This article was downloaded by: [McGill University Library] On: 27 August 2012, At: 03:45 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: <u>http://www.tandfonline.com/loi/gpss20</u>

Part 7: Synthesis of Some New 1,5-Benzodiazepines Fused with Different Heterocyclic Moieties

A. M. El-Sayed ^a , A. Khodairy ^a , H. Salah ^a & H. Abdel-Ghany ^a

^a Chemistry Department, South Valley University, Sohag, Egypt

Version of record first published: 24 Feb 2007

To cite this article: A. M. El-Sayed, A. Khodairy, H. Salah & H. Abdel-Ghany (2007): Part 7: Synthesis of Some New 1,5-Benzodiazepines Fused with Different Heterocyclic Moieties, Phosphorus, Sulfur, and Silicon and the Related Elements, 182:4, 711-722

To link to this article: <u>http://dx.doi.org/10.1080/10426500601087301</u>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <u>http://www.tandfonline.com/page/terms-and-conditions</u>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be

independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.



Part 7: Synthesis of Some New 1,5-Benzodiazepines Fused with Different Heterocyclic Moieties

A. M. El-Sayed A. Khodairy H. Salah H. Abdel-Ghany Chemistry Department, South Valley University, Sohag, Egypt

3-cyano-1,11-dihydro-4,5-diphenyl-2-thioxopyrido[2,3-b](1,5)benzodiazepine **2** and 3-(2'-cyano-1'-phenyl-2'-ethanethiocarboxamide)-4-phenyl-1(H)(1,5)benzodiazepin-2-one **3** were prepared via the reaction of 1,3-dihydro-4-phenyl-(1,5)benzodiazepin-2-one **1** with benzylidenecyanothioacetamide. Compound **2** was treated with halo compounds to give the corresponding S-alkylated compounds 4_{a-e} , which underwent as intramolecular ring closure to thieno[3,2.5,6]pyrido[2,3b](1,5)benzodiazepines 5_{a-e} under PTC conditions. One-pot synthesis of compounds 5_{a-e} was achieved via the reaction of compound **2** with the appropriate halo compound under PTC conditions. Compound **1** and 1-ethyl-4-phenyl-(1,5)benzodiazepin-2-one **9** were treated with carbon disulfide or phenylisothiocyanate and active nitriles to afford 4-thioxothiopyrano[4,3-b](1,5)benzodiazepines **8** and **10–14**. Treatment of compound **10** with phenylisothiocyanate or acetic anhydride yielded oxazino- and pyrimido[4,5-b]thiopyrano-[4',3'-b'](1,5)benzodiazepine **15** and **17**. The reaction of compound **1** with elemental sulfur and active nitriles yielded thieno[3,2-b](1,5)benzodiazepines **18–21**, respectively.

Keywords 3-cyano-1,11-dihydro-4,5-diphenyl-2-thioxopyrido[2,3-b](1,5)benzodiazepine; 1-ethyl-1(H)-4-phenyl-1,5-benzodiazepin-2-one; PTC

INTRODUCTION

Benzodiazepines and their polycyclic derivatives are used in pharmaceutical and biological chemistry,¹ where they are used as antitumor agents,² nevirapine analgnes,³ and anti–HIV-1 (Human Immunodeficiency Virus) agents.³ They also are screened for in vitro cytotoxicity against a number of cancer cell lines,⁴ such as colon cancer, breast cancer, lung cancer, and bladder cancer.⁵ For all these reasones, we continue our laboratory work on the synthesis of fused and spiro

Received January 9, 2006; accepted March 14, 2006.

Address correspondence to H. Abdel-Ghany, Chemistry Department, Faculty of Science, South Valley University, 82524 Sohag, Egypt. E-mail: khodairy@yahoo.com

benzodiazepines.^{6–11} So herein we report the synthesis of pyrido[2,3-b]-,thiopyrano[4,3-b]-, and thieno[3,2-b](1,5)benzodiazepines.

