

Isoindoles and dihydroisoquinolines by gold-catalyzed intramolecular hydroamination of alkynes

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The title compounds are enantioselectively synthesized in just two preparative steps, making use of the Ugi-four-component reaction with an amino acid as chiral component, followed by a gold-catalyzed hydroamination.

Gold-catalyzed reactions have emerged as powerful and highly selective tools for organic synthesis in recent years:¹ obviously gold

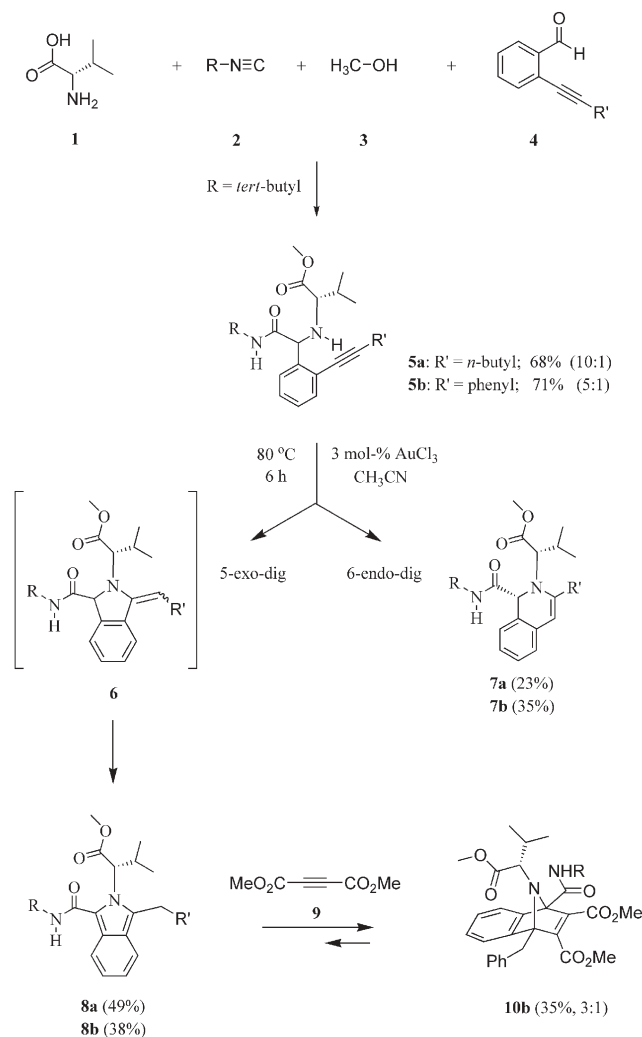
complexes exhibit a high reactivity as rather soft Lewis-acids, especially in terms of alkyne activation for nucleophilic attacks,² enabling these processes to proceed under very moderate reaction conditions. As part of our ongoing studies on gold-catalyzed annelation reactions³ we envisioned that a two step sequence of the Ugi-four-component reaction⁴ with a gold-catalyzed cyclization should offer a fascinating opportunity to build up chiral heterocycles with a high degree of complexity.

According to this general idea we applied L-valine (**1**) as chiral amine component and benzaldehydes **4**⁵ as source of the alkyne moiety. The condensation process of the four reaction components **1–4** proceeds somewhat sluggishly at room temperature within 6 to 9 days, building up the highly functionalized amines **5**, finally with satisfactory yield and diastereoselectivity (Scheme 1).

The hydroamination⁶ succeeded with just 3 mol-% gold chloride as catalyst in acetonitrile at 80 °C with overall yields above 70%: the dihydroisoquinolines **7** obviously are the result of a 6-*endo*-dig cyclization, whereas the isoindoles **8** derive from the 5-*exo*-dig products **6**, which of course isomerize under aromatization under the acidic reaction conditions.

Alkynes **5** were applied as purified main diastereomers and gave **7a** and **7b** as single diastereomers according to TLC and NMR, thus indicating configurational stability under the acidic reaction conditions. In crystals of **7b** there are two independent molecules with identical configurations in the unit cell. Knowing the absolute configuration of the L-valine chiral centres to be *S* our X-ray structure analysis of **7b** (Fig. 1) then established that the other chiral centres have the *R*-configuration.⁷

To our knowledge **8a** and **8b** are the first examples of chiral isoindoles, prompting us to test the face selectivity of the Diels–Alder reaction of **8b** with acetylene dicarboxylic acid dimethyl ester



Scheme 1 Combination of Ugi reaction, hydroamination and Diels–Alder reaction.

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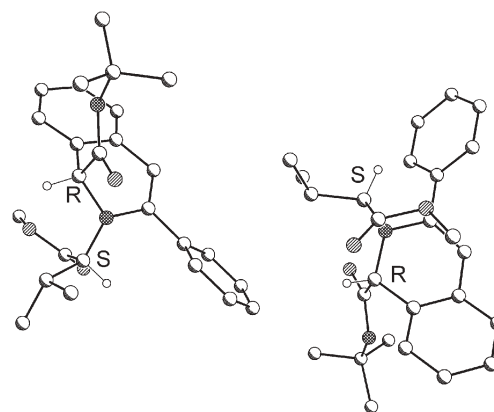


Fig. 1 Structure of the chiral isoquinoline **7b** in the crystal.

