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Synthesis and biological evaluations of P4-benzoxaborole-substituted macrocyclic inhibitors of HCV NS3 protease

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ABSTRACT

We disclose here a series of P4-benzoxaborole-substituted macrocyclic HCV protease inhibitors. These inhibitors are potent against HCV NS3 protease, their anti-HCV replicon potencies are largely impacted by substitutions on benzoxaborole ring system and P2* groups. P2* 2-thiazole-isoquinoline provides best replicon potency. The in vitro SAR studies and in vivo PK evaluations of selected compounds are described herein.

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Infection with Hepatitis C Virus (HCV) is a major cause of human liver disease throughout the world, affecting over 200 million individuals. In the US alone, an estimated 4.5 million Americans are chronically infected. HCV infection is responsible for 40–60% of all chronic liver disease cases and 30% of all liver transplants. The current standard of care for HCV infection is a combination of injectable pegylated interferon- α (PEG IFN- α) plus oral ribavirin, which is effective in only about 50% of genotype-1 patients achieving sustained viral response.¹ This protocol has been associated with side effects including neuropsychiatric events, flu-like symptoms and hematological toxicities.² Therefore, there has been tremendous interest in the development of more effective therapeutics in treating HCV infection. One of the validated targets is HCV NS3/4A serine protease.³

Extensive efforts in the discovery of HCV NS3 protease inhibitors have resulted in a number of drug candidates at various stages of clinical development.⁴ The two most advanced compounds, VX-950 (telaprevir) and SCH-503034 (boceprevir), provided an

* Corresponding author. E-mail address: charles.ding@gmail.com (C.Z. Ding). early proof of concept in suppressing the virus and are currently undergoing Phase III clinical trials.⁵ Newer protease inhibitors with improved potency, different mode of binding interaction with the protease enzyme and pharmacokinetic properties have emerged. These inhibitors, such as danoprevir (ITMN-191),⁶ TMC-435,⁷ BMS-791325⁸ (Fig. 1), and vaniprevir (MK-7009)⁹ are in clinical development representing structural diversity set of promising HCV protease inhibitors.

This initial excitement about the potential novel HCV treatments has been somewhat dampened by a quick emergence of enzyme resistance to these agents.¹⁰ The opportunity for other chemical classes of HCV protease inhibitors with better resistance profile still exists. In our search for novel HCV protease inhibitors, we considered benzoxaborole as the P4 moiety, the fourth aminoacid residue from carboxyl terminus. Benzoxaboroles (core structure shown in compound **2** in Scheme 1) are a chemical class of organoboron compounds that have excellent physicochemical and biological properties.¹¹ They are metabolically stable, and exhibit good water solubility. Docking studies of compound **4** to the enzyme active site suggest that benzoxaborole moiety with suitable orientation and linkage can potentially interact with active site polar aminoacid residues: Ser122, Arg123, Arg155 and

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Scheme 1. General syntheses of benzoxaborole-substituted macrocyclic HCV inhibitors. Reagents and conditions: (a) TFA, DCM, rt basic work-up, 83%; (b) triphosgene, TEA, THF, -40 °C; then 2, 12, 21 and 22, 15–30%; for 10, HATU, DIEA, 26, DMF, rt 50%; (c) *p*-O₂N-Ph-OCOCI, 25, ACN, rt 90%; (d) amine 1, ACN, 60 °C; 50%.

Asp168 of HCV NS3/4A serine protease. These additional interactions may provide better resistance profile than the known inhibitors that do not have. Syntheses of the target compound **4** and its analogs as well as their biological evaluations are reported herein.



Scheme 2. Preparation of 6-hydroxybenzoxaborole. Reagents and conditions: (a) Tf_2O , pyridine, -78 °C to rt; (b) (pinB)₂, PdCl₂(dppf), KOAc, dioxane, 80 °C; (c) NaBH₄, MeOH, rt; (d) HCl; (e) BBr₃, CH₂Cl₂, -78 °C to rt.

P4-Benzoxaborole-substituted macrocyclic compounds based on ITMN-191 scaffold were prepared using a general scheme as



Scheme 3. Preparation of substituted 6-aminobenzoxaboroles. Reagents and conditions: (a) for **18a**: NaBH₄, MeOH; for **18b**: MeMgBr, THF, 0 °C, \sim 93%; (b) B(iPrO)₃, *n*-BuLi, THF, -70 °C, 40–47%; (c) fuming HNO₃, -45 °C, 80–90%; (d) Raney Ni, hydrazine monohydrate, MeOH, rt, 70–80%.



