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PAPER

Effective 1,5-, 1,6- and 1,7-remote stereocontrol in reactions of alkoxy- and hydroxy-substituted allylstannanes with aldehydes[†]

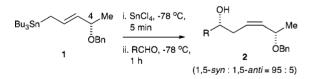
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Alk-2-enylstannanes with 4-, 5- and 6-alkoxy- or -hydroxy-substituents are transmetallated stereoselectively with tin(IV) halides to generate allyltin trihalides which react with aldehydes to give (*Z*)-alk-3-enols with useful levels of 1,5-, 1,6- and 1,7-stereocontrol. Alk-2-enylstannanes with a stereogenic centre bearing a hydroxy or alkoxy group at the 4-, 5- or 6-position, react with overall (*Z*)-1,5-, 1,6- and 1,7-syn-stereoselectivity with respect to the hydroxy and alkoxy substituents. The analogous reactions of alkoxy- and -hydroxyalk-2-enylstannanes with a methyl bearing stereogenic centre at the 4- or 5-position react with overall (*Z*)-1,5- and 1,6-*anti*-stereoselectivity with respect to the hydroxy and methyl substituents.

Introduction

4-Benzyloxypent-2-enylstannane **1** is transmetallated by tin(IV) chloride to generate an allyltin trichloride which reacts with aldehydes to give the (*Z*)-1,5-*syn*-5-benzyloxyalk-3-en-1-ols **2** with excellent 1,5-stereocontrol.¹ This stereoselectivity usually dominates the intrinsic stereochemical bias of a chiral aldehyde and iterative reactions using an octa-2,7-dienylstannane have been used to prepare aliphatic 1,5,9,13-polyols and ethers with useful stereoselectivity.²



It was of interest to see whether remote stereocontrol was also observed for tin(IV) halide mediated reactions of alk-2enylstannanes which had alkoxy and hydroxy substituents at the 5- and 6-positions. We now report full details of the 1,5- and 1,6stereocontrol observed in the tin(IV) halide promoted reactions between aldehydes and the 5-alkoxy(hydroxy)alk-2-enylstannanes **I** and **II**, and 1,5-, 1,6- and 1,7-stereocontrol found for the analogous reactions with aldehydes of the 6-alkoxy(hydroxy)alk-2-enylstannanes **III–V**, see Fig. 1.^{3,4}

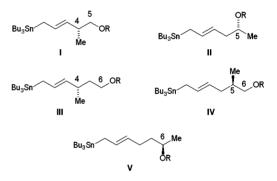


Fig. 1 Alk-2-enylstannanes selected for study.

Results and discussion

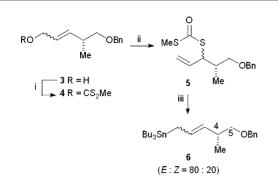
1,5-Stereocontrol using 5-alkoxyalk-2-enylstannanes I

The (*R*)-5-benzyloxy-4-methylpent-2-enylstannane **6** was synthesized as outlined in Scheme 1. Treatment of (*R*)-5-benzyloxy-4methylpent-2-enol (**3**)⁵ (e.e. >85%, Mosher's), an 80:20 mixture of (*E*)- and (*Z*)-isomers, with sodium hydride, carbon disulfide and methyl iodide gave the xanthate **4**. This on heating in toluene under reflux isomerised to give a 50:50 mixture of the epimeric dithiocarbonates **5**, which were converted into the 5benzyloxypent-2-enylstannane **6**, as an 80:20 mixture of (*E*)- and (*Z*)-isomers, by heating with tributyltin hydride in benzene under free radical conditions.

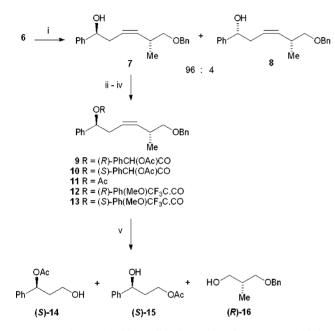
Tin(iv) chloride was added to the mixture of (*E*)- and (*Z*)isomers of the alkenylstannane **6** in dichloromethane at -78 °C to effect transmetallation. After 10 min, benzaldehyde was added and after a further hour the reaction was worked up to give a mixture of the 1,5-*anti*- and 1,5-*syn*-(*Z*)-hex-3-en-1-ols **7** and **8** in a ratio of 96:4, see Scheme 2. (*E*)-Alk-3-enols were not isolated and were only very minor products (<2%).

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 $[\]dagger$ Electronic supplementary information (ESI) available: full experimental details for the chemistry outlined in Schemes 3, 4, 5, 8, 10, 11, 12, 17, and 18 and all compounds listed in Table 1–6 together with copies of representative ¹H and ¹³C NMR spectra. See DOI: 10.1039/c0ob01084g



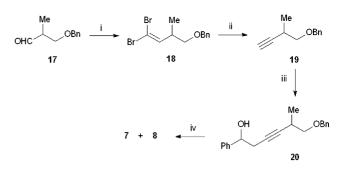
Scheme 1 Preparation of stannane 6. Reagents and conditions: i, NaH, r.t., 1 h, CS₂, r.t., 3 h, then MeI, r.t. 15 h [76%; (E) : (Z) = 80 : 20]; ii, heat in toluene, 15 h (*ca.* 100%; 50 : 50 mixture); iii, Bu₃SnH, AIBN (trace), benzene, heat, 2.5 h [86%; (E) : (Z) = 80 : 20].



Scheme 2 Tin(IV) chloride mediated reaction of stannane 6 with benzaldehyde. Reagents and conditions: i, $SnCl_4$, -78 °C, 10 min, then PhCHO, 1 h (83%; 7:8=96:4); ii, (*R*)- or (*S*)-*O*-acetyl-mandelic acid, DCC, DMAP (cat), DCM, r.t., 15 h (9, 83%; 10, 88%); iii, Ac₂O, DMAP (cat), Et₃N, DCM, r.t., 50 h (89%); iv, (*S*)- or (*R*)-Mosher's acid chloride, py., CCl₄, r.t. (12, 86%; 13, 51%); v, O₃, CHCl₃, -60 °C, 30 min, DMS, r.t., then NaBH₄, MeOH, r.t., 3 h [(*S*)-14, 29%].

The major and minor products 7 and 8 could be distinguished by ¹H NMR, their ratio being estimated by integration of the doublets at $\delta 0.91$ and 0.95 attributed to their 5-CH₃ groups. To confirm that the two products were the epimers 7 and 8, a mixture was prepared by reacting racemic styrene oxide with the racemic alkyne **19**,⁶ which was available in two steps from 3-benzyloxy-2methylpropanal **17**, see Scheme 3. This gave the hex-3-ynol **20** as a mixture of diastereoisomers and hydrogenation using Lindlar's catalyst gave a 50:50 mixture of the epimeric alcohols 7 and 8. Comparison with the products from the tin(IV) chloride mediated reaction of stannane **6** with benzaldehyde confirmed that the (*Z*)-1,5-*syn*-epimer **8** was the minor product.

The 3,4-vinylic coupling constant of the major product 7 from the stannane reaction was 10 Hz consistent with the (Z)-geometry. The 1,5-*anti*-configuration was assigned by comparison of the ¹H



Scheme 3 Synthesis of a mixture of the epimeric alcohols 7 and 8. Reagents and conditions: i, CBr₄, Ph₃P, Zn powder, DCM, r.t., 24 h, then add **17**, r.t., 2.5 h (74%); ii, *n*-BuLi in hexane, -78 °C, 2 h, then r.t., 1 h (65%); iii, **19**, *n*-BuLi, THF, -78 °C, 20 min, BF₃. Et₂O, -78 °C, 20 min, add styrene oxide, -78 °C, 1.5 h (14%); iv, Lindlar's cat., H₂, r.t., 72 h (72%, a 50 : 50 mixture of 7 and 8).

NMR spectra of the (*R*)- and (*S*)-acetylmandelates **9** and **10**^{7.8} and was confirmed by ozonolysis of the acetate **11** with a reductive work-up, which gave the (*S*)-3-acetoxy-3-phenylpropanol (*S*)-**14**, $[\alpha]_{D} -79$ (*c* 0.61 in CHCl₃)^{1a,9} and a mixture of the primary acetate (*S*)-**15**, formed by 1,3-migration of the acetyl group, and 3-benzyloxy-2-methylpropanol (*R*)-**16**, derived from the other fragment of the ozonolysis. The ¹⁹F NMR spectra of the Mosher's derivatives **12** and **13** showed two products, in a ratio of *ca*. 91:9, the discrepancy from the 96:4 ratio of epimers **7** and **8** as estimated by ¹H NMR being attributed to the e.e. of 85% of the alkenylstannane **6** (*cf*. the e.e. of alcohol **3**).

Tin(iv) chloride mediated reactions of allylstannane **6** with several aldehydes were investigated and the results are shown in Table 1. In all cases, useful diastereoselectivities in favour of the (*Z*)-1,5-*anti*-products were observed even for reactions of the stannane with enantiomeric chiral aldehydes. ¹H NMR confirmed the (*Z*)-configuration of the double-bonds of the major products and the diastereoselectivities of the reactions of the stannane with 4-methoxy- and 4-chloro-benzaldehyde.

 $Table 1 \quad 1,5\text{-}Stereocontrol in tin({\tt IV}) chloride mediated reactions between aldehydes and allylstannane 6$

	RCHO 6 -SnCl₄ -78 ℃	R R Me	`OBn
R	1,5-anti-product ^a	Yield (%)	1,2-anti: 1,5-syn
Ph 4-MeOC ₆ H ₄ 4-ClC ₆ H ₄ Me ₂ CH Et Me $\underbrace{5}_{2}$ $\underbrace{5}_{2}$ $\underbrace{5}_{2}$ $\underbrace{5}_{2}$	7 21 22 23 24 25	83 65 67 80 69 70	96:4 96:4 ^b 96:4 ^c 95:5 ^d 95:5 ^b 95:5 ^d
Me OBn	26	73	95:5 ^d

^{*a*} In all cases, the vicinal ¹H NMR vinylic coupling constant was *ca.* 10 Hz. ^{*b*} Ratio by ¹H NMR and from Mosher's derivatives; configuration by analogy. ^{*c*} Ratio by ¹H NMR; configuration by analogy. ^{*d*} Ratio by ¹H NMR and/or Mosher's; configuration assigned using *O*-acetylmandelates. The ¹⁹F spectra of the Mosher's derivatives of the product **21** from 4-methoxybenzaldehyde were also consistent with the 1,5anti-configuration. Configurations were assigned to the major products **23**, **25** and **26** from 2-methylpropanal and (R)- and (S)-2-benzyloxypropanal by comparison of the ¹H NMR spectra of their O-acetylmandelates, and the ¹⁹F spectra of their Mosher's derivatives, which also indicated the 1,5-stereoselectivity.

The minor product in each case was assigned the (Z)-1,5-synconfiguration by analogy with the formation of the (Z)-1,5-synisomer **8** in the reaction with benzaldehyde.

The use of other Lewis acids to promote reactions of allylstannane **6** with aldehydes was briefly investigated. With dibutyltin dichloride, titanium(IV) chloride, aluminium(III) chloride and boron trifluoride diethyletherate, complex mixtures of products were obtained. A modest, 40%, yield of the 1,5-*anti*- and 1,5*syn*-products **7** and **8**, ratio 95:5, was obtained using butyltin trichloride. However, with tin(IV) bromide, improved 1,5-(Z)*anti*: 1,5-(Z)-*syn* stereoselectivities of *ca*. 99:1 were obtained for reactions of the allylstannane **6** with the arylaldehydes, and improved stereoselectivities for reactions with 2-methylpropanal and propanal. No (*E*)-isomers of the products were detected in the product mixtures from these reactions.

The regio- and stereo-selectivities of these tin(IV) halide promoted reactions of stannane **6** with aldehydes are consistent with a kinetically controlled stereoselective intramolecular transmetallation, see Fig. 2. This generates the allyltin trihalide **28** in which the vinyl and methyl substituents are *trans*-disposed, *i.e.* pseudoequatorial, with respect to the 5-membered ring formed by co-ordination of the oxygen of the benzyloxy group with the electron deficient tin. The allyltin trihalide can then react with an aldehyde *via* the chair-like six-membered transition state, **29** in which the substituent next to the tin adopts the axial position to give the 1,5-*anti*-(Z)-products after an aqueous work-up.

To check the compatibility of this chemistry with a 2-alkyl substituent in the allylstannane, an (E,Z)-mixture of the 5-*tert*-butyldimethylsilyl-2,4-dimethylpent-2-enylstannane **33** was prepared, see Scheme 4. Thus the alcohol **30** (e.e. *ca.* 50%¹¹) was converted into the xanthate **31**, which on heating underwent a

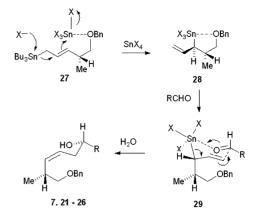


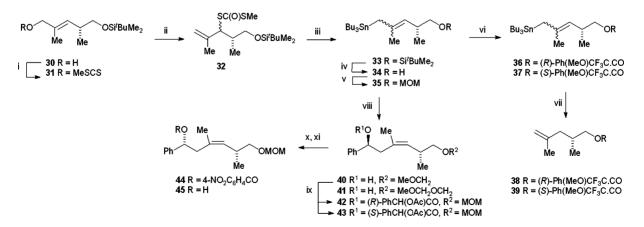
Fig. 2 Mechanism proposed to account for the stereoselectivity of tin(IV) halide promoted reactions of allylstannane **6** and aldehydes.

3,3-sigmatropic rearrangement to give the dithiocarbonate **32** as a 60:40 mixture of epimers. On reaction with tributyltin hydride under free radical conditions, these were converted into the stannane **33**. Following desilylation, *O*-alkylation of the resulting hydroxystannane **34** then gave the 5-(methoxymethoxy)pent-2-enylstannane **35**.

To check its optical purity, hydroxystannane **34** was converted into the Mosher's derivatives **36** and **37**, but the ¹⁹F NMR spectra of these were complicated by the presence of geometrical isomers. However, protonolysis gave the pent-4-enyl esters **38** and **39**, which by ¹⁹F NMR were estimated to have e.e.s of *ca*. 50%.

The tin(IV) chloride promoted reaction of the alk-2enylstannane **35** with benzaldehyde proceeded with excellent stereoselectivity to give the 1,5-*anti*-(Z)-hex-3-en-1-ol **40** (69%). Just one minor side-product was isolated and was identified as the 6-(methoxymethoxy)hex-3-enol **41** (6%) attributed to traces of (methoxymethoxy)methyl chloride in the MOM-chloride used to prepare the stannane **35**.

The (Z)-configuration of the major product 40 was confirmed by NOE, a significant enhancement of 4-H being observed on irradiation of the 3-CH₃. The configuration at C(1) was established



Scheme 4 Synthesis of 2,4-dimethylpent-2-enylstannanes and their reactions with benzaldehyde. Reagents and conditions: i, NaH, CS₂, MeI (94%); ii, heat in toluene, 18 h (75%; a 60:40 mixture of epimers); iii, Bu₃SnH, AIBN, benzene 3 h; iv, TBAF (88% from the dithiocarbonate **32**; a 55:45 mixture of geometric isomers); v, MeOCH₂Cl, 'Pr₂NEt (79%); vi, (*S*)- or(*R*)-Mosher's acid chloride, py (**36**, 38%; **37**, 66%); vii, HBr, EtOH (87–93%); viii, SnCl₄, -78 °C, 10 min then PhCHO, -78 °C, 1 h (**40**, 69%, **41**, 6%); ix, (*R*)- or (*S*)-*O*-acetylmandelic acid, DCC, DMAP (**42** and **43**, both *ca*. 100%); x, 4-NO₂C₆H₄CO₂H, Ph₃P, DEAD (33%); xi, NaOH, MeOH (72%).

 Table 2
 1,5-Stereocontrol in tin(IV) halide mediated reactions between aldehydes and allylstannanes 34 and 35

$R^{1}CHO \xrightarrow{34- \text{ or } 35-SnX_{4}}_{-78 \text{ °C}} R^{1} \xrightarrow{OH \text{ Me}}_{H^{2}} OR^{2}$						
R ¹	Stannane	$SnX_{4}(X)$	1,5-anti-producta	\mathbb{R}^2	Yield (%)	1,5-anti: 1,5-syn ^a
Ph	35	Cl	40	МОМ	69	≥98:2
4-MeOC ₆ H ₄	"	"	46	"	49	≥98:2
ⁱ Pr	"	"	47	"	57	≥98:2
MeCH=CH	"	"	48	"	68	≥98:2
Ph	34	Br	49	Н	92	≥98:2
МеСН=СН	"	"	50	"	89	≥98:2
^{<i>a</i>} Ratio by ¹ H NMR.						

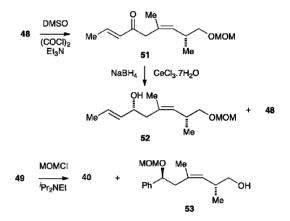
by comparison of the ${}^{1}H$ NMR chemical shifts of the *O*-acetylmandelates **42** and **43**.

To establish the stereoselectivity of this reaction, the 1,5-*anti*product **40** was converted into its 1,5-*syn*-diastereoisomer **45** via a Mitsunobu reaction¹² followed by saponification of the inverted 4-nitrobenzoate **44**, see Scheme 4. Epimers **40** and **45** could be distinguished by NMR, for example, the doublet due to the 5-CH₃ group was seen at $\delta 0.87$ in the ¹H NMR spectrum of the 1,5-*anti*product **40** and at $\delta 1.03$ in the ¹H NMR spectrum of the 1,5-*syn*epimer **45**. The 1,5-*syn*-alkenol **45** was not detected in the product mixture from the reaction of allylstannane **35** with benzaldehyde, and so the 1,5-*anti*: 1,5-*syn* selectivity was estimated to be $\geq 98: 2$.

The tin(IV) chloride promoted reactions of the allylstannane 35 with methoxybenzaldehyde, 2-methylpropanal and (E)-but-2-enal were also highly stereoselective in favour of the (Z)-1,5anti-alkenols 46-48, see Table 2. Structures were assigned to these products by analogy with the stereoselective formation of the (Z)-1,5-alkenol 40 in the reaction with benzaldehyde. (Z)-1,5-syn-Epimers were not detected from these reactions but to check that 1,5-anti- and 1,5-syn-epimers could be distinguished, the (Z)-1,5-anti-product 48 from the reaction with (E)-but-2enal was oxidised, and the resulting ketone 51 reduced to give a mixture of the epimeric alkenols 48 and 52. These were clearly different by ¹H and ¹³C NMR and the (Z)-1,5-syn-epimer 52 was not detected in the product mixture from the tin(IV) chloride promoted reaction of stannane 35 and (E)-but-2-enal. The only side-products isolated from these reactions were the minor 6-(methoxymethoxy)alk-3-en-1-ols due to the impurity in the MOM-protected stannane 35.

The free hydroxystannane **34** was also found to react, after transmetallation with tin(IV) bromide, with benzaldehyde and (*E*)-but-2-enal to give the 1,5-*anti*-(*Z*)-alkenols **49** and **50** with excellent yields, 92 and 89% and diastereoselectivities, >99:1. The (*Z*)-1,5-*anti*-configuration assigned to the product **49** from the reaction of stannane **34** with benzaldehyde was confirmed by MOM-protection, which gave the known 1,5-*anti*-alkenol **40** together with a small amount of its regioisomer **53**. The structure of the product **50** was assigned by analogy.

5-Alkoxy-4-methylpent-2-enyl(tributyl)stannanes, cf. generic structure I, appear to undergo highly stereoselective reactions with aldehydes on transmetallation with tin(IV) halides to give 1,5-*anti*-(Z)-alk-3-enols. This stereoselectivity, which can be promoted by either tin(IV) chloride or tin(IV) bromide, is compatible with an

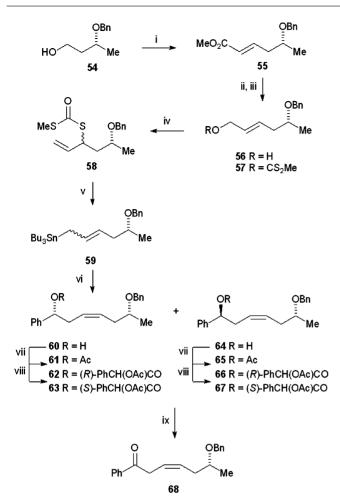


additional 2-methyl substituent, with different alkoxy substituents, and even with an unprotected 5-hydroxyl group in the stannane. The next phase of this work was to study analogous reactions of 5-substituted hexenylstannanes, cf. structure **II**, for 1,6-remote stereocontrol.

1,6-Stereocontrol using 5-substituted alkenylstannnanes II

The 5-benzyoxyhex-2-enylstannane **59** was synthesized to evaluate the ability of a 5-benzyloxy substituent to deliver 1,6-stereocontrol in reactions with aldehydes, see Scheme 5. (*R*)-3-Benzyloxybutanol **54** was converted into the 5-benzyloxyhex-2-enol **56** via oxidation,¹³ a Wittig reaction to give the (*E*)-ester **55** together with 5% of its (*Z*)-isomer, and reduction. Heating the corresponding xanthate **57** gave the dithiocarbonate **58** which was converted into the stannane **59**, as a 2 : 1 mixture of (*E*)- and (*Z*)-isomers, by treatment with tributyltin hydride under free-radical conditions.

However, tin(IV) chloride and bromide promoted reactions of stannane **59** with benzaldehyde gave mixtures of the 1,6-*syn*- and 1,6-*anti*-products **60** and **64**. These products were shown to be epimers since oxidation of the mixture gave a single ketone **68**. Their configurations at C(1) were established by comparison of the ¹H NMR spectra of their (*R*)-and (*S*)-*O*-acetylmandelates **62/63** and **66/67**. The alkene geometry of the 1,6-*syn*-product **60** was established by the 3,4-vinylic coupling constant of *ca*. 10–11 Hz observed for the acetate **61** and the *O*-acetylmandelates **62** and **63**. The ¹H NMR spectrum of the (*S*)-*O*-acetylmandelates **64** and **65** and **66**.



