

# A Ruthenium-Mediated Asymmetric Hydrogenation Approach to the Synthesis of Discodermolide Subunits

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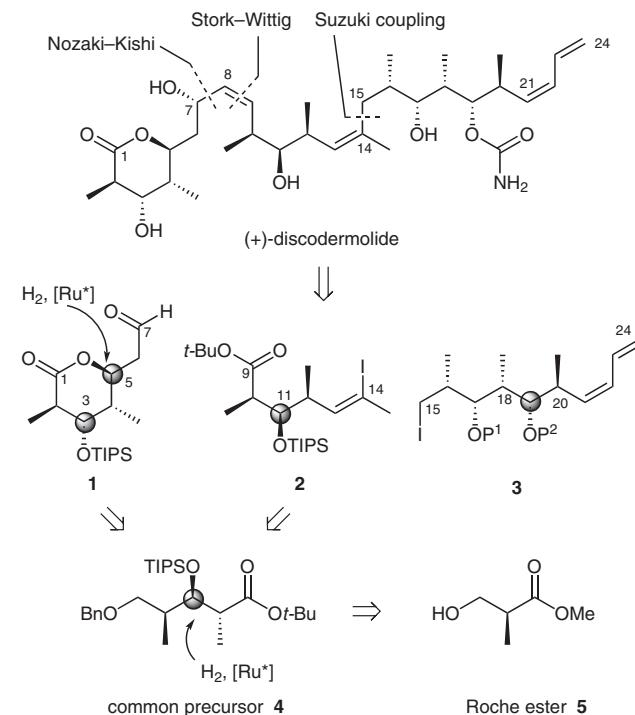
**Abstract:** The C1–C7 and C9–C14 subunits of (+)-discodermolide have been synthesized using ruthenium-SYNPHOS-mediated asymmetric hydrogenation reactions of  $\beta$ -keto esters to set the C3, C5 and C11 hydroxy-bearing stereocenters with very high levels of diastereoselectivity.

**Key words:** asymmetric catalysis, hydrogenation, ruthenium, total synthesis, natural products

The polypropionate-derived marine metabolite discodermolide was first isolated by Gunasekera and co-workers<sup>1</sup> in 1990 from the Caribbean deep-sea sponge *Discodermia dissoluta*. Preliminary biological evaluations showed this compound to possess potent immunosuppressive activity.<sup>2</sup> It was later demonstrated that discodermolide exhibited excellent microtubule-stabilizing capabilities<sup>3</sup> as well as synergism with taxol.<sup>4</sup> This promising biological profile and the potential as chemotherapeutic agent<sup>5</sup> have prompted considerable synthetic efforts toward the total synthesis of discodermolide.<sup>6</sup>

As part of our ongoing projects involving the synthesis of biologically relevant natural products via ruthenium-promoted asymmetric hydrogenation,<sup>7</sup> we were particularly interested in the linear polypropionate backbone of discodermolide, because the C2–C4, C10–C12, and C18–C20 motifs appeared ideally suited for construction through hydrogenation reactions. From a retrosynthetic point of view, all of the reported total syntheses disconnect discodermolide into three major fragments of equal complexity, each subunit possessing a *syn,anti* methyl-hydroxy–methyl stereotriad. On this basis, we chose to disconnect discodermolide into compounds **1**, **2**, and **3**, corresponding to C1–C7,<sup>8</sup> C9–C14 and C15–C24 fragments of the natural product, as depicted in Scheme 1. Formation of the C14–C15 bond would result from a Suzuki coupling reaction between (*Z*)-alkenyl iodide **2** and compound **3**. The C7–C8 bond would be created by a Nozaki–Kishi reaction between aldehyde **1** and a (*Z*)-vinyl iodide derived from an ester at C9.

