

## Cycloaddition Reactions

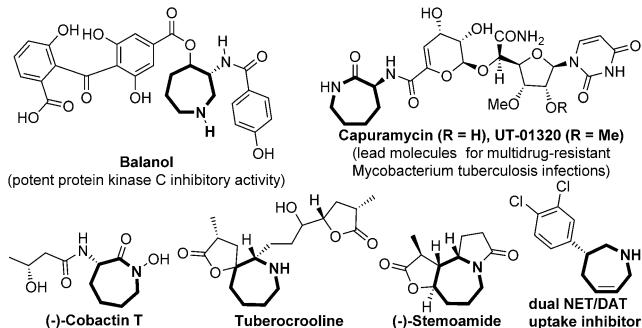
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## [5+2] Cycloaddition of 2-(2-Aminoethyl)oxiranes with Alkynes via Epoxide Ring-Opening: A Facile Access to Azepines

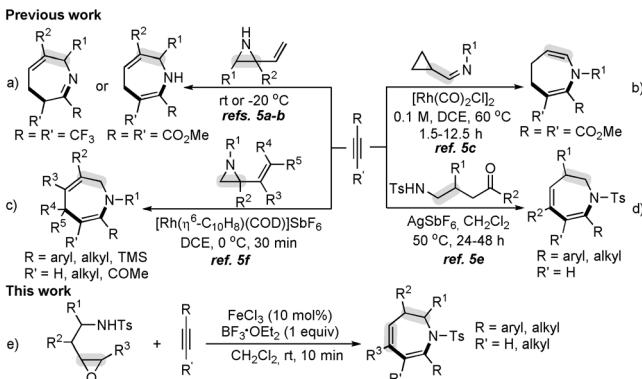
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**Abstract:** A new  $\text{FeCl}_3$  and  $\text{BF}_3\text{-OEt}_2$  co-catalyzed tandem hetero-[5+2] cycloaddition of 2-(2-aminoethyl)oxiranes with a wide range of alkynes, including terminal alkynes and alkyl-substituted internal alkynes is presented. This is the first example of rapid and facile production of diverse 2,3-dihydro-1*H*-azepines through a sequence of epoxide ring-opening, annulation, and dehydroxylation with broad substrate scope and exquisite selectivity control.

Access to seven-membered ring systems, especially seven-membered *N*-heterocyclic systems, have gained growing interest among the synthetic community owing to the continuing identification of the seven-membered-ring-containing natural products with the appealing pharmacological and pesticidal activities.<sup>[1,2]</sup> Attractive *N*-heterocyclic systems include azepine derivatives,<sup>[3]</sup> which are unique structural motifs presented in many biologically active natural products and pharmaceuticals, such as balanol,<sup>[3e,f]</sup> capuramycin and its derivatives,<sup>[3g]</sup> (-)-cobactin T,<sup>[3h-i]</sup> tuberocrooline,<sup>[3j,k]</sup> (-)-stemoamide,<sup>[3l-n]</sup> and (*S*)-3-(3,4-dichlorophenyl)-2,3,4,7-tetrahydro-1*H*-azepine<sup>[3o]</sup> (Scheme 1). Accordingly, considerable efforts has been devoted to the development of new efficient methods for assembling azepines.<sup>[4-6]</sup> Among them, cycloaddition reactions,<sup>[1,2,4]</sup> including the hetero-[5+2] cycloaddition reactions with alkynes,<sup>[5]</sup> are particularly fascinating for straightforward building the azepine frameworks. Despite the obvious synthetic utility, approaches through the hetero-[5+2] cycloaddition reactions with alkynes are less abundant and remain limited to the substrate scope with regard to both the five-atom units and the alkynes.<sup>[5]</sup> Pioneering work of the [5+2] cycloaddition reaction relied on the use of 2,3-divinylaziridine as a five-atom unit under metal-free conditions, but only two electron-poor alkynes were investigated for the synthesis of two azepines (Scheme 2a).<sup>[5a,b]</sup> To access



Scheme 1. Examples of important azepine derivatives.



