## 3-HYDROXYLATION OF GIBBERELIN A<sub>12</sub>-ALDEHYDE IN *GIBBERELLA FUJIKUROI* STRAIN REC-193A\*

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Abstract—When grown on PDL medium for 11 days the strain REC-193A of *Gibberella fujikuroi* produces the usual range of gibberellins and *ent*-kaurenoid metabolites. After 3–5 days under the same conditions of culture, this slow growing strain produces virtually none of these metabolites. These short term cultures were found to convert gibberellin  $A_{12}$ -aldehyde into gibberellins  $A_{12}$  (8·3%),  $A_{14}$  (45%),  $A_4$  (*ca*. 17%) and  $A_7$  (*ca*. 6%). Under identical conditions of culture gibberellin  $A_{12}$  was largely unmetabolised. These results show that 3-hydroxylation is the first step in the conversion of gibberellin  $A_{12}$ -aldehyde into gibberellins  $A_{14}$ ,  $A_4$  and  $A_7$ .

Most of the published work on the biosynthesis of the gibberellins (GAs) in Gibberella fujikuroi has been conducted with high yielding strains such as ACC 917. This paper concerns the 3-hydroxylation of  $GA_{12}$ -aldehyde (Ia) in the slow growing strain REC-193A which is wild-type for GA production and a mutant for colonial growth and amylase activity.<sup>1</sup> This strain was obtained<sup>1</sup> as a random ascospore from the perithecium of a strain of G. fujikuroi originally isolated from rice plants in Japan. Subsequent papers will deal with mutants that block specific steps in the GA biosynthetic pathway.

 $GA_{12}$ -aldehyde (Ia) is probably the initial *ent*-gibberellane from the *in vivo* contraction of *ent*-7*a*-hydroxykaur-16-en-19-oic acid (II). It occurs in cultures<sup>2</sup> of *G. fujikuroi* ACC 917 which convert it efficiently and specifically<sup>3</sup> into  $GA_3$  (III). This aldehyde (Ia) is also formed<sup>4</sup> from mevalonic acid lactone in a cell-free preparation from the seed of *Cucurbita pepo*. In cultures of *G. fujikuroi* ACC 917 Cross *et al.*<sup>3</sup> found that the incorporation of [17-<sup>14</sup>C]-GA<sub>12</sub>-aldehyde into [17-<sup>14</sup>C]-GA<sub>3</sub> was much higher (15·4%) than that (0·7%) of the corresponding acid,  $GA_{12}$  (V). From this observation and the non-incorporation of  $GA_9$  (VIII) into  $GA_3$  (III) Cross *et al.*<sup>3</sup> concluded that 3-hydroxylation in the biosynthesis of  $GA_3$  (III) occurred before oxidation to  $GA_{12}$  (V) and before the formation of the 19,10lactone. Direct evidence is now presented in support of this view.

When strain REC-193A was grown in shake-flask culture on a potato-dextrose liquid (PDL) medium for 11 days, the usual range of GAs and related *ent*-kaurenoids was produced

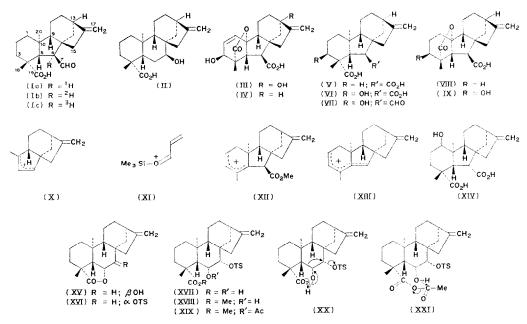
\* Part VI in the series "Fungal Products". For Part V see (1973) J. Chem. Soc. Perkin I in press.

<sup>&</sup>lt;sup>1</sup> PHINNEY, B. O. and FUKUYAMA, M. (1973) in preparation.

<sup>&</sup>lt;sup>2</sup> HANSON, J. R., HAWKER, J. and WHITE, A. F. (1972) J. Chem. Soc. Perkin I 1892.

<sup>&</sup>lt;sup>3</sup> CROSS, B. E., NORTON, K. and STEWART, J. C. (1968) J. Chem. Soc. C, 1054.

