

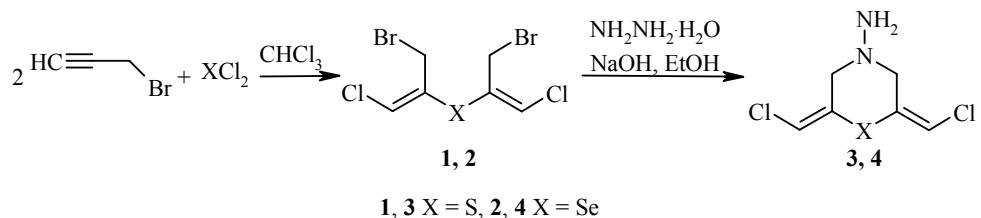
BIS-*E*-CHLOROMETHYLIDENE DERIVATIVES OF 4-THIO- AND 4-SELENOMORPHOLINAMINES

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It is known that the reaction of bis(2-haloethyl) sulfides with substituted amines gives 4-thiomorpholine and its *N*-substituted derivatives [1], while the reaction with hydrazine hydrate yields 4-thiomorpholinamine [1, 2]. 4-Selenomorpholinamine is unknown, but 4-selenomorpholine and a series of its *N*-substituted derivatives have been reported [3, 4].

We have developed a general method for the preparation of the previously unknown bis-*E*-chloromethylidene derivatives of 4-thiomorpholinamine **3** and 4-selenomorpholinamine **4**. The method is composed of two stages, the first of which is a stereo- and regioselective reaction of sulfur or selenium dichloride with propargyl bromide, occurring as an anti-Markovnikov addition to give the *E,E*-bis(3-bromo-1-chloro-1-propen-2-yl) sulfide (**1**) or selenide (**2**) in high yields [5, 6]. The second stage is a chemoselective nucleophilic substitution of the bromine atoms in the bromomethylene fragments of sulfide **1** and selenide **2** with hydrazine hydrate in the presence of NaOH to give the heterocycles **3** and **4** in high yields. This second stage includes the alkylation of one hydrazine amino group with two bromomethyl groups of the divinylchalcogenides **1** and **2**, giving the unsymmetrical 1,1-disubstituted hydrazine derivatives, which is a typical feature of the reaction between hydrazine and alkyl halides [7].



The presence in compounds **1-4** of a =CHCl fragment is indicated by the value of the direct coupling constant ($^1J_{\text{CH}} = 199\text{-}203 \text{ Hz}$) in the ^{13}C NMR spectra, pointing to a direct carbon–halogen bond [8]. The structure of selenide **2** [5] and the 4-selenomorpholinamine **4** was also confirmed by the presence of a direct spin-spin coupling between the selenium atom and the C= carbon atom ($^1J_{\text{C-Se}} = 101\text{-}113 \text{ Hz}$) in the ^{13}C NMR spectra. The *EE* configuration of the divinylchalcogenides **1** and **2**, as well as the heterocycles **3** and **4**

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follows from the fine structure in the ^{13}C NMR spectra characterized by a *trans*-vicinal spin-spin coupling between the vinyl proton and the carbon atom of the CH_2 group ($^3J_{\text{CH}} = 6.2\text{-}7.0$ Hz) [9]. The ^{15}N NMR spectra of heterocycles **3** and **4** clearly reveal the signals of the two different nitrogen atoms, and in the case of the heterocycle **3** it is possible to observe the fine structure in the NH_2 ^{15}N NMR signal due to a spin-spin interaction of the nitrogen atom and proton. The mass spectra of heterocycles **3** and **4** are characterized by intensive molecular ions and the ion fragments $[\text{M}-\text{NH}_2]^+$ and $[\text{M}-\text{Cl}]^+$.

Hence the two-stage method we have discovered offers the possibility of preparing novel and promising chloromethylidene derivatives of 4-thiomorpholinamine and 4-selenomorpholinamine from the available propargyl bromide, sulfur and selenium dichlorides, and hydrazine hydrate. These are intermediate products for further functionalization, due to the presence of chlorine atoms on a double bond and free amino groups in their structure.

^1H , ^{13}C , ^{77}Se , and ^{15}N NMR spectra were recorded on a Bruker DPX-400 instrument (400, 100, 76, and 40 MHz, respectively) using CDCl_3 as solvent. Chemical shifts were referred to TMS (for ^1H and ^{13}C nuclei), Me_2Se (for the ^{77}Se nuclei at 0.0 ppm), or MeNO_2 (for the ^{15}N nuclei at 0.0 ppm). Mass spectra were recorded on a Shimadzu QP5050A instrument (EI, 70 eV) with an SPB-5ms column (60 m). Elemental analysis was carried out on a Thermo Finnigan EA 1112 analyzer.

