	12.27
	DMF	66	522.61	64.10	3.32	10.90	12.40
<b>15c</b>	274–76	Yellowish brown	C <sub>32</sub> H <sub>19</sub> N <sub>5</sub> O <sub>2</sub> S <sub>2</sub>	67.47	3.36	12.29	11.26
	DMF	63	569.67	67.29	3.25	12.10	11.40
<b>15d</b>	198–200	Yellowish brown	C <sub>27</sub> H <sub>16</sub> N <sub>4</sub> O <sub>2</sub> S <sub>2</sub>	65.84	3.27	11.37	13.02
	DMF	59	492.58	65.900	3.35	11.50	12.90
<b>14e</b>	243–45	Brown	C <sub>32</sub> H <sub>18</sub> N <sub>4</sub> O <sub>2</sub> S <sub>2</sub>	69.30	3.27	10.10	11.56
	DMF	69	554.65	69.40	3.00	9.95	11.65
<b>15f</b>	265–67	Brown	C <sub>30</sub> H <sub>16</sub> N <sub>4</sub> O <sub>2</sub> S <sub>3</sub>	64.27	2.88	9.99	17.16
	DMF	64	560.68	64.00	2.60	10.10	17.00
<b>15g</b>	265–268	Brown	C <sub>36</sub> H <sub>20</sub> N <sub>4</sub> O <sub>2</sub> S <sub>2</sub>	71.51	3.33	9.27	10.60
	DMF	67	604.71	71.30	3.50	9.40	10.70
<b>15h</b>	>320	Brown	C <sub>38</sub> H <sub>20</sub> N <sub>4</sub> O <sub>3</sub> S <sub>2</sub>	70.79	3.13	8.69	9.95
	DMF	61	644.74	70.90	3.10	8.50	10.20
<b>17</b>	272–75	Brown	C <sub>19</sub> H <sub>12</sub> N <sub>2</sub> OS <sub>3</sub>	59.97	3.18	7.36	25.28
	DMF	62	380.51	59.70	3.00	7.50	25.40



**SYNTHESIS OF 3-(4-NAPHTHO[1,2-*d*]FURAN-2-yl)-1,3-HIAZOL-2-yl)-2H-CHROMEN-2-IMINE (3) AND (2E)-2-(4-NAPHTHO[1,2-*d*]FURAN-2-yl(1,3-THIAZOL-2-yl))-3-PHENYLPROP-2-ENENITRILE (5)**

Method A: Thiazolylacetonitrile **2** (1.45 g, 5 mmol) and salicylaldehyde or benzaldehyde (5 mmol) in ethanol (20) containing a catalytic amount of piperidine were stirred at room temperature for 2 h. The resulting solid was collected and recrystallized to give **3** and **5**, respectively (Table I).

Method B: Equimolar amounts of 2-(aminothioxomethyl)-3-phenylprop-2-enenitrile<sup>20</sup> (**6**) and naphtho[1,2-*d*]furan-2-ylethan-1-one (**1**) (5 mmol) in acetic acid (15 mL) was boiled under reflux for 2 h. The resulting solid was collected and crystallized to give a product identical (mp., mixed mp, and spectra) with **5**

**SYNTHESIS OF 3-(4-NAPHTHO[1,2-*d*]FURAN-2-yl)-1,3-THIAZOLE-2-yl)-2H-CHROMEN-2-ONE (4)**

A mixture of **3** (0.5 g) and hydrochloric acid (5 mL, 6 M) was stirred at room temperature for 1 h. The resulting solid was collected and recrystallized from acetic acid to give substituted coumarin **4** (Table I).

**SYNTHESIS OF (2E)-3-AZA-3-(ARYLAMINO)-2-NAPHTHO[1,2-*d*]FURAN-2-yl(1,3-THIAZOL-2-yl))-2-ENENITRILE 7a-c**

Method A: The appropriate arenediazonium chlorides (5 mmol) was added dropwise to a cold ethanol solution, containing thiazolylacetonitrile **2** (1.45 g, 5 mmol) and sodium acetate (1 g), (50 mL) at 0–5°C while stirring. The mixture was stirred for 3 h at 0–5°C. The resulting solid was collected and crystallized from dioxane to give **7a-c** (Tables I and II).

Method B: A mixture of the appropriate (2E)-2-aminothioxomethyl)-3-aza)-3-(arylamino)prop-2-enenitrile<sup>21</sup> **8a-c** and 2-Bromo-1-naphtho[1,2-*d*]furan-2-ylethan-1-one (**1**) (5 mmol each) in ethanol (30 mL) was boiled under reflux for 2 h. The resulting solid, which formed after cooling, was collected and crystallized from acetic acid to afford a product identical in all respects (mp., mixed mp., and spectra) with products obtained in method A.

