Accepted Manuscript

Title: Synthesis and pharmacological properties of naturally occurring prenylated and pyranochalcones as potent anti-inflammatory agents



Author: Kongara Damodar Jin-Kyung Kim Jong-Gab Jun

PII:S1001-8417(16)00058-9DOI:http://dx.doi.org/doi:10.1016/j.cclet.2016.01.043Reference:CCLET 3571To appear in:Chinese Chemical LettersReceived date:4-8-2015Revised date:4-12-2015Accepted date:15-1-2016

Please cite this article as: K. Damodar, J.-K. Kim, J.-G. Jun, Synthesis and pharmacological properties of naturally occurring prenylated and pyranochalcones as potent anti-inflammatory agents, *Chinese Chemical Letters* (2016), http://dx.doi.org/10.1016/j.cclet.2016.01.043

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

Graphical Abstract

Synthesis and pharmacological properties of naturally occurring prenylated and pyranochalcones as potent anti-inflammatory agents

Kongara Damodar^a, Jin-Kyung Kim^b, Jong-Gab Jun^{a*}

^aDepartment of Chemistry and Institute of Natural Medicine, Hallym University, Chuncheon 200-702, Korea

^b Department of Biomedical Science, College of Natural Science, Catholic University of Daegu, Gyeungsan-Si 700-702, Korea



A facile approach for the synthesis of naturally occurring prenylated and pyranochalcones is achieved using Claisen-Schmidt condensation as a key step. Subsequently, their anti-inflammatory study revealed that the chalcones bearing the prenyl group at 3- and/or 5-position on ring A (acetophenone moiety), *i.e.*, **1-4** and **7** show weak, or no inhibition activity, whereas chalcones having the prenyl group only on ring B (aldehyde part), *i.e.*, **5**, **6** and **8** show significant activity on the production of inflammatory mediated NO with no cytotoxicity.

Original article

Synthesis and pharmacological properties of naturally occurring prenylated and pyranochalcones as potent anti-inflammatory agents

Kongara Damodar^a, Jin-Kyung Kim^b, Jong-Gab Jun^{a*}

^aDepartment of Chemistry and Institute of Natural Medicine, Hallym University, Chuncheon 200-702, Korea ^bDepartment of Biomedical Science, College of Natural Science, Catholic University of Daegu, Gyeungsan-Si 700-702, Korea

ARTICLE INFO

ABSTRACT

Article history: Received 4 August 2015 Received in revised form 4 December 2015 Accepted 15 January 2016 Available online

Keywords: Prenylated chalcone Pyranochalcone Claisen-Schmidt condensation Anti-inflammatory Nitric oxide (NO) An efficient approach has been developed for the synthesis of naturally occurring prenylated chalcones *viz.* kanzonol C (1), stipulin (2), crotaorixin (3), medicagenin (4), licoagrochalcone A (5) and abyssinone D (6) along with the pyranochalcones paratocarpin C (7), anthyllisone (8) and 3-*O*-methylabyssinone A (9). The key step of the synthesis is a Claisen-Schmidt condensation. Subsequently, their anti-inflammatory effects were investigated in lipopolysaccharides (LPSs)-induced RAW-264.7 macrophages. Of the synthesized chalcones, compounds 5 (IC₅₀= 10.41 µmol/L), 6 (IC₅₀= 9.65 µmol/L) and 8 (IC₅₀= 15.34 µmol/L) show remarkable activity with no cytotoxicity. Compound 9 (IC₅₀= 4.5 µmol/L) exhibits maximum (83.6%) nitric oxide (NO) inhibition, but shows slight cytotoxicity. The results reveal that the chalcones bearing the prenyl group at 3- and/or 5-position on ring A (acetophenone moiety), *i.e.*, **1-4** and **7** show weak, or no inhibition activity, whereas chalcones having the prenyl group only on ring B (aldehyde part), *i.e.*, **5**, **6** and **8** show significant activity on the production of inflammatory mediated NO with no cytotoxicity.

