LETTERS 2009 Vol. 11, No. 3 653–655

ORGANIC

Asymmetric Construction of Quaternary Carbon Centers by Sequential Conjugate Addition of Lithium Amide and in Situ Alkylation: Utility in the Synthesis of (–)-Aspidospermidine

Mayuko Suzuki, Yoshito Kawamoto, Takeo Sakai, Yasutomo Yamamoto, and Kiyoshi Tomioka*

Graduate School of Pharmaceutical Sciences, Kyoto University, Yoshida, Sakyo-ku, Kyoto 606-8501, Japan

tomioka@pharm.kyoto-u.ac.jp

Received November 29, 2008

ABSTRACT



Chiral diether ligand-controlled asymmetric conjugate addition of a lithium amide to cyclopentenecarboxylate and subsequent in situ alkylation gave a chiral cyclopentane derivative bearing a quaternary carbon with high enantio- and diastereoselectivity. The cyclopentane derivative was converted successfully to (-)-aspidospermidine.

Synthetically useful asymmetric nitrogen—carbon bond formations that are otherwise difficult to accomplish have been achieved through conjugate addition reactions of both chiral¹ and achiral² lithium amides to enoates.³ Since the resulting lithium enolate intermediate is a powerful nucleophile, subsequent in situ alkylation is quite promising. Thus, potentially, useful one-pot transformations to produce adjacent asymmetric carbon centers with concomitant installation of vicinal N–C and C–C bonds could be achieved.⁴ Of particular interest is the asymmetric construction of quaternary carbons in tandem fashion. Given our success in the development of chiral diether-mediated asymmetric conjugate additions of lithium arylmethyl- and allyltrialkylsilylamides,⁵ we envisioned a tandem asymmetric conjugate addition– alkylation strategy for the construction of a quaternary carbon center. Herein, we describe a highly efficient enantioselective construction of a quaternary carbon center with up to 97% ee and over 98% de and the use of the product in the asymmetric total synthesis of (–)-aspidospermidine.

 ⁽a) Furukawa, M.; Okawara, T.; Terawaki, Y. Chem. Pharm. Bull.
 1977, 25, 1319–1325. (b) Hawkins, J. M.; Fu, G. C. J. Org. Chem. 1986, 51, 2820–2822. (c) Rico, J. G.; Lindmark, R. J.; Rogers, T. E.; Bovy, P. R. J. Org. Chem. 1993, 58, 7948–7951. (d) Enders, D.; Wahl, H.; Bettray, W. Angew. Chem., Int. Ed. Engl. 1995, 34, 455–457. (e) Sewald, N.; Hiller, K. D.; Helmreich, B. Liebigs Ann. 1995, 925–928. (f) Leroux, M.-L.; Gall, T.; Mioskowski, C. Tetrahedron: Asymmetry 2001, 12, 1817–1823. (g) Bull, S. D.; Davies, S. G.; Robert, P. M.; Savory, E. D.; Smith, A. D. Tetrahedron 2002, 58, 4629–4642.

⁽²⁾ Yamamoto, Y.; Asano, N.; Uyehara, T. J. Am. Chem. Soc. 1992, 114, 5427-5429.

⁽³⁾ Review of the asymmetric conjugate addition of chiral lithium amide: Davies, S. G.; Smith, A. D.; Price, P. D *Tetrahedron: Asymmetry* **2005**, *16*, 2833–2891.

⁽⁴⁾ Our recent example: Sakai, T.; Kawamoto, Y.; Tomioka, K. J. Org. Chem. 2006, 71, 4706–4709.

^{(5) (}a) Doi, H.; Sakai, T.; Iguchi, M.; Yamada, K.; Tomioka, K. J. Am. Chem. Soc. 2003, 125, 2886–2887. (b) Sakai, T.; Doi, H.; Kawamoto, Y.; Yamada, K.; Tomioka, K. Tetrahedron Lett. 2004, 45, 9261–9263. (c) Doi, H.; Sakai, T.; Yamada, K.; Tomioka, K. Chem. Commun. 2004, 1850–1851. (d) Sakai, T.; Doi, H.; Tomioka, K. Tetrahedron 2006, 62, 8351–8359. (e) Sakai, T.; Yamada, K.; Tomioka, K. Chem. Asian J. 2008, 3, 1486–1493.

