



Original article

Optimization of 5-hydroxytryptamines as dual function inhibitors targeting phospholipase A₂ and leukotriene A₄ hydrolase

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ABSTRACT

Dual function inhibitors targeting phospholipase A₂ (PLA₂) and leukotriene A₄ hydrolase (LTA₄H) may balance the arachidonic acid (AA) metabolic network and be used as new anti-inflammatory drugs. In previous study, we discovered multi-target drugs towards the AA metabolic network, among which a dual-target inhibitor (**JMC08-4**) for human nonpancreatic secretory phospholipase A₂ (hnps-PLA₂) and human leukotriene A₄ hydrolase (LTA₄H-h) was found. Based on the structure of compound **JMC08-4**, new dual-target inhibitors were designed assisted by molecular docking. In this report, a series of 5-hydroxytryptamine compounds were synthesized; and most of these title compounds showed more potent inhibitory activity than compound **JMC08-4** in the *in vitro* bioassay against these two enzymes. The best one inhibited hnps-PLA₂ and LTA₄H-h with IC₅₀ values of $9.2 \pm 0.5 \mu\text{M}$ and $2.4 \pm 1.4 \mu\text{M}$, respectively.

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1. Introduction

Phospholipase A₂ (PLA₂) and leukotriene A₄ hydrolase (LTA₄H) enzymes are important anti-inflammatory targets. PLA₂ hydrolyzes the acylester bond at the sn-2 position of phosphoglycerides and liberates free fatty acids (predominantly arachidonic acid) and lysophospholipids. Over expression of PLA₂ plays an important role in the inflammatory process [1,2]. The proinflammatory substance leukotriene B₄ (LTB₄) derived from arachidonic acid (AA) via 5-lipoxygenase (5-LOX) pathway. LTA₄H catalyzes the rate-limiting step in the biosynthesis of LTB₄, and therefore is a potent inducer of inflammation [3]. PLA₂, LTA₄H, and other key inflammation-related proteins in the AA metabolic network, such as cyclooxygenase 2 (COX-2) and 5-LOX, have been studied as targets for anti-inflammatory drugs. Many inhibitors of these targets have been used in treating acute and chronic inflammation. However, inhibiting only one target of the metabolic pathways can possibly shift the metabolism of AA towards other pathways, which may cause side effects. For example, selective COX-2 inhibitors were reported to have cardiovascular side effects in clinical trials [4]. The reason is that the inhibitors increased the production of leukotrienes (LTs) on one hand and decreased the generation of

prostaglandins (PGs) on the other hand. Multi-target drug discovery has become an increasingly perspective to overcome disadvantages of single-target based drugs [5,6]. In anti-inflammatory drug research, some multi-target inhibitors have been reported, including combinations of COX/5-LOX [7–10], 5-LOX/thromboxane A₂ synthase (TxA₂S) [11], thromboxane A₂ receptor (TxA₂R)/TxA₂S [12], 5-LOX/microsomal prostaglandin E synthase (mPGES) [13], and COX/LTA₄H [14].

Although multi-target drugs have advantages, there have been no interchangeable strategies to design this new kind of drugs and the discovery of it was often serendipitous [15]. In a previous study, we developed a computational algorithm for finding multiple target optimal intervention (MTOI) solutions in a disease network. We applied MTOI to the AA network and a number of optimal combinations of target intervention (MTOI solutions) were found, such as: COX-2/LTA₄H, COX-2/5-LOX, PGES/LTA₄H, PLA₂/LTA₄H, PLA₂/COX-2/LTA₄H, and PLA₂/5-LOX/COX-2, which are both effective in controlling inflammation mediators and safe with minimal side effects [16]. We also developed a strategy that combined receptor-based common pharmacophores with molecular docking, which was successfully applied to screen for dual-target inhibitors against hnps-PLA₂ and LTA₄H-h [17]. Compound 1,2-amino-3-[5-(benzyloxy)-1H-indol-3-yl] propanoic acid (**JMC08-4**) (Fig. 1), whose binding conformations in proteins accommodate the common pharmacophores, is used as lead compound against the two enzymes. We explored structural optimizations of the **JMC08-**

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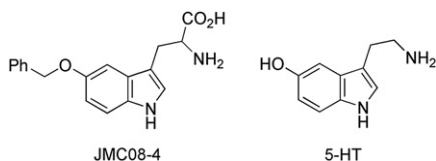


Fig. 1. Structure of JMC08-4 and 5-HT.

4, in relation to the binding pocket requirements of both enzymes, 5-hydroxytryptamine (serotonin, 5-HT, Fig. 1) analogs were then selected for studying.

The 5-HT analogs are less difficult in synthesis and possess potential druggability. The compound 5-HT is a neurotransmitter that modulates neural activity and a wide range of neuropsychological processes [18]. Some of 5-HT analogs have been reported as agonists or antagonists of serotonin receptors. Peroutka et al. reported that 5-benzyloxytryptamine (**4a** in this article) is a known ligand of 5-hydroxytryptamine receptors [19], and Buzzi et al. found it can mediate blockage of neurogenic plasma extravasation in rat duramater [20]. We hope to develop 5-HT analogs as new anti-inflammatory drugs. A series of 5-HT analogs have been designed and tested for their inhibition against LTA₄H-h and hnps-PLA₂.

2. Results and discussion

2.1. Design of 5-HT analogs

Compound JMC08-4 was docked into the substrate binding site of LTA₄H-h and hnps-PLA₂ using Autodock 4.0 (Fig. 2). In Fig. 2A, we found that there was hydrophobic space in the L-shaped pocket of LTA₄H-h when compound JMC08-4 bound there. It suggested that increasing the length of compound JMC08-4's hydrophobic chain could enhance the interaction between this compound and LTA₄H-h. Fig. 2B is the binding pocket of hnps-PLA₂. We can see that using a larger aromatic group to replace the 5-benzyloxy group of compound JMC08-4 (e.g. compound in Fig. 3B), the binding of the ligand to hnps-PLA₂ was almost unchanged.