Scheme 2. [5+2] Cycloaddition for the synthesis of azepines.

more azepines, Wender and co-workers first developed a rhodium-catalyzed hetero-[5+2] cycloaddition of cyclopropyl imines with dimethyl acetylenedicarboxylate (Scheme 2b).<sup>[5c]</sup> Recently, we reported a new silver-catalyzed tandem [5+2] cycloaddition of  $\gamma$ -amino ketones with simple terminal alkynes, which provided a practical access to 2,3-dihydro-1*H*-azepines (Scheme 2d).<sup>[5e]</sup> Interestingly, Zhang and co-workers recently documented a new rhodium catalysis to extend the [5+2] cycloaddition to various vinylaziridines and alkynes leading to 2,5-dihydro-1*H*-azepines (Scheme 2c).<sup>[5f]</sup> Thus, new efficient strategies, especially involving the use of new readily available five-atom units, for the intermolecular [5+2] cycloaddition with alkynes would be warmly welcomed.

Herein, we report a new iron and  $\text{BF}_3\text{-OEt}_2$  co-catalyzed intermolecular [5+2] cycloaddition of 2-(2-aminoethyl)oxiranes with alkynes for producing diverse 2,3-dihydro-1*H*-azepines (Scheme 2e); this reaction proceeds via epoxide ring-opening by C–O bond cleavage,<sup>[7,8]</sup> annulation and dehydroxylation cascades, and represents the first example

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of utilizing oxiranes for the [5+2] intermolecular cycloaddition with alkynes.

We started our studies to explore optimal reaction conditions for the [5+2] cycloaddition between 4-methyl-*N*-(2-(oxiran-2-yl)ethyl)benzenesulfonamide (**1a**) and phenylacetylene (**2a**) (Table 1). We were pleased to find that

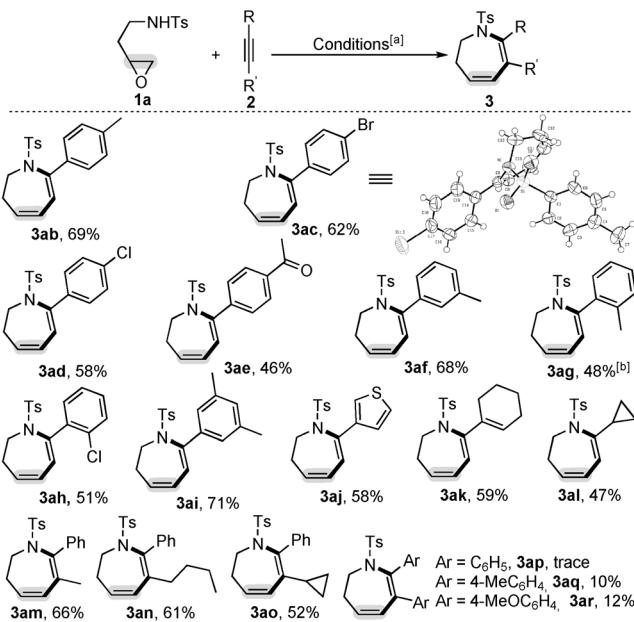
**Table 1:** Screening of optimal reaction conditions.<sup>[a]</sup>

Entry	Variation from the standard conditions	Yield <sup>[d]</sup> [%]
1 <sup>[b]</sup>	none	74
2	without $\text{BF}_3\text{-OEt}_2$	51
3	without $\text{FeCl}_3$	60
4	$\text{FeCl}_3$ (5 mol %)	68
5	$\text{FeCl}_3$ (15 mol %)	61
6	$\text{BF}_3\text{-OEt}_2$ (0.8 equiv)	69
7	$\text{BF}_3\text{-OEt}_2$ (1.2 equiv)	68
8	$\text{FeCl}_2$ instead of $\text{FeCl}_3$	57
9	$\text{Fe}(\text{OTf})_3$ instead of $\text{FeCl}_3$	67
10	$\text{CuCl}_2$ instead of $\text{FeCl}_3$	54
11	$\text{InCl}_3$ or $\text{YbCl}_3$ instead of $\text{FeCl}_3$	49
12	at 0°C	71
13	$\text{CH}_2\text{ClCH}_2\text{Cl}$ instead of $\text{CH}_2\text{Cl}_2$	70
14	MeCN instead of $\text{CH}_2\text{Cl}_2$	< 5
15 <sup>[c]</sup>	none	60

[a] Reaction conditions: **1a** (0.1 mmol), **2a** (0.2 mmol),  $\text{FeCl}_3$  (10 mol %),  $\text{BF}_3\text{-OEt}_2$  (1 equiv),  $\text{CH}_2\text{Cl}_2$  (2 mL), room temperature, 10 min. [b] The reaction gave the same yield under air or argon atmosphere. [c] **1a** (1 mmol) and 30 min. A side-product, 2-(phenylacetynyl)-1-tosylpyrrolidine (**4aa**), was obtained in 15% yield. [d] Yield of isolated product.