Herein we report the preparation of key discodermolide intermediates **1** and **2**. In our retrosynthetic plan, both fragments would be prepared from a common precursor **4**

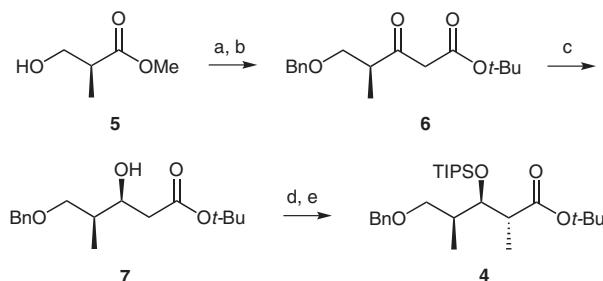


Scheme 1 Retrosynthetic analysis for (+)-discodermolide

which in turn would be derived from methyl (*S*)-3-hydroxy-2-methylpropionate (Roche ester) **5**. Among the seven stereocenters of the target subunits, three would be created via ruthenium-mediated asymmetric hydrogenation<sup>9</sup> of  $\beta$ -keto esters using the atropisomeric ligand SYNPHOS<sup>10</sup> as the chiral diphosphine.

The common precursor **4** was readily prepared in five steps starting from commercially available Roche ester **5**. After protection of the hydroxy function as a benzyl ether, subsequent chain extension with lithio *tert*-butyl acetate<sup>11</sup> delivered the  $\beta$ -keto ester **6** required for the asymmetric hydrogenation reaction (Scheme 2).

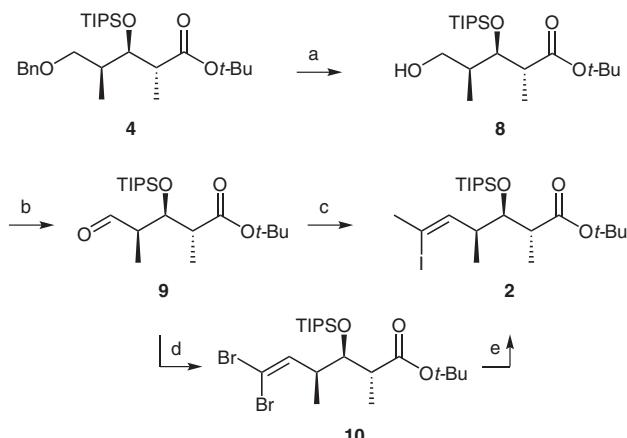
The stereoselective reduction of the ketone function was accomplished using 2 mol% of the chiral complex  $\{\text{Ru}[(R)\text{-SYNPHOS}]\text{Br}_2\}$  prepared *in situ* from commercially available  $[\text{Ru}(\text{cod})(2\text{-methylallyl})_2]$ <sup>12</sup> and afforded  $\beta$ -hydroxy ester **7** in high yield and with excellent diastereoselectivity (99% de, as determined by HPLC analysis). Subsequent diastereoselective methylation of **7** using the Fráter–Seebach conditions<sup>13</sup> (>95% de, only one diastereomer detected by  $^1\text{H}$  NMR) followed by protection of the



**Scheme 2** Synthesis of the common precursor (**4**). *Reagents and conditions:* (a)  $\text{BnOC(=NH)CCl}_3$ ,  $\text{TfOH}$ ,  $\text{CH}_2\text{Cl}_2$ , r.t., 3.5 h, 85%; (b) LDA,  $t\text{-BuOAc}$ , THF,  $-40^\circ\text{C}$ , 3 h, 80%; (c)  $\text{H}_2$  (75 bar), 2 mol%  $\{\text{Ru}[(R)\text{-SYNPHOS}] \text{Br}_2\}$ ,  $t\text{-BuOH}-\text{MeOH}$  (4:1), 50 °C, 24 h, 96%, 99% de; (d) LDA, THF,  $-40^\circ\text{C}$  to 0 °C, 1 h; then HMPA, MeI, THF,  $-10^\circ\text{C}$  to r.t., 3 h, 73%, 95% de; (e) TIPSOTf, 2,6-lutidine,  $\text{CH}_2\text{Cl}_2$ ,  $-30$  to  $-15^\circ\text{C}$ , 4 h, quant.

hydroxy function then furnished the common precursor **4** which was thereby obtained in 48% overall yield from **5**.

