Stereoselective Synthesis of β , γ -Unsaturated Ketones by Acid-Mediated Julia-Type Transformation from 2-(1-Hydroxyalkyl)-1-alkylcyclopropanols

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Abstract: An efficient transformation of 2-(1-hydroxyalkyl)-1alkylcyclopropanols, obtained from α,β -unsaturated ketones, to β , γ -unsaturated ketones was achieved by trifluoroacetic acid (TFA)-mediated reaction.

Key words: β , γ -unsaturated ketone, methylene insertion, cyclopropanol, carbocation, stereoselective

In 1960, Julia reported acid-catalyzed isomerization of (1-hydroxyalkyl)cyclopropanes to (E)-homoallylic alcohols.1 This reaction has been studied intensively and applied for various natural product synthesis.^{2,3} When it was applied to 2-(1-hydroxyalkyl)-1-alkylcyclopropanols as starting materials, β , γ -unsaturated ketones were obtained via carbocation intermediate (Scheme 1).4,5 This transformation, however, has not been studied well, because the starting materials are difficult to obtain. We had reported stereoselective and stereospecific cyclopropanations of α,β -epoxyketones with bis(iodozincio)methane to give 2-(1-hydroxyalkyl)-1-alkylcyclopropanols.⁶ With these cyclopropane derivatives, we studied the transformation into β , γ -unsaturated ketones.



Scheme 1 Julia-type transformation of 2-(1-hydroxyalkyl)-1alkylcyclopropanol into (E)- β , γ -unsaturated ketone

As shown in Table 1, treatment of 2-(1-hydroxyethyl)-1phenylcyclopropanol (1a) with a catalytic amount of various acids resulted in moderate E/Z selectivity (entries 1– 3). Among them, TFA gave the best diastereoselectivity. Increasing the amount of TFA improved the yield and diastereoselectivity (entries 4 and 5). We found that treatment with five equivalents of TFA gave the desired β , γ unsaturated ketone with high diastereoselectivity (entry 6).7

Other examples are shown in Table 2. Although a reaction of tertiary alcohol 1b at 0 °C gave the desired product 2b in 70%, it proceeded sluggishly (entry 1). The reaction at

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Table 1 Acid-Mediated Transformation of 1a into 2a

$\begin{array}{c} HO \\ Ph \\ \hline 1a \end{array} \begin{array}{c} OH \\ CH_2Cl_2 \\ Ph \\ \hline 2a \end{array} \begin{array}{c} O \\ Ph \\ \hline 2a \end{array}$					
Entry	Acid (equiv) ^a	Time (h)	Temp (°C)	Yield (%) ^b	E/Z ^c
1	TFA (0.2)	24	r.t.	67	75:25
2	TsOH (0.2)	6	r.t.	89	68:32
3	TfOH (0.2)	6	r.t.	76	62:38
4	TFA (1.0)	24	r.t.	90	88:12
5	TFA (2.0)	24	r.t.	90	92:8
6	TFA (5.0)	0.5	0	90	95:5

^a TFA: trifluoroacetic acid, TsOH: *p*-toluenesulfonic acid, TfOH trifluoromethanesulfonic acid.

^b Yield was determined by ¹H NMR using bromoform as internal standard.

^c The *E*/*Z* ratio was determined by ¹H NMR analysis.

-78 °C, however, improved the yield (entry 2). Trisubstituted alkene 2d was obtained with high diastereoselectivity from 1d (entry 4). From bicyclic compound 1f, ringexpanded (E)-cycloalkene 2f was obtained in excellent yield (entry 6). In the case of a primary alcohol, such as 1g, this acid-mediated transformation did not proceed efficiently because of the lack of stability of the primary cation intermediate (entry 7). Treatment of 1g with mesyl chloride and pyridine,⁸ however, gave the corresponding β , γ -unsaturated ketone **2g** in good yield (Scheme 2).

These substrates, 2-(1-hydroxyalkyl)-1-alkylcyclopropanols, are available from α , β -epoxyketones with bis(iodozincio)methane⁶ in one step; α,β -epoxyketones can easily be prepared from α,β -unsaturated ketones with ba-



Scheme 2 Formation of β , γ -unsaturated ketones via mesylate elimination



^a Stereochemistry was determined by NOE analysis.

^b Reaction carries out at -78 °C.

^c Reaction carried out at r.t. for 24 h.