Scheme 4. Preparation of amine 25 and acid 26. Reagents and conditions: (a) NaBH₄, MeOH, 0 °C, 99%; (b) $B(iPrO)_3$, BuLi, THF, -78 °C, 50%; (c) LAH, THF, rt, 46%; (d) concn HCl, reflux, 85%

shown in Scheme 1. ITMN-191 was prepared by following a patent procedure.¹² The P4 BOC group was removed by treatment with TFA in dichloromethane to give amine **1**. The coupling of the amine to 6-aminobenzoxaboroles (**2**, **21** and **22**) providing urea compounds (**4**, **8** and **9**) was accomplished via an isocyanide intermediate, which was prepared by treatment of compound **1** with triphosgene in the presence of triethylamine at low temperature. The intermediate was not isolated, 6-aminobenzoxaboroles were added in situ. This reaction sequence provided urea compounds in modest 30% isolated yield.¹³ The carbamate-linked P4-benzox-aborole-substituted macrocyclic compounds (**5** and **6**) were generated going through the same isocyanate intermediate reacting with hydroxybenzoxaboroles instead (**11** and **12**). The synthesis of urea compound **7** involves conversion of 6-aminomethylbenzoxaborole

25 to its *p*-nitrophenyl carbamate **3**, followed by reaction of it with macrocyclic amine **1** in somewhat higher (50%) yield. The amide **10** was prepared in 50% by coupling of 6-benzoxaborole carboxylic acid **26** with macrocyclic amine **1** in the presence of HATU, DIEA in DMF at room temperature.

The functionalized benzoxaborole compounds necessary for the preparation of benzoxaborole-substituted macrocyclic HCV inhibitors were prepared as follows. The 6-aminobenzoxaborole **2** (R^1 =NH₂, R= R^2 =H) was prepared according to a published procedure.¹⁴ Both 5- and 6-hydroxybenzoxaboroles (**11** and **12**) were prepared in the same fashion starting from separate starting material 5 or 4-methoxysalicylaldehyde, as illustrated for 6-hydroxybenzoxaborole in Scheme 2. 4-Methoxysalicylaldehyde was converted to its triflate **14** by treatment with triflic anhydride in the presence of pyridine at low temperature. Boronation was accomplished by pinacol diborate in the presence of a palladium catalyst to provide boronobenzaldehyde **15**. Reduction with sodium borohydride and acidic work up gave 6-methoxybenzoxaborole **16**. Cleavage of the methoxy group by BBr₃ provided 6-hydroxybenzoxaborole **12**.

6-Amino-5-fluorobenzoxaboroles **21** and **22** were prepared as shown in Scheme 3 from 2-bromo-5-fluorobenzaldehyde **17**. Reduction with sodium borohydride or addition of methyl magnesium bromide to **17** produced benzyl alcohol **18a** or **18b** in high yield. Halogen-metal exchange reaction and deprotonation were accomplished using two equivalents of *n*-butyllithium, the resulting dianion was reacted with triisopropyl borate and acidic work-up providing benzoxaboroles **19a** and **19b**. Nitration of both compounds with fuming nitric acid produced 6-nitro derivatives



Scheme 5. Preparation of isoquinoline- and quinoline-based HCV inhibitors. Reagents and conditions: (a) KOtBu, DMF, rt; then, **27a** or **27b**, 60–70%; (b) HCl(g), dioxane, rt, ~85%; (e) triphosgene, Et₃N, CH₂Cl₂, -40 °C, then, compound **2** or **25**, 20–30%.

20a and **20b** with high regio-selectivity and yield. Raney nickel reduction of the two nitro compounds provided 6-amino-7-fluor-obenzoxaboroles **21** and **22**.

6-Aminomethylbenzoxaboroles **25** and benzoxaborole-6-carboxylic acid **26**¹⁵ were prepared as shown in Scheme **4**. 2-Bromo-4-cyanobenzaldehyde was converted to 6-cyanobenzoxaborole **24** by following the same sequence of reactions as shown in Scheme **3** for preparation of benzoxoborole **19a** and **19b**. The aminomethylbenzoxaborole **25** was obtained from reduction of compound **24** using lithium aluminium hydride, while compound **26** was obtained from hydrolysis of compound **24** with concn HCl.

Inhibitors with P2* groups other than isoindoline (ITMN-191 scaffold) were prepared as described in Scheme 5 through heteroarvl displacement reactions. Prerequisite chloro-isoquinoline 27a and chloro-quinoline 27b were prepared by following the published procedures.¹⁶ The other coupling partner hydroxymacrocycle **28** was also prepared according to a published reference procedure.¹⁷ Heteroaryl displacement reaction of compound **28** with either 27a or 27b was achieved in good yield by treatment of 28 with slightly more than two equivalents of potassium tert-butoxide in DMF at room temperature. Both products 29a and 29b were treated with HCl in dioxane to produce aminomacrocycles 30a and 30b. Linking of either 30a or 30b with both 6-aminobenzoxaborole 2 and 6-aminomethylbenzoxaborole 25 was accomplished by the same reaction sequence as described in Scheme 1. Compounds 31 and 32 are products of amine 30a coupled to 6-aminobenzoxaborole 2 and aminomethylbezoxaborole 25; while 33 and 34 are products of amines 30b coupled with the same benzoxaboroles.