treatment of oxirane **1a** with alkyne **2a**, 10 mol % of  $\text{FeCl}_3$  and 1 equiv of  $\text{BF}_3\text{-OEt}_2$  in  $\text{CH}_2\text{Cl}_2$  at room temperature for 10 min was preferred to furnish the desired product **3aa** in 74% yield (entry 1). The reaction was successful when using  $\text{FeCl}_3$  or  $\text{BF}_3\text{-OEt}_2$  alone, albeit giving diminishing yields (entries 2 and 3). A screen of the amount of both  $\text{FeCl}_3$  and  $\text{BF}_3\text{-OEt}_2$  revealed a combination of 10 mol % of  $\text{FeCl}_3$  and 1 equiv of  $\text{BF}_3\text{-OEt}_2$  as the best option (entries 1 and 4–7). A series of other Lewis acids, such as  $\text{FeCl}_2$ ,  $\text{Fe}(\text{OTf})_3$ ,  $\text{CuCl}_2$ ,  $\text{InCl}_3$  and  $\text{YbCl}_3$ , were subsequently examined: each of which exhibited catalytic activity, but was less efficient than  $\text{FeCl}_3$  (entries 1 and 8–11). A lower reaction temperature (0°C) slightly affected the reaction in terms of yield (entry 12). While  $\text{CH}_2\text{ClCH}_2\text{Cl}$  was proved to be a highly reactive medium (entry 13), MeCN had a rather lower reactivity (entry 14). Gratifyingly, the reaction scale up to 1 mmol of oxirane **1a** was successfully performed, providing **3aa** in moderate yield (entry 15).

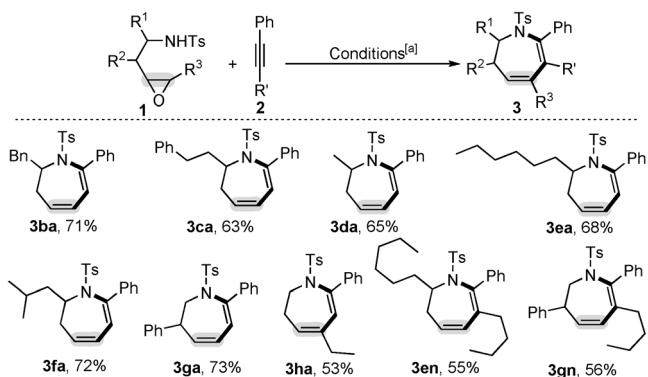
With the optimal reaction conditions in hand, we set out to investigate the scope of this intermolecular [5+2] cycloaddition protocol with respect to alkenes (**2**) (Scheme 3). Initial screening revealed that the optimal conditions were compatible with a wide range of terminal alkynes, namely arylalkynes (**2b–i**), heteroarylalkyne (**2j**), 3-enyne (**2k**) and aliphatic alkyne (**2l**). In the presence of oxirane **1a**,  $\text{FeCl}_3$  and



**Scheme 3.** Variation of the alkynes (**2**). [a] Reaction conditions: **1** (0.2 mmol), **2** (0.4 mmol),  $\text{FeCl}_3$  (10 mol %),  $\text{BF}_3\text{-OEt}_2$  (1 equiv; 0.2 mmol),  $\text{CH}_2\text{Cl}_2$  (2 mL), room temperature and 10 min. [b] A side product, 1-(*o*-tolyl)-2-(1-tosylpyrrolidin-2-yl)ethan-1-one (**5ag**), was obtained in 23% yield.

$\text{BF}_3\text{-OEt}_2$ , several substituents, including Me, Br, Cl, and MeCO groups, on the aryl ring were well tolerated, and their nature of electron and position had an obviously effect on the yields (**3ab–ah**). While alkyne **2b** with an electron-donating Me group on the phenyl ring delivered **3ab** in 69% yield, alkyne **2e** having an electron-withdrawing MeCO group led a decrease in the yield (46%; **3ae**). Alkyne **2b** with a high active Br group was also converted to **3ab** in 62% yield.<sup>[9]</sup> In the case of methyl-substituted arylalkynes **2b**, **2f** and **2g**, the reactivity decreased from *para* to *meta* to *ortho* substitution in terms of yields (**3ab**, **3af** and **3ag**). Gratifyingly, diMe-substituted arylalkyne **2i** or 3-ethynylthiophene **2j** were viable for producing the corresponding products **3ai** and **3aj**. The optimal conditions were found to be tolerated the alkene and the cyclopropyl ring, thus giving products **3ak** and **3al** in moderate yields. Encouraged by the resulted described above, a number of internal alkynes were examined (**3am–ar**). Alkyl-substituted internal alkynes **2m–o** were viable substrates for the reaction, but 1,2-diaryllalkynes **2p–r** showed lower reactivity: 1,2-Diphenylalkyne **2p** had no reactivity, and other alkynes, 1,2-di-*p*-tolylethyne **2q** and 1,2-bis(4-methoxyphenyl)ethyne **2r**, delivered **3aq** and **3ar** in lower yields.

