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Metal-free Catalyzed Cyclization of N-methoxybenzamides to Construct Quaternary Carbon-Containing Isoindolinones

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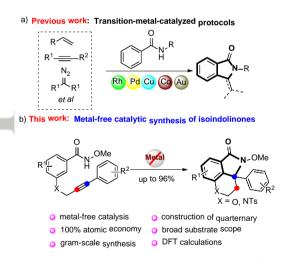
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Summary of main observation and conclusion Through the intramolecular cyclization of *N*-methoxybenzamides, a simple and efficient method for onstructing valuable isoindolinones under metal-free conditions was developed. The reaction was featured by employing low-cost catalyst, simple operation, 100% atomic economy and excellent regioselectivity. Moreover, a detailed computational study on the reaction system has been performed to clarify the mechanism. This protocol tolerated a variety of functional groups and provided a metal-free protocol for the synthesis of chromane- or tetrahydroquinoline-fused isoindolinones in good yields.

Background and Originality Content

Among the large family of indoles, isoindolinone derivatives are remarkable owing to their prevalence in natural products, drugs nd functional materials. For example, indoprofen is used in antiinflammatory drugs, falipamil demonstrates anti-ischemic properties, chlortalidone has a diuretic effect and pestalachloride A is identified as an antifungal drug.¹Therefore, the synthesis of this class of compounds caused a great interest of chemists, and various synthetic methods have been developed.²

Scheme 1 The synthetic methods for isoindolinones.



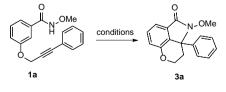
Remarkable progress has been made in the Rh(III)-, Pd(II)-, I(II)-, Co(II)- and Au(I)-catalyzed C–H bond functionalizations, which provides a step- and atom-economical route to isoindolinones from less functionalized substrates (Scheme 1a).³ For instance, the group of Ma reported the Rh(III)-catalyzed synthetic route of isoindolinones via a sequential C–H activation/

^aE-mail:wenlirong@qust.edu.cn;liming928@qust.edu.cn ^bE-mail:*shaofeinee@163.com* allene formation/cyclization pathway.⁴ We disclosed a Co/Agcatalyzed C-H alkynylation/annulation with alkynes for the construction of 3-methyleneisoindolinones.⁵ Despite the fact that significant development has been achieved on transition metals promoting research of the cyclization of alkynes based on C-H functionalizations or cascade reactions, an alternative route relying on metal-free strategy to construct target products is still appealing due to green and economy features.⁶

Triflimide (HNTf₂) as a special catalyst has been achieved uniquely outstanding performance in a range of organic transformations, including C-C and C-heteroatom formation processes because of its strong acidity and good compatibility with organic solvents.⁷ In these cases, it has been extensively used as a super Brønsted acid to catalyze Friedel-Crafts reactions,⁸ aldol reactions,⁹ and cycloaddition reactions.¹⁰ Especially, Ye and coworkers have developed a variety of HNTf2-initiated cascade cyclizations of ynamides for the synthesis of functionalize heterocycle compounds based on keteniminium ions.¹¹ Inspired by these successful results and as part of our interest in green chemistry,¹² we envisioned that HNTf₂ might also active the simple alkynes, thus allowing the facile and practical access to a variety of highly functionalized isoindolinones. Herein, we describe the realization of intramolecular HNTf₂-initiated cascade cyclization of alkynes, leading to chromane-fused isoindolinones in good yields, providing a green and economical alternative among the previously applicable metal-catalyzed protocols (Scheme 1b).