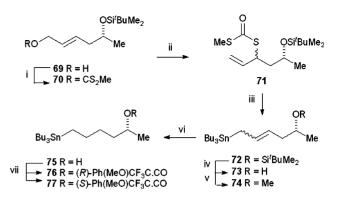
Scheme 5 Chemistry of the 5-benzyloxyhex-2-enylstannane **59**. Reagents and conditions: i, DMSO, $(COCI)_2$, DCM, $-50 \,^{\circ}$ C, 15 min, Et₃N, $-50 \,^{\circ}$ C, 5 min, r.t., 20 min, then MeO₂C. CHPPh₃, DCM, r.t., 15 h [70% plus 5% of its (*Z*)-isomer]; ii, DIBAL-H, hexane, DCM, $-78 \,^{\circ}$ C, 3 h (97%); iii, NaH, benzene, r.t., 1 h, CS₂, r.t., 3 h, MeI, r.t., 15 h (88%); iv, toluene, heat, 15 h (*ca.* 100%, 1:1); v, Bu₃SnH, AIBN (trace), benzene, heat, 3 h [67%; (*E*): (*Z*) = 2:1]; vi, SnCl₄ or SnBr₄, $-78 \,^{\circ}$ C, 2–20 min, PhCHO, $-78 \,^{\circ}$ C, 1 h (44–76%); vii, Ac₂O, DMAP (cat), Et₃N, DCM, r.t., 15 h (**61**, 87%; **65**, 88%); viii, (*R*)- or (*S*)-*O*-acetylmandelic acid, DCC, DMAP (cat), DCM, r.t., 15 h (**62**, 82%; **63**, 77%; **66**, 80%; **67**, 87%); ix, DMSO, (COCI)₂, DCM, $-50 \,^{\circ}$ C, 15 min, 'Pr₃NEt, $-50 \,^{\circ}$ C, 5 min (49%).

proportions of the two products varied, typically with only modest stereoselectivity in favour of the 1,6-*syn*-epimer **60**.

The *tert*-butyldimethylsilyloxyhex-2-enylstannane **72** was prepared to see whether stannanes analogous to the 5benzyloxyhexenylstannane **59**, but with different *O*-substituents, reacted with aldehydes with better remote stereocontrol, see Scheme 6.

Xanthate **70**, available from the alcohol **69**, rearranged on heating to give the dithiocarbonate **71** as a mixture of epimers. Reaction with tributyltin hydride then gave the stannane **72** as a 2:1 mixture of (*E*)- and (*Z*)-isomers. Desilylation was achieved using tetrabutylammonium fluoride to give the 5-hydroxyhex-2-enylstannane **73**, which was methylated to give the 5-methoxyhex-2-enylstannane **74**.

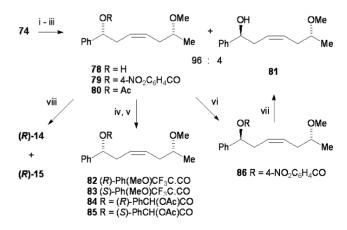
The optical purity of stannane **73** was checked by reduction using di-imide to give the 5-hydroxyhexylstannane **75**. This was



Scheme 6 Preparation of 5-hydroxyhex-2-enylstannane 73. Reagents and conditions: i, NaH, benzene, 40 °C, 2 h, CS₂, r.t., 3 h, MeI, r.t., 15 h (92%); ii, toluene, heat, 15 h (*ca.* 100%, 1 : 1); iii, Bu₃SnH, AIBN (trace), benzene, heat, 3.5 h [85%, (*E*) : (*Z*) = 2 : 1]; iv, TBAF, THF, r.t., 15 h (88%); v, NaH, THF, 35 °C, 2 h, MeI, r.t., 15 h (85%); vi, TsNHNH₂, NaOAc, EtOH, heat, 4 h (83%); vii, (*S*)- or (*R*)-Mosher's acid chloride, py., r.t. 15 h (76, *ca.* 100%; 77, 70%).

converted into its Mosher's derivatives **76** and **77**. A comparison of the ¹⁹F NMR spectra of these confirmed that the e.e. of stannane **73** was >98%, see Scheme 6.

Transmetallation of the 5-methoxyhex-2-enylstannane **74** using tin(IV) bromide generated an allyltin tribromide, which reacted with benzaldehyde to give a mixture of the inseparable (Z)-1,6-*syn*-and (Z)-1,6-*anti*-6-methoxyhept-3-enols **78** and **81**, ratio 91 : 9, see Scheme 7. The use of tin(IV) chloride resulted in significantly lower stereoselectivity, **78** : **81** = 70 : 30. The major product was converted into the minor *via* a Mitsunobu reaction with inversion using 4-nitrobenzoic acid followed by saponification of the resulting ester **86**. The two alcohols were distinctly different by NMR and the product ratio from the allylstannane reaction could be measured by integration of peaks in its ¹³C NMR spectrum. Interestingly the 4-nitrobenzoate **79** prepared by esterification of the major alcohol **78** and its epimer **86** from the Mitsunobu reaction, could not be

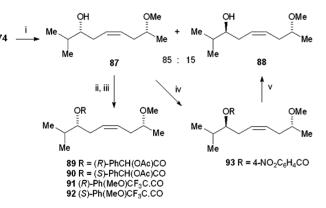


Scheme 7 1,6-Stereocontrol in the reaction of stannane 74 and benzaldehyde. Reagents and conditions; i, SnBr₄, DCM, -78 °C, 10 min, PhCHO, -78 °C, 1 h (78%; 78 : 81 = 91 : 9); ii, O₂NC₆H₄COCl, DMAP, Et₃N, DCM, r.t., 72 h (79, 88%); iii, Ac₂O, DMAP, Et₃N, DCM, r.t., 15 h (92%); iv, (*S*)- or (*R*)-Mosher's acid chloride, py (82, 97%; 83, 78%); v, (*R*)- or (*S*)-*O*-acetylmandelic acid, DCC, DMAP, DCM, r.t. 15 h (84, 80%; 85, 87%); vi, 78, 4-O₂NC₆H₄CO₂H, Ph₃P, DEAD, tol, -35 °C to r.t., 2 h (80%); vii, NaOH, MeOH, r.t., 2 h (84%); viii, O₃, CHCl₃, -78 °C, 15 min, DMS, r.t., then NaBH₄, MeOH, DCM, r.t., 15 h [(*R*)-14, 58%; (*R*)-15, 15%].

distinguished by NMR. The high optical purity of the stannane, and hence of the products in this series, meant that the ratio of the epimers **78** and **81** could also be estimated using the ¹⁹F spectra of the Mosher's derivatives **82** and **83**.

The 1,6-syn-configuration assigned to the major product **78** was established by ozonolysis of its acetate **80**, which gave (R)-3-acetoxy-3-phenylpropanol (R)-**14** together with the product (R)-**15** of acetate migration.^{1a} Comparison of the ¹H NMR spectra of the (R)-and (S)-O-acetylmandelates **84** and **85** supported this assignment. The vinylic coupling constant of 11 Hz in the ¹H NMR spectra of the Mosher's derivatives **82** and **83** and the O-acetylmandelates **84** and **85** confirmed the (Z)-geometry of the alkene.

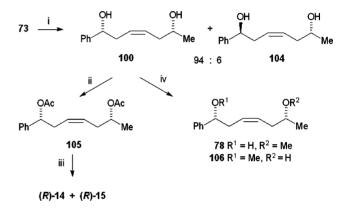
The tin(IV) bromide promoted reaction of the 5-methoxyhexenylstannane **74** with 2-methylpropanal was also stereoselective, the (*Z*)-3,8-*syn*-8-methoxynon-5-en-3-ol **87** being the major product, see Scheme 8. To check that the major and minor products were epimers, the major alkenol **87** was converted into the minor **88** using a Mitsunobu reaction followed by saponification of the resulting ester **93**. Minor differences were seen in the ¹H and ¹³C NMR spectra of these epimers. The C(3) configuration of the major product **87** was established by comparison of the ¹H NMR data of *O*-acetylmandelates **89** and **90**. The reaction diastereoselectivity was estimated by ¹³C NMR spectra of the product mixture and from the ¹⁹F NMR spectra of the Mosher's derivatives **91** and **92**. The 10.5 Hz vinylic coupling constant in the ¹H NMR spectrum of the (*S*)-Mosher's derivative **92** confirmed the (*Z*)-geometry.



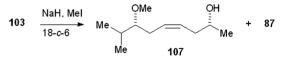
Scheme 8 1,6-Stereocontrol in the reaction of stannane 74 and 2-methylpropanal. Reagents and conditions: i, SnBr₄, -78 °C, 10 min, 'PrCHO, -78 °C, 1 h (72%; 87:88 = 85:15); ii, (*R*)- or (*S*)-*O*-acetylmandelic acid, DCC, DMAP, DCM, r.t. 15 h (89, 81%; 90, 51%); iii, (*S*)- or (*R*)-Mosher's acid chloride, py (91, 89%; 92, 98%); iv, 4-O₂NC₆H₄CO₂H, Ph₃P, DEAD, tol, -35 °C to r.t., 18 h (45%); v, NaOH, MeOH, r.t., 2.5 h (92%).

Tin(v) bromide promoted reactions of the 5-methoxyhexenylstannane 74 with other aldehydes were investigated, see Table 3. In all cases the (Z)-1,6-*syn*-isomers were the major products with useful stereoselectivity for achiral aldehydes. Matching and mismatching was observed for the chiral aldehydes, (R)- and (S)-2-benzyloxypropanal, with Felkin–Anh rather than chelation control being preferred. The structures of the products from these reactions were assigned by analogy with the products 78/81 and 87/88 from the reactions between alkenylstannane 74 and benzaldehyde or 2-methylpropanal. Mosher's derivatives and Oacetylmandelates were used to establish the reaction selectivities and the configurations of the hydroxyl bearing stereogenic centres in the products.

The 5-hydroxyhex-2-enylstannane 73 was also found to undergo stereoselective reactions with aldehydes in favour of (Z)-1,6syn-products after transmetallation with tin(IV) halides. With benzaldehyde, a 90:10 mixture (59%) of the 1,6-syn- and 1,6anti-epimers 100 and 104 was obtained using tin(IV) chloride, with a better yield (72%) and stereoselectivity, 94:6, being obtained using tin(IV) bromide, see Scheme 9. The ratios of the products 100 and 104 were determined by integration of the benzylic protons observed at $\delta 4.68$ for the 1.6-syn-epimer 100, and at $\delta 4.77$ for the minor 1,6-anti-isomer 104. The configuration of the major product 100 at the benzylic position, C(1), was established by ozonolysis of its bis-acetate 105 which, after reduction, gave the primary alcohol (R)-14 (61%), together with the secondary alcohol (R)-15 (13%).^{1a} O-Methylation also gave a mixture of the regioisomeric monomethyl ethers 78 and 106, the former being identical to the major product from the reaction of the 5-methoxyhex-2enylstannane 74 with benzaldehyde. The minor product 104 was identified by analogy with earlier work.



Scheme 9 1,6-Stereocontrol in the reaction of the 5-hydroxyhexenylstannane 73 and benzaldehyde. Reagents and conditions: i, SnX_4 , -78 °C, 10 min, PhCHO, -78 °C, 1 h (with $SnCl_4$; 59%, 100:104 = 90:10; with $SnBr_4$, 72%, 100:104 = 94:6); ii, Ac₂O, Et₃N, DMAP, DCM, r.t., 15 h (91%); iii, O₃, CHCl₃, -78 °C, 15 min, DMS, r.t., then NaBH₄, MeOH, DCM, r.t., 15 min [(*R*)-14, 61%, (*R*)-15, 13%]; iv, NaH, 18-crown-6, THF, r.t., 15 min, diol 100, r.t., 45 min, MeI, r.t., 15 h (78, 11%, 106, 45%).



Tin(IV) bromide promoted reactions of the 5-hydroxyhex-2enylstannane **73** and other aldehydes were investigated, see Table 3. In all cases useful levels of (*Z*)-1,6-*syn*-stereoselectivity were observed. The product ratios in these cases were determined by ¹H and ¹³C NMR. The structure of the major product **103** from 2-methylpropanal was confirmed by *O*-methylation, which gave the regioisomeric monomethyl ethers **107** and **87**, the latter being identical to the major product from the reaction of 2methylpropanal with the 5-methoxyhexenylstannane **74**.

The (Z)-1,6-*syn*-stereoselectivity observed for the tin(IV) bromide mediated reactions of the 5-hydroxy- and 5-methoxy-hex-2-enylstannanes **73** and **74** with aldehydes is consistent with participation of the allyltin tribromides **108** in which the vinyl
 Table 3
 1,6-Stereocontrol in tin(IV) bromide mediated reactions between aldehydes and allylstannanes 73 and 74

$\overline{\mathbf{R}^1}$	Stannane	-78 °C R' 1,6- <i>syn</i> -product"	R ²	Yield (%)	16 aug 16 aug
K	Stannane	1,0- <i>syn</i> -ploduct	K	1 leid (76)	1,6-syn:1,6-ant
Ph	74	78	Me	78	96:4
4-MeOC ₆ H ₄	"	94	"	65	93:7 ^b
$4-ClC_6H_4$	"	95	"	61	92:8 ^b
Me	"	96	"	69	91:9 ^c
Et	"	97	"	67	84:16 ^b
Me ₂ CH	"	87	"	72	85:15
Me 55 II OBn	"	98	n	88	75:25 ^d
Me کر OBn	11	99	n	73	90 : 10 ⁴
Ph	73	100	Н	72	94:6
$4-O_2NC_6H_4$	"	101	"	71	94:6 ^c
Me	"	102	"	60	96:5 ^c
Me ₂ CH	"	103	"	59	93:7

^{*a*} In all cases the (*Z*)-configuration was confirmed by vicinal vinylic ¹H couplings of *ca.* 10 Hz observed either for the products or for a derivative of the products. ^{*b*} Ratio from ¹H NMR, ¹³C NMR and Mosher's ¹⁹F spectra; relative ¹⁹F chemical shifts for the Mosher's derivatives were consistent with the 1,6-*syn*-configuration. ^{*c*} Ratio by ¹H and ¹³C NMR; structure by analogy. ^{*d*} Ratio from ¹H NMR, ¹³C NMR and Mosher's ¹⁹F spectra; ¹H data of *O*-acetylmandelates and the relative ¹⁹F chemical shifts for the Mosher's derivatives established the 1,6-*syn*-configuration.

and methyl groups are *cis*-disposed, *i.e.* pseudo-equatorial, about the five-membered ring formed by co-odination of the methoxy or hydroxy group with the electron deficient tin, see Fig. 3. Reaction of these allyltin tribromides with an aldehyde *via* the six-membered, chair-like transition state **109**, in which the group next to tin adopts the axial position, would then account for the overall (Z)-1,6-*syn*-stereocontrol. The use of tin(IV) bromide rather than tin(IV) chloride led to improved 1,6-*syn*-stereoselectivity in these reactions. Better stereoselectivity was also observed for the 5-hydroxy- and 5-methoxy-alkenylstannanes **73** and **74**, in comparison with that found for the 5-benzyloxyhexenylstannane **59**.

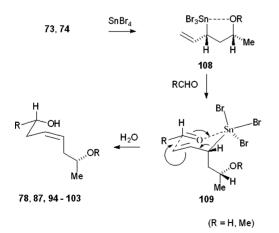
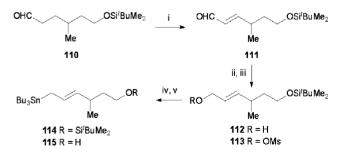


Fig. 3 Mechanism proposed to account for the 1,6-*syn*-stereoselectivity in reactions of allylstannanes **73** and **74** with aldehydes.

Having studied reactions of 5-alkoxy(hydroxyl)alk-2-enylstannanes for 1,5- and 1,6-stereocontrol, it was decided that 6hydroxyalkenylstannanes should be studied, see generic structures **III**, **IV** and **V**, for 1,5-, 1,6- and 1,7-stereocontrol. Hydroxy substituted alkenylstannanes were selected for study, since the hydroxyl group had proved to be an effective co-ordinating group for 1,6-stereocontrol in 5-substituted systems.

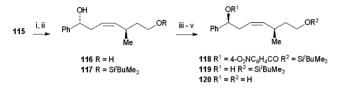
1,5-Stereocontrol using 4-methyl-6-hydroxyhex-2-enylstannnane III (R = H)

The racemic aldehyde **110** was available in two steps from citronellol¹⁴ and was dehydrogenated to give the unsaturated aldehyde **111**, see Scheme 10.¹⁵ Reduction gave the alcohol **112** which was converted to the stannane **114** by displacement of its mesylate **113**.¹⁶ Desilylation then gave the racemic 6-hydroxy-4-methylhexenylstannane **115**.



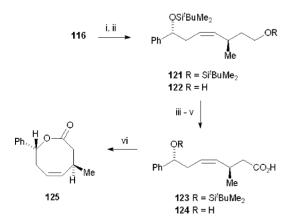
Scheme 10 Synthesis of the racemic 6-hydroxy-4-methylhexenylstannane 115. Reagents and conditions; i, $(EtO)_2P(O)OCH_2CH=CH_2$, $Pd(Ac)_2$, NaHCO₃, THF, heat, 72 h (95%); ii, DIBAL-H, THF, -78 °C to -45 °C, 3 h (84%); iii, MsCl, DCM, Et₃N, -7 °C, 10 min; iv, Bu₃SnLi, THF, -78 °C, 2 h, r.t., 15 h (53% from 112); v, TBAF, THF, r.t., 15 h (72%).

The reaction between the racemic 6-hydroxy-4-methylhex-2enylstannane **115** and benzaldehyde mediated by tin(IV) bromide gave the (*Z*)-1,5-*anti*-hept-3-ene-1,7-diol **116** (64%) with excellent stereoselectivity. With tin(IV) chloride, a lower yield, 40%, and slightly reduced stereoselectivity, 95:5, was observed. The *syn*-epimer **120** was prepared from diol **116** by selective silylation of the primary hydroxyl group and esterification of the secondary hydroxyl group of the resulting monosilyl ether **117** using a Mitsunobu reaction to give the inverted 4-nitrobenzoate **118**. Saponification and desilylation gave the inverted 1,5-*syn*-diol **120**, see Scheme 11. Epimers **116** and **120** could be distinguished by ¹H and ¹³C NMR, and the 10.5 Hz vinylic coupling observed for the 1,5-*syn*-epimer **120** were present in the product mixture from the allylstannane reaction.



Scheme 11 1,5-Stereocontrol in the reaction of 6-hydroxyhexenyl-stannane 115 and benzaldehyde. Reagents and conditions: i, SnBr₄, -78 °C, 5 min, PhCHO, -78 °C, 1 h (64%, 116 : 120 \geq 98 : 2); ii, TBSCl, imid., DCM, r.t., 30 min (67%); iii, 4-O₂NC₆H₄CO₂H, PPh₃, DIAD, 0 °C, 10 min, r.t., 30 min (63%); iv, NaOH, MeOH, r.t., 2 h; v, TBAF, THF, r.t., 15 h (89%).

As the compounds in this series were racemic, ozonolysis or *O*-acetylmandelates couldn't be used to establish their relative configuration. The 1,5-*anti*-diol **116** was therefore taken through to lactone **125** to see whether transannular nOe measurements would confirm the assigned stereochemistry, see Scheme 12. The diol was protected as its bis-*tert*-butyldimethylsilyl ether **121** which was selectively monodesilylated using HF-pyridine to alcohol **122**. Oxidation *via* the corresponding aldehyde gave the carboxylic acid **123** and, after desilylation to give the hydroxyacid **124**, lactonisation¹⁷ gave the lactone **125**. This confirmed the (*Z*)-geometry assigned to the double-bond, but nOe studies did not establish the relative configuration of the stereogenic centres. The 1,5-*anti*-configuration was therefore assigned to products in this



Scheme 12 Synthesis of lactone 125. Reagents and conditions: i, TBSCl, imid., DMAP, r.t., 18 h (82%); ii, HF-py., THF, 18 h (72%); iii, Dess—Martin periodinane, NaHCO₃, DCM, r.t., 1 h; iv, NaClO₂, NaH₂PO₄, 2-methylbut-2-ene, 'BuOH, H₂O, r.t., 2 h (76% from 122); v, HCl, MeOH, r.t., 2 h (80%); vi, 2,4,6-Cl₃C₆H₂COCl, Et₃N, THF, r.t., 2 h, then DMAP, toluene, reflux, 4.5 h (60%).

series on the basis of the supposed mechanism and by analogy to the other series, *vide infra*.

The 6-hydroxy-4-methylhexenylstannane **115** was then reacted with a range of aldehydes using tin(IV) bromide to effect transmetallation, and the results obtained are shown in Table 4. In all cases high levels of stereoselectivity were observed by NMR, and in some cases, by glc. The 1,5-*anti*-configuration of the products was assigned by analogy with earlier work and the (*Z*)-geometry of the double-bond was confirmed by the vinylic ¹H coupling constants which were in the range of 10–11 Hz.

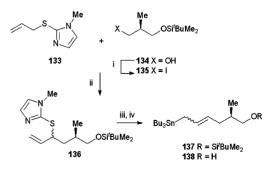
 $Table \ 4 \quad 1,5 \ Stereo control \ in \ tin({\rm IV}) \ bromide \ mediated \ reactions \ between \ aldehydes \ and \ allyl \ stannane \ 115$

RCHO	115-SnBr₄ R	OH	OH
R	1,5-anti-product ^a	Yield (%)	1,5-anti: 1,5-syn
Ph	116	64	99:1
4-MeOC ₆ H ₄	126	61	99 : 1 ^b
$4-ClC_6H_4$	127	74	98:2 ^b
$3-ClC_6H_4$	128	64	99:1 ^b
$4-O_2NC_6H_4$	129	62	99:1 ^b
Me ₂ CH	130	63	91:9 ^c
ⁿ Pr ²	131	67	96:4 ^c
МеСН=СН	132	76	97 : 3 ^c

^{*a*} In all cases the vicinal vinylic coupling constant was *ca.* 10 Hz. ^{*b*} Ratio by ¹H NMR. ^{*c*} Ratio by glc.

1,6-Stereocontrol using 5-methyl-6-hydroxyhex-2-enylstannnane IV (R = H)

A synthesis of the 6-hydroxy-5-methylhex-2-enylstannane **138** is outlined in Scheme 13. Following lithiation, the allyl sulfide **133**¹⁸ was alkylated using iodide **135**, which is available from alcohol **134**, to give a mixture of the epimeric alkylated sulfides **136**. Reaction of this mixture with tributyltin hydride under free-radical conditions gave the 6-silyloxyhexenylstannane **137** which was deprotected to give the 6-hydroxy-5-methylhexenylstannane **138**, (E):(Z) = 2:1, see Scheme 13. In this two step synthesis of alk-2-enylstannanes from alkyl halides, the lithiated allylsulfide **139** is being used as the synthetic equivalent of 3-(tributylstannyl)propenyllithium **140**,¹⁹ see Fig. 4.



Scheme 13 Synthesis of the 6-hydroxy-5-methylhexenylstannane 138. Reagents and conditions: i, (a) MsCl, Et₃N, DCM, -20 °C to r.t., 1 h (*ca.* 100%) (b) NaI, acetone, heat, 18 h (70%); ii, 133, *n*BuLi, THF, HMDA, -78 °C, 30 min, 135, -78 °C, 2.5 h (42%); iii, Bu₃SnH, AIBN, benzene, heat, 18 h [79%, (*E*): (*Z*) = 2:1]; iv, TBAF, THF, r.t., 18 h (79%).