<sup>&</sup>lt;sup>4</sup> GRAEBE, J. E., BOWEN, D. H. and MACMILLAN, J. (1972) Planta 102, 261.



These metabolites were conveniently identified by combined GC-MS of the derivatized total extract from the culture filtrates; those identified by GC-MS of the methylated extract and of the methylated-trimethylsilylated extract are listed in Table 1. When REC-193A was grown on PDL medium for 5 days or less, little or no GAs were produced as shown by the GC trace of the methylated extract (Fig. 1a). When grown on PDL containing GA<sub>12</sub>-aldehyde (Ia) this substrate was rapidly metabolised (Fig. 1b) mainly to GA<sub>14</sub> (VI) and slightly to GA<sub>12</sub> (V); these products from GA<sub>12</sub>-aldehyde were identified by GC-MS of the methylated extract from 5-day cultures. In contrast GA<sub>12</sub> (V) was largely unmetabolized (Fig. 1c) by REC-193A, grown under identical conditions of culture. These cultures were able to metabolize at least 5.5 mg GA<sub>12</sub>-aldehyde (Ia) per 100 ml culture medium in 5 days. A time course study of this metabolism from 1 to 5 days showed that GA<sub>12</sub> (V) and GA<sub>14</sub> (VI) were initially formed in equal amount. However, after 3 days GA<sub>12</sub> (V) had reached a maximum level and thereafter only the GA<sub>14</sub> concentration increased until the GA<sub>12</sub>-aldehyde (Ia) had been completely metabolized.

 Table 1. Metabolites of Gibberella fujikuroi REC-193A identified by

 GC-MS of the methylated and methylated trimethylsilylated extract from 11-day cultures

GA <sub>1</sub> GA <sub>4</sub> GA <sub>7</sub> GA <sub>9</sub> GA <sub>12</sub>	GA <sub>13</sub> GA <sub>14</sub> GA <sub>25</sub>	Fusaric acid 4-Hydroxyphenyl acetic acid ent-6,7-secokaur-16-ene-6,7,19-trioic acid ent-7-oxo-6,7-secokaur-16-ene-6,19-dioic acid
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The specific conversion of  $GA_{12}$ -aldehyde into  $GA_{14}$  (VI) was confirmed by using [6-<sup>2</sup>H]-GA<sub>12</sub>-aldehyde (Ib); GC-MS of the methylated extract from a 5-day culture (see Fig. 1b) established that  $GA_{12}$  (V) and  $GA_{14}$  (VI) were formed without dilution of the [<sup>2</sup>H]-label. The percentage conversion of  $GA_{12}$ -aldehyde was determined from a feed of

[6-<sup>3</sup>H]-GA<sub>12</sub>-aldehyde (Ic). After 3 days' culture, unlabelled GA<sub>12</sub> (V) and unlabelled GA<sub>14</sub> (VI) were added to the total extract which was then fractionated by partition chromatography on Sephadex G25 with the solvent system  $C_6H_6$ -AcOH-H<sub>2</sub>O(8:5:3) (see Ref. 5).

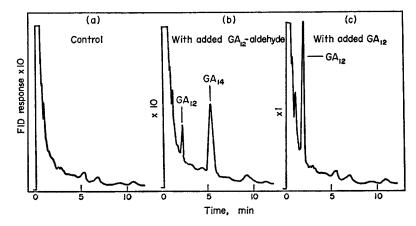


Fig. 1. GC of methylated  $10^{-3}$ -aliquots of unfractionated extracts from 5-day cultures of Giberella fujikuroi REC-193A.