With compound **4** in hand, we could address the preparation of the C1–C7 (**1**) and C9–C14 (**2**) fragments of discodermolide. Thus, compound **2** was obtained in three steps from **4** after deprotection of the primary hydroxy function, followed by oxidation to the aldehyde and (*Z*)-alkenyl iodide introduction under Zhao conditions<sup>14</sup> (Scheme 3).

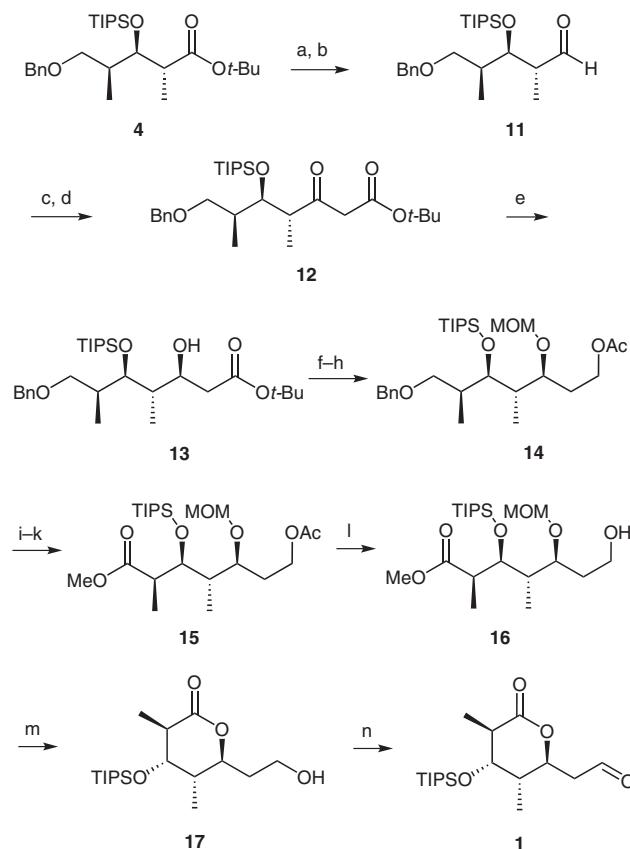


**Scheme 3** Synthesis of the C9–C14 (**2**) subunit of discodermolide. *Reagents and conditions:* (a)  $\text{H}_2$  (1 atm),  $\text{Pd/C}$  (5%),  $\text{THF}$ , r.t., 10 min, 79%; (b) Dess–Martin periodinane,  $\text{CH}_2\text{Cl}_2$ , 0 °C to r.t., 4 h, 83%; (c)  $\text{Ph}_3\text{PEtI}$ ,  $n\text{-BuLi}$ ,  $\text{THF}$ ,  $-78^\circ\text{C}$  to  $-20^\circ\text{C}$ , 20 min; then  $\text{NaHMDS}$ ,  $\text{THF}$ , 2 min; then **9**,  $\text{THF}$ ,  $-78^\circ\text{C}$  to r.t., 4.5 h, 12%; (d)  $\text{CBr}_4$ ,  $\text{PPh}_3$ ,  $\text{Et}_3\text{N}$ ,  $\text{CH}_2\text{Cl}_2$ , r.t., 30 min, 84%; (e)  $\text{MeLi}$ ,  $\text{CuI}$ ,  $\text{Et}_2\text{O}$ , 0 °C then r.t., 30 min, then **10**,  $\text{Et}_2\text{O}$ ,  $-90$  to  $-70^\circ\text{C}$ , 1 h, then  $\text{I}_2$ ,  $\text{THF}$ , r.t., 45 min, 72%.