RESULTS AND DISCUSSION

Recently, Khodairy¹⁰ reported that the reaction of 1,3-dihydro-4phenvl-1.5-benzodiazepin-2-one **1**¹² with some vlidenecvanothioacetamides and sodium ethoxide gave corresponding 3-cyano-2thioxopyrido[2,3-b][1,5]-benzodiazepines. We report herein other trials for the synthesis of 3-cvano-1.11-dihydro-4.5-diphenyl-2-thiopyrido [2,3-b][1,5]benzodiazepine via the reaction of 1,3-dihydro-4-phenyl-1,5-benzodiazepin-2-one 1 with benzylidenecyanothioacetamide and piperidine as a basic catalyst, where a mixture of 3-cyano-1,11-dihydro-4.5-diphenyl-2-thioxopyrido[2,3-b](1,5)-benzodiazepine 2 and 2-cyano-3(4-phenyl-2-oxo-1,2-dihydro-3H-(1,5)benzodiazepin-3-ylidene)-3-phenylpropanimidothioic acid 3 was obtained in a 20% and 70% yield. respectively. The IR spectrum of compound 2 showed new absorption bands at 3385 cm⁻¹ and 2214 cm⁻¹ corresponding to NH and CN groups, respectively, while its ¹H NMR spectrum revealed the presence of multiplet signals at δ 8.70–7.00 ppm for the 14 H aromatic protons and a singlet signal at δ 9.50 ppm for the new NH group.

Treatment of compound 2 with active halo compounds, namely ethyl chloroacetate, chloroacetonitrile, and phenacyl bromide in the presence of odium acetate as a catalyst, afforded 2-carbethoxymethylthio-3cyano-4,5-diphenyl-11(H)pyrido[2,3-b](1,5)benzodiazepine 4a, 3-cyano-2-cyanomethy-Ithio-4,5-diphenyl-11(H)pyrido[2,3-b](1,5) benzodiazepine 4_b, and 2-benzoyl-methylthio-3-cyano-4,5-diphenyl-11(H)pyrido [2,3-b](1,5) benzodiazepine 4_c , respectively. Using the PTC technique $(dioxane/potassium carbonate/tetrabutyl-ammonium bromide [TBA<math>\beta$]), compounds 4_{a-c} underwent intramolecular cyclization into 3-amino-2carbethoxy-4,5-diphenyl-11(H)thieno[2,3-b]pyrido-[2',3'-b'](1,5)benzodiazepine 5_a, 3-amino-2-cyano-4,5-diphenyl-11(H)thieno-[2,3-b]pyrido [2',3'-b'](1,5)benzodiazepine 5_b, and 3-amino-2-benzoyl-4,5-diphenyl-11(H)thieno[2,3-b]pyrido[2',3'-b'](1,5)benzodiazepine $\mathbf{5_c}$, respectively. Compounds **5**_{a-c}, 3-amino-2-carboxamido-4,5-diphenyl-11(H)thieno [2,3-b]pyrido [2',3'-b'](1,5)benzodiazepine 5_d, and 3-amino-2-phenylcarboxamido-4,5-diphenyl-11(H)thieno[2,3-b]pyrido[2',3'-b'](1,5)benzodiazepine 5_e were synthesized directly in a one-pot step via the reaction of compound 2 with ethyl chloroacetate, chloroacetonitrile, phenacyl bromide, chloroacetamide, and chloroacetanilide, respectively, under PTC conditions (dioxane/potassium carbonate/TBAB). IR spectra of compounds $\mathbf{5}_{a-e}$ showed characteristic absorption bands at 3463-3215 cm⁻¹ due to the NH₂ group and at 1721 cm⁻¹ and 1671 cm⁻¹

for the CO groups, with disappearance of the absorption band for the CN group. ¹HNMR spectra of compounds $\mathbf{5_{a-e}}$ represented the characteristic broad signal at δ 5.70–5.10 ppm due to the NH₂ group along with two singlet signals at δ 8.10 and δ 4.60 ppm due to the NH and CONH₂ groups. The reaction of compound **2** with ethyl cyanoacetate under PTC conditions yielded a mixture of ethyl 3-{[3-cyano-4,5-diphenyl-11H-pyrido(2,3-b)(1,5)benzodiazepin-2-yl]thio}-3-iminopropanoate **6** and 4-amino-3-cyano-5,6-diphenyl-2-oxo-12(H)thiopyrano[2,3-b]pyrido [2',3'b'](1,5)benzo-diazepine **7**, respectively (c.f. Scheme 1 and Table I).