Molecular docking results of compound JMC08-4 gave clues on the further optimization of LTA₄H/PLA₂ dual function inhibitors. Using compound JMC08-4 backbone, a series of 5-HT analogs were designed as LTA₄H/PLA₂ inhibitors. First, 5-hydroxyl group of 5-HT was combined with different aromatic groups to enhance the hydrophobicity. One of them is **4f**, which was docked into both enzymes as shown in Fig. 3. The larger aromatic group of **4f** interacted with the residues Leu369, Leu367 and Pro382 of the pocket, the interaction increased hydrophobicity as we prediction. To further enhance the activity of these compounds, the 3-ethylamine

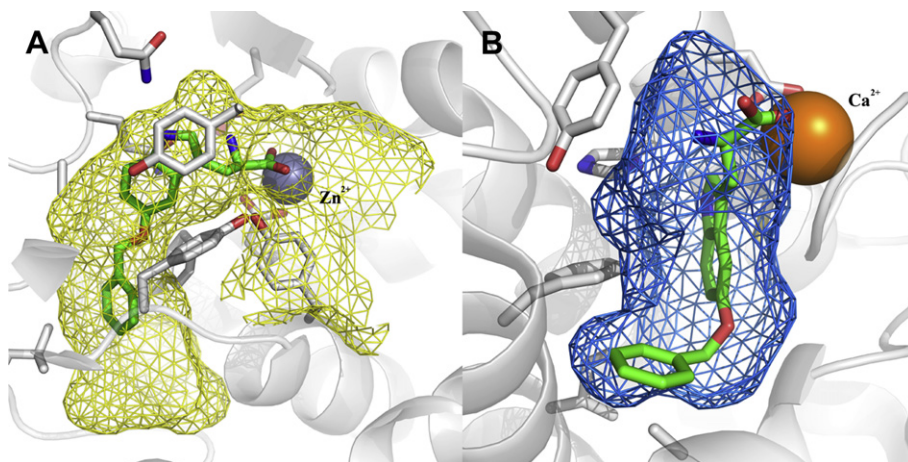


Fig. 2. The binding models of compound JMC08-4. A, compound JMC08-4 with LTA₄H; B, compound JMC08-4 with sPLA₂.

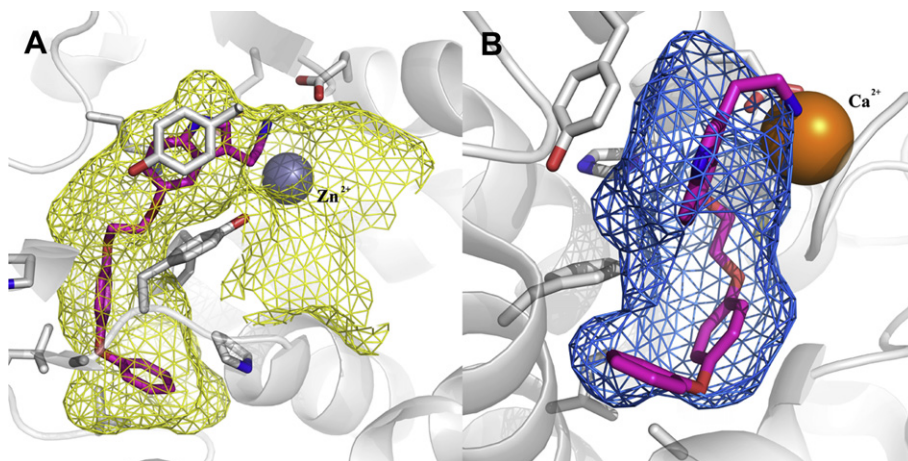


Fig. 3. The binding models of a molecule with a longer hydrophobic group (**4f**) (predicted by molecular docking). A, compound **4f** with LTA₄H; B, compound **4f** with sPLA₂.

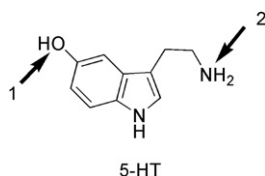


Fig. 4. Design of new inhibitors. 1, modification with large hydrophobic groups; 2, modification with functional groups that coordinate with metal atoms.

group of 5-substituted-tryptamine was combined with either functional groups possessing negative charge or atoms coordinating with metal atoms (e.g. carboxyl).

For designed potential 5-HT analogs, the two major ways of optimization are: 1) modification with large hydrophobic groups; 2) addition of functional groups that coordinate with metal atoms (Fig. 4).

2.2. Synthesis of 5-HT analogs

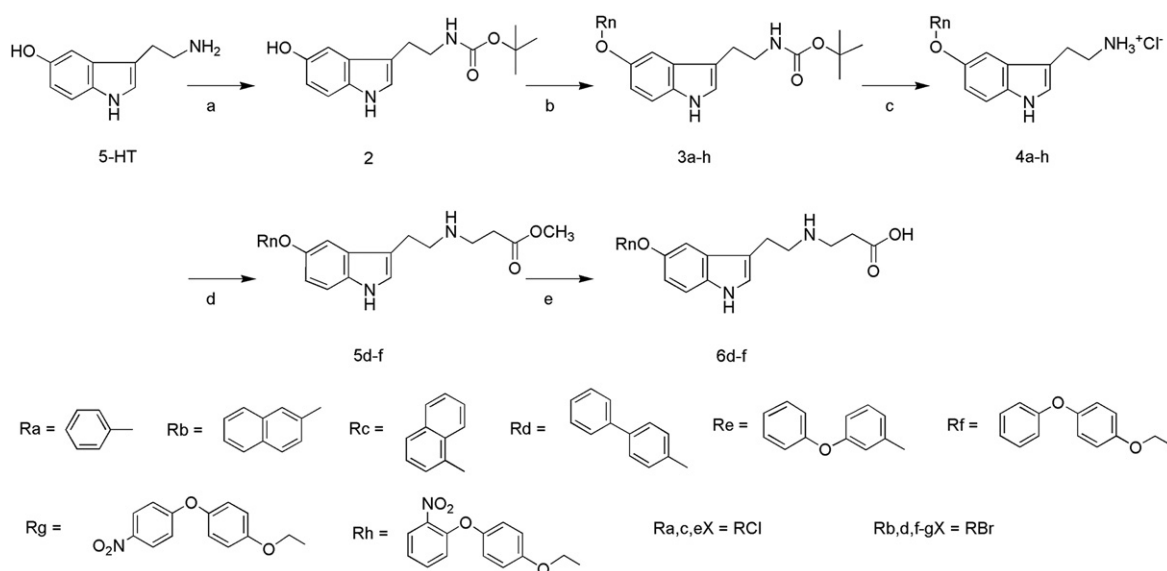
The fourteen 5-HT analogs (**4a–h**, **5d–f**, and **6d–f**) were prepared by multi-step organic synthesis (Scheme 1). Primary amino group of 5-HT was first protected by BOC [21]. Then, in the

presence of Cs_2CO_3 , the intermediate N-BOC-5-HT (**2**) and different electrophiles R_nX (X is a halogen) reacted in DMF to get intermediates **3a–h**. The title compounds **4a–h** were obtained by removing the BOC anhydride in the ethyl acetate solution of hydrochloride. These monohydrochloride compounds were converted to the title compounds **5d–f** through nucleophilic conjugate addition by treating with methacrylate in methanol. Another kind of title compounds **6d–f** were obtained from **5d–f** by hydrolysis in an acetone solution of NaOH.