This Zhao olefination reaction has been employed by Smith,<sup>15</sup> Marshall,<sup>16</sup> and the Novartis group<sup>17</sup> on related substrates in their syntheses of discodermolide, delivering the corresponding (*Z*)-vinyl iodides in modest yields. In our hands, however, the reaction proved even more troublesome and provided **2**<sup>18</sup> in only 12% yield. To circumvent this restrictive step, aldehyde **9** was converted into dibromoolefin **10**<sup>19</sup> in order to install the critical (*Z*)-alkenyl iodide using the procedure reported by Tanino and

Miyashita.<sup>20</sup> Thus, treatment of **10** with an excess amount of  $\text{Me}_2\text{CuLi}$  (4.5 equiv) generated stereoselectively a (*Z*)-vinyl copper species which was trapped with iodine to afford **2** in 72% yield along with 14% of the *E*-isomer. It should be noted that this particular reaction required extensive investigation of the experimental parameters before achieving a satisfactory result. Indeed, variations in reaction time, temperature, or  $\text{Me}_2\text{CuLi}$  amount resulted invariably in lower yields of **2**. Not only lower conversions were usually observed but the corresponding 1,1-dimethylalkene as well as the (*Z*)-alkenyl bromide were also formed in addition to the (*E*)-alkenyl iodide. Starting from the common precursor **4**, the C9–C14 (**2**) subunit of discodermolide was thereby generated in four steps in 40% overall yield.

Preparation of the C1–C7 subunit (**1**) began with a four-step sequence from **4** to afford the  $\beta$ -keto ester **12** required for the diastereoselective hydrogenation reaction to install the C5 hydroxy-bearing stereocenter (Scheme 4).



**Scheme 4** Synthesis of the C1–C7 (**1**) subunit of discodermolide. *Reagents and conditions:* (a) DBAL-H, toluene,  $-78^\circ\text{C}$  to  $-60^\circ\text{C}$ , 2 h, 78%; (b) Dess–Martin periodinane,  $\text{CH}_2\text{Cl}_2$ , 0 °C to r.t., 3.5 h, 95%; (c) LDA,  $t\text{-BuOAc}$ ,  $\text{THF}$ ,  $-78^\circ\text{C}$ , 15 min, 99%; (d) Dess–Martin periodinane,  $\text{CH}_2\text{Cl}_2$ , 0 °C to r.t., 1.5 h, quant.; (e)  $\text{H}_2$  (80 bar), 4 mol%  $\{[\text{RuCl}(\text{R})\text{-SYNPHOS}](\mu\text{-Cl})_3\}[\text{Me}_2\text{NH}_2]$ ,  $\text{MeOH}$ , r.t., 22 h, 94%, 99% de; (f)  $\text{MOMCl}$ , DIPEA,  $\text{CH}_2\text{Cl}_2$ , 0 °C to r.t., 22 h, 82%; (g)  $\text{LiAlH}_4$ ,  $\text{Et}_2\text{O}$ , r.t., 3.5 h, 97%; (h)  $\text{Ac}_2\text{O}$ , DMAP, pyridine,  $\text{CH}_2\text{Cl}_2$ , r.t., 30 min, 99%; (i)  $\text{H}_2$  (1 atm),  $\text{Pd/C}$  (5%),  $\text{THF}$ , r.t., 25 min, quant.; (j)  $\text{NaIO}_4$ ,  $\text{RuCl}_3$ ,  $\text{MeCN-CCl}_4-\text{H}_2\text{O}$ , r.t., 2 h; (k)  $\text{MeI}$ ,  $\text{K}_2\text{CO}_3$ ,  $\text{DMF}$ , r.t., overnight, 76% (2 steps); (l)  $\text{K}_2\text{CO}_3$ ,  $\text{MeOH}$ , r.t., 30 min, 84%; (m) PPTS,  $t\text{-BuOH}$ , 80 °C, overnight, 58%; (n) Dess–Martin periodinane,  $\text{CH}_2\text{Cl}_2$ , 0 °C to r.t., 6 h, 69%.