9. As the result we obtained a 3 : 1 mixture of the diastereomers of **10b** as an oily product: further separation and purification failed because of partial decomposition during flash chromatography. In test tube experiments also the Diels–Alder reactions with *N*-methyl and *N*-4-tolyl maleimides were examined: the resulting cycloadducts were even more sensitive than **10b**, clearly undergoing retro-Diels–Alder reaction at room temperature on silica. The retro reaction obviously profits from the electronic stabilization of isoindoles **8** by the amide substituent, which is part of the conjugated donor–acceptor system with the isoindole nitrogen. Normally retro-Diels–Alder reactions of isoindole adducts take place at much more elevated temperatures.⁸

In summary, the combination of Ugi reaction and gold catalysis offers a valuable opportunity for building up chiral dihydroisoquinolines and isoindoles, which are highly functionalized and of interest for further stereoselective transformations.

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Notes and references

- For reviews on gold catalysis, see: (a) G. Dyker, *Angew. Chem.*, 2000, **112**, 4407; G. Dyker, *Angew. Chem., Int. Ed.*, 2000, **39**, 4237; (b) A. S. K. Hashmi, *Gold Bull.*, 2004, **37**, 51; (c) A. Hoffmann-Röder and N. Krause, *Org. Biomol. Chem.*, 2005, **3**, 387; (d) A. S. K. Hashmi, *Angew. Chem., Int. Ed.*, 2005, **54**, 6990.
- Current examples of gold catalyzed reactions of alkynes: (a) Y. Luo and C. J. Li, *Chem. Commun.*, 2004, 1930; (b) Z. Shi and C. He, *J. Am. Chem. Soc.*, 2004, **126**, 13596; (c) S. T. Staben, J. J. Kennedy-Smith and F. D. Toste, *Angew. Chem.*, 2004, **116**, 5464; S. T. Staben, J. J. Kennedy-Smith and F. D. Toste, *Angew. Chem., Int. Ed.*, 2004, **43**, 5350; (d) X. Yao and C.-J. Li, *J. Am. Chem. Soc.*, 2004, **126**, 6884; (e) A. S. K. Hashmi, M. Rudolph, J. P. Weyrauch, M. Wölfe, W. Frey and J. W. Bats, *Angew. Chem.*, 2005, **117**, 2858; A. S. K. Hashmi, M. Rudolph, J. P. Weyrauch, M. Wölfe, W. Frey and J. W. Bats, *Angew. Chem., Int. Ed.*, 2005, **44**, 2798; (f) Y. Harrak, C. Blaszykowski, M. Bernard, K. Cariou, E. Mainetti, V. Mouries, A.-L. Dhiman, L. Fensterbank and M. Malacria, *J. Am. Chem. Soc.*, 2004, **126**, 8656; (g) C. Nieto-Oberhuber, M. P. Muñoz, E. Buñuel, C. Nevado, D. J. Cárdenas and A. M. Echavarren, *Angew. Chem.*, 2004, **116**, 2456; C. Nieto-Oberhuber, M. P. Muñoz, E. Buñuel, C. Nevado, D. J. Cárdenas and A. M. Echavarren, *Angew. Chem., Int. Ed.*, 2004, **43**, 2402; (h) M. R. Luzung, J. P. Markham and F. D. Toste, *J. Am. Chem. Soc.*, 2004, **126**, 10858; (i) V. Mamane, T. Gress, H. Krause and A. Fürstner, *J. Am. Chem. Soc.*, 2004, **126**, 8654; (j) K. Miki, T. Yokoi, F. Nishino, Y. Kato, Y. Washitake, K. Ohe and S. Uemura, *J. Org. Chem.*, 2004, **69**, 1557; (k) T. Yao, X. Zhang and R. C. Larock, *J. Am. Chem. Soc.*, 2004, **126**, 11164; (l) S. Antonietti, E. Genin, V. Michelet and J. P. Genêt, *J. Am. Chem. Soc.*, 2005, **127**, 9976; (m) N. Asao, K. Sato, Menggenbateer and Y. Yamamoto, *J. Org. Chem.*, 2005, **70**, 3682; (n) L. Zhang and S. A. Kozmin, *J. Am. Chem. Soc.*, 2005, **127**, 6962; (o) C.-G. Yang and C. He, *J. Am. Chem. Soc.*, 2005, **127**, 6966; (p) J. Zhu, N. P. Grigoriadis, J. P. Lee and J. A. Porco, Jr, *J. Am. Chem. Soc.*, 2005, **127**, 9342; (q) J. P. Markham, S. T. Staben and F. D. Toste, *J. Am. Chem. Soc.*, 2005, **127**, 9708; (r) A. S. K. Hashmi and P. Sinha, *Adv. Synth. Catal.*, 2004, **346**, 432; (s) A. S. K. Hashmi, M. Rudolph, J. P. Weyrauch, M. Wölfe, W. Frey and J. W. Bats, *Angew. Chem., Int. Ed.*, 2005, **44**, 2798.