We next turned our attention to explore the generality of 2-(2-aminoethyl)oxiranes **1** in the presence of alkyne **2a**,  $\text{FeCl}_3$  and  $\text{BF}_3\text{-OEt}_2$  (Scheme 4). The optimal conditions were applicable to a wide range of 2-(2-aminoethyl)oxiranes (**1**) bearing diverse substituents at the different position (**3ba–ha**). 2-Benzyl-substituted 2-(2-aminoethyl)oxirane **1b** could be smoothly converted into the desired product **3ba** in 71% yield. PhCH<sub>2</sub>CH<sub>2</sub>-substituted 2-(2-aminoethyl)oxirane **1c** and Me-substituted 2-(2-aminoethyl)oxirane **1d** were suitable for the synthesis of **3ca** and **3da** in moderate yields. Using other

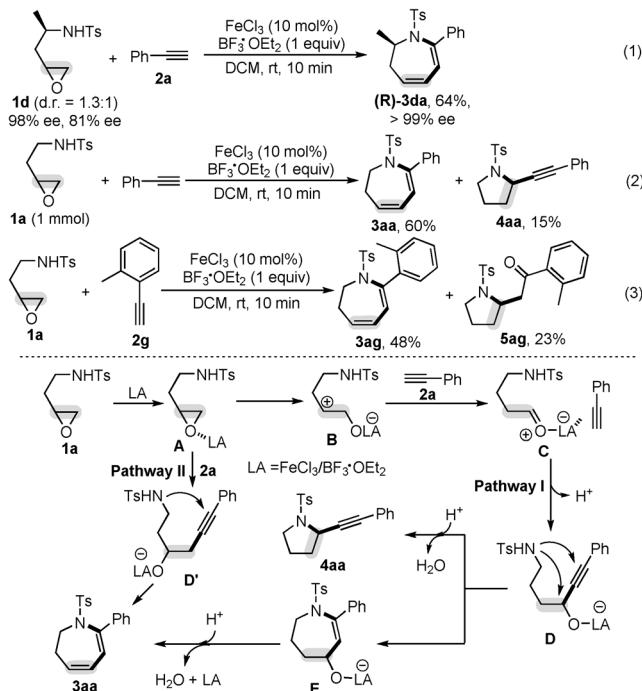


**Scheme 4.** Variation of the 2-(2-aminoethyl)oxiranes 1. [a] Reaction conditions: see Table 1 and Scheme 3.

2-alkyl-substituted oxirane **1e** and **1f** successfully assembled **3ea** and **3fa**. For 2-(2-aminoethyl)oxirane **1g** having a Ph group on the 2-position of the ethyl moiety, the reaction generated **3ga** in 73 % yield. Pleasingly, 2-(2-aminoethyl)oxirane **1h** with an ethyl group on the 3-position of the oxirane moiety was a suitable substrate for the cycloaddition (**3ha**). Substrates **1e** and **1g** also exhibited high reactivity with internal alkyne **1n**, thus building **3en** and **3gn** in 55 % and 56 % yield, respectively.

Gratifyingly, (*R*)-**1d** with 1.3:1 d.r. could be smoothly converted into the desired product (*R*)-**3da** in 64 % yield with >99 % ee [Eq. (1); Scheme 5]. Notably, two five-membered-ring side-products **4aa** [Eq. (2); Scheme 5] and **5ag** [Eq. (3); Scheme 5] were obtained (Table 1 and Scheme 3).