First, conversion of **4** into aldehyde **11** was performed using a hydride reduction–Dess–Martin oxidation sequence. Addition of lithio *tert*-butyl acetate to **11** afforded the corresponding  $\beta$ -hydroxy ester as a mixture of diastereomers (7:3 in favor of the undesired Felkin *syn*-product) which was directly subjected to Dess–Martin oxidation, leading to compound **12** in quantitative yield. The following hydrogenation reaction was initially performed in methanol at 80 °C under 80 bar of hydrogen, using 2 mol% of the chiral complex {Ru[(*R*)-SYNPHOS]Br<sub>2</sub>}. However, under these conditions, the expected  $\beta$ -hydroxy ester **13** was obtained in only 52% yield along with recovered starting material as well as some transesterified hydrogenated product. Variation of the reaction conditions (temperature, hydrogen pressure, reaction time, solvent, and/or catalyst loading) did not allow any improvement. As the use of {Ru[(*R*)-SYNPHOS]Br<sub>2</sub>} in the hydrogenation reaction failed to afford reasonable yields of **13**, we tried an Ikariya-type complex<sup>21</sup> bearing (*R*)-SYNPHOS as a ligand.<sup>22</sup> The reaction was conducted in methanol at room temperature under 80 bar of hydrogen using 1 mol% of the complex [{RuCl[(*R*)-SYNPHOS]}{ $\mu$ -Cl]<sub>3</sub>][Me<sub>2</sub>NH<sub>2</sub>]. Under these conditions, the corresponding  $\beta$ -hydroxy ester was obtained in 68% yield with no transesterification product observed. Finally, using 4 mol% of the ruthenium complex under the otherwise aforementioned conditions, the hydrogenation reaction proceeded in a satisfactory 94% yield and with a high level of diastereoselectivity (99% de, as determined by HPLC analysis). Having compound **13** in hand, a few functional transformation steps remained in order to obtain aldehyde **1**. Thus, after MOM protection, the ester function was reduced to the corresponding alcohol which was acylated to furnish compound **14**. The benzyl ether was then converted into ester **15** by deprotection of the primary alcohol, RuCl<sub>3</sub>/NaIO<sub>4</sub> oxidation and methylation of the resulting carboxylic acid. After hydrolysis of the acetate function, access to compound **1** required a lactonization step. To this end, **16** was treated with a catalytic amount of PPTS in refluxing *tert*-butanol, which resulted in cleavage of the MOM ether and subsequent lactonization at the C5 position. Finally, Dess–Martin oxidation of **17** completed the synthesis of the C1–C7 fragment (**1**)<sup>23</sup> of discodermolide which was thereby prepared in 14 steps from **4** in 14% overall yield.

In conclusion, discodermolide subunits **1** and **2** have been efficiently prepared from a common precursor **4**, possessing a *syn,anti* methyl–hydroxy–methyl stereotriad. The C3, C5, and C11 hydroxy-bearing stereocenters have been set through ruthenium-promoted asymmetric hydrogenation reactions using SYNPHOS as a ligand. This route provides good yields and high levels of diastereoselectivity. Efforts are currently under way to complete the C15–C24 subunit of discodermolide as well.