Results and Discussion

Table 1 Optimization of reaction conditions for 3a.ª



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Report

	Entry	Catalyst 2	Solvent	Temp (℃)	Yield (%) ^b
	1	AgOTf	DCE	100	29
	2	Cu(OTf) ₂	DCE	100	33
	3	Zn(OTf) ₂	DCE	100	36
	4	Sc(OTf)₃	DCE	100	40
	5	In(OTf)₃	DCE	100	51
	6	Yb(OTf)₃	DCE	100	39
	7	CF ₃ COOH	DCE	100	49
	8	HOTf	DCE	100	63
	9	$HNTf_2$	DCE	100	73
•	10	HNTf ₂	DCM	100	64
	11	HNTf ₂	toluene	100	43
	12	HNTf ₂	MeOH	100	trace
	13	HNTf ₂	EtOH	100	trace
	14	$HNTf_2$	HFIP	100	46
<u> </u>	15	HNTf ₂	DCE	80	60
	16	HNTf ₂	DCE	120	71
	17 °	HNTf ₂	DCE	100	80
	18 ^d	HNTf ₂	DCE	100	81
-	19 ^e	HNTf ₂	DCE	100	82
	20	-	DCE	100	n.r. ^f

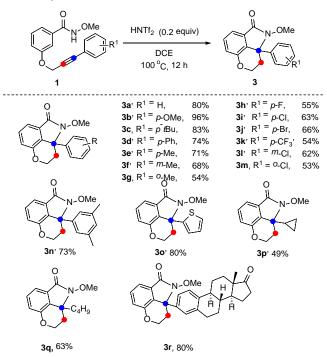
teaction conditions: 1a (0.1 mmol), catalyst 2 (0.01 mmol), solvent (2 mL), in air, 12 h. ^b Isolated yields. ^c 20 mol% of catalyst 2 was used. ^d25% mol of catalyst 2 was used. ^e100% mol of catalyst 2 was used. ^fn.r. = no reaction, DCE = 1,2-dichloroethane, HFIP = hexafluoroisopropanol.

N-methoxy-3-((3-phenylprop-2-yn-1-At the outset. yl)oxy)benzamide 1a was employed as the model substrate to ptimize the reaction conditions, and the results were summarized in Table 1. To our delight, the desired product 3a was furnished in atmosphere at 100 °C (Table 1, entry 1). We then screened other metal catalysts including copper-, zinc-, scandium-, indiumand ytterbium-salts (Table 1, entries 2-6) and found that In(OTf)₃ c uld efficiently catalyze the cascade cyclization, delivering the uesired product 3a in 51% yield (Table 1, entry 5). Further investigation of Brønsted acids revealed that HNTf2 could fectively promote the reaction with a yield of 73% (Table 1, entries 7-9). The solvents have a significant influence on the transformation (Table 1, entries 10-14) and the protic solvents such MeOH, EtOH and HFIP gave inferior yields. The reaction cemperature was studied and the results showed that increasing or decreasing the temperature was not conducive to the reaction able 1, entries 15 and 16). Gratifyingly, by simple applying 20 mol% of HNTf₂ as a catalyst, the yields of **3a** could be further increased to 80%. The further increasement of the catalyst did not significantly promote the transformation with a better yield (Table 1, entries 18-19). A control experiment showed that HNTf₂ was essential to the transformation, and the reaction did not take place without it (Table 1, entry 20).

Under the existing conditions (Table 1, entry 17), the cyclization reaction of various alkyne-tethered *N*-methoxybenzamides **1** were explored. As shown in Scheme 2, the transformations proceeded smoothly and gave the corresponding products isoindolinones **3a**-**3r** with good to excellent yields. *N*-methoxybenzamides **1** with electron-donating substituents achieved excellent yields of 71–96% (**3b-3e**). However, *N*-methoxybenzamides **1** with electron-withdrawing groups such as F, Cl, Br and CF₃ showed slightly less

reactive to provide 54–66% yields (**3h-3k**). For *meta*-substituted and *ortho*-substituted benzamides, the reaction is affected by the steric hindrance of the aromatic ring, both the electron-donating

Scheme 2 Synthesis of isoindolinones 3.^{*a,b*}

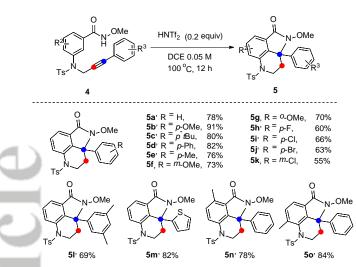


 a Reaction conditions: 1 (0.20 mmol), HNTf_2 (0.04 mmol), DCE (4 mL), 100 °C, 12 h. b Isolated yields.