The fractions were monitored by scintillation radio-counting. The  $[6^{-3}H]$ -GA<sub>12</sub>, recovered from fractions 19-25, was crystallized to constant activity which showed 8.3% incorporation of the [<sup>3</sup>H]-label from the substrate (Ic). The [6-<sup>3</sup>H]-GA<sub>14</sub>, recovered from fractions 35-42 contained 45% of the total label from the substrate (Ic). Fractions 48-55 which contained 17.3% of the label were methylated and shown by GC-MS to contain mainly GA<sub>4</sub> (IX) and two unidentified minor products. Similar GC-MS analysis of fractions 60-70 which contained 6.6% of the label revealed the presence of only GA<sub>7</sub> (IV). Fractions 31-39 contained 16% of the label; GC-MS of a methylated aliquot showed the presence of  $GA_{14}$ (VI) and a major component with the same MW as GA14; several minor but unidentified components were also detected. Combined gas chromatography-radio counting (GC-RC) of this fraction, essentially by the method of Belham and Neal,<sup>6</sup> showed that the major component and at least two of the incompletely resolved minor components were radioactive. GC-MS of the methylated-trimethylsilylated fractions 31-39 showed that the major component formed a mono-TMSi ether. The MS of this derivative was similar to that of MeGA<sub>14</sub> TMSi and indicated that this major component was isomeric with GA<sub>14</sub>. The presence of the hydroxy group in ring A was indicated by the base peak at m/e 231 which corresponds to the ion (X) obtained from cleavage of the 1, 10- and 4,5-bonds; the related MeGA<sub>14</sub> and MeGA<sub>18</sub> TMSi ethers show<sup>7</sup> significant ions at m/e 231 and 319 respectively which correspond to analogous fragmentations. An intense ion at m/e 129 (XI) is present which is characteristic<sup>7</sup> of 3-TMSi ethers of GAs, but which could also be formed from 1-TMSi ethers by a similar fragmentation. The absence of a 3-hydroxyl group is suggested by the absence of significant peaks at m/e 298 and 239 which are present in the MS of MeGA<sub>14</sub> and MeGA14 TMSi ether and which can be ascribed to the conjugated ions (XII) and

<sup>&</sup>lt;sup>5</sup> PITEL, D. W., VINING, L. C. and ARSENAULT, G. P. (1971) Can. J. Biochem. 49, 183.

<sup>&</sup>lt;sup>6</sup> BELHAM, J. E. and NEAL, G. E. (1972) Anal. Biochem. 45, 6.

<sup>&</sup>lt;sup>7</sup> BINKS, R., MACMILLAN, J. and PRYCE, R. J. (1969) Phytochemistry 8, 271.

(XIII). Furthermore, the formation of a  $3\alpha$ -hydroxy GA is biogenetically improbable while 1-hydroxylation is known, for example, in the fungal GA<sub>16</sub>. It is therefore suggested that the major component in fractions 31-39 is a 1-hydroxy GA (XIV).

The results from the feed of  $[6-{}^{2}H]$ -GA<sub>12</sub>-aldehyde show that this aldehyde is converted into GA<sub>12</sub> (V), GA<sub>14</sub> (VI), GA<sub>4</sub> (IX) and GA<sub>7</sub> (IV) in short term cultures of *G. fujikuroi* REC-193A. The non-metabolism of GA<sub>12</sub> (V) under identical culture conditions could be due to the inability of the diacid to penetrate. However, this is unlikely since GA<sub>12</sub> is rapidly metabolised by another mutant of *G. fujikuroi* (unpublished results). It is concluded therefore that GA<sub>12</sub> is probably not on the direct pathway between GA<sub>12</sub>-aldehyde (Ia) and GA<sub>3</sub> (III). Thus 3-hydroxylation occurs before oxidation of GA<sub>12</sub>-aldehyde (Ia) to GA<sub>12</sub> (V) and before formation of the 19,10-lactone bridge. The inferred intermediacy of GA<sub>14</sub>aldehyde (VII) will be substantiated in a forthcoming publication. Our results also provide the first demonstration that the enzymes required for GA synthesis are present in the trophophase<sup>8</sup> of *G. fujikuroi* cultures; all previous biosynthetic studies relate to the idiophase.