1. Introduction

In multi-cellular organisms, inflammation is an early, protective, homeostatic response of a host against a pathogenic challenge [1] and is indicative of either acute or chronic inflammation. In normal conditions, this process is automatically regulated by the limiting expression levels of pro-inflammatory cytokines, but under pathological conditions, macrophage stimulation leads to an increase of nitric oxide (NO) production. NO, short-lived free radical, can regulate various physiological functions in the cardiovascular, nervous and immune system [2]. Its Endogenous secretion from L-arginine is catalyzed by a family of nitric oxide synthase (NOS) enzymes *viz.* neuronal NOS (nNOS), endothelial NOS (eNOS) and inducible NOS (iNOS). The first two are constitutively expressed and can generate physiologically vital amounts of NO involved chiefly in nerve function and blood regulation whereas the later one (*i.e.* iNOS) produces larger amounts (nano molar) in response to various proinflammatory stimuli. Overproduction of NO causes cell damage because of its highly reactive nature. Therefore, effective control of NO accumulation by iNOS inhibition represents a beneficial therapeutic strategy.

Nonsteroidal anti-inflammatory drugs (NSAIDs) and classical steroidal anti-inflammatory drugs (SAIDs) are currently used to treat acute inflammation. Treatment of chronic inflammation with these SAIDs and NSAIDs is not absolutely successful due to unexpected side effects associated with these developed compounds. Hence, there is a need for the identification and development of safe, effective and novel anti-inflammatory agents.

Bacterial lipopolysaccharides (LPSs) are the major outer surface membrane components present in almost all Gram-negative bacteria and can induce the production of inflammatory mediators including iNOS in diverse eukaryotic species ranging from insects to humans [3]. Therefore, reducing the expression levels of LPS-inducible inflammatory mediators is a promising method to attenuate a variety of disorders derived from inflammation triggered by activated macrophages. RAW 264.7 is a murine macrophage cell line which has been established as an excellent model to screen anti-inflammatory activity of bioactive compounds.

Chalcones, as members of the flavonoid family, are a distinguished class of naturally occurring, bioactive compounds with 1,3diaryl-2-propen-1-one skeleton. They are abundantly present in edible plants and are important precursors in the biosynthesis of flavonoids and isoflavonoids [4]. Primitive therapeutic applications of these plant-related, secondary metabolites can be associated with the thousand-year old use of plants and herbs for the treatment of different medical disorders. These small and non-chiral chemical templates possess a conjugated double bond and an entirely delocalized π -electron system on both benzene rings which gives the

* Corresponding author.

E-mail address: jgjun@hallym.ac.kr

compounds non-linear optical properties [5]. Recently, chalcones have been a subject of great interest around the globe in view of their availability in nature, effortless synthesis, accessible structural modifications and multifarious biological activities. Various natural and non-natural chalcones have been investigated as anti-inflammatory [6], antioxidant [7], anticancer [8], antiprotozoal [9], antimicrobial [10], antiviral [11], antibacterial [12], antihyperglycemic [13], antiplatelet aggregation [14], antiangiogenic [15], antiulcerative [16], antiubercular [17], and antiplasmodial [18] agents. They have also shown inhibitory effects on several enzymes [19].

Continuing our interest on the synthesis and biological evaluation of chalcones as anti-inflammatory agents [20], herein we describe the synthesis of natural prenylated and pyranochalcones using a Claisen–Schmidt condensation as a key step and present the assessment of their anti-inflammatory effects.

Natural prenylated chalcones under the current study *viz*. kanzonol C (1) [21], stipulin (2) [22], crotaorixin (3) [23], medicagenin (4) [24], licoagrochalcone A (5) [25] and abyssinone D (6) [26] were isolated from *Glycyrrhiza eurycarpa*, *Dalbergia stipulacea*, *Crotalaria orixensis*, *Crotalaria medicaginea*, *Glycyrrhiza glabra*, and *Erythrina abyssinica*, respectively (Fig. 1). Pyranochalcones include in this investigation are paratocarpin C (7) [27], anthyllisone (8) [28] and 3-*O*-methylabyssinone A (9) [29] which were isolated from *Paratocarpus Venezosa* Zoll, *Anthyllis hermanniae* and *Lonchocarpus nicou*, respectively (Fig. 1). Synthesis of these interesting compounds was not yet been reported, except for compounds **4** [30] and **5** [31].