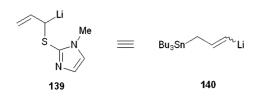
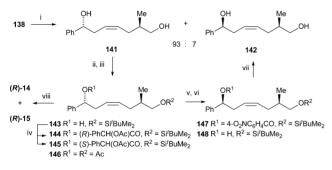


Fig. 4 The synthetic equivalence of the lithiated allylsulfide 139 and 3-(tributylstannylpropenyl)lithium 140.

The tin(IV) bromide mediated reaction of the 5methylhexenylstannane 138 with benzaldehyde was stereoselective in favour of the (Z)-1,6-anti-heptenediol 141, 1,6anti: 1,6-syn = 93:7, see Scheme 14. With tin(IV) chloride, the stereoselectivity dropped to 84:16 and the yield to 62%. To establish the stereoselectivity of this reaction, the (Z)-1,5-antiproduct 141 was converted into its 1,6-svn-epimer 142 by selective protection as its monosilyl ether 143, inversion of configuration at C(1) using a Mitsunobu reaction, hydrolysis of the ester 147 and desilylation of the resulting (Z)-1,6-syn-silyloxyalcohol 148. The 1,6-anti- and 1,6-syn-diols 141 and 142 were distinguishable by ¹H NMR and their ratio was measured by integration of 1-H, which was observed at δ 4.65 for the *anti*-diol 141 and at $\delta 4.71$ for its *syn*-epimer **142**. The configuration of the major product 141 at C(1) was initially assigned on the basis of the ¹H NMR chemical shifts of the (R)- and (S)-O-acetylmandelates 144 and 145 prepared from the monosilylether 143, and was confirmed by ozonolysis of the diacetate 146 which gave the (R)-3-acetoxy-3-phenylpropanol (R)-14 together with the product (*R*)-15 of acetate migration.^{1a}

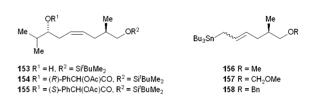


Scheme 14 1,6-Stereocontrol in the reaction of 6-hydroxyhexenylstannane 138 and benzaldehyde. Reagents and conditions: i, SnBr₄, DCM, -78 °C, 10 min, PhCHO, -78 °C, 1 h (84%; 141 : 142 = 93 : 7); ii, TBSCl, Et₃N, DMAP, DCM, r.t., 3 h (92%); iii, Ac₂O, Et₃N, DMAP, DCM, r.t., 18 h (90%); iv, (*R*)- or (*S*)-*O*-acetylmandelic acid, DCC, DMAP, DCM, r.t. 18 h (144, 90%; 145, 80%); v, 4-O₂NC₆H₄CO₂H, Ph₃P, DEAD, toluene, -35 °C, 90 min (69%); vi, NaOH, MeOH, r.t., 3 h (73%); vii, TBAF, THF, r.t., 18 h (88%); viii, O₃, DCM, -78 °C, 40 min, DMS, then DCM, NaBH₄, MeOH, r.t., 2 h [(*R*)-14, 34%; (*R*)-15, 16%].

Tin(v) bromide mediated reactions of the allylstannane **138** with several aldehydes were investigated, and the results are summarised in Table 5. In all cases, useful stereoselectivities in favour of the (*Z*)-1,6-*anti*-isomers were observed as estimated by ¹H NMR. The 1,6-*anti*-configuration of the major product from the reaction of the alkenylstannane **138** with 2-methylpropanal was established by comparison of the ¹H NMR spectra of the (*R*)- and (*S*)-*O*-acetylmandelates **154** and **155** prepared from the monosilyl ether **153**. The structures of the other products were assigned by analogy.

R ¹ CH0	0 − 138-SnBr₄ − -78 °C		Ме U ОН
\mathbf{R}^1	1,6-anti-product ^a	Yield (%)	1,6-anti: 1,6-syn
Ph	141	84	93:7
4-MeOC ₆ H ₄	149	58	91:9 ^b
$4-O_2NC_6H_4$	150	76	91:9 ^b
Et	151	56	93:7 ^c
Me ₂ CH	152	80	91: 9 ^d

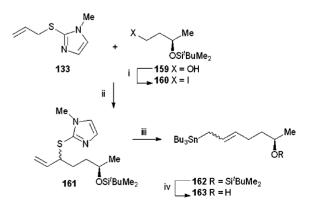
^{*a*} In all cases the vicinal vinylic coupling was 10 Hz. ^{*b*} Ratio by ¹H NMR. ^{*c*} Ratio by pulse delayed ¹³C NMR. ^{*d*} Ratio by pulse delayed ¹³C NMR; stereochemistry by comparison of the ¹H NMR spectra of the (*R*)- and (*S*)-*O*-acetylmandelates **153** and **154**.



The 6-hydroxyhexenylstannane **138** was also converted into its methyl, MOM and benzyl ethers **156–158**. However, tin(IV) bromide mediated reactions of these with benzaldehyde were less stere-oselective than those observed for the 6-hydroxyhexenylstannnane **138**, the 1,6-*anti*:1,6-*syn* ratios typically being of the order of 60:40, although (*Z*)-alkenols remained the dominant products.

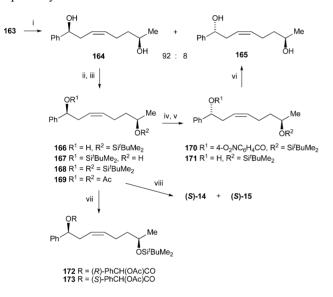
1,7-Stereocontrol using 6-hydroxyhept-2-enylstannnane V (R = H)

1,7-Remote stereocontrol is rare in open-chain systems and so the chemistry of 6-hydroxyhept-2-enylstannane **163**, in which there is potential for 1,7-stereocontrol determined by the stereogenic centre at C(6), was investigated. This stannane was prepared by alkylation of the allylsulfide **133** using the alkyl iodide **160** prepared from the alcohol **159**²⁰ see Scheme 15. Reaction of the alkylated sulfide **161** with tributyltin hydride under free-radical conditions gave the 6-hydroxyheptenylstannane **163** after desilylation of the silyl ether **162**.



Scheme 15 Synthesis of the 6-hydroxyhept-2-enylstannane 163. Reagents and conditions: i, Ph₃P, imid., ether, MeCN, I₂, 0 °C, r.t., 2.5 h (75%); ii, *n*BuLi, hexane, 133, -78 °C, 20 min, 160, -78 °C, 2.5 h (83%, a 1 : 1 mixture of epimers); iii, Bu₃SnH, AIBN (trace), benzene, heat, 1.5 h (92%); iv, TBAF, THF, r.t., 50 h [80%, (E):(Z) = 2:1].

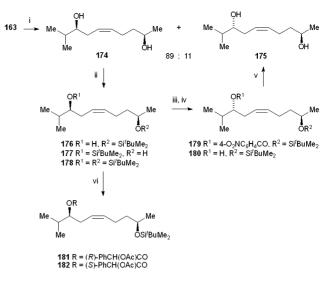
The tin(IV) bromide mediated reaction of the heptenylstannane 163 with benzaldehyde gave the (Z)-1,7-syn and (Z)-1,7-antiproducts 164 and 165 with useful stereoselectivity, 164:165 =92:8, together with a third product accounting for ca. 2%of the product mixture, which was not identified. To confirm that the minor product from the reaction of the stannane 163 with benzaldehyde was the (Z)-1,7-anti-isomer 165, an authentic sample was prepared, see Scheme 16. Silvlation of the 1,7-syn-diol 164 gave the regioisomeric monosilyl ethers 166 (63%) and 167 (5%), together with the bis-silvl ether 168 (15%). Inversion of the monosilyl ether 166 at C(1) using a Mitsunobu reaction provided the 1,7-anti-silyloxy ester 170 which was converted to the 1,7anti-diol 165 by saponification and deprotection of the resulting monosilyl ether 171. Diols 164 and 165 were distinguishable by ¹H and ¹³C NMR *e.g.* 1-H was observed at δ 4.72 and at δ 4.63 in the ¹H NMR spectra of the 1,7-syn and 1,7-anti-isomers 164 and 165, respectively.



Scheme 16 1,7-Stereocontrol in the reaction of 6-hydroxyhept-2-enylstannane 163 and benzaldehyde. Reagents and conditions: i, 163, SnBr₄, DCM, -78 °C, 10 min, add PhCHO, -78 °C, 1 h (72%, 164:165 = 92:8); ii, TBSCl, imid., DMF, r.t., 15 h (166, 63%; 167, 5%; 168, 15%); iii, Ac₂O, Et₃N, DMAP, DCM, r.t., 15 h (96%); iv, 4-O₂NC₆H₄CO₂H, Ph₃P, DEAD, toluene, -35 °C, r.t., 2 h (74%); v, NaOH, MeOH, r.t., 2 h (93%); vi, TBAF, THF, r.t., 72 h (96%); vii, (*R*)or (*S*)-*O*-acetylmandelic acid, DCC, DMAP, DCM, r.t. 18 h (172, 78%; 173, 83%); viii, O₃, DCM, -78 °C, 15 min, DMS, then DCM, NaBH₄, MeOH, r.t., 15 min [(*S*)-14, 27%; (*S*)-15, 13%].

The 1,7-syn-configuration was assigned to the major product **164** by comparison of the ¹H NMR spectra of the (*R*)- and (*S*)-O-acetylmandelates **172** and **173** prepared from the monosilyl ether **166**, and was confirmed by ozonolysis of the bis-acetate **169** which gave (*S*)-3-acetoxy-3-phenylpropan-1-ol (*S*)-**14** and the (*S*)-3-acetoxy-1-phenylpropan-1-ol (*S*)-**15**, formed by 1,3-acetate migration,^{1a} see Scheme 16. The (*Z*)-configuration of the doublebond was consistent with the 10.5 Hz vinylic ¹H coupling observed for the 1,7-syn-diol **164**.

The tin(iv) bromide promoted reaction of the 6-hydroxyhex-2-enylstannane **163** with 2-methylpropanal was also stereoselective in favour of the (*Z*)-1,7-*syn*-product **174**, (*Z*)-1,7-*syn*diol **174**: (*Z*)-1,7-*anti*-diol **175** = 89 : 11, see Scheme 17. A small



Scheme 17 1,7-Stereocontrol in the reaction of 6-hydroxyhept-2-enylstannane 163 and 2-methylpropanal. Reagents and conditions: i, 163, SnBr₄, DCM, -78 °C, 10 min, add 'PrCHO, -78 °C, 1 h (63%, 174:175 = 89:11); ii, TBSCl, imid., DMF, r.t., 15 h (176, 65%; 177, 1%; 178, 9%); iii, 4-O₂NC₆H₄CO₂H, Ph₃P, DEAD, toluene, -35 °C, r.t., 5 h (61%); iv, NaOH, MeOH, r.t., 3 h; v, TBAF, THF, r.t., 50 h (66% from 179); vi, (*R*)- or (*S*)-*O*-acetylmandelic acid, DCC, DMAP, DCM, r.t. 18 h (181, 95%; 182, 74%).

amount, ca. 2%, of a third product was also isolated but was not identified.

To check the structure of the minor product **175**, a sample was prepared by silylation of the major 1,7-*syn*-diol **174**. This gave a mixture of the regioisomeric monosilyl ethers **176** (65%) and **177** (1%), together with the bis-silyl ether **178** (9%). A Mitsunobu reaction of the major monosilyl ether **176** using 4-nitrobenzoic acid gave the inverted ester **179**, which was converted into the 1,7-*anti*-diol **175** by saponification and desilylation of the monosilyl ether **180**. The 1,7-*syn*- and 1,7-*anti*-diols **174** and **175** were difficult to distinguish by ¹H NMR but were distinguishable by ¹³C NMR. Pulse delayed ¹³C NMR was used to measure their ratio in the product mixture from the allylstannane reaction. The configuration of the hydroxyl bearing stereogenic carbon in the monosilyl ether **176** was established by comparison of the ¹H NMR spectra of the (*R*)- and (*S*)-*O*-acetylmandelates **181** and **182**.

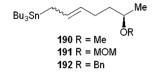
Tin(IV) bromide promoted reactions of the 6-hydroxyhept-2enylstannane **163** with several aromatic and aliphatic aldehydes were investigated, and the results obtained are shown in Table 6. In all cases useful levels of stereoselectivity in favour of the (*Z*)-1,7-*syn*-products were observed. The structures of the major products were consistent with spectroscopic data. The 1,7-*syn*diastereoselectivity and structures of the minor products were assigned by analogy with the reactions of stannane **163** with benzaldehyde and 2-methylpropanal. Product ratios were measured by ¹H NMR for the products from aromatic aldehydes and by ¹³C pulse delayed NMR for the products from aliphatic aldehydes.

The tin(IV) chloride mediated reaction of the 6hydroxyheptenylstannnane **163** with benzaldehyde gave the *syn*and (Z)-1,7-*anti*-products **164** and **165**, but the stereoselectivity was only 75:25 in favour of the 1,7-*syn*-diastereoisomer **164**. The hydroxyheptenylstannane **163** was also converted into the

RCHO	163 -SnBr₄ → -78 °C	R H	Me OH
R	1,7-syn-product ^a	Yield (%)	1,7-syn:1,7-anti
Ph 4-MeOC ₆ H ₄ 4-ClC ₆ H ₄ 2-naphth. Me ₂ CH Me Et Me ₂ CHCH ₂	164 183 184 185 174 186 187 188	72 47 71 65 63 36 61 58	92:8 89:11 ^b 92:8 ^b 93:7 ^b 89:11 90:10 ^c 91:9 ^c 85:15 ^c

^{*a*} Minor product assumed to be the (*Z*)-1,7–isomer by analogy with the reactions of benzaldehyde and 2-methylpropanal with stannane **163**; a third very minor product, *ca.* 1%, also detected but not identified. ^{*b*} Ratio by ¹H NMR. ^{*c*} Ratio by pulse-delayed ¹³C NMR.

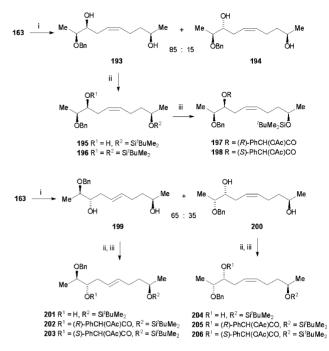
6-methoxy-, 6-(methoxymethoxy)- and 6-benzyloxystannanes **190–192**, but preliminary studies of tin(IV) halide promoted reactions of these with aldehydes gave mixtures of products, and so were not studied further.



Tin(iv) bromide promoted reactions of the 6-hydroxyheptenylstannane **163** with the chiral aldehydes (*S*)- and (*R*)-2benzyloxypropanal were also investigated, see Scheme 18. With the (*S*)-enantiomer, the (*Z*)-1,7-*syn*- and (*Z*)-1,7-*anti*-products **193** and **194** were obtained, the 1,7-*syn*: 1,7-*anti* ratio of 85:15 being measured by ¹³C pulse delayed NMR. The geometry of the doublebond of the major product **193** followed from its vicinal vinylic ¹H NMR coupling constant (10.5 Hz) and its 1,7-*syn*-configuration was established by comparison of the ¹H NMR spectra of the (*R*)- and (*S*)-*O*-acetylmandelates **197** and **198** prepared from the monosilyl ether **195**. The structure of the minor product was assigned by analogy with the identification of the (*Z*)-1,7-*anti*isomers **165**/175 as the minor products from the reaction of stannane **163** with benzaldehyde and 2-methylpropanal.

The tin(v) bromide promoted reaction of the (R)-2benzyloxypropanal and the (R)-6-hydroxyheptenylstannane **163** was less stereoselective and gave two separable products, ratio 65:35. Unexpectedly, the major product was found by spin decoupling to have a vinylic coupling constant of 15 Hz, indicative of an (E)-alkene. To establish their configuration at the newly introduced hydroxyl bearing carbon, following selective monosilylation, both products were converted into (R)- and (S)-O-acetylmandelates, see Scheme 18. By comparison of the ¹H NMR spectra of the O-acetylmandelates **202**/**203** and **205**/**206** derived from each silyl ether, and taking into account the double-bond geometries as indicated by ¹H NMR, the major product was identified as the (E)-1,7-*anti*-diol **199** and the minor as the (Z)-1,7-*anti*-diol **200**.

The reaction of the (R)-6-hydroxyheptenylstannane 163 with (R)-2-benzyloxpropanal was the only example found during the



Scheme 18 Reactions of (*S*)- and (*R*)-2-benzyloxypropanal with the 6-hydroxyhept-2-enylstannane 163. Reagents and conditions: i, SnBr₄, -78 °C, 10 min, (*S*)- or (*R*)-MeCH(OBn)CHO, -78 °C, 1 h (193, 194, 56%; 199, 200, 67%); ii, TBSCl, imid., DMF. r.t., 15 h (195, 61%; 196, 5%; 201, 55%; 204, 33%); iii, (*R*)- or (*S*)-*O*-acetylmandelic acid, DCC, DMAP, DCM, r.t. 18 h (197, 78%; 198, 75%; 202, 78%; 203, 75%; 205, 89%; 206, 88%).

course of this work where the major product from a tin(IV) halide promoted reaction of a heterofunctionalised alk-2-enylstannane had an (E)-double-bond. It is likely that the formation of the (E)-isomer **199** is indicative of an open-chain transition structure in the reaction of the intermediate allyltin tribromide with the aldehyde, but why this particular combination of reagents proceeds *via* an open-chain process rather than *via* the more usually observed 6-membered chair-like transition structure is not clear.

Apart from this exception, the stereoselectivities observed for all the tin(IV) bromide promoted reactions of the 6-hydroxyalk-2enylstannanes 115, 138 and 163 with aldehydes are consistent with kinetically controlled, stereoselective transmetallations generating allyltin tribromides which react with aldehydes via six-membered chair-like transition structures, see Fig. 5. It is proposed that the stereogenic centre in the alk-2-enylstannane controls the facial selectivity of the transmetallation to give the allyltin tribromides 207, 210 and 213, in which the methyl and ethenyl substituents are pseudoequatorial about the six-membered oxastannane ring formed by co-ordination of the hydroxyl group with the electron deficient tin. These allyltin tribromides can then react with aldehydes via the six-membered chair-like transition structures 208, 211 and 214 to give the (Z)-1,5-anti-, the (Z)-1,6-antiand the (Z)-1,7-syn-products 209, 212 and 215, respectively. The formation of the (Z)-double-bonds in these products is consistent with the known preference of a group next to tin to adopt the axial position in analogous chair-like transition structures. This preference together with the preference of the R group of the aldehyde to adopt the equatorial position, explains how the allyltin

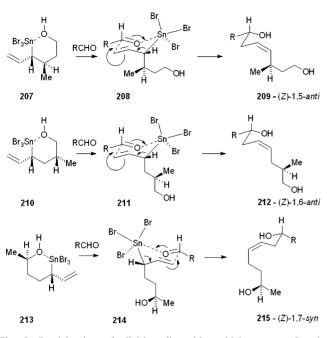


Fig. 5 Participation of allyltin tribromides which account for the stereoselectivity observed for the tin(IV) bromide promoted reactions of allylstannanes 116, 138 and 163 with aldehydes.

tribromides 207, 210 and 213, deliver the remote stereocontrol which is observed, *cf*. the generalised structures 209, 212 and 215

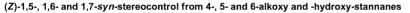
in Fig. 5 with the structures of the products obtained from alk-2enylstannanes **115**, **138** and **163** in Tables 4, 5 and 6.

Summary and conclusions

This work has shown that allylstannanes with an alkoxy or hydroxy substituent at the 4-, 5- or 6-position react with tin(IV) halides at -78 °C to generate intermediates, believed to be allyltin trihalides, which react with aldehydes with useful levels of 1,5-, 1,6- and 1,7- stereocontrol. Fig. 6 shows an overview of the results obtained. Interestingly, stannanes 1,73/74 and 163 with a stereogenic carbon bearing a secondary alkoxy- or hydroxy-substituent at either the 4-, 5- or 6-position react with overall (*Z*)-1,5-, (*Z*)-1,6- and (*Z*)-1,7-*syn*-stereocontrol. In contrast, stannanes 6, 34/35, 115 and 138 with a terminal alkoxy- or hydroxy-substituent and a stereogenic carbon bearing a methyl substituent at either the 4- or 5-position, react with overall (*Z*)-1,5- or (*Z*)-1,6-*anti*-stereocontrol.

Details of mechanistic studies including studies into the trapping of the allyltin trihalides **28** and **216** and evidence for kinetic control of the transmetallation step will be reported in full elsewhere. However, the overall remote stereoselectivity which is reported here is believed to be due to two factors:

(a) the kinetically controlled, diastereofacially selective transmetallation of substituted allylstannanes by the tin(IV) halide to form allyltin trihalides with high levels of stereoselectivity,²¹ and



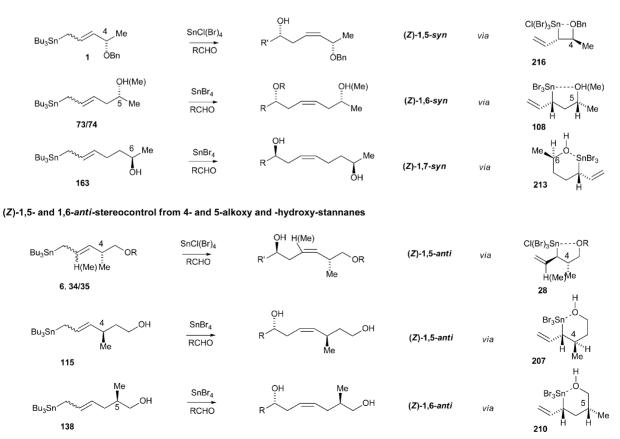


Fig. 6 Overview of remote stereocontrol using alkoxy and hydroxy-alk-2-enylstannanes.