groups and the electron-withdrawing groups have inferior results with moderate yields (**3f**, **3g**, **3l** and **3m**). The reaction of the disubstituted benzamide **1n** proceeded smoothly, and the corresponding product **3n** was obtained in 73% yield. To our delight, heterocyclic substrate **1o** underwent smoothly to produce product **3o** in 80% yield. Additionally, the reaction with the alkyl-substituted substrates gave the products (**3p** and **3q**) in 49% and 63% yields, respectively. It is worth noting that natural product derivative **1r** could also be employed to provide corresponding product **3r** in a yield of 80%.

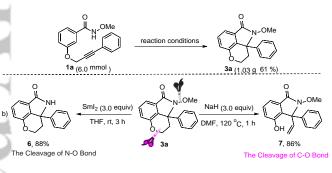
Next, the synthesis of tetrahydroquinoline-fused isoindolinones was also explored under the optimized reaction conditions by employing the aniline-typed *N*-methoxybenzamides **4** (Scheme 3). For R³ groups, *N*-methoxybenzamides **4** with electron-donating substituents achieved good yields of 76-91% (5b-5e). Although the electronic effect on the aryl rings reduced the yields to some extent, the desired products 5h-5j were still generated with yields ranging from 60% to 66%. Compared to para-substituted benzamides, the yields of ortho- and meta-substituents were slightly reduced (5f, 5g and 5k). The cyclization strategy was also compatible with disubstituted benzamide 4I, and the product 5I was got in 69% yield. Thiophene-substituted benzamide was also successfully converted to the corresponding heterocyclic product **5m** in 82% yield. For R² groups, when the methyl substituent was installed at ortho or para position of benzene ring, the desired isoindolinones were obtained in good yields (5n and 5o)

Scheme 3 Synthesis of isoindolinones 5.^{a,b}

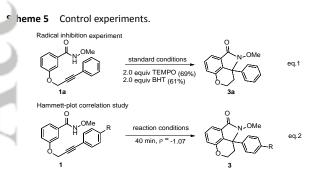


[•]Reaction conditions: **4** (0.20 mmol), HNTf₂ (0.04 mmol), DCE (4 mL), 100 °C, 1 h. ^bIsolated yields.





To demonstrate the practical application of the present protocol, gram-scale experiment was carried out for the reaction of **1a** (6 mmol) under the optimized reaction conditions, and **3a** was brained in 61% yield (Scheme 4a). The product **3a** could be efficiently deprotected by treating with Sml₂ in THF, providing **6** in $\cos y$ yield, while the cleavage of C-O bond of **3a** by treatment with NaH in DMF produced the 4-hydroxyisoindolin-1-one **7** in 86% yield, v hich might have anti-HIV¹³ and anti-inflammatory activities (Scheme 4b).¹⁴



The control experiments were performed to gain some understanding of the mechanism (Scheme 5). The addition of radical inhibitor into the reaction system provided **3a** smoothly, indicating that the radical pathway may not be involved in the process (Scheme 5, eq.1). Moreover, the reaction of different *para*-substituted *N*-methoxybenzamides **1** was studied (Scheme 5, eq.2). In the Hammett curve, $lg(k_x/k_H)$ is linear with σ , and the negative value of ρ (-1.07) indicates that the reaction is facilitated by

electron-donating groups (see Supporting information for details). $^{\rm 15}$

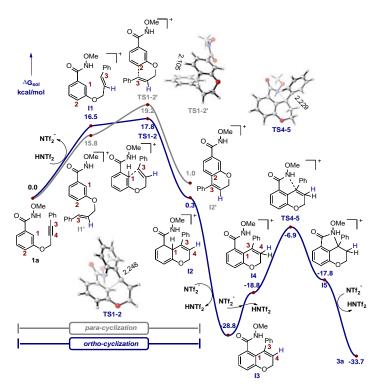


Figure 1 Calculated Gibbs free energy profiles for the possible reaction mechanism in kcal mol $^{\rm 1}$