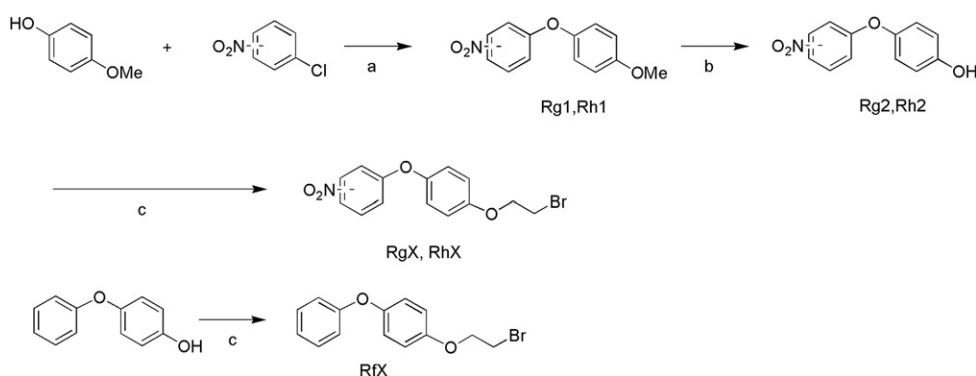
In the synthesis of compounds **3a–h**, some electrophilic reagents (**Rf–hX**) were not commercially available. Scheme 2 shows the preparations of these complicated electrophiles. In the presence of KOH, 4-methoxyphenol and chloronitrobenzene was melt together and condensed to get compounds **Rg1** and **Rh1**. Stirred in dichloromethane solution of BBr_3 , compounds **Rg1** and **Rh1** were converted to **Rg2** and **Rh2**. When **Rg2**, **Rh2** and phenoxyphenol were further refluxed in 1,2-dibromoethane of KOH, corresponding intermediates **RfX**, and **Rg–hX** were got.

2.3. Evaluation of inhibitory activities

All the title compounds were subjected to *in vitro* inhibitory activities against $\text{LTA}_4\text{H-h}$ and hnpS-PLA_2 using the following



Scheme 1. Synthesis of compounds **4a–h**, **5d–f**, and **6d–f**. (a), BOC anhydride, NaHCO_3 , NaCl, DCM; (b), R_nX , Cs_2CO_3 , DMF; (c), HCl, EtOAc, rt; (d), methyl acrylate, Na_2CO_3 , MeOH; (e), NaOH, acetone, water.



Scheme 2. Preparation of complicated electrophiles **Rf–hX**. (a), KOH; (b), BBr_3 , CH_2Cl_2 ; (c), 1,2-dibromoethane, KOH.

Table 1
Inhibitory activity of compounds **4a–h**, **5d–f**, and **6d–f**.

Compound	R1	R2	LTA ₄ H IC ₅₀ (μM)		PLA ₂ IC ₅₀ (μM)
			Aminopeptidase	Epoxide hydrolase	
Bestatin	—	—		4.2 ± 0.5	>100
NI101	—	—	>100	>100	0.88 ± 0.15
JMC08-4	—	—	17.3 ± 0.2	231 ± 4	42.1 ± 1.9
4a		H	13 ± 1	>100	52 ± 5
4b		H	8.7 ± 0.8	>100	25 ± 5
4c		H	30 ± 5	>100	27 ± 9
4d		H	2.1 ± 0.2	18 ± 4	4.4 ± 0.9
4e		H	8.9 ± 0.5	>100	15.3 ± 0.4
4f		H	1.3 ± 0.2	15 ± 3	11.5 ± 0.9
4g		H	41.8 ± 0.9	>100	5.3 ± 0.6
4h		H	6.8 ± 0.6	>100	8.6 ± 0.9
5d		COOMe	5.8 ± 0.5	7.6 ± 0.5	38 ± 5
5e		COOMe	9.27 ± 0.09	15.8 ± 5.3	25 ± 6
5f		COOMe	0.67 ± 0.03	2.4 ± 1.4	9.2 ± 0.5
6d		COOH	7.08 ± 0.01	3.8 ± 0.7	53 ± 12
6e		COOH	9.9 ± 0.7	>100	58 ± 9
6f		COOH	5.2 ± 0.1	>100	71 ± 12

methods. The inhibitory activity results are summarized in Table 1. Limited by solubility, only compounds with 50% inhibition below 100 μM were tested for IC₅₀ values.

2.3.1. Against LTA₄H

LTA₄H is a ubiquitously distributed 69 kDa zinc-containing cytosolic enzyme. It possesses both aminopeptidase activity and epoxide hydrolase activity [22], and the substrate binding pockets of these two activities are largely overlap. The activity assay of LTA₄H aminopeptidase is more sensitive to inhibitor binding and less expensive. When the weak inhibitors showed no reliable response in the epoxide hydrolase activity assay, the activity data of LTA₄H aminopeptidase

could be used to evaluate inhibitors' binding affinities. Using previously reported methods [23,24], we measured LTA₄H inhibitory activities of these compounds against both aminopeptidase and epoxide hydrolase, with bestatin as the positive control.

2.3.2. Against PLA₂

Testing inhibitory activity of the compounds on hnp-PLA₂ was based on a continuous fluorescence assay and carried out in a 96-well plate, using 1-anilinonaphthalene-8-sulfonate (ANS) as an interfacial probe [25]. 2-(1-Benzyl-5-methoxy-2-methyl-1H-indol-3-yl) acetamide (NI101), one reported hnp-PLA₂ inhibitor [26], was used as the positive control.

2.4. Structure activity relationships

As shown in Table 1, most compounds displayed more potent inhibitory activity against LTA₄H-h and hnpS-PLA₂, than compound **JMC08-4** did.

In the series of secondary amine compounds (compounds **4a–f**), compounds **4d** and **4f** show more potent inhibitory activity than the other compounds against LTA₄H-h and hnpS-PLA₂, with IC₅₀ values in the magnitude of 10 μM. These results suggested that compounds with longer hydrophobic groups would show higher activities than those with smaller ones, which were consistent with the molecular docking results. However, compounds **4g** and **4h** with nitro-substituted aromatic groups did not show good inhibition against LTA₄H because these groups were not fit well into the L-shaped narrow pocket.