Gibberellin A<sub>12</sub>-aldehyde (Ia), required for the above studies, was prepared from ent- $6\beta$ ,7a-dihydroxykaur-16-en-19-oic acid 19,6-lactone (XV), essentially as described by Cross et al.<sup>3</sup> The overall yield was significantly increased in two ways. First the efficiency of the ring contraction of the intermediate tosylate (XVI) to GA12-aldehyde by t-BuOK in t-BuOH was increased by 5–10% by the addition of 10% H<sub>2</sub>O.<sup>9</sup> Secondly the by-product (XVII) from this ring contraction was converted in fair yield into GA12-aldehyde (Ia) by refluxing collidine or by sodium hydride in tetrahydrofuran at room temperature. The mechanism of the ring contraction of the hydroxy-acid (XVII) is of interest since Cross et al.<sup>3</sup> observed that the corresponding Me ester (XVIII) underwent ring contraction with difficulty and in minute (4%) yield with NaOMe in refluxing MeOH in contrast to the 35% yield from the 6-acetate (XIX). We have found that the Me ester (XVIII) is also stable to NaH in tetrahydrofuran. The depicted mechanism (XX) for the ring contraction of the acid (XVII) may account for these results. Steric hindrance to the abstraction of the 6-hydroxy proton in the Me ester (XVIII) can be circumvented in the acid (XVII) by internal attack of the carboxyl anion. The sterically less demanding attack by base at the carbonyl carbon of the 6-acetate (XIX) would account for the observed rearrangement of the latter. These steric arguments are supported by the ease of acetylation of the acid (XVII), observed in the present studies and explicable by intramolecular acetylation via the mixed anhydride (XXI), compared to the difficult acetylation of the Me ester (XVIII).<sup>3,10</sup>

[6<sup>-2</sup>H]- and [6<sup>-3</sup>H]-gibberellin A<sub>12</sub>-aldehydes (Ib) and (Ic) were prepared from GA<sub>12</sub>aldehyde (Ia) in two ways: (a) with NaOMe in MeO<sup>2</sup>H or MeO<sup>3</sup>H, and (b) more conveniently in boiling tetrahydrofuran containing NaOMe and <sup>2</sup>H<sub>2</sub>O or <sup>3</sup>H<sub>2</sub>O. [6<sup>-3</sup>H]-Gibberellin A<sub>12</sub>-aldehyde (Ic) was purified by chromatography on Sephadex LH20]using the bi-phasic solvent system, PE–EtOAc–AcOH–MeOH–H<sub>2</sub>O (50:15:10:10:2), a method developed in this laboratory by C. M. Wels (unpublished results). The [6<sup>-3</sup>H]-aldehyde (Ic) had a specific activity of 0.89  $\mu$ Ci  $\mu$ mol<sup>-1</sup>. [6<sup>-2</sup>H]-Gibberellin A<sub>12</sub>-aldehyde (Ib) was shown by MS to contain *ca*. 90%-[<sup>2</sup>H<sub>1</sub>] which was located at the 6-position by NMR. The high field doublet, present in the unlabelled aldehyde (Ia) at  $\tau$  6·78 and part of the AM system of the 5,6-protons, was absent from the spectrum. The GA<sub>12</sub>-aldehydes (Ia–c) were stable as crystalline solids but were oxidized to GA<sub>12</sub> (V), slowly as gums and more rapidly in

<sup>&</sup>lt;sup>8</sup> BU'LOCK, J. D. (1967) Essays in Biosynthesis and Microbial Development, Wiley, London.

<sup>&</sup>lt;sup>9</sup> HANSON, J. R. personal communication.

<sup>&</sup>lt;sup>10</sup> HANSON, J. R. (1966) Tetrahedron 22, 2877.

solution. [6-<sup>2</sup>H]-Gibberellin  $A_{12}$ -aldehyde (Ib) was preparatively oxidized by Jones reagent to  $GA_{12}$  (V) without loss of label providing another example<sup>11</sup> that oxidation of aldehydes with chromium trioxide does not proceed via the enol.

## EXPERIMENTAL

Provenance of G. fujikuroi REC-193A. Strain REC-193A was obtained<sup>1</sup> from a wild-type, N-481, of G. fujikuroi, originally isolated from rice seedlings on Honshu Island, Japan. During sub-culturing of N-481, perithecia appeared fortuitously. REC-193A was one of the strains that came from the germination of 500 ascospores which were collected randomly from these perithecia. REC-193A is thus homocaryotic in origin.

Culture conditions. Sub-master cultures of G. fujikuroi REC-193A, grown on slopes of potato-dextrose agar, were used to innoculate potato-dextrose liquid<sup>12</sup> medium (100 ml) in 500 ml conical flasks. These cultures, grown at 25° for 3 days on a reciprocal shaker, were successively sub-cultured, to ensure homogeneity, by transferring innocula (1 ml) to fresh medium (100 ml) in 500 ml conical flasks. After 3 such transfers, the cultures were grown at 25° for 3–11 days with, or without, the following substrates (added in Me<sub>2</sub>CO to the hot autoclaved media before the third transfer): (a) GA<sub>12</sub>-aldehyde (Ia; 0.5–5.5 mg); (b) [6-<sup>2</sup>H]-GA<sub>12</sub>-aldehyde (Ic; *ca.* 100  $\mu$ g); and (d) GA<sub>12</sub> (V; 0.5–5.5 mg).