SCHEME 1

August 2012
t 03:45 27
Library] at
Jniversity
[McGill [
vnloaded by
Dov

Compound
New
f the
Data o
ctral]
d Spe
al an
alytic
I An
TABLE

TABLE	I Analytical	and Sp	ectral Data	of the P	New Co	unoduu	ds		
Product	МР	Vield	Mole Form	Analy	tical Data	a ^b Calcd/Fo	pun	IR	¹ HNMB
No.	(∘C)a	(%)	(Mol. Wt.)	С	Н	Ν	S	$(\mathrm{cm}^{-1})^c$	$\vartheta \; (\mathbf{ppm})^d$
2	180 dioxane	66	$C_{25}H_{16}N_4S$ (404.49)	74.23 74.50	3.99 3.86	$13.85 \\ 13.96$	7.93 7.89	3309, 3196 (2NH); 2214 (CN); 1119 (C=S)	10.00 (s, 1H, NH); 9.50 (s, 1H, NH); 8.70-7.00 (m, 14H)
en en	214 dioxane	42	$C_{25}H_{18}N_4OS (422.50)$	71.07 71.26	4.29 4.42	13.26 13.00	7.59 7.34	$3385, 3290, 3184$ (NH, NH $_2$); 2205 (CN); 1669 (CO), 1122	arom.) 10.20 (s, 1H, NH); 8.30–7.00 (m, 14H, arom.); 6.60–6.50 (br, 1H, NH); 3.50 (s, 1H,
$4_{\rm a}$	200 ethanol	70	$C_{29}H_{22}N_4O_2S$ (490.58)	71.00 71.23	4.52 4.70	11.42 11.23	$6.54 \\ 6.41$	(U=S) 3218 (NH); 2978, 2922 (CH _{aliph} .); 2210 (CN); 1719 (CO)	CH), 1.7(s, 1H, SH) 7.55 (s, 1H, NH); 7.40–6.80 (m, 14H, arom.); 4.25–3.80 (q, 2H, CH ₂): 3.20 (s, 2H, CH ₂);
$4_{ m b}$	240 CHCl ₃	74	$C_{27}H_{17}N_5S$ (443.53)	$73.12 \\ 73.15$	3.86 3.93	$\begin{array}{c} 15.79\\ 15.86\end{array}$	7.23 7.50	3221 (NH); 2980 (CH _{aliph} .); 2205 (CN)	1.45-1.00 (t, 3H, CH ₃) 10.55 (s, 1H, NH); $8.65-7.60$ (m, 14H, arom.); 4.50 (s, 2H, CH)
$4_{\rm c}$	222 ethanol	55	$C_{33}H_{22}N_4OS$ (522.62)	75.84 75.89	4.24 4.38	$10.72 \\ 10.84$	$6.13 \\ 6.22$	3343(NH); 2921 (CH _{aliph} .); 2208 (CN);	CH2) 8.30-7.15 (m, 20H, arom. + NH); 4.85 (s, 2H, CH2)
$5_{\rm a}$	220 ethanol	82	$C_{29}H_{22}N_4O_2S$ (490.58)	71.00 71.16	4.52 4.39	11.42 11.56	6.54 6.32	1510 (CO) 3415, 3345, 3221 (NH, NH ₂); 2934 (CH _{aliph} .); 1710 (CO)	8.30 (s, 1H, NH); $7.65-6.40$ (m, 14H, arom.); 5.20 (s, 2H, NH ₂); $4.20-3.80$ (g, 2H, OH ₂); $4.20-3.80$ (g, 2H, OH ₂); $4.20-3.60$ (g,
б _b	268 benzene	60	$C_{27}H_{17}N_5S$ (443.53)	73.12 73.26	$3.86 \\ 3.74$	$\begin{array}{c} 15.79\\ 15.66\end{array}$	7.23 7.42	3415, 3334, 3233 (NH, NH ₂); 2192 (CN)	CH2); 1.500-0.50 (1, 5H, CH3) 8.40 (s, 1H, NH); 8.10-6.95 (m, 14H, arom.); 5.20 (s, 2H,
5 _c	230 CHCI ₃	43	${ m C}_{33}{ m H}_{22}{ m N}_4{ m OS}\ (522.62)$	75.84 75.69	4.24 4.29	$10.72 \\ 10.78$	6.13 6.29	3456, 3356, 3240 (NH, NH ₂); 1670 (CO)	$_{\rm (MH2)}^{\rm (MH2)}$ 10.35 (s, 1H, NH); 8.00–7.20 (m, 19H, arom.); 5.30 (s, 2H, NH2)