Compounds **5d–f** showed stronger LTA₄H inhibitory activities than compounds **4a–h** due to their functional groups coordinating with metal atoms. The interaction between the zinc ion and oxygen atoms of the methyl propionate group may cause the enhancement of their binding affinities. In contrast, compounds **6d–f**, with the propionic acid group at position 2 showed little improvement in inhibitory activity against LTA₄H. Propionic acid group possessed both negative charge and atoms that could coordinate with metal atoms and was supposed to improve the binding affinity of compounds **6d–f**. It may have been caused by the intramolecular hydrogen bonds between the carboxyl group and the amino group.

Except for compound **5f**, all compounds with a substitution at position 2 had decreased inhibitory activity against PLA₂ about one order of magnitude. It is probably caused by unwanted collisions appeared between the inhibitors and some residues of PLA₂.

Though multi-target inhibitors generally do not have high affinities, they have potential beneficial effects, such as avoiding drug resistance and toxicity, and stabilizing “sick” cells [27]. For example, noncompetitive antagonists of N-methyl-D-aspartate receptors only inhibit multiple targets weakly, but exhibit advantages of reducing side effects [28–31]. We would therefore expect that these multi-target inhibitors may be more effective to control inflammation than single-target inhibitors. Of course, these compounds need to be further studied to characterize their overall effects at system level.

3. Conclusion

In this report, we aimed to identify potent dual function inhibitors by improving the binding affinities of a known moderate inhibitor of PLA₂ and LTA₄H, compound **JMC08-4**. Following docking result and structural information of this compound, two major optimizations were made: 1) hydrophobic group at position 5 of indole backbone was enlarged to increase the hydrophobic interaction; and 2) functional group that could coordinate with metal atoms was added at proper site. These optimizations turned out to be effective. Fourteen 5-HT analogs were successfully designed and synthesized. Most of these compounds showed more potent inhibitory activity against LTA₄H-h and hnpS-PLA₂ than compound **JMC08-4**. The best potential compound is **5f**, which inhibited LTA₄H-h with an IC₅₀ value of 2.4 ± 1.4 μM, increasing nearly three orders of magnitude than compound **JMC08-4**. It also improved inhibitory activity against hnpS-PLA₂, the IC₅₀ value is reach to 9.2 ± 0.5 μM.

4. Experimental

4.1. Chemistry

The reagents and solvents were commercially available and purified according to conventional methods. All reactions were monitored by thin layer chromatography; using silica gel 60 F-254

aluminum sheets and UV light (254 and 366 nm) for detection. All title compounds gave satisfactory ¹H NMR, ¹³C NMR, elemental analyses, and mass spectrometry analyses. The ¹H NMR and ¹³C NMR spectra were measured on a Bruker-400 M spectrometer using TMS as internal standard. Elemental analyses were performed on an ElementarVario EL instrument. Mass spectra were recorded on a VG-ZAB-MS spectrometer. Title compounds **4a–f** possessed a purity of >99.5% as verified by elemental analyses after comparison with calculated values.

4.1.1. General procedure for the synthesis of N-(butyloxycarbonyl)serotonin (**2**)

Serotonin hydrochloride (1.07 g, 5 mmol) was suspended in 10 mL of chloroform. Sodium bicarbonate (0.42 g, 10 mmol) in 7 mL of water, 1 g of sodium chloride, and di-tert-butyloxycarbonyl anhydride (1.1 g, 5 mmol) in 2.5 mL of chloroform were added. The mixture was refluxed for 3 h. The reaction mass was filtered and aqueous phase was washed with chloroform. The combined organic phases were washed with water and brine and dried over magnesium sulfate. The solvent was removed under reduced pressure to produce a solid product. Yield: 90%. ¹H NMR (CDCl₃): δ 7.97 [s (br), 1H], 7.26 (d, 1H), 7.01 (d + s, 2H), 6.80 (dd, 1H), 4.66 (s (br), 1H), 3.42 (t + m, 2H), 2.87 (t, 2H), 1.44 (s, 9H).

4.1.2. General procedure for the preparation of 1-(4-methoxyphenoxy)-2-nitrobenzene (**R_{h1}**)

A mixture of 4-methoxyphenol (2.0 g, 16 mmol) and anhydrous potassium hydroxide (1.0 g, 18 mmol) was heated at 150 °C and stirred. After the mixture melted (about 10 min), 1-chloro-2-nitrobenzene (2.0 g, 13 mmol) was added and the mixture was stirred at 170 °C for 2 h, while being monitored by TLC. When the reaction was complete, the reaction mass was poured into 50 mL of 3% potassium hydroxide aqueous solution and stirred at room temperature for about 2 h and then cooled to 4 °C. The solid was filtered with suction and washed with water. The crude product was recrystallized by ethanol to afford yellow solid 2.8 g (yield 90%). ¹H NMR (400 MHz, DMSO-d₆): δ 7.93 (1H, d), 7.44 (1H, m), 7.26–6.90 (6H, m), 3.82 (3H, s).

R_{g1} was synthesized by a similar procedure, which was monitored by TLC.

4.1.3. General procedure for the preparation of 4-(2-nitrophenoxy)phenol (**R_{h2}**)

R_{h1} (1.0 g, 4 mmol) was stirred in CH₂Cl₂ (20 mL) at 0 °C, and 0.57 mL (6 mmol) BBr₃ in 20 mL CH₂Cl₂ was slowly dropped into the solution above. After being stirred at room temperature for approximately 3 h, the mixture was poured into H₂O. The organic layer was washed with brine, dried over Na₂SO₄, and concentrated at reduced pressure. The residue was separated by flash chromatography (silica gel, the eluting solvent ranged from hexane to EtOAc/hexane = 1:2) as a yellow solid which weighed 0.87 g (yield 92%). ¹H NMR (400 MHz, DMSO-d₆): δ 9.50 (1H, s), 8.00 (1H, d), 7.60 (1H, m), 7.24 (1H, m), 6.97–6.80 (5H, m).

R_{g2} was synthesized by a similar procedure, which was monitored by TLC.