(b) the subsequent reaction of the allyltin trihalides with aldehydes *via* six-membered chair-like transition structures in which the group next to tin adopts an axial position.²²

The allylstannanes used in this work were mixtures of (E)- and (Z)-isomers, typically (E): (Z) = 70: 30. No attempts were made to study reactions of the (E)- and (Z)-isomers separately although studies of the analogous chemistry of alk-2-enylgermanium compounds indicated that (Z)-isomers were transmetallated with better stereoselectivity.²³ To explain the high levels of stereocontrol in the transmetallation step, even for 5- and 6-substituted alk-2-envlstannanes, it is suggested that the tin halide initially coordinates to the alkoxy or hydroxy group and is then delivered to the double-bond of the allylstannane via an intramolecular process. For example, formation of the allyltin trichloride 28 from both the (E)- and Z)-isomers of the (R)-4-methylpent-2enylstannane 6 via intramolecular delivery of the trichlorotin moiety to the double-bond places the methyl substituent in the less hindered "outside" position, see transition structures 217 and 219 in Fig. 7. Formation of the isomeric allyltin trichloride 221 in which the vinyl and methyl groups are *cis*-disposed with respect to the 5-membered oxastannane ring would require the methyl group to be in the more hindered "inside" position, see transition structures 218 and 220. Note that the stereochemistry of the tin(IV) halide mediated transmetallation of allylstannanes by tin(IV) halides has not been defined. An antarafacial process is depicted in Fig. 7 but an alternative 6-membered, cyclic, suprafacial process could be involved. However, the explanation given for the introduction of the stereogenic centre bearing the tin into the intermediate allyltin trihalides would still be valid.21

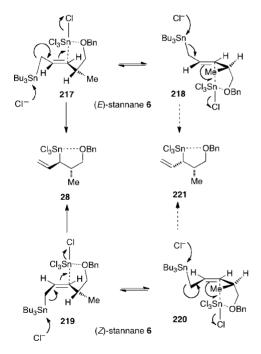
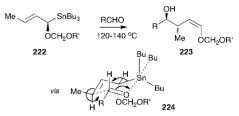


Fig. 7 Transmetallation of the (E)- and (Z)-isomers of the alk-2-envlstannane **6**.

In the second step, the reaction of the allyltin trihalide with the aldehyde, the high fidelity for formation of (Z)-double-bonds in the products is consistent with the participation of 6-membered chair-

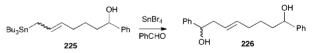
like transition structures, in which the tin is trigonal bipyramidal, *i.e.* transition structures **29**, **109**, **208**, **211** and **214**.

1-Substituted alk-2-enylstannanes are known to react with aldehydes on heating to give (Z)-alkenes; for example the highly stereoselective synthesis of the (Z)-1,2-*anti*-alkenes **223** from the non-catalysed reactions of 1-alkoxybut-2-enylstannanes **222** with aldehydes. The six-membered chair-like transition state **224** in which the group α to the tin is in the axial position, has been suggested to explain the stereoselectivity observed in these reactions.²²



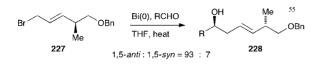
The transition structures 29, 109, 208, 211 and 214 which involve a 1-substituted allyltin trichloride reacting with an aldehyde with formation of an (Z)-alkene, are analogous to the transition structure 224 proposed for the thermal process. In particular, is suggested that the alkoxy or hydroxy group, which initially directed the transmetallation, is no longer bound to the tin in these transition structures. The high preference for (Z)-alkene formation, especially for reactions in which the co-ordinating group is at the 6position in the allylstannane, is difficult to reconcile with transition structures in which the tin is hexacoordinated, *i.e.* octahedral.

This remote stereocontrol using tin(IV) halide mediated reactions of alkoxy and hydroxy-functionalised alk-2-enylstannanes would appear to be generally useful and reliable with better stereoselectivity obtained using tin(IV) bromide and 6-hydroxyalk-2-enylstannanes for 1,6- and 1,7-stereocontrol. Preliminary studies to extend this work to include 1,8-stereocontrol using 7-hydroxy-7-phenylhept-2-enylstannane **225** were unsuccessful, in that a mixture of the (*E*)-1,8-syn-and 1,8-*anti*-diphenyloct-3-ene-1,8diols **225** was obtained (81%, 2:1). However, 1,8- and 1,9stereocontrol has been achieved by combining 1,5-stereocontrol as discussed here, with subsequent stereoselective 2,3- and 3,3sigmatropic rearrangements.^{24,25}



This chemistry has the limitation that it involves the use of organotin reagents, indeed one equivalent of tributyltin chloride or bromide is produced as the side-product in the transmetallation step. Analogous allylsilanes undergo transmetallation with tin(IV) halides but, under the longer reaction times required, the intermediate allyltin trihalides begin to equilibrate with their epimers and only modest stereoselectivities are observed.²⁶ However, allylgermanes undergo transmetallation at -78 °C relatively quickly, and 1,5- and 1,6-remote stereocontrol has been shown for reactions with aldehydes of 5- and 6-alkoxy and -hydroxy-alk-2-enylgermanes. Indeed when using tin(IV) bromide and (*Z*)-alk-2-enylgermanes higher stereoselectivities can be obtained than those found using the corresponding allylstannanes.²³

Bismuth tri-iodide also effects transmetallation of the allylstannane $\mathbf{6}$ to generate an allylbismuth intermediate which reacts with aldehydes with useful 1,5-stereocontrol, albeit in favour of (E)-alk-3-enols.²⁷ Similar stereoselectivity has also been observed for the bismuth(0) promoted reactions of 5-alkoxy-4-methylpent-2-enyl bromides with aldehydes, a tin-free process, for example the formation of the (E)-alkenols **228** from the bromide **227** and a reduced bismuth species.²⁸



Finally, 1,5-stereocontrol has also been observed for tin(IV) halide promoted reactions of amino- and phenylthio-substituted alkenylstannanes with aldehydes, and for reactions of 4- and 5-alkoxyalk-2-enylstannanes with reactive imines.²⁹⁻³¹ Details of this work and of the application of remote stereocontrol using tin(IV) halide mediated reactions of allylstannanes to complete syntheses of the natural products, patulolide C,²⁴ epothilones³² and pamamycin 607,³³ will be reported in full elsewhere.

Experimental

General experimental procedures

NMR spectra were recorded on Varian Unity 500, Bruker AC-300 or Varian XL spectrometers and IR spectra on a Perkin–Elmer 1710FT spectrometer. For minor diastereoisomers, >10%, present in mixtures, just the distinctive peaks are reported. Low resolution chemical ionisation (CI) and electron impact (EI) mass spectra were recorded on a Kratos MS25 mass spectrometer coupled to a DS 55 data system or on a VG Trio 2000 mass spectrometer. Fast atom bombardment (FAB) mass spectra and all high resolution mass spectra were recorded on a Kratos Concept 1S mass spectrometer coupled to a Mach 3 data system. Compounds containing tin or chlorine showed characteristic clusters of isotope peaks in their mass spectra.

Optical rotations were recorded on an Optical Activity AA-100 polarimeter at 589 nm using chloroform as the solvent at ambient temperature. Analytic high performance liquid chromatography was carried out with a Waters Z module, $10 \text{ cm} \times 8 \text{ mm}$ cartridge, C18 5 m stationary phase and detection by ultraviolet absorption using a Perkin–Elmer IC-480 detector at 255 nm. Semi-preparative high performance liquid chromatography was carried out using a Gilson 303 pump (with manometric module), Dynamax 83-211-C column 25 cm $\times 10$ mm, 8 m silica, detection with a Gilson 131 refractive index detector and Gilson 115 UV detector at 254 nm. Chromatography refers to flash chromatography and was carried out using Merck silica 60H (40–60 m, 230–300 mesh) or May and Baker Sorbsil C60 silica gel (40–60 m) as the stationary phase.

Petrol refers to light petroleum which distils between 40 °C and 60 °C. Tin(IV) chloride was dried with phosphorus pentoxide and distilled. Ether refers to diethyl ether. Brine refers to saturated aqueous sodium chloride. All solvents were dried and distilled before use.

Alcohol **3** was prepared as described in the literature.⁵ Its e.e. was estimated by ozonolysis with a reductive work-up to give (*R*)-3-benzyloxy-2-methylpropanol shown to have an e.e. of >85% by Mosher's derivatisation.

General procedure for the reaction of an alkoxyalk-2-enylstannane with an aldehyde in the presence of a Lewis acid

The tin(IV) halide (1.044 M in DCM, 958 μ l, 1.00 mmol) was added to the stannane (1.0 mmol) in DCM (10 ml) at -78 °C. After 10 min, the aldehyde (3.48 M in DCM, 287 μ l, 1.00 mmol) was added and the mixture was maintained at -78 °C for 1 h. Saturated aqueous sodium hydrogen carbonate (5 ml) was added and the mixture allowed to warm to room temperature. Ether (50 ml) and water (50 ml) were added and the organic phase was washed with aqueous ammonia (10%, 50 ml) and brine (50 ml), then dried (MgSO₄). Concentration under reduced pressure and chromatography of the residue gave the products as colourless oils.

General procedure for the preparation of Mosher's derivatives³⁴

The alcohol (0.10 mmol) in carbon tetrachloride (300 μ l) was added to the 2-methoxy-2-phenyl-3,3,3-trifluoropropanoyl chloride (0.20 mmol) in pyridine (300 μ l) at room temperature and the mixture stirred until no starting material remained, typically 1–1.5 h (TLC). 3-Dimethylaminopropylamine (400 μ l) was added and the clear solution was stirred for 10 min. Ether (25 ml) and water (25 ml) were added and the organic phase was washed with dilute aqueous hydrogen chloride (1 M, 20 ml), saturated aqueous sodium carbonate (20 ml) and brine (20 ml), then dried (MgSO₄). Concentration under reduced pressure gave the Mosher's ester as a colourless oil.

General procedure for the preparation of O-acetylmandelates

2-Acetoxy-2-phenylacetic acid (0.30 mmol), DCC (0.30 mmol) and 4-N,N-dimethylaminopyridine (0.005 mmol) were added to the alcohol (0.10 mmol) in DCM (2 ml) and the mixture stirred at room temperature for 15 h. DCM (15 ml) and water (15 ml) were added and the organic phase washed with dilute aqueous hydrogen chloride (3.5 M, 10 ml), saturated aqueous sodium hydrogen carbonate (10 ml) and brine (10 ml), then dried (MgSO₄). Concentration under reduced pressure left a solid which was absorbed onto silica. Chromatography gave the Oacetylmandelate as a colourless oil.

O-[(*R*)-5-Benzyloxy-4-methylpent-2-enyl]-*S*-methyl dithiocarbonate 4

Alcohol 3 [(E):(Z) = 80:20; 3.39 g, 16.46 mmol) in benzene (16 ml) was added to sodium hydride (723 mg; 60% in mineral oil, 18.08 mmol) in benzene (16 ml) at 5 °C. After 1 h at room temperature, carbon disulfide (5.00 g, 65.76 mmol) was added at 5 °C and the mixture was stirred at room temperature for 3 h. Methyl iodide (9.34 g, 65.76 mmol) was added and the mixture stirred for 15 h then filtered through celite washing the filter-cake with DCM (120 ml). The organic extracts were washed with brine (80 ml), dried (MgSO₄) and concentrated under reduced pressure. Chromatography of the residue using petrol: ether (15:1) as eluent gave the title compound 4 (3.72 g, 76%) as a yellow oil, a 4:1 mixture of (E)- and (Z)-isomers, $[\alpha]_{\rm D}$ -1.9 (c = 0.49) (Found: M⁺ + H, 297.0986. $C_{15}H_{21}O_2S_2$ requires M, 297.0983); v_{max}/cm^{-1} 2856, 1453, 1217, 1060, 970 and 737; $\delta_{\rm H}$ (300 MHz, C₆D₆) 0.90 (0.6 H, d, J 7, 4-CH₃), 0.98 (2.4 H, d, J 7, 4-CH₃), 2.21 (0.6 H, s, SCH₃), 2.22 (2.4 H, s, SCH₃), 2.39 (0.8 H, m, 4-H), 2.74 (0.2 H, m, 4-H),

3.07–3.21 (2 H, m, 5-H₂), 4.33 (0.4 H, s, PhCH₂), 4.34 (1.6 H, s, PhCH₂), 4.98 (1.6 H, d, *J* 5.5, 1-H₂), 5.18 and 5.26 (each 0.2 H, dd, *J* 14, 7, 1-H), 5.43 (0.2 H, t, *J* 11.5, 3-H), 5.52–5.69 (1.8 H, m, 2-H and 3-H) and 7.17–7.36 (5 H, m, ArH); $\delta_{\rm C}$ (75 MHz, C₆D₆) 16.4, 17.3, 18.6, 33.1, 36.8, 69.8, 72.9, 74.3, 74.6, 122.5, 122.7, 127.7, 128.3, 139.0, 140.0 and 215.8; *m/z* (CI, NH₃) 314 (M⁺ + 18, 33%), 297 (M⁺ + 1, 82) and 189 (100).

S-[(3*RS*,4*S*)-5-Benzyloxy-4-methylpent-1-en-3-yl]-*S*-methyl dithiocarbonate 5

The xanthate **4** (3.72 g, 12.57 mmol) was heated under reflux in toluene (150 ml) for 15 h. Concentration under reduced pressure gave the title compound **5** (3.72 g, 100%) as a yellow oil, a 1 : 1 mixture of epimers, $[\alpha]_{\rm D}$ +1.5 (c = 0.52) (Found: M⁺, 296.0909). C₁₅H₂₀O₂S₂ requires *M*, 296.0905); $v_{\rm max}/\rm cm^{-1}$ 2856, 1647, 1454, 1101, 924, 870 and 737; $\delta_{\rm H}$ (300 MHz, CDCl₃) 0.98 (3 H, d, *J* 7, 4-CH₃), 2.18 (1 H, m, 4-H), 2.42 (1.5 H, s, SCH₃), 2.43 (1.5 H, s, SCH₃), 3.29–3.46 (2 H, m, 5-H₂), 4.44–4.52 (3 H, m, PhCH₂ and 3-H), 5.11–5.33 (2 H, m, 1-H₂), 5.80 (1 H, m, 2-H), 7.27–7.37 (5 H, m, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃) 13.2, 13.3, 14.3, 14.5, 37.6, 37.8, 51.2, 51.5, 72.7, 72.7, 73.2, 73.4, 117.3, 118.2, 127.7, 127.8, 128.5, 134.8, 136.3, 138.6, 188.8 and 188.9; m/z (CI, NH₃) 314 (M⁺ + 18, 5%), 297 (M⁺ + 1, 24) and 91 (100).

(R)-5-Benzyloxy-4-methylpent-2-enyl(tributyl)stannane 6

Tributyltin hydride (4.80 g, 16.49 mmol) and azobis-isobutyronitrile (135 mg, 0.825 mmol) were added to a degassed solution of the dithiocarbonate 5 (3.72 g, 12.57 mmol) in benzene (125 ml) and the solution was heated under reflux for 2.5 h. After concentration under reduced pressure, chromatography of the residue using petrol: ether (50: 1 + 1% triethylamine) as eluent gave the title compound 6 (5.18 g, 86%) as a colourless oil, a 4:1 mixture of (E)- and (Z)-isomers), $[\alpha]_D$ –6.3 (c = 0.79); (Found: M⁺ - C₄H₉, 423.1723. C₂₁H₃₅O¹²⁰Sn requires M, 423.1710; $v_{\text{max}}/\text{cm}^{-1}$ 2956, 2925, 1455, 1095, 962 and 734; $\delta_{\rm H}$ (300 MHz, CDCl₃) 0.78– 0.98 (15 H, m, $3 \times \text{SnCH}_2\text{CH}_2\text{CH}_2\text{CH}_3$), 1.03 (3 H, m, 4-CH₃), 1.22-1.75 (14 H, m, $3 \times \text{SnCH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ and $1-\text{H}_2$), 2.48 (0.8) H, m, 4-H), 2.78 (0.2 H, m, 4-H), 3.24 (1 H, dd, J 8, 6, 5-H), 3.46 (1 H, m, 5-H'), 4.46–4.57 (2 H, m, PhCH₂), 4.89 (0.2 H, dd, J 10.5, 9.5, 3-H), 5.16 (0.8 H, dd, J 15, 7, 3-H), 5.53-5.67 (1 H, m, 2-H) and 7.25–7.38 (5 H, m, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃) (*E*)-isomer 9.3, 13.9, 14.5, 17.9, 27.5, 29.3, 37.2, 73.1, 76.2, 127.6, 127.7, 128.1, 128.5, 129.3 and 139.0; m/z (EI) 423 (M⁺ – 57, 54%), 291 (33) and 91 (100).

(1S, 5R, 3Z)-6-Benzyloxy-5-methyl-1-phenylhex-3-en-1-ol 7

Following the general procedure, stannane **6** (500 mg, 1.044 mmol), tin(1V) chloride in DCM (1.044 M, 1.00 ml, 1.044 mmol) and benzaldehyde in DCM (4.16 M, 250 µl, 1.044 mmol), with chromatography using petrol:ether (3:1) as eluent, gave the title compound **7** (257 mg, 83%) as a colourless oil, a 97:3 mixture of epimers, $[\alpha]_D$ –88 (c = 0.36); (Found: M⁺ + NH₄, 314.2108. C₂₀H₂₈NO₂ requires *M*, 314.2120); v_{max}/cm^{-1} 3422, 2958, 2926, 1454, 1090, 1029, 913, 877, 740 and 700; δ_H (300 MHz, CDCl₃) major epimer **7**, 0.91 (3 H, d, *J* 7.5, 5-CH₃), 2.39 (1 H, m, 2-H), 2.60 (1 H, dt, *J* 13, 9.5, 2-H'), 2.88 (1 H, m, 5-H), 3.16 (1 H, t, *J* 8.5, 6-H), 3.35 (1 H, dd, *J* 8.5, 5.5, 6-H'), 3.42 (1 H, br s,

OH), 4.56 (2 H, s, PhCH₂), 4.68 (1 H, dd, *J* 8, 3.5, 1-H), 5.36 (1 H, t, *J* 10, 4-H), 5.58 (1 H, td, *J* 10, 6, 3-H) and 7.30–7.40 (10 H, m, ArH); minor epimer **8**, 0.95 (3 H, d, *J* 7.5, 5-CH₃), 4.50 (2 H, s, PhCH₂); $\delta_{\rm C}$ (75 MHz, CDCl₃) 17.4, 32.5, 38.7, 73.2, 73.5, 74.9, 125.9, 126.4, 127.4, 127.9, 128.1, 128.5, 128.6, 136.9, 138.2 and 145.0; *m/z* (CI, NH₃) 314 (M⁺ + 18, 36%), 296 (M⁺, 10) and 279 (M⁺ – 17, 100).

(1*S*,5*R*,3*Z*)-1-[(*R*)-2-Acetoxy-2-phenylacetoxy]-6-benzyloxy-5-methyl-1-phenylhex-3-ene 9

Following the general procedure, the alcohol 7 (25 mg, 0.084 mmol) and (*R*)-2-acetoxy-2-phenylacetic acid, after chromatography using petrol: ether (3:1) as eluent, gave the title compound **9** (33 mg, 83%) as a colourless oil, $[\alpha]_D$ –95 (*c* = 0.37) (Found: M⁺ + NH₄, 490.2597. C₃₀H₃₆NO₅ requires *M*, 490.2593); $v_{\text{max}}/\text{cm}^{-1}$ 2924, 1744, 1453, 1371, 1229, 1206, 1173, 1056 and 735; δ_{H} (300 MHz, CDCl₃) 0.78 (3 H, d, *J* 7, 5-CH₃), 2.13 (3 H, s, O₂CCH₃), 2.42–2.68 (3 H, m, 5-H and 2-H₂), 3.12 (2 H, d, *J* 8, 6-H₂), 4.43 (2 H, s, PhCH₂), 5.03 (1 H, dt, *J* 10.5, 7, 3-H), 5.12 (1 H, dd, *J* 10.5, 9.5, 4-H), 5.75 (1 H, t, *J* 7, 1-H), 5.95 (1 H, s, 2'-H) and 7.20–7.50 (15 H, m, ArH); *m/z* (CI, NH₃) 490 (M⁺ + 18, 100%), 296 (17) and 279 (42).

$(1S,\!5R,\!3Z)\!-\!1\!-\![(S)\!-\!2\!-\!Acetoxy\!-\!2\!-phenylacetoxy]\!-\!6\!-benzyloxy\!-\!5\!-methyl\!-\!1\!-phenylhex\!-\!3\!-ene$ 10

Following the general procedure, the alcohol 7 (25 mg, 0.084 mmol) and (*S*)-2-acetoxy-2-phenylacetic acid, after chromatography using petrol: ether (3:1) as eluent, gave the title compound **10** (35 mg, 88%) as a colourless oil; $[\alpha]_D - 1.5 (c = 0.48)$; (Found: M⁺ + NH₄, 490.2581. C₃₀H₃₆NO₅ requires *M*, 490.2593); $v_{\text{max}}/\text{cm}^{-1}$ 2854, 1745, 1454, 1372, 1232, 1207, 1175, 1058 and 738; δ_{H} (300 MHz, CDCl₃) 0.83 (3 H, d, *J* 7, 5-CH₃), 2.19 (3 H, s, O₂CCH₃), 2.49–2.73 (3 H, m, 5-H and 2-H₂), 3.15–3.28 (2 H, m, 6-H₂), 4.48 (2 H, s, PhCH₂), 5.22–5.36 (2 H, m, 3-H and 4-H), 5.73 (1 H, t, *J* 7, 1-H), 5.99 (1 H, s, 2'-H) and 6.90–7.40 (15 H, m, ArH); *m/z* (CI, NH₃) 490 (M⁺ + 18, 100%), 296 (17) and 279 (39).

(1S,5R,3Z)-1-Acetoxy-6-benzyloxy-5-methyl-1-phenylhex-3-ene 11

Triethylamine (182 mg, 1.802 mmol), 4-N,N-dimethylaminopyridine (4 mg, 0.033 mmol) and acetic anhydride (65 mg, 0.637 mmol) were added to the alcohol 7 (103 mg, 0.348 mmol) in DCM (2 ml) and the solution stirred for 50 h. DCM (20 ml) and water (20 ml) were added and the organic phase was washed with dilute aqueous hydrogen chloride (3.5 M, 15 ml) and brine (15 ml) then dried (MgSO₄). Concentration under reduced pressure and chromatography of the residue using petrol: ether (3:1) as eluent, gave the title compound 11 (105 mg, 89%) as a colourless oil, $[\alpha]_{D}$ -68 (c = 0.50); (Found: M⁺ + NH₄, 356.2227. C₂₂H₃₀NO₃ requires M, 356.2226; v_{max} /cm⁻¹ 2854, 1744, 1372, 1234, 1092, 1028 and 738; $\delta_{\rm H}$ (300 MHz, CDCl₃) 0.86 (3 H, d, J 7, 5-CH₃), 2.05 (3 H, s, O₂CCH₃), 2.52–2.78 (3 H, m, 5-H and 2-H₂), 3.18–3.29 (2 H, m, 6-H₂), 4.48 (2 H, s, PhCH₂), 5.23–5.36 (2 H, m, 3-H and 4-H), 5.76 (1 H, t, J 7, 1-H) and 7.25–7.40 (10 H, m, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃) 17.7, 21.4, 32.7, 34.8, 73.1, 75.2, 75.7, 124.3, 126.8, 127.6, 127.7, 128.1, 128.5, 128.5, 135.8, 138.8, 140.4 and Downloaded by UNIVERSITY OF SOUTH AUSTRALIA on 28 June 2012 ublished on 25 February 2011 on http://pubs.rsc.org | doi:10.1039/C0OB01084G 170.4; *m/z* (CI, NH₃) 356 (M⁺ + 18, 37%), 296 (25), 279 (75) and 85 (100).