5 _d	204 pet. Ether	80	$C_{27}H_{19}N_5OS (461.54)$	70.26 70.39	4.15 4.28	15.17 15.32	6.95 6.87	3463, 3370, 3329, 3197 (NH, 2NH ₂); 1683 (9CO)	10.20 (s, 1H, NH); $8.30-6.90$ (m, 14H, arom.); 5.70 (s, 2H, NH ₀); 465 (s, $2H$ NH ₆)
5e	280 ethanol	60	$C_{33}H_{23}N_5OS$ (537.64)	$73.72 \\ 73.91$	4.31 4.50	$13.03 \\ 13.23$	$5.96 \\ 5.72$	(200) 3461, 3392, 3316, 3215 $(2NH, NH_2)$; 1680 (CO)	9.15(8, 1H, NH); 8.10(8, 1H, NH); 7.80-6.90(m, 19H, NH); 7.80-6.90(m, 19H, 19H, 19H, 19H, 19H)
9	124 benzene	40	$C_{30}H_{23}N_5O_2S$ (517.60)	69.61 69.83	4.48 4.91	13.53 13.26	6.19 6.02	3397, 3202 (2NH); 2204 (CN); 1700 (CO)	8.40 (s, 1H, NH); 7.45 (s, 1H, NH); 7.45 (s, 1H, NH); $7.30-6.60$ (m, 14H, arom.); $4.30-3.60$ (q, 2H, CH $_2$); $3.10(s, 2H, CH_2)$; $3.10(s, 2H, CH_2)$; $1.15-0.95$ (t, $3H$ CH $_2$); $3.10(s, 2H, CH_2)$; $1.15-0.95$ (t, $3H$ CH $_2$);
2	188 dioxane	23	$C_{28}H_{17}N_5OS$ (471.53)	71.32 71.51	3.63 3.83	$14.85 \\ 14.50$	$6.80 \\ 6.84$	$3424, 3318, 3215$ (NH, NH $_2$); 2216 (CN); 1636 (CO)	
œ	189 dioxane	78	$C_{19}H_{12}N_4S_2$ (360.45)	$63.31 \\ 63.53$	3.35 3.11	$15.54 \\ 15.77$	$17.79 \\ 17.59$	3430, 3330, 3194 (NH, NH ₂); 2202 (CN)	8.02(s, 1H, NH); 7.85–6.90 (m, 9H, arom.); 5.20–4.60 (br, 2H. NH ₂
10	200 ethanol	50	$C_{21}H_{16}N_4S_2$ (388.51)	64.92 64.83	4.15 4.00	14.42 14.60	16.51 16.55	3304, 3163 (NH ₂); 2211 (CN)	9.60 (s, 2H, NH ₂); 8.00–6.95 (m, 9H, arom.); 3.25–2.85 (q, 2H, CH ₂); 1.40–1.05 (t, 3H, CH ₂); CH ₂); 1.40–1.05 (t, 3H, CH ₂)
п	185 benzene	43	$C_{21}H_{18}N_4OS_2$ (406.52)	62.05 62.15	4.46 4.55	$13.78 \\ 13.50$	15.77 15.43	3427, 3340 , $3258(2NH_2); 1645 (CO)$	9.50 (s, 2H, NH ₂); 8.25–7.05 (m, 9H, arom.); 4.65 (s, 2H, NH ₂); 1.50–1.15 (q, 2H, NH ₂); 1.50–1.15 (q, 2H, CH ₂); CH ₂ ; 1.00–0.80 (t, 3H, CH ₃)
12	130 benzene	48	$C_{21}H_{15}N_{3}OS_{2}$ (389.49)	64.76 64.88	3.88 3.49	10.79 10.96	16.46 16.51	2209 (CN); 1667 (CO)	7.90–7.15 (m, 9H, arom.); 4.50 (s, 1H, CH); 4.00–3.80 (q, 2H, CH ₂); 1.30–1.10 (t, 3H, CH ₂)
13	140 methanol	.72	$C_{27}H_{21}N_5S$ (447.55)	72.46 72.31	4.73 4.85	15.65 15.69	7.16 7.09	3315, 3210 (NH ₂); 2206 (CN)	$\begin{array}{c} 8.00-7.00 \ (m, 14H, arom.); \\ 4.90 \ (s, 2H, NH_2); 2.90-2.40 \\ (q, 2H, CH_2); 1.10-0.80 \ (t, 3H, CH_2); 1.10-0.80 \ (t, 3H, CH_2); 0.10-0.80 \ (t,$
14	170 pet. ether	66	${ m C}_{27}{ m H}_{20}{ m N}_4{ m S}_2\ (464.60)$	69.80 69.71	4.34 4.30	12.06 12.18	$13.80 \\ 13.77$	2209 (CN)	8.20–6.85 (m, 14H, arom.); 8.20–6.85 (m, 14H, arom.); 4.40 (s, 1H, CH); $2.75–2.35$ (q, 2H, CH ₂); $1.15-0.90$ (t, 3H CH ₂);
15	218 ethanol	89	$C_{23}H_{18}N_4OS_2$ (430.54)	64.16 64.30	4.21 4.09	13.01 13.30	14.89 14.67	3290 (NH); 2930 (CH _{aliph} .); 2203 (CN); 1696 (CO)	8.80 (s. 1H, NH); 8.30–7.05 (m, 9H, arom.); 3.30–3.50 (q, 2H, CH ₂); 2.50 (s, 3H, CH ₃); 1.45–1.00 (t, 3H, CH ₃) (Continued on next page)

