A NEW SYNTHESIS OF 2', 3'-DIDEOXYNUCLEOSIDES FOR AIDS CHEMOTHERAFY

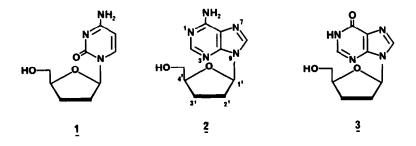
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SUMMARY: Dideoxynucleosides were prepared in high optical purity from L-glutamic acid. The condensation reactions between activated 2,3-dideoxypentofuranoses and silylated purines or pyrimidines afforded separable β/α mixtures of dideoxynucleosides.

2',3'-Dideoxynucleosides have found utility mainly as reagents for DNA sequencing¹, and consequently are commercially available in only small amounts at very high prices². The recent discovery³ that some dideoxynucleosides selectively inhibit <u>in vitro</u> the infectivity and cytopathic effect of the HTLV-III (HIV) virus has sparked renewed interest in this class of compounds, and dideoxycytidine, <u>1</u>, is now being clinically evaluated as treatment against Acquired Immune Deficiency Syndrome (AIDS). A general and economically attractive synthesis of dideoxynucleosides has therefore become an important goal.

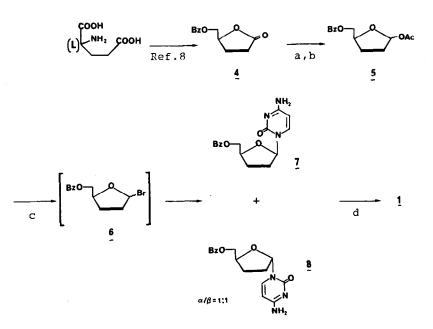
2',3'-Dideoxynucleosides are typically synthesized from 2'-deoxynucleosides <u>via</u> Barton-type deoxygenation reactions⁴, or from intact nucleosides by multistep routes involving deoxygenation reactions to 2',3'-unsaturated dideoxynucleosides⁵, which are then hydrogenated. Approaches through ketonucleosides⁶ and a photoreductive process⁷ have also been described.



We envisioned producing dideoxycytidine (ddC), <u>1</u>, dideoxyadenosine (ddA), <u>2</u>, and dideoxyinosine (ddI), <u>3</u>, from a stereochemically less complex, and much cheaper, precursor, <u>i.e.</u> L-glutamic acid (Scheme I).

The known, stereospecific deamination-lactonization reaction, followed by standard manipulations⁸, gave lactone 4. Selective reduction⁹ with disiamylborane followed by oxidative hydrolysis and acetylation gave a good yield of acetate 5 as a mixture of anomers¹⁰.

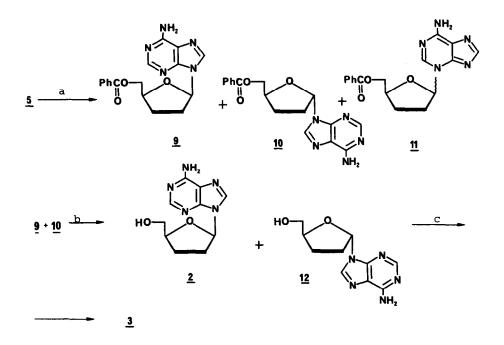
Scheme I



Reagents: (a) disiamylborane, THF, then H_2O_2 , $50^{\circ}C$; 94% (b) Ac₂O, pyridine, RT; 92% (c) Me₃SiBr, CH₂Cl₂, 0°C, then bis-silylcytosifie, -78°C -> RT; 44% yield for <u>7</u> after chromatography; (d) NH₃/MeOH quant.

Treatment with trimethylsilylbromide¹¹ cleanly afforded a mixture of anomeric bromides¹² <u>6</u>. Addition of silylated cytosine to this solution gave a separable mixture¹³ of protected ddC and its α anomer in a 1:1 ratio¹⁴. Standard deprotection afforded ddC¹⁵. A similar approach (Scheme II) was successful for purine nucleosides. The coupling reaction was carried out at room temperature in order to minimize formation of a side-product, presumably the N-3 alkylation $\operatorname{product}^{16} \frac{11}{11}$.

Scheme II



Reagents: (a) TMSBr, $0^{\circ}C$; then bis-silyladenine, RT; 63% yield for 2 + 10 (b) NH₃, MeOH, RT, quant. yield for 2 + 12 (c) adenosine deaminase, H₂O; 33% yield for 3 + 12 crystallization.