Analysis of metabolites. (a) From 11-day cultures without added substrate. A portion  $(10^{-3})$  of the unfractionated EtOAc extract from the culture filtrate (100 ml) was methylated  $(CH_2N_2)$  then trimethyl silylated with hexamethyldisilazane-trimethylsilylchloride (1:1) in pyridine. The derivatized metabolites were examined by GC-MS on a Varian MAT CH7 fitted with a Biemann-Watson molecular separator. The GC conditions were: 2% QF-1 on Gaschrome Q (100-120 mesh) in a glass column (213 cm  $\times 1.6$  mm i.d.) with an He flow-rate of 7 ml min<sup>-1</sup> and temp. programmed from 210° at 2° min<sup>-1</sup>. The compounds were identified by direct comparison of the MS with those of authentic compounds.<sup>7</sup>

(b) From 5-day cultures without added substrate. A portion  $(10^{-3})$  of the total extract was methylated and examined by GC on 2% QF-1 on Gaschrome Q (80–100 mesh) in a glass column (150 × 0.47 cm i.d.) isothermally at 200° with N<sub>2</sub> flow-rate of 75 ml min<sup>-1</sup>. The FID trace is shown in Fig. 1.

(c) From 5-day cultures with added [6-<sup>1</sup>H]- and [6-<sup>2</sup>H]- $GA_{12}$ -aldehydes. The crude extracts from each substrate were separately examined by GC as in (a) after methylation and by GC-MS after methylation and after methylation-trimethylsilylation as in (b) except that a 2% SE33 column was used. The GA<sub>12</sub> and GA<sub>14</sub> from the unlabelled substrate were identified from the published<sup>7</sup> MS of these derivatives. The MS of the derivatives of [6-<sup>2</sup>H]-GA<sub>12</sub> and -GA<sub>14</sub> showed the same [<sup>2</sup>H]-content as the [6-<sup>2</sup>H]-GA<sub>12</sub>-aldehyde.

(d) From a 3-day culture with added  $[6-^{3}H]-GA_{12}$ -aldehyde. The filtrate from a shake-flask (20 ml) culture grown in the presence of  $[6^{-3}H]$ -GA<sub>12</sub>-aldehyde (ca. 101 µg, 628 000 decomp. min<sup>-1</sup>) was extracted with EtOAc. Unlabelled GA12 (9.93 mg) and GA14 (11.54 mg) were added to the extract which was dissolved in the minimum vol. of the aqueous phase of the two phase system  $C_6H_6$ -AcOH-H<sub>2</sub>O (8:5:3) and absorbed in Sephadex G25 (fine, 200 mg). The latter was then added to a column (136  $\times$  1.5 cm) of Sephadex G25 (fine) which had been swollen with the aqueous phase and equilibrated with the organic phase of the above solvent system. The column was eluted with the organic phase (flow-rate 1.25 ml min<sup>-1</sup>) and fractions (5 ml) were collected. Aliquots were evaporated to dryness and counted with 32% efficiency in a Tracerlab Corumatic 200 Scintillation Counter in 5% Butyl-PBD in toluene-methoxycthanol (3:2). Fractions 19-25 were pooled and the purple pigment was removed from the recovered material by TLC on silica gel with EtOAc-PE-AcOH (50: 50:1). The band at  $R_{1}0:3$  was eluted with EtOAc and the recovered material was crystallized from Me<sub>2</sub>CO-PE to constant activity to give GA<sub>12</sub> m.p. 245-248°; 5250 decomp. mg<sup>-1</sup> min<sup>-1</sup> (8.3% incorporation); one peak by GC of a methylated aliquot. Fractions 40-47 were pooled and treated as above. The material recovered from TLC [silica gel; EtOAc-PE-AcOH (60:40:1)] at  $R_f$  0.25 crystallized from MeCOEt-PE to constant activity to yield GA14 m.p. 221-223°, 24 537 decomp. mg<sup>-1</sup> min<sup>-1</sup> (45% incorporation); one peak by GC of Me ester, Fractions 48-55 (108 603 decomp. min<sup>-1</sup>; 17.3% incorporation) were bulked. GC of a methylated aliquot showed a major peak, identified as GA<sub>4</sub> Me ester by GC-MS under the conditions specified in (b). Fractions 60-70 (41 500 decomp. min<sup>-1</sup>; 6.6%) were bulked. GC of a methylated aliquot showed only one peak identified as GA7 Me ester by GC-MS under the conditions specified in (b). Fractions 31-39 (99 800 decomp. min<sup>-1</sup>; 16% incorporation) were bulked and an aliquot was examined by GC-MS as described in (b) as the Me and MeTMSi derivatives. The major peak which was shown to be radioactive by the method of Belham and Neal<sup>6</sup> had the following MS. Me ester m/e (%) 376 (M<sup>+</sup>, 4), 344(100), 330(10), 329(23), 269(9), 227(6), and 164(15); Me ester TMSi ether m/e (%) 448 (M<sup>+</sup>,11), 433(10), 416(85), 401(10), 287(59), 259(26), 231(100) and 129(57).