4.1.4. General procedure for the preparation of 1-(4-(2-bromoethoxy)phenoxy)benzene (**R_{pX}**)

The compound 4-phenoxyphenol (250 mg, 1.34 mmol) and anhydrous potassium hydroxide (150 mg, 2.66 mmol) were refluxed in 1,2-dibromoethane (2 mL) for 8 h. The mixture was washed with water and brine, and then dried over magnesium sulfate. After being concentrated at reduced pressure, the residue was separated by flash chromatography (silica gel, the eluting solvent ranged from hexane to EtOAc/hexane = 3:7) as a colorless

liquid and weighed 313 mg (yield 80%). ^1H NMR (400 MHz, CDCl_3): δ 7.31 (2H, t + s), 7.06 (1H, t), 6.98 (6H, m), 4.28 (2H, t), 3.65 (2H, t).

R_gX and **R_hX** were synthesized by a similar procedure, which was monitored by TLC.

4.1.5. General procedure for the preparation of tert-butyl 2-(5-(2-(4-phenoxyphenoxy) ethoxy)-1H-indol-3-yl) ethylcarbamate (**3f**)

A mixture of **2** (126 mg, 0.457 mmol) and **R_pX** (147 mg, 0.502 mmol) was stirred in anhydrous DMF (4 mL) in the presence of Cs_2CO_3 (140 mg, 0.430 mmol) at 90 °C for 8 h. The mixture was poured into ethyl acetate (20 mL) and washed with water and brine, then dried over magnesium sulfate. After being concentrated at reduced pressure, the residue was separated by flash chromatography (silica gel, the eluting solvent ranged from EtOAc/hexane = 1:9 to EtOAc/hexane = 2:3) as a colorless oil which weighed 120 mg (yield 59%). ^1H NMR (400 MHz, CDCl_3): δ 7.94 (s (br), 1H), 7.26 (m, 2H), 7.10 (s, 1H), 7.04 (m, 2H), 6.97 (m, 7H), 4.61 (s (br), 1H), 4.35 (dd, 4H), 3.45 (s (br), 2H), 2.91 (t, 2H), 1.43 (s, 9H).

The compounds **3a**, **3c**, and **3e–h** were synthesized by a similar procedure, which was monitored by TLC.

4.1.6. General procedure for the preparation of tert-butyl 2-(5-(naphthalen-2-ylmethoxy)-1H-indol-3-yl) ethylcarbamate (**3b**)

A mixture of compound **2** (292 mg, 1.06 mmol) and 2-(bromomethyl) naphthalene (141 mg, 1.00 mmol) was stirred in anhydrous DMF (4 mL) in the presence of Cs_2CO_3 (420 mg, 1.5 mmol) at room temperature for 8 h. The mixture was poured into ethyl acetate (20 mL) and washed with water and brine, then dried over magnesium sulfate. After being concentrated at reduced pressure, the residue was separated by flash chromatography (silica gel, the eluting solvent ranged from EtOAc/hexane = 1:9 to EtOAc/hexane = 2:3) as a colorless oil which weighed 290 mg (yield 66%). ^1H NMR (400 MHz, CDCl_3): δ 8.11 (s (br), 1H), 7.92 (s, 1H), 7.84 (m, 3H), 7.58 (d, 1H), 7.47 (m, 2H), 7.22 (d, 1H), 7.18 (s, 1H), 6.97 (m, 2H), 5.25 (s, 2H), 4.65 (s (br), 1H), 3.43 (d, 2H), 2.89 (t, 2H), 1.44 (s, 9H).

The compound **3d** was synthesized by a similar procedure, which was monitored by TLC.

4.1.7. General procedure for the preparation of 2-(5-(benzyloxy)-1H-indol-3-yl) ethanaminium chloride (**4a**)

The compound **3a** (183 mg, 0.5 mmol) was dissolved in anhydrous ethyl acetate saturated with hydrogen chloride (6 mL). A white precipitate appeared while the solution was stirred and bubbled with dry HCl gas for 45 min. The solid was filtered with suction and washed with ethyl acetate, and the crude product was recrystallized by water as a white powder which weighed 166 mg (yield 90%).

The compounds **4b–h** were synthesized by a similar procedure.

4.1.7.1. 2-(5-(Benzyloxy)-1H-indol-3-yl) ethanaminium chloride (4a**)**. White solid, yield 90%, ^1H NMR (400 MHz, DMSO- d_6): δ 10.87 (s, 1H), 8.16 (s, 3H), 7.49 (d, 2H), 7.29 (m, 6H), 6.82 (dd, 1H), 5.11 (s, 2H), 3.01 (m, 4H); ^{13}C NMR (DMSO- d_6): δ 23.08 (CH_2), 69.83 (O- CH_2), 101.74, 109.27, 111.81, 112.14, 124.04, 127.15 (C_{indole}), 127.60 (C_{ph}), 127.69, 128.31 (2C, C_{ph}), 131.57 (C_{indole}), 152.08 (O- C_{indole}); ESI-HRMS m/z 267.14865. Anal. Calcd for $\text{C}_{17}\text{H}_{19}\text{ClN}_2\text{O}$: C, 67.43; H, 6.32; N, 9.25. Found: C, 67.09; H, 6.27; N, 9.21.

4.1.7.2. 2-(5-(Naphthalen-2-ylmethoxy)-1H-indol-3-yl) ethanaminium chloride (4b**)**. White solid, yield 67%, ^1H NMR (400 MHz, DMSO- d_6): δ 10.88 (s, 1H), 8.15 (s, 3H), 8.04 (s, 1H), 7.93 (m, 3H), 7.63 (dd, 1H), 7.52 (m, 2H), 7.29 (dd, 2H), 7.21 (d, 1H), 6.87 (dd, 1H), 5.29 (s, 2H), 3.03 (m, 4H); ^{13}C NMR (DMSO- d_6): δ 23.09 (CH_2), 70.00 (O- CH_2), 101.95, 109.28, 111.84, 112.15, 124.09 (C_{indole}), 125.85, 125.98, 126.16, 126.23 (C_{N}), 127.18 (C_{indole}), 127.56, 127.75, 127.90

(C_{N}), 131.63 (C_{indole}), 132.46, 132.80, 135.38 (C_{N}), 152.09 (O- C_{indole}); ESI-HRMS m/z 317.16435. Anal. Calcd for $\text{C}_{21}\text{H}_{21}\text{ClN}_2\text{O}$: C, 71.48; H, 6.00; N, 7.94. Found: C, 71.01; H, 6.00; N, 7.83.