^d Corresponding α , β -unsaturated ketone was also obtained in 25%.

sic hydroperoxide. So the net transformation can be regarded as methylene insertion between a carbonyl group and an alkenyl group of α , β -unsaturated ketone (Scheme 3).

In conclusion, we showed the stepwise transformation of α , β -unsaturated ketones to (*E*)- β , γ -unsaturated ketones via cyclopropanation. Further applications for organic synthesis, especially natural product synthesis, are being studied in our laboratory.



Scheme 3 The entire transformation

General Procedure for the Transformation into $\beta{,}\gamma{\text{-}}\text{Unsaturated}$ Ketone

To a solution of 2-(1-hydroxyalkyl)-1-alkylcyclopropanol (0.5 mmol) in CH₂Cl₂ (2 mL), TFA (2.5 mmol) was added at 0 °C. The mixture was stirred at 0 °C for 30 min, then the solvent was removed under reduced pressure. The residue was purified by SiO₂ column chromatography (eluting with hexane–EtOAc) to give the corresponding β , γ -unsaturated ketone.

cis-2-(1-Hydroxyethyl)-2-methyl-1-phenylcyclopropanol (1d)

¹H NMR (500 MHz, C₆D₆): δ = 7.29–7.26 (m, 2 H), 7.18–7.14 (m, 2 H), 7.10–7.07 (m, 1 H), 4.03 (q, *J* = 7.0 Hz, 1 H), 2.73 (br s, 1 H), 2.62 (br s, 1 H), 1.29 (d, *J* = 7.0 Hz, 3 H), 0.83 (d, *J* = 5.5 Hz, 1 H), 0.80 (d, *J* = 5.5 Hz, 1 H), 0.70 (s, 3 H). ¹³C NMR (125 MHz, C₆D₆): δ = 141.2, 128.7, 128.3, 127.4, 70.6, 65.6, 32.0, 22.3, 19.6, 14.9. HRMS: *m*/z calcd for C₁₂H₁₆O₂: 192.1150; found: 192.1146.

(E)-3-Methyl-1-phenyl-3-penten-1-one (2d)

¹H NMR (500 MHz, CDCl₃): δ = 7.99–7.97 (m, 2 H), 7.57–7.53 (m, 1 H), 7.47–7.43 (m, 2 H), 5.39 (qqt, *J* = 7.0, 1.5, 1.5 Hz, 1 H), 3.47 (t, *J* = 1.0 Hz, 2 H), 1.69 (dt, *J* = 1.0, 1.0 Hz, 3 H), 1.64 (dtq, *J* = 7.0, 1.5, 1.5 Hz, 3 H). ¹³C NMR (125 MHz, CDCl₃): δ = 198.8, 136.9, 132.9, 130.0, 128.5, 128.3, 123.6, 49.2, 16.2, 13.6. HRMS: *m/z* calcd for C₁₂H₁₄O: 174.1045; found: 174.1042.

(E)-3-Cyclohexadecen-1-one (2f)

¹H NMR (500 MHz, CDCl₃): δ = 5.58–5.47 (m, 2 H), 3.03 (d, J = 6.5 Hz, 2 H), 2.49 (t, J = 7.0 Hz, 2 H), 2.09 (dt, J = 6.5, 6.0 Hz, 2 H), 1.59 (tt, J = 7.0, 7.0 Hz, 2 H), 1.42–1.21 (m, 18 H). ¹³C NMR (125 MHz, CDCl₃): δ = 210.4, 135.6, 123.2, 47.7, 40.9, 32.1, 28.2, 27.4, 27.2, 27.1, 26.9, 26.5, 26.2, 26.1, 25.6, 22.3. HRMS: *m/z* calcd for C₁₆H₂₈O: 236.2140; found: 236.2136.

1-Phenyl-3-buten-1-one (2g)

To a solution of 2-hydroxymethyl-1-phenylcyclopropanol (**1g**, 0.5 mmol) in CH_2Cl_2 (2 mL), pyridine (2 mmol), and MsCl (0.6 mmol) was added at r.t. The mixture was stirred at r.t. for 3 h, then sat. aq solution of NaHCO₃ was added. The organic layer was extracted with EtOAc, and the combined organic layers were dried over Na₂SO₄. The solvent was removed under reduced pressure, and the residue was purified by SiO₂ column chromatography eluting with hexane and EtOAc to give 1-phenyl-3-buten-1-one (**2g**, 86%, 0.43 mmol).

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equiv of H_2O to this system did not also change the ratio. These experiments mean that the diastereoselectivity does not come from isomerization of the initial product.

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