 Table 1. Tandem Asymmetric Conjugate Addition-Alkylation^a

Mes Li	-TMS 2 + CO ₂ <i>t</i> -Bu 3	Ph MeO 1 toluei -78 °C, -60 °C,	Ph OMe ne 1.5 h 1.5 h	R-X (HMPA) THF -40 °C, 1 h	Mes NH E R 4	⊃₂ <i>t</i> -Bu
entry	R-2	X	4	yield (%)	ee (%)	de (%)
1	$PhCH_2Br$		4a	90	97	>98
2	$H_2C=CH$	$\mathrm{CH}_{2}\mathrm{Br}$	4b	93	95	>98
3	EtI^{b}		4 c	94	95	>98
4	MeI		4d	93	95	>98
a		0				

^{*a*} All reactions were performed using **2** (3 equiv), **1** (3.6 equiv), and **3** (1 equiv). In the alkylation step, R–X (2 equiv) was used in entries 1, 2, and 4. ^{*b*} For ethylation, EtI (10 equiv) and HMPA (6 equiv) were used.

We began our studies with allylation of the lithium enolate intermediate generated by conjugate addition of lithium N-mesitylmethyl-N-TMS-amide 2 to tert-butyl cyclopentenecarboxylate 3 in the presence of C_2 -symmetric chiral diether 1 in toluene at -78 °C (Table 1, entry 2).^{5d} Treatment of a toluene solution of this lithium enolate with allyl bromide did not yield the expected allylation product 4b (R = $CH_2CH=CH_2$), but instead yielded the conjugate addition product 4 (R = H) with 96% ee. Upon addition of THF to a toluene solution of the lithium enolate at -78 °C, the allylation process proceeded at -40 °C for 1 h, giving the desired product **4b** as nearly a single diastereomer in 92% yield. However, the enantioselectivity dropped to 90% ee. We overcame this problem by additional stirring at -60 °C for 1.5 h during the conjugate addition step (to complete the asymmetric conjugate addition, before the addition of THF, which accelerates the racemic conjugate addition reaction), giving the allylation product 4b with 95% ee and over 98% de in 93% yield.

C-Benzylation and *C*-methylation also proceeded satisfactorily to give the corresponding quaternary carbon products **4a** (R = Bn) and **4d** (R = Me) with up to 97% ee as nearly single diastereomers (entries 1 and 4). *C*-Ethylation with ethyl iodide required the coaddition of HMPA to give the product **4c**⁶ (R = Et) with 95% ee in 94% yield (entry 3).

Ready conversion of the secondary amine product to the primary amine was demonstrated by successively treating **4a** with NCS for *N*-chlorination (87%); *N*-dibutylamino-DBU for elimination to the imine, and; hydroxylamine for transoximation (51%, two steps). These successive reactions gave the β -amino acid derivative **5** (Scheme 1).^{5d,7}

Elimination of the amino function of 5 to give olefin 6 was also performed by sequential *N*-dimethylation with formalin–sodium cyanoborohydride, *m*-CPBA oxidation to

Scheme 1. Conversion of 4 to 5 and 6



N-oxide, and thermal Cope elimination⁸ in THF at 60 °C in 84% overall yield (Scheme 1). We also obtained olefin **6** from **4a** via *N*-methylation, *m*-CPBA oxidation, and Cope elimination⁹ with dialuminium trioxide¹⁰ in 33% overall yield.