4.1.7.3. 2-(5-(Naphthalen-1-ylmethoxy)-1H-indol-3-yl) ethanaminium chloride (4c**)**. White solid, yield 89%, ^1H NMR (400 MHz, DMSO- d_6): δ 10.85 (s, 1H), 8.15 (d, 1), 7.92–8.02 (m, 6H), 7.72 (d, 1H), 7.55 (m, 3H), 7.36 (s, 1H), 7.28 (d, 1H), 7.21 (d, 1H), 6.86 (m, 1H), 5.29 (s, 2H), 3.01 (m, 4H); ^{13}C NMR (DMSO- d_6): δ 23.12 (CH_2), 69.50 (O- CH_2), 101.84, 109.29, 111.85, 112.17, 124.09 (C_{indole}), 125.36, 125.88, 126.17, 126.59 (C_{N}), 127.21 (C_{indole}), 127.56, 127.75, 127.90, 128.40, 131.23 (C_{N}), 131.65 (C_{indole}), 133.20 (C_{N}), 152.08 (O- C_{indole}); ESI-HRMS m/z 317.16454. Anal. Calcd for $\text{C}_{21}\text{H}_{21}\text{ClN}_2\text{O}$: C, 71.48; H, 6.00; N, 7.94. Found: C, 71.00; H, 6.02; N, 7.75.

4.1.7.4. 2-(5-(4-Phenylbenzyloxy)-1H-indol-3-yl) ethanaminium chloride (4d**)**. White solid, yield 91%, ^1H NMR (400 MHz, DMSO- d_6): δ 10.88 (s, 1H), 8.16 (s (br), 3H), 7.68 (dd, 4H), 7.59 (d, 2H), 7.47 (t, 2H), 7.37 (t, 1H), 7.28 (d, 1H), 7.21 (d, 2H), 6.84 (dd, 1H), 5.17 (s, 2H), 2.98 (m, 4H); ^{13}C NMR (DMSO- d_6): δ 23.11 (CH_2), 69.45 (O- CH_2), 101.76, 109.29, 111.81, 112.16, 124.06 (C_{indole}), 126.63 (4C, C_{ph}), 127.17 (C_{indole}), 127.56 (C_{ph}), 128.31, 128.91 (2C, C_{ph}), 131.59 (C_{indole}), 136.97, 139.47, 139.85 (C_{ph}), 152.05 (O- C_{indole}); ESI-HRMS m/z 343.18014. Anal. Calcd for $\text{C}_{23}\text{H}_{23}\text{ClN}_2\text{O}$: C, 72.91; H, 6.12; N, 7.39. Found: C, 72.89; H, 6.10; N, 7.27.

4.1.7.5. 2-(5-(3-Phenoxybenzyloxy)-1H-indol-3-yl) ethanaminium chloride (4e**)**. White solid, yield 88%, ^1H NMR (400 MHz, DMSO- d_6): δ 10.87 (s, 1H), 8.11 (s (br), 3H), 7.43 (m, 3H), 7.18 (m, 6H), 6.98 (dd, 3H), 6.80 (d, 1H), 5.11 (s, 2H), 2.96 (dd, 4H); ^{13}C NMR (DMSO- d_6): δ 23.07 (CH_2), 69.42 (O- CH_2), 101.95, 109.26, 111.77, 112.12 (C_{indole}), 117.40, 117.61 (C_{ph}), 118.69 (2C, C_{ph}), 122.47, 123.51 (C_{ph}), 124.10, 127.17 (C_{indole}), 129.99 (C_{ph}), 130.04 (2C, C_{ph}), 131.66 (C_{indole}), 140.10 (C_{ph}), 151.92 (O- C_{indole}), 156.47, 156.73 (O- C_{ph}); ESI-HRMS m/z 359.17468. Anal. Calcd for $\text{C}_{23}\text{H}_{23}\text{ClN}_2\text{O}_2$: C, 69.95; H, 5.87; N, 7.09. Found: C, 69.62; H, 5.86; N, 7.09.

4.1.7.6. 2-(5-(2-(4-Phenoxyphenoxy) ethoxy)-1H-indol-3-yl) ethanaminium chloride (4f**)**. White solid, yield 90%, ^1H NMR (400 MHz, DMSO- d_6): δ 10.88 (s, 1H), 8.14 (s, 3H), 7.32 (m, 3H), 7.32 (m, 3H), 7.20 (dd, 2H), 7.04 (m, 5H), 6.93 (d, 2H), 6.79 (dd, 1H), 4.00 (s, 4H), 3.03 (m, 4H); ^{13}C NMR (DMSO- d_6): δ 23.07 (CH_2), 66.96, 79.00 (O- CH_2), 101.37, 109.32, 111.65, 112.20 (C_{indole}), 115.78, 117.29, 120.69 (2C, C_{ph}), 122.59 (C_{ph}), 124.09, 127.19 (C_{indole}), 129.85 (2C, C_{ph}), 131.60 (C_{indole}), 149.53 (C_{ph}), 152.05 (O- C_{indole}), 154.77, 157.94 (O- C_{ph}); ESI-HRMS m/z 389.18504. Anal. Calcd for $\text{C}_{24}\text{H}_{25}\text{ClN}_2\text{O}_3$: C, 67.84; H, 5.93; N, 6.59. Found: C, 67.48; H, 5.94; N, 6.53.

4.1.7.7. 2-(5-(2-(4-(4-Nitrophenoxy) phenoxy) ethoxy)-1H-indol-3-yl) ethanaminium chloride (4g**)**. Pale yellow solid, yield 63%, ^1H NMR (400 MHz, DMSO- d_6): δ 10.88 (s, 1H), 8.26 (d, 2H), 8.21 (s, 3H), 7.28 (d, 1H), 7.13 (m, 8H), 6.80 (d, 1H), 4.36 (s, 4H), 3.03 (m, 4H); ^{13}C NMR (DMSO- d_6): δ 23.07 (CH_2), 66.85, 79.06 (O- CH_2), 101.38, 109.32, 111.64, 112.20 (C_{indole}), 116.11, 116.58, 121.97 (2C, C_{ph}), 124.09 (C_{indole}), 126.12 (2C, C_{ph}), 127.19, 131.61 (C_{indole}), 141.42, 147.42 (C_{ph}), 152.03 (O- C_{indole}), 154.97, 163.72 (O- C_{ph}); ESI-HRMS m/z 434.17094.