- (a) G. Dyker, D. Hildebrandt, J. Liu and K. Merz, *Angew. Chem.*, 2003, **115**, 4536; G. Dyker, D. Hildebrandt, J. Liu and K. Merz, *Angew. Chem., Int. Ed.*, 2003, **42**, 4399; (b) G. Dyker and D. Hildebrandt, *J. Org. Chem.*, 2005, **70**, 6093.
- (a) I. Ugi, S. Lohberger, R. Karl, B. M. Trost, I. Fleming and C. H. Heathcock, The Passerini and Ugi reaction, in *Comprehensive Organic Chemistry*, Vol. 2, Ch. 4.6, pp. 1083–1109, Pergamon Press, Oxford, 1991; (b) A. Demharter, W. Hörl, E. Herdtweck and I. Ugi, *Angew. Chem.*, 1996, **108**, 185; A. Demharter, W. Hörl, E. Herdtweck and I. Ugi, *Angew. Chem., Int. Ed.*, 1996, **35**, 173; (c) A. Dömling and I. Ugi, *Angew. Chem.*, 2000, **112**, 3301–3344; A. Dömling and I. Ugi, *Angew. Chem., Int. Ed.*, 2000, **39**, 3168; (d) I. Ugi, W. Hörl, C. Hanusch-Kompa, T. Schmid and E. Herdtweck, *Heterocycles*, 1998, **47**, 965; (e) G. Dyker, *Angew. Chem.*, 1997, **109**, 1777; G. Dyker, *Angew. Chem., Int. Ed. Engl.*, 1997, **36**, 1700; (f) G. Dyker, K. Breitenstein and G. Henkel, *Tetrahedron: Asymmetry*, 2002, **13**, 1929.
- (a) K. Sonogashira, Y. Tohda and N. Hagihara, *Tetrahedron Lett.*, 1975, **50**, 4467; (b) S. Thorand and N. Krause, *J. Org. Chem.*, 1998, **63**, 8551; (c) G. Dyker, W. Stirner and G. Henkel, *Eur. J. Org. Chem.*, 2000, 1433.
- Some current articles on hydroamination: (a) T. E. Müller, M. Grosche, E. Herdtweck, A.-K. Pleier, E. Walter and Y.-K. Yan, *Organometallics*, 2000, **19**, 170; (b) E. Mizushima, T. Hayashi and M. Tanaka, *Org. Lett.*, 2003, **5**, 3349; (c) G. B. Bajracharya, Z. Huo and Y. Yamamoto, *J. Org. Chem.*, 2005, **70**, 4883; (d) Y. Luo, Z. Li and C.-J. Li, *Org. Lett.*, 2005, **7**, 2675.
- G. M. Sheldrick, *SHELXTL-97*, University of Göttingen, 1997. *Crystal data for 7b*: C₂₆H₃₂N₂O₃, M = 420.54, triclinic, space group P1, *a* = 9.187(14), *b* = 11.389(18), *c* = 12.80(2) Å, α = 105.30(13), β = 93.62(10), γ = 105.23(8)°, *V* = 1234(3) Å³, *Z* = 2, $2\theta_{\max}$ = 50°, 5711 measured reflections, 512 parameters, μ = 0.074 mm⁻¹, *R*1 = 0.0509 for 2835 observed reflections (*I* > 2σ(*I*)), *wR*2 = 0.1214 for all reflections; The intensity data were collected on a Bruker-axs-SMART 1000 diffractometer (Mo-K α radiation, λ = 0.7107 Å, *T* = 203 K). The structure was solved by direct methods and refined by full matrix least squares using SHELXTL-97. All non-hydrogen atoms were refined using anisotropic thermal parameters; hydrogen atoms were included by use of a riding model and fixed isotropic thermal parameters. CCDC 289676. For crystallographic data in CIF or other electronic format see DOI: 10.1039/b516017k.
- (a) P. Sohar, F. Miklos, A. Csampai and G. Stajer, *J. Chem. Soc., Perkin Trans. 1*, 2001, **5**, 558; (b) R. P. Kreher and N. Kohl, *Angew. Chem.*, 1984, **96**, 507; R. P. Kreher and N. Kohl, *Angew. Chem., Int. Ed.*, 1984, **23**, 517; (c) D. E. Remy and F. H. Bissett, *J. Org. Chem.*, 1978, **43**, 4469; (d) J. Bornstein, D. E. Remy and J. E. Shields, *J. Chem. Soc., Chem. Commun.*, 1972, **20**, 1149.