Asymmetric conjugate addition of lithium allylamide 7^{5c} to **3** and in situ benzylation gave **8**, which was then treated without purification with aq HF for protodesilylation to provide **9** (Scheme 2). Olefin isomerization of **9** with Wilkinson's catalyst in aqueous acetonitrile at reflux and simultaneous hydrolysis of the imine gave primary amine **5** in 90% yield. In this benzylation, we obtained **9** with 98% ee in 55% yield, which is in stark contrast to 85% ee for the protonation product in the conjugate addition of **7**.^{5c} Kinetic enantioenrichment in the benzylation step may be responsible for this high enantioselectivity. In fact, almost complete *C*-allylation and *C*-methylation gave the corresponding products with 85% ee in reasonably high yields of 88% and 94%, respectively.

Scheme 2. Asymmetric Conjugate Addition of Allylamine and Benzylation



The asymmetric total synthesis of aspidospermidine 18 has been considered to be a touchstone of the methodology for

⁽⁶⁾ The stereochemistry of asymmetric conjugate addition has been reported in ref 5b. The relative configuration of **4c** was assigned by an NOE experiment. Ooi, T.; Miki, T.; Taniguchi, M.; Shiraishi, M.; Takeuchi, M.; Maruoka, K. *Angew. Chem., Int. Ed.* **2003**, *42*, 3796–3798.

⁽⁷⁾ Katoh, T.; Watanabe, T.; Nishitani, M.; Ozeki, M.; Kajimoto, T.; Node, M. *Tetrahedron Lett.* **2008**, *49*, 598–600.

⁽⁸⁾ Grainger, R. S.; Patel, A. Chem. Commun. 2003, 1072-1073.

⁽⁹⁾ Albini, A. Synthesis 1993, 263–277.

⁽¹⁰⁾ Brand, M.; Drewes, S. E.; Loizou, G.; Roos, G. H. P. Synth. Commun. 1987, 17, 795-802.

Scheme 3. Total Synthesis of (-)-Aspidospermidine 18



the asymmetric construction of quaternary carbons.^{11,12} The utility of the quaternary carbon product **4** was demonstrated by the asymmetric total synthesis of (-)-**18** (Scheme 3).¹³

(11) Reviews of total synthesis: (a) Saxton, J. E. In *The Alkaloids*;
Cordell, G. A., Ed.; Academic Press: New York, 1998; Vol. 51, Chapter 1.
(b) Toyota, M.; Ihara, M. *Nat. Prod. Rep.* **1998**, 327–340, and references therein.

(12) Recent examples of total synthesis of nonracemic aspidospermidine:
(a) Padwa, A.; Price, A. T. J. Org. Chem. 1998, 63, 556-565. (b) Kobayashi,
S.; Peng, G.; Fukuyama, T. Tetrahedron Lett. 1999, 40, 1519-1522. (c)
He, F.; Bo, Y.; Altom, J. D.; Corey, E. J. J. Am. Chem. Soc. 1999, 121,
6771-6772. (d) Kozmin, S. A.; Iwama, T.; Huang, Y.; Rawal, V. H. J. Am.
Chem. Soc. 2002, 124, 4628-4641. (e) Marino, J. P.; Rubio, M. B.; Cao,
G.; Dios, A. J. Am. Chem. Soc. 2002, 124, 13398-13399. (f) Fukuda, Y.;
Shindo, M.; Shishido, K. Org. Lett. 2003, 5, 749-751. (g) Iyengar, R.;
Schildknegt, K.; Morton, M.; Aube, J. J. Org. Chem. 2005, 70, 1064510652. (h) Pearson, W. H.; Aponick, A. Org. Lett. 2006, 8, 1661-1664. (i)
Ishikawa, H.; Elliott, G. I.; Velcicky, J.; Choi, Y.; Borger, D. L. J. Am.
Chem. Soc. 2006, 128, 10596-10612. (j) Ishikawa, T.; Kudo, K.; Kuroyabu,
K.; Uchida, S.; Kudoh, T.; Saito, S. J. Org. Chem. 2008, 73, 7498-7508.