2. Experimental

All chemicals were obtained from commercial suppliers and were used without further purification unless noted otherwise. All solvents used for reactions were freshly distilled from proper dehydrating agents under nitrogen gas. All solvents used for chromatography were purchased and directly used without further purification. The ¹H NMR spectra were recorded at Varian Mercury-300 MHz FT-NMR and 75 MHz for ¹³C NMR, with the chemical shift (δ) reported in parts per million (ppm) downfield relative to TMS and the coupling constants (*J*) quoted in Hz. CDCl₃/CD₃OD/CD₃COCD₃ was used as solvent and an internal standard. Mass spectra were recorded using Agilent-5977E spectrometer. Melting points were measured on a MEL-TEMP II apparatus and were uncorrected. Thin-layer chromatography (TLC) was performed on DC-Plastikfolien 60, F₂₅₄ (Merck, layer thickness 0.2 mm) plasticbacked *silica* gel plates and visualized by UV light (254 nm) or staining with *p*-anisaldehyde. Chromatographic purification was carried out using Kieselgel 60 (60-120 mesh, Merck).

2.1 General procedure for Claisen-Schmidt condensation reaction

To a stirred solution of acetophenone (0.25 mmol) and aromatic aldehyde (0.3 mmol, 1.2 equiv.) in MeOH (2 mL) and H₂O (1 mL) was added KOH (0.309 g, 5.5 mmol, 22 equiv.) and the mixture was stirred at room temperature for 72 h. After completion of the reaction, H₂O (25 mL) was added and extracted with EtOAc (3×25 mL). The combined organic layer was washed with brine (2×40 mL), dried over anhydrous Na₂SO₄ and concentrated in vacuo. The crude was purified by column chromatography (EtOAc/hexane = 1/20 to 1/5) to obtain the allyl protected chalcone.

2.2 General procedure for allyl group deprotection

To a stirred solution of di- or tri-allyloxy chalcone (0.15 mmol) in anhydrous MeOH (2.5 mL) were added K_2CO_3 (0.124 g, 0.9 mmol, 6 equiv.) and Pd(PPh_3)_4 (2 mmol%) at room temperature and degassed for 2 min. The reaction mixture was stirred at 60° C for 1-1.5 h. After completion of the reaction, solvent was removed *in vacuo*. H₂O (15 mL) was added to the crude, neutralized with slow addition of 1 mol/L HCl (1.5 mL) at 0 °C and then extracted with EtOAc (3 × 30 mL). The combined organic layer was washed with brine (2 × 40 mL), dried over anhydrous Na₂SO₄ and concentrated *in vacuo*. The crude was purified by column chromatography (EtOAc/hexane = 1/5 to 1/1, v/v) to obtain the pure chalcone.

Physical and spectroscopic characterization data of the compounds described in this article were given in Supporting information.

3. Results and discussion

3.1 Chemistry

Our approach for the synthesis of the chalcones **1-9** is outlined in Schemes 1-4. The synthesis commenced with the prenylation of 2,4-dihydroxyacetophenone (**10**) following the literature procedure [31b] (Scheme 1).

Treatment of compound **10** with BF_3 . Et_2O followed by 2-methyl-but-3-en-2-ol addition at room temperature afforded compounds **11-13**. Subsequently, allyl protection of **10** along with the prenylated acetophenones **11-13** was accomplished with Cs_2CO_3/NaI and allyl bromide in DMF at 60° C and the resulting products **14-17** were each obtained in high yields.

Next, 4-hydroxybenzaldehyde (18) and vanillin (19) were transformed to their prenyl derivatives (Scheme 2). Treatment of 18 and 19 with 3,3-dimethylallyl bromide using 1 mol/L NaOH afforded the aldehydes 20 and 21, respectively. Later, aldehydes 18-21 were reacted with allyl bromide in the presence of Cs_2CO_3/NaI to generate the allyl protected aldehydes 22-25 in high yields, respectively.