4.1.7.8. 2-(5-(2-(4-(2-Nitrophenoxy) phenoxy) ethoxy)-1H-indol-3-yl) ethanaminium chloride (4h**)**. Pale yellow solid, yield 74%, ^1H NMR (400 MHz, DMSO- d_6): δ 10.87 (s, 1H), 8.11 (m, 3H), 7.40 (q, 3H), 7.18 (m, 6H), 6.98 (dd, 3H), 6.81 (d, 1H), 5.11 (s, 4H), 3.00 (dd, 4H); ^{13}C NMR (DMSO- d_6): δ 23.06 (CH_2), 66.87, 79.06 (O- CH_2), 101.38, 109.32, 111.64, 112.20 (C_{indole}), 115.98 (2C, C_{ph}), 119.20 (C_{ph}), 120.71 (2C, C_{ph}), 123.17 (C_{ph}), 124.07 (C_{indole}), 125.45 (C_{ph}),

127.19, 131.61 (C_{indole}), 134.78, 140.50, 148.49, 150.48 (C_{ph}), 152.03 (O–C_{indole}), 155.48 (O–C_{ph}); ESI-HRMS *m/z* 434.17016.

4.1.8. General procedure for the preparation of methyl 3-(2-(5-(4-phenylbenzyloxy)-1H-indol-3-yl) ethylamino) propanoate (**5d**)

Compound **4d** (378 mg, 0.679 mmol), Cs₂CO₃ (111 mg, 0.575 mmol), and methyl acrylate (68 μ L, 0.68 mmol) were stirred in methanol (2 mL) at room temperature for 12 h. The mixture was poured into ethyl acetate and washed with water and brine, then dried over magnesium sulfate. After being concentrated at reduced pressure, the residue was separated by flash chromatography (silica gel, the eluting solvent ranged from CH₂Cl₂ to MeOH/CH₂Cl₂ = 1:5) as a colorless oil which weighed 226 mg (yield 78%). The compounds **5e** and **5f** were synthesized by a similar procedure.

4.1.8.1. Methyl 3-(2-(5-(4-phenylbenzyloxy)-1H-indol-3-yl) ethylamino) propanoate (5d**)**. Yield 78%. ¹H NMR (400 MHz, CDCl₃): δ 8.05 (s, 1H), 7.59 (m, 6H), 7.44 (t, 2H), 7.34 (t, 1H), 7.25 (m, 1H), 7.17 (d, 1H), 6.97 (m, 2), 5.15 (s, 2H), 3.62 (s, 3H), 2.93 (m, 6H), 2.50 (t, 2H); ¹³C NMR (CDCl₃): δ 25.79, 34.63, 44.98, 49.76 (CH₂), 51.55, 70.77 (O–C), 102.46, 111.86, 112.92, 113.73, 122.82 (C_{indole}), 127.13 (4C, C_{ph}), 127.29 (C_{indole}), 127.87 (C_{ph}), 128.10, 128.78 (2C, C_{ph}), 131.78 (C_{indole}), 136.75, 140.75, 140.93 (C_{ph}), 153.11 (O–C_{indole}), 173.17 (C=O); ESI-HRMS *m/z* 429.21640 [M + H].

4.1.8.2. Methyl 3-(2-(5-(3-phenoxybenzyloxy)-1H-indol-3-yl) ethylamino) propanoate (5e**)**. Yield 72%. ¹H NMR (400 MHz, CDCl₃): δ 8.03 (s, 1H), 7.28 (m, 5H), 7.11 (dd, 3H), 6.96 (t, 5H), 5.08 (s, 2H), 3.62 (s, 3H), 2.91 (m, 6H), 2.50 (t, 2H); ¹³C NMR (CDCl₃): δ 25.73, 34.59, 44.96, 49.71 (CH₂), 51.56, 70.61 (O–C), 102.53, 111.84, 112.91, 113.67 (C_{indole}), 117.94, 118.09 (C_{ph}), 118.94 (2C, C_{ph}), 122.28, 122.83 (C_{ph}), 123.28, 127.83 (C_{indole}), 129.75 (2C, C_{ph}), 129.83 (2C, C_{ph}), 131.79 (C_{indole}), 139.79 (C_{ph}), 152.94 (O–C_{indole}), 156.13, 156.45 (O–C_{ph}), 173.17 (C=O); ESI-HRMS *m/z* 445.21206 [M + H].

4.1.8.3. Methyl 3-(2-(5-(4-phenylbenzyloxy)-1H-indol-3-yl) ethylamino) propanoate (5f**)**. Yield 69%. ¹H NMR (400 MHz, CDCl₃): δ 8.03 (s, 1H), 7.27 (m, 3H), 7.12 (d, 1H), 6.97 (m, 9H), 4.35 (dq, 4H), 3.62 (s, 3H), 2.94 (m, 6H), 2.51 (t, 2H); ¹³C NMR (CDCl₃): δ 25.74, 34.60, 44.97, 49.77 (CH₂), 51.57, 67.36, 67.64 (O–C), 102.42, 111.88, 112.91, 113.70 (C_{indole}), 115.88, 117.68, 120.78 (2C, C_{ph}), 122.48 (C_{ph}), 122.86, 127.86 (C_{indole}), 129.62 (2C, C_{ph}), 131.86 (C_{indole}), 150.46 (C_{ph}), 152.89 (O–C_{indole}), 155.08, 158.44 (O–C_{ph}), 173.17 (C=O); ESI-HRMS *m/z* 475.22250 [M + H].

4.1.9. General procedure for the preparation of 3-(2-(5-(4-phenylbenzyloxy)-1H-indol-3-yl) ethylamino) propanoic acid (**6d**)

The compound **5d** (226 mg, 0.527 mmol) and sodium hydroxide (80 mg, 2 mmol) were dissolved in a mixture solution (acetone and water, v:v = 3:1) and stirred at room temperature for 12 h. The solution was acidified to pH = 1 with 1 M hydrochloric acid and extracted with ethyl acetate three times. The organic layer was washed with brine, then dried over Na₂SO₄, concentrated at reduced pressure, and the residue was separated by flash chromatography (silica gel, the eluting solvent ranged from CH₂Cl₂/MeOH = 5:1 to MeOH) as white solid which weighed 107 mg (yield 47%). ¹H NMR (400 MHz, DMSO-d₆): δ 10.84 (s, 1H), 7.68 (t, 4H), 7.58 (d, 2H), 7.47 (t, 2H), 7.37 (t, 1H), 7.28 (d, 2H), 7.20 (s, 1H), 6.85 (d, 1H), 5.17 (s, 2H), 3.15 (d, 4H), 3.04 (m, 2H), 2.68 (t, 2H); ¹³C NMR (DMSO-d₆): δ 21.89, 30.61, 42.66, 47.11 (CH₂), 69.50 (O–C), 101.90, 109.14, 111.82, 112.15, 124.03 (C_{indole}), 126.62 (4C, C_{ph}), 127.12 (C_{indole}), 127.42 (C_{ph}), 128.30, 128.91 (2C, C_{ph}), 131.60 (C_{indole}), 136.97, 139.48, 139.84 (C_{ph}), 152.06 (O–C_{indole}), 172.08 (C=O); ESI-HRMS *m/z* 415.20104 [M + H].