August 2012
5
ŝ
4
3
at
$\overline{}$
Library
niversity
5
[McGill
_y c
ql
loade
umo
D

TABLE I Analytical and Spectral Data of the New Compounds (Continued)

Product	МР	Vield	Mole Form	Analy	tical Data	a ^b Calcd/Fe	pund	TR	¹ HNMB.
No.	(C) ^a	(%)	(Mol. Wt.)	С	Н	N	ß	$(\mathrm{cm}^{-1})^c$	$\vartheta (\mathbf{ppm})^d$
16	272 ethanol	61	$\mathrm{C}_{23}\mathrm{H}_{18}\mathrm{N}_4\mathrm{OS}_2$	64.16	4.21	13.01	14.89	3270 (NH); 2924	8.95 (s, 1H, NH); 8.20–7.25 (m,
			(430.54)	64.00	4.33	13.21	15.04	(CH _{aliph} .)	9H, arom.), 4.00–3.50 (q,
								·	$2H, CH_2$); 2.35 (s, $3H, CH_3$);
									1.50 - 1.10 (t, 3H, CH ₃)
17	243 methanol	59	$C_{28}H_{21}N_5S_3$	64.22	4.04	13.37	18.37	3200, 3140 (2NH);	8.70 (s, 1H, NH); 8.30–6.90 (m,
			(523.69)	64.07	4.31	13.01	18.52	$2926(CH_{alinh.})$	15H, arom. $+ NH$);
									3.80-3.15 (q, 2H, CH ₂);
									$1.60 - 1.05 (t, 3H, CH_3)$
18	360 benzene	30	$\mathrm{C}_{18}\mathrm{H}_{12}\mathrm{N}_{4}\mathrm{S}$	68.33	3.82	17.71	10.13	3416, 3310, 3240 (NH,	9.00–7.25 (m, 10H, arom.+
			(316.38)	68.52	4.00	17.95	10.01	NH_2); 2201 (CN)	NH); 4.45–4.00 (br, 2H,
									$\rm NH_2)$
19	212 ethanol	67	$\mathrm{C}_{18}\mathrm{H}_{14}\mathrm{N}_{4}\mathrm{OS}$	64.65	4.22	16.75	9.59	3436, 3320, 3200, 3105	10.60 (s, 1H, NH); 8.25–7.20
			(334.39)	64.35	4.00	16.94	9.89	$(NH, 2NH_2)$; 1669	(m, 9H, arom.); 5.50 (s, 2H,
								(CO)	$\rm NH_2$); 5.00 (s, 2H, $\rm NH_2$)
20	180 ethanol	57	$C_{20}H_{17}N_3O_2S$	66.10	4.71	11.56	8.82	3220, 3205, 3110 (NH,	11.05 (s, 1H, NH); $8.30-7.00$
			(363.43)	66.32	4.52	11.71	8.97	NH_2); 2918 (CH _{aliph} .);	(m, 9H, arom.); 4.30 (s, 2H,
								1681 (CO)	NH_2); 3.70–3.20 (q, 2H,
									CH_2); 1.60–1.10 (t, 3H, CH_3)
21	120 benzene	20	$C_{18}H_{11}N_{3}OS$	68.12	3.49	13.24	10.10	3196 (NH); 2926	8.85 (s, 1H, NH); 8.30–7.05 (m,
			(317.36)	68.41	3.60	13.10	10.31	$(CH_{aliph.})$; 2203 (CN) ;	9H, arom.); 4.40 (s, 1H, CH)
								1700 (CO)	
^a Uncor	rected.								

 b Satisfactory microanalysis obtained, C, ± 0.35 ; H, ± 0.4 ; N, ± 0.2 ; S, ± 0.2 .