Ozone was bubbled through a solution of acetate 11 (103 mg, 0.308 mmol) in chloroform (10 ml) at -60 °C for 30 min. The solution was purged with oxygen for 10 min, dimethyl sulfide (191 mg, 3.08 mmol) was added and the solution was allowed to warm to room temperature. Following concentration under reduced pressure, the residual oil was dissolved in DCM (5 ml) and added dropwise to a suspension of sodium borohydride (100 mg, 2.63 mmol) in methanol (5 ml) at 0 °C. After 3 h at room temperature, dilute aqueous hydrogen chloride (1 M, 5 ml) was added at 0 °C. Upon warming to room temperature, DCM (25 ml) and water (25 ml) were added, and the aqueous phase was extracted with DCM (25 ml). The organic extracts were washed with brine (30 ml), dried (MgSO₄) and concentrated under reduced pressure. Chromatography of the residue using petrol: ether (1:1)as eluent gave a mixture of 3-benzyloxy-2-methylpropanol 16 and (S)-3-acetoxy-1-phenylpropan-1-ol (S)-15 (2:1, 42 mg), and (S)-3acetoxy-3-phenylpropan-1-ol (S)-14 (17 mg, 29%), as a colourless oil, $[\alpha]_{\rm D}$ -79 {c = 0.61; lit.^{1*a*} for (*R*)-14, $[\alpha]_{\rm D}$ +79.2} (Found: M⁺ + NH₄, 212.1294. C₁₁H₁₈NO₃ requires *M*, 212.1287); *v*_{max}/cm⁻¹ 3415, 2956, 1736, 1374, 1240, 1048 and 760; $\delta_{\rm H}$ (300 Mhz, CDCl₃) 1.98– 2.18 (5 H, m, 2-H₂ and O₂CCH₃), 3.65 (2 H, dd, J 7, 5 Hz, 1-H₂), 5.96 (1 H, dd, J 9, 5, 3-H) and 7.25–7.40 (5 H, m, ArH); $\delta_{\rm c}$ (75 MHz, CDCl₃) 21.4, 39.7, 59.0, 73.4, 126.6, 128.3, 128.8, 140.4 and 171.1; m/z (CI, NH₃) 212 (M⁺ + 18, 100%), 152 (95) and 134 (98).

(1*S*,5*R*,3*Z*)-6-Benzyloxy-1-[(*R*)-2-methoxy-2-phenyl-3,3,3-trifluoropropanoyloxy]-5-methyl-1-phenylhex-3-ene 12

Following the general procedure, alcohol 7 (31 mg, 0.105 mmol) and (*S*)-2-methoxy-2-phenyl-3,3,3-trifluoropropanoyl chloride gave the title compound **12** (46 mg, 86%) as a colourless oil, $[\alpha]_D$ –14 (*c* = 0.48) (Found: M⁺ + NH₄, 530.2498. C₃₀H₃₅F₃NO₄ requires *M*, 530.2518); *v*_{max}/cm⁻¹ 2928, 2851, 1749, 1454, 1272, 1170, 1017 and 762; δ_H (300 MHz, CDCl₃) 0.84 (3 H, d, *J* 7, 5-CH₃), 2.57–2.83 (3 H, m, 5-H and 2-H₂), 3.21 (2 H, d, *J* 7.5, 6-H₂), 3.53 (3 H, s, OCH₃), 4.46 (2 H, s, PhCH₂), 5.30–5.37 (2 H, m, 3-H and 4-H), 5.91 (1 H, t, *J* 7, 1-H) and 7.20–7.45 (15 H, m, ArH); δ_F (470 MHz, CDCl₃) –73.0, –73.2, ratio 9:91; *m/z* (CI, NH₃) 530 (M⁺ + 18, 92%), 279 (87), 189 (69) and 85 (100).

(1*S*,5*R*,3*Z*)-6-Benzyloxy-1-[(*S*)-2-methoxy-2-phenyl-3,3,3-trifluoropropanoyloxy]-5-methyl-1-phenylhex-3-ene 13

Following the general procedure, alcohol 7 (26 mg, 0.088 mmol) and (*R*)-2-methoxy-2-phenyl-3,3,3-trifluoropropanoyl chloride, after chromatography using petrol : ether (3 : 1) as eluent, gave the title compound **13** (23 mg, 51%) as a colourless oil, $[\alpha]_D -70$ (c = 0.83) (Found: M⁺ + NH₄, 530.2528. C₃₀H₃₅F₃NO₄ requires *M*, 530.2518); v_{max}/cm^{-1} 2851, 1748, 1454, 1271, 1170, 1121, 1018 and 718; δ_H (300 MHz, CDCl₃) 0.83 (3 H, d, *J* 7, 5-CH₃), 2.56–2.78 (3 H, m, 5-H and 2-H₂), 3.16 (2 H, d, *J* 6.5, 6-H₂), 3.46 (3 H, s, OCH₃), 4.46 (2 H, s, PhCH₂), 5.16–5.25 (2 H, m, 3-H and 4-H), 5.98 (1 H, t, *J* 7, 1-H) and 7.30–7.50 (15 H, m, ArH); δ_F (470 MHz, CDCl₃) -73.0, -73.2, ratio 95 : 5; m/z (CI, NH₃) 530 (M⁺ + 18, 55%) and 279 (100).

O-[(5*R*,2*E*)-5-*tert*-Butyldimethylsilyloxyhex-2-en-1-yl] *S*-methyl dithiocarbonate 70

Allylic alcohol 69 (2.98 g, 12.96 mmol) in benzene (15 ml) was added to sodium hydride (571 mg, a 60% dispersion in mineral oil, 14.27 mmol) in benzene (12.5 ml) at room temperature. After 2 h at 40 °C, carbon disulfide (3.94 g, 51.84 mmol) was added at 5 °C and the mixture was stirred at room temperature for 3 h. Methyl iodide (7.37 g, 51.90 mmol) was added at 5 °C and the mixture stirred for 15 h at room temperature. The mixture was filtered through celite and the filtercake was washed with DCM (100 ml). The organic extracts were washed with brine (75 ml), dried (MgSO₄) and concentrated under reduced pressure to yield the title compound 70 (3.82 g, 92%) as a yellow oil, $[\alpha]_D$ -4.7 (c = 0.80) (Found: M⁺ + H, 321.1389. C₁₄H₂₉O₂S₂Si₂ requires M, 321.1378; v_{max} /cm⁻¹ 2929, 1254, 1217, 1185, 1059, 836 and 775; δ_{H} $(300 \text{ MHz}, C_6 D_6) 0.07 [6 \text{ H}, \text{ s}, \text{Si}(\text{CH}_3)_2], 1.04 [9 \text{ H}, \text{ s}, \text{Si}(\text{CH}_3)_3],$ 1.07 (3 H, d, J 6, 6-H₃), 1.97-2.18 (2 H, m, 4-H₂), 2.22 (3 H, s, SCH₃), 3.68 (1 H, m, 5-H), 5.02 (2 H, d, J 6.5, 1-H₂) and 5.56 and 5.72 (each 1 H, dt, J 15.5, 6.5, 2-H or 3-H); $\delta_{\rm C}$ (75 MHz, C₆D₆) -4.6, -4.4, 18.2, 18.8, 23.6, 26.0, 42.8, 68.2, 74.2, 125.3, 134.5 and 216.0; m/z (CI, NH₃) 321 (M⁺ + 1, 63%), 189 (97), 159 (100) and 132 (84).

S-[(*3RS*,5*R*)-5-*tert*-Butyldimethylsilyloxyhex-1-en-3-yl] *S*'-methyl dithiocarbonate 71

Xanthate **70** (3.82 g, 11.93 mmol) was heated under reflux in toluene (120 ml) for 15 h. Concentration under reduced pressure gave the title compound **71** (3.82 g, 100%) as a yellow oil, a 1 : 1 mixture of epimers (Found: M^+ + H, 321.1376. $C_{14}H_{29}O_2S_2S_1$; requires M, 321.1378); v_{max}/cm^{-1} 2930, 1649, 1255, 1047, 865, 836 and 775; δ_H (300 MHz, CDCl₃) 0.05 and 0.06 (each 3 H, s, SiCH₃), 0.89 and 0.91 [each 4.5 H, s, SiC(CH₃)₃], 1.16 (3 H, d, *J* 6, 6-H₃), 1.64–1.88 (2 H, m, 4-H₂), 2.41 and 2.42 (each 1.5 H, s, SCH₃), 3.89 (1 H, m, 5-H), 4.29 (1 H, m, 3-H), 5.06–5.32 (2 H, m, 1-H₂) and 5.80 (1 H, m, 2-H); δ_C (75 MHz, CDCl₃) -4.8, -4.6, -4.2, -4.0, 12.9, 13.0, 18.0, 23.8, 24.0, 25.9, 25.9, 29.7, 43.2, 43.8, 45.1, 45.8, 65.7, 66.0, 115.9, 116.9, 137.4, 138.0, 188.5 and 188.6; *m/z* (CI, NH₃) 321 (M⁺ + 1, 26%) and 189 (100).

(R)-5-tert-Butyldimethylsilyloxyhex-2-enyl(tributyl)stannane 72

Tributyltin hydride (2.27 g, 7.81 mmol) and a-azo-bis-isobutyronitrile (51 mg, 0.311 mmol) were added to a degassed solution of the dithiocarbonate 71 (2.0 g, 6.25 mmol) in benzene (85 ml) and the solution was heated to reflux for 3.5 h. Following concentration under reduced pressure, chromatography of the residue using petrol (with 2% triethylamine) as eluent afforded the title compound 72 (2.66 g, 85%) as a colourless oil, a 2:1 mixture of (*E*)- and (*Z*)-isomers, $[\alpha]_{D}$ +2.3 (*c* = 1.16) (Found: M⁺ – C₄H₉, 447.2109. C₂₀H₄₃OSi¹²⁰Sn requires M, 447.2104); v_{max} /cm⁻¹ 2928, 1463, 1376, 1254, 1129, 1090, 1004, 836 and 775; $\delta_{\rm H}$ (300 MHz, CDCl₃) 0.05 and 0.07 (each 3 H, s, SiCH₃), 0.84–0.94 [24 H, m, 3× SnCH₂CH₂CH₂CH₃ and SiC(CH₃)₃], 1.11 (2 H, d, J 6, 6-H₃), 1.13 $(1 \text{ H}, d, J 6, 6\text{-H}_3), 1.24\text{--}1.74 (14 \text{ H}, m, 3 \times \text{SnCH}_2\text{C}H_2\text{C}H_2\text{C}H_3)$ and 1-H₂), 1.98-2.23 (2 H, m, 4-H₂), 3.73 (0.66 H, m, 5-H), 3.81 (0.34 H, m, 5-H), 5.08 (0.33 H, dt, J 10.5, 7, 2-H), 5.18 (0.67 H, dt, J 15, 7, 2-H) and 5.57 (1 H, m, 3-H); $\delta_{\rm C}$ (75 MHz, CDCl₃) –4.6, -4.5, 9.2, 9.4, 10.7, 13.7, 14.4, 18.2, 23.2, 23.5, 25.9, 27.3, 29.1,

37.4, 43.3, 69.0, 69.4, 120.5, 122.0, 129.7 and 131.5; *m*/*z* (EI) 447 (M⁺ – 57, 86%), 445 (M⁺ – 57, 70) and 291 (100).

(R)-5-Hydroxyhex-2-enyl(tributyl)stannane 73

Tetrabutylammonium fluoride in THF (1 M, 10.4 ml, 10.4 mmol) was added to the silvl ether 72 (3.39 g, 6.74 mmol) in THF (10 ml) and was stirred at room temperature for 15 h. Water (20 ml) was added and the mixture was stirred for a further 1 h. The mixture was extracted with ether $(4 \times 20 \text{ ml})$ and the organic extracts were washed with brine (75 ml), dried (MgSO₄) and concentrated under reduced pressure. Chromatography of the residue using petrol: ether (10:1 + 1% triethylamine) as solvent, gave the title compound 73 (2.31 g, 88%) as a colourless oil, a 2:1 mixture of (*E*)- and (*Z*)-isomers), $[\alpha]_D$ +2.3 (*c* = 0.70) (Found: M⁺ – C₄H₉, 333.1236. $C_{14}H_{29}O^{120}Sn$ requires M, 333.1239); v_{max}/cm^{-1} 3351, 2925, 1463, 1376, 1073 and 961; $\delta_{\rm H}$ (300 Mz, CDCl₃) 0.76–0.95 (15 H, m, 3×SnCH₂CH₂CH₂CH₃), 1.18 (2 H, d, J 6.5, 6-H₃), 1.22 $(1 \text{ H}, d, J 6, 6\text{-H}_3), 1.25\text{--}1.87 (14 \text{ H}, m, 3 \times \text{SnCH}_2\text{C}H_2\text{C}H_2\text{C}H_3)$ and 2-H₂), 2.00–2.26 (2 H, m, 4-H₂), 3.72 (0.7 H, m, 5-H), 3.81 (0.3 H, m, 5-H), 5.16 and 5.69 (each 1 H, m, 2-H or 3-H); $\delta_{\rm C}$ (75 MHz, CDCl₃) 9.2, 9.4, 10.8, 13.7, 14.5, 22.5, 22.8, 27.3, 29.1, 36.9, 42.8, 67.5, 67.9, 119.2, 120.8, 132.0 and 133.8; m/z (EI) 333 (M⁺ - 57, 1%), 331 M⁺ - 57, 1), (291 (9), 269 (17), 135 (43) and 84 (100).

(R)-5-Methoxyhex-2-enyl(tributyl)stannane 74

Hydroxyalkenylstannane 73 (516 mg, 1.326 mmol) in THF (5 ml) was added to a suspension of sodium hydride (64 mg of a 60% dispersion in mineral oil, 1.60 mmol) in THF (1 ml) at room temperature. After 2 h at 35 °C, methyl iodide (941 mg, 6.630 mmol) was added at room temperature and the suspension was stirred for 15 h. Water (5 ml) was added and the mixture was extracted with ether $(4 \times 15 \text{ ml})$. The organic extracts were washed with brine (50 ml), dried (MgSO₄) and concentrated under reduced pressure. Chromatography of the residue using petrol: ether (40:1 + 1% triethylamine) gave the title compound 74 (452 mg, 85%) as a colourless oil, a 2:1 mixture of the (E)- and (Z)-isomers, $[\alpha]_{D}$ +7.3 (c = 1.90) (Found: M⁺ - C₄H₉, 347.1398. $C_{15}H_{31}O^{120}Sn$ requires M, 347.1397); v_{max}/cm^{-1} 2925, 1463, 1376, 1129, 1099 and 960; $\delta_{\rm H}$ (300 MHz, CDCl₃) 0.75–0.96 (15 H, m, 3×SnCH₂CH₂CH₂CH₃), 1.11 (2 H, d, J 6, 6-H₃), 1.14 (1 H, d, J 6, 6-H₃), 1.22–1.84 (14 H, m, $3 \times \text{SnCH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ and 1-H₂), 1.98–2.35 (2 H, m, 4-H₂), 3.25 (1 H, m, 5-H), 3.31 (2 H, s, OCH₃), 3.33 (1 H, s, OCH₃), 5.08 (0.34 H, dt, J 10.5, 7, 2-H), 5.18 (0.66 H, dt, J 15, 7.5, 2-H) and 5.60 (1 H, m, 3-H); $\delta_{\rm C}$ (75 MHz, CDCl₃) 9.2, 9.4, 10.7, 13.7, 14.4, 18.8, 19.1, 27.3, 29.1, 33.6, 39.4, 55.9, 56.0, 77.4, 119.7, 121.3, 130.2 and 131.8; m/z (EI) 347 (M⁺ – 57, 13%), 345 (M⁺ – 57, 9), 291 (20), 269 (11) and 59 (100).

(R)-5-Hydroxyhexyl(tributyl)stannane 75

Toluene 4-sulfonyl hydrazide (239 mg, 1.28 mmol) and then sodium acetate (105 mg, 1.28 mmol) were added to the allylstannane **73** (100 mg, 0.257 mmol) in ethanol (10 ml) and the solution was heated under reflux. After 2 h, further portions of toluene 4-sulfonyl hydrazide (239 mg, 1.28 mmol) and sodium acetate (105 mg, 1.28 mmol) were added and the solution was heated under reflux for a further 2 h. Following concentration under reduced pressure, hexane (25 ml) and water (25 ml) were added.

The organic phase was washed with brine (20 ml), dried (MgSO₄) and concentrated under reduced pressure. Chromatography of the residue using petrol : ether (5 : 1 + 1% triethylamine) as the eluent, gave the title compound **75** (83 mg, 83%) as a colourless oil, $[\alpha]_{\rm D}$ –3.7 (c = 0.86) (Found: M⁺ – C₄H₉, 335.1393. C₁₄H₃₁O¹²⁰Sn requires *M*, 335.1396); $v_{\rm max}/{\rm cm}^{-1}$ 3339, 2925, 1462, 1376 and 1072; $\delta_{\rm H}$ (300 MHz, CDCl₃) 0.74–0.97 (17 H, m, 1-H₂ and 3 × SnCH₂CH₂CH₂CH₃), 1.22 (3 H, d, *J* 6, 6-H₃), 1.25–1.65 (18 H, m, 2-H₂, 3-H₂, 4-H₂ and 3 × SnCH₂CH₂CH₂CH₃) and 3.83 (1 H, m, 5-H); $\delta_{\rm C}$ (75 MHz, CDCl₃) 8.8, 9.0, 13.7, 23.5, 27.1, 27.4, 29.3, 30.5, 39.0 and 68.2; *m/z* (EI) 335 (M⁺ – 57, 100%), 333 (M⁺ – 57, 80) and 177 (74).

(*R*)-5-[(*R*)-2-Methoxy-2-phenyl-3,3,3-trifluoropropanonyloxy]hexyl(tributyl)stannane 76

Following the general procedure, alcohol **75** (14 mg, 0.036 mmol) and (*S*)-2-methoxy-2-phenyl-3,3,3-trifluoropropanoyl chloride afforded the title compound **76** (22 mg, 100%) as a colourless oil, $[\alpha]_{\rm D}$ +14.5 (*c* = 0.94) (Found: M⁺ – C₄H₉, 551.1794. C₂₄H₃₈F₃O₃¹²⁰Sn requires *M*, 551.1794); *v*_{max}/cm⁻¹ 2925, 1746, 1454, 1270, 1170, 1123, 1020 and 716; $\delta_{\rm H}$ (500 MHz, C₆D₆) 0.86–1.10 (15 H, m, 3 × SnCH₂CH₂CH₃O₃), 1.12 (3 H, d, *J* 6, 6-H₃), 1.30–1.76 (20 H, m, 3 × SnCH₂CH₂CH₂CH₃ and 1-H₂, 2-H₂, 3-H₂ and 4-H₂), 3.57 (3 H, s, OCH₃), 5.19 (1 H, m, 5-H), 7.13–7.24 (3 H, m, ArH) and 7.80–7.84 (2 H, m, ArH); $\delta_{\rm F}$ (470 MHz, C₆D₆) –73.0; *m*/*z* (EI) 551 (M⁺ – 57, 21%), 549 (M⁺ – 57, 15) and 317 (100).

(*R*)-5-[(*S*)-2-Methoxy-2-phenyl-3,3,3-trifluoropropionyloxy]hexyl(tributyl)stannane 77

Following the general procedure, alcohol **75** (37 mg, 0.095 mmol) and (*R*)-2-methoxy-2-phenyl-3,3,3-trifluoropropanoyl chloride afforded the title compound **77** (40 mg, 70%) as a colourless oil, $[\alpha]_{\rm D}$ –36.5 (*c* = 1.62) (Found: M⁺ – C₄H₉, 551.1788. C₂₄H₃₈F₃O₃¹²⁰Sn requires *M*, 551.1794); *v*_{max}/cm⁻¹ 2925, 1746, 1454, 1270, 1169, 1123, 1081, 1021 and 715; $\delta_{\rm H}$ (500 MHz, C₆D₆) 0.86–1.08 (15 H, m, 3 × SnCH₂CH₂CH₂CH₃), 1.16 (3 H, d, *J* 6, 6-H₃), 1.20 – 1.75 (20 H, m, 3 × SnCH₂CH₂CH₂CH₂CH₃ and 1-H₂, 2-H₂, 3-H₂ and 4-H₂), 3.57 (3 H, s, OCH₃), 5.20 (1 H, m, 5-H), 7.14–7.23 (3 H, m, ArH) and 7.80–7.84 (2 H, m, ArH); $\delta_{\rm F}$ (470 MHz, C₆D₆) –72.9; *m*/*z* (CI, NH₃) 551 (M⁺ – 57, 7%), 549 (M⁺ – 57, 5), 308 (56) and 30 (100).

(1R,6R,3Z)-6-Methoxy-1-phenylhept-3-en-1-ol 78

Following the general procedure, stannane **74** (450 mg, 1.12 mmol), tin(IV) bromide (1.044 M in DCM, 1.07 ml, 1.12 mmol) and benzaldehyde (321 µl, 1.12 mmol), after chromatography using petrol: ether (2:1) as eluent, gave the title compound **78** (193 mg, 78%) as a colourless oil, a 91:9 mixture of epimers, $[\alpha]_D$ +95 (c = 0.90) (Found: M⁺ – OH, 203.1434. C₁₄H₁₈O requires *M*, 203.1436); v_{max}/cm^{-1} 3417, 2972, 1493, 1453, 1135, 1086, 1054, 760 and 701; δ_H (300 MHz, CDCl₃) 1.16 (3 H, d, *J* 6, 7-H₃), 2.12 (1 H, dt, *J* 14, 5, 2- or 5-H), 2.30–2.50 (2 H, m, 2 × 2- or 5-H), 2.58 (1 H, m, 2- or 5-H), 3.15 (1 H, br s, OH), 3.25–3.40 (4 H, m, 6-H and OCH₃), 4.71 (1 H, dd, *J* 8.5, 4.5, 1-H), 5.50–5.66 (2 H, m, 3-H and 4-H) and 7.23–7.39 (5 H, m, ArH); δ_C (75 MHz, CDCl₃) major epimer **78** 18.9, 34.5, 37.8, 56.1, 73.5, 76.5, 125.8, 127.3, 127.4, 128.3, 129.3 and 144.6; minor epimer **81** 18.6, 34.0

Downloaded by UNIVERSITY OF SOUTH AUSTRALIA on 28 June 2012 ublished on 25 February 2011 on http://pubs.rsc.org | doi:10.1039/C0OB01084G and 37.3; *m*/*z* (CI, NH₃) 238 (M⁺ + 18, 1%), 220 (M⁺, 9), 203 (M⁺ - 17, 96), 171 (98), 121 (66) and 59 (100).