Aldehyde part of pyranochalcones 7-9 was obtained from aldehydes 18, 20 and 19 respectively (Scheme 3). Reaction of aldehydes 18-20 with 3-chloro-3-methyl-1-butyne using K_2CO_3/KI and catalytic CuI in acetone under reflux gave compounds 26-28 which were subsequently converted to compounds 29-31, respectively.

Keeping the allyl protected acetophenones (14-17) and aromatic aldehydes (22-25 and 29-31) in hand, next we executed the Claisen-Schmidt condensation (Scheme 4). Treatment of the suitable acetophenone with 1.2 equiv. of aromatic aldehyde using excess KOH in MeOH:H₂O (2:1) at room temperature produced the allyl protected chalcones (32-40) in good to high yields. Finally, allyl group deprotection was achieved using K₂CO₃ and catalytic Pd(PPh₃)₄ (2 mol%) in anhydrous MeOH at 60° C to furnish the desired natural prenylated and pyranochalcones 1-9. All the products 1-9 were structurally confirmed by their spectral (¹H NMR & ¹³C NMR and MS) data.

3.2. Pharmacology

To examine the anti-inflammatory effect of the synthesized chalcones 1-9, we selected an *in vitro* model with murine RAW 264.7 macrophage cell line, which is an established model for anti-inflammatory drug screening. We assayed the ability of compounds 1-9 to decrease NO release after LPS stimulation. The RAW 264.7 cells were incubated with LPS and 1 μ mol/L & 10 μ mol/L concentrations of compounds 1-9 for 24 h. Later, culture media were harvested and the amount of NO was measured. Compounds 1-4 and 7 exhibited weak, or no suppression of NO production at 10 μ mol/L level (Table 1). On the other hand, chalcones 5, 6, 8 and 9 were shown to exhibit a moderate to good level of activity.

Next, cell viability was analyzed to check whether the inhibitory effect was due to cytotoxicity. All of the prepared chalcones displayed no toxic effects, except compound **9**, which showed a slight cytotoxicity at 10 μ mol/L level (Table 1). The IC₅₀ values of these natural chalcones **1-9** were evaluated by using GraphPad Prism 4.0 software and the values were 56.75, 35.40, 41.39, 29.09, 10.41, 9.65, 52.71, 15.34 and 4.50 μ mol/L, respectively. Based upon the results, it can be concluded that chalcones bearing the prenyl group only on ring B (aldehyde part), *i.e.*, **5**, **6** and **8** are fruitful to show good anti-inflammatory activity by effective inhibition of NO production with no cytotoxicity.

4. Conclusion

We have developed a simple and efficient approach for the synthesis of naturally occurring prenylated and pyranochalcones **1-9** using the Claisen-Schmidt condensation as a key step. Later, their *anti*-inflammatory effects were evaluated in lipopolysaccharides (LPSs)-stimulated RAW-264.7 macrophages. Of the chalcones prepared in this study, compounds **5**, **6** and **8** showed remarkable activities with no cytotoxicity. Compound **9** (IC₅₀ = 4.5 μ mol/L) exhibited maximum (83.6%) nitric oxide (NO) inhibition, but showed a slight cytotoxicity. The results revealed that the chalcones bearing the prenyl group at 3- and/or 5-position on ring A (acetophenone part), *i.e.*, **1-4** and **7** showed weak or no inhibition activities, whereas chalcones holding prenyl group only on ring B (aldehyde part), *i.e.*, **5**, **6** and **8** showed significant activities on the production of inflammatory mediated NO with no cytotoxicity.

Acknowledgment

This research was financially supported by Priority Research Centers Program through the National Research Foundation of Korea (NRF) funded by the Ministry of Education, Science and Technology (NRF-2009-0094071).