Biological evaluations of the compounds were done both using HCV protease and replicon assays. Results are shown in Table 1. The protease inhibitory IC_{50} 's were determined using a FRET assay with HCV NS3/4A 1a protease domain.¹⁸ The replicon EC_{50} 's were determined using a replicon luciferase cell-based assay.¹⁹ The initially designed compound **4** proved equipotent than danoprevir (ITMN-191) with enzymatic activity IC_{50} of 0.4 nM. While **4** was potent in the genotype 1b assay (EC_{50} 3.7 nM), the P4 structural



Figure 1. Representative HCV protease inhibitors in clinical development.

Table 1

HCV protease NS3/4 1a IC_{50} and replicon EC_{50} values for ITMN191-based P4-benzoxaborole-substituted macrocyclic inhibitors as compared to ITMN191





^a FRET assay with HCV NS3 1a protease domain. Values are means of duplicate or triplicate experiments; errors are usually within ±10%.

change turned out to be detrimental to compound's potency against genotype 1a replicon activity (EC₅₀ 159 nM). Encouraged by this result, we set out to further explore the linker SAR. Changing the urea linker to its corresponding carbamate (compound 5), the potency dropped off in both enzymatic or replicon assays. Compound **6** is a regioisomer of compound **5**, its potency was even worse. However, homologation of the urea linker (in compound 7) restored enzyme activity and, compared to lead compound 4, improved replicon EC₅₀ by 4 fold for genotype 1b, although similar for genotype 1a. The amide linker in compound 10 provided similar potency or slightly inferior to compound 4. Next, we explored substitutions on benzoxaborole ring system by improving lipophilicity, reducing polar surface area (PSA) of the ring while maintaining the oxaborole functionality, aimed to improve replicon potency for genotype 1a. We kept the urea linker for this exploration because it showed best potency and ease of synthesis. Compound 8



Table 2

HCV protease NS3/4 1a IC₅₀ and replicon EC₅₀ values for isoquinoline and quinoline P2* inhibitors as compared to TMC435

is a fluoro derivative of compound **4** with excellent enzyme potency and improvement in both genotype 1a and 1b replicon activity by 2 to 3-fold. The compound **9** with both methyl and fluoro substitutions on benzoxaborole ring improved further cellular potency in genotype 1a. Preliminary SAR in this series strongly suggests that it should be possible to further optimize cellular potency by further modifications in the benzoxaborole ring.

We decided to explore the impact of P2* groups on the anti-HCV potency of P4-benzoxoborole substituted macrocyclic compounds. We picked representative isoquinoline and quinolone groups from other classes of HCV protease inhibitors, namely BMS-791325 and TMC-435. The in vitro profiles of these compounds are summarized in Table 2. Interestingly, compounds **31** and **32** with isoquinoline P2* have similar enzyme and replicon potency profile to isoindo-line P2*. The more bulky, elaborated quinoline P2*-containing compounds **33** and **34** showed better replicon potency, although enzymatic potency fell off. We believe that some of the differences between the enzyme and replicon potency may originate from factors that differentiate both assays, such as different transcellular transport.

Selected compounds were evaluated in rats for their pharmacokinetic parameters, blood samples from both jugular and portal vein were drawn and drug concentrations were measured. The results are shown in Table 3. The oral absorption was calculated from the portal vein drug concentrations and oral bioavailability was calculated from jugular vein drug concentrations as compared to the drug concentrations after IV administration. ITMN-191 exhibits calculated 17.4% absorption and 20% oral bioavailability in rats.

Table 3

Physicochemical properties and oral PK parameters of selected benzoxaborolemacrocyclic inhibitors

Compds	MW	c log P	PSA	Oral PK in rats ^a	
				% Absorption	% BA
ITMN191	732	5.6	181	17.4	20
4	807	4.5	213	0.9	0.5
7	821	4.7	213	0.6	0.7
34	939	7.7	226	<lod< td=""><td><lod< td=""></lod<></td></lod<>	<lod< td=""></lod<>

^a % Absorption was calculated from portal vein drug concentration, while % BA was from jugular vein.

However, benzoxaborole-substituted macrocylic inhibitors **4**, **7** and **34** displayed minimal to undetectable level of absorption and oral bioavailability. We noticed good water solubility of these inhibitors compared to ITMN191 when the PK samples were made, however, their permeability and absorption are almost certainly limited by their high molecular weight and high polar surface area (PSA).

In summary, we have designed and synthesized a series of P4-benzoxaborole-substituted macrocyclic HCV protease inhibitors. We suggest that the benzoxaborole moiety can be a useful moiety towards developing compounds retaining potency against resistant enzymes.^{20a} These compounds exhibited potent inhibitory activity against HCV NS3/4 protease. Their cellular replicon potencies were impacted by substitutions on the benzoxaborole ring system and P2* groups, but even a limited exploration with compounds **8** and **9** suggest that further potency optimization should be possible. These compounds had high polar surface area (PSA), which may initially limit their oral absorption and bioavail-ability. However, our results with a related series suggest that relatively simple structural changes can bring these molecules back into more drug-like parameters.^{20b}

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