**SYNTHESIS OF THIOANILID 9**

Phenyl isothiocyanate (0.65 g (0.6 mL), 5 mmol) was added to a mixture of thiazolylacetonitrile (1.45 g, 5 mmol) **2** and potassium hydroxide

TABLE II <sup>1</sup>H NMR Spectra of Some Newly Synthesized Compounds

Comp. no.	Spectral data
<b>2</b>	<sup>1</sup> H NMR: 4.22 (s, 3H), 7.51–8.19 (m, 8H) IR: 2252 (CN)
<b>4</b>	IR: 1660 (C=N), 1604 (C=C)
<b>5</b>	IR: 1712 (CO), 1604 (C=C)
<b>7a</b>	<sup>1</sup> H NMR: 7.5–8.12 (m, 13H), 8.21 (s, 1H) IR: 3342 (NH), 2214 (CN) Mass: 378 (100%), 223 (4.2%), 195 (42%), 152 (48.8%).
<b>7b</b>	IR: 3349 (NH), 221 (CN)
<b>7c</b>	<sup>1</sup> H NMR: 2.37 (s, 3H), 7.26–8.20 (m, 12H) and 14.05 (s, br., 1H). IR: 3349 (NH) and 2221 (CN).
<b>9</b>	IR: 3350 (NH), 2214 (CN), 1651 (C=N), 1569 (C=C)
<b>10</b>	IR: 3323 (NH), 2177 (CN), 1623 (C=N)
<b>11</b>	<sup>1</sup> H NMR: 4.10 (s, br., 2H), 6.70–7.67 (m, 14H), 9.32 (s, br., 1H). IR: 3350, 3280, 3150 (NH <sub>2</sub> , NH), 3050 (CH), 1620 (C=N), 1596 (C=C).
<b>15a</b>	<sup>1</sup> H NMR: 2.27 (s, 3H), 7.31–8.10 (m, 13H) and 12,53 (s, br., 1H). IR: 2191 (CN), 1720 (CO), 1666 (C=N)
<b>15b</b>	<sup>1</sup> H NMR: 1.56 (t, 3H), 4.55 (s, 2H), 7.60–8.29 (m, 13H) and 12,53 (s, br., 1H). IR: 2191 (CN), 1743 (CO), 1624 (C=N)
<b>15d</b>	<sup>1</sup> H NMR: 2.20 (s, 3H), 6.46 (d, 4H), 6.72 (s, 1H), 7.01–7.67 (m, 10H), 8.21 (s, 1H). <sup>13</sup> C NMR: 23, 89, 102, 111, 115, 118, 121, 124, 127, 128, 129, 143, 146, 149, 150, 153, 154, 155, 194. IR: 2198 (CN), 1685 (CO), 1596 (C=C)
<b>15f</b>	<sup>1</sup> H NMR: 2.40 (s, 3H), 7.50–8.32 (m, 13) IR: 2191 (CN), 1674 (CO)
<b>15g</b>	IR: 2191 (CN), 1680 (CO), 1635 (C=N)
<b>17</b>	<sup>1</sup> H NMJR: 1.51 (s, 1H), 2.25 (s, 3H), 6.71–7.76 (m, 8H).

(0.28 g, 5 mmol) in *N,N*-dimethylformamide (15 mL) while stirring at room temperature until potassium hydroxide dissolve completely, and the reaction mixture was continued stirring for 1 h. The reaction mixture was poured onto water (30 mL) and acidifies with acetic acid. The resulting solid was collected and recrystallized from dioxin to give thioanilide **9** (Table I).

**SYNTHESIS OF (2E)-3-METHYLTHIO-2-(4-NAPHTHO[1,2-*d*]-FURAN-2-yl)(1,3-THIAZOL-2-yl))-3-(PHENYLAMINO)PROP-2-ENENITRILE (**10**)**

Phenyl isothiocyanate (0.65 g, 0.6 mL), 5 mmol) was added to a mixture of thiazolylacetonitrile (1.45 g, 5 mmol) **2** and potassium hydroxide

(0.28 g, 5 mmol) in *N,N*-dimethylformamide (15 mL) while stirring at room temperature until potassium hydroxide dissolve completely, and the reaction mixture was stirred for 1 h, then Iodomethane (0.71 g, 5 mmol) was added dropwise while stirring. The reaction mixture was stirred for 2 h and the resulting solid was collected and recrystallized from *N,N*-dimethylformamide to give **10** (Table I).

### SYNTHESIS OF [5-AMINO-4-(4-NAPHTHO[1,2-*d*]FURAN-2-yl)(1,3-THIAZOL-2-yl)PYRAZOL-3-yl]-PHENYLAMINE (**11**)

A mixture of **10** (2.18 g, 5 mmol) and hydrazine hydrate (1 m, 99%) in ethanol (20 mL) was boiled under reflux for 4 h. The resulting solid was collected and crystallized from *N,N*-dimethylformamide to give **11** (Table I).