(13) Recent examples of total synthesis of racemic aspidospermidine:
(a) Callaghan, O.; Lampard, C.; Kennedy, A. R.; Murphy, J. A. J. Chem. Soc., Perkin Trans. 1 1999, 8, 995–1002. (b) Toczko, M. A.; Heathcock, C. H. J. Org. Chem. 2000, 65, 2642–2645. (c) Patro, B.; Murphy, J. A. Org. Lett. 2000, 2, 3599–3601. (d) Banwell, M. G.; Smith, J. A. J. Chem. Soc., Perkin Trans. 1 2002, 2613–2618. (e) Banwell, M. G.; Lupton, D. W.; Willis, A. C. Aust. J. Chem. 2005, 58, 722–737. (f) Sharp, L. A.; Zard, S. Z. Org. Lett. 2006, 8, 831–834. (g) Coldham, I.; Burrell, A. J. M.; White, L. E.; Adams, H.; Oram, N. Angew. Chem., Int. Ed. 2007, 46, 6159–6162. (h) Callier-Dublanchet, A.-C.; Cassayre, J.; Gagosz, F.; Quiclet-Sire, B.; Sharp, L. A.; Zard, S. Z. Tetrahedron 2008, 64, 4803–4816.

The ee of the ethylation product 4c (95% ee) was increased to >99% ee in 85% recovery by recrystallization of its hydrochloride from ethyl acetate. Successive methylation of the hydrochloride salt of 4c, N-oxidation with m-CPBA, and Cope elimination with Al_2O_3 in *tert*-butyl alcohol¹⁴ gave olefin 11 in 65% yield along with 12 (11%) and 13 (11%). Lithium aluminum hydride reduction of ester 11 to alcohol 14, TPAP oxidation to the aldehyde, Wittig olefination for one-carbon elongation, hydrolysis to the aldehyde, and sodium borohydride reduction gave alcohol 15 in 79% yield (five steps from 11). These steps should be carefully carried out because the products all have low boiling points and can easily be distilled off. Oxidative cleavage of olefin 15 with osmium tetroxide and sodium metaperiodate yielded a lactol, which was then oxidized to δ -lactone 16, bearing three differently oxidized oxygen functionalities¹⁵ in 85% yield. Amide formation with lactone 16 and tryptamine in *n*-butyl alcohol at reflux gave 17 in 75% yield. The remaining transformations were a modification of the Harley-Mason's protocol. Sulfuric acid treatment of 17 induced Pictet-Spengler cyclization. Simultaneous rearrangement occurred at reflux for 2.5 h. Lithium aluminum hydride reduction in THF at reflux for 1.5 h completed the synthesis of (-)-aspidospermidine **18** in 37% yield.¹⁶

In summary, we have developed an efficient method for constructing a chiral quaternary carbon by tandem asymmetric conjugate addition of a lithium amide to an enoate and its subsequent in situ alkylation. Total synthesis of (–)aspidospermidine was successfully demonstrated as a touchstone of the strategic application of the asymmetric conjugate amination—alkylation protocol.

Acknowledgment. This research was partially supported by a Grant-in-Aid for Young Scientists (B), a Grant-in-Aid for Scientific Research in Priority Areas "Advanced Molecular Transformations of Carbon Resources", a Grant-in-Aid for Scientific Research (A), and the Targeted Proteins Research Program of the Ministry of Education, Culture, Sports, Science, and Technology, Japan. T.S. acknowledges a JSPS fellowship.

Supporting Information Available: Experimental procedures and characterization data of new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

OL802759J

⁽¹⁴⁾ Zavada, J.; Krupicka, J.; Sicher, J. Collect. Czech. Chem. Commun. 1966, 31, 4273–4285.

⁽¹⁵⁾ Node, M.; Nagasawa, H.; Fuji, K. J. Org. Chem. 1990, 55, 517–521.

^{(16) (}a) Harley-Mason, J.; Kaplan, M. J. Chem. Soc., Chem. Commun. **1967**, 915–916. (b) Schultz, A. G.; Pettus, L. J. Org. Chem. **1997**, 62, 6855–6861.