The compounds **6e** and **6f** were synthesized by a similar procedure.

4.1.9.1. 3-(2-(5-(3-Phenoxybenzyloxy)-1H-indol-3-yl) ethylamino) propanoic acid (6e**)**. White solid, yield 37%, ¹H NMR (400 MHz, DMSO-d₆): δ 10.77 (s, 1H), 7.39 (dd, 3H), 7.24 (m, 2H), 7.13 (dd, 4H), 7.01 (d, 2H), 6.94 (d, 1H), 6.78 (dd, 1H), 5.10 (s, 2H), 2.98 (t, 4H), 2.89 (m, 2H), 2.20 (t, 2H); ¹³C NMR (DMSO-d₆): δ 23.37, 44.32, 47.18 (CH₂), 69.36 (O–C), 101.93, 110.35, 111.70, 112.03 (C_{indole}), 117.40, 117.60 (C_{ph}), 118.67 (2C, C_{ph}), 122.44, 123.49 (C_{ph}), 123.82, 127.24 (C_{indole}), 129.97 (C, C_{ph}), 130.03 (3C, C_{ph}), 131.62 (C_{indole}), 140.16 (C_{ph}), 151.80 (O–C_{indole}), 156.48, 156.71 (O–C_{ph}), 173.66 (C=O); ESI-HRMS *m/z* 431.19692 [M + H].

4.1.9.2. 3-(2-(5-(2-(4-Phenoxyphenoxy) ethoxy)-1H-indol-3-yl) ethylamino) propanoic acid (6f**)**. White solid, yield 40%, ¹H NMR (400 MHz, DMSO-d₆): δ 10.75 (s, 1H), 7.35 (t, 2H), 7.25 (d, 1H), 7.17 (s, 1H), 7.05 (d, 6H), 6.93 (d, 2H), 6.77 (d, 1H), 4.32 (s, 4H), 2.99 (t, 2H), 2.89 (t, 4H), 2.19 (t, 2H); ¹³C NMR (DMSO-d₆): δ 23.44, 30.64, 43.58, 44.32, 47.23 (CH₂), 66.85, 66.98 (O–C), 110.43, 111.56, 112.09, 114.53 (C_{indole}), 115.78, 117.29, 120.69 (2C, C_{ph}), 122.56 (C_{ph}), 123.82, 127.29 (C_{indole}), 129.85 (2C, C_{ph}), 132.60 (C_{indole}), 149.51 (C_{ph}), 151.94 (O–C_{indole}), 154.45, 157.95 (O–C_{ph}), 173.03 (C=O); ESI-HRMS *m/z* 461.20681 [M + H].

4.2. Biochemical methods

4.2.1. Inhibition assay of aminopeptidase activity of LTA₄H-h

The aminopeptidase activity was determined as described previously [17]. The enzyme was incubated at 37 °C in 96-well microtiter plates in 50 mM Tris–HCl, pH 8.0, containing 100 manacles. The reactions were initiated by the addition of Ala-p-nitroanilides as substrate. The formation of p-NA was then monitored at 37 °C for 15 min at 405 nm using a multi well ultraviolet spectrometer (Spectra Max 190, Molecular Devices). For IC₅₀ determinations, 0.125 mM substrate was used. All reactions containing inhibitors were performed after incubating inhibitors with enzymes for 2 min. The final concentration of DMSO was 5% (v/v). The initial reaction rates at different inhibitor concentrations were used for calculating the IC₅₀ values.

4.2.2. Inhibition assay of hydrolase activity of LTA₄H-h

The epoxide hydrolase activity was determined using an ELISA assay to quantify the amount of LTB₄ [17]. LTA₄ methyl ester (Cayman Chemical) was hydrolyzed in cold acetone with 50 mM NaOH (20%, v/v) under an inert atmosphere of nitrogen. To determine the effect of small molecules on the epoxide hydrolase activity of LTA₄H-h, 300 ng of enzyme was incubated with compounds in reaction buffer (10 mM sodium phosphate, pH 7.4, 4 mg/mL BSA, 5% v/v DMSO) for 15 min at 37 °C. The reaction was initiated by the addition of LTA₄ (150 nM final assay concentration) and incubated for another 10 min at 37 °C. 25 μ L sample was added to 500 μ L assay buffer without DMSO to stop the reaction and then further diluted fivefold. LTB₄ was quantified in the diluted sample by a commercially available LTB₄ ELISA kit (Cayman Chemical).

4.2.3. Inhibitory activity assay of hnpS-PLA₂

Reaction buffer (150 μ L in 50 mM Tris–HCl, pH 8.0, 100 mM NaCl, 5 μ g/mL bovine serum albumin (BSA), and 10 mM ANS), 20 μ L of substrate stock solution [2 mM 1,2-dimyristoyl-sn-glycero-3-phosphocholine (DMPC)] and 2 mM of sodium deoxycholate in water, sonicated for 2 min, 10 μ L of CaCl₂ stock solution (100 mM), and 10 μ L of inhibitor stock solution (dissolved in DMSO) were incubated at 25 °C for 10 min. Reactions were started by adding 10 μ L of hnpS-PLA₂ stock solution (3.2 μ g/mL) and monitored by

excitation at 377 nm and emission at 470 nm using a multiwell fluorometer (SYNERGY 4, BioTek). The initial reaction rates at different inhibitor concentrations were used for calculating the IC₅₀ value.

4.3. Molecular docking

All new compounds were docked into the substrate binding site of LTA₄H and PLA₂. The crystal structures of the two enzymes were retrieved from the Protein Data Bank (LTA₄H, PDB ID: 3CHR [32]; PLA₂, PDB ID: 1DB4 [33]). Molecular docking with flexible ligands and rigid receptors was performed using AutoDock 4.00 software. Conformational searching used the Lamarckian genetic algorithm (LGA). Number of GA runs was set to 100, population size to 300, maximum number of energy evaluations to 12,500,000, and maximum number of generations to 27,000. Docked states of each small molecule were clustered with rms of 2.0 Å. The lowest energy conformation in the largest cluster was taken as the binding state.

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Appendix A. Supplementary material

Supplementary material associated with this article can be found in the online version, at <http://dx.doi.org/10.1016/j.ejmech.2012.10.057>. These data include MOL files and InChIKeys of the most important compounds described in this article.

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