Propargyl bromide was obtained by the bromination of propargyl alcohol using PBr_3 [10].

E,E-Bis(3-bromo-1-chloro-1-propen-2-yl) sulfide (1) and **E,E-bis(3-bromo-1-chloro-1-propen-2-yl) selenide (2)** were prepared using methods [5, 6] in 80 and 90% yields, respectively.

(2E,6E)-2,6-Bis(chloromethylidene)thiomorpholin-4-amine (3). Sulfide **1** (0.682 g, 2 mmol) in EtOH (9 ml) was slowly added dropwise to a solution of NaOH (0.200 g, 5 mmol) in hydrazine hydrate (3.100 g, 62 mmol) at 0°C under an argon atmosphere. The mixture was stirred for 8 h at 20°C, diluted with water (9 ml), and extracted with CHCl_3 . The extract was dried over MgSO_4 . Solvent was removed under reduced pressure. The product obtained was purified by column chromatography on silica gel using $\text{EtOAc}-\text{CHCl}_3$ (5:95) as eluent. Yield 0.314 g (74%), a yellow oil. ^1H NMR spectrum, δ , ppm (J , Hz): 3.40 (2H, s, NH_2); 3.87 (4H, d, $^4J = 0.9$, 2 CH_2); 6.27 (2H, t, $^4J = 0.9$, 2 =CH). ^{13}C NMR spectrum, δ , ppm (J , Hz): 57.5 (tdt, $^1J = 140.8$, $^3J = 6.3$, $^3J = 4.4$, 2 CH_2); 115.5 (dt, $^1J = 199$, $^3J = 4.8$, 2C-Cl); 128.9 (dt, $^2J = 4.0$, $^2J = 4.0$, C-S-C). ^{15}N NMR spectrum, δ , ppm (J , Hz): -315.2 (t, $^1J = 3.9$, NH_2); -295.0 (s, N). Mass spectrum, m/z (I_{rel} , %): (^{35}Cl , ^{15}N , ^{32}S): 212 [M]⁺ (36), 195 [$\text{M}-\text{NH}_2$]⁺ (3), 177 [$\text{M}-\text{Cl}$]⁺ (38), 175 (100), 158 (8), 145 (18), 131 (16), 114 (3), 92 (34), 71 (43), 69 (34), 45 (57), 39 (71). Found, %: C 34.50; H 3.91; Cl 33.86; N 13.20; S 15.54. $\text{C}_6\text{H}_8\text{Cl}_2\text{N}_2\text{S}$. Calculated, %: C 34.13; H 3.82; Cl 33.59; N 13.27; S 15.19.

(2E,6E)-2,6-Bis(E-chloromethylidene)selenomorpholin-4-amine (4). Compound **4** was prepared similarly to compound **3** from NaOH (0.150 g, 3.8 mmol), hydrazine hydrate (0.500 g, 10.0 mmol), and the selenide **2** (0.970 g, 2.5 mmol) in EtOH (10 ml) by stirring over 12 h and subsequent isolation by column chromatography on silica gel (eluent $\text{EtOAc}-\text{CHCl}_3$, 5:95). Yield 0.433 g (67%), a yellow oil. ^1H NMR spectrum, δ , ppm (J , Hz): 3.15 (2H, s, NH_2); 3.98 (4H, d, $^4J = 1.3$, 2 CH_2); 6.38 (2H, t, $^4J = 1.3$, 2 =CH). ^{13}C NMR spectrum, δ , ppm (J , Hz): 58.2 (tdt, $^1J = 140.8$, $^3J = 6.2$, $^3J = 4.8$, 2 CH_2); 117.2 (dt, $^1J = 200.1$, $^3J = 4.6$, 2C-Cl); 124.4 (d, $^3J = 3.6$, $^1J_{\text{Se}-\text{C}} = 101.0$, C-Se-C). ^{77}Se NMR spectrum, δ , ppm: 404.4. ^{15}N NMR spectrum, δ , ppm (J , Hz): -314.2 (NH_2); -298.3 (N). Mass spectrum, m/z (I_{rel} , %): (^{35}Cl , ^{15}N , ^{80}Se): 260 [M]⁺ (6), 243 [$\text{M}-\text{NH}_2$]⁺ (1), 225 [$\text{M}-\text{Cl}$]⁺ (9), 197 (1), 193 (4), 159 (10), 140 (8), 119 (15), 39 (100). Found, %: C 28.34; H 3.58; Cl 27.24; N 11.19; Se 30.11. $\text{C}_6\text{H}_8\text{Cl}_2\text{N}_2\text{Se}$. Calculated, %: C 27.93; H 3.13; Cl 27.48; N 10.86; Se 30.60.

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