### SYNTHESIS OF 2,3-DIHYDRO-1,3,4-THIADIAZOLES **15a-h**

Method A: Phenyl isothiocyanate (0.65 g (0.6 mL), 5 mmol) was added to a mixture of thiazolylacetonitrile (1.45 g, 5 mmol) **2** and potassium hydroxide (0.28 g, 5 mmol) in *N,N*-dimethylformamide (15 mL) while stirring at room temperature until potassium hydroxide dissolved completely and the reaction mixture was stirred for 30 min. The appropriate hydrazoneoyl halides **12a-h** (5 mmol) were added and stirred for 3 h. The resulting solid was collected and recrystallized from the appropriate solvent to afford **15a-h** (Table I).

Method B: Triethylamine (0.5 g, (0.75 mL), 5 mmol) was added dropwise to a mixture of the appropriate hydrazoneoyl halides **12a-h** and **17** (1.9 g, 5 mmol) in ethanol (20 mL) while stirring. Stirred was continued for 1 h and the resulting solid was collected and crystallized to give products identical in all respects (mp., mixed mp., and spectra) with those obtained in method A.

### SYNTHESIS OF 3-METHYLTHIO-2-(4)-NAPHTHO[1,2-*d*]FURAN-2-yl(1,3-THIAZOL-2-yl)-3-THIOXOPROPANENITRILE (**17**)

A mixture of thiazolylacetonitrile **2** (1.45 g, 5 mmol), potassium hydroxide (0.28 g (5 mmol) and carbon disulfide (0.47 g, 5 mmol) in *N,N*-dimethylformamide (15 mL) was stirred for 6 h at room temperature. Iodometane (0.71 g, 5 mmol) was added dropwise to the above mixture and stirred for 1 h. The resulting solid was collected and crystallized from acetic acid to give **17** (Tables I and II)

## REFERENCES

- [1] P. A. Lowe, In *Heterocyclic Chemistry*, H. Suschitzky and O. Methcohn (eds.), Chemical Society, London, Vol. 1, pp. 119–139 (1980).
- [2] J. V. Metzger, Thiazoles and their benz Derivatives. In *Comprehensive Heterocyclic Chemistry*, A. R. Katritzky and C. W. Rees (eds.), Pergamon Press, vol. 6, pp. 328 (1984).
- [3] H. D. Brown, *US Pat. 3 278 547*, 1966; [*Chem. Abstr.* **65**, 18593 (1966)].
- [4] S. P. Singh and S. Segal, *Indian J. Chem.*, **27B**, 941 (1988).
- [5] Y. Usui, *Yakugaku Zasshi*, **89**, 099 (1969) [*Chem. Abstr.* **71**, 69601 (1969)].
- [6] P. Goursot and E. F. Westrum, Jr. *J. Chem. Eng. Data*, **14**, 1 (1969).
- [7] N. M. Rateb and A. O. Abdelhamid, *Heteroatom. Chem.*, **15**, 107 (2004).
- [8] A. O. Abdelhamid and N. S. Beshai, *Chemistry. An Indian J.*, **1**, 32 (2003).
- [9] A. O. Abdelhamid, B. A. M. Abdel-Wahab, and A. A. Al-Atoom, *Phosphorus, Sulfur, and Silicon and the Related Elements*, **179**, 601 (2004).
- [10] N. A. Abdel-Riheem, N. M. Rateb, A. A. Al-Atoom, and A. O. Abdelhamid, *Heteroatom Chem.*, **14**, 421 (2003).
- [11] N. M. Rateb, N. A. Abdel-Riheem, A. A. Al-Atoom, and A. O. Abdelhamid, *Phosphorus, Sulfur, and Silicon and The Related Element*, **178**, 1101 (2002).
- [12] G. Fravel, *Bull. Soc. Chim. Fr.*, **31**, 150 (1904).
- [13] N. E. Eweiss and A. Osman, *Tetrahedron Lett.*, 1169 (1979).
- [14] A. S. Shawali and A. O. Abdelhamid, *Bull. Chem. Soc. Jpn.*, **49**, 321 (1976).
- [15] A. S. Shawali and A. Osman, *Tetrahedron*, **27**, 2517 (1971).
- [16] A. O. Abdelhamid and F. H. H. El-Shiatey, *Phosphorus, Sulfur, Silicon, and the Related Elements*, **39**, 45 (1988).
- [17] H. M. Hassaneen, A. S. Shawali, N. E. Elwan, and N. M. Abounada, *Sulfur Letters*, **14**, 41 (1972).
- [18] R. H. Wiley and G. H. Garboe, *J. Am. Chem. Soc.*, **78**, 624 (1956).
- [19] A. O. Abdelhamid and M. A. M. Alkhodshi, *Phosphorus, Sulfur, and Silicon, and the Related Elements*, in press (2004).
- [20] J. S. A. Brunskill, A. De, and O. F. Ewing, *J. Chem. Soc, Perkin Trans I*, 629 (1978).
- [21] S. A. Mansour, W. M. Eldeib, S. E. Abdou, and H. A. Daboun, *Sulfur Letter*, **6**, 181 (1987).