(1*R*,6*R*,3*Z*)-6-Methoxy-1-(4-nitrobenzoyloxy)-1-phenylhept-3-ene 79

Triethylamine (27 mg, 0.27 mmol), 4-N,N-dimethylaminopyridine (2 mg, 0.016 mmol) and 4-nitrobenzoyl chloride (59 mg, 0.318 mmol) were added to the alcohol 78 (23 mg, 0.105 mmol) in DCM (2 ml) and the mixture stirred at room temperature for 72 h. Ether (20 ml) and water (20 ml) were added, and the organic phase was washed with aqueous hydrogen chloride (3.5 M, 15 ml) and brine (15 ml) then dried (MgSO₄). After concentration under reduced pressure, chromatography of the residue using petrol ether (4:1) as eluent gave the title compound 79 (34 mg, 88%) as a colourless oil, $[\alpha]_{\rm D}$ -18 (c = 1.58 (Found: M⁺ + NH₄, 387.1914. $C_{21}H_{27}N_2O_5$ requires M, 387.1919); v_{max}/cm^{-1} 2972, 1725, 1607, 1529, 1345, 1273, 1102 and 720; $\delta_{\rm H}$ (300 MHz, CDCl₃) 1.11 (3 H, d, J 6 7-H₃), 2.16 and 2.29 (each 1 H, dt, J 14.5, 7, 5-H), 2.75 and 2.92 (each 1 H, dt, J 14.5, 7, 2-H), 3.28 (1 H, m, 6-H), 3.33 (3 H, s, OCH₃), 5.47–5.62 (2 H, m, 3-H and 4-H), 6.06 (1 H, t, J 7, 1-H), 7.31–7.49 (5 H, m, ArH) and 8.23–8.36 (4 H, m, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃), 18.9, 34.1, 34.5, 56.1, 76.4, 77.3, 123.5, 125.2, 126.6, 128.3, 128.6, 129.4, 130.8, 135.8, 139.6, 150.6 and 163.9; m/z (CI, NH₃) 387 (M⁺ + 18, 8%), 203 (100) and 171 (56).

(1R,6R,3Z)-1-Acetoxy-6-methoxy-1-phenylhept-3-ene 80

Triethylamine (402 mg, 3.98 mmol), 4-N,N-dimethylaminopyridine (5 mg, 0.041 mmol) and acetic anhydride (162 mg, 1.59 mmol) were added to the alcohol 78 (175 mg, 0.795 mmol) in DCM (4 ml) and the mixture stirred at room temperature for 15 h. DCM (35 ml) and water (35 ml) were added, and the organic phase washed with aqueous hydrogen chloride (3.5 M, 25 ml), saturated aqueous sodium hydrogen carbonate (25 ml) and brine (15 ml) then dried (MgSO₄). After concentration under reduced pressure, chromatography of the residue using petrol: ether (3:1) as eluent gave the title compound 80 (192 mg, 92%) as a colourless oil, $[\alpha]_{\rm D}$ +34 (c = 0.7); Found: M⁺ + NH₄, 280.1917. $C_{16}H_{26}NO_3$ requires M, 280.1913); v_{max}/cm^{-1} 2972, 1741, 1373, 1237, 1094, 1024 and 700; $\delta_{\rm H}$ (300 MHz, C₆D₆) 1.07 (3 H, d, J 6, 7-H₃), 1.74 (3 H, s, O₂CCH₃), 2.12, 2.31, 2.56 and 2.76 (each 1 H, dt, J 14, 7, 2-H or 5-H), 3.12 (1 H, m, 6-H), 3.16 (3 H, s, OCH₃), 5.50 and 5.63 (each 1 H, m, 3- or 4-H), 6.02 (1 H, t, J 7, 1-H), and 7.10–7.37 (5 H, m, ArH); $\delta_{\rm C}$ (75 MHz, C₆D₆) 19.1, 20.7, 34.4, 34.8, 55.8, 75.5, 76.6, 125.9, 127.0, 128.6, 129.3, 140.9 and 169.4; m/z (CI, NH₃) 280 M⁺ + 18, 50%), 220 (71), 203 (100) and 171 (100).

Ozone was bubbled through a solution of the acetoxyalkene **80** (100 mg, 0.382 mmol) in chloroform (5 ml) at -78 °C for 15 min. The solution was purged with oxygen for 10 min, then dimethyl sulfide (170 mg, 2.73 mmol) was added and the mixture allowed to warm to room temperature. After concentration under reduced pressure, the residual oil was dissolved in DCM (2 ml), and methanol (2 ml) and sodium borohydride (100 mg, 2.63 mmol) were added at 0 °C. After 1.5 h at room temperature aqueous hydrogen chloride (1 M, 4 ml) was added at 0 °C. After warming to room temperature DCM (25 ml) and water (25 ml) were added, and the aqueous phase was extracted with DCM (25 ml). The

organic extracts were washed with brine (30 ml), dried (MgSO₄) and concentrated under reduced pressure. Chromatography of the residue using petrol : ether (2 : 1 to 1 : 2) as eluent gave (*R*)-1-phenyl-3-acetoxypropan-1-ol (*R*)-**15** (11 mg, 15%) as a colourless oil, $[\alpha]_{\rm D}$ +15 (c = 0.49) (Found: M⁺, 194.0943. C₁₁H₁₄O₃ requires *M*, 194.0943); $v_{\rm max}/{\rm cm}^{-1}$ 3438, 2957, 1738, 1368, 1243, 1041 and 702; $\delta_{\rm H}$ (300 MHz, CDCl₃) 2.01–2.45 (6 H, m, O₂CCH₃, 2-H₂ and OH), 4.16 (1 H, dt, *J* 11.5, 5.5, 3-H), 4.35 (1 H, ddd, *J* 11.5, 7.5, 5.5, 3-H'), 4.83 (1 H, dd, *J* 8, 5.5, 1-H) and 7.30–7.42 (5 H, m, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃) 21.0, 38.0, 61.6, 71.3, 125.8, 127.8, 128.6, 143.9 and 171.3; *m/z* (CI, NH₃) 212 (M⁺ + 18, 4%), 193 (M⁺ – 1, 7) and 177 (M⁺ – 17, 100). The second fraction contained (*R*)-3-acetoxy-3-phenylpropan-1-ol (*R*)-**14** (43 mg, 58%) as a colourless oil, $[\alpha]_{\rm D}$ +69 (c = 1.35; lit.^{1a,9} +79.2), all other data as for the (*S*)-enantiomer (*S*)-**14** prepared earlier.

(1S,6R,3Z)-6-Methoxy-1-phenylhept-3-en-1-ol 81

Sodium hydroxide (57 mg, 1.43 mmol) was added to the 4nitrophenyl ester 86 (80 mg, 0.216 mmol) in methanol (6 ml). After 2 h at room temperature water (15 ml) was added and the mixture extracted with ether $(2 \times 20 \text{ ml})$. The organic extracts were washed with brine (30 ml), dried (MgSO₄) and concentrated under reduced pressure. Chromatography of the residue using petrol: ether (2:1) as eluent gave the title compound 81 (40 mg, 84%) as a colourless oil, $[\alpha]_D - 72$ (c = 1.78) (Found: M⁺, 220.1461. C₁₄H₂₀O₂ requires M, 220.1463); v_{max} /cm⁻¹ 3422, 3126, 2972, 1494, 1453, 1378, 1199, 1135, 1087, 1052, 760 and 702; $\delta_{\rm H}$ (300 MHz, CDCl₃) 1.18 (3 H, d, J 6.5, 7-H₃), 2.28 (2 H, t, J 6.5, 5-H₂), 2.58 (2 H, t, J 6.5, 2-H₂), 3.09 (1 H, br s, OH), 3.31-3.42 (4 H, m, 6-H and OCH₃), 4.78 (1 H, t, J 6.5, 1-H), 5.53 (1 H, dt, J 11, 7, 3-H), 5.64 (1 H, dt, J 11, 7, 4-H) and 7.26–7.42 (5 H, m, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃) major epimer 81 18.6, 34.0, 37.3, 56.1, 73.4, 76.4, 125.8, 127.1, 127.3, 128.3, 128.8 and 144.5; minor epimer 78 18.9, 34.5 and 37.8; m/z (CI, NH₃) 238 (M⁺ + 18, 38%), 220 (M⁺, 33) and 203 (100).

(1*R*,6*R*,3*Z*)-6-Methoxy-1-[(*R*)-2-methoxy-2-phenyl-3,3,3-trifluoropropanoyloxy]-1-phenylhept-3-ene 82

Following the general procedure, alcohol **78** (16 mg, 0.072 mmol) and (*S*)-2-methoxy-2-phenyl-3,3,3-trifluoropropanoyl chloride afforded the title compound **82** (31 mg, 97%) as a colourless oil, $[\alpha]_{\rm D}$ +52 (*c* = 1.35) (Found: M⁺ + NH₄, 454.2201. C₂₄H₃₁F₃NO₄ requires *M*, 454.2205); *v*_{max}/cm⁻¹ 2967, 1748, 1261, 1170, 1095, 1018, 801 and 700; $\delta_{\rm H}$ (300 MHz, CDCl₃) 1.08 (3 H, d, *J* 6, 7-H₃), 2.05 and 2.17 (each 1 H, dt, *J* 14.5, 7, 5-H), 2.62 and 2.76 (each 1 H, dt, *J* 14.5, 7, 2-H), 3.22 (1 H, m, 6-H), 3.31 and 3.49 (each 3 H, s, OCH₃), 5.36 (1 H, m, 3- or 4-H), 5.49 (1 H, dt, *J* 10.5, 7, 3- or 4-H), 6.02 (1 H, dd, *J* 7, 6.5, 1-H) and 7.31–7.48 (10 H, m, ArH); $\delta_{\rm F}$ (470 MHz, CDCl₃) –73.0, –73.1, ratio 91 : 9; *m*/*z* (CI, NH₃) 454 (M⁺ + 18, 8%), 203 (100) and 171 (62).

(1*R*,6*R*,3*Z*)-6-Methoxy-1-[(*S*)-2-methoxy-2-phenyl-3,3,3-trifluoropropanoyloxy]-1-phenylhept-3-ene 83

Following the general procedure, alcohol **78** (15 mg, 0.068 mmol) and (*R*)-2-methoxy-2-phenyl-3,3,3-trifluoropropanoyl chloride afforded the title compound **83** (23 mg, 78%) as a colourless oil, $[\alpha]_{\rm D}$ –4.9 (c = 0.93) (Found: M⁺ + NH₄, 454.2210. C₂₄H₃₁F₃NO₄ requires *M*, 454.2205); $v_{\rm max}/\rm{cm}^{-1}$ 2972, 1749, 1272, 1170, 1124,

Downloaded by UNIVERSITY OF SOUTH AUSTRALIA on 28 June 2012 Dublished on 25 February 2011 on http://pubs.rsc.org | doi:10.1039/C00B01084G 1017, 995, 719 and 699; $\delta_{\rm H}$ (300 MHz, CDCl₃) 1.09 (3 H, d, *J* 6, 7-H₃), 2.11 and 2.23 (each 1 H, dt, *J* 14.5, 7, 5-H), 2.64 and 2.82 (each 1 H, dt, *J* 14.5, 7, 2-H), 3.27 (1 H, m, 6-H), 3.32 and 3.58 (each 3 H, s, OCH₃), 5.48 (1 H, m, 3- or 4-H), 5.58 (1 H, dt, *J* 10.5, 7, 3- or 4-H), 5.96 (1 H, dd, *J* 7.5, 6.5, 1-H) and 7.23–7.47 (10 H, m, ArH); $\delta_{\rm F}$ (470 MHz, CDCl₃) –73.0, –73.2, ratio 10:90; m/z (CI, NH₃) 454 (M⁺ + 18, 19%), 220 (18) and 203 (100).

(1*R*,6*R*,3*Z*)-1-[(*R*)-2-Acetoxy-2-phenylacetoxy]-6-methoxy-1-phenylhept-3-ene 84

Following the general procedure, alcohol **78** (9 mg, 0.041 mmol) and (*R*)-2-acetoxy-2-phenylacetic acid, after chromatography using petrol: ether (2:1) as eluent, gave the title compound **84** (13 mg, 80%) as a colourless oil, $[\alpha]_D -31$ (c = 0.53) (Found: M⁺ + NH₄, 414.2265. C₂₄H₃₂NO₅ requires M, 414.2281); v_{max}/cm^{-1} 2931, 1747, 1373, 1232, 1208, 1175, 1089, 1056 and 699; δ_H (300 MHz, CDCl₃) 1.08 (3 H, d, *J* 6, 7-H₃), 2.07 (1 H, m, 5-H), 2.17 –2.30 (4 H, m, 5-H' and O₂CCH₃), 2.57 and 2.69 (each 1 H, dt, *J* 14.5, 7, 2-H), 3.24 (1 h, m, 6-H), 3.32 (3 H, s, OCH₃), 5.43 (1 H, m, 3- or 4-H), 5.54 (1 H, dt, *J* 10.5, 7, 3- or 4-H), 5.77 (1 H, t, *J* 7, 1-H), 6.02 (1 H, s, 2'-H) and 6.98–7.55 (10 H, m, ArH); m/z (CI, NH₃) 414 (M⁺ + 18, 4%), 341 (17), 203 (100), 171 (76) and 59 (88).

(1*R*,6*R*,3*Z*)-1-[(*S*)-2-Acetoxy-2-phenylacetoxy]-6-methoxy-1-phenylhept-3-ene 85

Following the general procedure, alcohol **78** (12 mg, 0.055 mmol) and (*S*)-2-acetoxy-2-phenylacetic acid, after chromatography using petrol: ether (4:1) as eluent, gave the title compound **85** (19 mg, 87%) as a colourless oil, $[\alpha]_D + 75 (c = 0.90)$ (Found: M⁺ + NH₄, 414.2287. C₂₄H₃₂NO₅ requires *M*, 414.2281); v_{max}/cm^{-1} 2972, 1747, 1373, 1232, 1176, 1089, 1056 and 699; δ_H (300 MHz, CDCl₃) 1.04 (3 H, d, *J* 6, 7-H₃), 1.92–2.14 (2 H, m, 5-H₂), 2.21 (3 H, s, O₂CCH₃), 2.42–2.63 (2 H, m, 2-H₂), 3.18 (1 H, m, 6-H), 3.24 (3 H, s, OCH₃), 5.17 (1 H, m, 3- or 4-H), 5.36 (1 H, dt, *J* 11, 7.5, 3- or 4-H), 5.79 (1 H, t, *J* 6.5, 1-H), 6.03 (1 H, s, 2'-H) and 7.18–7.55 (10 H, m, ArH); m/z (CI, NH₃) 414 (M⁺ + 18, 28%) and 203 (100).

(1*S*,6*R*,3*Z*)-6-Methoxy-1-(4-nitrobenzoyloxy)-1-phenylhept-3-ene 86

Diethyl azodicarboxylate (94 mg, 0.540 mmol) was added to a suspension of the alcohol 78 (79 mg, 0.359 mmol), triphenylphosphine (141 mg, 0.538 mmol) and 4-nitrobenzoic acid (90 mg, 0.539 mmol) in toluene (3 ml) at -35 °C and the mixture was allowed to warm to room temperature. After 1.75 h, ether (25 ml) and water (25 ml) were added. The organic phase was washed with brine (20 ml), dried (MgSO₄) and concentrated under reduced pressure. Chromatography of the residue using petrol: ether (6:1) as eluent gave the title compound 86 (106 mg, 80%) as a colourless oil, $[\alpha]_{D}$ +35.5 (c = 0.76) (Found: M⁺ + NH₄, 387.1917. C₂₁H₂₇N₂O₅ requires M, 387.1919); v_{max}/cm⁻¹ 2972, 1726, 1607, 1529, 1345, 1273, 1102, 720 and 700; $\delta_{\rm H}$ (300 MHz, CDCl₃) 1.12 (3 H, d, J 6 7-H₃), 2.16 (1 H, dt, J 14, 7, 5-H), 2.30 (1 H, dt, J 14.5, 6, 5-H), 2.75 and 2.92 (each 1 H, dt, J 14.5, 7, 2-H), 3.28 (1 H, m, 6-H), 3.32 (3 H, s, OCH₃), 5.45–5.63 (2 H, m, 3-H and 4-H), 6.08 (1 H, t, J 7, 1-H), 7.32–7.48 (5 H, m, ArH) and 8.24–8.35 (4 H, m, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃), 18.9, 34.1, 34.5, 56.1, 76.4, 77.3, 123.5, 125.3,

126.5, 128.3, 128.6, 129.3, 130.8, 135.8, 139.6, 150.6 and 163.9; $m/z~{\rm (CI,\,NH_3)}$ 387 (M+ + 18, 10%), 203 (100) and 171 (44).

N-Methyl-2-(prop-2-enylthio)imidazole 133

Allyl bromide (10 g, 82.64 mmol) was added dropwise to 2mercapto-1-methylimidazole (4.71 g, 41.32 mmol) and potassium carbonate (5.70 g, 41.32 mmol) in acetone (100 ml) and the mixture stirred for 3 h. Water (50 ml) was added and the stirring continued for a further 30 min. The mixture was extracted with ether and the organic extracts washed with brine and dried (MgSO₄). Following concentration under reduced distillation of the residue at 160 °C, 2 mm Hg, gave the title compound **133** (5.3 g, 83%) as a yellow oil (Found: M⁺, 154.0565. C₇H₁₀N₂S requires *M*, 154.0565); v_{max} /cm⁻¹ 3375, 1459, 1413, 1280, 1125, 920, 734 and 688; $\delta_{\rm H}$ (300 MHz, CDCl₃) 3.60 (5 H, m, NCH₃, CH₂S), 5.00 (2 H, m, 3'-H₂), 5.90 (1 H, ddt, *J* 17, 10, 7, 2'-H), 6.90 (1 H, d, *J* 1, ArH) and 7.05 (1 H, d, *J* 1, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃) 33.4, 37.8, 118.0, 122.4, 129.4, 133.6 and 140.8; *m*/*z* (EI) 154 (M⁺, 100%).

(S)-3-tert-Butyldimethylsilyloxy-1-iodo-2-methylpropane 135

Methanesulfonyl chloride (12.0 ml, 155.3 mmol) was added dropwise to the alcohol **134** (26.4 g, 129.4 mmol) and triethylamine (36 ml, 259 mmol) in DCM (200 ml) at -20 °C. The mixture was allowed to warm to ambient temperature over 1 h and was then washed with sodium hydroxide (10% w/v) and water then dried (MgSO₄). Concentration under reduced pressure gave the corresponding mesylate (36.6 g, 100%) as a colourless oil, $[\alpha]_D$ +4.1 (c = 6) (Found: M⁺ + H, 283.1401. C₁₁H₂₇O₄SSi requires *M*, 283.1399); v_{max} /cm⁻¹ 2956, 1473, 1358, 1255, 1177, 1094, 1044, 966, 839 and 779; *m/z* (CI, NH₃) 300 (M⁺ + NH₄, 8%), 283 (M⁺ + 1, 100) and 264 (60).

The mesylate (36.49 g, 129.4 mmol) and sodium iodide (58.2 g, 388.2 mmol) in acetone (500 ml) was heated under reflux for 18 h. Following concentration under reduced pressure, the residue was diluted with ethyl acetate and the solution washed with aqueous sodium thiosulfate (5% w/v) and brine then dried (MgSO₄). Concentration under reduced pressure and chromatography of the residue using petrol : ether (3 : 1) as eluent gave the title compound **135** (28.46 g, 70%) as a colourless oil, $[\alpha]_D$ +5.6 (c = 7.2, CH₂Cl₂) (Found: M⁺ + H, 315.0643. C₁₀H₂₄IOSi requires *M*, 315.0641); v_{max} /cm⁻¹ 2956, 2929, 2857, 1471, 1257, 1196, 1103, 839 and 777; δ_H (300 MHz, CDCl₃) 0.10 (6 H, s, 2 × CH₃Si), 0.92 [9 H, s, SiC(CH₃)₃], 0.98 (3 H, d, *J* 7, 2-CH₃), 1.65 (1 H, m, 2-H), 3.30 (2 H, m, 1-H₂), 3.42 (1 H, dd, *J* 10, 7, 3-H) and 3.55 (1 H, dd, *J* 10, 5, 3-H'); *m/z* (CI, NH₃) 315 (M⁺ + 1, 100%).

(5*R*)-6-*tert*-Butyldimethylsilyloxy-3-(*N*-methyl-2-imidazol-2ylthio)-5-methylhex-1-ene 136

Butyllithium in hexanes (1.6 M, 22 ml, 34.98 mmol) was added to the propenyl sulfide **133** (4.90 g, 31.8 mmol) and hexamethylphosphoric triamide (11.4 ml, 63.6 mmol) in THF (150 ml) at -78 °C and the mixture stirred for 30 min. The iodide **133** (10 g, 31.8 mmol) in THF (50 ml) was added and the mixture stirred at -78 °C for 2.5 h. Saturated aqueous ammonium chloride (50 ml) was added and the reaction warmed to ambient temperature then diluted with ether and washed with water and brine then dried (MgSO₄). After concentration under reduced pressure, chromatography of the residue using petrol: ethyl acetate (3:1) gave the title compound **136** (4.51 g, 42%) as a colourless oil, $[\alpha]_D$ +0.4 (c = 5.6) (Found: M⁺, 340.2009. C₁₇H₃₂N₂OSSi requires M, 340.2005); v_{max}/cm^{-1} 2955, 2930, 1460, 1280, 1254, 1094, 838 and 776; δ_H (300 MHz, CDCl₃) 0.05 (6 H, s, 2 × CH₃Si), 0.90 [9 H, s, SiC(CH₃)₃], 0.96 (3 H, d, J 7, 5-CH₃), 1.45 (1 H, m, 5-H), 1.85 (2 H, m, 4-H₂), 3.44 (2 H, m, 6-H₂), 3.68 (3 H, s, NCH₃), 3.90 (1 H, m, 3-H), 4.85 (2 H, m, 1-H₂), 5.72 (1 H, m, 2-H), 6.90 (1 H, d, J 1, ArH) and 7.10 (1 H, d, J 1, ArH); δ_C (75 MHz, CDCl₃) -5.4, -5.3, 16.0, 17.1, 18.3, 25.9, 33.6, 33.7, 33.8, 37.3, 37.7, 51.9, 51.9, 67.6, 68.1, 115.6, 116.2, 122.6, 129.7, 138.6, 139.6, 139.9 and 140.0; m/z (EI) 340 (M⁺, 5%), 95 (100); m/z (CI, NH₃) 341 (M⁺ + 1, 100%).