Moreover, density functional theory (DFT) calculations were performed to clarify the regioselectivity of the reaction (Figure 1; for the details see the Supporting Information). The initiation of the reaction involves the protonation of the alkynyl of 1a to form a cationic intermediate I1, where the electrophilic ortho-cyclization process between the carbon cation C3 and C1 takes place via transition state TS1-2 to afford I2.¹⁶ The barrier needed for this step is 17.8 kcal mol⁻¹, which is affordable under the current experimental conditions. The electrophilic para-cyclization transition state TS1-2' was also located and the barrier is 1.4 kcal mol⁻¹ higher than TS1-2, which agrees well with experimental results. The origin of the regioselectivity could be explained by the NBO charges in Figure 1. The NBO charges suggest that the electrostatic interaction between C1 and C3 in TS1-2 is stronger than that between C2 and C3 in TS1-2'. This means that the C3 carbon would prefer to attack the C1 atom rather than C2 atom of the phenyl ring, generating the corresponding ortho-cyclization intermediate 12. From 12, the proton transfer from C1 to C4 (stepwise deprotonation and protonation via I3, moreover, a trace amount of I3 was detected through HRMS) takes place with the help of NTf₂⁻/HNTf₂ species, leading the reaction to intermediate I4, where the last ring closing process via C-N bond formation occurs. This is the rate-determining step with a reaction barrier of 21.9 kcal mol⁻¹. The desired product 3a will be generated by the last deprotonation process. The whole reaction is exothermic about 33.7 kcal mol⁻¹ with a rate-determining barrier of 21.9 kcal mol⁻¹, agreeing well with the experimental observations.

Conclusions

In conclusion, we have developed an efficient metal-free intramolecular cyclization of *N*-methoxybenzamides, enabling facile access to chromane- or tetrahydroquinoline-fused

isoindolinones in good yields from easily available starting materials. This reaction was characterized by employing cheap catalyst, simple operation, broad substrate scope, 100% atomic economy and excellent regioselectivity. The gram-scale experiment rendered this method highly practical in organic synthesis. Moreover, a detailed computational study on the reaction system have been performed to clarify the mechanism. Further synthetic applications are currently undergoing in our laboratory.

Experimental

General experimental procedure for products 3a-3r and 5a-5o (using 3a as an example): To a stirred solution of *N*-methoxy-3-((3- μ -nenylprop-2-yn-1-yl)oxy)benzamide **1a** (56.2 mg, 0.2 mmol) in DCE (4 mL) was added HNTf₂ (11.2 mg, 0.04 mmol) at room t mperature in a sealing tube (10 mL). Next, the reaction mixture was sealed and heated at 100 °C for 12 h. Then, the solvent was evaporated in vacuum and purified by silica gel column cnromatography to afford pure desired product **3a** (45 mg, 80%) as a yellow solid.

Supporting Information

The supporting information for this article is available on the WWW under https://doi.org/10.1002/cjoc.2018xxxxx.

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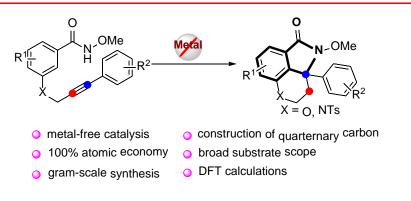
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Entry for the Table of Contents

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Metal-free Catalyzed Cyclization of Nmethoxybenzamides to Construct Quaternary Carbon-Containing Isoindolinones



Through the intramolecular cyclization of *N*-methoxybenzamides, a simple and efficient method for constructing valuable isoindolinones under metal-free conditions was developed. The reaction was featured by employing low-cost catalyst, simple operation, 100% atomic economy and excellent regioselectivity. Moreover, a detailed computational study on the reaction system has been performed to clarify the mechanism. This protocol tolerated a variety of functional groups and provided a metal-free protocol for the synthesis of chromane- or tetrahydroquinoline-fused isoindolinones in good yields.

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