Gibberellin  $A_{12}$ -aldehyde (Ia). (a) From ent-6 $\beta$ -hydroxy-7 $\beta$ -tosyloxykaur-16-en-19-oic acid 19,6-lactone (XVI). The tosylate (180 mg) in a mixture of KOH (1.8 g), t-BuOH (15ml) and H<sub>2</sub>O (1.5 ml) was refluxed for

<sup>11</sup> GOSWAMI, C. and BANERJI, K. K. (1970) Bull. Chem. Soc. Japan 43, 2643; ibid. (1972) 45, 2925.

<sup>12</sup> SPECTOR, C. and PHINNEY, B. O. (1908) Physiol. Plant. 21, 127.

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6 hr under N<sub>2</sub>. The reaction mixture was concentrated *in vacuo*, diluted with H<sub>2</sub>O, acidified to pH 2.5 and then extracted with EtOAc. The gum, recovered from the EtOAc, was purified by TLC on silica gel HF with EtOAc-PE-AcOH (50:50:1). Elution of the band at  $R_1$  0.5 (bright green colour with 5% H<sub>2</sub>SO<sub>4</sub> in EtOH at 120°) with EtOAc gave GA12-aldehyde (Ia), crystallized from Me2CO-PE in needles (66.5 mg), m.p. 159-162° (lit.<sup>3</sup> m.p. 159-163°) identified from the published IR; <sup>3</sup> 7 9.26 (3H, s, 20-H), 8.86 (3H, s, 18-H), 6.78 (1H, J 5.5, 13.0 Hz, 6-H), 5.26 (1H, br, 17-H), 5.14 (1H, br, 17-H), and 0.32 (1H, J 5.5 Hz, 7-H); m/e (% base peak) 316 (M<sup>+</sup>, 3), 298(15), 270(67), 241(100) and 239(22). The Me ester of GA<sub>12</sub> aldehyde. prepared with  $CH_2H_2$  had m/e (%) 330(M<sup>+</sup>, 7), 287(12), 270(43), 255(15), 241(100), and 239(19). In some experiments, substantial quantities of a second product,  $R_f 0.35$ , were obtained. Elution of this band with EtOAc and crystallization of the recovered gum from EtOAc-PE gave ent-68-hydroxy-78-tosyloxykaur-16en-19-oic acid (XVII), m.p. 161 5-162° (with decomp.) (Found: C. 66.3; H. 7.6; S. 6.3, C<sub>27</sub>H<sub>34</sub>O<sub>6</sub>S requires C, 66.4; H, 7.4; S, 6.5%; y<sub>max</sub> 3550, 1745, 1655, 1599 and 898 cm<sup>-1</sup>; 7 8.93 (3H, s, 20-H), 8.90 (3H, s, 18-H), 7.57 (3H, s, Ar-CH<sub>3</sub>), 5.60 (1H, br, 6-H), 5.47 (1H, d, J 3Hz, 7-H), 5.21 (2H, br, 17-H<sub>2</sub>), 2.74 (2H, d, J 8 Hz, ArH) and 2.21 (2H, d, J 8 Hz, ArH). The methyl ester (XVIII) crystallized from EtOAc-PE with m.p. 179.5-180° (lit.3,13 m.p. 190-192° and 186-188°); vmax 3365, 3040, 1691, 1658, 1598 and 887 cm<sup>-1</sup>; 7 8.95 (6H, s, 18-H and 20-H), 7.59 (3H, s, ArMe), 6.28 (3H, s, OMe), 5.72 (1H, br d, J 2 Hz, 6-H), 5.55 (1H, br d, J 2 Hz, 7-H), 5.22 (2H, br, 17-H<sub>2</sub>), 4.57 (1H, s, 6-OH, exchanged with D<sub>2</sub>O), 2.69 (2H, d, J 8 Hz, ArH) and 2.14 (2H, d, J 8 Hz, ArH). The acetate Me ester (XIX, 30 mg), prepared from the acid (34 mg), pyridine (10 ml) and Ac<sub>2</sub>O (3 ml) at 20° for 24 hr, followed by methylation with CH<sub>2</sub>N<sub>2</sub> had m.p. 177-5-178° (with decomp.) (lit,<sup>3</sup> m.p. 177-178°) with IR and NMR spectra identical to the published spectra.