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### References and Notes

- (a) Gunasekera, S. P.; Gunasekera, M.; Longley, R. E.; Schulte, G. K. *J. Org. Chem.* **1990**, *55*, 4912; correction: *J. Org. Chem.* **1991**, *56*, 1346. (b) Gunasekera, S. P.; Paul, G. K.; Longley, R. E.; Isbrucker, R. A.; Pomponi, S. A. *J. Nat. Prod.* **2002**, *65*, 1643.
- (a) Longley, R. E.; Caddigan, D.; Harmony, D.; Gunasekera, M.; Gunasekera, S. P. *Transplantation* **1991**, *52*, 650. (b) Longley, R. E.; Caddigan, D.; Harmony, D.; Gunasekera, M.; Gunasekera, S. P. *Transplantation* **1991**, *52*, 656.
- (a) Hung, D. T.; Chen, J.; Schreiber, S. L. *Chem. Biol.* **1996**, *3*, 287. (b) ter Haar, E.; Kowalski, R. J.; Hamel, E.; Lin, C. M.; Longley, R. E.; Gunasekera, S. P.; Rosenkranz, H. S.; Day, B. W. *Biochemistry* **1996**, *35*, 243. (c) Klein, L. E.; Freeze, B. S.; Smith, A. B. III.; Horwitz, S. B. *Cell Cycle* **2005**, *4*, 501.
- (a) Huang, G. S.; Lopez-Barcons, L.; Freeze, B. S.; Smith, A. B. III.; Goldberg, G. L.; Horwitz, S. B.; McDaid, H. M. *Clin. Cancer Res.* **2006**, *12*, 298. (b) Honore, S.; Kamath, K.; Braguer, D.; Horwitz, S. B.; Wilson, L.; Briand, C.; Jordan, M. A. *Cancer Res.* **2004**, *64*, 4957. (c) Martello, L. A.; McDaid, H. M.; Regl, D. L.; Yang, C.-P. H.; Meng, D.; Pettus, T. R. R.; Kaufman, M. D.; Arimoto, H.; Danishefsky, S. J.; Smith, A. B. III.; Horwitz, S. B. *Clin. Cancer Res.* **2000**, *6*, 1978. (d) Kowalsky, R. J.; Giannakakou, P.; Gunasekera, S. P.; Longley, R. E.; Day, B. W.; Hamel, E. *Mol. Pharmacol.* **1997**, *52*, 613.
- Smith, A. B. III.; Freeze, B. S.; LaMarche, M. J.; Sager, J.; Kinzler, K. W.; Vogelstein, B. *Bioorg. Med. Chem. Lett.* **2005**, *15*, 3623.
- (a) For a review of the total syntheses prior to 2003: Paterson, I.; Florence, G. J. *Eur. J. Org. Chem.* **2003**, 2193. (b) For a recent review on syntheses, construction and biological evaluation of analogues: Smith, A. B. III.; Freeze, B. S. *Tetrahedron* **2008**, *64*, 261. (c) For a recent review on total syntheses of discodermolide and dictyostatin: Florence, G. J.; Gardner, N. M.; Paterson, I. *Nat. Prod. Rep.* **2008**, *25*, 342. (d) For references to synthetic approaches to discodermolide, see ref. 6b.
- (a) Labeeuw, O.; Blanc, D.; Phansavath, P.; Ratovelomanana-Vidal, V.; Genet, J.-P. *Eur. J. Org. Chem.* **2004**, 2352. (b) Le Roux, R.; Desroy, N.; Phansavath, P.; Genet, J.-P. *Synlett* **2005**, 429. (c) Roche, C.; Desroy, N.; Haddad, M.; Phansavath, P.; Genet, J.-P. *Org. Lett.* **2008**, *10*, 3911.
- Synthetic approaches to discodermolide involving C1–C7 lactone subunit: (a) Miyazawa, M.; Oonuma, S.; Maruyama, K.; Miyashita, M. *Chem. Lett.* **1997**, *26*, 1191. (b) Misske, A. M.; Hoffman, H. M. R. *Tetrahedron* **1999**, *55*, 4315. (c) Yadav, J. S.; Abraham, S.; Reddy, M. M.; Sabitha, G.; Sankar, A. R.; Kunwar, A. C. *Tetrahedron Lett.* **2001**, *42*, 4713. (d) Day, B. W.; Kangani, C. O.; Avor, K. S. *Tetrahedron: Asymmetry* **2002**, *13*, 1161.
- For reviews on Ru-catalyzed asymmetric hydrogenation, see: (a) Ohkuma, T.; Kitamura, M.; Noyori, R. In *Catalytic Asymmetric Synthesis*, 2nd ed.; Ojima, I., Ed.; Wiley-VCH: New York, **2000**, *1*. (b) Noyori, R. *Angew. Chem. Int. Ed.* **2002**, *41*, 2008. (c) Kitamura, M.; Noyori, R. In *Ruthenium in Organic Synthesis*; Murahashi, S.-i., Ed.; Wiley-VCH: Weinheim, **2004**, *3*. (d) Genet, J.-P. *Acc. Chem. Res.* **2003**, *36*, 908.
- (a) Duprat de Paule, S.; Champion, N.; Ratovelomanana-Vidal, V.; Genet, J.-P.; Dellis, P. (b) Duprat de Paule, S.; Jeulin, S.; Ratovelomanana-Vidal, V.; Genet, J.-P.; Champion, N.; Dellis, P. *Tetrahedron Lett.* **2003**, *44*, 823. (c) Duprat de Paule, S.; Jeulin, S.; Ratovelomanana-Vidal, V.; Genet, J.-P.; Champion, N.; Dellis, P. *Eur. J. Org. Chem.*