Dideoxyinosine, another potentially useful antiviral agent, was prepared by subjecting the mixture of <u>9</u> and <u>10</u> to a solution of the enzyme adenosine deaminase¹⁷, which quantitatively deaminated the ß-anomer, leaving the α unreacted. Consequently ddI, <u>3</u>, was obtained by direct crystallization of the reaction crude from methanol¹⁸.

The methodology described in this letter should make a variety of 2',3'dideoxynucleosides easily available at very low cost.

REFERENCES AND NOTES:

- ^{a)} Present address: Pharmaceutical Research and Development Division, Bristol-Myers Company, 5 Research Parkway, Wallingford, CT 06492-7660
- 1. F. Sanger, S. Nicklen, A.R. Coulson Proc.Natl.Acad.Sci.USA, 1977, 74, 5463.
- 2. For example, Aldrich Chemical Company sells 2',3'-dideoxycytidine at \$42.00/100 mg.
- 3. H. Mitsuya, S. Broder Proc.Natl.Acad.Sci. USA, 1986, 83, 1911.
- 4. See, for example: E.J. Prisbe, J.C. Martin <u>Synth.Commun.</u>, 1985, <u>15</u>, 401.
- See, for example: (a) J.P. Horwitz, J. Chua, M. Noel, J.T. Donatti <u>J.Org.Chem.</u>, 1967, <u>32</u>, 817 (b) K.E. Pfitzner, J.G. Moffatt <u>J.Org.Chem.</u>, 1964, <u>29</u>, 1508. (c) M.J. Robins, F. Hansske, N.H. Low, J.I. Park <u>Tetrahedron Lett.</u>, 1984, <u>25</u>, 367 and references therein.
- 6. J. Thiem, D. Rasch Nucleosides & Nucleotides, 1985, 4, 487.
- I. Saito, H. Ikehira, R. Kasatani, M. Watanabe, T. Matsuura <u>J.Am. Chem.Soc.</u>, 1986, <u>108</u>, 3115.
- 8. M. Taniguchi, K. Koga, S. Yamada Tetrahedron, 1974, 30, 3547.
- P. Kohn, R.H. Samaritano, L.M. Lerner <u>J. Am. Chem. Soc.</u>, 1964, <u>86</u>, 1457. We thank Dr. D.G. Walker of this department for suggesting the use of the reagent.
- All new compounds gave satisfactory spectral and analytical data. The α/ß ratio for 5 was 55:45 and was assigned on the basis of NMR considerations. See: K.J. Stone, R.D. Little J.Am.Chem.Soc., 1985, 107, 2495.
- 11. J.W. Gillard, M. Israel <u>Tetrahedron Lett.</u>, 1981, <u>22</u>, 513
- 12. Use of 1.2 eq. of TMSBr was needed to achieve >90% conversion. The anomeric protons in the NMR (CDCl₃) at δ 6.80; 6.70 (α/β 60:40) were diagnostic. These anomeric bromides were too unstable for isolation.
- Separation was effected by preparative reverse phase chromatography (C-18, elution with water/methanol mixtures)
- 14. The β/α ratio is very sensitive to the substrate used and the experimental conditions, and some selectivity in favor of the β anomer can be achieved. We are currently looking for conditions that will maximize this ratio.
- 15. Dideoxycytidine so obtained had ¹H-NMR, m.p. and optical rotation in agreement with literature values. See ref. 5a.
- 16. Assignment was made on the basis of the ¹H-NMR and UV spectra. Also treatment of isolated <u>11</u> with acids gave in variable yield isomerization to the N-9 alkylated compounds <u>9</u> and <u>10</u>. For a precedent, see: M. Miyaki, S. Shimizu <u>Chem.Pharm.Bull.</u>, 1970, <u>18</u>, 732.
- 17. A.Bloch, M.J. Robins, J.R. McCarthy <u>J.Med.Chem.</u>, 1967, <u>10</u>, 908.
- 18. Dideoxyadenosine had $[\alpha]_{D}^{23}$ -25.3° (C 1.00, H₂O); lit. ¹⁹: $[\alpha]_{D}^{25}$ -25.2° (C 1.01, H₂O). Dideoxyinosine had IR (KBr) 1700 cm⁻¹; UV(H₂O): λ_{max} 240 nm. NMR (CDCl₃, 360 MHz) anomeric proton at δ 6.185 (dd, J = 6.8; 3.3 Hz); $[\alpha]_{D}^{2}$ -28.7° (C 1.00, DMF).
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