(b) From ent-6 $\beta$ -hydroxy-7 $\beta$ -tosyloxykaur-16-en-19-oic acid (XVII) and NaH. The tosylate (50 mg), NaH (125 mg; 50% dispersion in oil) and THF (25 ml) were stirred at 20° for 24 hr; H<sub>2</sub>O was added cautiously and the mixture was concentrated *in vacuo*. After the addition of 2 N HCl the product was recovered in EtOAc and purified as in (a) to give GA<sub>12</sub>-aldehyde (16.6 mg), identified by TLC and by GC-MS of the methyl ester.

(c) From ent-6 $\beta$ -hydroxy-7 $\beta$ -tosyloxykaur-16-en-19-oic acid (XVII) and collidine. The tosylate (900 mg) in dry collidine (70 ml) was refluxed for 24 hr under N<sub>2</sub>. After the addition of H<sub>2</sub>O and 2 N HCl, the product was recovered in EtOAc and chromatographed twice on a column of silica. Gibberellin A<sub>12</sub>-aldehyde (297 mg) was eluted with 15–20% EtOAc in PE and identified by TLC and by GLC of the methyl ester.

[6-<sup>2</sup>H]-Gibberellin  $A_{12}$ -aldehyde (Ib). (a) Gibberellin  $A_{12}$ -aldehyde (7 mg), MeO<sup>2</sup>H (0.5 ml) and NaOMe (60 mg) were heated at 110° for 18 hr in a sealed tube. After the addition of <sup>2</sup>H<sub>2</sub>O the solution was concentrated and acidified to pH 2.5. The product was recovered in EtOAc and purified by TLC on silica gel HF with Me<sub>2</sub>CO-PE-AcOH (30:70:1). The band at  $R_f$  0.45 was eluted with EtOAc to give [6-<sup>2</sup>H]-GA<sub>12</sub>- aldehyde (5 mg) containing *ca.* 90%-[<sup>2</sup>H] from the MS of the acid and Me ester; the NMR showed a singlet at 0.32 $\tau$  for 7-H, no 6-H signal at 6.78 $\tau$ . (b) Gibberellin  $A_{12}$ -aldehyde (4 mg), NaOMe (1.3 g), THF (2 ml), distilled from CaH<sub>2</sub>), and <sup>2</sup>H<sub>2</sub>O (1 ml) were boiled for 5 hr. After the addition of saturated aqueous KH<sub>2</sub>PO<sub>4</sub>, the product was recovered in EtOAc and was purified as in (a) to give [6-<sup>2</sup>H]-GA<sub>12</sub>- aldehyde (2 mg) containing *ca.* 90%-[<sup>2</sup>H] by MS.