- 2003**, 1931. (d) Duprat de Paule, S.; Jeulin, S.; Ratovelomanana-Vidal, V.; Genet, J.-P.; Champion, N.; Deschaux, G.; Dellis, P. *Org. Process Res. Dev.* **2003**, *7*, 399.
- (11) (a) Rathke, M. W.; Lindert, A. *J. Am. Chem. Soc.* **1971**, *93*, 2318. (b) Wolberg, M.; Ji, A.; Hummel, W.; Müller, M. *Synthesis* **2001**, 937.
- (12) Genet, J.-P.; Pinel, C.; Ratovelomanana-Vidal, V.; Mallart, S.; Cano de Andrade, M. C.; Laffitte, J. A. *Tetrahedron: Asymmetry* **1994**, *5*, 665.
- (13) (a) Fráter, G. *Helv. Chim. Acta* **1979**, *62*, 2825. (b) Seebach, D.; Wasmuth, D. *Helv. Chim. Acta* **1980**, *63*, 197.
- (14) Chen, J.; Wang, T.; Zhao, K. *Tetrahedron Lett.* **1994**, *35*, 2827.
- (15) (a) Smith, A. B. III.; Beauchamp, T. J.; LaMarche, M. J.; Kaufman, M. D.; Qiu, Y. P.; Arimoto, H.; Jones, D. R.; Kobayashi, K. *J. Am. Chem. Soc.* **2000**, *122*, 8654. (b) For an investigation of the mechanism of this transformation, see: Harimoto, H.; Kaufman, M. D.; Kobayashi, K.; Qiu, Y.; Smith, A. B. III. *Synlett* **1998**, 765.
- (16) Marshall, J. A.; Johns, B. A. *J. Org. Chem.* **1998**, *63*, 7885.
- (17) Mickel, S. J.; Sedelmeier, G. H.; Niederer, D.; Schuerch, F.; Grimler, D.; Koch, G.; Daeffler, R.; Osmani, A.; Hirni, A.; Schaer, K.; Gamboni, R.; Bach, A.; Chaudhary, A.; Chen, S.; Chen, W.; Hu, B.; Jagoe, C. T.; Kim, H.-Y.; Kinder, F. R.; Liu, Y.; Lu, Y.; McKenna, J.; Prashad, M.; Ramsey, T. M.; Repic, O.; Rogers, L.; Shieh, W.-C.; Wang, R.-M.; Waykole, L. *Org. Process Res. Dev.* **2004**, *8*, 101.
- (18) **Spectroscopic Data for Compound 2**  
 $[\alpha]_D^{25} +3.6$  (*c* 1.0,  $\text{CHCl}_3$ ).  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta = 5.22$  (dq,  $J = 9.1, 1.1$  Hz, 1 H), 4.32 (dd,  $J = 7.0, 4.1$  Hz, 1 H), 2.64 (qd,  $J = 7.1, 4.1$  Hz, 1 H), 2.43–2.50 (m, 1 H), 2.42 (d,  $J = 1.1$  Hz, 3 H), 1.45 (s, 9 H), 1.19 (d,  $J = 7.1$  Hz, 3 H), 1.10 (br s, 18 H), 0.98–1.15 (m, 3 H), 1.01 (d,  $J = 7.0$  Hz, 3 H).  $^1\text{H}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta = 173.0, 137.6, 100.3, 80.2, 75.7, 47.1, 45.3, 33.7, 28.1, 18.3, 16.1, 13.1, 10.9$ . IR (film): 1732, 1640, 1257, 883  $\text{cm}^{-1}$ . MS (DCI/ $\text{NH}_3$ ):  $m/z = 551$  [M + H] $^+$ , 472 [M +  $\text{NH}_4t\text{-Bu}$ ] $^+$ .