References

- [1] J. Quintans, Immunity and inflammation: The cosmic view, Immunol. Cell Biol. 72 (1994) 262-264.
- [2] S. Moncada, R.M. Palmer, E. A. Higgs, Nitric oxide: physiology, pathophysiology, and pharmacology, Pharmacol. Rev. 43 (1991) 109-142.
- [3] B. Hinz, K. Brune, Cyclooxygenase-2-10 years Later, J. Pharmacol. Exp. Ther. 300 (2002) 367-375.
- [4] D.I. Batovska, I.T. Todorova, Trends in utilization of the pharmacological potential of chalcones, Curr. Clin. Pharmcol. 5 (2010) 1-29.
- [5] (a) A.K. Singh, G. Saxena, R. Prasad, A. Kumar, Synthesis, characterization and calculated non-linear optical properties of two new chalcones, J. Mol. Str. 1017 (2012) 26-31;

(b) E.D. D'silva, G.K. Podagatlapalli, S.V. Rao, et al., New, high efficiency nonlinear optical chalcone co-crystal and structure-property relationship, Cryst. Growth Des. 11 (2011) 5362-5369.

[7] J.F. Stevens, C.L. Miranda, B. Frei, D.R. Buhler, Inhibition of peroxynitrite-mediated LDL oxidation by prenylated flavonoids: The α,β-unsaturated keto functionality of 2'-hydroxychalcones as a novel antioxidant pharmacophore, Chem. Res. Toxicol. 16 (2003) 1277-1286.

[8] D.K. Mahapatra, S.K. Bharti, V. Asati, Anti-cancer chalcones: Structural and molecular target perspectives, Eur. J. Med. Chem. 98 (2015) 69-114.

[9] F. Lunardi, M. Guzela, A.T. Rodrigues, et al., Trypanocidal and leishmanicidal properties of substitution containing chalcones, Antimicrob. Agents Chemother. 47 (2003) 1449-1451.

[10] M. Ritter, R.M. Martins, D. Dias, M.P. Pereira, Recent advances on the synthesis of chalcones with antimicrobial activities: A brief review, Lett. Org. Chem. 11 (2014) 498-508.

[11] J.Y. Park, H.J. Jeong, Y.M. Kim, et al., Characteristic of alkylated chalcones from Angelica keiskei on influenza virus neuraminidase inhibition, Bioorg. Med. Chem. Lett. 21 (2011) 5602-5604.

[12] S.F. Nielsen, T. Boesen, M. Larsen, et al., Antibacterial chalcones-bioisosteric replacement of the 4'-hydroxy group, Bioorg. Med. Chem. 12 (2004) 3047-3054.

^[6] S.J. Won, C.T. Liu, L.T. Tsao, et al., Synthetic chalcones as potential anti-inflammatory and cancer chemopreventive agents, Eur. J. Med. Chem. 40 (2005) 103-112.

[13] C.T. Hsieh, T.J. Hsieh, M. El-Shazly, et al., Synthesis of chalcone derivatives as potential anti-diabetic agents, Bioorg. Med. Chem. Lett. 22 (2012) 3912-3915.

[14] L.M. Zhao, H.S. Jin, L.P. Sun, et al., Synthesis and evaluation of antiplatelet activity of trihydroxychalcone derivatives, Bioorg. Med. Chem. Lett. 15 (2005) 5027-5029.

- [15] L. Varinska, M. van Wijhe, M. Belleri, et al, Anti-angiogenic activity of the flavonoid precursor 4-hydroxychalcone, Eur. J. Pharmacol. 691 (2012) 125-133.
- [16] K.V. Shashidhara, S.R. Avula, V. Mishra, et al., Identification of quinoline-chalcone hybrids as potential antiulcer agents, Eur. J. Med. Chem. 89 (2015) 638-653.
- [17] F. Macaev, V. Boldescu, S. Pogrebnoi, G. Duca, Chalcone scaffold based antimycobacterial agents, Med. Chem. 4 (2014) 487-493.
- [18] M. Larsen, H. Kromann, A. Kharazmi, S. F. Nielsen, Conformationally restricted anti-plasmodial chalcones, Bioorg. Med. Chem. Lett. 15 (2005) 4858-4861.
- [19] (a) O. Nerya, R. Musa, S. Khatib, et al., Chalcones as potent tyrosinase inhibitors: the effect of hydroxyl positions and numbers, Phytochemistry 65 (2004) 1389-1395;

(b) S. Iwata, N. Nagata, A. Omae, et al., Inhibitory effect of chalcone derivativess on recombinant human aldose reductase, Biol. Pharm. Bull. 22 (1999) 323-325.