Consequently, the possible mechanisms outlined in Scheme 5 were proposed to understand the current tandem [5+2] cycloaddition reaction.<sup>[7,8]</sup> Both FeCl<sub>3</sub> and BF<sub>3</sub>·OEt<sub>2</sub> act



**Scheme 5.** Control experiments and possible reaction mechanisms.

as Lewis acids to form the carbocation intermediate **B** by coordinating with the oxygen atom in 2-(2-aminoethyl)oxirane **1a**,<sup>[8]</sup> which then undergoes ring-opening of the intermediate **B** and complex with alkyne **2a** afford the oxocarbenium ion intermediate **C** (Pathway I). Electrophilic *anti*-addition of the intermediate **C** across the C–C triple bond in alkyne **2a** selectively gives the vinyl cation intermediate **D**, followed by annulation to produce the seven-membered ring intermediate **E**, which is supported by the formation of the five-membered-ring side-products **4aa** [Eq. (2)] and **5ag** [Eq. (3)]. Finally, elimination of the intermediate **E** assembles the desired product **3aa** and releases H<sub>2</sub>O.

Although it is difficult to synthesize the five-membered-ring side-products **4aa** [Eq. (2)] and **5ag** [Eq. (3)] from intermediate **D'**, we cannot rule out the possible Pathway II. Notably, the results show that using a combination of FeCl<sub>3</sub> and BF<sub>3</sub>·OEt<sub>2</sub> as Lewis acids is necessary to offer the desired products **3** with high yields, and it might be because FeCl<sub>3</sub> can promote the ring-opening process in this tandem cycloaddition process, and a stoichiometric amount of BF<sub>3</sub>·OEt<sub>2</sub> can make the reaction faster and promoted the final elimination process.

In summary, we have developed a novel tandem hetero-[5+2] cycloaddition of 2-(2-aminoethyl)oxiranes with alkynes for the synthesis of 2,3-dihydro-1*H*-azepines by means of the FeCl<sub>3</sub> and BF<sub>3</sub>·OEt<sub>2</sub> co-catalysis. The reaction features a broad scope with respect to a wide range of both substituted 2-(2-aminoethyl)oxirane and alkynes. Importantly, the reaction employs two simple and inexpensive Lewis acids, FeCl<sub>3</sub> and BF<sub>3</sub>·OEt<sub>2</sub>, as co-catalysts to achieve exquisite regio- and chemocontrol, thus provides a new utilization of oxiranes in the intermolecular cycloaddition reaction for the rapid and practical construction of diverse 2,3-dihydro-1*H*-azepines. Further studies on the applications of this tandem cycloaddition strategy in heterocycle synthesis are currently underway in our laboratory.

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**Keywords:** azepines · cycloaddition · Lewis acids · oxiranes · ring opening

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## Communications

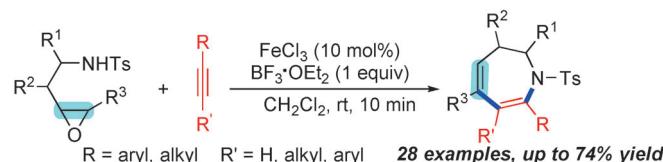


## Cycloaddition Reactions

C. Hu, R.-J. Song, M. Hu, Y. Yang,  
J.-H. Li,\* S. Luo\*

## [5+2] Cycloaddition of 2-(2-

Aminoethyl)oxiranes with Alkynes via Epoxide Ring-Opening: A Facile Access to Azepines



**Access all azepines:** An  $\text{FeCl}_3$  and  $\text{BF}_3\cdot\text{OEt}_2$  co-catalyzed tandem hetero-[5+2] cycloaddition of 2-(2-aminoethyl)-oxiranes with alkynes is presented. This method provides a rapid and practical

access to 2,3-dihydro-1*H*-azepines with exquisite chemo- and regiocontrol. The reaction is simple and has broad substrate scope and excellent functional-group tolerance.

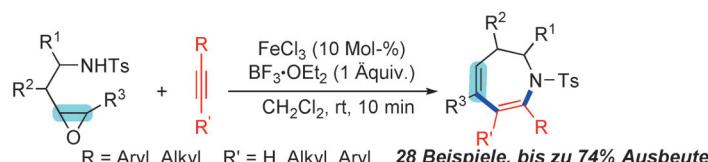


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und praktischen Zugang zu 2,3-Dihydro-1*H*-azepinen mit exzellenter Chemo- und Regiokontrolle. Die Reaktion ist einfach, hat einen breiten Substratbereich und toleriert zahlreiche funktionelle Gruppen.

## Communications

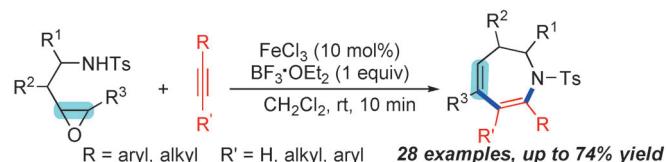


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