 $^{c}\,\mathrm{Measured}$ by a Nicolet FT-IR 710 spectrophotometer.

^dMeasured by a Varian EM 360 L spectrometer at 60 MHZ using TMS as a internal standard and DMSO as a solvent.

Compound 1 was allowed to react with carbon disulfide and malononitrile in the presence of triethylamine as a basic catalyst 2-amino-1-cyano-11(H)-5-phenyl-4-thioxothiopyrano[4,3to afford bl(1.5)benzodiazepine 8. Moreover, the reaction of 1-ethyl-3H-4-phenyl-1,5-benzodiazepin-2-one 9^{10} with carbon disulfide and active nitriles, namely malononitrile, cyanothioacetamide, cyanoacetamide, or ethyl cyanoacetate in the presence of triethylamine as a basic catalyst, yielded 2-amino-1-cyano-11-ethyl-5-phenyl-4thioxothiopyrano[4,3-b](1,5)benzodiazepine **10**, 2-amino-11-ethyl-5phenyl-4-thioxo-thiopyran[4,3-b](1,5)benzodiazepine-1-carboxamide 11. 1-cyano-11-ethyl-2-oxo-5-phenyl-4-thioxothiopyrano[4,3and b](1,5)benzodiazepine 12, respectively. In analogy, compound 9 was treated with a mixture of phenyl isothiocyanate and malononitrile or cyanothioacetamide in presence of triethylamine as a basic catalyst to give 2-amino-1-cvano-3,5-diphenyl-11-ethyl-4-thioxopyrido[4,3b](1,5)benzodiazepine 13 and 1-cyano-3,5-diphenyl-2,4-dithioxo-11ethyl-pyrido[4,3-b](1,5)benzodiazepine 14, respectively. The reaction pathway was suggested to be a preliminary formation of carbanion of the CH_{2 benzodiazepine} group, which was added to the C=S bond followed by a nucleophilic attack of the SH group or the NH group at the CN. CO, and CS groups followed by condensation of the active methylene and the $C=O_{benzodiazepine}$ group with the eliminaton of H_2S molecule in case of cyanothioacetamide, water molecule in case of cyanoacetamide, or ethanol molecule in case of ethyl cyanoacetate. The IR spectra of compounds 8 and 10-14 exhibited new absorption bands at 3430-3163 cm^{-1} for the NH₂ group and 2211–2202 cm^{-1} for the CN group. The ¹HNMR spectra of these compounds showed the disappearance of the signal corresponding to the CH_{2 benzodiazepine} group and exihibted multiplet signals at δ 8.00–7.00 for aromatic protons, a broad signal at δ 5.20–4.30 ppm for the NH₂ group, and a singlet signal at δ 4.1 for the CH group, respectively.

Treatment of compound **10** with acetic anhydride along with pyridine gave 2-acetylamino-1-cyano-11-ethyl-5-phenyl-4-thioxothiopyrano[4,3b](1,5)-benzodiazepine **15**, which was converted into 13-ethyl-1-imino-3-methyl-7-phenyl-6-thioxo(1,3)oxazino[4,5-b]thiopyrano[4',3'-b'] (1,5) benzodiazepine **16** in boiling pyridine. The cyclization of compound **10** into 2,7-diphenyl-3,6-dithioxo-13-ethyl-1-imino-4(H)-pyrimido[4,5b]thiopyrano[4',3-b,](1,5)benzodiazepine **17** was achieved by treating it with phenylisothiocyanate. The IR spectra of compounds **16** and **17** showed the absence of absorption bands corresponding to the NH₂ and CN groups and revealed a new absorption band at 3270–3140 cm⁻¹ corresponding to NH groups. ¹HNMR spectra were consistent of the proposed structures. Furthermore, the reaction of elemental sulfur and active methylene, namely malononitrile, cyanothioacetamide, cyanoacetamide, or ethyl cyanoacetate with compound **1** in presence of triethylamine as a basic catalyst, gave 2-amino-1-cyano-4-phenyl-10(H)thieno[3,2-b] (1,5)benzodiazepine **18**,¹⁰ 2-amino-1-carboxamido-4-phenyl-10(H)thieno[3,2-b](1,5)benzodiazepine **19**, 2-amino-1-carbethoxy-4-phenyl-10 (H)thieno[3,2-b](1,5)benzodiazepine **20**, and 1-cyano-1, 10-dihydro-2-oxo-4-phenylthieno[3,2-b](1,5)benzodiazepine **21**, respectively. IR spectra of compounds **18–21** exhibited new absorption bands at 3436–3105 cm⁻¹ for the NH₂ group, 2201 and 2203 cm⁻¹ for the CN group in compounds **18** and **20**, and 1700–1669 cm⁻¹ for C=O groups in compounds **19–21**. ¹HNMR spectra of these compounds showed the disappearance of the signal specific for the CH_{2 benzodiazepine} group.