[20] (a) S.J. Kim, C.G. Kim, S.R. Yun, et al., Synthesis of licochalcone analogues with increased anti-inflammatory activity, Bioorg. Med. Chem. Lett. 24 (2014) 181-185;

(b) J.H. Jeon, M.R. Kim, J.G. Jun, Concise synthesis of licochalcone A through water-accelerated [3,3]-sigmatropic rearrangement of an aryl prenyl ether, Synthesis (2011) 370-376.

- [21] T. Fukai, J. Nishizawa, T. Nomura, Five isoprenoid-substituted flavonoids from Glycyrrhiza eurycarpa, Phytochemistry 35 (1994) 515-519.
- [22] P. Bhatt, R. Dayal, Stipulin, a prenylated chalcone from *Dalbergia stipulacea*, Phytochemistry 31 (1992) 719-721.
- [23] T. Narender, S.K. Tanvir, M.S. Rao, et al., Prenylated chalcones isolated from *Crotalaria genus* inhibits in vitro growth of the human malaria parasite *Plasmodium falciparum*, Bioorg. Med. Chem. Lett. 15 (2005) 2453-2455.
- [24] G.V.R. Rao, P.S. Rao, K.R. Raju, A prenylated chalcone from Crotalaria medicaginea, Phytochemistry 26 (1987) 2866-2868.
- [25] Y. Asada, W. Li, T. Yoshikawa, Isoprenylated flavonoids from hairy root cultures of Glycyrrhiza glabra, Phytochemistry 47 (1998) 389-392.
- [26] L. Cui, P.T. Thuong, H.S. Lee, et al., Four new chalcones from Erythrina abyssinica, Planta Med. 74 (2008) 422-426.
- [27] Y. Hano, N. Itoh, A. Hanaoka, et al.; Paratocarpins A E, five new isoprenoid-substituted chalcones from Paratocarpus venenosa Zoll., Heterocycles 41 (1995) 191-198.
- [28] L. Pistelli, K. Spera, G. Flamini, et al., Isoflavonoids and chalcones from Anthyllis hermanniae, Phytochemistry 42 (1996) 1455-1458.
- [29] M.A. Lawson, M. Kaouadji, A.J. Chulia, A single chalcone and additional rotenoids from *Lonchocarpus nicou*, Tetrahedron Lett. 51 (2010) 6116-6119.
 [30] (a) G.V. Rao, B.N. Swamy, V. Chandregowda, G.C. Reddy, Synthesis of (±)-abyssinone I and related compounds: Their anti-oxidant and cytotoxic activities, Eur. J. Med. Chem. 44 (2009) 2239-2245;

(b) A. Maiti, M. Cuendet, V.L. Croy, et al.; Synthesis and biological evaluation of (±)-abyssinone II and its analogues as aromatase inhibitors for chemoprevention of breast cancer, J. Med. Chem. 50 (2007) 2799-2806.

[31] (a) H.-M. Wang, L. Zhang, J. Liu, et al., Synthesis and anti-cancer activity evaluation of novel prenylated and geranylated chalcone natural products and their analogs, Eur. J. Med. Chem. 92 (2015) 439-448;

(b) N. Tadigoppula, V. Korthikunta, S. Gupta, et al., Synthesis and insight into the structure-activity relationships of chalcones as antimalarial agents, J. Med. Chem. 56 (2013) 31-45.



- 1 R^1 , R^3 = prenyl, R^2 , R^4 = H (kanzonol C)
- 2 R^1 , R^4 = prenyl, R^2 , R^3 = H (stipulin)
- 3 $R^1 = OMe$, $R^3 = prenyl$, R^2 , $R^4 = H$ (crotaorixin)
- 4 R^1 , $R^2 = H$, R^3 , $R^4 = prenyl$ (medicagenin)
- 5 R^1 = prenyl, R^2 , R^3 , R^4 = H (licoagrochalcone A)
- 6 R^1 = prenyl, R^2 = OMe, R^3 , R^4 = H (abyssinone D)

Fig. 1. Structures of naturally occurring prenylated and pyranochalcones (1-9).