[6-<sup>3</sup>H]-*Gibberellin*  $A_{12}$ -*aldehyde* (Ic). (a) GA<sub>12</sub>-aldehyde (80 mg), MeO<sup>3</sup>H (0.5 ml, *ca.* 25 mCi) and NaOMe (60 mg) were heated at 100° for 24 hr in a sealed tube. Tritiated water (0.25 ml) was added to the solution which was reduced in volume *in vacuo* and acidified. The product, recovered in EtOAc, was purified by TLC on silica gel with Me<sub>2</sub>CO-PE-HOAc (30:70:1); recovery of the material at  $R_f$  0.6 in EtOAc gave crude [6-<sup>3</sup>H]-GA<sub>12</sub>-aldehyde (56 mg, *ca.* 6100 decomp. min<sup>-1</sup>  $\mu g^{-1}$ ) which was purified by liquid-chromatography as described in (c). (b) GA<sub>12</sub>-aldehyde (230 mg), THF (20 ml, distilled from CaH<sub>2</sub>, NaOMe (2 g) and <sup>3</sup>H<sub>2</sub>O (1 ml, *ca.* mCi) were refluxed under N<sub>2</sub> for 7.5 hr. The reaction mixture was added to satd. aq. KH<sub>2</sub>PO<sub>4</sub> which was extracted with EtOAc. The product was subjected to TLC as in (a) to give crude [6-<sup>3</sup>H]-GA<sub>12</sub>-aldehyde (163 mg, *ca.* 6000 decomp. min<sup>-1</sup>  $\mu g^{-1}$ ) which was further purified as in (c). (c) *Purification by Sephadex chromatography.* To a column (100 × 1.5 cm) of Sephadex LH20 swollen in the aqueous phase of the bi-phasic system PE-EtOAc-AcOH-MeOH-H<sub>2</sub>O (50:15:10:10:2) was added [6-<sup>3</sup>H]-GA<sub>12</sub>-aldehyde (30·6 mg, 202 × 10<sup>6</sup> decomp. min<sup>-1</sup>) in the minimum vol. of the aqueous phase. The column was eluted with the organic phase and 10 ml fractions were collected. [6-<sup>3</sup>H]-Gibberellin A<sub>12</sub>-aldehyde, obtained from fractions 16-23, was crystallized from Me<sub>2</sub>CO-PE to constant radioactivity (5 mg, 6210 decomp. min<sup>-1</sup>  $\mu g^{-1}$ ).

Gibberellin  $A_{12}$  (V). Gibberellin  $A_{12}$ -aldehyde (Ia, 33·3 mg) in acetone (5 ml) was stirred with Jones reagent (0.06 ml) at  $-10^{\circ}$  for 2 hr. MeOH was then added to the solution which was concentrated *in vacuo*, diluted with H<sub>2</sub>O and acidified to pH 2·5. The product recovered in EtOAc was purified by TLC on silica gel HF with EtOAc-PE-AcOH (50:50:1). Elution of the band at  $R_f 0.3$  gave  $GA_{12}$  (16 mg) m.p. 245–247·5° (from Me<sub>2</sub>CO-PE) (lit. m.p. 244–247°), identified by MS of Me ester.<sup>7</sup>

 $[6^{2}H]$ -Gibberellin  $A_{12}$ .  $[6^{2}H]$ -GA<sub>12</sub>-aldehyde (6.8 mg) in acetone (1 ml) was treated with Jones reagent (0.012 ml) as above to give  $[6^{-2}H]$ -GA<sub>12</sub> (6.8 mg). GC-MS of the Me ester showed *ca*. 90% [<sup>2</sup>H<sub>1</sub>]-content.

<sup>13</sup> GALT, R. H. B. and HANSON, J. R. (1965) J. Chem. Soc. 1565.

 $[6^{-3}H]$ -Gibberellin  $A_{12}$ .  $[6^{-3}H]$ -Ga<sub>12</sub> was obtained from a tritiation of GA<sub>12</sub>-aldehyde in which the sealed tube exploded during heating. A portion of the gummy product recovered from the portective metal tube was chromatographed on a column (100 × 1.5 cm) of Sephadex LH20 as described above for  $[6^{-3}H]$ -GA<sub>12</sub>-aldehyde. In addition to  $[6^{-3}H]$ -GA<sub>12</sub>-aldehyde in fractions 16–23,  $[6^{-3}H]$ -GA<sub>12</sub> was recovered from fractions 30–40 and crystallized from Me<sub>2</sub>CO–PE to constant activity (4900 decomp. min<sup>-1</sup>  $\mu$ g<sup>-1</sup>).

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