(5*R*)-6-*tert*-Butyldimethylsilyloxy-5-methylhex-2enyl(tributyl)stannane 137

A solution of the hex-2-en-3-yl sulfide **136** (3.42 g, 10.05 mmol), tributyltin hydride (5.4 ml, 20.1 mmol) and azobis-*iso*-butyronitrile (150 mg, 5% w/w) in degassed benzene (100 ml) was heated under reflux for 18 h. After concentration under reduced pressure, chromatography of the residue using petrol containing 0.5% triethylamine as eluent gave the title compound **137** (4.11 g, 79%) as a colourless oil, $[\alpha]_D$ +0.9 (c = 3.7, CH₂Cl₂) (Found: M⁺ – C₄H₉, 461.2262. C₂₁H₄₅OSi¹²⁰Sn requires *M*, 461.2262); v_{max} /cm⁻¹ 2956, 2928, 2856, 1463, 1255, 1094, 838 and 775; δ_H (300 MHz, CDCl₃) 0.10 (6 H, s, 2 × CH₃Si), 0.80–1.00 [27 H, m, 5-CH₃, SiC(CH₃)₃, 3 × SnCH₂CH₂CH₂CH₃), 1.30–1.80 (16 H, m, 5-H, 1-H₂, 4-H, 3 × SnCH₂CH₂CH₂CH₃), 2.12 (1 H, m, 4-H'), 3.45 (2 H, m, 6-H₂) 5.20 (1 H, m, 3-H) and 5.52 (1 H, m, 2-H); *m/z* (EI) 461 (M⁺ – 57, 37%), 459 (M⁺ – 57, 30%), 291 (40), 289 (30) and 75 (100).

(5R)-6-Hydroxy-5-methylhex-2-enyl(tributyl)stannane 138

Tetrabutylammonium fluoride in THF (1 M, 15.9 ml, 15.9 mmol) was added to the stannane **137** (4.11 g, 7.93 mmol) in THF (16 ml) and the solution stirred for 18 h. After concentration under reduced pressure, chromatography of the residue using petrol: ether (5:1 containing 1% triethylamine) gave the title compound **138** (2.54 g, 79%) as a colourless oil, a 2:1 mixture of (*E*)- and (*Z*)-isomers (Found: M⁺ – C₄H₉, 347.1400. C₁₅H₃₁O¹²⁰Sn requires *M*, 347.1397); v_{max}/cm^{-1} 3334, 2956, 2925, 1462, 1377, 1038, 960 and 874; $\delta_{\rm H}$ (300 MHz, CDCl₃) 0.85–1.00 (18 H, m, 5-CH₃, 3 × SnCH₂CH₂CH₂CH₂CH₃), 1.30–1.65 (13 H, m, 5-H, 3 × SnCH₂CH₂CH₂CH₃), 1.72 (2 H, d, *J* 8.5, 1-H₂), 1.80–2.15 (2 H, m, 4-H₂), 3.50 (2 H, m, 6-H₂), 5.15 (0.33 H, m, 2-H), 5.25 (0.67 H, dt, *J* 15, 7.5, 2-H) and 5.60 (1 H, m, 3-H); *m/z* (EI) 347 (M⁺ – 57, 36%) 345 (M⁺ – 57, 30), 291 (55), 289 (40) and 177 (100).

(1R,6R,3Z)-6-Methyl-1-phenylhept-3-ene-1,7-diol 141

Following the general procedure, tin(IV) bromide (1 M in DCM, 230 µl, 0.23 mmol), stannane **138** (92 mg, 0.23 mmol) in DCM (2 ml) and benzaldehyde in DCM (3 M, 76 µl, 0.23 mmol), after chromatography using ether : petrol (2 : 1 + 1% triethylamine) as eluent, gave the title compound **141** (42 mg, 84%) as a colourless oil, a 93 : 7 mixture of epimers, $[\alpha]_D$ +60 (c = 3.2) (Found: M⁺ + NH₄, 238.1809. C₁₄H₂₄NO₂ requires *M*, 238.1807); v_{max}/cm^{-1} 3346, 2955, 1494, 1454, 1035, 876, 759 and 701; δ_H (500 MHz, CDCl₃) major epimer **141** 0.87 (3 H, d, *J* 7, 6-CH₃), 1.63 (1 H, m, 6-H),

1.88 and 2.18 (each 1 H, dt, *J* 13, 7, 5-H), 2.35 (1 H, dt, *J* 15, 4, 2-H), 2.58 (1 H, dt, *J* 14, 8, 2-H'), 2.67 and 3.06 (each 1 H, br s, OH), 3.36 (1 H, dd, *J* 12, 4.5, 7-H), 3.46 (1 H, dd, *J* 11, 6, 7-H'), 4.65 (1 H, dd, *J* 9, 4, 1-H), 5.40–5.55 (2 H, m, 3-H and 4-H) and 7.22–7.32 (5 H, m, ArH); minor 1,6-*syn*-epimer **142** 2.05 (2 H, t, *J* 7, 5-H₂) and 4.69 (1 H, dd, *J* 8,5, 1-H); $\delta_{\rm C}$ (75 MHz, CDCl₃) 17.0, 31.0, 35.7, 37.4, 66.8, 73.8, 125.8, 126.4, 127.5, 128.4, 131.3 and 144.4; *m/z* (CI/NH₃) 238 (M⁺ + 18, 5%), 221 (M⁺ + 1, 6), 220 (M⁺, 22) and 203 (M⁺ – 17, 100).

(1S,6R,3Z)-6-Methyl-1-phenylhept-3-ene-1,7-diol 142

TBAF in THF (1 M, 134 µl, 0.134 mmol) was added to the silyl ether **148** (30 mg, 0.09 mmol) in THF (1 ml) and the solution stirred for 18 h. After concentration under reduced pressure, chromatography of the residue using ether : petrol (2 : 1) as eluent, gave the title compound **142** (17 mg, 88%) as a colourless oil, $[\alpha]_D$ –48 (c = 1.7) (Found: M⁺ + NH₄, 238.1809. C₁₄H₂₄NO₂ requires M, 238.1807); v_{max}/cm^{-1} 3333, 2960, 1454, 1261, 1030, 799, 759 and 700; δ_H (500 MHz, CDCl₃) 0.88 (3 H, d, *J* 7, 6-CH₃), 1.68 (1 H, oct, *J* 7, 6-H), 2.05 (2 H, t, *J* 7, 5-H₂), 2.46 (1 H, dt, *J* 13.5, 5.5, 2-H), 2.75 (1 H, dt, *J* 13.5, 7, 2-H'), 3.40 (2 H, m, 7-H₂), 4.71 (1 H, dd, *J* 8, 4.5, 1-H), 5.40–5.60 (2 H, m, 3-H and 4-H) and 7.20–7.35 (5 H, m, ArH); δ_C (75 MHz, CDCl₃) 16.4, 30.6, 35.6, 37.1, 67.0, 73.8, 125.8, 126.3, 127.5, 128.4, 130.7 and 144.2; *m/z* (CI, NH₃) 238 (M⁺ + 18, 18%), 221 (M⁺ + 1, 27), 220 (M⁺, 100) and 203 (M⁺ – 17, 98).

(1*R*,6*R*,3*Z*)-7-*tert*-Butyldimethylsilyloxy-6-methyl-1-phenylhept-3-en-1-ol 143

tert-Butyldimethylsilyl chloride (101 mg, 0.673 mmol) was added to the alcohol 141 (148 mg, 0.673 mmol), triethylamine (103 ml, 0.74 mmol) and 4-N,N-dimethylaminopyridine (5 mg) in DCM (3 ml). The mixture was stirred for 18 h, diluted with DCM and washed with water and brine then dried (MgSO₄). After concentration under reduced pressure, chromatography of the residue using petrol: ether (5:1) as eluent, gave the title compound 143 (176 mg, 92%) as a colourless oil, $[\alpha]_{\rm D}$ +32 (c = 2.0) (Found: M^+ + H, 335.2374. $C_{20}H_{35}O_2Si$ requires *M*, 335.2406); v_{max} /cm⁻¹ 3358, 2955, 2929, 1471, 1389, 1255, 1091, 837, 776 and 700; $\delta_{\rm H}$ (300 MHz, CDCl₃) 0.10 (6 H, s, 2 × CH₃Si), 0.87 (3 H, d, J 6.5, 6-CH₃), 0.94 [9 H, s, (CH₃)₃CSi], 1.67 (1 H, m, 6-H), 1.90 (1 H, dt, J 14, 7, 5-H), 2.15 (1 H, d, J 3, OH), 2.20 (1 H, m, 5-H'), 2.55 (2 H, m, 2-H₂), 3.45 (2 H, m, 7-H₂), 4.75 (1 H, m, 1-H), 5.55 (2 H, m, 3-H and 4-H) and 7.30–7.45 (5 H, m, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃) -5.3, 16.5, 18.4, 26.0, 31.0, 36.3, 37.4, 67.9, 73.9, 125.7, 125.9, 127.5, 128.4, 131.9 and 144.2; *m/z* (CI, NH₃) 335 (M⁺ + 1, 5%), 334 (M⁺, 2) and 317 (M⁺ – 17, 100).

(1*R*,6*R*,3*Z*)-1-[(*R*)-2-Acetoxy-2-phenylacetoxy]-7-*tert*-butyldimethylsilyloxy-6-methyl-1-phenylhept-3-ene 144

Following the general procedure, alcohol **143** (27 mg, 0.081 mmol), after chromatography using petrol : ether (5 : 1) as eluent, gave the title compound **144** (37 mg, 90%) as a colourless oil, $[\alpha]_D$ –18 (c = 3.7) (Found: M⁺ + NH₄, 528.3135. C₃₀H₄₆NO₅Si requires *M*, 528.3145); $v_{\text{max}}/\text{cm}^{-1}$ 2955, 2929, 1751, 1372, 1207, 1175, 1085, 1059, 838, 776 and 698; δ_{H} (300 MHz, CDCl₃) 0.05 (6 H, s, 2 × CH₃Si), 0.83 (3 H, d, *J* 7, 6-CH₃), 0.92 [9 H, s, SiC(CH₃)₃], 1.60

(1 H, m, 6-H), 1.78 (1 H, dt, J 14.5, 7.5, 5-H), 2.10 (1 H, dt, J 13, 7, 5-H'), 2.22 (3 H, s, CH₃CO₂), 2.60 (2 H, m, 2-H₂), 3.40 (2 H, dd, J 6, 1.5, 7-H₂), 5.45 (2 H, m, 3-H and 4-H), 5.76 (1 H, t, J 7, 1-H), 6.01 (1 H, s, 2'-H) and 6.95–7.45 (10 H, m, ArH); $\delta_{\rm c}$ (75 MHz, CDCl₃) –5.3, 16.4, 18.3, 20.7, 25.9, 30.8, 34.3, 36.2, 67.8, 74.5, 77.2, 124.3, 126.1, 127.7, 127.8, 127.9, 128.7, 129.2, 131.5, 133.7, 139.5, 167.9 and 170.3; m/z (CI, NH₃) 528 (M⁺ + 18, 5%) and 317 (100).

(1*R*,6*R*,3*Z*)-1-[(*S*)-2-Acetoxy-2-phenylacetoxy]-7-*tert*-butyldimethylsilyloxy-6-methyl-1-phenylhept-3-ene 145

Following the general procedure, alcohol **141** (27 mg, 0.081 mmol), after chromatography using petrol : ether (5 : 1) as eluent, gave the title compound **145** (33 mg, 80%) as a colourless oil, $[\alpha]_D$ +54 (c = 3.3) (Found: M⁺ + NH₄, 528.3113. C₃₀H₄₆NO₅Si requires M, 528.3145); v_{max}/cm^{-1} 2956, 1748, 1373, 1232, 1176, 1084, 1060, 838, 758 and 698; δ_H (300 MHz, CDCl₃) 0.05 (6 H, s, 2 × CH₃Si), 0.78 (3 H, d, J 6.5, 6-CH₃), 0.93 [9 H, s, SiC(CH₃)₃], 1.55 (1 H, oct, J 6.5, 6-H), 1.70 and 1.95 (each 1 H, dt, J 14, 6.5, 5-H), 2.20 (3 H, s, CH₃CO₂), 2.55 (2 H, m, 2-H₂), 3.36 (2 H, d, J 6.7, 1-H), 6.05 (1 H, s, 2'-H) and 7.30–7.55 (10 H, m, ArH); δ_C (75 MHz, CDCl₃) –5.3, 16.4, 18.3, 20.7, 26.0, 29.7, 30.7, 34.1, 36.1, 67.7, 74.6, 124.2, 126.3, 127.7, 128.0, 128.4, 128.7, 129.2, 131.3, 134.0, 139.5, 168.2 and 170.2; m/z (CI, NH₃) 528 (M⁺ + 18, 8%) and 317 (100).

(1R,6R,3Z)-1,7-diacetoxy-6-methyl-1-phenylhept-3-ene 146

Acetic anhydride (195 µl, 2.07 mmol) was added to the diol 141 (114 mg, 0.52 mmol), triethylamine (72 μ l, 5.2 mmol) and 4-N,N-dimethylaminopyridine (10 mg) in DCM (5 ml) and the solution stirred for 18 h. Water was added, the aqueous phase was extracted with DCM and the organic extracts were washed with brine and dried (MgSO₄). After concentration under reduced pressure, chromatography using petrol: ether (4:1) as eluent gave the title compound 146 (141 mg, 90%) as a colourless oil, $[\alpha]_{\rm D}$ +27 (c = 3.4); (Found: M⁺ + NH₄, 322.2032. C₁₈H₂₈NO₄ requires M, 322.2018); $v_{\rm max}$ /cm⁻¹ 2962, 1739, 1373, 1237, 1035 and 701; $\delta_{\rm H}$ (300 MHz, CDCl₃) 0.90 (3 H, d, J 7, 6-CH₃), 1.85 (2 H, m, 5-H₂), 2.07 and 2.11 (each 3 H, s, CH₃CO₂), 2.12 (1 H, m, 6-H), 2.55 and 2.70 (each 1 H, dt, J 14.5, 7.5, 2-H), 3.87 and 3.94 (each 1 H, dd, J 11, 6, 7-H), 5.35–5.55 (2 H, m, 3-H and 4-H), 5.79 (1 H, t, J 7, 1-H) and 7.30–7.40 (5 H, m, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃) 16.6, 21.0, 21.3, 31.0, 32.9, 34.4, 68.7, 75.5, 125.6, 126.5, 128.0, 128.4, 130.2, 140.2, 170.3 and 171.2; m/z (CI, NH₃) 322 (M⁺ + 18, 18%) and 245 (100).

Following the procedure outlined for the ozonolysis of acetate **11**, bis-acetate **146** (106 mg, 0.349 mmol), after chromatography using petrol:ether (5:2) as eluent, gave (*R*)-3acetoxy-1-phenylpropan-1-ol (*R*)-**15** (11 mg, 16%) as a colourless oil, $[\alpha]_D$ +31 (c = 1), and a mixture of (*R*)-3-acetoxy-3phenylpropan-1-ol (*R*)-**14** and (*R*)-4-acetoxy-3-methylbutan-1-ol, which were separated by reverse phase semi-preparative HPLC using acetonitrile:water (1:1, flow rate 15 ml min⁻¹) as eluent to give (*R*)-3-acetoxy-3-phenylpropan-1-ol (*R*)-**14**^{1a} (23 mg, 34%) as a colourless oil, $[\alpha]_D$ +73 (c = 2), and (*R*)-4-acetoxy-3-methylbutan-1-ol (5 mg, 10%) as a colourless oil, $[\alpha]_D$ –3.5 (c=0.5) (Found: M⁺ + NH₄, 164.1285. C₇H₁₈NO₃ requires *M*, 164.1287); *v*_{max}/cm⁻¹ 3410, 2917, 1737, 1245, 1037 and 756; $\delta_{\rm H}$ (300 MHz, CDCl₃) 1.02 (3 H, d, *J* 7, 3-CH₃), 1.40–1.75 (3 H, m, OH, 2-H₂), 2.00 (1 H, oct, *J* 7, 3-H), 2.10 (3 H, s, CH₃CO₂), 3.75 (2 H, m, 1-H₂) and 3.97 (2 H, m, 4-H₂); *m/z* (CI/NH₃) 164 (M⁺ + 18, 100%) and 147 (M⁺ + 1, 25).

(1S,6R,3Z)-7-tert-Butyl
dimethylsilyloxy-6-methyl-1-phenylhept-3-en-1-ol 148

Following the procedure outlined for the synthesis of 4nitrobenzoate **86**, alcohol **143** (98 mg, 0.293 mmol), after chromatography using petrol : ether (30 : 1) as eluent, gave the inverted 4-nitrobenzoate **147** (97 mg, 69%) as a colourless oil, $[\alpha]_D$ +19.5 (c = 1.7); v_{max}/cm^{-1} 2955, 1728, 1530, 1344, 1272, 1102, 838, 777 and 720; δ_H (300 MHz, CDCl₃) 0.05 (6 H, s, 2 × CH₃Si), 0.86 (3 H, d, *J* 7, 6-CH₃), 0.92 [9 H, s, (CH₃)₃CSi], 1.60 (1 H, m, 6-H), 1.86 and 2.20 (each 1 H, dd, *J* 14, 7, 5-H), 2.72 and 2.90 (each 1 H, dt, *J* 15, 7.5, 2-H), 3.41 (2 H, d, *J* 6, 7-H₂), 5.50 (2 H, m, 3-H and 4-H), 6.06 (1 H, t, *J* 7, 1-H), 7.30–7.50 (5 H, m, ArH) and 8.30 (4 H, m, ArH); δ_C (75 MHz, CDCl₃) –5.3, 16.5, 18.3, 25.9, 30.9, 34.4, 36.2, 67.6, 123.5, 124.4, 126.6, 128.3, 128.6, 130.8, 131.9, 135.9, 139.6, 150.6 and 163.9.

Following the procedure outlined for the synthesis of alcohol **81**, nitrobenzoate **147** (75 mg, 0.157 mmol) gave the title compound **148** (38 mg, 73%) as a colourless oil, $[\alpha]_D$ –16 (c = 3.7) (Found: M⁺ + H, 335.2404. C₂₀H₃₅O₂Si requires *M*, 335.2406); v_{max}/cm^{-1} 3385, 2955, 2929, 1471, 1255, 1091, 837, 776 and 700; δ_H (300 MHz, CDCl₃) 0.05 (6 H, s, 2 × CH₃Si), 0.90 (3 H, d, *J* 7, 6-CH₃), 0.94 [9 H, s, (CH₃)₃CSi], 1.68 (1 H, oct, *J* 6.5, 6-H), 1.90 (1 H, dt, *J* 15, 7.5, 5-H), 2.20 (2 H, m, 5-H' and OH), 2.55 (2 H, m, 2-H₂), 3.45 (2 H, m, 7-H₂), 4.75 (1 H, dd, *J* 7.5, 5.5, 1-H), 5.55 (2 H, m, 3-H and 4-H) and 7.30–7.45 (5 H, m, ArH); δ_C (75 MHz, CDCl₃) –5.3, 16.5, 18.4, 26.0, 30.9, 36.3, 37.4, 67.8, 73.9, 125.8, 125.9, 127.5, 128.4, 131.8 and 144.2; m/z (CI, NH₃) 335 (M⁺ + 1, 8%), 334 (M⁺, 2) and 317 (M⁺ – 17, 100).

(3RS,6R)-6-tert-Butyldimethylsilyloxy-3-(N-methylimidazol-2ylthio)hept-1-ene 161

Butyllithium in hexanes (1.6 M, 19.6 ml, 31.36 mmol) was added to the propenyl sulfide 133 (4.43 g, 28.77 mmol) in THF (150 ml) at -78 °C. After 20 min, the 3-tert-butyldimethylsilyloxy-1-iodobutane 160 (9.04 g, 28.79 mmol) in THF (50 ml) was added and the solution maintained at -78 °C for 2.5 h. Saturated aqueous ammonium chloride (20 ml) was added and the mixture was allowed to warm to room temperature. Water (150 ml) and ether (200 ml) were added, the aqueous phase was extracted with ether (100 ml) and the organic extracts were washed with brine (250 ml), dried (MgSO₄) and concentrated under reduced pressure. Chromatography using petrol:ether (1:3) as eluent afforded the title compound 161 (8.09 g, 83%) as a colourless oil, a mixture of epimers, $[\alpha]_D$ +0.6 (c = 0.86) (Found: M⁺ + H, 341.2080. C₁₇H₃₃N₂OSSi requires *M*, 341.2083); *v*_{max}/cm⁻¹ 2955, 2930, 2857, 1458, 1255, 1131, 1078, 1047, 1005, 916, 836 and 775; $\delta_{\rm H}$ (300 MHz, CDCl₃) 0.05 (1.5 H, s, SiCH₃), 0.07 (4.5 H, s, 3 × SiCH₃), 0.90 [9 H, s, SiC(CH₃)₃], 1.14 (3 H, d, J 6, 7-H₃), 1.45–1.90 (4 H, m, 4-H₂ and 5-H₂), 3.69 (3 H, s, NCH₃), 3.82 (2 H, m, 3-H and 6-H), 4.91 (2 H, m, 1-H₂), 5.76 (1 H, m, 2-H) and 6.95 and 7.11 (each 1 H, s, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃) -4.7, -4.4, 18.1, 23.7, 23.9, 25.9, 30.3, 30.3, 33.8, 37.0, 37.1, 53.8, 53.8, 68.1, 68.2, 116.3, 122.6, 129.6, 138.7 and 140.1; *m*/*z* (CI, NH₃) 341 (M⁺ + 1, 100%), 301 (14) and 115 (42).

(R)-6-tert-Butyldimethylsilyloxyhept-2-enyl(tributyl)stannane 162

Tributyltin hydride (13.85 g, 47.59 mmol) and azo-bis-isobutyronitrile (390 mg, 2.38 mmol) were added to a degassed solution of the alkenyl sulfide 161 (8.09 g, 23.79 mmol) in benzene (240 ml) and the solution heated under reflux for 1.5 h. After concentration under reduced pressure, chromatography of the residue using petrol (+ 1% triethylamine) as eluent gave the title compound 162 (11.27 g, 92%) as a colourless oil, a 2:1 mixture of (*E*)- and (*Z*)-isomers, $[\alpha]_{D}$ -7.1 (*c* = 1.19) (Found: M⁺ - C₄H₉, 461.2259. $C_{21}H_{45}OSi^{120}Sn$ requires *M*, 461.2262; v_{max}/cm^{-1} 2957, 2928, 2856, 1463, 1254, 1133, 1083, 1035, 960, 836 and 774; $\delta_{\rm H}$ (300 MHz, CDCl₃) 0.06 (6 H, s, 2 × SiCH₃), 0.82–0.97 [24 H, m, 3×SnCH₂CH₂CH₂CH₃ and SiC(CH₃)₃], 1.16 (2 H, d, J 6.5, 7-H₃), 1.18 (1 H, d, J 6, 7-H₃), 1.25–1.80 (16 H, m, 1-H₂, 5-H₂ and $3 \times$ SnCH₂CH₂CH₂CH₃), 2.02 (2 H, m, 4-H₂), 3.81 (1 H, m, 6-H) and 4.95–5.65 (2 H, m, 2-H and 3-H); $\delta_{\rm C}$ (75 MHz, CDCl₃) –4.7, –4.3, 9.1, 9.3, 10.5, 11.2, 13.8, 14.1, 18.2, 23.5, 23.8, 25.9, 27.4, 29.0, 29.2, 39.2, 40.3, 68.3, 68.5, 124.0, 125.2, 128.3 and 129.1; *m/z* (CI, NH₃) 461 (M⁺ - 57, 32%), 459 (M⁺ - 57, 28), 308 (100) and 306 (65).