EXPERIMENTAL

Synthesis of Compounds 2 and 3: General Procedure

A mixture of compound 1 (0.01 mol, 2.36 g), benzylidenecyanothioacetamide (0.01 mol, 1.88 g), and piperidine (1 mL) was refluxed in dioxane (20 mL) for 4 h. On cooling, the formed precipitate was filtered off and crystallized to give compound **3**. The filtrate was poured into a mixture of water and HCl (50: 3 v/v), and the solid product was filtered off, washed with water, and crystallized to give compound **2** (cf. Scheme 1, Table I).

Synthesis of Compounds 4_{a-c} : General Procedure

A mixture of compound 2 (0.005 mol, 2.02 g); 0.005 mol of the appropriate halocompound; ethyl chloroacetate (0.54 mL), chloroacetonitrile (0.31 mL), or phenacyl bromide (0.99 g); and sodium acetate (0.005 mol, 0.41 gm) in ethanol (20 mL) was refluxed for 2 h. The precipitate that obtained on cooling was filtered off, washed with water, and crystallized from the appropirate solvent (cf. Scheme 1, Table I).

Synthesis of Compounds 5_{a-c} : Method A (General Procedure)

To a solution of the appropriate compound 4_{a-c} (0.01 mol) in dioxane (20 mL), anhydrous potassium carbonate (3 g), and TBAB (0.003 g) were added. The reaction mixture was stirred for 5 h at 60°C until the completion of the reaction (TLC). The reaction mixture was filtered off, and the filtrate evaporated in vacuo. The residual solid was washed with water and crystallized from the appropriate solvent (cf. Scheme 1, Table I).

Synthesis of Compounds 5_{a-e}: Method B (General Procedure)

A mixture of anhydrous potassium carbonate (3 g); dry dioxane (30 mL); compound 2(0.005 mol, 2.02 g); the appropirate halocompound; ethylchloroacetate (0.54 mL), chloroacetonitrile (0.31 mL), phenacyl bromide (0.99 g) chloroacetamide (0.46 g), or chloroacetanilide (0.85 g); and TBAB (0.003 g) was stirred for 5 hr at 60°C until the completion of the reaction (TLC). The reaction mixture was filtered off, and the filtrate evaporated in vacuo. The residual solid was washed with water and crystallized from the appropriate solvent (cf. Scheme 1, Table I).

Synthesis of Compounds 6 and 7: General Procedure

A mixture of anhydrous potassium carbonate (3 g), dry dioxane (30 mL), compound **2** (0.005 mol, 2.02 g), ethyl cyanoacetate (0.005 mol, 0.53 mL), and TBAB (0.003 g) was stirred for 4 h at 60°C until the completion of the reaction (TLC). The reaction mixture was filtered off, and the filtrate evaporated in vacuo. The residual solid was washed with water and crystallized to give compound **6**. The precipitate (carbonate layer) was dissolved in water (50 mL) and acidified by HCl, and the solid product was filtered off, washed with water, and crystallized to give compound **7** (cf. Scheme 1, Table I).