Scheme 1. Synthesis of prenyl substituted acetophenones. Reagents and conditions: (a) 2-methyl-but-3-en-2-ol, BF₃.Et₂O, 1,4-dioxane, r.t., 1 h. (b) allyl bromide, Cs₂CO₃/NaI, DMF, 60 °C, 5 h.



- 7 $R^1 = H, R^2 = prenyl$ (paratocarpin C)
- 8 R^1 = prenyl, R^2 = H (anthyllisone)
- 9 $R^1 = OMe, R^2 = H$ (3-*O*-methylabyssinone A)



Scheme 2. Synthesis of prenyl substituted benzaldehydes. Reagents and conditions: (a) 3,3-dimethylallyl bromide, 1 mol/L NaOH, 0° C – r.t., 2 h. (b) allyl bromide, Cs₂CO₃/NaI, DMF, 60° C, 2 h.



reflux, 4.5 h. (b) N, N-diethylaniline, sealed tube, 210 °C, 1.5 h.



Scheme 4. Synthesis of naturally occurring prenylated and pyranochalcones. Reagents and conditions: (a) KOH, MeOH/H₂O (2:1), r.t., 72 h, 58%-90%. (b) K₂CO₃, Pd(PPh₃)₄, MeOH, 60 $^{\circ}$ C, 1-1.5 h, 51%-89%.

Table 1.

Anti-inflammatory activities of naturally occurring prenylated and pyranochalcones 1-9.

Compd	NO Production (% inhibition) ^{a,b}		Proliferation ^c		IC ₅₀ (µmol/L)
	1 µmol/L	10 µmol/L	I µmol/L	10 µmol/L	
Medium	$1.13 \pm 0.87 \ (98.87)^{***}$	1.13 ± 0.87 (98.87)***	99.77 ± 4.23	99.77 ± 4.23	
1	131.54 ± 2.6 (-31.54) ***	124.48 ± 10.29 (-24.48)***	112.35 ± 15.75	93.56 ± 5.91	56.75
2	124.14 ± 1.43 (-24.14) ***	103.13 ± 5.1 (-3.13)	108.03 ± 11.72	$157.63 \pm 8.06*$	35.40
3	124.98 ± 6.49 (-24.98) ***	99.72 ± 11.36 (0.28)	112.35 ± 15.75	113.36 ± 8.80	41.39
4	116.27 ± 2.95 (-16.27) ***	82.26 ± 0.8 (17.74)***	108.03 ± 9.41	99.25 ± 2.71	29.09
5	125.30 ± 4.79 (-25.3)***	53.58 ± 2.41 (46.42)***	99.24 ± 0.63	111.37 ± 11.13	10.41
6	120.10 ± 2.03 (-20.1)***	39.59 ± 4.31 (60.41)***	99.16 ± 3.26	103.85 ± 2.0	9.65
7	124.2 ± 2.91 (-24.2)***	129.9 ± 2.52 (-29.9)***	105.45 ± 6.96	97.43 ± 2.22	52.71
8	119.20 ± 0.64 (-19.2)***	58.09 ± 0.94 (41.91)***	108.47 ± 2.86	98.24 ± 10.12	15.34
9	107.21 ± 3.51 (-7.21)**	16.45 ± 0.89 (83.55)***	106.66 ± 5.82	$92.74 \pm 1.40*$	4.50
LPS	$99.99 \pm 3.76 (0.01)$	$99.99 \pm 3.76 (0.01)$			

^a The results are reported as mean value \pm SEM for n = 3. Statistical significance is based on the difference when compared with LPS-treated groups (**P < 0.01, ***P < 0.001).

^b Inhibition is based on LPS as shown in parenthesis.

^cThe results are reported as mean value \pm SEM for n = 3. Statistical significance is based on the difference when compared with Medium groups (*P < 0.05).