(R)-6-Hydroxyhept-2-enyl(tributyl)stannane 163

TBAF in THF (1 M, 10.3 ml, 10.30 mmol) was added to the silvl ether 162 (1.77 g, 3.42 mmol) and the solution stirred at room temperature for 50 h. Water (10 ml) was added, the mixture was stirred for 1 h, then extracted with ether $(4 \times 15 \text{ ml})$. The organic extracts were washed with brine (50 ml), dried (MgSO₄) and concentrated under reduced pressure. Chromatography of the residue using petrol : ether (2:1 + 1% triethylamine) as eluent gave the title compound 163 (1.10 g, 80%) as a colourless oil, a 2:1 mixture of (*E*)- and (*Z*)-isomers, $[\alpha]_D$ –3.4 (*c* = 1.08) (Found: M⁺ $-C_4H_9$, 347.1397. $C_{15}H_{31}O^{120}Sn$ requires M, 347.1397); v_{max}/cm^{-1} 3331, 2957, 2925, 2872, 1463, 1376, 1075 and 960; $\delta_{\rm H}$ (300 Mz, $CDCl_3$) 0.85–1.00 (15 H, m, 3 × $SnCH_2CH_2CH_2CH_3$), 1.21 (2 H, d, J 6, 7-H₃), 1.24 (1 H, d, J 6, 7-H₃), 1.26–1.80 (16 H, m, 3 × SnCH₂CH₂CH₂CH₃, 1-H₂ and 4-H₂), 2.10 (2 H, m, 5-H₂), 3.85 (1 H, m, 6-H), 5.11 (0.33 H, dt, J 10.5, 7, 2- or 3-H), 5.25 (0.67 H, dt, J 15, 7, 2- or 3-H) and 5.60 (1 H, m, 2- or 3-H); $\delta_{\rm C}$ (75 MHz, CDCl₃) 9.1, 9.4, 10.5, 11.2, 13.8, 14.2, 23.4, 27.4, 29.2, 39.2, 39.6, 67.9, 68.2, 123.5, 124.9, 129.1 and 129.9; *m/z* (EI) 347 (M⁺ - 57, 72%), 345 (M^+ – 57, 55), 291 (39), 289 (40) and 233 (15).

(1S,7R,3Z)-1-Phenyloct-3-ene-1,7-diol 164

Following the general procedure, stannane **163** (500 mg, 1.24 mmol), tin(IV) bromide (545 mg, 1.24 mmol) in DCM (1 ml) and benzaldehyde (132 mg, 1.24 mmol) in DCM (0.5 ml), after chromatography using petrol: ether (1:2 + 1% triethylamine) as eluent gave the title compound **164** (196 mg, 72%) as a colourless oil, a 92:8 mixture of epimers, $[\alpha]_D -71$ (c = 0.82) (Found: M+, 220.1456. C₁₄H₂₀O₂ requires *M*, 220.1463); v_{max}/cm^{-1} 3348, 2965, 2925, 1453, 1129, 1081, 1053, 759 and 701; δ_H (500 MHz, CDCl₃) 1.12 (3 H, d, *J* 6, 8-H₃), 1.41 (2 H, m, 6-H₂), 2.10 (4 H, m, 2- or 5-H₂ and 2×OH), 2.51 (2 H, t, *J* 7, 2- or 5-H₂), 3.73 (1 H, m, 7-H),

4.72 (1 H, t, J 6, 1-H), 5.35 and 5.51 (each 1 H, dt, J 10.5, 7, 3-H or 4-H) and 7.24–7.32 (5 H, m, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃) major epimer **164** 23.6, 23.9, 37.1, 38.6, 67.5, 73.7, 125.1, 125.9, 127.4, 128.4, 133.0 and 144.1; minor 1,7-*anti*-epimer 37.4 and 38.4; *m/z* (CI, NH₃) 221 (M⁺ + 1, 35%), 204 (100) and 186 (85).

(1R,7R,3Z)-1-Phenyloct-3-ene-1,7-diol 165

TBAF in THF (1 M, 0.74 ml, 0.74 mmol) was added to the silvl ether 171 (82 mg, 0.246 mmol) in THF (0.3 ml) and the solution stirred at room temperature for 72 h. Water (2 ml) was added and the mixture stirred for 1 h, then extracted with ether $(4 \times 5 \text{ ml})$. The organic extracts were washed with brine (10 ml), dried (MgSO₄) and concentrated under reduced pressure. Chromatography of the residue using petrol: ethyl acetate (1:1) as eluent gave the title compound 165 (52 mg, 96%) as a colourless oil, $[\alpha]_{\rm D}$ +58 (c = 2.22) (Found: M⁺, 220.1462. C₁₄H₂₀O₂ requires M, 220.1463); $v_{\rm max}/{\rm cm}^{-1}$ 3358, 2965, 2924, 1453, 1130, 1081, 1054, 759 and 700; δ_H (500 MHz, CDCl₃) 1.11 (3 H, d, J 6.5, 8-H₃), 1.41 (2 H, m, 6-H₂), 2.00 and 2.24 (each 1 H, m, 2- or 5-H), 2.34 (1 H, dt, J 14, 5, 2- or 5-H), 2.58 (1 H, dt, J 14, 8, 2- or 5-H), 2.82 (2 H, br s, 2 × OH), 3.72 (1 H, m, 7-H), 4.63 (1 H, dd, J 8.5, 4, 1-H), 5.44 (2 H, m, 3-H and 4-H) and 7.21–7.34 (5 H, m, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃) major 1,6-anti-epimer 165 23.5, 37.4, 38.3, 66.6, 73.7, 125.6, 125.7, 127.4, 128.2, 132.3 and 144.3; minor 1,7-syn-epimer 164 23.6, 37.1 and 38.6; m/z (CI, NH₃) 238 (M⁺ + 18, 12%), 220 (M⁺, 38), 203 (M⁺ - 17, 100) and 185 (87).

(1*S*,7*R*,3*Z*)-7-*tert*-Butyldimethylsilyloxy-1-phenyloct-3-en-1-ol 166, (2*R*,8*S*,5*Z*)-8-*tert*-butyldimethylsilyloxy-8-phenyloct-5-en-2-ol 167 and (1*S*,7*R*,3*Z*)-1,7-bis-(*tert*-butyldimethylsilyloxy)-1-phenyloct-3-ene 168

Imidazole (78 mg, 1.15 mmol) and tert-butyldimethylsilyl chloride (173 mg, 1.15 mmol) were added to the diol 164 (230 mg, 1.045 mmol) in N,N-dimethylformamide (1 ml) and the mixture stirred at room temperature for 15 h. Water (2 ml) was added and the mixture extracted with ether $(4 \times 10 \text{ ml})$. The organic extracts were washed with brine (30 ml), dried (MgSO₄) and concentrated under reduced pressure. Chromatography of the residue using petrol: ether (7:1 to 1:3) as eluent afforded the title compound 168 (72 mg, 15%) as a colourless oil, $[\alpha]_{\rm D}$ -25 (c = 1.12) (Found: M⁺ – H, 447.3117. C₂₆H₄₇O₂Si₂ requires M, 447.3115); v_{max}/cm⁻¹ 2957, 2929, 2858, 1472, 1382, 1255, 1137, 1091, 836 and 775; $\delta_{\rm H}$ (300 MHz, CDCl₃) -0.07, 0.07, 0.08 and 0.09 (each 3 H, s, SiCH₃), 0.92 and 0.93 [each 9 H, s, SiC(CH₃)₃], 1.13 (3 H, d, J 6, 8-H₃), 1.25-1.55 (2 H, m, 6-H₂), 1.85-2.18 and 2.45 (each 2 H, m, 2-H₂ or 5-H₂), 3.78 (1 H, m, 7-H), 4.70 (1 H, t, J 6, 1-H), 5.44 (2 H, m, 3-H and 4-H) and 7.21-7.36 (5 H, m, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃) -4.9, -4.7, -4.3, 18.2, 18.3, 23.7, 23.8, 25.9, 25.9, 38.9, 39.6, 68.3, 75.1, 125.7, 126.0, 126.7, 127.9, 131.5 and 145.3; m/z (FAB, Xe) 447 (M⁺ – 1, 1%), 433 (2), 315 (6) and 221 (100). The second fraction contained the title compound 166 (219 mg, 63%) as a colourless oil, $[\alpha]_{\rm D}$ -61 (c = 0.88) (Found: M⁺ – OH, 317.2295. C₂₀H₃₃OSi requires M, 317.2301); v_{max} /cm⁻¹ 3365, 2957, 2929, 2857, 1255, 1137, 1091, 1051, 1005, 836, 775 and 700; $\delta_{\rm H}$ (500 MHz, CDCl₃) 0.09 (6 H, s, 2×SiCH₃), 0.94 [9 H, s, SiC(CH₃)₃], 1.15 (3 H, d, J 6, 8-H₃), 1.43 (2 H, m, 6-H₂), 1.95–2.25 (2 H, m, 2- or 5-H₂), 1.98 (1 H, d, J 2.5,

OH), 2.57 (2 H, m, 2- or 5-H₂), 3.82 (1 H, m, 7-H), 4.74 (1 H, ddd, J 7.5, 5.5, 2.5, 1-H), 5.43 and 5.60 (each 1 H, m, 3-H or 4-H) and 7.28–7.44 (5 H, m, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃) –4.7, –4.3, 18.2, 23.7, 23.8, 25.9, 37.3, 39.5, 68.3, 73.9, 124.7, 125.9, 127.5, 128.4, 133.4 and 144.1; m/z (CI, NH₃) 317 (M⁺ - 17, 78%), 221 (70), 185 (100) and 143 (75). The third fraction contained the title compound 167 (18 mg, 5%) as a colourless oil, $[\alpha]_{\rm D}$ -29 (c = 0.52) (Found: M⁺ – OH, 317.2310. C₂₀H₃₃OSi requires M, 317.2301); $v_{\rm max}$ /cm⁻¹ 3333, 2957, 2929, 2857, 1255, 1088, 1070, 836, 777 and 700; $\delta_{\rm H}$ (300 MHz, CDCl₃) –0.08 and 0.06 (each 3 H, s, SiCH₃), 0.93 [9 H, s, SiC(CH₃)₃], 1.18 (3 H, d, J 6, 8-H₃), 1.28 (1 H, d, J 5, OH), 1.40 (2 H, m, 6-H₂), 2.04 and 2.49 (each 2 H, m, 2-H₂ or 5-H₂), 3.73 (1 H, m, 7-H), 4.71 (1 H, t, J 6, 1-H), 5.46 (2 H, m, 3-H and 4-H) and 7.20–7.35 (5 H, m, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃) –4.9, -4.7, 18.3, 23.5, 23.7, 25.9, 38.9, 67.7, 75.0, 126.0, 126.2, 126.9, 127.9, 131.1 and 145.3; m/z (CI, NH₃) 335 (M⁺ + 1, 1%), 317 (M⁺ - 17, 8), 220 (75), 203 (85), 185 (60) and 102 (100). The fourth fraction contained recovered starting material 164 (33 mg, 14%).

(1S,7R,3Z)-1,7-Diacetoxy-1-phenyloct-3-ene 169

Triethylamine (500 mg, 4.95 mmol), 4-N,N-dimethylaminopyridine (6 mg, 0.049 mmol) and acetic anhydride (202 mg, 1.98 mmol) were added to the diol 164 (109 mg, 0.495 mmol) in DCM (3 ml) and the solution stirred at room temperature for 15 h. DCM (35 ml) and water (35 ml) were added, and the organic phase washed with aqueous hydrogen chloride (3.5 M, 25 ml), saturated aqueous sodium hydrogen carbonate (25 ml) and brine (15 ml) then dried (MgSO₄). After concentration under reduced pressure, chromatography of the residue using petrol: ether (3:1) as eluent gave the title compound 169 (145 mg, 96%) as a colourless oil, $[\alpha]_{\rm D}$ -39 (c = 1.10) (Found M⁺ + NH₄, 322.2021. $C_{18}H_{28}NO_4$ requires *M*, 332.2018); v_{max}/cm^{-1} 2937, 1736, 1373, 1238, 1024 and 701; $\delta_{\rm H}$ (300 MHz, CDCl₃) 1.22 (3 H, d, J 6.5, 8-H₃), 1.33–1.68 (2 H, m, 6-H₂), 2.01 (2 H, m, 2- or 5-H₂), 2.05 and 2.10 (each 3 H, s, CH₃CO₂), 2.54 and 2.68 (each 1 H, dt, J 14.5, 7, 2- or 5-H), 4.87 (1 H, m, 7-H), 5.36 and 5.47 (each 1 H, dt, J 10.5, 7, 3-H or 4-H), 5.77 (1 H, t, J 6.5, 1-H) and 7.28–7.40 (5 H, m, ArH); δ_C (75 MHz, CDCl₃) 19.9, 21.3, 21.4, 23.4, 34.3, 35.6, 70.5, 75.5, 124.5, 126.6, 127.9, 128.4, 131.9, 140.2, 170.2 and 170.7; *m/z* (CI, NH₃) 322 M⁺ + 18, 1%), 245 (26) and 185 (100).

Ozone was bubbled through a solution of the alkene 169 (107 mg, 0.352 mmol) in DCM (5 ml) at -78 °C for 15 min. The solution was purged with oxygen for 10 min, then dimethyl sulfide (200 mg, 3.23 mmol) was added and the solution allowed to warm to room temperature. After concentration under reduced pressure, the residue was dissolved in DCM (2 ml) and methanol (2 ml) and sodium borohydride (82 mg, 2.16 mmol) were added at 0 °C. After 15 min at room temperature, aqueous hydrogen chloride (1 M, 2 ml) was added at 0 °C. After warming to room temperature DCM (25 ml) and water (25 ml) were added, and the aqueous phase was extracted with DCM (25 ml). The organic extracts were washed with brine (30 ml), dried (MgSO₄) and concentrated under reduced pressure. Chromatography of the residue using petrol: ether (2:1)gave (S)-3-acetoxy-1-phenylpropan-1-ol (S)-15 (9 mg, 13%) as a colourless oil, $[\alpha]_D$ -25 (c = 0.73) and a 3:1 mixture of (S)-3acetoxy-3-phenylpropan-1-ol (S)-14 and (R)-4-acetoxypent-1-ol (58 mg). HPLC of this mixture allowed the isolation of (S)-3acetoxy-3-phenylpropan-1-ol (S)-14 (18 mg, 27%) as a colourless

oil, $[\alpha]_D - 72$ (c = 0.92)^{1a} with spectroscopic data identical to those of a sample prepared earlier.

(1*R*,7*R*,3*Z*)-7-*tert*-Butyldimethylsilyloxy-1-phenyloct-3-en-1-yl 4-nitrobenzoate 170

Diethyl azodicarboxylate (149 mg, 0.856 mmol) was added to a suspension of alcohol 166 (190 mg, 0.569 mmol), triphenylphosphine (223 mg, 0.853 mmol) and 4-nitrobenzoic acid (142 mg, 0.853 mmol) in toluene (4.5 ml) at -35 °C and the mixture was allowed to warm to room temperature. After 2 h, ether (30 ml) and water (30 ml) were added and the organic phase was washed with brine (30 ml) and dried (MgSO₄). Concentration under reduced pressure gave a solid which was absorbed onto silica. Chromatography using petrol: ether (10:1) as eluent afforded the title compound 170 (203 mg, 74%) as a colourless oil, $[\alpha]_{\rm D}$ -30 (c = 1.39) (Found: M⁺ – H, 482.2370. C₂₇H₃₆NO₅Si requires M, 482.2363); v_{max}/cm⁻¹ 2929, 2856, 1727, 1608, 1530, 1344, 1272, 1102, 837, 775 and 720; $\delta_{\rm H}$ (300 MHz, CDCl₃) 0.08 and 0.09 (each 3 H, s, SiCH₃), 0.92 [9 H, s, SiC(CH₃)₃], 1.13 (3 H, d, J 6, 8-H₃), 1.40 (2 H, m, 6-H₂), 1.90–2.25 (2 H, m, 2- or 5-H₂), 2.74 and 2.90 (each 1 H, dt, J 14.5, 7, 2- or 5-H), 3.79 (1 H, m, 7-H), 5.39 and 5.55 (each 1 H, dt, J 10.5, 7.5, 3-H or 4-H), 6.07 (1 H, t, J 7, 1-H), 7.30–7.50 (5 H, m, ArH) and 8.22–8.37 (4 H, m, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃), -4.7, -4.3, 18.1, 23.8, 25.9, 34.3, 39.4, 68.2, 77.4, 123.3, 123.6, 126.6, 128.3, 128.6, 130.8, 133.4, 135.9, 139.6, 150.6 and 163.9; m/z (FAB, Xe) 482 (M⁺ - 1, 2%), 317 (26), 221 (100), 185 (64), 143 (84), 129 (83) and 115 (80).

(1*R*,7*R*,3*Z*)-7-*tert*-Butyldimethylsilyloxy-1-phenyloct-3-en-1-ol 171

Sodium hydroxide (101 mg, 2.53 mmol) was added to the ester 170 (175 mg, 0.362 mmol) in methanol (10 ml) and, after 2 h at room temperature, water (15 ml) was added. The mixture was extracted with ether $(2 \times 20 \text{ ml})$ and the organic extracts were washed with brine (30 ml), dried (MgSO₄) and concentrated under reduced pressure to afford the title compound 171 (112 mg, 93%) as a colourless oil, $[\alpha]_D$ +15 (c = 0.98) (Found: M⁺ – OH, 317.2305. $C_{20}H_{33}$ OSi requires M, 317.2301); v_{max}/cm^{-1} 3359, 2957, 2929, 2857, 1255, 1137, 1091, 1051, 1001, 836, 775 and 700; $\delta_{\rm H}$ (300 MHz, CDCl₃) 0.11 (6 H, s, 2 × Si(CH₃), 0.95 [9 H, s, SiC(CH₃)₃], 1.16 (3 H, d, J 6, 8-H₃), 1.45 (2 H, m, 6-H₂), 1.90-2.25 (3 H, m, 2- or 5-H₂ and OH), 2.55 (2 H, m, 2- or 5-H₂), 3.82 (1 H, m, 7-H), 4.75 (1 H, dd, J 7.5, 5.5, 1-H), 5.43 and 5.61 (each 1 H, m, 3-H or 4-H) and 7.26–7.42 (5 H, m, ArH); $\delta_{\rm C}$ (75 MHz, CDCl₃) -4.7, -4.3, 18.2, 23.7, 23.8, 25.9, 37.4, 39.5, 68.3, 74.0, 124.8, 125.9, 127.5, 128.4, 133.5 and 144.1; m/z (CI, NH₃) 317 (M⁺ – 17, 100%), 221 (51) and 185 (68).

(1*S*,7*R*,3*Z*)-1-[(*R*)-2-Acetoxy-2-phenylacetoxy]-7-*tert*-butyldimethylsilyloxy-1-phenyloct-3-ene 172

Following the general procedure, alcohol **166** (25 mg, 0.075 mmol) and (*R*)-2-acetoxy-2-phenylacetic acid, after chromatography using petrol: ether (4:1) as eluent, gave the title compound **172** (30 mg, 78%) as a colourless oil, $[\alpha]_D$ –61 (c = 1.36) (Found: M⁺ + NH₄, 528.3142. C₃₀H₄₆NO₅Si requires *M*, 528.3145); v_{max}/cm^{-1} 2929, 1750, 1373, 1231, 1208, 1174, 1138, 1084, 1058, 1003, 836, 775 and 698; δ_H (300 MHz, CDCl₃) 0.06 and 0.08 (each 3 H, s,

SiCH₃), 0.92 [9 H, s, SiC(CH₃)₃], 1.10 (3 H, d, J 6, 8-H₃), 1.17–1.44 (2 H, m, 6-H₂), 1.73–2.01 (2 H, m, 2- or 5-H₂), 2.21 (3 H, s, CH₃CO₂), 2.42–2.65 (2 H, m, 2- or 5-H₂), 3.73 (1 H, m, 7-H), 5.06 and 5.33 (each 1 H, m, 3-H or 4-H), 5.80 (1 H, t, J 7, 1-H), 6.03 (1 H, s, 2'-H) and 7.30–7.56 (10 H, m, ArH); m/z (CI, NH₃) 528 (M⁺ + 18, 28%), 384 (25) and 317 (100).

(1*S*,7*R*,3*Z*)-1-[(*S*)-2-Acetoxy-2-phenylacetoxy]-7-*tert*-butyldimethylsilyloxy-1-phenyloct-3-ene 173

Following the general procedure, alcohol **166** (26 mg, 0.078 mmol) and (*S*)-2-acetoxy-2-phenylacetic acid, after chromatography using petrol: ether (4:1) as eluent, gave the title compound **173** (33 mg, 83%) as a colourless oil, $[\alpha]_D$ +11.5 (*c*=1.38) (Found: M⁺ + NH₄, 528.3142. C₃₀H₄₆NO₅Si requires *M*, 528.3145); *v*_{max}/cm⁻¹ 2929, 1750, 1373, 1232, 1207, 1175, 1138, 1084, 1058, 836, 775 and 698; δ_H (300 MHz, CDCl₃) 0.07 and 0.09 (each 3 H, s, SiCH₃), 0.92 [9 H, s, SiC(CH₃)₃], 1.13 (3 H, d, *J* 6, 8-H₃), 1.24–1.51 (2 H, m, 6-H₂), 1.81–2.15 (2 H, m, 2- or 5-H₂), 2.22 (3 H, s, CH₃CO₂), 2.57 and 2.68 (each 1 H, dt, *J* 14.5, 7, 2- or 5-H), 3.77 (1 H, m, 7-H), 5.33 and 5.51 (each 1 H, m, 3-H or 4-H), 5.88 (1 H, t, *J* 7, 1-H), 6.05 (1 H, s, 2'-H) and 6.99–7.46 (10 H, m, ArH); *m/z* (CI, NH₃) 528 (M⁺ + 18, 100%), 414 (28) and 317 (73).

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