- (19) **Spectroscopic Data for Compound 10**  
 $[\alpha]_D^{25} +5.5$  (*c* 1.05,  $\text{CHCl}_3$ ).  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta = 6.23$  (d,  $J = 9.8$  Hz, 1 H), 4.28 (dd,  $J = 6.8, 4.1$  Hz, 1 H), 2.52–2.67 (m, 2 H), 1.47 (s, 9 H), 1.13 (d,  $J = 7.2$  Hz, 3 H), 1.10 (s, 18 H), 1.04–1.13 (s, 3 H), 1.07 (d,  $J = 7.0$  Hz, 3 H).  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta = 172.7, 140.8, 88.8, 80.7, 75.1, 46.9, 42.0, 28.2, 18.2, 15.5, 13.1, 9.4$ . IR (film): 1733, 1460, 1255, 882  $\text{cm}^{-1}$ . MS (DCI/ $\text{NH}_3$ ):  $m/z = 529$  [M + H] $^+$ . Anal. Calcd for  $\text{C}_{21}\text{H}_{40}\text{Br}_2\text{O}_3\text{Si}$ : C, 47.73; H, 7.63. Found: C, 47.81; H, 7.51.
- (20) Tanino, K.; Arakawa, K.; Satoh, M.; Iwata, Y.; Miyashita, M. *Tetrahedron Lett.* **2006**, *47*, 861.
- (21) (a) Ikariya, T.; Ishii, Y.; Kawano, H.; Arai, T.; Saburi, M.; Yoshikawa, S.; Akutagawa, S. *J. Chem. Soc., Chem. Commun.* **1985**, 922. (b) Ohta, T.; Tonomura, Y.; Nozaki, K.; Takaya, H.; Mashima, K. *Organometallics* **1996**, *15*, 1521. (c) Mashima, K.; Nakamura, T.; Matsuo, Y.; Tani, K. *J. Organomet. Chem.* **2000**, *607*, 51.
- (22) Jeulin, S.; Champion, N.; Dellis, P.; Ratovelomanana-Vidal, V.; Genet, J.-P. *Synthesis* **2005**, 3666.
- (23) **Spectroscopic Data for Compound 1**  
 $[\alpha]_D^{25} -2.5$  (*c* 0.8,  $\text{CHCl}_3$ ).  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta = 9.85$  (dd,  $J = 2.5, 1.3$  Hz, 1 H), 4.91 (ddd,  $J = 10.5, 7.4, 4.0$  Hz, 1 H), 3.87 (t,  $J = 2.3$  Hz, 1 H), 2.80 (td,  $J = 7.5, 2.3$  Hz, 1 H), 2.78 (td,  $J = 16.7, 4.0, 1.3$  Hz, 1 H), 2.68 (ddd,  $J = 16.7, 7.4, 2.5$  Hz, 1 H), 2.04–2.15 (m, 1 H), 1.28 (d,  $J = 7.5$  Hz, 3 H), 0.98–1.10 (m, 24 H).  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta = 199.5, 173.1, 76.3, 75.0, 46.5, 44.4, 33.6, 18.1, 16.7, 13.9, 12.7$ . IR (film): 2731, 1735, 1223, 883  $\text{cm}^{-1}$ . MS (DCI/ $\text{NH}_3$ ):  $m/z = 360$  [M +  $\text{NH}_4$ ] $^+$ , 343 [M + H] $^+$ .

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