Synthesis of Compounds 8 and 10–14: General Procedure

An equimolar amount (0.01 mole) of compound 1 (2.36 g) or compound 9 (2.64 g) in ethanol (20 mL), carbondisulfide (1.14 mL), or phenylisothiocyanate (1.3 mL), along with triethylamine (2 mL) were added. The reaction mixture was stirred at r. t. for 2 h, and the suitable active methylene, namely malononitrile (0.66 g), cyanothioacetamide (1 g), cyanoacetamide (0.8 g), ethyl cyanoacetate (1.1 mL), and dimethylformamide (2 mL), was added. The reaction mixture was refluxed for 4 h. After cooling, the reaction mixture was poured into water and HCl (100: 5 v/v). The solid product was filtered off, washed with water, and crystallized from the appropirate sovent (cf. Scheme 2, Table I)

Synthesis of Compound 15

A mixture of compound **10** (0.001 mol, 0.388 g), acetic anhydride (0.001 mol, 0.1 mL), and dry pyridine (20 mL) was refluxed for 1 h The reaction mixture was poured into ice-cold water. The separated solid was collected by filtration, washed with water, and crystallized (cf. Scheme 2, Table I).





SCHEME 3

Synthesis of Compound 16

A solution of compound **15** (0.001 mol, 0.43 g) in dry pyridine (20 mL) was refluxed for 5 h. The reaction mixture was poured into ice-cold water containing few drops of HCl. The separated solid was collected by filtration and crystallized (cf. Scheme 2, Table I).

Synthesis of Compound 17

A mixture of compound **10** (0.001 mol, 0.388 g), phenyl isothiocyanate (0.001 mol, 0.12 mL), and dry pyridine (20 mL) was refluxed for 10 h. The reaction mixture was poured into ice-cold water. The separated solid was collected by filtration and crystallized (cf. Scheme 2, Table I)

Synthesis of Compounds 18–21: General Procedure

To a stirred solution of compound 1 (0.01 mol, 2.36 g) in dry dioxane (20 mL), sulphur (0.01 mol, 0.32 g) and triethylamine (0.4 mL) were added. The reaction mixture was refluxed for 1 h, and then 0.01 mol of the appropirate active methylene, namely malononitrile (0.66 g),

cyanothioacetamide (1 g), cyanoacetamide (0.88 g), or ethyl cyanoacetate (1.13 mL), was added. The reaction mixture was refluxed for 4 h. After cooling, the solid precipitate was filtered off, washed with water, and crystallized from the appropirate solvent. The filtrate was evaporated in vacuo, and the residual solid was washed with water, filtered off, dried, and crystallized to give compound **20** (cf. Scheme 2, Table I).

REFERENCES

- [1] Y. Cui, X. B. Tang, C. X. Shao, and N. H. Sun, Chinese J. Chem. Soc., 23, 589 (2005).
- [2] R. Janciene, L. Kosychova, V. Bukelskiene, V. Domkus, Z. Stumbevicute, V. Ragaleviciene, et al., Arzneimittel-forschung, 52, 475 (2002); C. A., 138, 100346d (2003).
- [3] M. Dibraccio, G. Grossi, G. Roma, L. Vargiu, M. Mura, and E. M. Marongiu, *Europ. J. Med. Chem.*, **36**, 935 (2001).
- [4] I. A. O'Neil, S. Thomposon, S. B. Kalindjio, and T. C. Jenkins, *Tetrahydron Lett.*, 44, 7809 (2003).
- [5] W. Namrocka, B. Sztuba, A. Opolsk, J. Wietrzyk, M. W. Kowalska, and T. Glowiak, Arch. Pharm., 334, 3 (2001).
- [6] H. Abdel-Ghany, A. M. El-Sayed, A. A. Sultan, and A. K. El-Shafei, Synth. Comm., 20, 893 (1990).
- [7] A. M. El-Sayed, H. Abdel-Ghany, and A. M. M. El-Saghier, Synth. Comm., 29, 3561 (1999).
- [8] H. Abdel-Ghany, A. M. El-Sayed, A. Khodairy, and H. Salah, Synth Comm., 31, 2523 (2001).
- [9] A. Khodairy, A. M. El-Sayed, H. Abdel-Ghany, and H. Salah, J. Chinese Chem. Soc., 50, 1195 (2003).
- [10] A. Khodairy, Phosphorous, Sulfur, and Silicon, 180, 1893 (2005).
- [11] A. Khodairy, A. M. El-Sayed, H. Salah, and H. Abdel-Ghany, Synth. Comm., submitted.
- [12] W. Reid and P. Stahlofen, Chem. Ber., 90, 825 (1957).